

Chapter 6

Understanding Process Conditions

Chemical Engineering Department
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Heuristics

- Temperature

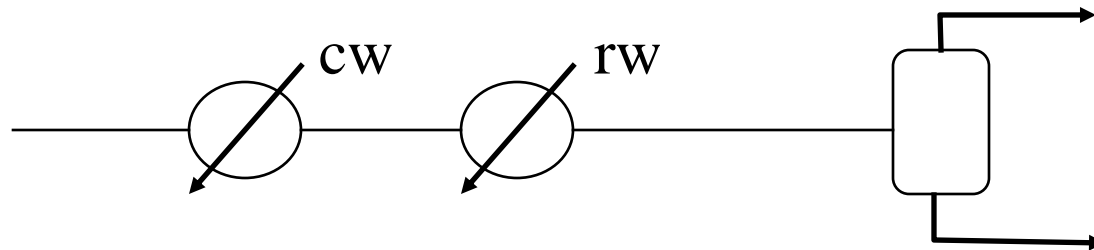
< 40°C	> 250°C	> 400°C
Require Refrigeration	Require Fired Heater	Special M.O.C

- Pressure

< 1 atm	> 10 atm
need Vacuum	Thick Walls - \$

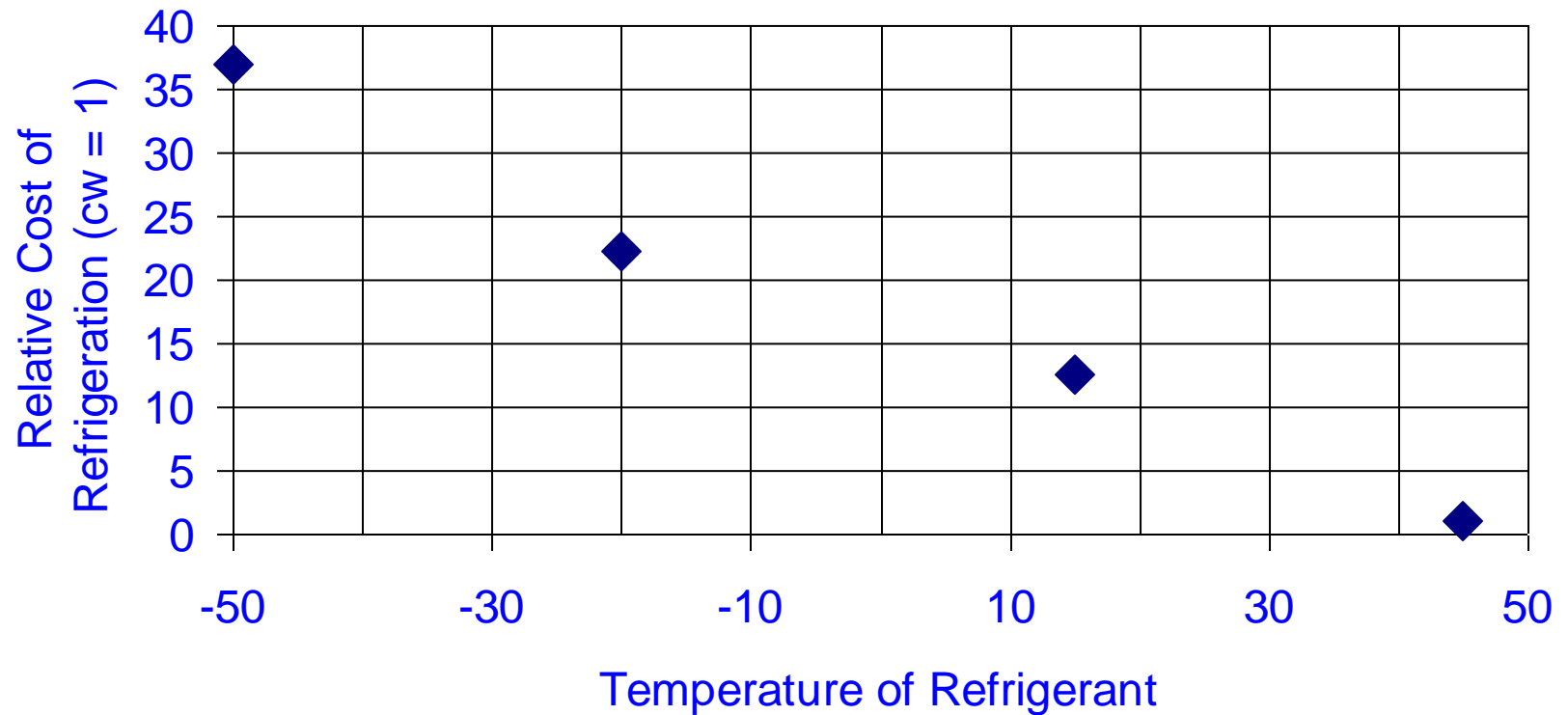
Temperature

- $T < 40^{\circ}\text{C}$ – Refrigeration
 - Use as much cooling water as Possible



- Operating Costs (Table 8.3)
 - Cooling Water (30-40°C) \$0.354/ GJ
