

■■■■■ EXPERIMENTAL INVESTIGATION OF A SOLAR LiCl DRYER SYSTEM

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ABSTRACT

Solar LiCl air dryer system has been investigated theoretically and experimentally. It consists of three parts: the main dryer part with liquid desiccant solution (LiCl) cycle, the solar heating system with sensible and latent heat stores, and the ground source.

The dryer system includes dehumidifier to decrease air humidity and the regeneration section that was designed to increase the concentration of LiCl solution. For the experiments, the supply heat energy in the heat storage tank was an electrical heater that provides the required energy similar to the solar heating system with PCM material investigated by the author (1). Also, cold water was used for the experiments instead of the ground source for cooling the LiCl solution in the heat exchanger before entering the dehumidifier.

In this paper, the experimental results will be presented. The theory of the experimental setup with the solar heating system with PCM material and the ground source is investigated.

Keywords: Solar energy, Phase Change material (PCM) tank, Liquid desiccant solution (LiCl) Ground source.

1. INSTRUCTION

Air conditioning systems consume a considerable amount of energy. Renewable energy like solar energy would be a good choice to be used in the thermodynamic cycle. Crofoot and Harisson (2), Andrusiak, Harrison, (3) considered using solar air conditioning systems.

The main part of the whole cycle is the dryer part and also choosing the working fluid in the dryer cycle.

According to Liu et al. (4) finding, in the dehumidification process, LiCl compared to LiBr in the same condition (the same vapor pressure and inlet temperature, rather than the same crystal temperature) had a better mass transfer performance in the same desiccant mass flow rate condition. they reported that the COP of using LiCl and LiBr in the cycle is similar. Additionally, initial cost of LiCl solution price is 18% lower than LiBr.

Khan and Martinze (5) developed a simple model to predict the performance of liquid desiccant absorber. They showed that the simple model had a good agreement with the detailed model.

Phase Change Material tanks have been used extensively in various literature. Ismail (7) investigated a numerical model to simulate a latent storage tank composed of spherical capsules filled with PCM placed inside a cylindrical tank filled with a working fluid circulation system to charge and discharge energy in the heat storage tank. The influence of charging and discharging time of the PCM tank were investigated numerically and experimentally. Water Heating System was also improved with the use of Phase Change Materials (8-9).

Ground source can be used for cooling the working fluid temperature. Bernardi et al. (10) investigated that it is useful to improve a cycle and increasing COP of whole cycle by using ground source.

In the current paper LiCl dryer has been investigated experimentally and theoretically combined with solar heating system with PCM tank and also ground source cycle.

2. Experimental Investigation

1. Governing equation

1.1. Dehumidifier and Regeneration

According to Liu et al. (4) paper, the equivalent humidity ratio of liquid desiccant ω_e is illustrated with the equilibrium status of air, as shown in Eq. (1).

$$\omega_e = 0.622 P_s B - P_s$$

Eq. 1

Moisture removal rate in the mass transfer performance of the dehumidifier or regenerator is shown in Eq. (2).

$$m_w = m_a \omega_{a,in} - \omega_{a,out}$$

Eq. (2)

Eq. (3) expresses the relation of moisture removal rate in the dehumidifier. Additionally, the regressed moisture removal rates using LiCl are shown in Eq. (4). Meanwhile, in Eq. (5), the volumetric mass transfer coefficient is illustrated.

$$m_w \text{LiCl, dehum} = 9.4542 \times 10^8 \times m_a 0.406 \omega_a 2.2478 \cdot m_s, in 0.6499 \cdot t_s, in - 2.3911 \cdot \xi_{in} 1.7919$$

Eq. 3

$$m_w \text{LiCl, regen} = 6.602 \times 10^{-8} \times m_a 0.1649 \cdot \omega_a - 0.4617 \cdot m_s, in 0.4818 \cdot t_s, in 3.223 \cdot \xi_{in} - 2.9341$$

Eq. 4

$$h_{ma} \text{LiCl, regen} = 4.6645 \times \nu_a 0.6562 \times \nu_s 0.5694 \times \phi_{e, in} - 0.2280$$

Eq. 5

In the regeneration or dehumidifier, on the analytical solution of the heat and mass transfer procedure in packed-bed, the mass transfer efficiency ε_m at diverse flow patterns is the function of heat capacity ratio of air to desiccant m^* and mass transfer unit NTU_m , as shown in Eq. (6), Eq. (7) and Eq. (8) investigated in (4) and (11-12).

