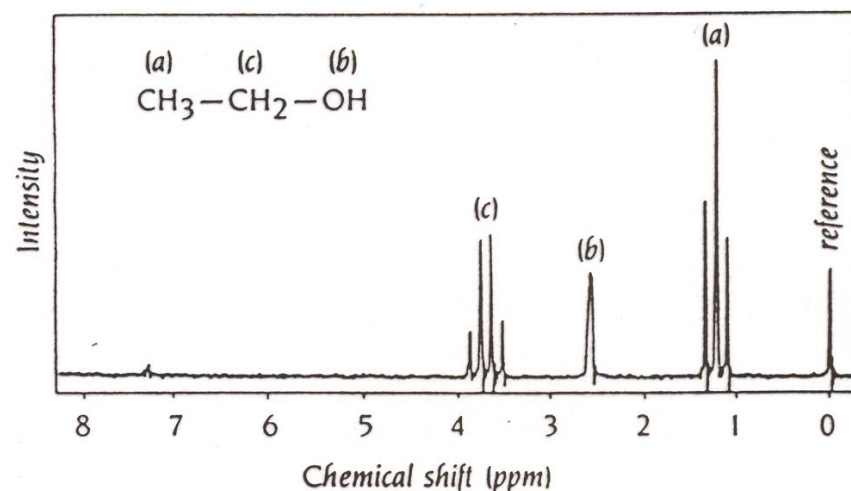


NMR Spectroscopy: Intro

NMR: Sophomore Organic Chemistry

- Sophomore organic NMR:

- 1D, ^1H spectra
- small organic molecules
- one signal for each type of ^1H
- signals split into doublets, triplet, quartets, etc., due to presence of ^1H nuclei on neighboring carbon atoms (“coupling”)
- frequencies (“chemical shifts”) of signals determined by electron density



- This course:

- what gives rise to the signals; microscopic and macroscopic
- what determines signal frequencies, intensities and multiplicities (chemical shifts/couplings/relaxation)
- how are spectra acquired and processed (pulsed Fourier transform methods)
- how can we manipulate nuclear magnetism to get other types of information (complex pulse sequences)
- 2D methods / chemical shift correlation (COSY, TOCSY)
- the nuclear Overhauser effect (NOE, NOESY)
- etc.

NMR Active Nuclei

- Most elements are comprised of one or more isotopes that are NMR active

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
<u>H</u>																<u>H</u>	<u>He</u>
<u>Li</u>	<u>Be</u>											<u>B</u>	<u>C</u>	<u>N</u>	<u>O</u>	<u>F</u>	<u>Ne</u>
<u>Na</u>	<u>Mg</u>											<u>Al</u>	<u>Si</u>	<u>P</u>	<u>S</u>	<u>Cl</u>	Ar
<u>K</u>	<u>Ca</u>	<u>Sc</u>	<u>Ti</u>	<u>V</u>	<u>Cr</u>	<u>Mn</u>	<u>Fe</u>	<u>Co</u>	<u>Ni</u>	<u>Cu</u>	<u>Zn</u>	<u>Ga</u>	<u>Ge</u>	<u>As</u>	<u>Se</u>	<u>Br</u>	<u>Kr</u>
<u>Rb</u>	<u>Sr</u>	<u>Y</u>	<u>Zr</u>	<u>Nb</u>	<u>Mo</u>	<u>Tc</u>	<u>Ru</u>	<u>Rh</u>	<u>Pd</u>	<u>Ag</u>	<u>Cd</u>	<u>In</u>	<u>Sn</u>	<u>Sb</u>	<u>Te</u>	<u>I</u>	<u>Xe</u>
<u>Cs</u>	<u>Ba</u>	*	<u>Hf</u>	<u>Ta</u>	<u>W</u>	<u>Re</u>	<u>Os</u>	<u>Ir</u>	<u>Pt</u>	<u>Au</u>	<u>Hg</u>	<u>Tl</u>	<u>Pb</u>	<u>Bi</u>	Po	At	Rn
Fr	Ra	**	Rf	Ha	Sg	Ns	Hs	Mt									
* <u>La</u> Ce <u>Pr</u> <u>Nd</u> Pm <u>Sm</u> <u>Eu</u> <u>Gd</u> <u>Tb</u> <u>Dy</u> <u>Ho</u> <u>Er</u> <u>Tm</u> <u>Yb</u> <u>Lu</u> ** Ac Th Pa <u>U</u> Np Pu Am Cm Bk Cf Es Fm Md No Lr																	

