

**QUT SCHOOL OF CHEMISTRY AND PHYSICS  
PVB303 NUCLEAR AND PARTICLE PHYSICS**

**PRACTICAL 2: SECULAR EQUILIBRIUM**

**SEMESTER 2, 2020**

**Aims:**

To investigate the phenomenon of secular equilibrium and to model radioactive decay processes in a physically realistic way.

Due to COVID-19 restrictions, this prac will be computer-based. You will use a first-principles model of the underlying physical process (as opposed to a mathematical solution) to simulate experimental datasets that would ordinarily be measured in the lab. You will then fit the datasets and determine the apparent decay rate constants.

**Theory:**

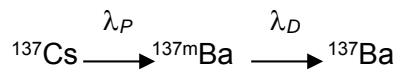
1. Secular equilibrium

Secular equilibrium results when:

- (1) a parent isotope decays to a radioactive daughter isotope, and
- (2) the half-life time  $T_D$  of the daughter is significantly shorter than that of the parent,  $T_P$  (e.g.  $T_P > 10000 T_D$ ).

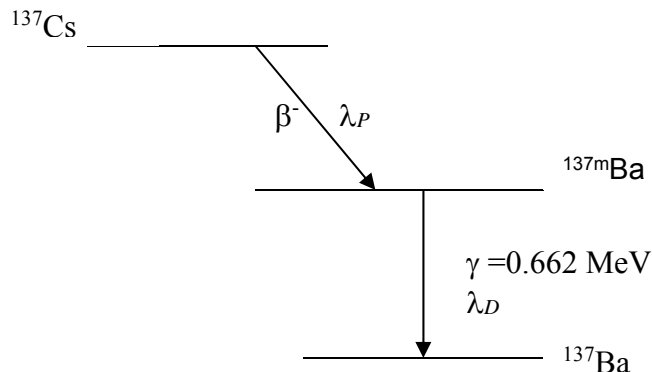
In this situation the production of the daughter isotope is slow, while the decay of the newly produced daughter is fast. As a result, a steady-state (secular-equilibrium) concentration of the daughter isotope is quickly established.

The decay



provides an example of a system where secular equilibrium occurs. The half-life times of  $^{137}\text{Cs}$  and  $^{137\text{m}}\text{Ba}$  are  $T_P \sim 30\text{y}$  and  $T_D \sim 153\text{s}$ , respectively. As a result, the production of  $^{137\text{m}}\text{Ba}$  is very slow, but the decay of the newly produced  $^{137\text{m}}\text{Ba}$  is fast; and the concentration of  $^{137\text{m}}\text{Ba}$  in a  $^{137}\text{Cs}$  sample quickly reaches a steady-state value.

The energy level scheme representing this decay is



where the decay rate constants  $\lambda$  are related to the respective half-life times  $T$  as  $\lambda = (\ln 2)/T$ . The instantaneous rates of decay or production of the three nuclides follow first-order kinetics:

$$\begin{aligned}\frac{d[^{137}\text{Cs}]}{dt} &= -\lambda_P [^{137}\text{Cs}] \\ \frac{d[^{137m}\text{Ba}]}{dt} &= \lambda_P [^{137}\text{Cs}] - \lambda_D [^{137m}\text{Ba}] \quad \dots (1) \\ \frac{d[^{137}\text{Ba}]}{dt} &= \lambda_D [^{137m}\text{Ba}]\end{aligned}$$

where [...] denotes the concentration (or the absolute amount) of the respective nuclide.

