# 第五單元蛋白質功能 - 酵素 Enzymes

- 5.1 酵素的特性與類型 Enzyme properties and classification
- 5.2 酵素催化反應的能量變化 Free energy change in enzyme catalyzed reaction
- 5.3 酵素動力學 Enzyme kinetics
- 5.4 酵素動力學 兩個受質的反應 Bisubstrate reactions
- 5.5 酵素與抑制物 Enzyme and inhibitors
- 5.6 調控酵素活性的方式 Allosteric regulation

## 學習目標:

- 1. 瞭解酵素的特性
- 2. 暸解酵素催化反應過程的能量變化圖
- 3. 暸解酵素動力學 (kinetics)
- 4. 暸解 Inhibitor 的種類及對 V<sub>max</sub>, K<sub>m</sub> 的影響
- 5. 熟悉調控(regulation)酵素活性的方式

## 天堂筆記:

- 1. Properties of enzymes
  - Higher reaction rate
  - Milder reaction condition
  - Greater reaction specificity
  - Capacity for regulation

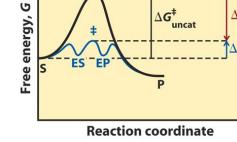


Figure 1 Reaction coordinate diagram.

- 2. Enzyme as catalyst (催化劑)
  - Ribozyme (catalytic RNA)
  - Protein
    - Holoenzyme (holoprotein) = Prosthetic group + Apoenzyme (apoprotein)
      - Prosthetic group: coenzyme (organic molecule) or cofactor (metal ion).

#### 3. Enzyme classification

Classification	Type of Reaction Catalyzed	
1. Oxidoreductase, 氧化還原酶	Oxidation-reduction reactions, A- + B \improx A + B+	
2. Transferase, 轉移酶	Transfer of functional groups, $A-B+C \Longrightarrow A+B-C$	
3. Hydrolase, 水解酶	Hydrolysis reactions, A-B + H₂O    ← A-H + B-OH	
4. Lyase, 裂解酶	Group elimination to form double bonds, $X \stackrel{Y}{\longrightarrow} A=B + X-Y$	
5. Isome <u>r</u> ase, 異構酶	Isomerization, $\stackrel{X}{\underset{A-B}{\bigvee}}\stackrel{Y}{\longleftarrow}\stackrel{X}{\underset{A-B}{\bigvee}}$	
6. Ligase, 接合酶, (synthetase)	Bond formation coupled with ATP hydrolysis, A + B $\Longrightarrow$ A-B	

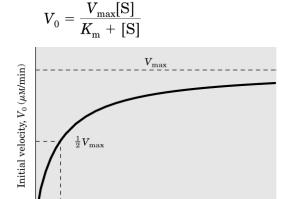
- 4. Enzymes stabilize transition state (Figure 1)  $\rightarrow \Delta G^{\dagger} \downarrow \rightarrow$  Reaction rate  $\uparrow$ 
  - E: enzyme; S: substrate; ES: E-S complex
  - Standard free energy change (△G, 自由能變化) → Equilibrium 平衡
  - Activation energy (△G<sup>‡</sup>, 活化能) → Reaction rate 反應速率
  - Binding energy ( $\triangle G_B$ ): result from multiple weak E-S interactions.

### 5. Enzyme kinetics 酵素動力學

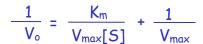
- Enzyme kinetics 瞬素動力学 **E** 又稱 steady-state kinetics, Michaelis-Menten kinetics  $E + S \rightleftharpoons E + P$
- Assumptions:
  - When  $[S] \gg [E]$ ,  $V_0$  is determined at the beginning of the reaction when the [P] is infinitely small and the back reaction  $(S \leftarrow P)$  can be ignored.
  - [ES] is constant
- Hyperbolic plot 雙曲線 (Vo vs [S])
  - $K_m = [S]$ , when  $V_o = \frac{1}{2} V_{max}$
  - $V_o = V_{max}$ , when [S] >>  $K_m$
  - $V_o \propto [S]$ , when  $[S] \ll K_m$

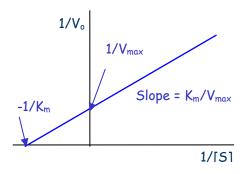
#### 6. Lineweaver-Burk plot, double reciprocal plot, 1/V<sub>o</sub> vs 1/[S]

- X-intercept:  $1/V_{max}$
- Y-intercept: -1/K<sub>m</sub>



Substrate concentration, [S] (mM)