 - Refrigerated Water (5-15°C) \$4.43/ GJ

Temperature



Temperature

- $T > 250^{\circ}\text{C}$ – hp steam @ 260°C (600 psig)
 - Need a molten salt / Dowtherm loop
 - Fired Heaters are very expensive
 - Compare - vaporizer

$$Q = 10,000 \text{ kW}$$

$$U = 1000 \text{ W/m}^2\text{K}$$

$$\Delta T = 30^{\circ}\text{C}$$

$$A = \frac{(10 \times 10^6)}{(10^3 \times 30)} = 333 \text{ m}^2$$

Temperature

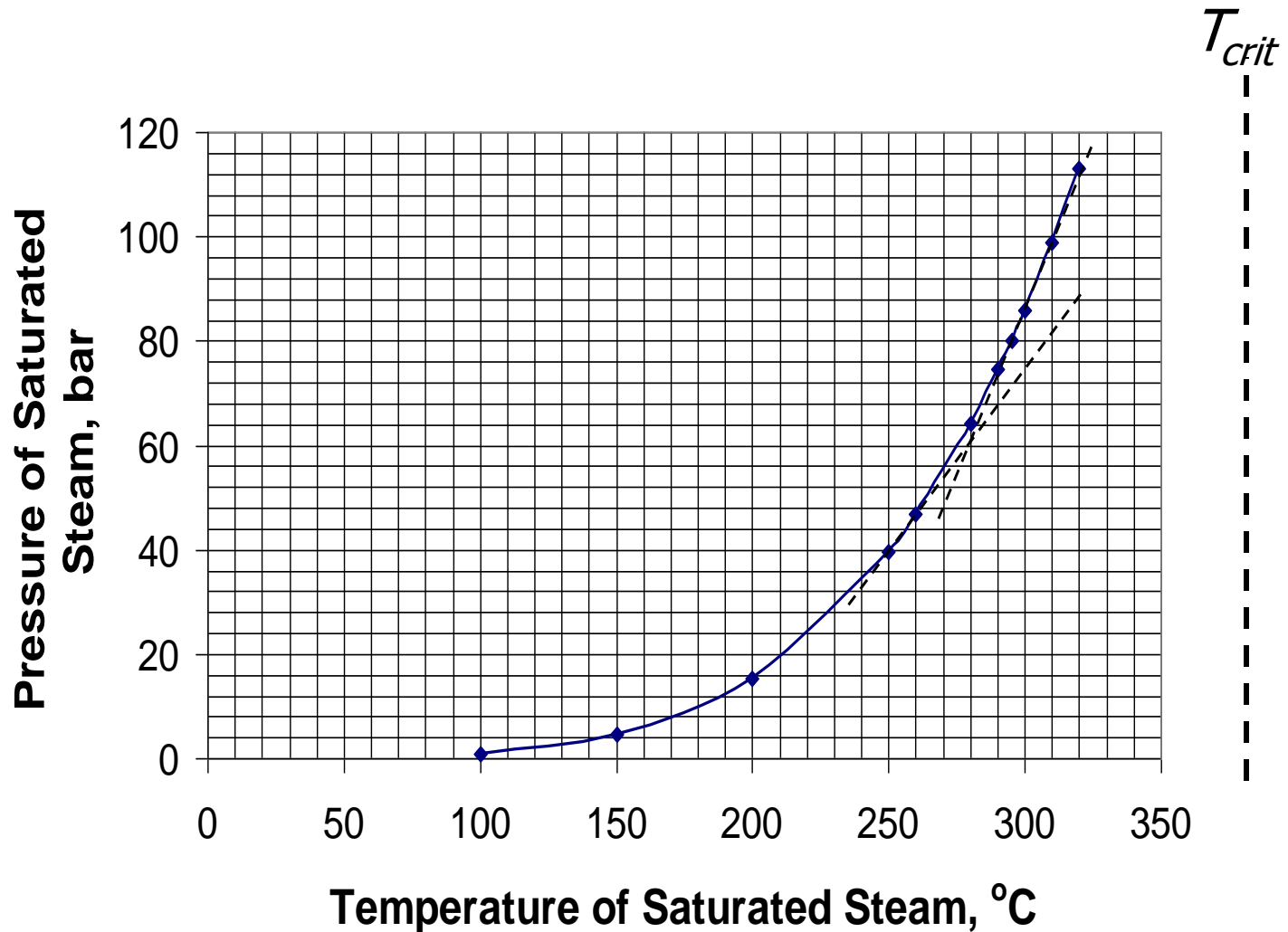
- C_{BM}
 - Heat Exchanger = \$ 1.70×10^5
 - Fired Heater = \$ 1.81×10^6
- $T > 400^\circ\text{C}$
 - M.O.C. is very Important