$$\varepsilon_m = \omega_{a,in} - \omega_{a,out} / \omega_{a,in} - \omega_{e,in} = f(m^*, NTU_m)$$

Eq.6

$$m^* = C_{p,e} \cdot m_a C_{p,s} \cdot m_s = C_{p,e} \cdot \rho_a \cdot V_a C_{p,s} \cdot \rho_s \cdot V_s$$

Eq. 7

where $C_{p,e} = dh/dt_s$

$$NTU_m = h_{mma} = h_{ma} \cdot V_{ma}$$

Eq. 8

The higher mass transfer efficiency is brought the moisture removal rate as shown in Eq. (9).

$$m_w = \varepsilon_m \cdot m_a \omega_{a,in} - \omega_{e,in}$$

Eq. 9

Eq. (10) infers that the coefficient of performance of the liquid desiccant system is determined by the cooling capacity of the processed air Q_a divided by the regeneration heat supplied into the regenerator Q_{hot} .

$$COP = Q_a / Q_{hot} = \frac{m_a \cdot (\omega_{a,in} - \omega_{a,out}) C_{p,wh}}{m_{wh} \cdot (t_{wh,in} - t_{wh,out})}$$

Eq.10

1.2. Solar thermal system

In the following equations, the solar collector efficiency and PCM tank efficiency are shown.

1) Solar collector: In the Eq. (13), solar collector efficiency is calculated where the solar collector heat transfer rate is calculated in Eq. (12).

$$Q_u = m_{ccw} T_{ic} - T_{oc}$$

Eq.(12)

$$\eta_c = Q_u / A_c G_c$$

Eq.13

2) PCM tank: For PCM tank efficiency shown in Eq. (16), PCM heat transfer rate (Eq. (14)) was divided by the maximum possible heat transfer rate that the PCM tank is able to save. In addition, data logger receives experimental data every 30 minutes. Therefore Eq. (15) needs to be divided by the period of 30 minutes in order to convert it to kW.

$$Q_s = m_{tcw} T_{it} - T_{ot}$$

Eq. 14

$$Q_p = m_{pcm} c_{pcm} T_{pcm} - T_a$$

Eq. 15

$$\eta t = 30 \times 60 Q_s Q_p$$

Eq. 16

2.1.2 Ground source temperature estimation

According to previous research investigation (13), T_g is the undisturbed ground temperature expressed in °K. It can be calculated as:

$$T_g X_s, t = T_g - A_s \exp(-X_s \pi 365 \alpha) \cos(2\pi 365 t - t_0 - X_s 2365 \pi \alpha - 321.8 + 273)$$

Eq. 17

where X_s , t , T_g , A_s , α and t_0 are the soil depth in feet, the day of year, annual surface soil temperature, the annual surface temperature amplitude ($T_{\max} - T_{\min}$), the soil thermal diffusivity¹, and a phase constant expressed in days, respectively. Eqs. (18-19) are the minimum and maximum ground temperatures for any depth, respectively.

$$T_{g,\min} = T_g - A_s \exp(-X_s \pi 365 \alpha)$$

Eq. 18

$$T_{g,\max} = T_g + A_s \exp(-X_s \pi 365 \alpha)$$

Eq. 19

According to Kavanaugh et al. (14), for vertical ground source systems, X_s can be fixed equal to the average depth in Eq. (18) and Eq. (19). Temperature can be estimated approximately as equal to the mean annual surface soil temperature, T_g .

¹ $\alpha = k / \rho C_p$ where k is the thermal conductivity in BTU/ hr lb °F, ρ is the density in lb / ft³ and C_p is the specific heat in BTU/ lb °F

For calculating ground heat exchanger length Eq. (20) in the Ground Source Heat Pump (GSHP) Project, Model of International Ground Source Heat Pump Association (IGSHPA) (13) is utilized as the method. The required Ground Heat Exchanger (GHX) length L_c establish on cooling necessity:

$$L_c = \frac{q_{d,cool}}{COP_c + 1} \frac{COP_c R_p + R_s F_c T_{ewt,max} - T_g,max}{T_g,max}$$

Eq. 20

Where COP_c is the design cooling coefficient of performance (COP) of the heat pump system, F_c is the part load factor for cooling, R_s is the soil thermal resistance, R_p is the pipe thermal resistance, $T_{g,max}$ is the maximum undisturbed ground temperature, and $T_{ewt,max}$ is the maximum design entering water temperature at the heat pump.

