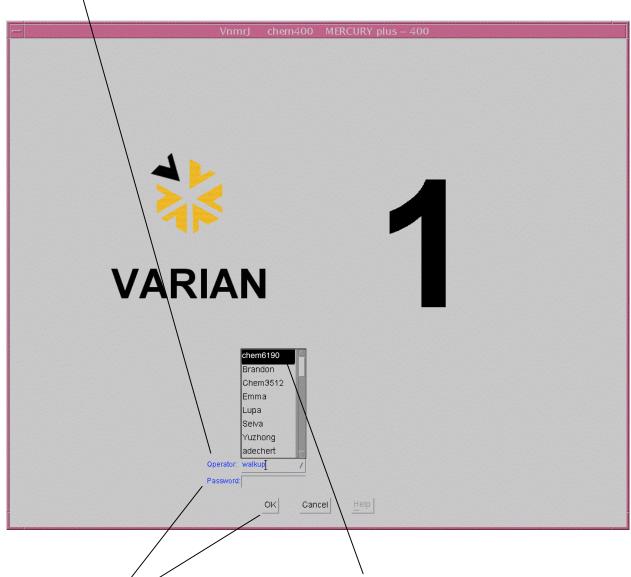
Lab #1 1D ¹H NMR Spectroscopy

- operation of the 400 MHz instrument using automated sample insertion (robot) and automated locking and shimming
- collection of 1D ¹H spectra
- retrieving data, peak picking, peak integration, plotting

<u>Login</u>

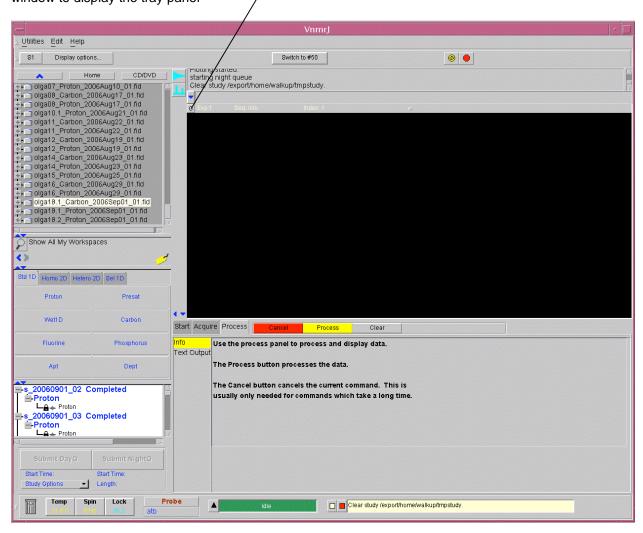
-the **operator** screen should be in view when you first sit down at the spectrometer console:



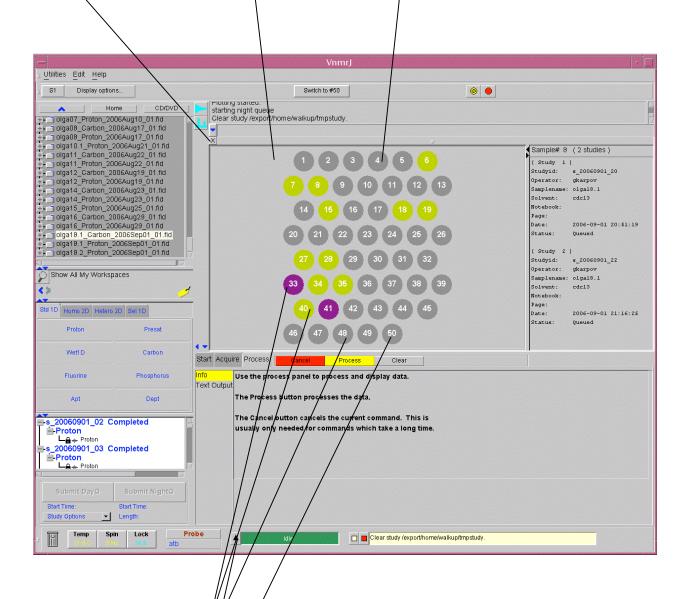
- -from the list of operators (pull-down menu), select chem6190
- -there is no password for this account
- -click **OK**

Sample slot selection

-if the **tray panel** is not present, click the "**O**" located above and to the left of the black spectral display window to display the tray panel