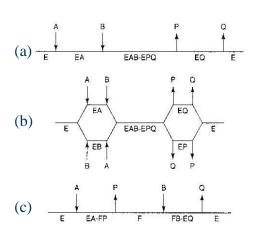
7. Characteristics of an enzyme:

 $K_{
m m}$ 

- K<sub>m</sub>: affinity of enzyme for substrate
- *k*<sub>cat</sub>: turnover number 轉換數,
  - k<sub>cat</sub> = V<sub>max</sub> / [E<sub>T</sub>], 最大反應速率/總酵素濃度
  - The number of  $S \rightarrow P$  in a given unit of time when the E is saturated with S.
  - Used to compare catalytic efficiency for different substrates of the same enzyme
- $\blacksquare$   $k_{\text{cat}}/K_{\text{m}}$ : the specificity constant,
  - Used to compare catalytic efficiency of different enzymes

#### 8. Bi-substrate and Bi-product reaction (Bi-Bi reaction)

- $\blacksquare$  A + B  $\leftrightarrow$  P + Q, catalyzed by enzyme E
- Sequential displacement (form ternary complex)
  - Compulsory order (ordered Bi-Bi)
    - (a) LDH (pyruvate + NADH  $\rightarrow$  lactate + NAD<sup>+</sup>)
  - Random order (random Bi-Bi)
    - (b) Creatine kinase (ATP + Cr  $\rightarrow$  PCr + ADP)
- Ping-Pong reaction (double displacement reaction, no ternary complex)
  - (c) Aminotransferases



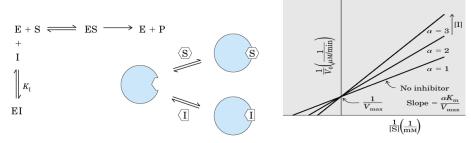
- 9. Irreversible inhibitor (destroy the active site):
  - Group specific reagent: react with specific R-group of a.a.
    - DIFP (diisopropylphosphofluoridate) inhibits chymotrypsin (serine protease), and acetylcholinesterase.
  - Suicide inhibitor (mechanism-based)
    - Penicillin →acts by covalently modifying transpeptidase (suicide inhibitor)

#### 10. Reversible inhibitor (I, 抑制物)

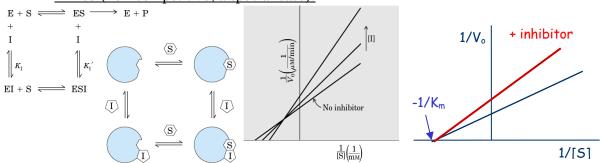
Inhibitor type	Binding site on enzyme	Kinetic effect
Competitive	Specifically at the catalytic (active site).	V <sub>max</sub> unchanged,
	Inhibitor has a similar structure as the substrate.	K <sub>m</sub> increased
	Inhibition is reversed by substrate	
Noncompetitive	I binds E or ES complex other than the active site.	V <sub>max</sub> decreased,
(mixed type)	Inhibition can not be reversed by substrate.	K <sub>m</sub> unchanged
Uncompetitive	I binds only to ES complex other than the active site.	V <sub>max</sub> decreased,
	Inhibition can not be reversed by substrate.	K <sub>m</sub> decreased

## Competitive:

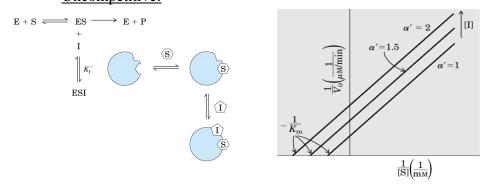
- Methanol vs ethanol and alcohol dehydrogenase
- CO vs O<sub>2</sub>



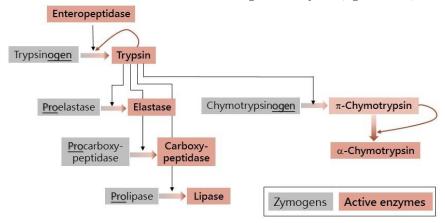
Mixed (non-competitive, a special case):



