- Parent material and climate interact to control soil C dynamics on annual to centennial
- timescales through the development of poorly crystalline metal oxides
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# PARENT MATERIAL AND CLIMATE INTERACTIONS CONTROL SOIL C DYNAMIC ${f 2}$

10 Abstract

Lorem ipsum...

Parent material and climate interact to control soil C dynamics on annual to centennial timescales through the development of poorly crystalline metal oxides

# Key messages:

- Climate explains more variance in  $\Delta^{14}$ C at the soil surface; parent material explains more variance at depth
- Interaction of parent material and climate explains more variance in bulk soil and respired  $\Delta^{14}$ C than either factor alone
- Poorly crystalline metal oxide content is highly correlated with the difference between bulk soil and respired  $\Delta^{14}C$

21 Introduction

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Understanding the response of soil carbon stocks to current and future changes in 22 climate requires insight into both the physical and environmental factors governing soil 23 carbon dynamics. Climate, and in particular temperature, has been found to be the most 24 important variable for explaining the age of soil carbon in topsoil at local to global scales 25 [Frank et al., 2012; Mathieu et al., 2015; Shi et al. 2020]. Yet our current understanding of 26 soil organic matter decomposition underscores the importance of physical mechanisms that 27 can affect the temperature sensitivity of decomposition, such as the interaction between soil organic matter and minerals [Davidson et al., 2000; Davidson and Janssens, 2006; Rasmussen 29 et al., 2006; Lehmann and Kleber, 2015;]. The effect of mineral-organic associations on the temperature sensitivity of soil organic matter has been addressed in several modeling studies 31 [Abramoff et al., 2019; Riley et al., 2016; Ahrens et al., 2015; Woolf and Lehmann, 2019]. These models typically invoke Michaelis Menten kinetics in addition to an Arrhenius-type temperature response in order to account for energy and substrate limitations on decomposition rates for soil organic matter found in association with minerals [Tang and 35 Riley, 2019; Ahrens et al., 2020. Studies comparing the relative importance of different soil mineral assemblages versus temperature for explaining soil C dynamics over time in situ are

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scarce, yet critical for testing these model-based findings. We designed the current study to
quantify the relative importance of climatic versus mineralogical mechanisms of soil organic
C persistence across a combined gradient of mean annual temperature (MAT) and parent
material in order to provide insight into the relevant time scales associated with these
mechanisms.

The relevance of soil minerals for mediating soil organic matter protection has been 43 found to be a function of the specific minerals present, rather than the amount of clay or total mineral surface area [Rasmussen et al. 2018a, "Beyond Clay"]. Soil mineral assemblages are dynamic, developing over time as primary minerals inherited from parent material weather to form reactive, poorly crystalline secondary minerals, which in turn eventually weather to increasing stable crystalline species [Slessarev 2022, others...]. Soils enriched in poorly crystalline metal oxides, such as Al and Fe oxyhydroxides, are known to be of particular importance for the accumulation and persistence of soil C [Torn 1997, Maseiello 2004. The amount of these minerals present is directly related to parent material, but is also 51 a function of primary mineral weathering rates [Slesseray et al. 2022]. Due to the strong 52 effect of climate on weathering rates, different soil mineral assemblages can form from the 53 same parent material under different climatic regimes (Rasmussen et al. 2018b). Conversely, similar mineral assemblages can be found among soils developed on different parent materials given adequate time for weathering and similar vegetation and climate [Graham and O'Geen 2010 "Soil mineralogy trends in California landscapes"]. These differential outcomes demonstrate that climatic and mineralogical controls on soil C cycling are not independent, but interact over the centennial to millenial time scales of soil development.

The strength and sorptive capacity of soil minerals is dependent on ligand exchange,
which is a function of not only surface area and charge, but more specifically the density of
accessible hydroxyl groups (Kaiser and Guggenberger, 2003; Rasmussen et al., 2018a
"Beyond Clay"; Kleber et al., 2015). Poorly crystalline metal oxides are particularly enriched

in hydroxyl groups, and batch sorption/desorption experiments have shown that the mineral-organic interactions between pedogenic metal oxide-rich clays are stronger than 65 those with siloxane-rich phyllosilicate clays (Kahle et al., 2004). Furthermore, the reactive properties of pedogenic metal oxides can also facilitate lower strength interactions with soil 67 organic matter through multivalent cation bridging (Kleber et al., 2007). This high reactivity of poorly crystalline Fe oxides is also implicated in the observations that poorly 69 crystalline Fe is correlated with aggregate stability but crystalline Fe oxide content is not Duiker et al., 2003). The importance of pedogenic metal oxides for explaining both soil C 71 concentration and bulk soil radiocarbon ( $\Delta^{14}C_{bulk}$ ) is confirmed by the findings of 72 Rasmussen et al. (Soil systems, 2018a) [other citations?], who observed that oxalate 73 extractable iron was the best predictor of both properties. However, the relevance of mineral-organic associations with specific mineral phases such as poorly crystalline metal oxides or crystalline metal oxides for specific timescales of soil C turnover is poorly studied.

Radiocarbon (14C) is a useful tracer for soil C dynamics over time scales ranging from 77 annual to millennial. The use of <sup>14</sup>C to measure timescales of soil carbon decomposition is 78 reliant on our knowledge of the ratio of  $^{12}\text{C}/^{14}\text{C}$  in the atmosphere. Once CO<sub>2</sub> is fixed into 79 organic matter via photosynthesis, this ratio starts to shift as <sup>14</sup>C is preferentially lost due to radioactive decay. Changes in the <sup>12</sup>C/<sup>14</sup>C ratio due to radioactive decay are detectable at 81 the relatively longer timescales of hundreds to thousands of years. However, we can detect changes in <sup>14</sup>C with nearly annual resolution for the so-called "bomb-C" period, which began 83 with the deployment and atmospheric testing of nuclear weapons in the mid- $20^{\rm th}$  century. This pulse of "bomb-C" led to a doubling of atmospheric <sup>14</sup>C concentration prior to the ban on above-ground nuclear tests in 1963. The level of <sup>14</sup>C in the atmosphere returned to pre-bomb levels around 2020 (Hua et al., 2021), thus archived samples now represent the best opportunity to construct a high-resolution time series of the bomb-C pulse as it moves through different soil organic matter pools.

Soil is an open system, and this has important implications for the interpretation of 90 radiocarbon measurements of soil C. For most soils, the majority of carbon that enters the 91 soil leaves relatively quickly, with only a small fraction persisting (Sierra et al. 2018). 92 Radiocarbon measurements of bulk soil C ( $\Delta^{14}C_{bulk}$ ) typically capture the signal from these 93 persistent pools, while more transient pools dominate measurements of C leaving the soil via heterotrophic respiration (Trumbore 2000). Here we define "transient" for C cycling on 95 annual to decadal timescales, while we use "persistent" to refer to C cycling on centennial to 96 millennial timescales. A diagnostic feature of the radiocarbon measurements of  $\Delta^{14}C_{bulk}$  and heterotrophically respired CO<sub>2</sub> ( $\Delta^{14}C_{respired}$ ) is that when these two signals are the same it indicates that all of the C in the soil has an equal probability of being decomposed by microbes and the system is homogenous. However, when  $\Delta^{14}C_{bulk}$  and  $\Delta^{14}C_{respired}$  are 100 substantially different this indicates the presence of both labile and persistent pools of soil C. 101

We turned to the western slope of the Sierra Nevada Mountains, USA to compare and 102 contrast the effects of climate and mineral assemblage on soil C dynamics. Drawing on 103 earlier studies in this region (Jenny et al., 1949; Trumbore et al., 1996; Dahlgren et al., 1997; 104 Rasmussen et al., 2006), we selected soils similar in age and vegetation along a combined 105 gradient of parent material (granite, andesite, basalt) and MAT (12°C = "warm", 9°C = 106 "cool", 5°C = "cold"). The climate gradient also represents a weathering gradient, with 107 poorly developed soils at the cold climate sites, intermediately developed soils at the cool 108 climate sites, and highly weathered soils at the warm climate sites (Harradine and Jenny, 109 1958; Rasmussen et al., 2010). Previous work at these sites confirmed strong differences in 110 mineral assemblages along both the parent material and climate gradients, making these 111 sites an ideal setting for probing the relative influence of climatic and mineralogical factors 112 on soil C dynamics (Rasmussen et al., 2006; others?). 113