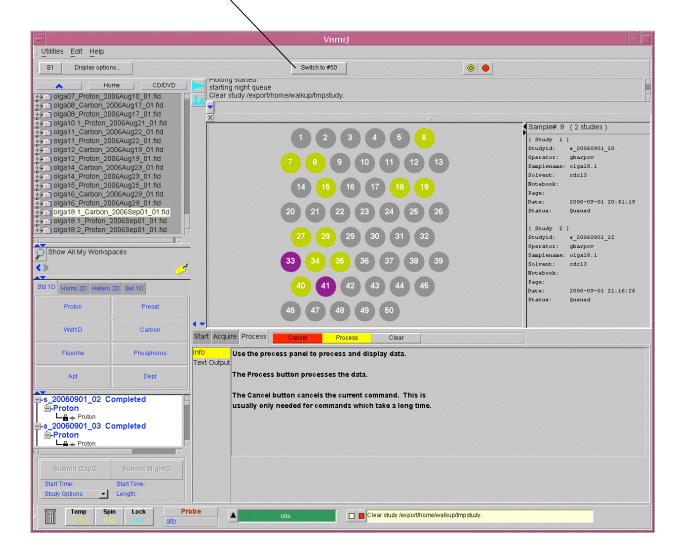


- -the tray panel represents the sample tray located on top of the white platform next to the magnet
- -the X in the upper left of the sample tray will toggle the display back to the black spectral display window
- -the numbered circles on the tray panel correspond to sample slots in the sample tray



- -sample slots that appear in color on the tray panel are NOT available for use
 - -also, sample **slot 50** /s **never** available for use (this slot is reserved for a standard sample)
- -sample slots that appear **gray** on the tray panel **may** be available for use, but this must be confirmed (see below)
- -you also can click on this arrow which will pop up a small window that tells you what sample currently is in the magnet

- -click the switch to #50 button at the top of the screen
 - -if nothing happens, this means that any of the gray slots can be used
 - -if there is a sample in the magnet already, the robot will take it out and replace it with sample 50. At this point, all **gray slots** are **now available** for use (*the robot is slow, so be patient*)
- -choose one of the available (gray) slots. You must confirm that this slot is empty by checking the sample tray



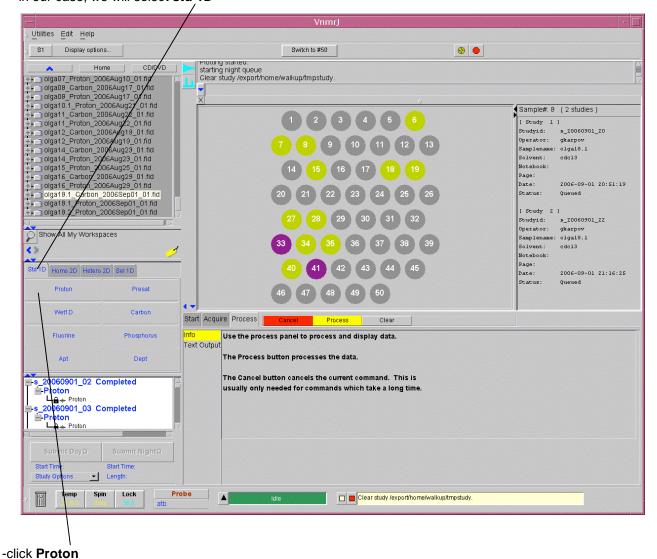
Sample placement

- -hold the sample tube near the middle of the tube and in a vertical position
- -push the tube through the hole on the large end of a sample spinner turbine
 - -NEVER push on the top of the tube
- -the bottom of the tube should be pushed slightly through the bottom (small) end of the spinner turbine
- -set the tube/spinner assembly in the black sample depth gauge in the sample tray
- -push the tube (by its sides, NOT from the top) down until it stops
 - -the tube is now properly positioned in the spinner turbine
- -now you can place your tube/spinner assembly in one of the open slots (on the sample tray, the **slot number** is to the **right** of the slot/hole)

Experiment selection

-you can select from among the experiment tabs the one corresponding to the experiment that you will be acquiring

-in our case, we will select std 1D

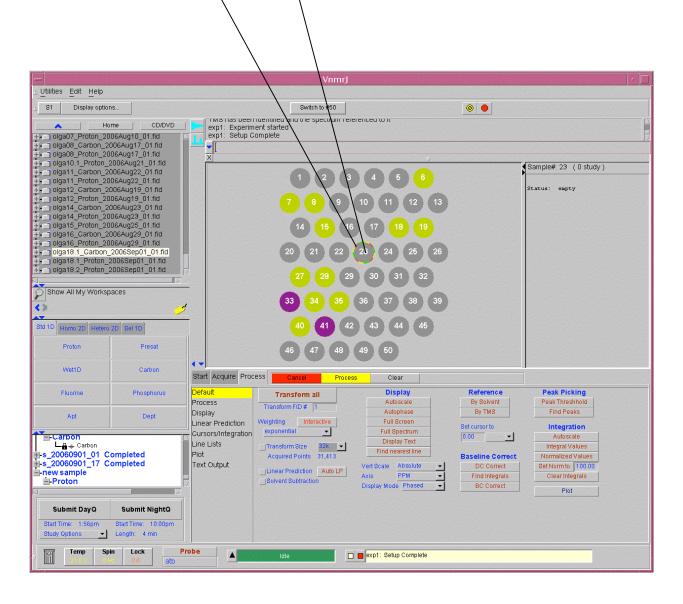


-this will load the appropriate parameters for collecting a simple 1D ¹H experiment