<http://bouman.chem.georgetown.edu/NMRpt/NMRpertbl.html>

History*

1940s	-First observation of nuclear magnetic resonance in solids and liquids (1945)
1950s	-Development of chemical shifts and spin-spin coupling constants as structural tools
1960s	-Superconducting magnets -Pulse Fourier transform approach developed -Nuclear Overhauser effect used for structure determination
1970s	-Computer control
1980s	-Development of multipulse and two dimensional techniques -Automation
1990s	-Pulsed field gradients -Coupled (“hyphenated”) methods (i.e. LC-NMR)
2000-present	-High-sensitivity cryogenic probes -Very high fields (> 900 MHz)

*From Claridge, “High-Resolution NMR Techniques in Organic Chemistry”



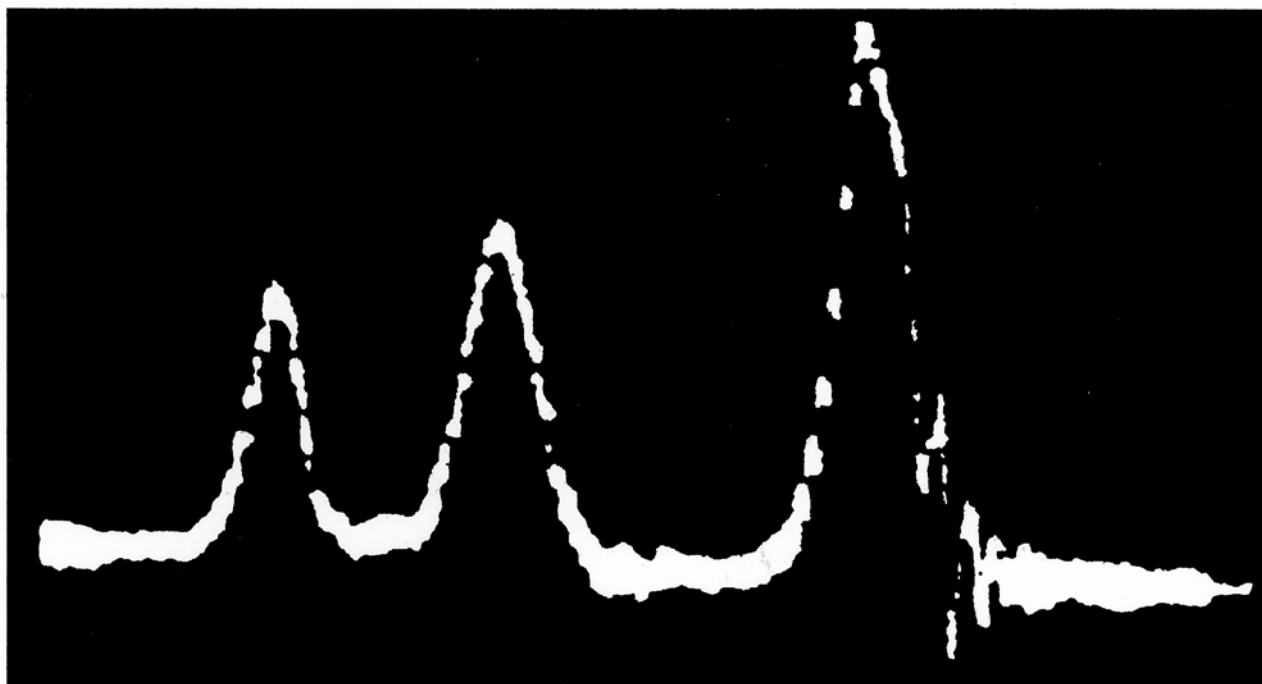
Recognition

- 1944 – Isidor Isaac Rabi - Nobel Prize in Physics
 - "for his resonance method for recording the magnetic properties of atomic nuclei"
- 1952 – Felix Bloch and Edward Mills Purcell – Nobel Prize in Physics
 - "for their development of new methods for nuclear magnetic precision measurements and discoveries in connection therewith"