The method that the model used in the Natural Resources Canada (RETScreen) GSHP Project Model used to model the COP and the capacity as a function of the entering fluid temperature uses a quadratic polynomial correlation:

$$COP_{actual} = COP_{baseline} K_0 + K_1 T_{ewt} + K_2 T_{ewt}^2$$

Eq. (21)

$$K_0 = 1.53105836$$

$$K_1 = -2.29609500 \times 10^{-2}$$

$$K_2 = 6.87440000 \times 10^{-5}$$

Where COP_{actual} is the actual COP of the heat pump, $COP_{baseline}$ is the nominal COP of the heat pump (e.g. measured at standard rating conditions, 0°C for heating and 25°C for cooling), k_i are correlation coefficients listed and Q_c is the capacity of the heat pump for cooling or heating.

The maximum and minimum design temperature of entering water are pinto out in Eqs. (22, 23) :

$$T_{ewt,\min} = T_{g,\min} - 1.5 \text{ oF} - 321.8$$

Eq. 22

$$T_{ewt,\max} = \min(T_{g,\max} + 20 \text{ oF}, 110 \text{ oF}) - 321.8$$

Eq. 23

Because the model was also designed to be used in permafrost, the 20°F minimum entering water temperature limitation was not implemented.

The part load factor (F) is calculated by divided the full load hours during the design month by the total number of hours in that month, as seen by the GHX Eq. (24). It can be evaluated as:

$$F = \frac{q_{\max}}{24} \quad \text{Eq. 24}$$

Where q_{\max} and q are the peak load for the month and the average load respectively.

3. Method

In the dryer cycle, LiCl as the working fluid circulated in the dehumidifier to absorb air humidity. To have LiCl with high concentration, regeneration part is designed in the dryer cycle. Heat resource is provided by electrical heater. Later, instead of the electric heater, solar collectors and PCM tank were used theoretically. Fig. 1 shows a schematic diagram of the LiCl dryer with solar heating system which consists of 3 parts:

1. Solar collector and heat storage tank:

During the day, the solar collectors are exposed to the solar radiation and thus produce hot water. Hot water is divided between the heat storage tank (HST) and PCM tank (Fig. 2). Heat storage tank is a heat exchanger that transfers the heat energy from solar collectors to dryer cycle. Additionally, dryer cycle does not need constant heat energy because sometimes the percentage of concentration of the solution is enough not to regenerate.

During day, solar collectors produce heat energy in the PCM tank and can be discharged during night in dryer cycle when it is demanded. Additionally, thermo physical properties of the PCM are shown in TABLE. 1.

TABLE. 1: THERMO PHYSICAL PROPERTIES OF THE PCM (PARAFFIN).

Melting point	55-59 °c
Heat storage capacity	178 (Kj/kg)
Solid Density	900 (m3/kg)
Liquid Density	770(m3/kg)
Thermal conductivity	2. (W/m.K)

3.

2. Drying system with LiCl working fluid:

4.

5. Whole dryer system is shown in Fig. 3. where in the dehumidifier part, LiCl solution splashes on the cellulose pad and a fan blows ambient air on the cellulose pad guided by the duct channel. Therefore, dried air leaves the pad outlet to be used in air conditioning system. In the regeneration part, Heat Storage Tank (HST tank) is applied between the solar collector and regeneration part of the dryer cycle. It should be mentioned that when LiCl concentration decreases, valves number 12, 19 will be closed and valves 13, 18 are opened to regenerate in the regeneration). The hot LiCl solution (valve 13) enters the cellulose pad and splashes on it by the Pump to create high LiCl concentration. When the system has enough LiCl concentration, valves 12 and 19 are opened and valves 13 and 18 are closed until the LiCl concentration becomes low. The LiCl concentration is controlled manually by measuring the LiCl and opening and closing the valves.

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8.



9.

10. Fig. 2: HST and PCM tank.

11.

12.