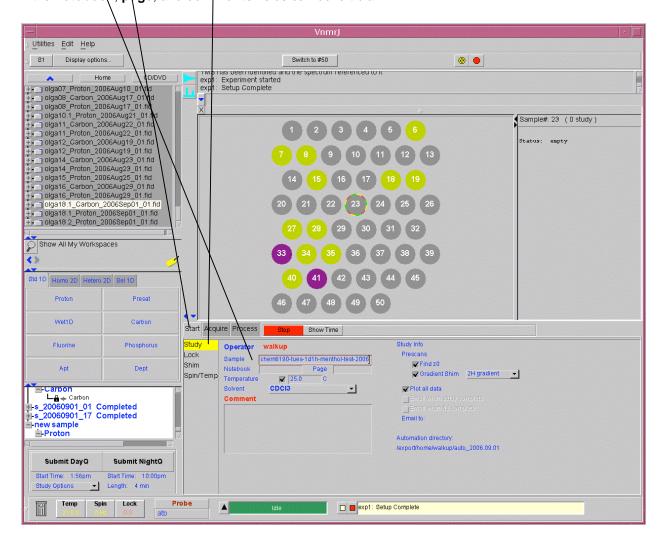
-in window below, **new sample** and **Proton** will appear (these will be colored yellow, indicating that the sample has not yet been submitted, and the experiment has not begun)

-on the **tray panel** click on the **appropriate** slot (an empty one that you have chosen)



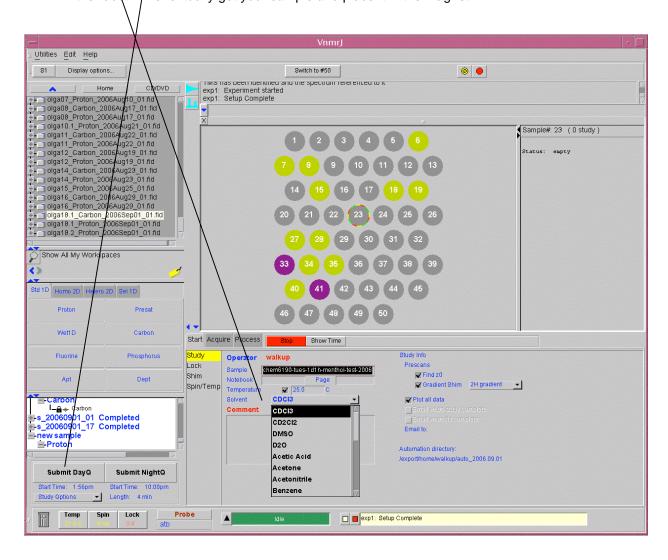


- -click the start tab, and the study option
- -in the Sample field, enter a descriptive sample name
 - -no capital letters or spaces, no special characters other than dash (-) and underscore (_)
 - -for\insta\nce, "chem61\frac{9}{0-1d1h-090306}"
- -the notebook, page, and comments fields can be left blank



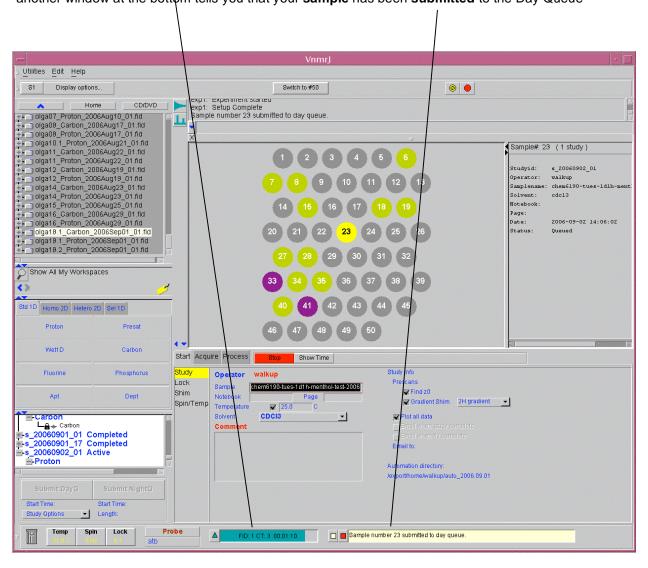
Setting parameters and submitting the sample