(1) At long times ( $t \gg T_D$ ), a time-independent steady-state concentration of  $^{137m}\text{Ba}$  is established, where the rate of its production equals the rate of decay:

$$\frac{d[^{137m}\text{Ba}]}{dt} = \lambda_P [^{137}\text{Cs}] - \lambda_D [^{137m}\text{Ba}] = 0 \quad \dots (2)$$

From Eq. (2), the steady-state concentration of  $^{137m}\text{Ba}$  can be found:

$$[^{137m}\text{Ba}]_{\text{sec}} = \frac{\lambda_P}{\lambda_D} [^{137}\text{Cs}] \quad \dots (3)$$

This is the condition of secular equilibrium between the two radioactive species. Experimentally, this corresponds to a  $^{137}\text{Cs}/^{137}\text{Ba}$  sample that was prepared sufficiently long ago to allow the steady state to be established. In this regime, the activity of  $^{137m}\text{Ba}$  will be ideally constant when measured over a period of time that is significantly shorter than  $T_P$ .

(2) The other possible experimental scenario is a fresh sample of the parent isotope  $^{137}\text{Cs}$ , where the concentration of the daughter isotope ( $^{137m}\text{Ba}$ ) is initially zero. In this situation the concentration of  $^{137m}\text{Ba}$  is time-dependent and builds up from the initial value to the steady-state value. Provided that  $\lambda_D \gg \lambda_P$  and the concentration of  $^{137}\text{Cs}$  changes very slowly, the kinetics of this build-up is described by the equation

$$[^{137m}\text{Ba}](t) = \frac{\lambda_P}{\lambda_D} [^{137}\text{Cs}] (1 - e^{-\lambda_D t}) \quad \dots (4)$$

At  $t \gg T_D$ ,  $[^{137m}\text{Ba}]$  approaches the secular equilibrium value given by Eq. (3).

(3) If the  $^{137}\text{Cs}$  sample initially contains some  $^{137m}\text{Ba}$  (but less than the secular-equilibrium amount), then Eq. (4) needs to be modified to take that into account. In this situation the kinetics of the build-up of  $[^{137m}\text{Ba}]$  is described by the equation

$$[^{137m}\text{Ba}](t) = [^{137m}\text{Ba}]_0 e^{-\lambda_D t} + \frac{\lambda_P}{\lambda_D} [^{137}\text{Cs}](1 - e^{-\lambda_D t}) \quad \dots (5)$$

where  $[^{137m}\text{Ba}]_0$  is the initial concentration of the daughter isotope. According to Eq. (5), at  $t = 0$   $[^{137m}\text{Ba}] = [^{137m}\text{Ba}]_0$ , while at  $t \gg T_D$   $[^{137m}\text{Ba}]$  approaches the secular equilibrium value. The two assumptions from scenario (2) are also implicit in Eq. (5), i.e.  $\lambda_D \gg \lambda_P$  and that the concentration of  $^{137}\text{Cs}$  changes very slowly.

(4) If  $^{137m}\text{Ba}$  is isolated from the parent isotope, then the rate of its fresh production becomes zero and it only undergoes decay. In this situation its kinetics is described by a decaying exponent:

$$[^{137m}\text{Ba}](t) = [^{137m}\text{Ba}]_0 e^{-\lambda_D t} \quad \dots (6)$$

where  $[^{137m}\text{Ba}]_0$  is the initial concentration of the daughter isotope.

## 2. Modeling of radioactive decay

Mathematically, radioactive decay is a probabilistic process: the decay of a given individual nucleus is driven by probabilistic variables rather than a deterministic mathematical function. In addition to this, the detection process itself is not 100% efficient: there is a certain probability (usually significantly less than 1) that a given act of decay would be successfully detected. Furthermore, the detector can make a false-positive reading whereby a detection is made when no decay had actually occurred; this is most typically due to hardware noise. A physically realistic modeling of the measurement process needs to consider all of these factors. A Matlab code implementing the simulation procedure is given in the Appendix.

