# **Characteristics Of Enols and Enolates**

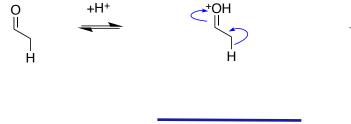
from chapter(s) \_\_\_\_\_ in the recommended text

#### A. Introduction

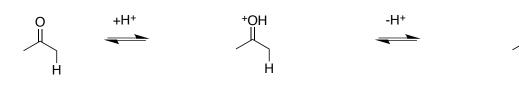
#### **B. Enols Form Under Acidic Conditions**

#### **Mechanism Of Formation**

protonation enol.

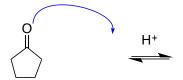






enol form

enol



enol form

## **Deuterium Exchange**

deuterons.

deuterated carbonyl

deuterated product

O-deuterated carbonyl

O-deuterated enol

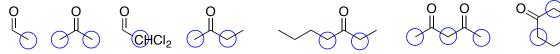


C-deuterated ketone

O-deuterated carbonyl

enol form

dideuterated ketone













tautomerism;

## **Enols Of 1,2- And 1,3-Dicarbonyl Compounds** enol form

10<sup>6</sup>: 1. enol

reason: Enolization of 1,3-cyclohexanedione forms conjugation between carbonyl and C=C which is stabilized by resonance, while the acetone does not have resonance effect.

reason: Compared to acetone, one carbonyl group in the 2,3-butanedione acts as electron withdrawing group that enhances acidity of  $\alpha$ -H.

$$0 \downarrow 0 \downarrow 0 \downarrow 0 \downarrow 0$$

## **Enols Of Other Carbonyl Compounds**

$$H_2O$$

enol

## **Keto-Enol Tautomers Of Other Compound Types**

keto

### C. Enolates Form Under Basic Conditions

#### **Mechanism Of Formation**

enolate enol

#### **Resonance Structures Of Enolates**

more more

It is not easy *N*-anions.

#### D. Effects Of Enolization

#### **Racemization**

#### optically active

#### achiral

optically active

#### can racemize

## **Double Bond Migration**

### Migration

enol

keto

keto

enol

fructose