- -do not change the Temp (temperature setting), or the Spin setting, or the Lock setting
- -for **solvent**, select **CDCI3** (deuterated chloroform)
- -click Submit\to Day Queue
 - -wait..\.....the robot is slow
 - -the slot in the tray panel corresponding to your sample will turn yellow
 - -the robot will eventually get your sample and place it in the magnet



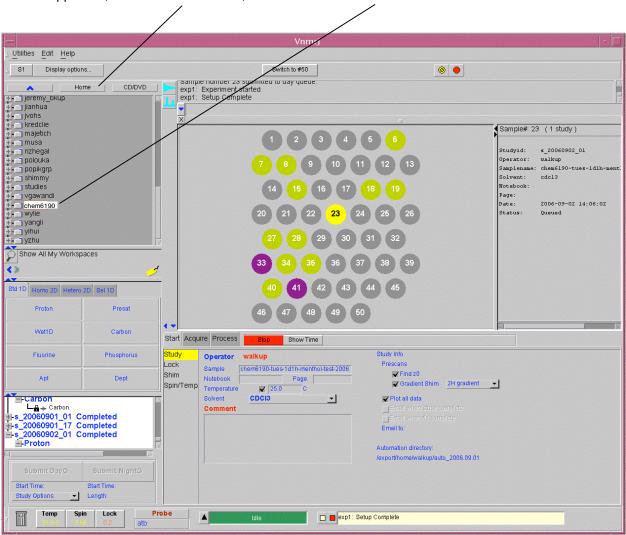
Locking and shimming

-in the automated mode, locking on the solvent and shimming (gradient shimming) are automated -the **window** at the **bottom of the screen** keeps you informed on the locking and shimming progress -another window at the bottom tells you that your **sample** has been **submitted** to the Day Queue



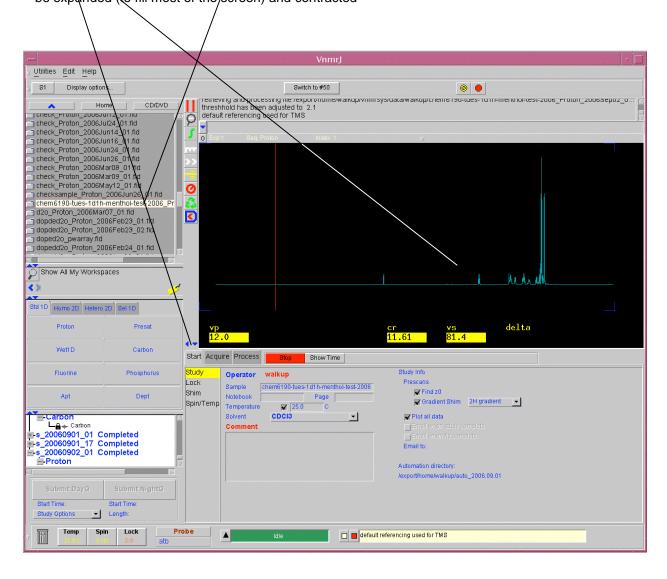
Retrieving the data and displaying the spectrum

-at the upper left, click the **Home** button, then double-click **chem6190**.......



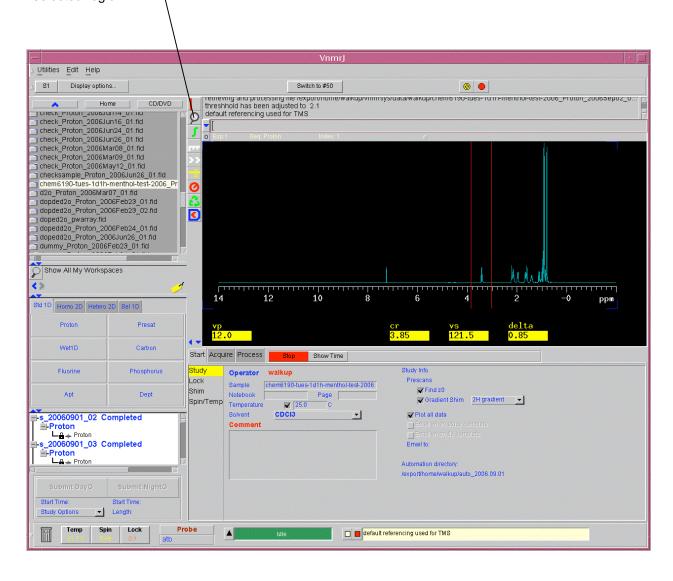
.....then double-click on the filename that you gave to your data

- -the **spectrum** should appear (in the window previously occupied by the tray panel)
- -the **two small blue arrow buttons** (•) at the lower left of the spectral window permit this window to be expanded (to fill most of the screen) and contracted