Suppose that a single nucleus of a radioactive species has the probability  $p_1$  to decay and the probability  $q_1 = 1 - p_1$  to survive over a time step  $\Delta t$ . For an elementary first-order process with the rate constant  $\lambda$ ,  $p_1 = 1 - \exp(-\lambda \Delta t)$ . Then, in an ensemble of  $N$  such nuclei, the probability of exactly  $Y$  nuclei decaying during  $\Delta t$  is given by a binomial distribution:

$$P_1(Y) = p_1^Y (1 - p_1)^{(N-Y)} \frac{N!}{Y! (N-Y)!} \quad \dots (7)$$

Here  $0 \leq Y \leq N$ , i.e. in principle any number of nuclei between 0 and  $N$  could decay within a single time step. However, for a large  $N$  and small  $p_1$ , the probability of  $Y$  being close to  $N$  is vanishingly small, and on average  $\langle Y \rangle = p_1 N$  nuclei will decay each time step  $\Delta t$ . In most time steps the number of decays will not be exactly  $\langle Y \rangle$  but would deviate from the average number according to the binomial distribution  $P_1$ .

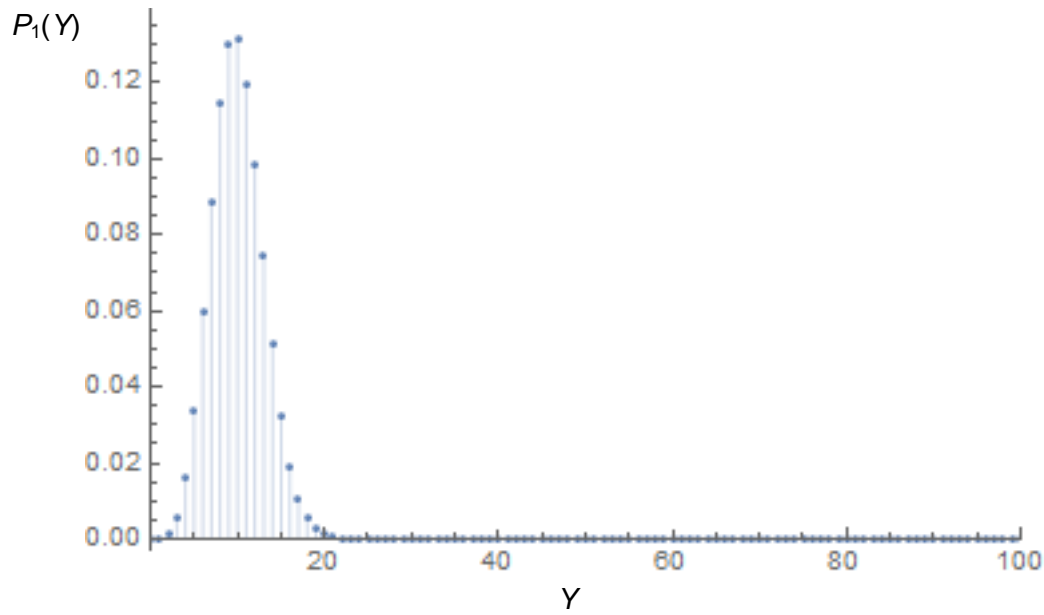
For computational efficiency, the Matlab code in the Appendix approximates the binomial distributions of  $^{137}\text{Cs}$  and  $^{137m}\text{Ba}$  decay numbers by the respective Gaussian distributions when  $N$  is large:

```
decays1 = double(ncs)*p1 + normrnd(0, sqrt(double(ncs)*p1*(1-p1)))
decays2 = double(nba)*p2 + normrnd(0, sqrt(double(nba)*p2*(1-p2)))
```

(The binomial-variate function `binornd(j,z)` is very slow when  $j$  is large, but a Gaussian variate provides a very good approximation of the binomial distribution in this limit.)

When  $N$  is small, a true binomial distribution is used:

```
decays2 = binornd(double(nba),p2)
```



Probability distribution  $P_1$  for  $N = 100$  and  $p_1 = 0.1$ . The extreme values  $Y = 0$  and  $Y > 25$  are highly improbable. On average  $100 \times 0.1 = 10$  nuclei would decay in each time step in this scenario. However, the number of decays will be exactly 10 only in  $\sim 13\%$  of the time steps; 95% of the time the number of decays in a single time step will be a binomially distributed random number between 5 and 16.

