# C-oxidation BDE Energy Report for: 189_lovastatin-out

This report covers the results for BDE calculations performed for: 189_lovastatin-out. Oxidation propensity is established using C-H Bond Dissociation Enthalpies (BDE). The lower the C-H BDE values the higher the propensity for C-oxidation. Details for the DFT calculations and overall workflow are explained at the end of this document

Bond Dissociation Energies (kcal/mol)
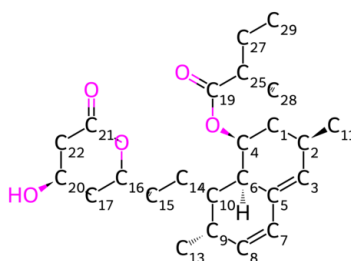
Atom	BDE (kcal/mol)	Propensity
C1	97.15	Low
C2	70.82	High
C3	109.16	Low
C4	97.4	Low
C6	81.17	High
C7	109.92	Low
C8	109.41	Low
C9	77.94	High
C10	94.92	Low
C11	101.68	Low
C13	100.63	Low
C14	97.31	Low
C15	97.87	Low
C16	91.0	Moderate
C17	97.52	Low
C20	91.22	Moderate

C22	91.66	Moderate
C25	85.89	High
C27	96.25	Low
C28	101.69	Low
C29	100.34	Low

Risk Scale:

kcal/mol   88   94

| High | Medium | Low |
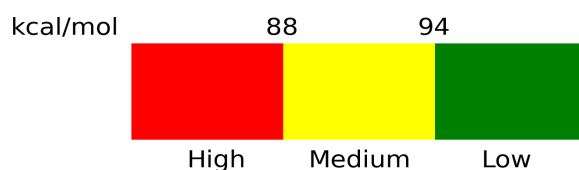
Calculation details and output files

Conformational search calculations were performed only for the base ground state molecule. The lowest energy conformer was selected to generate radicals and run optimization DFT calculations. DFT calculations were performed using Gaussian with B3LYP level of theory and 6-31G(d,p) basis set. The BDE protocol was adapted from: *Lienard, P., Gavartin, J., Boccardi, G., & Meunier, M. (2015). Predicting drug substances autoxidation. Pharmaceutical research, 32, 300-310.*