We were able to construct a time series of both  $\Delta^{14}C_{bulk}$  and  $\Delta^{14}C_{respired}$  at these sites by combining data from samples newly collected in 2019 with data from archived samples

collected in 2001 and 2009. Such a time series provides a crucial constraint for determining 116 the trajectory of bomb-derived <sup>14</sup>C concentrations over time (Beem-Miller et al., 2020; 117 Stoner et al., 2020). Whether bomb-C concentrations are increasing or decreasing with time 118 depends on both on the distribution of soil C among pools with different cycling rates as well 119 as the year in which the soil was sampled, and thus cannot be easily determined from 120 observations at a single point in time (Baisden et al. 2003). Using the radiocarbon time 121 series in combination with previously determined mineralogical data, we were able to test 122 several hypotheses regarding the roles of mineralogical versus climatic factors in determining 123 both overall cycling rates and the dynamics of transiently cycling soil C. 124

We can expect soils with large stocks of persistent soil C to have depleted values of 125  $\Delta^{14}C_{bulk}$  relative to soils dominated by fast cycling soil C. If climate proves more important 126 than parent material for determining soil C persistence, than we would expect to see large 127 differences in  $\Delta^{14}C_{bulk}$  among climate regimes when comparing soils within a given parent 128 material, but minimal differences among parent materials when comparing soils within the 129 same climate regime. However, if parent material proves more important than climate for 130 soil C persistence, we would expect the opposite trends in  $\Delta^{14}C_{bulk}$ : differences would be 131 greater among parent materials within a given climate regime than among climate regimes 132 within a given parent material. Alternatively, if persistent soil C is associated with specific 133 soil minerals, we would expect an interactive effect of parent material and climate on 134  $\Delta^{14}C_{bulk}$ . For example, if soil C persistence is due to the association of soil organic matter 135 with poorly crystalline metal oxides, we would expect to observe the most depleted  $\Delta^{14}C_{bulk}$ 136 values where the combination of parent material and climate factors has led to the greatest 137 abundance of these specific soil minerals. 138

Soil C found in association with minerals is typically older than organic matter found in free particulate forms [cite]. Accordingly, we might expect climate to be the dominant factor controlling the amount and cycling rates of C in transiently cycling soil C pools, with mineral factors being less relevant at these shorter timescales. If this hypothesis is correct,
we would expect to see greater differences in  $\Delta^{14}C_{respired}$  among different climate regimes and
within a given parent material than we would among different parent materials within the
same climate regime. Furthermore, we would expect warmer climate soils to have  $\Delta^{14}C_{respired}$ values closer to the atmosphere than colder climate soils, due to faster decomposition rates
in the actively cycling soil C pools, and accordingly, we would expect  $\Delta^{14}C_{respired}$  to change
more over time at the warmer climate sites than at the colder climate sites.

149 Methods

## Site descriptions

We collected samples from nine sites across a combined gradient of parent material and 151 climate in the Sierra Nevada Mountains of California (Fig. 1). Parent material changes 152 from basalt to andesite to granite along the north-south axis of the cordillera, while MAT 153 decreases as a function of increaing elevation along the east-west axis. Total mean annual precipitation (MAP) ranges from 910 to 1400 mm yr<sup>-1</sup> across the sites. Precipitation 155 increases slightly with elevation (Fig. 1), and falls mainly as rain at lower elevations (< 156 1400 m), but mainly as snow at higher elevations (> 1800 m). The andesitic and basaltic 157 parent materials receive slightly more precipitation on average than the granitic soils, with 158 MAP values of 1330 ( $\pm$  75) mm yr<sup>-1</sup>, 1160 ( $\pm$  175) mm yr<sup>-1</sup>, and 1000 ( $\pm$  85) mm yr<sup>-1</sup> for 159 the andesite, basalt, and granite transects, respectively. 160

Vegetation at the study sites is typical of the Sierra Mixed Conifer habitat (Parker, I., and W. J. Matyas. 1981. CALVEG: a classification of Californian vegetation. U.S. Dep. Agric., For. Serv., Reg. Ecol. Group, San Francisco). All of the sites are forested and dominated by conifers, although the exact species composition changes along with climate. Tree species at the lowest elevation warm climate sites are predominantly *Pinus ponderosa* mixed with lesser amounts of *Quercus* spp. The canopy species at the mid-elevation cool climate sites consist primarily of *Abies concolor* and *Pinus lambertiana*, while *Abies* 

magnifica dominates at the high elevation cold climate sites. Species present at all sites include Calocedrus decurrens in the canopy, the shrubs Arctostaphylos spp., Chamaebatia foliolosa, and Ceanothus spp. in the understory, and patchy ground cover of grasses and forbs (Fig. 1).

Ecosystem	Dominant Vegetation	Elevation (m a.s.l.)	MAP (mm yr <sup>-1</sup> )	MAT (°C)	Parent Material	Soil Taxonomy	
PP	Pinus ponderosa	920–1400	80–130	10–13	GR	fine-loamy, mixed, semiactive, mesic Ultic Haploxeralf	
	Pinus lambertiana		Mostly rain		BS	fine, kaolinitic, mesic Xeric Haplohumult	
	Quercus kelloggii				AN	fine, parasesquic, mesic Andic Palehumult	
WF	Abies concolor	1500–1800	80–130	8–10	GR	coarse-loamy, mixed, superactive, mesic Humic Dystroxerept	
	Pinus ponderosa		Rain/snow		BS	loamy-skeletal, mixed, superactive, mesic Typic Haploxerept	
	Pinus lambertiana				AN	medial-skeletal, amorphic, mesic Humic Haploxerand	
RF	Abies magnifica	2200–2400	100–130	5–6	GR	mixed, superactive, frigid Dystric Xeropsamment	
	Pinus jeffreyi		snow		BS	sandy-skeletal, mixed, superactive, frigid Typic Xerorthent	
					AN	medial-skeletal, amorphic, frigid Humic Vitrixerand	

Figure 1. NB: this is a placeholder—copied verbatim from Rasmussen et al. 2018 Dominant vegetation, climate parameters, and soil taxonomy for sampled ecosystems. MAP—mean annual precipitation; MAT—Ecosystem abbreviations: PP—ponderosa pine; WF—white fir; RF—red fir. Dominant vegetation is listed in order of over-story dominance. Parent material abbreviations: GR—granite; BS—basalt; AN—andesite.

#### Sample collection

Site locations were initially established in 2001 by C. Rasmussen (Rasmussen et al., 2006), and resampled in 2009 (Rasmussen et al., 2018) and 2019 (this study). Three replicate pits were dug to the depth of the B horizon (2001 sampling), or down to the C horizon (2009 and 2019 sampling). Samples were collected from the pit sidewalls by horizon

in 2001 and 2009, and by 0.1 m increments in 2019. For the 2019 resampling, we located the sites using GPS and geospatial coordinates recorded during site establishment. Prior to sample collection we compared the soil profiles to the pedon descriptions from the 2001 sampling campaign to confirm the profiles matched. In 2019 we also measured the depth of the litter layer and collected representative litter samples from each site. However, we only focus on the upper mineral soil layers in this study (0 to ca. 0.3m) as the 2001 sampling was restricted to this depth range.