Why not use High-Pressure Steam?

P_{sat} (bar)		T_{sat} (°C)
15.2		200
39.7		250
46.9		260
64.2		280
86.0		300
74.5		290
80.1		295
98.8		310
113.0		320

Graph of Saturated
Steam vs. Pressure

Why not use High-Pressure Steam?



Pressure

- Vacuum
 - Slightly Higher Cost due to Stiffening Rings
 - Large Equipment
 - Air Leaks
- High Pressure
 - Thick Walls - \$
 - H₂ Embrittlement
 - Safety

Minimum Wall Thickness

$$t = \frac{PR}{SE - 0.6P} + CA$$

Wall thickness, t = m , design pressure, P = bar , vessel radius, R = m

S = Design Stress (Max Allowable Working Pressure, bar) this is a function of material and temperature

E = Weld Efficiency(~ 0.9)

CA = Corrosion Allowance (0.00315 to 0.00625 m)

Minimum Wall Thickness

- Look at 36 inch Diameter Vessel with a CA of $\frac{1}{4}$ in made of CS with $S = 13,700$ psi

<u>P</u>	<u>t (m)</u>	<u>t / CA</u>
14.7 = 1 barg	0.0069	1.09
58.8 = 4 barg	0.0085	1.34
147 = 10 barg	0.0118	1.86

- As $P > 10$ then $t > CA$

What About S vs. T ?