- 1991 – Richard Ernst – Nobel Prize in Chemistry
 - "for his contributions to the development of the methodology of high resolution nuclear magnetic resonance (NMR) spectroscopy"
- 2002 – Kurt Wuthrich – Nobel Prize in Chemistry
 - "for his development of nuclear magnetic resonance spectroscopy for determining the three-dimensional structure of biological macromolecules in solution"
- 2003 – Paul Lauterbur and Sir Peter Mansfield – Nobel Prize in Physiology and Medicine
 - "for their discoveries concerning magnetic resonance imaging"

History

First published 'high-resolution' ^1H NMR spectrum (1951)*

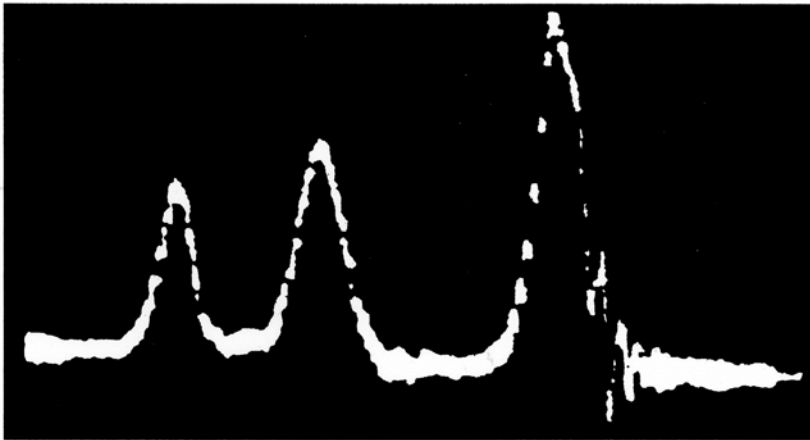
-spectrum of ethanol demonstrated that NMR spectra were a source of structural information (now, chemists got interested)



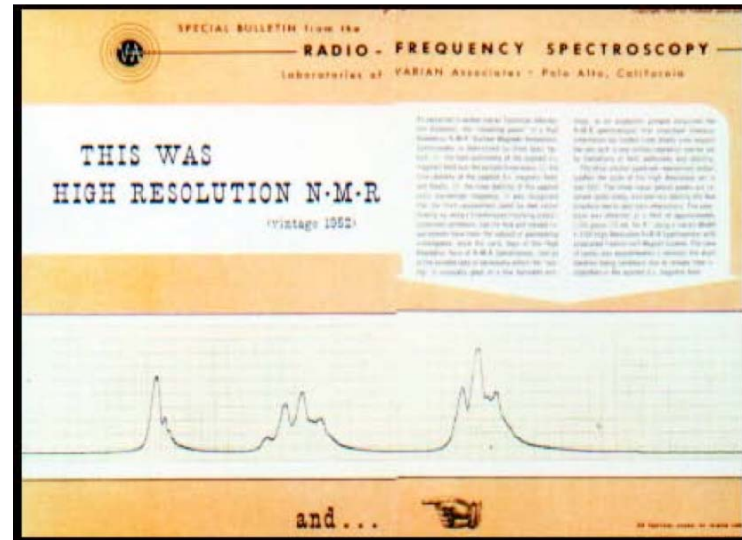
*Arnold, J. T., Dharmatti, S. S., and Packard, M. E., (1951) **Chemical Effects on Nuclear Induction Signals from Organic Compounds.** *J. Chem. Phys.* **19**, 507.