Once a decay has occurred, it may or may not be picked up by the detector. The detectors you are using in the pracs have detection efficiency  $f \sim 2\%$  - i.e., on average they detect only 2% of the decaying nuclei. The distribution of true-positive detections is also a binomial distribution: If  $Y$  decays are known to have occurred, then the number of true-positive detected events could be anywhere between 0 and  $Y$ , with the average value being  $fY$ . In the Matlab code in the Appendix, this is represented by the line

```
detector = binornd(round(decays2),eff)
```

Besides true positives, the detector inevitably generates some false-positive readings. The distribution of the number of false positives can be complicated because there may be multiple physical mechanisms by which a false-positive reading could arise. Here, for the sake of simplicity we will assume this distribution also to be binomial and empirically fine-tune its shape to match the experimental observations. In the Matlab code in the Appendix, the number of false positives is represented by the term

```
binornd(round(dt*noise),0.7)
```

## Objectives:

- (1) First, you will investigate the decay of  $^{137m}\text{Ba}$  isolated from its parent isotope. Simulate the kinetics of the decay and determine (by least-squares fitting) the value of  $\lambda_D$  as if you were analysing the results of an experimental measurement.
- (2) You will then simulate the transient buildup of  $^{137m}\text{Ba}$  to the secular-equilibrium value and again determine  $\lambda_D$  by least-squares fitting.
- (3) Compare the two "measured" values of  $\lambda_D$  and discuss the results.

## Experimental notes:

- 1) A NaI [ TI ]  $\gamma$ -ray detector used to measure the activity of a  $^{137}\text{Cs}$  source is in fact measuring the decay of  $^{137m}\text{Ba}$  since  $^{137}\text{Cs}$  emits beta particles which are not detected by the NaI detector.
- 2) The count rate (C) is proportional to activity of  $^{137m}\text{Ba}$ . Hence Eq. (4) may be written as

$$C_D(t) = C_{D_{\text{sec}}} (1 - e^{-\lambda_D t}) \quad \dots (8)$$

- 3)  $\text{Cs}^+$  and  $\text{Ba}^{2+}$  can be separated due to the differences in their chemical properties: one is a monovalent ion, while the other is divalent. The separation can be achieved by eluting the sample through an ion-exchange resin located within the Cs/Ba generator:  $\text{Ba}^{2+}$  is eluted, while  $\text{Cs}^+$  is retained by the resin. The Cs/Ba generator used in this practical can therefore be used to provide a "fresh" sample of  $^{137}\text{Cs}$ . However, the separation is not 100% efficient and in practice there will always be some  $^{137m}\text{Ba}$  present in the  $^{137}\text{Cs}$  sample at  $t = 0$ .

## Procedure: Simulation and Analysis

You will need MatLab for the simulations. MatLab is available for free to QUT students and can be installed on your personal (non-QUT) computer. Download it from QUT IT Help and install it on your computer. Alternatively, you can use an already-installed copy of MatLab at one of the SEF computer labs.

1. Set the initial amounts of both  $^{137}\text{Cs}$  and  $^{137m}\text{Ba}$  to zero: `n0cs = uint64(0)` and `n0ba = uint32(0)`. Run the MatLab simulation. This will generate the detector noise, i.e. readings that contain only false positives. Save this in Excel-readable CSV files.
2. Set the initial amounts of both  $^{137}\text{Cs}$  and  $^{137m}\text{Ba}$  to the values corresponding to the sample in a steady-state equilibrium: `n0cs = uint64(5e14)` and `n0ba = uint32(5e14*lambda_p/lambda_d)`. Run the MatLab simulation. Save the results in CSV files.
3. Emulate the decay of  $^{137m}\text{Ba}$  in the eluate, i.e. the decay of the daughter  $^{137m}\text{Ba}$  nuclide separated from the parent  $^{137}\text{Cs}$ : Set `n0cs = uint64(0)` and `n0ba = uint32(5e14*lambda_p/lambda_d)`, and run the MatLab simulation. Save the resulting decay curve in CSV files.

