## Escuela Rafael Díaz Serdán Ciencias y Tecnología: Química

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Educación para la vida

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3° de Secundaria

Unidad 3 2022-2023

Preparación para Examen de la Unidad 3

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## Tabla Periódica de los Elementos Químicos

|                 | 1             |                            |              |               |              |               |              |               |              |               |              |               |             |
|-----------------|---------------|----------------------------|--------------|---------------|--------------|---------------|--------------|---------------|--------------|---------------|--------------|---------------|-------------|
| 1<br>H<br>1.008 |               |                            |              |               |              |               |              |               |              |               |              |               |             |
| 3               | 4             | 1                          |              |               |              |               |              |               |              |               |              | 5             | 6           |
| Li<br>6.941     | Be<br>9.012   |                            |              |               |              |               |              |               |              |               |              | B<br>10.81    | C<br>12.01  |
| 11              | 12            | 1                          |              |               |              |               |              |               |              |               |              | 13            | 14          |
| Na<br>22.990    | Mg<br>24.305  |                            |              |               |              |               |              |               |              |               |              | Al<br>26.981  | Si<br>28.08 |
| 19              | 20            | 21                         | 22           | 23            | 24           | 25            | 26           | 27            | 28           | 29            | 30           | 31            | 32          |
| K<br>39.098     | Ca<br>40.078  | Sc<br>44.956               | Ti<br>47.88  | V<br>50.941   | Cr<br>51.996 | Mn<br>54.938  | Fe<br>55.847 | Co<br>58.933  | Ni<br>58.69  | Cu<br>63.546  | Zn<br>65.39  | Ga<br>69.72   | G<br>72.0   |
| 37              | 38            | 39                         | 40           | 41            | 42           | 43            | 44           | 45            | 46           | 47            | 48           | 49            | 50          |
| Rb<br>85.468    | Sr<br>87.62   | Y<br>88.906                | Zr<br>91.224 | Nb<br>92.906  | Mo<br>95.94  | Tc<br>(98)    | Ru<br>101.07 | Rh<br>102.905 | Pd<br>106.42 | Ag<br>107.868 | Cd<br>112.41 | In<br>114.82  | Sı<br>118.  |
| 55              | 56            | 57                         | 72           | 73            | 74           | 75            | 76           | 77            | 78           | 79            | 80           | 81            | 85          |
| Cs<br>132.905   | Ba<br>137.33  | *La<br>138.905             | Hf<br>178.49 | Ta<br>180.948 | W<br>183.85  | Re<br>186.207 | Os<br>190.2  | Ir<br>192.22  | Pt<br>195.08 | Au<br>196.967 | Hg<br>200.59 | Tl<br>204.383 | P           |
| 87              | 88            | 89                         | 104          | 105           | 106          | 107           | 108          | 109           | 110          | 111           | 112          | 113           | 11          |
| Fr<br>(223)     | Ra<br>226.025 | <sup>†</sup> Ac<br>227.028 | Rf<br>(261)  | Db (262)      | Sg<br>(263)  | Bh<br>(262)   | Hs<br>(265)  | Mt<br>(266)   | Ds<br>(271)  | Rg<br>(272)   | Cn           | Nh            | F           |

|             | 58      | 59      | 60                  | 61      | 62                  | 63     | 64                  | 65                  | 66                  | 67      | 68     |
|-------------|---------|---------|---------------------|---------|---------------------|--------|---------------------|---------------------|---------------------|---------|--------|
| *Lantánidos | Се      | $\Pr$   | $\operatorname{Nd}$ | Pm      | $\operatorname{Sm}$ | Eu     | $\operatorname{Gd}$ | $\operatorname{Tb}$ | Dy                  | Но      | Er     |
|             | 140.12  | 140.908 | 144.24              | (145)   | 150.36              | 151.96 | 157.25              | 158.925             | 162.50              | 164.930 | 167.26 |
|             | 90      | 91      | 92                  | 93      | 94                  | 95     | 96                  | 97                  | 98                  | 99      | 100    |
| †Actínidos  | Th      | Pa      | U                   | Np      | Pu                  | Am     | Cm                  | $\operatorname{Bk}$ | $\operatorname{Cf}$ | Es      | Fm     |
|             | 232.038 | 231.036 | 238.029             | 237.048 | (244)               | (243)  | (247)               | (247)               | (251)               | (252)   | (257)  |

## Vocabulario

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