# Physics 231

Outline for Day 1
Attendance and a list of units
Discuss syllabus
Units & Measurements

# P231 Week 1: measurements

#### Goals of Week 1:

- Learn about base and derived units
- Learn dimensions and dimensional analysis
- Understand measurements and their errors
- Learn how to propagate errors
- > Understand how  $\mu$ ,  $\sigma$ , and  $\sigma_{\mu}$  are related to measurements and errors

#### P231 Week 1: measurements

# Units

#### **Base Units**

#### **Derived Units**

Mechanical

Quantity	MKS unit	cgs unit	
mass	ko (kilogram	kg (kilogram) g	

length m (meter) cm time s (second) s

Other

miles/hour km/s mol/liter kg m/s<sup>2</sup> etc. etc.

etc.

Quantity
temperature
current
amount of matter
luminous intensity cd (candela)

#### P231 Week 1: measurements

# **Dimensions**

<u>dimension</u>: "the manifoldness with which the fundamental units of time, length, and mass are involved in determining the units of other physical quantities."

dimension: a measureable quantity



For mechanical base units ...

For some derived units ...

Mechanical

Quantity	Dimension
mass	M
length	$\mathbf{L}$
time	T

$$[miles/hours] = L/T$$

$$[km/s] = L/T$$

$$[knot] = L/T$$

$$[L (liter)] = L^3$$

$$[kg m/s^2] = ML/T^2$$

# **Dimensional Analysis**

- a way to figure out if an equation is correct
- allows you to decide which equation to use.

Ex. 1) Is this equation dimensionally correct?

$$ma = \frac{1}{2} mv^2$$

where m=mass, v=speed, a=acceleration=L/T<sup>2</sup>

Soln:  $[ma]=ML/T^2$  and  $[\frac{1}{2} mv^2]=ML^2/T^2$  since  $ML/T^2 \neq ML^2/T^2$  the equation cannot be correct.

Ex. 2) Is this equation dimensionally correct?

$$y = at^2$$

where y=position (L), t=time, a=acceleration=L/T<sup>2</sup>

Soln: [y]=L, [at²]=L/T² \* T²=L. since L = L, the equation is dimensionally correct. However, the equation is still wrong! How?

<sup>†</sup>1/2 is a dimension less constant

# **Dimensional Analysis** (cont)

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Ex. 3) How long does it take to drive 20 miles (to Lima) at a constant 60 mph? Soln: Let v=speed (L/T), d=distance (L) and t=time (T). Possible (linear) equations: t=v*d, t=v/d, t=d/v check dimensions: L^2/T 1/T T so: t=d/v = 20/60 = 1/3 hr or 20 minutes.
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measurement: the act or result of measuring

Example: use a plastic ruler to measure a shoe's length to be  $L=12.0\pm0.1$  inches.

Example: use a Vernier caliper to measure the same shoe length to be  $L=12.13\pm0.04$  inches.

#### Notice:

- •A measurement consists of a *number*, an *error* (or uncertainty, or tolerance), and a *unit*. 3 things!
- •The number of significant digits shown is related to the error.
- •The caliper is more *precise* than the ruler.
- •It is uncertain which measurement device is more accurate.

Accuracy and precision

- i. <u>accuracy:</u> how close the a measurement is to some accepted "true" value
- ii. <u>precision:</u> how repeatable the measurement is with a given

instrument and technique

# Measurements -accuracy and precision

Example: two bathroom scales.

Step on and off them repeatedly in a consistent way.

digital scale analog (yellow) scale

155.1 lbs 150. lbs

155.0 148

155.1 149

155.2

155.3

Q: Which scale is more precise?

Q: Which scale has the greater "spread" in values?

Q: Which scale is most accurate?

You go to the doctors office and they tell you 157.2 lbs.

Q: Which scale is most accurate?

Q: Which scale is more precise?





Significant figures or (significant digits)

-- a way of suggesting precision.

<u>significant figure</u>: any digit of a number that is known with certainty; any digit of a number beginning with the leftmost non-zero digit and ending with the rightmost non-zero digit (or a zero considered to be the exact value.

#### **Examples:**

- 1) 4,567,000 4
- 2) 4.567 0 5
- 3) 4,567,000 6
- 4) 4,567,000. 7
- 5) 0.03450 4
- 6) 30.003

#### Notes:

- 1. The number left of a decimal point is significant.
- 2. errors should have 1 significant figure.
- 3. For homework after week 1, give answers to 3 or 4 significant digits.
- 4. The <u>least significant digit</u> (LSD) is the rightmost significant digit.
- 5. the weights from the yellow scale should not be quoted to more than the 1's place.