#### 184 Incubations

Laboratory soil incubations were performed on composite samples from the three 185 replicate pedons sampled at each site in 2001 and 2019. We omitted the 2009 samples from 186 the incubation experiment to save on time and analysis costs, and because sample material 187 was only available from a single profile at each site. We composited and incubated each 188 depth increment separately in 1 L glass mason jars fitted with air tight sampling ports in the 189 lids. Incubations were performed in duplicate. Prior to the start of incubations we adjusted 190 the soil moisture content to 60% of water holding capacity (WHC). Samples from 2001 were 191 air-dried prior to archiving, and therefore we also air-dried the freshly collected soils from 192 2019 in order to control for the known effects of drying and rewetting on  $\Delta^{14}C_{resvired}$ 193 (Beem-Miller et al., 2021). We defined WHC as the gravimetric water content of 194 water-saturated soil placed in mesh-covered (50µm) tubes (50ml) weighed after draining for 195 30 minutes on a bed of fine sand. Following rewetting we allowed the soils to respire for one week before closing the jars. Incubations proceeded until CO<sub>2</sub> concentrations in the jar 197 headspace reached approximately 10,000 ppm, at which point we collected a 400 ml gas 198 subsample for radiocarbon analysis. Gas samples were collected with pre-evacuated 199 stainless-steel (Restec) vacuum canisters. All incubations were performed in the dark at 200 20°C. 201

#### Soil Physical Analyses and Mineral Characterization

Data on soil particle size distributions, bulk density, and mineral characterization were 203 obtained from previously published analyses of samples collected at the study sites in 2001 204 and 2009 (Rasmussen et al. 2006, Rasmussen et al., 2018). Both qualitative and quantitative 205 approaches were used to characterize soil mineral assemblages, including X-ray diffraction 206 (XRD) for the clay (<2 µm) fraction, atomic absorption spectroscopy, and non-sequential 207 selective dissolution. These previous analyses revealed that the dominant mineral species in 208 the soils of the highly weathered warm climate zone were similar across parent materials, but differed substantially across parent materials at the less weathered cool and cold climate sites. Mineral assemblages at the warm climate sites are dominated by 1:1 clays and large accumulations of crystalline iron oxides [Dahlgren et al. 1997 "Soil development along an 212 elevational transect in the western Sierra Nevada, California"; Rasmussen et al. 2007 "Soil 213 Genesis and Mineral Transformation Across an Environmental Gradient on Andesitic Lahar"; 214 Rasmussen et al. 2010 "Basalt weathering and pedogenesis across an environmental gradient 215 in the southern Cascade Range, California, USA"]. In contrast, the cool and cold climate 216 and esitic soils contain high concentrations of poorly crystalline short-range order (SRO) 217 minerals such as allophane and iron oxyhydroxides. The cool and cold climate basaltic soils 218 contain intermediate amounts of SRO minerals, while the granitic soils lack SRO minerals 219 almost entirely, but are rich in quartz and contain relatively more hydroxyl-interlayered 220 vermiculite than soils from the other lithologies. 221

The previous work at these sites showed that poorly crystalline metal oxide content was the best predictor of both C abundance and  $\Delta^{14}C_{bulk}$ . Accordingly, our analyses focus on the relationship between radiocarbon measurements and the abundance of poorly crystalline versus crystalline metal oxides as well as 2:1 versus 1:1 clays, but not the whole suite of mineralogical data. For simplification, we use the sum of ammonium-oxalate extractable aluminum and half of the ammonium-oxalate extractable Fe selectively dissolved

from bulk soils as a proxy for the abundance of poorly and non-crystalline metal oxides, and the difference of dithionite-citrate extractable Fe and ammonium-oxalate extractable Fe for the abundance of crystalline Fe in the main text. We present the results of additional regression analyses looking at the specific relationships between  $\Delta^{14}C_{bulk}$ ,  $\Delta^{14}C_{respired}$ , and the concentration of Fe or Al extracted with ammonium-oxalate, Fe extracted with dithionite-citrate, and Al extracted with sodium-pyrophosphate in the supplemental information.

# <sup>235</sup> Carbon, Nitrogen, and Radiocarbon Analysis

Total carbon and nitrogen content was determined by dry combustion (2019 samples: Vario Max, Elementar Analysensysteme GmbH) on finely ground soils (2019 samples: Retch MM400). For radiocarbon analysis of 2001 and 2019 samples, we first purified CO<sub>2</sub> from combusted soil samples (bulk soils) and incubation flask samples (respired CO<sub>2</sub>) on a vacuum line using liquid N<sub>2</sub>. Following purification, samples were graphitized with an iron catalyst under an H<sub>2</sub> enriched atmosphere at 550°C. Radiocarbon content was then measured by accelerator mass spectrometry (Micadas, Ionplus, Switzerland) at the Max Planck Institute for Biogeochemisty (Steinhof, 2017). See Rasmussen et al. (2018) for details of C, N, and radiocarbon analysis of the 2009 samples.

We report radiocarbon values using units of  $\Delta^{14}$ C, defined as the deviation in parts per thousand of the ratio of  $^{14}$ C $^{-12}$ C from that of the oxalic acid standard measured in 1950. This unit also contains a correction for the potential effect of mass-dependent fractionation by normalizing sample  $\delta^{13}$ C to a common  $\delta^{13}$ C value of -25 per mil (Stuiver & Polach, 1977). Values with  $\Delta^{14}$ C >0 indicate the presence of "bomb" C produced by atmospheric weapons testing in the early 1960s. Values with  $\Delta^{14}$ C < 0 indicate the influence of radioactive decay of  $^{14}$ C, which has a half-life of 5730 years.

# 252 Spline fitting

We used a spline function to compare soil properties from samples collected from 253 different depth intervals in different years and at different sites. We were motivated to use 254 consistent depth increments across sites when resampling in 2019 because of the strong 255 correlation between depth and  $\Delta^{14}$ C observed in the 2009 dataset, a correlation also noted in 256 numerous other studies (Mathieu et al., 2015; Shi et al., 2020; others...). We fit a 257 mass-preserving quadratic spline to the 2001 and 2009 profiles in order to convert soil 258 property data to the equivalent depth increments sampled in 2019 (Bishop et al., 2001). We 259 performed the spline fitting with the mpspline function of the GSIF package in R, using a  $\lambda$ 260 value of 0.1 (Hengl 2019). 261

#### 262 Statistical analysis

We used a linear modeling approach to assess the relative explanatory power of climate 263 versus parent material on the observed variation in  $\Delta^{14}$ C, as well as potential interactions 264 between these two factors. We constructed separate models for  $\Delta^{14}C_{bulk}$  and  $\Delta^{14}C_{respired}$  but 265 with the same equation structure (Eq. 1). For each model we considered the two-way 266 interaction between parent material and climate as well as the three-way interaction with 267 time (Eq. 1). For ease of interpretation, we considered the effect of depth by modeling each depth layer separately (0-10 cm, 10-20 cm, 20-30 cm). We also made pairwise comparisons of  $\Delta^{14}C_{bulk}$  and  $\Delta^{14}C_{respired}$  across sites and within years, and across years for individual sites. We assessed the significance of the temporal trend for pairwise combinations of parent material and climate using the emmtrends function of the emmeans package (Lenth, 2021). We corrected for multiple comparisons using Tukey's honestly significant mean difference.

#### Eq. 1

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$$\Delta^{14}C = \alpha + \beta_1(Parent\_material) \times \beta_2(Climate) \times \beta_3(Year) + \varepsilon$$

Where  $\alpha$  is the intercept term, the  $\beta$  terms are coefficients, and  $\varepsilon$  is random error.