History

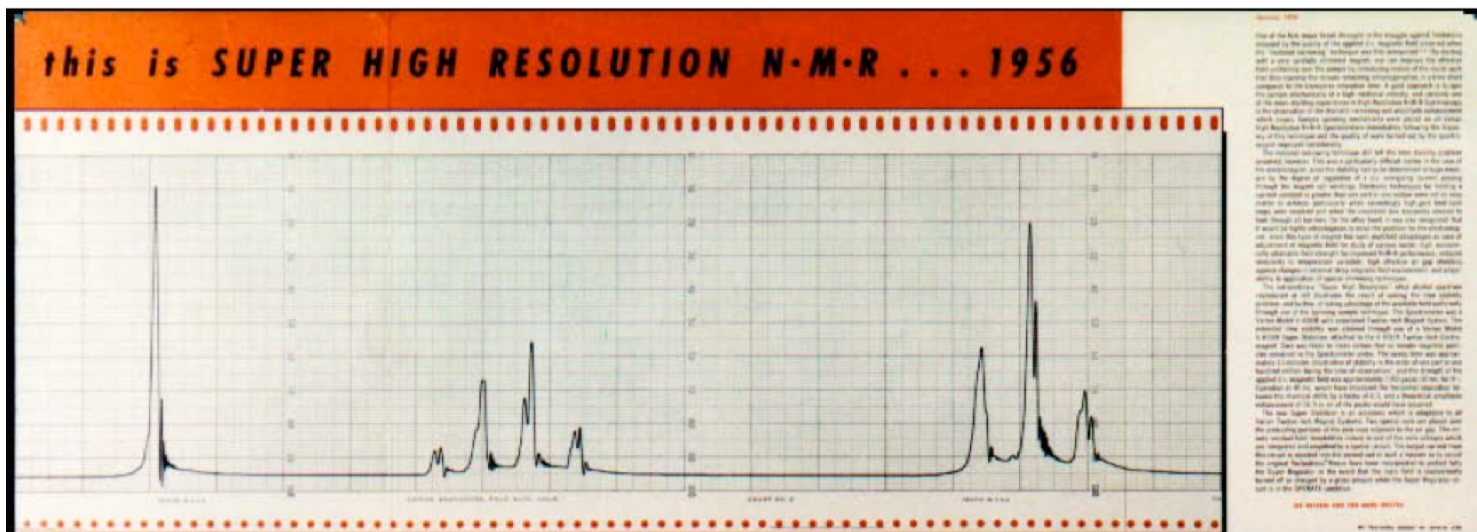
Ethanol - 1951



Ethanol - 1952

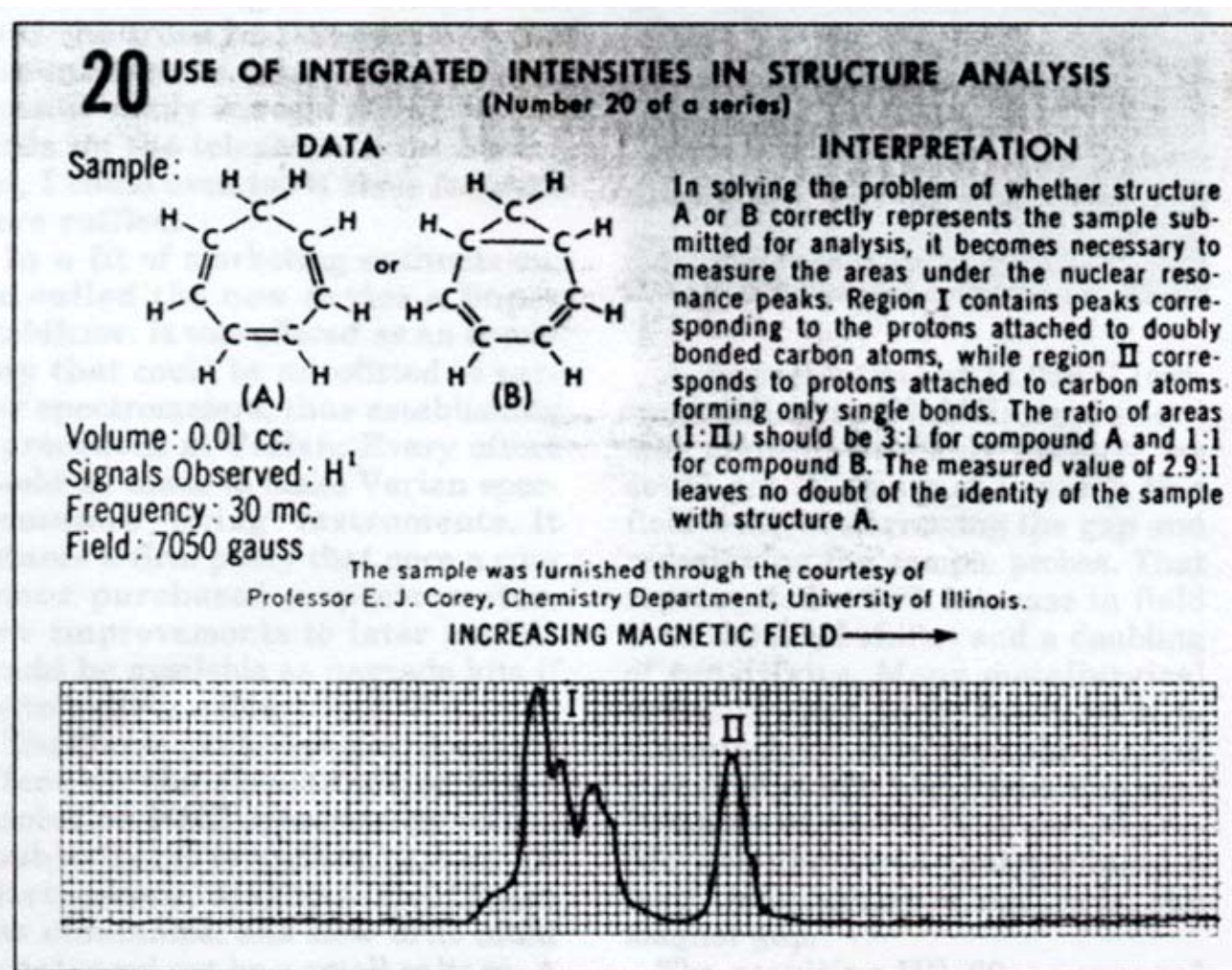


Ethanol - 1956