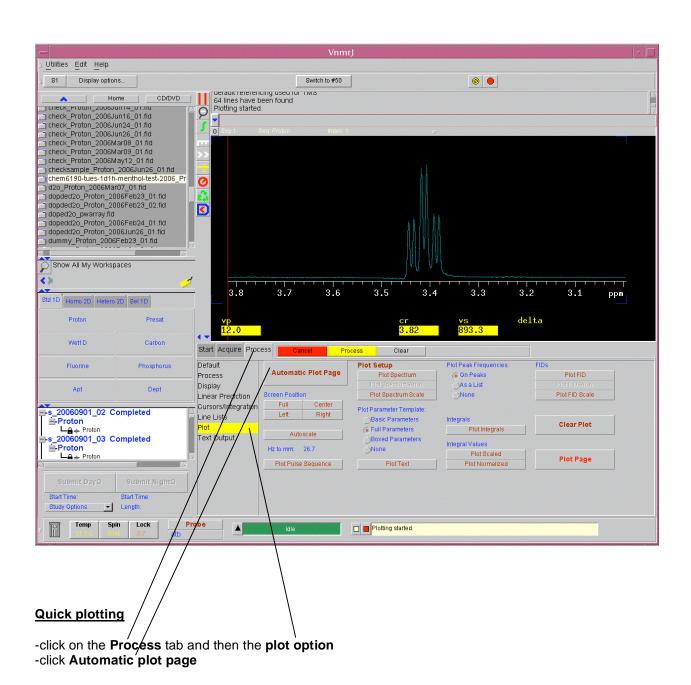
Zooming in and out

- -to zoom in on a particular spectral region or peak of interest, place the **arrow cursor** at the **left edge** of this region and click the **left mouse button** (or drag the red cursor to the left side of the region of interest with the left mouse button), then move the **arrow cursor** to the **right edge** and click the **right mouse button**
- -then click the **magnifying glass icon** (on the **vertical menu**) to toggle between the full display and the selected region \



Adjusting scale

-to adjust the scale/intensity, place the arrow cursor on a peak, hold down the middle mouse button, and move the mouse straight forward or backward (*NOT* sideways)



Chemical shift referencing

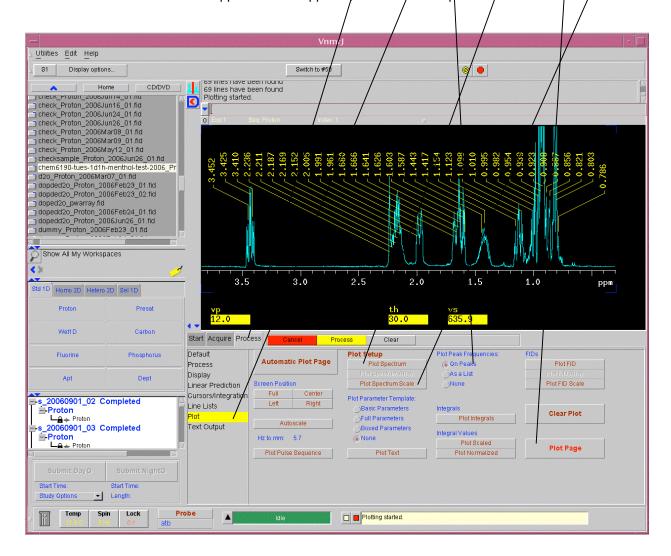
- -the ¹H chemical shifts should be referenced relative to TMS at 0.0 ppm
 - -when CDCl₃ is used as the solvent, there is a small amount of residual CHCl₃ in the sample, and the ¹H signal of CHCl₃ resonates at exactly 7.27 ppm relative to TMS
 - -thus, we will reference the spectrum relative to this CHCl₃ peak (we will force the computer to recognize the chemical shift of this peak as 7.27 ppm).
- -first, zoom in on the small signal at approximately 7.2 ppm
- -click the Process tab, then select default
- -set a red cursor (with the left mouse button) on the **center of the CHCI₃ signal** and click then **find** nearest line (under the **Display** options)
- -in the Set cursor to field, select ppm from the drop down menu, enter 7.27 in the box and hit veturn Utilities Edit Help S1 Display options. Switch to #50 **®** neignt= 74:55**6**5 mm = rrequency= 7:22007 ppm linewidth = 1:25614 Hz = digital resolution = 0:391275 Hz Reference frequency 7:27004 ppm, (2909.1 Hz) check_Proton_2006Jun16_01.fid check_Proton_2006Jun24_01.fid check_Proton_2006Jun26_01.fid check_Proton_2006Mar08_01.fid ____check_Proton_2006Mar09_01.fid ncheck Proton 2006May12 01.fid checksample_Proton_2006Jun26_01.fid chem6190-tues-1d1h-menthol-test-2006 F] d2o_Proton_2006Mar07_01.fid] dopded2o_Proton_2006Feb23_01.fid] dopded2o_Proton_2006Feb23_02.fid doped2o_pwarray.fid dopedd2o_Proton_2006Feb24_01.fid dopedd2o_Proton_2006Jun26_01.fid dummy_Proton_2006Feb23_01.fid Show All My Workspaces <≥ 7.270 7.265 7.255 7.285 7.280 7.275 7.260 ppm Std 1D Homo 2D Hetero 2D Sel 1D Proton delta Carbon Clear Peak Picking Fluorine Phosphorus Default Display eference Process Peak Threshhold Transform FID # 1 Autophase Display Apt Weighting Interact Linear Prediction Integration Cursors/Integration none Full Spectrum ppm 🔻 20060901_02 Completed Line Lists Display Text __Transform Size Integral Values ■ Proton Find nearest line Plot Acquired Points 12 791 **Baseline Correct** Normalized Values Text Output Set Norm to 100.00 Vert Scale DC Correct __Linear Prediction Auto LP Proton La Proton Find Integrals Clear Integrals |Solvent Subtraction Plot Start Time Start Time Study Options Length Lock Probe \blacktriangle Reference frequency 7.27004 ppm, (2909.1 Hz)