We also considered the relationship between  $\Delta^{14}C_{bulk}$  and  $\Delta^{14}C_{respired}$  to gain insight 276 into potential differences in soil C dynamics and persistence mechanisms across our sites 277 (Sierra et al. 2018). We modeled the effects of parent material (Eq. 2) and climate (Eq. 3) 278 on this relationship separately, as we did not have an adequate number of observations to 279 consider the interaction between these two explanatory variables. For this analysis we used 280  $\Delta^{14}$ C measurements made on samples collected in 2001 and 2019, and data from all depths. 281 The three-way interactions of  $\Delta^{14}C_{bulk}$  and the explanatory variables (parent material or 282 climate) were not significant with either depth or time for either Eq. 2 or Eq. 3, so we did 283 not include either depth or time as variables in the models. 284

Eq. 2

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$$\Delta^{14}C_{respired} = \alpha + \beta_1(\Delta^{14}C_{bulk}) \times \beta_2(Parent\_material) + \epsilon$$

286 Eq. 3

$$\Delta^{14}C_{respired} = \alpha + \beta_1(\Delta^{14}C_{bulk}) \times \beta_2(Climate) + \epsilon$$

We assessed the relative importance of poorly crystalline versus crystalline iron 287 minerals in protecting soil C from microbial decomposition by regressing  $\Delta^{14}$ C against the 288 concentrations of ammonium-oxalate extractable iron, ammonium-oxalate extractable 289 aluminum, pyrophosphate extractable aluminum, and dithionite-citrate extractable iron (Eq. 290 4). We fit the model for  $\Delta^{14}C_{bulk}$ ,  $\Delta^{14}C_{respired}$ , and the difference between  $\Delta^{14}C_{respired}$  and 291  $\Delta^{14}C_{bulk}$  ( $\Delta^{14}C_{bulk\text{-}respired}$ ). We used  $\Delta^{14}C$  data from 2001, 2009, and 2019 for the  $\Delta^{14}C_{bulk}$ 292 model, but only data from 2001 and 2019 for the  $\Delta^{14}C_{respired}$  and  $\Delta^{14}C_{bulk-respired}$  models. Selective dissolution was only performed on the soils collected in 2001, but these data were assumed to be comparable for the other time points as they reflect weathering processes 295 operating at timescales beyond the 18-year duration of this study. The regression analysis 296 conducted with Eq. 4 was done for the combined depth increment of 0 to 0.3 m, as 297 extracted metal concentrations did not change substantially over this increment. Combining 298

depth increments allowed us to control for the depth dependence of  $\Delta^{14}$ C as well as simplify interpretation of the data. In order to obtain values for the necessary data over the 0 to 0.3 m depth increment we computed mass-weighted estimates of extractable metal concentrations, carbon mass-weighted means of  $\Delta^{14}$ C<sub>bulk</sub> and flux-weighted means of  $\Delta^{14}$ C<sub>respired</sub>; these calculations were made prior to determining  $\Delta^{14}$ C<sub>bulk-respired</sub>.

Eq. 4

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$$\Delta^{14}C = \alpha + \beta_1(Metal_x) + \beta_2(time) + \varepsilon$$

Where  $\alpha$  is the intercept term,  $\beta$  is the coefficient for each factor in the model,  $Metal_x$  is the concentration of selectively dissolved metal oxides, time is the year of sampling, and  $\varepsilon$  is random error.

Results

#### 309 Soil carbon concentrations and flux rates

We observed both parent material and climate effects on soil organic C (SOC)

concentration. SOC concentrations were similar among parent materials for the warm

climate sites (**Fig. 2**, **a**), while at the cool and cold climate sites (**Fig. 2**, **b**, **c**) the

andesitic soils had higher SOC concentrations than either the basaltic or granitic soils. The

basaltic and granitic soils had similar SOC concentrations across climate zones, while the

cool and cold climate andesitic soils were enriched in C relative to the warm climate soils.

Soils showed a similar decrease in SOC concentration with depth across all sites (**Fig. 2**).

We did not observe significant changes in SOC concentration over time for the majority of our sites. We saw the most substantial variation in SOC concentration between 2001, 2009, and 2019 in the surface layers (0-0.1 m), with significant differences observed between years for the warm climate andesitic and basaltic sites, and for the deeper (0.1-0.2 m, 0.2-0.3 m) depth layers at the cold climate andesite site (SI Fig. 1). We observed similar trends in

SOC stocks across sites as with SOC concentration, despite lower bulk densities in the
andesitic sites with high C concentrations (SI Fig. 2). However, we did not assess soil
carbon stock changes over time in this study as we did not measure bulk density in 2019,
and earlier sampling campaigns did not incorporate adequate spatial replication for making
such an assessment (Johnson et al., 1990; Schrumpf et al., 2013).

Flux rates of heterotrophic respiration differed among parent materials and among 327 climate zones. When compared on a carbon basis (mg  $\mathrm{CO}_2$  g  $\mathrm{C}^{\text{-}1}$  d<sup>-1</sup>), flux rates tended to 328 be higher for the andesitic soils than soils from either basaltic or andesitic soils, particularly 329 at depth (SI Fig. 2). The exceptions to this trend were the surface (0-0.1 m, 2001 and 330 2019) and near-surface (0.1-0.2 m, 2019) soils from the warm climate sites (SI Fig. 2). 331 Respiration rates for granitic and basaltic soils tended to decrease with decreasing MAT 332 (warm > cool > cold); however, we did not see any clear trend in respiration rates with 333 respect to climate for the andesitic soils (SI Fig. 2). 334

## 335 Radiocarbon depth profiles

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 $\Delta^{14}C_{bulk}$  covaried with both parent material and climate. We observed 336 the most enriched  $\Delta^{14}C_{bulk}$  at the warm climate sites, indicating a preponderance of 337 relatively young, fast-cycling C. However, contrary to what would be expected from the 338 decomposition-temperature relationship, we observed the oldest soil C (i.e. most depleted 339  $\Delta^{14}C_{bulk}$  values) at the intermediate MAT cool climate sites rather than at the cold climate sites (Fig. 3, b, c) (Fig. 3, a). When comparing  $\Delta^{14}C_{bulk}$  from different parent materials 341 within a given climate zone,  $\Delta^{14}C_{bulk}$  of andesitic soils tended to be the most depleted, while 342 the granitic soils tended to be the most enriched (Fig. 3). We focus on the 2019 data here 343 for simplicity, but  $\Delta^{14}C_{bulk}$  profiles showed similar patterns in both 2001 (SI Fig. 2) and 344 2009 (Rasmussen et al., 2018). 345

Analysis of variance for  $\Delta^{14}C_{bulk}$  revealed significant two-way interactions between

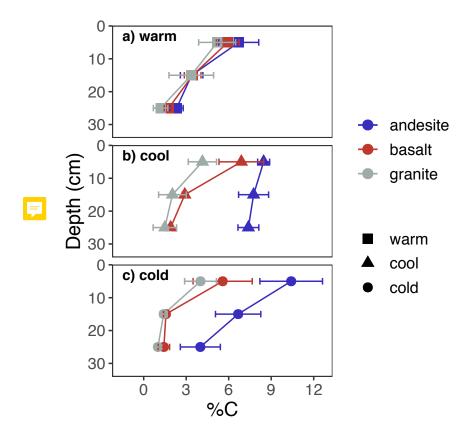


Figure 2. Profiles of soil C concentration. Points show mean of 2001, 2009, and 2019 data; error bars show  $\pm 2 \text{SE}$ .

parent material and climate at all depths (Table 1). This interaction was evident in the differences in  $\Delta^{14}C_{bulk}$  that we observed among parent materials within each climate zone. 348 We observed the greatest differences in  $\Delta^{14}C_{bulk}$  among parent materials at the warm and 349 cool sites (Fig. 3, a, b), while  $\Delta^{14}C_{bulk}$  was similar among parent materials at the coldest 350 sites (Fig. 3, c). We also found depth to be an important factor influencing the relative 351 importance of climate versus parent material effects on  $\Delta^{14}C_{bulk}$ . Although  $\Delta^{14}C_{bulk}$  declined 352 with depth for all sites, climate explained more of the variance in  $\Delta^{14}C_{bulk}$  in the uppermost 353 soil layer (0-10 cm) whereas parent material explained more in the bottom two layers (10-20 354 cm, 20-30 cm) (**Table 1**). 355

Heterotrophically respired  $CO_2$ . The patterns we observed in  $\Delta^{14}C_{respired}$  were similar to those we observed in  $\Delta^{14}C_{bulk}$  (Fig. 3). We found climate to the only significant

factor for explaining the variance observed in  $\Delta^{14}C_{respired}$  in the uppermost soil layer (0-0.1 m), while at the deepest depth (0.2-0.3 m) parent material was more important than climate (Table 1). Overall, we found the two-way interaction between parent material and climate explained more of the variance in  $\Delta^{14}C_{respired}$  than it did in  $\Delta^{14}C_{bulk}$  (Table 1, F values).