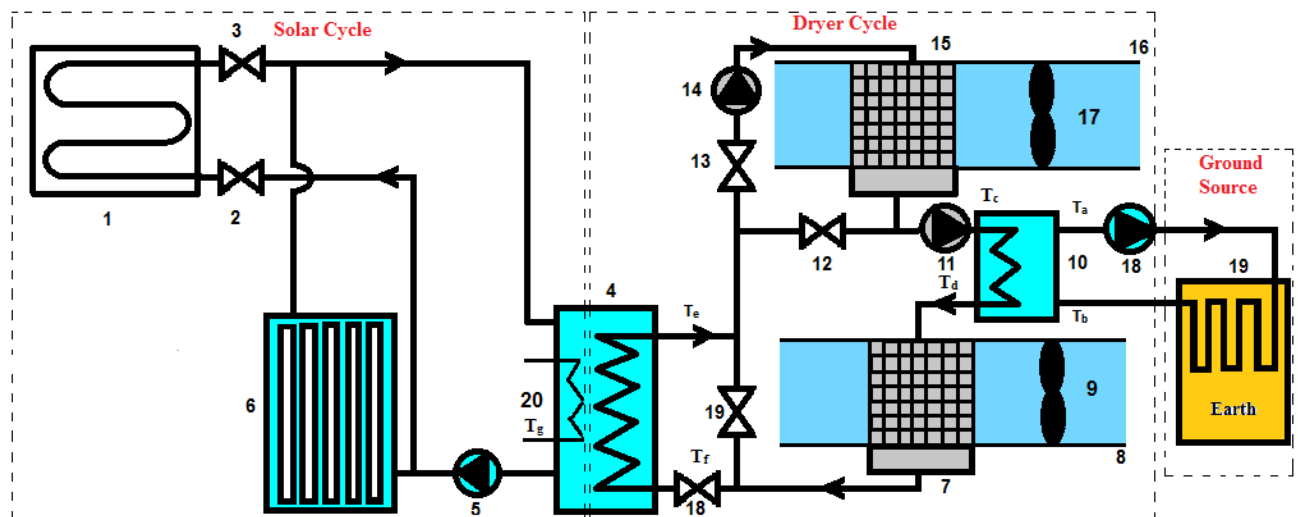


13. Fig. 3: Dryer system with Liquid desiccant LiCl.

14.

15.

16.



17. Number

18. Name

19.

20. Number

21. Name

22. 1

23. Collector

24.

25. 8, 16

26. Air Channel

27. 2, 3,

28. Valve

29.

30. 9

31. Fan

12, 13,1 8,19			,		
			1		
			7		
32. 4	33. Heat storage tank (HST)	34.	35. 1 0	36. Heat exchanger B	
37. 5, 11, 14,1 8	38. Pump	39.	40. 1 9	41. Ground Source	
42. 6	43. PCM tank	44.	45. 2 0	46. Electrical Heater	
47. 7, 15	48. Cellulose pad	49.	50.	51.	

52. Fig. 1: Whole solar dryer cycle with use of ground source.

53.

54. TABLE 2: LICL PROPERTIES.

55. Air		56. LiCl solution		57. Wa	58. Fluid
59. 3 0	60. 7 0	61. 3 0	62. 70	63. 20	64. Temperatu re (°C)
65. 1 . 8	66. 0 . 9 9 8	67. 1 2 6 0	68. 126	69. 100	70. Density (kg/m ³)
71. 1 . 0 0 5 8	72. 1 . 0 0 9	73. 2 . 7	74. 2.8	75. 4.1	76. Specific heat (KJ/kg °C)
77. 1 . 7 4 6 × 1 0 ⁻⁵	78. 2 . 0 7 × 1 0 . 5	79. 3 . 5 × 1 0 ⁻³	80. 2.5×10 ⁻³	81. 10.006×10 ⁻⁵	82. Viscosity(P a.Sec)

83. 0	84. 0	85. 0	86. 0.8	87. 0.5	88. Thermal Conductivity
.	.	.			
0	0	5			
2	3	5			
6					89. (W/m.K)
90. 0	91. 0	92. -	93. -	94. 7.0	95. Prandtl Number
.	.				
7	6				
0	9				
8	7				

96.

97.

3. Using Geo source for cooling LiCl solution.

98.

99. If LiCl liquid desiccant temperature decreases, it can absorb much more humidity, therefore, ground source system which has water as a working fluid is used. It can be used in the heat exchanger B in the dryer cycle. The dryer system would be effective because it does not need to have cooling tower in the cycle especially when the cycle is in the humid climate. Finally, LiCl solution with high concentration comes back in dehumidifier part after heat exchanger B outlet and it can absorb the air humidity.