Which digit is the LSD in each of the above? Which place is occupied by the LSD in the above?

**Errors** (uncertainty, tolerance)
-- the best way to quantify precision.

How do you determine an error on a measurement?

- a) From the number of significant figures?
  Not good. There is NO universally accepted rule for deriving errors from significant digits.
  - Ex.) engineers say 32.4 means 32.4+-0.05
- b) By looking at the smallest "tickmarks" on your instrument. Tolerance is usually ½ of the tickmark spacing.
- c) By considering how difficult it is to use the instrument. Ex. using a stopwatch.
- d) By repeating the measurement many times and finding the spread of measurements. ("standard deviation") BEST!

Mistake, not error.

## Measurements

**Errors** types of errors

### random, independent errors

- related to the <u>precision</u> of the measurement
- the type determined in the previous slide
- depends on the number of measurements



#### systematic errors

- related to the <u>accuracy</u> of the measurement
- an effect that shifts all measurements away from the true value. Ex) the zeropoint was set wrong on the yellow scale. If the zero point was increase by 5 lbs, it would give accurate readings
- use your best understanding and judgement of the measurement process to estimate.
  - Ex) you are measuring the volume of a balloon.
    - You understand that the answer will depend on the pressure and temperature inside and outside of the balloon.

**Errors** ways of mathematically expressing errors

#### absolute errors

-- 155 +- 8 lbs has an absolute error of 8 lbs

### fractional errors

-- 155 +- 8 lbs has a fractional error of 0.052

#### percentage errors

-- 155 +- 8 lbs has a percentage error of 5.2%

#### **Error Propagation**

How do you figure out the error for a number that was calculated from numbers with errors?

I. If only significant figure are shown:



- a) Addition and subtraction: the final answer should have its LSD in the same place as the least precise input measurement
  - Ex) 5800 m + 121 m = 5900 m
  - Ex) 612800 s + 2011.5 s = 614,800 s
  - Ex) 220. 115 = 105
- b) Multiplication and division: the final answer should have the same number of sig figs as the input number with the fewest sig. figs.
  - Ex)  $2000 \times 15.143 = 30,000$
  - Ex)  $382,500 \times 11. = 4,200,000 \pmod{4,207,500}$
  - Ex)  $520 / 3 = 200 \pmod{173.3}$

### Error Propagation - cont.

- II. If errors are explicitly shown
- a) Addition and subtraction:
  - 1) simple way: add error

Ex) 
$$580.\pm 2 \text{ m} + 121 \pm 3 \text{ m} = 701.\pm 5$$

(This is an overestimate.)

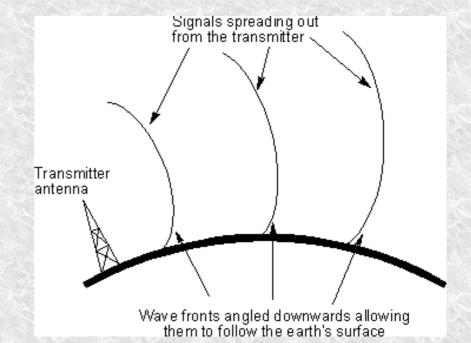
2) correct way: add errors "in quadrature"

Ex) 
$$580.\pm 2 \text{ m} + 121 + -3 \text{ m} = 701.\pm \text{ e}$$

where 
$$e = \sqrt{(2)^2 + (3)^2} = \sqrt{13} = 3.61$$

- b) Multiplication and division:
  - 1) simple way: "adding the percentages" Ex) Hwk Problems 2-5 ...
  - 2) correct way: add fractional errors in quadrature. (We will use the Simple way instead.)

Note: the LSD of the answer must match the LSD of the error! Note: the number of sig figs in the final answer does not have to be the same as the least precise input number, ala prev slide.



#### **Errors and statistics**

Mean 
$$\mu = \frac{\sum x_i}{N}$$

Standard Deviation 
$$\sigma = \sqrt{\frac{\sum (x_i - \mu)^2}{(N-1)}}$$

Standard Deviation of the mean

$$\sigma_{\mu} = \frac{\sigma}{\sqrt{N}}$$

#### Normal or Gaussian distribution