Peak picking

- -peak picking means determining what the chemical shifts are for the peaks in the spectrum
- -first, zoom in on the peaks in your spectrum (the region from about 0 to about 4 ppm)
- -first we need to set the **intensity threshold** (chemical shifts will be returned for all of the peaks with intensities above this threshold)
 - -from the **vertical menu** at the left of the spectral display window, select the yellow **threshold icon** and move the **yellow line** up/down on the spectrum to the desired height
- -click the **Process tab**, then the **Default** option, then click the **Find Peaks** button (under **Peak Picking**)
 -the peak positions will be displayed on the screen
 - -you can remove the displayed yellow lines and peak positions by selecting **Process/Display** and then clicking on **Full** under **Screen Position**

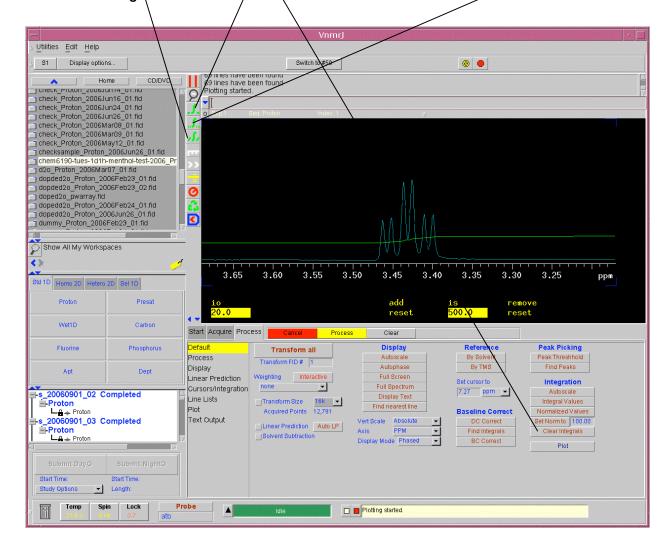


-in order to print the peak positions (hardcopy), under Process/Plot, click Plot Spectrum, then Plot Spectrum Scale, then under Plot Peak Frequencies, select On/Peaks. Then click Plot Page /-if you want to return to a display without the peak positions, select Process/Display, and then click Full under Screen Position. If the ppm scale disappears/click the/PPM option under Axis