100.

101.

102.

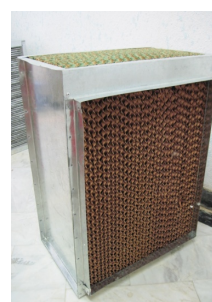
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105.

106.

107.



108. Fig. 4: Cellulose pad and the heat exchangers used in LiCl dryer system.

4. Experimental test

109.

110. Humidity, temperature and air velocity in LiCl dryer are measured by SAMWON model SU-50313, model TESLINK RS-232 and Model Smart Sensor AR846 respectively.

111.

112. The solar thermal system had different setup located in a different location. Solar system was set up in the latitude of 35.39, longitude of 53.2 and altitude of 1127.29m. In this experimental work, solar radiation is measured using Kipp & Zonen pyranometer. The data logger has LM35 temperature sensors with the temperature accuracy of $\pm 0.1^\circ\text{C}$.

113.

5. Results and Discussion

114.

115. The desired air mass flow rate assumed to be 0.54 kg/s and according to Eqs. (3) and (4), the LiCl

solution flow rate in the dryer cycle is calculated to be 0.23 kg/s depending on the moisture removal and also moisture content created by dryer the cycle. The parameters measured in this experiment are inlet and outlet temperature, humidity and flow rates both in the air side, water side and LiCl solution side. Table 3 shows the measured data. The result shows that the ambient air humidity can be decreased well from 66% and temperature of 32°C down to 30% of humidity with 35°C of the delivered air. There is no significant change in the air temperature.

116.

117. TABLE 3. DATA OF Whole LiCl DRYER AT SAME TIME

118.Ambient air temperature	119
	32°
120.Ambient air humidity	121
	66
122.Outlet air temperature in dehumidifier	123
	32.
124.Outlet air humidity in dehumidifier	125
	30
126.Humidity air	127
	66
128.Flow of inside air the humidifier and Regeneration part	129
	0.5
130.Outlet humidity air in regeneration	131
	80
132.Outlet temperature air in regeneration	133
	35°

134.LiCl solution flow 135
0.2

136. Inlet water heat exchanger 137
B(T_b) 18°

138. Outlet water heat exchanger 139
B(T_a) 29°

140.Water in heat exchanger B 141
0.1

142.Inlet LiCl solution heat exchanger B(T_c) 143
50°

144.Outlet LiCl solution heat exchanger B(T_d) 145
41°

146.Inlet LiCl heat storage tank(T_f) 147
40°

148.Outlet LiCl heat storage tank (T_g) 149
52

150.Average water temperature inside heat storage tank 151
58

152.

153. According to Table 3, the measured heat storage tank temperature with water was measured to be 58°C. The current heat storage tank had an electric heater. However, solar thermal system can be used to provide the required temperature. Auxiliary supply system with electric heater can still be connected to the system.

However, less energy can be used to heat up and keep the heat storage tank constant. Therefore, solar thermal system with PCM material is introduced to the system. According to Fig. 5, solar radiation, inlet and outlet temperature of solar collector are shown in a specific sunny day.

154.

155. The inlet and outlet solar collector temperatures are changed with delay because of absorbing energy by PCM tank material that release the heat smoothly to the water tank heat store in order to stabilize the temperature.

156.

157.

158. Fig. 5: Inlet and outlet temperatures from the solar collector (to the left), solar radiation (to the right).

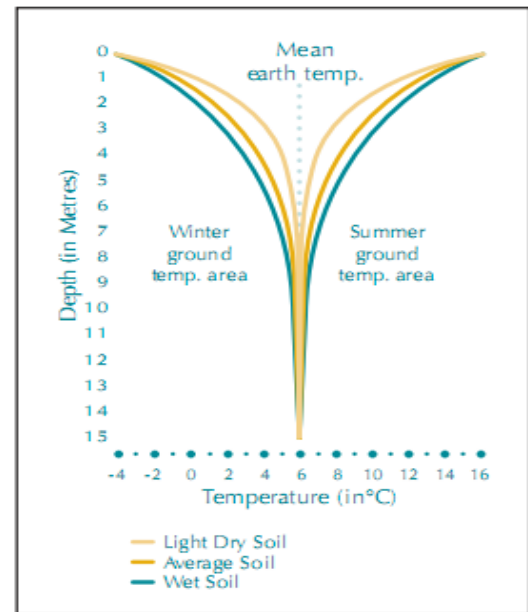
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161. Fig. 6: Heat storage tank temperature versus the time (hr).