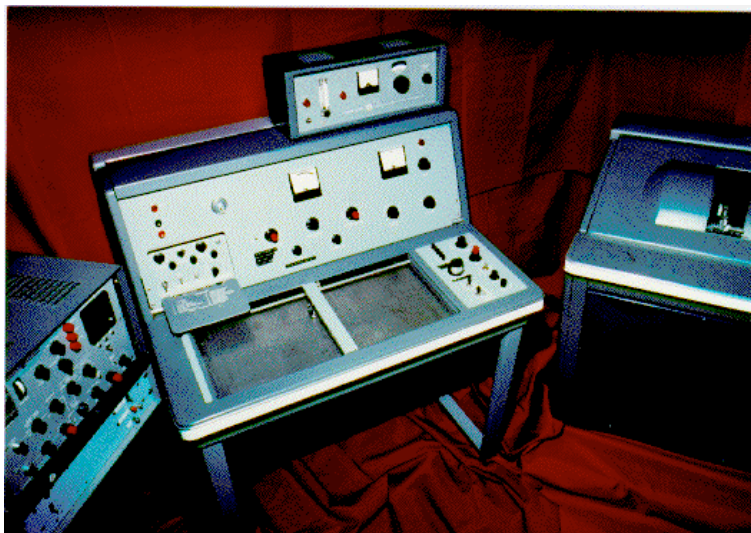
History

The first chemical problem solved by NMR
-signal integration discriminated between
possible synthesis products