The effect of climate on  $\Delta^{14}C_{respired}$  appeared to be moderated by parent material.

Accordingly, we did not observe significant differences in  $\Delta^{14}C_{respired}$  among the andesitic soils when compared across climate zones at any depth (SI Table XX Tukey results for emm).

In contrast,  $\Delta^{14}C_{respired}$  diverged substantially between climate zones for the basaltic and granitic soils, particularly for the 0.1-0.2 m and 0.2-0.3 m depth layers (**Fig. 3**). Overall,  $\Delta^{14}C_{respired}$  values across sites were most similar at the soil surface (0-0.1 m), and most divergent at the intermediate depth (0.1-0.2 m) (**Fig. 3**).

#### Radiocarbon time series

Temporal trends in bulk and respired  $\Delta^{14}$ C reflect the degree to which soil C is exchanging with C fixed from the atmosphere. The average annual decline in  $\Delta^{14}$ C atmospheric CO<sub>2</sub> between 2001 and 2009 for the northern hemisphere was -5.13 per mille yr<sup>-1</sup> (Graven et al., 2017; Sierra, 2018) (**Fig. 4, dotted lines**). Therefore, changes in  $\Delta^{14}$ C of soil C that parallel the atmospheric trend must be exchanging relatively rapidly compared to those that change little over the same time period.

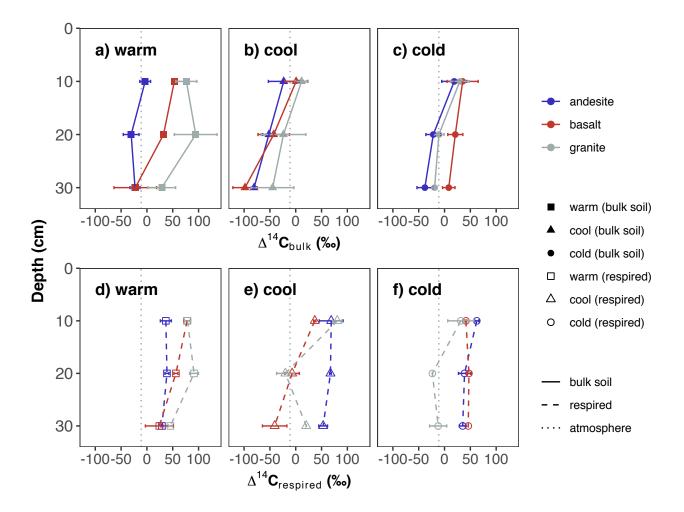


Figure 3. Depth profiles of  $\Delta^{14}C_{bulk}$  and  $\Delta^{14}C_{respired}$  in 2019. Top panels show bulk data, bottom panels show respired data. Panels (a) and (d) show data from the warm climate sites, (b) and (e) from the cool climate sites, and (c) and (f) from the cold climate sites. Black vertical lines show  $\Delta^{14}C$  of the atmosphere in 2019. Points show the mean of three replicate profiles for bulk soil, and the mean of laboratory duplicates for respired  $CO_2$ . Error bars show  $\pm 1$  SD for bulk soils and the minimum and maximum for respired  $CO_2$ .

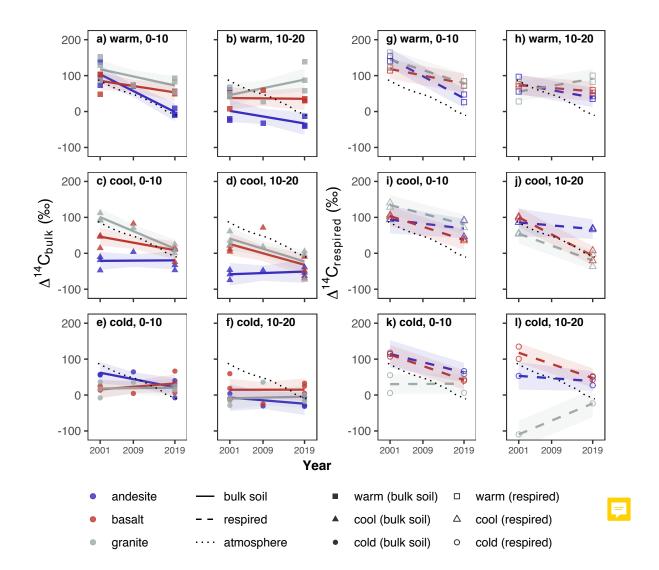


Figure 4. Temporal trends in  $\Delta^{14}$ C for 0-10 cm and 10-20 cm depth layers. Panels a-f show  $\Delta^{14}$ C bulk data; from left, the first column (panels a, c, and e) show 0-10 cm data, and the second column (panels b, d, and f) shows 10-20 cm data. Panels g-l show  $\Delta^{14}$ C respired data; the third column from left (panels g, i, k) shows 0-10 cm data, and the rightmost column (panels h, j, and l) shows 10-20 cm data. Points show observed data; lines show linear trend estimates for marginal means; ribbons show 95% confidence intervals for trends. Dotted line shows atmospheric  $\Delta^{14}$ C.

Table 1  $ANOVA~for~\Delta^{14}C_{\rm bulk}~and~\Delta^{14}C_{\rm respired}$ 

			Bulk soil			Respiration		
Depth	Predictor	df	F	p	df	F	p	
0-10cm	Parent material	2	12.00	< .001	2	0.04	0.958	
	Climate	2	32.34	< .001	2	14.02	< .001	
	Year	1	32.03	< .001	1	75.29	< .001	
	Parent material:Climate Parent material:Year Climate:Year Parent material:Climate:Year Residuals		8.75	< .001	4	7.90	0.001	
			2.38	0.105	2	1.93	0.177	
			6.61	0.003	2	2.26	0.137	
			5.19	0.002	4	3.75	0.024	
					16			
20-30cm	Parent material	2	15.58	< .001	2	0.92	0.421	
	Climate	2	11.61	< .001	2	0.77	0.483	
	Year	1	1.30	0.26	1	0.65	0.434	
	Parent material:Climate	4	1.71	0.165	4	4.33	0.019	
	Parent material:Year	2	1.56	0.222	2	0.86	0.446	
	Climate:Year	2	4.04	0.024	2	1.41	0.278	
	Parent material:Climate:Year	4	0.98	0.43	2	0.37	0.698	
	Residuals	44			13			

Bulk soil. We observed a significant three-way interaction between parent material, climate, and time at all three depths in the linear models (Eq. 1) for  $\Delta^{14}C_{bulk}$  (Table 1).

The change over time in  $\Delta^{14}C_{bulk}$  was also affected by depth, with greater differences seen between 2001 and 2019 in the uppermost soil layer than in the deeper layers (Fig. 4). We observed a significant decrease in  $\Delta^{14}C_{bulk}$  over time in both warm and cool climate granitic soils for the uppermost soil layer (0-0.1 m), and additionally for the warm climate andesitic soils (Fig. 4, a). In the deeper soil layers (0.1-0.2 m and 0.2-0.3 m), we only observed a significant change over time in  $\Delta^{14}C_{bulk}$  for the cool climate basalt and granite soils (Fig. 4,

c, d, 0.2-0.3 m data not shown).  $\Delta^{14}C_{bulk}$  of the cool climate andesitic soils remained essentially unchanged between 2001 and 2019 for all depths (**Fig. 4**, c, d, 0.2-0.3 m data not shown), underscoring the importance of the interaction between parent material and climate for explaining temporal trends in  $\Delta^{14}C_{bulk}$ .