- Look at Several Steels in Graph
- For CS $S \downarrow$ as $T > 400^\circ\text{C}$
- Must use Stainless Steel and $S \uparrow$
- For a given Pressure
 - $t \uparrow$ as $T \uparrow$

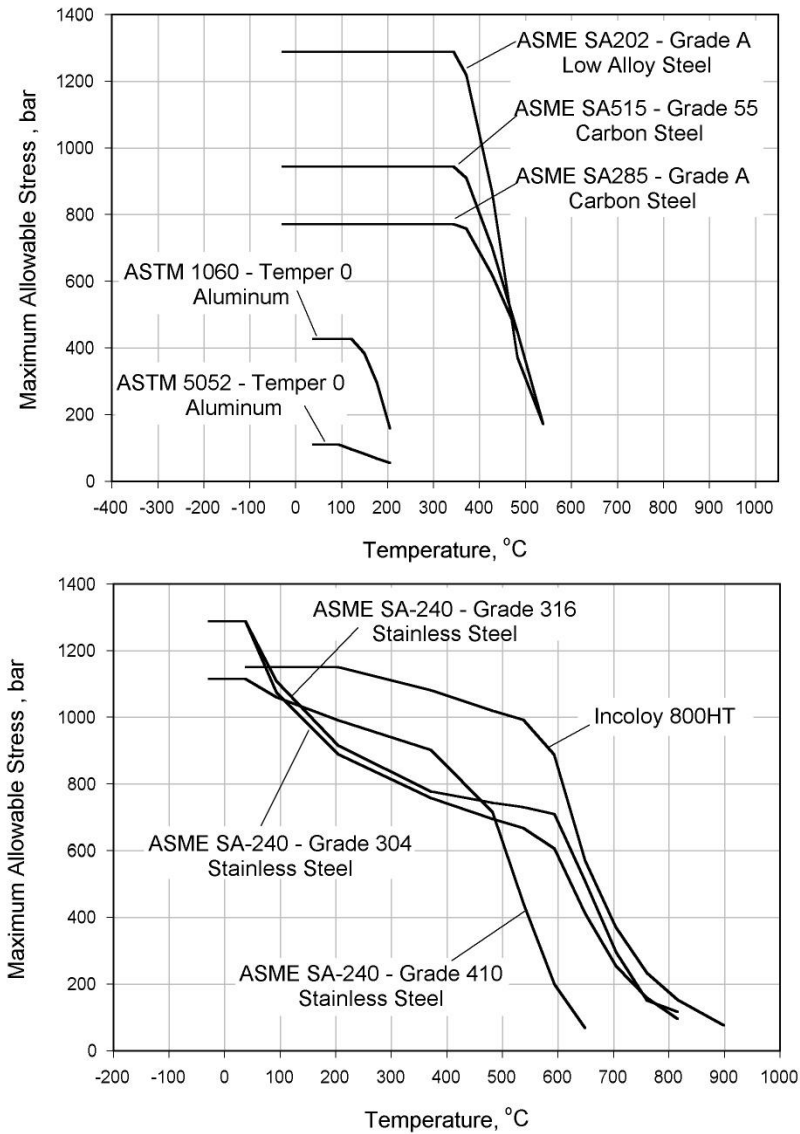
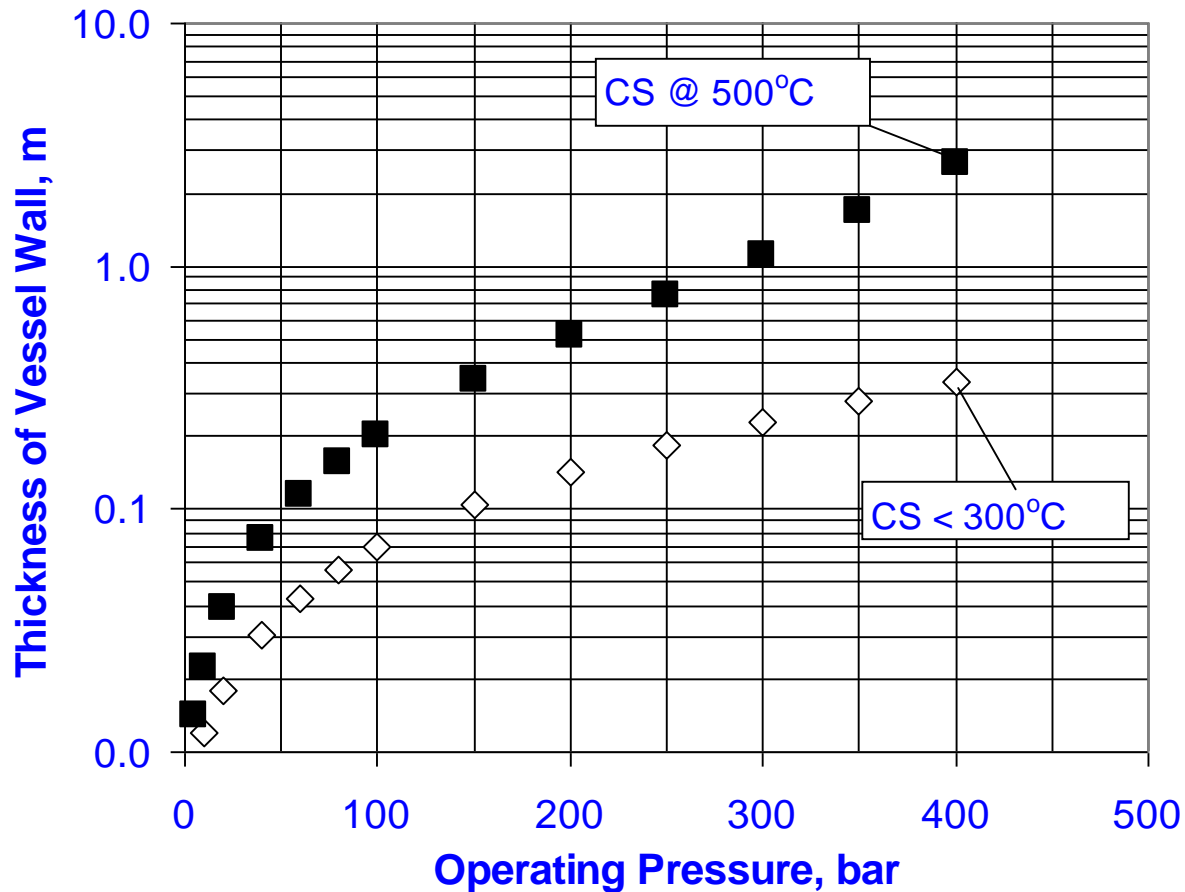