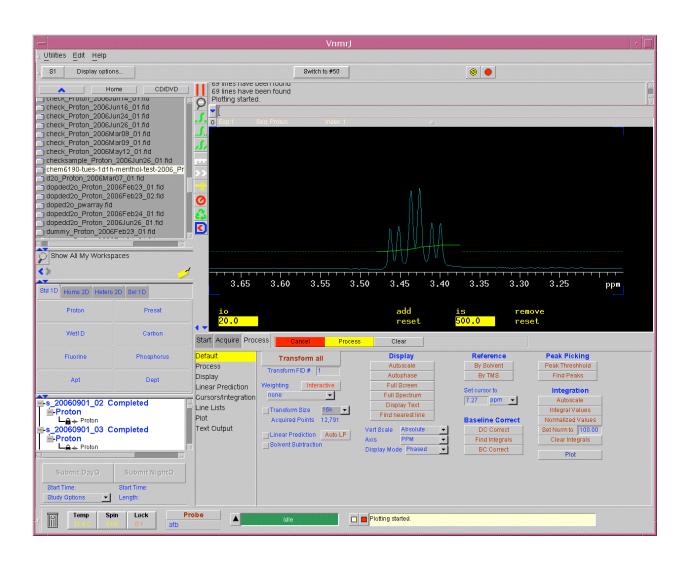


Integration

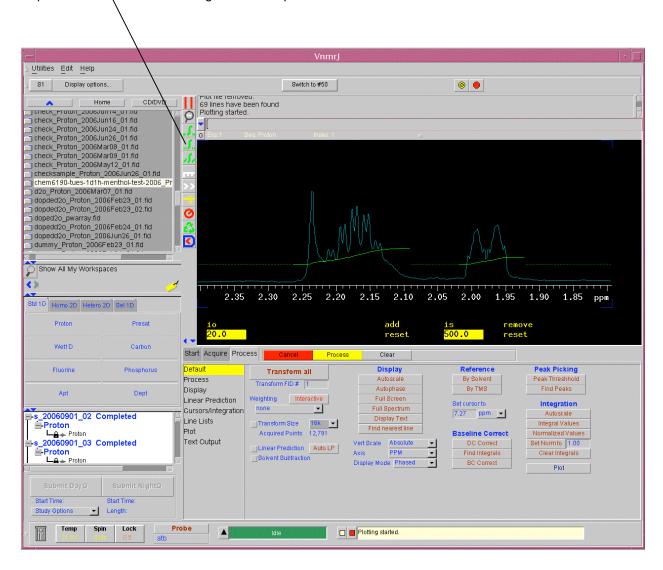
- -first, expand the display around the multiplet signal centered at about 3.43 ppm
- -on the vertical menu at the left is the integral, icon/button
- -as you click on this button you toggle through the **3 modes**: first **is part integral**, second is full integral, and third is **no integral**
- -select the part integral mode
 - -when this is selected, two more integral buttons will appear below the first: the **integral resets** and the **integral LvI/Tit**
- -under Process/default, click clear integrals (under Integration options)
- -then click the integral resets button



- -then, place the cursor just to the **left of the peak** (at the baseline) and **click** (left mouse button) Then place the cursor just to the **right of the peak** and **click** (again, left mouse button).
 - -you'll see that the green integral display changes to dashes except on the peak that you are integrating (which remains a solid green color)

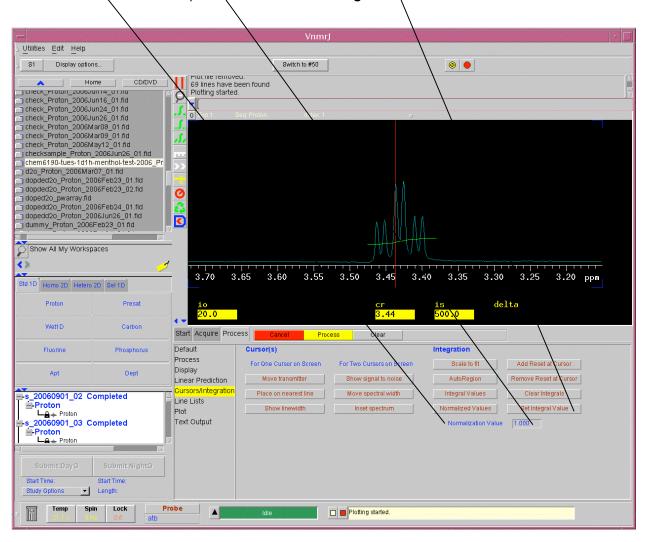


- -expand around the next two signals in the spectrum (the two at approximately 2.2 and 2.0 ppm)
- -click the integral resets button
- -click on the right and left sides of each of these signals, as you have done for the previous signal
- -repeat this for the rest of the signals in the spectrum



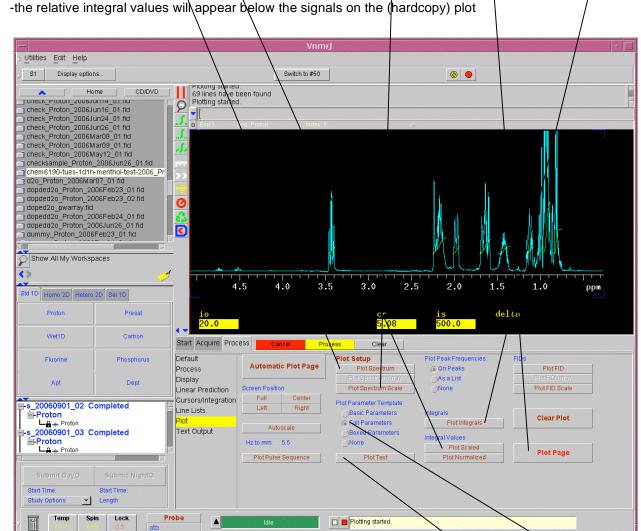
Integral calibration

- -the **integrals** of signals are **proportional to the number of protons** giving rise to the individual signals
- -it is convenient to **normalize the integration values** to simplify the analysis (enabling the analysis to be done by inspection)
- -select a well-resolved signal, and, if possible, one that you suspect results from a single ¹H nucleus
- -zoom on this peak, put the cursor on the peak, select Process then Cursors/Integration, and in the Normalization value field put 1.0. Then click Set Integal Value.