The relationship of  $\Delta^{14}C_{bulk}$  to atmospheric  $\Delta^{14}C$  also depended on the combination of parent material and climate. In 2001, the warm climate sites were the only sites where the basaltic and andesitic soils were enriched relative to the atmosphere, and this enrichment was only observed for the uppermost soil layer (**Fig. 4**). In contrast, granitic soils at both the warm and cool climate sites were enriched relative to the atmosphere in 2001 (**Fig. 4**). For the cold climate sites, where  $\Delta^{14}C_{bulk}$  was most similar, all three lithologies were depleted relative to atmospheric in both surface and subsoil layers in 2001 (**Fig. 4**, **e**, **f**).

We observed that  $\Delta^{14}C_{bulk}$  tended to decrease or remain unchanged between 2001 and 395 2019 across sites, but the rates of change in  $\Delta^{14}C_{bulk}$  were typically smaller than the 396 corresponding change in atmospheric  $\Delta^{14}$ C over the same period. Accordingly,  $\Delta^{14}$ C<sub>bulk</sub> 397 measured in 2019 tended to be enriched relative to the atmosphere at more sites, and also 398 more enriched at depth than in 2001. We observed surface soil  $\Delta^{14}C_{bulk}$  (0-0.1 m) in 2019 to 399 be enriched relative to the atmosphere at all sites except for the cool climate andesite soils 400 (Fig. 4; Fig. 3, d-f). Furthermore,  $\Delta^{14}C_{bulk}$  was enriched relative to the atmosphere down 401 to 30 cm at two of the sites in 2019: the warm climate granite soil (Fig. 3, d) and cold 402 climate basalt soil (Fig. 3, f).  $\Delta^{14}C_{bulk}$  at the cool climate andesite site was the most 403 depleted relative to the atmosphere at all time points (Fig. 4). 404

Heterotrophically respired CO<sub>2</sub>. Temporal trends in  $\Delta^{14}C_{respired}$  tended to be of greater magnitude than what we observed for  $\Delta^{14}C_{bulk}$  (Fig. 4, g-l). However, changes in  $\Delta^{14}C_{respired}$  between 2001 and 2019 still tended to be smaller in magnitude than the change observed in the atmosphere over this period (Fig. 4, g-l). In contrast to  $\Delta^{14}C_{bulk}$ ,  $\Delta^{14}C_{respired}$  values of both surface and near surface soils (0-0.2 m) were close to atmospheric

levels in 2001, while in 2019,  $\Delta^{14}C_{respired}$  tended to be enriched relative to the atmosphere, even for the deeper soil layers (**Fig. 4, g-l**).

We saw significant decreases in  $\Delta^{14}C_{respired}$  values of surface (0-0.1 m) soils at seven of 412 the nine sites, with the only exceptions being the cool climate and esitic and cold climate 413 granitic sites (Fig. 4, g-l). In absolute terms, the changes in  $\Delta^{14}C_{respired}$  over time in the 414 uppermost soil layer were greatest at the warm sites (-4  $\pm 2$  per mil yr<sup>-1</sup>), while changes were 415 similar for the cool and cold sites (-2.7  $\pm$ 1.2 per mil yr<sup>-1</sup>, and -2.2  $\pm$ 2.1 per mil yr<sup>-1</sup>, 416 respectively). When considered within parent materials, granitic soils showed the greatest decrease in  $\Delta^{14}C_{respired}$  over time at the warm climate site and the least change at the cold climate site. In contrast, the andesitic soils showed the least amount of change over time at 419 the cool climate sites, while changes over time in the basaltic soils were similar across all 420 three climate regimes. The magnitude of the change in  $\Delta^{14}C_{respired}$  over time tended to 421 decrease with depth for all soils. We observed significant negative trends over time for 422  $\Delta^{14}C_{respired}$  at only four of the nine sites for the 0.1-0.2 m layer (warm and site, cool basalt, 423 cool granite, and cold basalt) (Fig. 4), and only one site for the 0.2-0.3 m layer (cold basalt) 424 (SI Fig. XX).  $\Delta^{14}C_{respired}$  at the cool and esitic soils remained unchanged at all depths over 425 the study period. 426

We observed a significant increase in  $\Delta^{14}C_{respired}$  from 2001 to 2019 at only one site: the cold climate granitic soil (**Fig. 4**). This was also the only soil for which  $\Delta^{14}C_{respired}$  was more depleted than  $\Delta^{14}C_{bulk}$ . We observed this anomaly for the deeper soil layers in both 2001 and 2019. We observed  $\Delta^{14}C_{respired}$  values of -469.1-127.8 per mil for the 8-27 cm layer in 2001, compared to  $\Delta^{14}C_{bulk}$  values between -30.80 and -10.50. Similarly, we observed  $\Delta^{14}C_{respired}$  values of -396.70 and -23.50 per mil for the 10-20 cm layer in 2019, compared to -18.10 and 0.40 for  $\Delta^{14}C_{bulk}$  (SI Table X). However, since these anomalous values of  $\Delta^{14}C_{respired}$  were restricted to the deeper soil layers from this one site, and were consistent over time, the response appears to be a laboratory artifact unique to these soils. Accordingly, we have excluded these highly depleted samples from the statistical analyses.

# Relationship of bulk soil and respired $ext{CO}_2$ $\Delta^{14} ext{C}$

We assessed the relationship between  $\Delta^{14}C_{bulk}$  and  $\Delta^{14}C_{respired}$  using linear regression models for parent material (Eq. 3) and climate (Eq. 4). We observed  $\Delta^{14}C_{respired}$  to be enriched relative to  $\Delta^{14}C_{bulk}$  for almost all sites and all depths, but that the magnitude of the difference depended on both parent material and climate. Accordingly, y-intercepts for all models were equal to or greater than zero, indicating that respired  $CO_2$  in these soils is predominantly modern (i.e. < 60 years old) for all but the deepest soil layers, regardless of parent material or climate regime. We found the largest y-intercept values for the soils developed on andesitic parent material, and for soils in the cool climate zones, which indicates that  $CO_2$  respired from these soils is relatively enriched in decadally cycling bomb-C.

Slope values less than 1 in these models indicate that every per mille change in 448  $\Delta^{14}C_{bulk}$  is associated with a correspondingly smaller change in  $\Delta^{14}C_{respired}$ . This suggests that the process regulating persistence of soil C on long time scales is distinct from that 450 which regulates more transiently cycling soil C. Similar to what we found for the 451 y-intercepts, modeled slopes for the parent material-only model (Eq. 2) were smallest for 452 the andesitic soils (slope = 0.51, 95% CI = [0.22, 0.80]) (Fig. 5, a), and for cool climate soils in the climate-only model (Eq. 3) (slope = 0.61, 95% CI = [0.30, 0.91]) (Fig. 5, b). While we could not directly test the interaction of parent material and climate factors in these models owing to the limited number of observations, we observed that mean differences 456 in  $\Delta^{14}C_{bulk}$  and  $\Delta^{14}C_{respired}$  were substantially greater for the cool climate soils developed on 457 andesitic parent material than for the other sites. 458

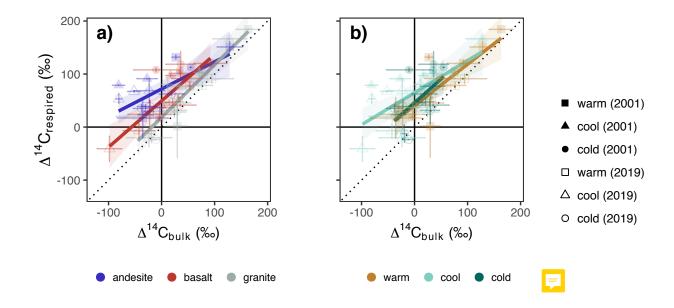


Figure 5. Parent material and climate effects on the relationship of  $\Delta^{14}C_{bulk}$  and  $\Delta^{14}C_{respired}$ .

a) Parent material model (Eq. 3) and b) Climate model (Eq. 4). Dotted line shows 1:1 relationship. Points show the mean of three replicate profiles for bulk soil, and the mean of laboratory duplicates for respired  $CO_2$ . Error bars show  $\pm 1$  SD for bulk soils and the minimum and maximum for respired  $CO_2$ . Respired  $CO_2$  from the cold granite site in 2001 was extremely depleted in  $\Delta^{14}C$  and thus is excluded for display purposes.