#### Uncompetitive:



- 11. Regulation of enzyme activity
  - pH, temperature
  - Control of enzyme availability
    - Rate of synthesis
    - Rate of degradation
    - Differential subcellular expression
  - Regulation type:
    - Isoenzyme, or isozyme (caused by gene duplication)
      - ♦ Properties:
        - Enzyme catalyze the same reaction
        - ♦ Different kinetic property: K<sub>m</sub>, V<sub>max</sub>
        - ♦ Different physical property: molecular weight
        - Different chemical property: amino acid sequence
        - ♦ Different tissue specificity: location
      - ♦ L-lactate dehydrogenase: lactate + NADH → pyruvate + NAD+
        - ♦ H. heart and M. muscle
        - ♦ LDH (a tetramer): HHHH, HHHM, HHMM, HMMM, MMMM
    - Allosteric (non-covalent)
      - ♦ Hemoglobin; Aspartate transcamoylase (ATCase)
    - Covalent modification
      - ♦ Phosphorylation-dephosphorylation (kinase-phosphatase)
    - Peptide bond cleavage (proteolytic cleavage)
      - ♦ Zymogen, proenzyme, proprotein (inactive precursor) → active enzyme
      - ♦ Digestive enzymes (proteases)
        - ◆ Trypsin formed by enteropeptidase (master activation step)
          - Coordinated control of digestive enzymes (figure below):



- ♦ Proteins in the blood-clotting cascade
  - Serine proteases: enzymes has serine in the active site
  - ♦ Zymogen: factors circulate in blood as *inactive form*
  - ♦ *Cascade* reaction:
    - $\diamond$  1<sup>st</sup> factor (small amount)  $\rightarrow$  end reaction (large amount)
    - ♦ Rapid response → limiting blood loss
- ♦ Other examples:
  - ◆ Procaspase → caspase, (programmed cell death, apoptosis)
  - ♦ Proinsulin → insulin (hormone)
  - ◆ Procollagen → collagen (Procollagenase → collagenase)
  - ♦ Timed tissue remodeling in development
    - ♦ Metamorphosis of a tadpole into a frog
    - ♦ Mammalian uterus after delivery

### 魔咒關鍵詞:

Holoenzyme (holoprotein) = Prosthetic group + Apoenzyme (apoprotein)

Prosthetic group, coenzyme or cofactor

Enzyme (E), Substrate (S), Enzyme-Substrate complex (E-S complex), Transition state

Standard free energy change ( $\triangle G$ ), Equilibrium

Activation energy ( $\triangle G^{\ddagger}$ )

Binding energy ( $\triangle G_B$ )

Enzyme kinetics (steady-state kinetics, Michaelis-Menten kinetics)

Hyperbolic plot, Vo, Km, Vmax

Lineweaver-Burk plot (double reciprocal plot)

 $k_{\rm cat}, k_{\rm cat}/{\rm K_m}$ 

Reversible inhibitor (I):

Competitive, Uncompetitive, Non-competitive

Regulation

Isoenzyme (isozyme)

Allosteric enzyme (non-covalent)

Covalent modification

Peptide bond cleavage (proteolytic cleavage)

Inactive precursor: zymogen, proenzyme, proprotein

Cascade reaction

## 魔法參考書目:

- 1. 台大莊榮輝教授教學網頁: http://juang.bst.ntu.edu.tw/BCbasics/index.htm
- **2.** Lehninger Principles of Biochemistry (2013), 6th ed, David L. Nelson, and Michael M. Cox, Freeman and Company, New York.
- **3.** Principles of Biochemistry (2013) 4th ed. Voet, Voet, and Pratt. Wiley.
- 4. Biochemistry, a short course. (2015) John L. Tymoczko, Jeremy M. Berg, Lubert Stryer (3<sup>rd</sup> ed) W.H. Freeman & Company.

#### 魔法練習題:

- 酵素對化學反應的反應平衡、反應速率、自由能變化(ΔG,)、活化能(ΔG<sup>\*</sup>)等參數有何影響?(增加、減少、或不變)。
- 2. 生化學家研究一個遵循Michaelis-Menten kinetic的酵素,得到下列數據:

Substrate conc.	Initial velocity
[S], μ <b>M</b>	V <sub>o</sub> (μmole/min)
1	49
2	96
8	349
50	621
100	676
1,000	698
5,000	699

- (a) 請作Michaelis-Menten 的圖 (反應速率V。對受質濃度[S]作圖)。
- (b) 請作Lineweaver-Burk 的圖 (1/Vo 對1/[S]作圖)。
- (c) 請問這個酵素的最大反應速率Vmax 是多少?Km是多少?
- (d) 加入抑制物X後再測量,得到 $V_{max}$  為700  $\mu$ mole/min,  $K_m$  為10  $\mu$ M, 請問X是屬於哪一類型的抑制物?這種抑制物的特徵為何?
- 3. 調控酵素活性的方式有哪些?