1961: First Widely (Wildly) Successful NMR Instrument

-Varian A-60 (60 MHz) NMR spectrometer

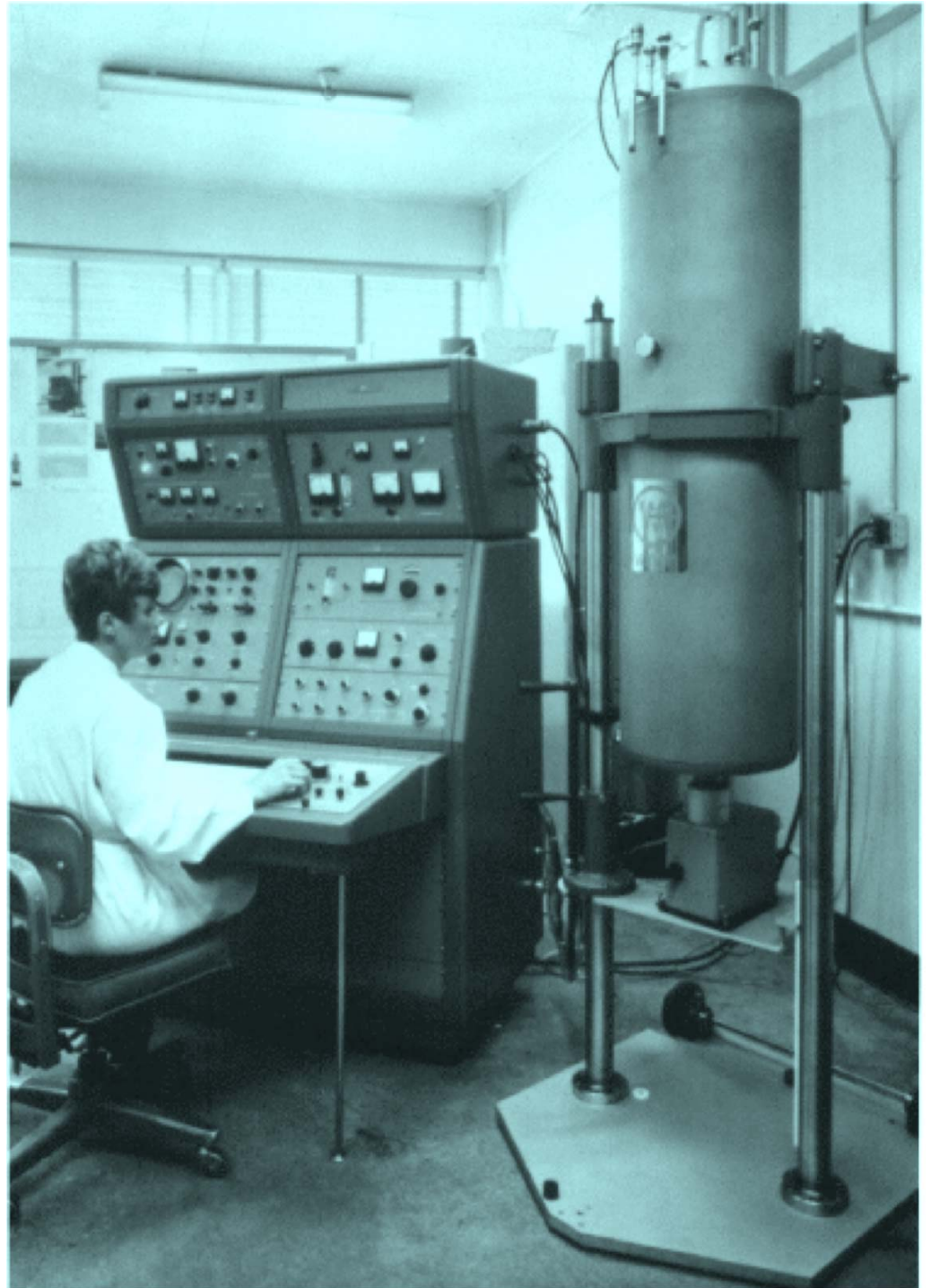


"I put a sample in the instrument, adjusted the resolution, and ran a spectrum on the precalibrated chart. It was perfect. But could the A-60 reproduce a spectrum with the fingerprint quality of the IR instrument? I moved the pen back to the start and restarted the scan. I was momentarily distracted, and when I looked back I saw only one line on the chart. "Why didn't the second scan run properly?" I asked. The answer came back, "It did!" Amazed and almost incredulous, I returned the pen three more times. It laid down five identical spectra with a single trace showing on the paper! **At that instant, I knew the field of organic chemistry would never be the same again.**" (from Shoolery, J.N. "NMR spectroscopy in the beginning." *Anal. Chem.* **1993**, 65(17), 731A-741A)