- Emulate the buildup of  $^{137m}\text{Ba}$  in a fresh  $^{137}\text{Cs}$  sample initially containing a small amount of  $^{137m}\text{Ba}$ : Set `n0cs = uint64(5e14)` and `n0ba = uint32(1e7)`, and run the MatLab simulation. Save the resulting buildup curve in CSV files.
- Plot the results of Step 3 on In-linear graph paper (or as a In-linear plot in MatLab) and determine the half-life time and the decay rate constant of  $^{137m}\text{Ba}$ .
- To plot the results of Step 4, you need to take into account the initial non-zero amount of  $^{137m}\text{Ba}$  in the fresh sample of  $^{137}\text{Cs}$ :

$$C_D(t) = C_{Dsec}(1 - e^{-\lambda_D t}) + x C_{Dsec} e^{-\lambda_D t} \quad \dots\dots\dots (9)$$

Eq. (9) is equivalent to Eq. (5).  $C_{Dsec}$  is the secular-equilibrium concentration of  $^{137m}\text{Ba}$  given by Eq. (3), and  $x$  is the fraction of the daughter nuclide present at  $t = 0$ :  $x = C_D(0)/C_{Dsec}$ . Estimate the appropriate values of  $C_{Dsec}$  and  $x$  based on the parameter values used in your simulations.

Once an estimate of  $x$  is obtained, Eq. (9) can be linearised as

$$\ln \left[ \frac{C_D(t) - C_{Dsec}}{C_{Dsec}} \times \frac{1}{x - 1} \right] = -\lambda_D t \quad \dots\dots\dots (10)$$

Plot your data from Step 4 on In-linear graph paper (or as a In-linear plot in MatLab) according to Eq. (10). [Caution: Do NOT use the code-generated file `Detector_Log.csv` for this purpose.] Determine the decay rate constant of  $^{137m}\text{Ba}$ . Compare it with the value measured in Step 5.

Your report must include the following:

- A brief background.
- A description of the simulation performed and the data analysis.
- The plots of the four simulated timecourses: noise, steady-state, the decay curve and the buildup curve of  $^{137m}\text{Ba}$ . Each dataset should be shown both in the original coordinates (counts vs time) and in the Log-linear coordinates,  $\ln(\text{counts})$  vs time.
- Least-squares fitting of the decay and the build-up, including a clear description of the model and the fitting parameters used, and the corresponding apparent decay rate constants.
- A discussion of the curves obtained, uncertainties of the measurements and sources of error and the degree of agreement of the two measured values of  $\lambda_D$ .
- Discuss whether the secular equilibrium model is applicable to the Cs/Ba system.
- Compare the levels of noise in the noise-only timecourse and the steady-state timecourse (Steps 1 and 2, respectively). Are the two noise levels equal or not? What is the explanation?
- Your least-squares fitted values of  $\lambda_D$  will most likely differ from each other as well as from the input value of  $\lambda_D$  (variable `lambda_d` in the MatLab code). Why is this?
- Eq. (6) being a monoexponential function suggests that the decay of  $^{137m}\text{Ba}$  (Step 3) should be a simple straight line in In-linear coordinates. However, the In-linear plot in Step 5 is curved upwards at long times. Why is this? Is this consistent with the experimental decay dataset on BB? What are the implications for the sensitivity of the experimental measurement?
- In the In-linear plot from Step 5, the scatter of data points increases as the time increases. Why is this?