Figure 5.5: Maximum Allowable Stresses for Materials of Construction as a Function of Operating Temperature (Data from Perry et al. [3], Chapter 10 and Ref [15])

Material of Construction

1 m Diameter Vessel made of SA285 - Grade A Carbon Steel



Material of Construction

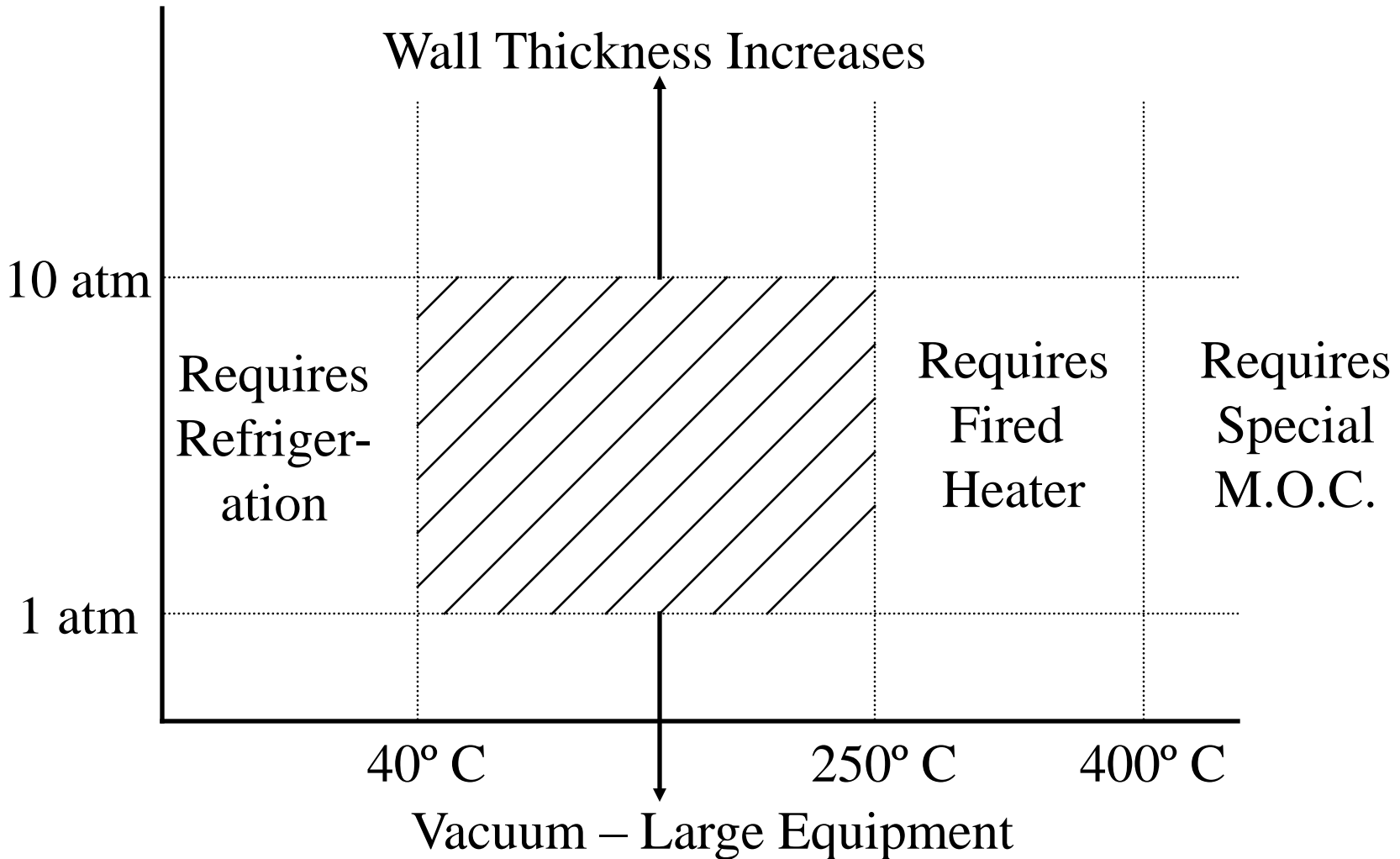
- Carbon Steel
 - Cheap
- Stainless Steel
 - Expensive
 - Better Chemical/Thermal Resistance
- What About $T = 700-900^{\circ}\text{C}$?
 - Insulate inside of Pipe
 - Metal – Refraction Lining

Conclusions

- $T < 40^{\circ}\text{C}$ – Refrigeration
- $T > 250^{\circ}\text{C}$ – Fired Heater or Furnace
- $T > 400^{\circ}\text{C}$ – M.O.C. Issues
- $P < 1 \text{ atm}$ – Vacuum and Large Equipment
- $P > 10 \text{ atm}$ – Cost



Operating Conditions



Do we ever operate outside these limits?

- Tables 6.1 – 6.3
 - Reactors and Separators
- Table 6.4
 - Other Equipment

Examples

- Example 1 – Acrylic Acid
 - Appendix B.9
 - Why does T-305 Operate with the top pressure at 0.07 bar?
 - Feed – 86.6 kmol/h Acrylic Acid – $nbp = 140^{\circ}\text{C}$
 - 6.1 kmol/h Acetic Acid – $nbp = 118^{\circ}\text{C}$

Table 6.2 – Reasons for using $P < 1$ atm

1. Obtain a gas phase for VLE
2. Temperature sensitive materials

Examples

- Example 2 – Separation of Propane
 - Typical depropanizer operates at 220 psig (16 bar)
 - why?

Table 6.2 – reasons for using $P > 10$ bar

1. Obtain a liquid phase for VLE