Superconducting Magnets

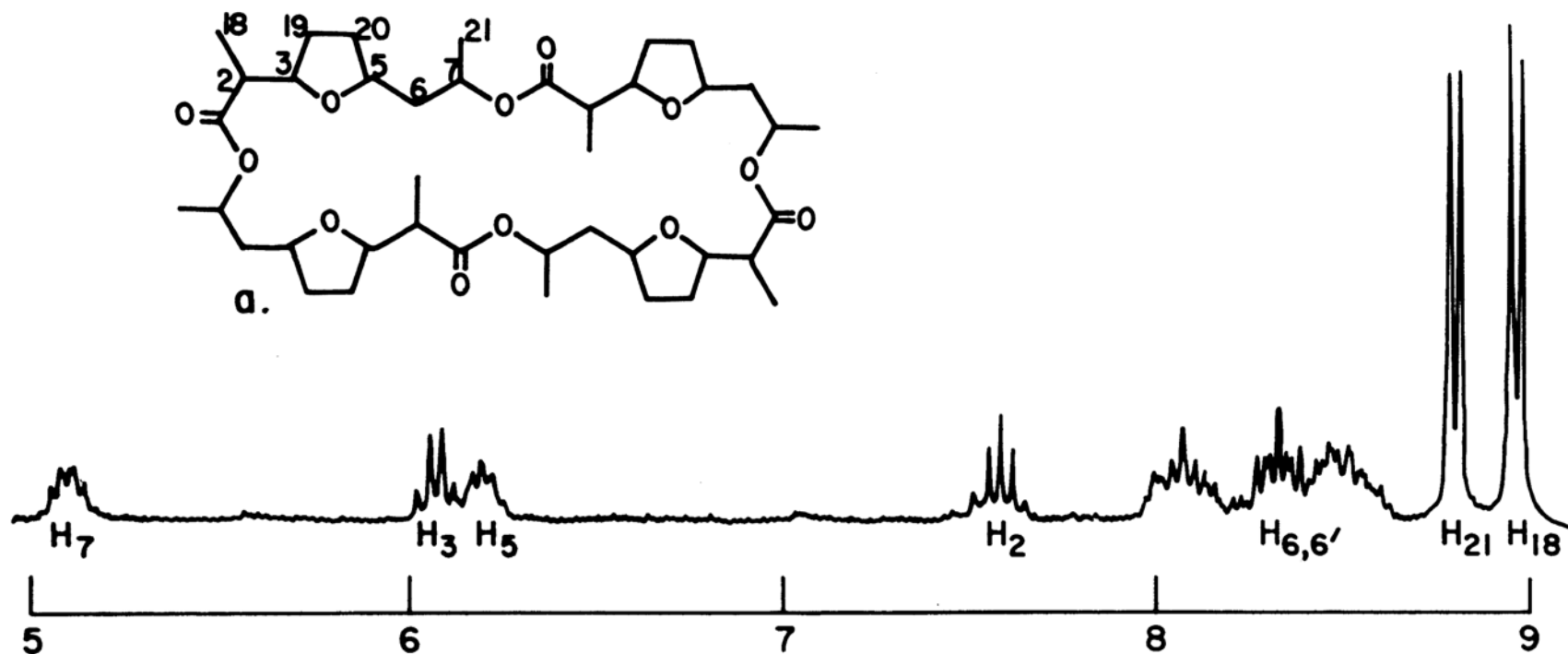
-the first high resolution superconducting NMR magnet (1964): the Varian HR-200, the HR-220 (shown here) and later the HR-300 high resolution NMR spectrometers.