#### 459 Mineral assemblages and radiocarbon

Mineral assemblage data is reported fully in Rasmussen et al. (2018). Here we focus on 460 the selective dissolution data with respect to the trends we observed in  $\Delta^{14}C_{bulk}$ ,  $\Delta^{14}C_{respired}$ , 461 and  $\Delta^{14}C_{respired-bulk}$ . We observed a significant negative correlation between  $\Delta^{14}C_{bulk}$  and the 462 concentration of oxalate extractable iron, oxalate extractable aluminum, and pyrophosphate 463 extractable aluminum (SI). For simplicity, we focus here on the sum of oxalate extractable 464 aluminum and half of the oxalate extractable iron as a proxy for non-crystalline metal oxide 465 abundance, and the difference of dithionite-citrate extractable iron and ammonium-oxalate 466 extractable iron as a proxy for crystalline Fe-oxide abundance. The relationship between 467 non-crystalline oxide abundance and  $\Delta^{14}C_{bulk}$  was highly significant (p <0.001), with the 468 model explaining 0 percent of the observed variation. In contrast, we did not find a 460 significant relationship between crystalline Fe-oxide abundance and  $\Delta^{14}C_{bulk}$ . 470

We observed a relationship between  $\Delta^{14}C_{respired}$  and poorly crystalline metal oxide 471 abundance that was significant at  $\alpha = 0.1$ , but not at  $\alpha = 0.05$  (p = 0.07). The marginal 472 significance of this relationship is not surprising given that this regression considered 473 flux-weighted means, and the majority of the CO<sub>2</sub> flux came from the 0-0.1 m soil layer 474 where climate was the dominant control on  $\Delta^{14}C_{respired}$ . However, we did see a highly 475 significant relationship between poorly crystalline metal oxide abundance and  $\Delta^{14}C_{respired-bulk}$ 476 (Fig. 6, a). As with  $\Delta^{14}C_{bulk}$ , there was no relationship with crystalline Fe-oxide abundance 477 for either  $\Delta^{14}C_{respired}$  or  $\Delta^{14}C_{respired-bulk}$  (Fig. 6, b). 478

Discussion

Climatic and mineralogical factors both play key roles in soil carbon persistence.

However, the relevance of parent material for explaining soil organic matter persistence in
soils with mixed mineralogies is still poorly explained. In the current study we illuminate
how parent material interactions with climate lead to the development of distinct mineral
assemblages, which in turn control the dynamics of soil C cycling at timescales ranging from

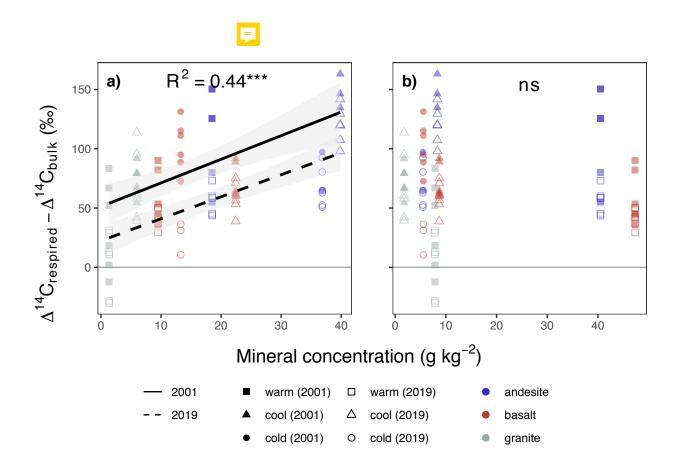


Figure 6. Relationship of poorly crystalline and crystalline minerals to the difference of  $\Delta^{14}C_{respired}$  and  $\Delta^{14}C_{bulk}$  ( $\Delta^{14}C_{respired-bulk}$ ). a) Poorly crystalline metal oxide content (oxalate-extractable aluminum + 1/2 oxalate-extractable iron), b) Crystalline Fe content (dithionite-extractable iron - oxalate-extractable iron). Points show mass-weighted mineral concentrations and carbon-weighted values of  $\Delta^{14}C_{respired-bulk}$  for 0-30cm profiles. Lines show linear model fits from Eq. 5.

annual to centennial. Our key findings are that soil mineral characteristics mediate climatic
controls on soil C cycling, and that mineralogical controls on soil C cycling are not limited to
soil C persistence on centennial timescales, but are relevant for C cycling on shorter
timescales as well.

Our results challenge the primacy of climatic controls on soil carbon persistence, 489 insofar as we observed the most depleted  $\Delta^{14}C_{bulk}$  in soils of the cool climate zone, not the 490 cold climate zone. We found that climate is a critical factor for explaining the dynamics of 491 both persistent and more transient soil C pools, as measured by proxy with  $\Delta^{14}C_{bulk}$  and 492  $\Delta^{14}C_{respired}$ , respectively. However, our results indicate that soil mineral characteristics 493 moderate the strength of the climate effect. This is particularly apparent at depth. These 494 results are supported by a recent study in which the authors used a depth-resolved model 495 with energy and substrate limitation prescribed by Michalis Menten kinetics to model 496 profiles of  $\Delta^{14}C_{bulk}$  (Ahrens et al. 2020). The authors found that site mean annual 497 temperature could only explain a minimal amount of variation in the observed radiocarbon profiles, while varying sorption potential in the model allowed for a superior fit.

Previous work at our study sites showed that poorly crystalline metal oxides were key to explaining both soil C accumulation and  $\Delta^{14}C_{bulk}$  values in these soils (Rasmussen et al., 2018). Our results confirm these findings and extend them to demonstrate a significant correlation between poorly crystalline metal oxide abundance and  $\Delta^{14}C_{respired-bulk}$  and a marginally significant correlation with  $\Delta^{14}C_{respired}$ . We focus here on  $\Delta^{14}C_{respired-bulk}$ , as this metric offers a unique insight into the magnitude of the difference between the cycling rates of persistent and more transiently cycling soil C.

We would expect the greatest differences between  $\Delta^{14}C_{respired}$  and  $\Delta^{14}C_{bulk}$  to be found in soils with a pool of old soil C protected from decomposition in some way, and another pool of soil C that is readily decomposed. In contrast, the smallest differences should occur for soils lacking strong soil C protection mechanisms, in which the majority of soil C has an

equal probability of being decomposed by microbes. Accordingly, we observed the smallest 511 differences between  $\Delta^{14}C_{respired}$  and  $\Delta^{14}C_{bulk}$  in the soils with the lowest concentration of 512 poorly crystalline metal oxides, while we observed the largest differences in the soils with the 513 highest concentrations of these minerals. We interpret these findings as providing evidence 514 for a key role of poorly crystalline metal oxides in protecting soil organic matter from 515 decomposition. The values of  $\Delta^{14}C_{bulk}$  in the subsurface layers (0.1-0.3 m) of the soils with 516 low poorly crystalline metal oxide content (e.g. granitic soils) were depleted relative to the 517 atmosphere, indicating persistent soil C, but we also observed similarly depleted values of 518  $\Delta^{14}C_{respired}$  in these soils. This suggests that soil C persistence in these soils is due to other 519 physical constraints on decomposition that are alleviated under laboratory incubation 520 conditions, for example transport or isolation [Gleixner et al., 2013]. 521

