		Γ.	acetate	CH ₃ COO	- TABL	TABLE OF POLYATOMIC IONS				oxalate $C_2O_4^{2-}$			PERIODIC TABLE OF IONS						
		- 1	arsenate	AsO ₄ ³⁻		lrogen phos	sphate	H ₂ PO ₄ -	perchlorate			1 LIVI	JDIO	ши	101	10111	,		
			arsenite	AsO ₃ ³ -		ogen carbo		HCO ₃	periodate	IO ₂			KEY						
			benzoate	C ₆ H ₅ COC		ogen oxala		HC_2O_4	permangan			atomic_ number	≥ 26 г.	3+ charge	<u>.</u>				
		- 1	borate	BO ₃ ³ -	1 3	ogen sulfa		HSO ₄	peroxide	O_2^2		number	iron (I	ion		4=	40		
1		Ι.	bromate	BrO ₃		ogen sulfic		HS ⁻	phosphate	PO		1.				17	18		
1			carbonate	CO_3^{2-}		ogen sulfit		HSO ₃	pyrophosph			symbo	I IC	Z+ (IUPAC)		1	2		
H ⁺			chlorate	ClO_3^-		ogen sum oxide	·C	OH-	sulfate	SO			iron (I	1)		H-	Не		
hydrogen	2		chloride	Cl ⁻		chlorite		ClO-	sulfite	SO		13	14	15	16	hydride	helium		
3	4	- 1	chlorite	ClO ₂ -	ioda			IO ₃ -	thiocyanate			5	6	7	8	9	10		
Li ⁺	Be^{2+}	- 1	chromate	CrO ₄ ² -	I	hydrogen pho	snhate	HPO ₄ ²⁻	thiosulfate	S_2		В	C	N3-	O^{2-}	F-	Ne		
lithium	beryllium		cyanate	CNO-	nitra	0 1	Spriace	NO ₃	POSITIVE POL			boron	carbon	nitride	oxide	fluoride	neon		
	Ů		cyanide	CN-	nitri			NO_2^-	ammonium				<u> </u>						
11	12		dichromate		I	osilicate		SiO ₄ ⁴	hydronium	H ₃ (13	14	15	16	17	18		
Na ⁺				01207	Ortin	Jameate		5104	nyaromam	113	9	Al^{3+}	Si	P ³⁻	S ²⁻	Cl-	Ar		
sodium	magnesium	3	4	5	6	7	8	9	10	11	12	aluminum	silicon	phosphide	sulfiide	chloride	argon		
19	20	21	²² Ti ⁴	+ 23 V3+	24 Cr3+	25 Mn ²⁺	²⁶ Fe	3+ 27 Co	$^{2+}$ 28 $^{2+}$ $^{2+}$	²⁹ Cu ²	₊ 30	31	32	33	34	35	36		
K+	Ca ²⁺	Sc^3	+ titanium (I	/) vanadium(Ⅲ)	chromium (III)	manganese(II)	iron (I	II) cobalt (copper (I	$2 \operatorname{Zn}^{2+}$	Ga ³⁺	$\mathrm{Ge^{4+}}$	As ³⁻	Se ²⁻	Br-	Kr		
potassium	calcium	scandi			Cr^{2+}	Mn ⁴⁺	Fe			Cu ⁺		gallium	germanium	l .	l	bromide	I. I		
potassium	Calcium	Scandi	titanium (II			manganese(IV)	iron (I	I) cobalt (III) nickel (III)	copper (l) Zinc	gamum	germanium	arseniue	Scieniuc	bronnae	Mypton		
37	38	39	40	⁴¹ Nb ⁵⁺	42	43	44 Ru	3+ 45	46 Pd ²⁺	47	48	49	50 Sn ⁴⁺	⁵¹ Sb ³⁺	52	53	54		
Rb ⁺	Sr ²⁺	Y ³ -	+ Zr ⁴⁺	niobium (V)	Mo^{6+}	Tc^{7+}	rutheniun		B+ paladium(II)	Ag^+	Cd ²⁺	In^{3+}	tin (IV)	antimony(III)	Te ²⁻	I-	Xe		
	strontium	yttriun	1			technetium	Ru		F	silver	cadmium	indium	Sn ²⁺	Sb ⁵⁺	telluride	1	xenon		
Tubiuiuiii		J		niobium(III)	J		rutheniun		paladium(IV)	511,01			tin (II)	antimony(V)	tondirac				
55	l	57	72	73	74	75	76	77	⁷⁸ Pt ⁴⁺	⁷⁹ Au ³	80 Hg ²⁺	81 Tl+	82 Pb ²⁺	83 Bi ³⁺	⁸⁴ Po ²⁺	85	86		
Cs ⁺	Ba ²⁺	La^3	$^{+}$ Hf $^{4+}$	Ta ⁵⁺	W^{6+}	Re^{7+}	Os	⁴⁺ Ir ⁴⁻	+ platinum(IV)	gold (III)	mercury (II	thallium (I)	lead (II)	bismuth(III)	polonium(II)	I At⁻	Rn		
cesium		lanthan		tantalum	tungsten	rhenium	osmiur	n iridium	Pt ²⁺	Au ⁺			Pb ⁴⁺	Bi ⁵⁺	Po ⁴⁺	astatide	radon		
									platinum(II)	gold (I)	mercury (I)	Tl ³⁺ thallium(III)	lead (IV)	bismuth(V)	polonium(IV)				
87		89						-	<u> </u>										
Fr ⁺	Ra ²⁺	Ac^3	3+																
francium	I . I	actiniu	m 58	59	60	61	62	Sm ³⁺ 63	Eu ³⁺ 64	65	66	67	68	69	70	Yb ³⁺	71		
				Ce ³⁺ Pr	3+ N	d ³⁺ Pn	3+ san	narium(III) euro	ppium (III) Gd	3+ Т	b^{3+} D	v ³⁺ Ho	o ³⁺ E1	r ³⁺ Tr	n ³⁺ ytte	rbium(III)	Lu ³⁺		
				I		nium promet		Sm ²⁺	Eu ²⁺ gadolin		oium dyspro	<i>y</i>		I		* 71 O . 1	lutetium		
samarium(II) europium (II)																			
			90	91 P2	a ⁵⁺ 92 [J6+ 93	94	Pu ⁴⁺ 95	Am ³⁺ 96	97	$3k^{3+}$ 98	99	100	101 _M	$[d^{2+}]^{102}$	2 No ²⁺ 1	03		
Γh^{4+} [protactinium(V)] uranium (VI) Γh^{3+} [plutonium(IIV) [americium(III)] Γh^{3+} [berkelium(III)] Γh^{3+} [Γh^{3+} [Γh^{3+} [mendelevium (III]] Γh^{3+} [mendelevium (IIII]] Γh^{3+} [mendelevium (III]] Γh^{3+} [mendelevium (IIIII]] Γh^{3+} [mendelevium (IIII]] Γh^{3+} [mendelevium (IIIII]] Γh^{3+} [mendelevium (IIII]] Γh^{3+} [mendelevium (IIIIIIIIII]] Γh^{3+} [mendelevium (IIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIII													Lr^{3+}						
													wrencium						
				protactini	um(IV) uraniui	n (IV)	plut	tonium(VI) ame	ericium(IV)	berkel	ium(IV)				ium (III) nob				