Advanced Plotting

- -expand around the peaks in your spectrum (0 4 ppm)
- -under Process/Plot, click Plot Spectrum and Plot Spectrum Scale (both under Plot Setup), Plot Integrals (under Integrals), Plot Scaled (under Integral Values), and then Plot Page



-try the above set of commands again, but in this case **include** clicking on **Plot Text** with the **Full Parameters** button selected

-this will **print the acquisition parameters** on the hardcopy plot. You will note that these parameters are printed on the left side of the spectrum, so it is often a good idea to leave some blank space to the left of the peaks in your spectrum before you plot the parameters (i.e., in our case, zoom in on the region from about 5 ppm to 0 ppm, as above, in which case there is "blank space" between the left edge of the spectrum and the peak at 3.5 ppm)

Logging out

- -it is best to remove your sample. Simply click the Switch to #50 button
- -in the extreme upper left of the screen click the **Utilities** button, and select **Change Operator** from the menu

Goals for Lab #1:

Get a good hardcopy plot for the spectrum of our sample/molecule (menthol in CDCl₃)

Reference the spectrum properly

Get the frequencies for as many signals as possible

Plot expansions of all signals with peak frequencies

- -the frequencies/chemical shifts are related to shielding of the nuclei, and are thus important for defining the identity of the nucleus giving rise to the signal
- -these frequencies will be necessary for calculating coupling constants
- -the coupling constants will assist in defining connectivity in the molecule

Get good integrals for each signal

- -normalize the integrals
- -the integrals will tell you how many ¹H nuclei give rise to the signal, therefore if the group is –CH, -CH₂, -CH₃, etc.

Plot the full spectrum with acquisition parameters

Exercises and Questions for Lab #1:

- 1. -Define the relationship (in general) between the spectral width, the number of points digitized/collected, the digital resolution and the acquisition time.
 - -Using the acquisition parameters for the spectrum that we collected, calculate the digital resolution in two ways. Do they agree?
 - -Describe two ways that you could improve the digital resolution (in general).
- -Based solely on your integration analysis, make an estimate of the number of hydrogens in your compound
 - -Does this estimate correspond to the expected number?
 - -Elaborate on this and attempt to account for any discrepancies.
- **3.** -On a plot of the full spectrum of your molecule, label the signals a, b, c, etc. from left to right (from low to high field).
 - -Determine the chemical shift for each signal.
 - -Construct a table such as the one shown below and begin to fill out the information (i.e. chemical shifts)
- -Look carefully each of the signals in the spectrum, and try to describe their multiplet structures (i.e. singlet?, doublet?, triplet?, doublets?, quartet?, triplet of doublets?)
 - -Do these descriptions of the multiplet structures make sense based on the known structure of menthol?
 - -Add the multiplicity information to your table, along with coupling constants.
- 5. -For the signals at ~3.4, ~2.2, ~1.95, and ~1.1 ppm, and for the very intense signals that are observed between ~1.0 and ~0.7 ppm, determine the coupling constants for the various couplings observed in each signal.
 - -What information concerning connectivity can you make based on your observed coupling constants?
- -Based on the information above, and based on knowledge of shielding/inductive effects, see if you can assign some of the signals in the spectrum to the various hydrogens in menthol -Use the numbering on the menthol molecule shown below.
 - -Provide as complete a justification as you can for these assignments.
- 7. -Discuss the differences (chemical shifts, couplings, resolution, etc.) that you might observe between menthol spectra acquired using 400 MHz and 600 MHz spectrometers.
- **8.** -Sketch the 1D ¹H spectrum of propionic acid (CH₃CH₂COOH) in D₂O with the correct multiplet structure and the correct relative intensities.

SAMPLE:

Our sample is $\sim 0.5 M$ menthol in CDCl₃

HO
$$\begin{array}{c}
2 & 1 \\
3 & \\
10 & 5 \\
7 & 6
\end{array}$$
menthol

SIGNAL (LABEL)	Chemical Shift (ppm)	Integral (normalized values)	Multiplicity	Coupling constants (Hz)	Connectivity / Assignments / Comments
а	3.43				
b	2.2				
С					
Etc.					