-niobium-titanium alloy wire wrapped around a core and bathed in liquid helium (4 K)



High Field NMR

-high field (200 MHz), but still 1D CW NMR



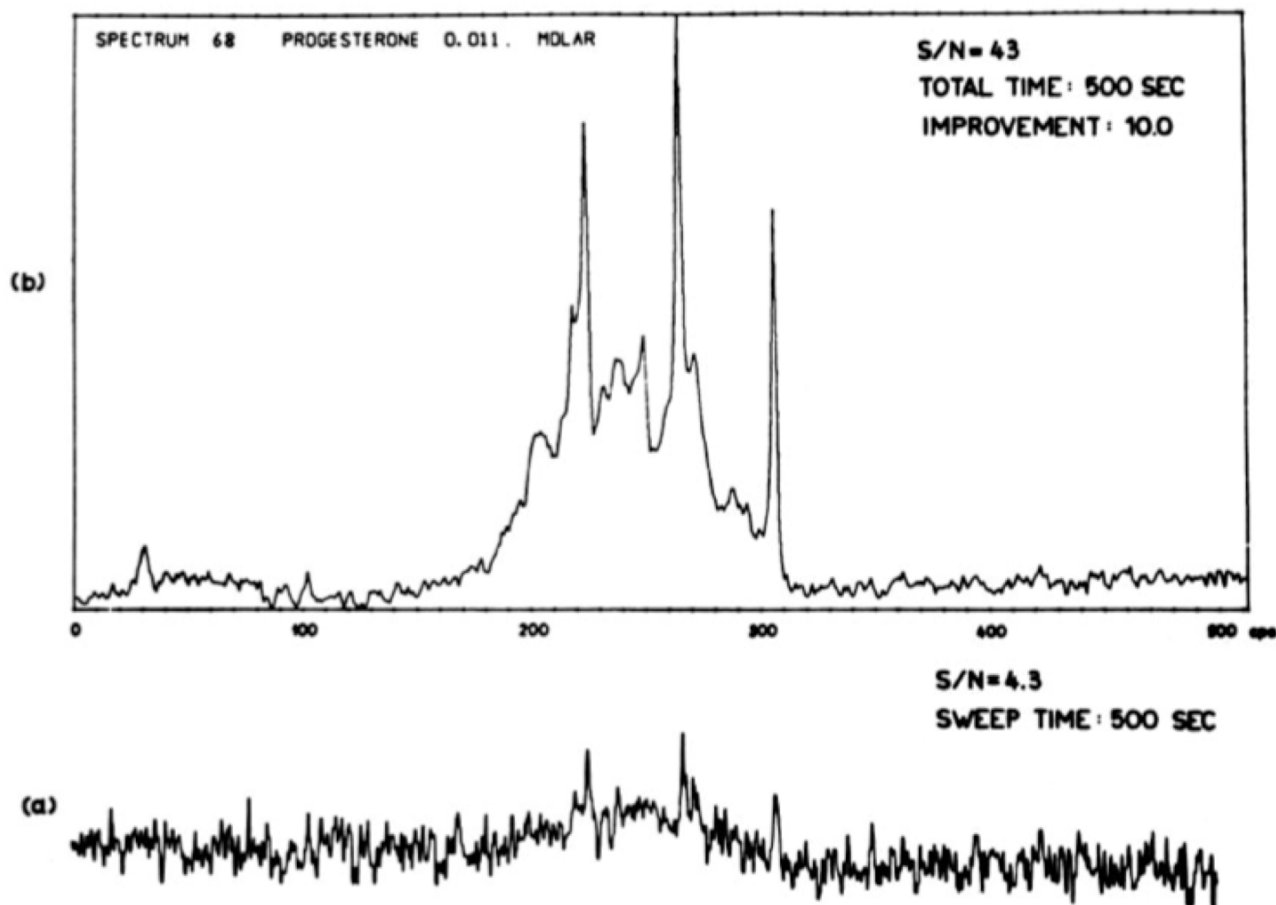
1966: Primitive Computer Control And Pulsed Fourier Transform NMR

-Richard Ernst in 1965/1966 at the helm



Pulsed Fourier Transform NMR

Richard Ernst & Wes Anderson, Rev. Sci. Instr. 37, 93 (1966)



Fourier transform (top) and conventional spectra of 0.011 M progesterone showing sensitivity enhancement by a factor ten

Modern Superconducting Magnets

