Tutorial Sheet-7, Physics-2 (15B11PH211), Even Semester 2019-20

- 1. Explain the following terms: (i) spontaneous emission, (ii) stimulated emission, (iii) metastable state, (iv) population inversion, and (v) optical pumping. [CO1]
- 2. (a) A certain ruby laser emits 1.00-J pulses of light whose wavelength is 694 nm. What is the minimum number of Cr^{3+} ions present in ruby laser? (b) In a Ruby laser, total number of Cr^{3+} ions is 2.8×10^{19} . If the laser emits radiation of wavelength 7000 Å. Calculate the energy of the laser pulse. [Ans: (a) 3.5×10^{18} ions, (b) 7.94 J] [CO2]
- 3. Find the ratio of populations of two states in a laser that produces light of wavelength 6328 Å at 27°C. [Ans: 1.1×10⁻³³] [CO2]
- **4.** Find the ratio of spontaneous emission to stimulated emission for a cavity of temperature 50 K and wavelength is 10⁻⁵ m. [Ans: 3.16×10¹²] [CO2]
- **5.** A 10 mW laser has efficiency of 1%. Suppose all input energy is utilized in pumping atoms from the ground state to the excited state, which is 20 eV above the ground state. Find how many atoms are promoted to the excited state in one second. [Ans: 3.12×10^{17}] [CO3]
- 6. The coherence length of sodium light is 2.945×10^{-2} m and its wavelength is 5890 Å. Calculate (i) the frequency, (ii) the number of oscillations corresponding to the coherence length, and (iii) the coherence time. [Ans: (i) 5.09×10^{14} Hz, (ii) 5×10^{4} , and (iii) 9.82×10^{-11} sec.]
- 7. The ruby laser has two states at 27°C and 227°C. If it emits radiation of wavelength 7000 Å, then calculate the ratio of relative population. [Ans: 1.25×10⁻¹²] [CO3]

Constants: $h = 6.62 \times 10^{-34} \text{ J-s}, k = 1.38 \times 10^{-23} \text{ J/K} = 8.61 \times 10^{-5} \text{ eV/K}.$