## APPENDIX: MatLab simulation code

The following code simulates, in a physically realistic way, the two-step decay sequence  $^{137}\text{Cs} \rightarrow ^{137\text{m}}\text{Ba} \rightarrow ^{137}\text{Ba}$  as well as the detection process for the  $^{137\text{m}}\text{Ba}$  decays. (See the manual section "Modeling of radioactive decay" for the description of the simulation procedure.)

The initial amounts of  $^{137}\text{Cs}$  and  $^{137\text{m}}\text{Ba}$  are the values of the variables `n0cs` and `n0ba`, respectively.

The output (the decay curve) is saved in the arrays:

`time` (list of time steps)

`detector` (the number of  $^{137\text{m}}\text{Ba}$  decays in a given time step)

`logdet` (ln of the number of decays)

Set the variables `n0cs` and `n0ba` to the appropriate values in order to simulate the four measurements described under the Procedure.

```
dt = 1;           % Time step = 1 s; you can change this if required
nt = 2000;        % Keep simulation time dt*nt = 2000 s

tp = 30.17*525948.766*60; % T1/2 for 137Cs (in s)
lambda_p = log(2)/tp;    % Decay rate constant for 137Cs (s^-1)
td = 2.552*60;           % T1/2 for 137mBa (in s)
lambda_d = log(2)/td;    % Decay rate constant for 137mBa (s^-1)
n0cs = uint64(5e14);      % Initial number of 137Cs nuclei
ncs = n0cs               % Current number of 137Cs nuclei
p1 = 1 - exp(-lambda_p*dt); % Probability of decay of a given 137Cs nucleus per time step

n0ba = uint32(5e14*lambda_p/lambda_d) % Initial number of 137mBa nuclei
nba = n0ba;                 % Current number of 137mBa nuclei
p2 = 1 - exp(-lambda_d*dt); % Probability of decay of a given 137mBa nucleus per time step
eff = 0.02;                 % Detector efficiency (2%)
pnoise = 0.8;               % Empirical noise parameters
noise = 10/pnoise;

times = [dt:dt:nt*dt];      % Array with the time axis values
detector = 0*times;         % Array with detector reads for each time step
logdet = detector;          % Log (ln) of the detector reads

for ii = 1:nt
    % The number of 137Cs decays in the current time step.
    % Matlab's binomial RN generator can't handle very large integers,
    % so we approximate a binomial variate by using Gaussian noise.
    % Since the number of decays must be non-negative,
    % any negative values are replaced with 0.
    decays1 = double(ncs)*p1 + normrnd(0, sqrt(double(ncs)*p1*(1-p1)));
    decays1 = max(decays1, 0);
    % Each 137Cs decay decreases the number of 137Cs nuclei
    % and increases the number of 137mBa nuclei:
    ncs = ncs - decays1
    nba = nba + decays1;

    % The number of 137mBa decays in the current time step.
```

```

% For 137mBa decay we use a true binomial variate when nba is small,
% otherwise use a Gaussian approximation (which is much faster):
if (nba < 10000)
    decays2 = binornd(double(nba),p2);
else
    decays2 = double(nba)*p2 + normrnd(0, sqrt(double(nba)*p2*(1-p2)));
    decays2 = max(decays2, 0);
end % End of If
% Each 137mBa decay decreases the number of 137mBa nuclei:
nba = nba - decays2
ii % Lazy progress indicator

% Detector reads include true positives and false positives:
detector(ii) = binornd(round(decays2),eff) + binornd(round(dt*noise),0.7);
logdet(ii) = log(detector(ii));
end % End of For

% Save the results: Change the directory and file names as appropriate
csvwrite("C:\KM\Classes\PVB303\Pracs\times.csv", transpose(times));
csvwrite("C:\KM\Classes\PVB303\Pracs\Detector.csv", transpose(detector));
csvwrite("C:\KM\Classes\PVB303\Pracs\Detector_Log.csv", transpose(logdet));

```