162. The minimum temperature for heat exchanger B inlet temperature in the LiCl dryer cycle is 18°C . Fig. 7 illustrates typical soil temperature variation reported by Kavanaugh and Rafferty (14). If the depth of the hole is higher, the soil temperature will decrease and finally become constant at 6°C . Therefore, ground source can be used in the heat exchanger B side in order to provide temperature less than 18°C . Ground source can be used especially in the humid climate where instead of the cooling tower, ground source can be used.

163.



164. Fig 7: Typical soil temperature variation (14).

165. _____

6. Conclusion

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167. Dryer system with LiCl solution as a working fluid can be investigated experimentally. The system showed that the dry ambient air with 0.54 kg/s from 66% to 30%.

168. The experimental investigation was carried out to test the PCM tank heat storage and water heat storage tank in order to use them for regeneration of the LiCl solution. The PCM tank can store solar energy during day and can deliver heat to the water heat storage tank for longer time during night, especially when the regeneration part is used in the dryer cycle during evening and night.

169.

170. In humid climate, it is difficult to benefit from cooling tower in the dryer cycle. Therefore ground source can be used in the dehumidification process.

171.

172.

173.					216.	217.	Mass	218.	219.	Soil
					ε_m	transfer efficiency,	%	A	thermal diffusivity	
7.	<u>Nom</u>	176.	177.		220.	221.	Relati	222.	223.	Phase
	<u>encl</u>				Φ	ve humidity, %		t_o	constant expressed in	days
	<u>ature</u>				224.	225.	Humi	226.	227.	Minim
174.					Ω	dity ratio, kg/kg		$T_{g,min}$	um ground	temperature, k
175.					228.	229.	Mass	230.	231.	Maxim
178.	179.	Mass	180.	181.	PCM	flow rate, kg/s		$T_{g,max}$	um ground	temperature, k
A	transfer area, m ²		C_{PCM}	Specific heat, kJ/						
				(kg°k)						
182.	183.	Specif	184.	185.	Melting*	232.	233.	234.	235.	Ground
a	ic area of the		T_{pcm}	g temperature PCM,		Heat	capacity ratio of air	Lc	heat exchanger length	
	packing per volume,	m ² /m ³		k		to desiccant,	dimensionless			
					236.	237.	Moist	238.	240.	design
186.	187.	Atmo	188.	189.	Air	ure removal rate,		COP_h	cooling coefficient of	performance (COP)
B	spheric pressure, Pa		T_a	temperature, °k	m_w	kg/s		239.	of the heat pump	system
190.	191.	Specif	192.	193.	PCM					
C_p	ic heat capacity, kJ/	(kg°k)	η_t	tank efficiency						
194.	195.	Entha	196.	197.		241.	242.	243.	244.	part
H	lpy, kJ/kg		T_g	ound	NTU _m	transfer unit,	dimensionless	F_c	load factor for	cooling
				198.	245.	P	246.	V	247.	R
				temperat			apor pressure,	s		248.
				ure			Pa			oil thermal
					249.	Q	250.	C	251.	R
				199.			ooling	p		252.
							capacity or			pipe thermal
							heating			resistance
200.	201.	Coeffi	202.	203.	Soil		capacity, kw			
COP	cient of performance	of the liquid	X_s	depth, m						
	desiccant system,				253.	t	254.	T	255.	T
							emperature, °C	g_{max}		256.
204.	205.	Conv	206.	207.	The					he maximum
h_m	ective mass transfer	coefficient, kg/(m ² s)	T	day of year, k						undisturbed
										ground
										temperature, k
208.	209.	Volu	210.	211.	Annual	257.	v	258.	V	259.
h_{ma}	metric mass transfer	coefficient, kg/(m ³ s)	T_g	surface soil	temperature, k			olume of the	ewt,max	T
								packed-bed		260.
								module, m ³		maximum
212.	213.	Super	214.	215.	The					design
N	ficial velocity, m/s	A_s		annual surface	temperature, k					entering water
				amplitude (T _{max} -T _{min}), m ²	261.	ξ	262.	M	263.	264.
							ass	OP _{actual}	C	he actual COP
							concentration			

[illegible]

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