The soils with the largest difference between  $\Delta^{14}C_{respired}$  and  $\Delta^{14}C_{bulk}$  were those that 522 had both strongly depleted values of  $\Delta^{14}C_{bulk}$  and values of  $\Delta^{14}C_{respired}$  that were enriched 523 relative to the atmosphere. If we employ a theoretical compartmental model to interpret this 524 finding, a possible scenario would involve one pool of centennial to millennially cycling soil C 525 whose signal dominates the  $\Delta^{14}C_{bulk}$  signal, and second pool of soil C enriched with C fixed 526 from the atmosphere in the years immediately following the bomb-C spike, e.g. 1964 to 1990. 527 whose signal dominates the  $\Delta^{14}C_{respired}$  signal. More complex model structures could 528 potentially be fit to these data, but the data provide clear evidence for the presence of at 529 least two distinct pools with strongly depleted C largely inaccessible to the microbial 530 community, and another pool enriched with decadally cycling C dominating the respiration 531 flux. This stands in contrast to the soils with  $\Delta^{14}C_{respired}$  values close to that of the 532 atmosphere, in which the respired signal is likely dominated by sub-decadally cycling C. 533

Laboratory studies on soil organic matter associations with poorly crystalline metal oxides such as goethite show that only a portion of the organic matter is so tightly bound as to resist desorption (Kaiser and Guggenberger, 2003). Further studies have demonstrated

that a portion of the organic matter can be mobilized by exchange with DOC (Leinemann et al., 2018). In this scenario the highly depleted  $\Delta^{14}C_{bulk}$  values observed in the soils with high concentrations of poorly crystalline minerals may derive from organic matter that is strongly sorbed or trapped in micropores, while the bomb-C enriched decadally cycling C observed in the respiration flux could derive from the more microbially accessible and DOC-exchangeable mineral associated soil C pool. However, the rates of change over time that we observed for  $\Delta^{14}C_{respired}$  indicate that annually cycling C is also an important component of soil organic matter at all of our sites.

Our finding that parent material explains more of the variation in  $\Delta^{14}C_{respired}$  at depth than climate suggests the role of soil minerals in regulating annually to decadally cycling soil C is of particular importance in deeper soil layers. DOC has been shown to move downward 547 through the soil profile via preferential sorption of new soil C inputs and corresponding 548 desorption of older DOC, which is then available to the microbial community (Kaiser and 549 Kalbitz, 2012). This process is mineral controlled, and while we did not test it directly, such 550 a process could explain why we see an increased importance of parent material, and the 551 interaction between climate and parent material, for explaining  $\Delta^{14}C_{respired}$  trends with 552 depth. 553

In contrast to what we observed for the poorly crystalline metal oxides, the lack of correlation we observed between crystalline Fe oxides and soil radiocarbon suggests that these minerals do not play an important role in explaining soil C persistence, at least in these soils. Other studies have shown that crystalline Fe oxides do protect soil C from microbial decomposition, but that the overall sorption capacity of these mineral species is low [Kahle et al., 2003]. We observed a large increase in the amount of iron dissolved from crystalline minerals at the warm sites relative to the cool or cold sites, along with a corresponding decrease in soil C concentration and relative enrichment in both  $\Delta^{14}C_{bulk}$  and  $\Delta^{14}C_{respired}$ . This increase in crystalline iron was also associated with a corresponding

decrease in the amount of poorly crystalline minerals. Together these trends suggest that
these soils have lost poorly crystalline minerals through leaching and transformation into
crystalline species. The patterns of C concentrations and associated SOC stocks,  $\Delta^{14}C_{bulk}$ ,
and  $\Delta^{14}C_{respired}$  observed across the climate/weathering gradient suggest that weathering of
poorly crystalline metal oxides leads to a reduction in soil carbon stocks, which is caused by
losses of old  $\Delta^{14}C$  depleted carbon associated with these minerals, and that this process is
relevant across a range of igneous parent materials.

The sensitivity of decomposition to temperature is of particular interest for 570 understanding how soil C dynamics may change under a warming climate. Comparing the 571 change in  $\Delta^{14}C_{respired}$  over time for the different climate zones across the different lithologies 572 provides insight into this question. Focusing on the near surface soils (0-0.1 m), where 573 climate effects are strongest, we observed that the rate of change in  $\Delta^{14}C_{respired}$  over time 574 was correlated with poorly crystalline metal oxide content. We observed a linear relationship 575 between MAT and the rate of change in  $\Delta^{14}C_{respired}$  over time for the granitic soils, which 576 had low concentrations of poorly crystalline metal oxides in all three climate zones. In 577 contrast,  $\Delta^{14}C_{respired}$  changed more over time in the coldest climate zone for the basaltic and 578 andesitic soils than in the cool climate zone with higher concentrations of poorly crystalline 579 metal oxides. Furthermore, the change in  $\Delta^{14}C_{respired}$  over time was similar for all three 580 parent materials at the warm climate zone with minimal poorly crystalline metal oxide 581 content. These findings provide evidence that the temperature sensitivity of soil organic matter decomposition may be attenuated in the presence of poorly crystalline minerals. If this is true, we would expect that potential increases in decomposition rates, and 584 accompanying carbon losses due to climate warming, would be greater in soils lacking 585 substantial concentrations of poorly crystalline metal oxides than in soils enriched in these 586 minerals. 587

588 Conclusion

Our study shows clearly that parent material and climate interact to control soil C 580 dynamics. This interaction was the key to explain trends in  $\Delta^{14}C_{bulk}$ , which is a proxy for 590 the mean age of soil C, and additionally in  $\Delta^{14}$ C<sub>respired</sub>, which reveals the relative 591 contributions of faster or more slowly cycling soil C to respiration. We observed that the 592 trends in both  $\Delta^{14}C_{bulk}$  and  $\Delta^{14}C_{respired}$  were best explained by climate at the soil surface, 593 but best explained by parent material in deeper soil layers. The overall importance of 594 climate was reflected in the changes in  $\Delta^{14}$ C we observed over time, both for bulk C and respired CO<sub>2</sub>. We saw greater changes at the soil surface than in the deeper layers, and in the highly weathered soils of the warm climate sites than in the poorly developed soils of the 597 cold sites. Yet contrary to the expected temperature-decomposition relationship, we saw the 598 most depleted  $\Delta^{14}C_{bulk}$  in the cool climate zone with intermediate MAT, which is also where 599 we saw the least amount of change in  $\Delta^{14}$ C over time at all depths. 600

The effect of the interaction between parent material and climate on soil C dynamics is 601 best explained by the development of distinct mineral assemblages via weathering, in 602 particular the formation and subsequent loss of poorly crystalline metal oxides. We confirm 603 the findings of other studies showing that poorly crystalline minerals are highly correlated 604 with the age of soil C, as measured by proxy with  $\Delta^{14}C_{bulk}$ . We extend this finding to show 605 that it is specifically the poorly crystalline metal oxide content that is correlated with 606  $\Delta^{14}C_{bulk}$ , and that the correlation with crystalline Fe content is not significant. Furthermore, 607 we provide mechanistic evidence for the protective effect of mineral-association on decomposition of soil organic matter by demonstrating that the difference between  $\Delta^{14}C_{bulk}$ and  $\Delta^{14}C_{respired}$  is highly correlated with poorly crystalline mineral content. Finally, the 610 time series data of  $\Delta^{14}C_{respired}$  provides preliminary evidence that the association of soil 611 organic matter with poorly crystalline metal oxides may attenuate the temperature 612 sensitivity of decomposition. 613

# PARENT MATERIAL AND CLIMATE INTERACTIONS CONTROL SOIL C DYNAMIGS

In future work we intend to quantify the timescales of soil carbon cycling in mineral and non-mineral associated pools with a compartmental modeling approach, using the radiocarbon time series presented here in addition to radiocarbon measurements of soil density and thermal fractions as constraints. With this modeling framework we also hope to test additional hypotheses regarding mineral association and temperature sensitivity of soil organic matter.