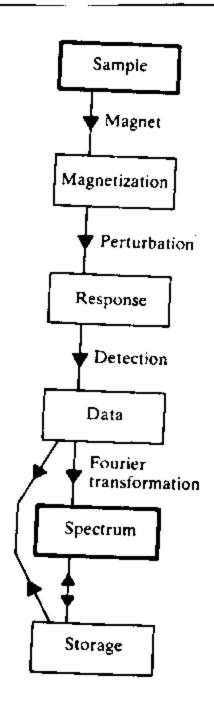
CHEM / BCMB 4190/6190/8189

Introductory NMR

Lectures 1 and 2



Nuclear Magnetism:

NMR is a manifestation of nuclear spin angular momentum (P).

$$P = h - \sqrt{I(I+1)}$$

I = angular momentum quantum numberh = Planck's constant/ 2_

Possible Values of I:

- 1. I=0, mass number (A) and atomic number (Z) are even.
- I= half-integral value, A is odd.
- 3. I= integral value, A is even and Z is odd.

For NMR:

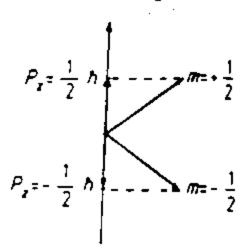
- I=0 are NMR-inactive
- I > 1/2, nuclei possess electric quadrupole moment due to non-spherical nuclear charge distribution. The lifetime of the magnetic states for quadrupoles in solution are much shorter than for I=1/2. This results in line broadening and they can be more difficult to study.
- I= 1/2 include ¹H, ¹³C, ¹⁵N and ³¹P.

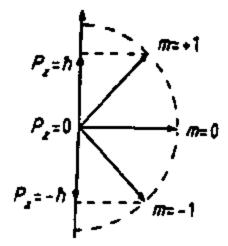
Due to restrictions of quantum mechanics only one of the three Cartesian coordinates can be specified.

$$P_{r} = hm$$

m = (-I, -I + 1, ..., I-1, I) - magnetic quantum number.

 I_z has 2I+1 values of m. This behavior is called directional quantization.





Nuclei with non-zero spin angular momentum also possess nuclear magnetic moment:

$$\mu = \gamma P$$

- $-\gamma$ in part determines the receptivity of a nucleus in NMR spectroscopy. It is a constant for a given nuclide and it is referred to as the gyromagnetic ratio.
- μ and **P** are usually in the same direction but one important exception is "N.

Therefore:

$$\mu_z = m\gamma A$$

More Quantitatively:

$$\mu = g \, \underline{ehI}$$

$$4\pi M c$$

M = mass

e = charge uniformly spread over the surface <math>c = speed of light.

** No simple model can predict or explain the actual magnetic moments of nuclei**

In absence of an external magnetic field: 2I + 1 states of m are equivalent

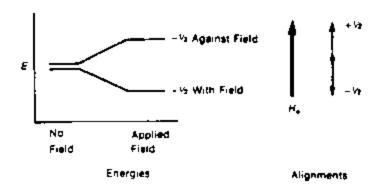
In presence of an external magnetic field, the spin states of a nucleus have energies given by:

$$E = \mu_z B_a = -I_z \gamma + B_o$$

B₀ = static magnetic field strength

The projection of angular momentum of the nuclei onto the z-axis of the laboratory frame results in 2I + 1 equally spaced energy levels, which are referred to as the Zeeman levels or Zeeman States.

For $I_z = 1/2$ we have m = +1/2 and m = -1/2 in which μ_z is either parallel or anti-parallel to the field direction.



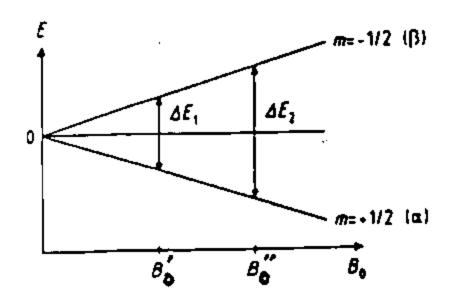
In quantum mechanics:

$$m = +1/2 = spin function \alpha$$

$$m = -1/2 = spin function \beta$$

Since we are restricted to single quantum transitions:

$$\Delta m = \pm 1$$



$$\Delta E = \gamma + B_0$$

What Effects ΔE and why is this important?

- gyromagnetic ratio
- strength of static magnetic field

At equilibrium the energy states are not equally populated and the relative populations is given by the Boltzmann distribution.

$$\frac{N_m}{N} = \exp\left(\frac{-E_m}{k_{\rm E}T}\right) / \sum_{m=-1}^{I} \exp\left(\frac{-E_m}{k_{\rm E}T}\right)$$

In our case, $E_m = -m + \gamma B_0$

$$= \exp\left(\frac{m\hbar\gamma B_0}{k_B T}\right) / \sum_{m=-1}^{l} \exp\left(\frac{m\hbar\gamma B_0}{k_B T}\right)$$

Since $m + \gamma B_O \ll kbT$ the exponential can be expanded to first order using Taylor series.

$$\approx \left(1 + \frac{m\hbar\gamma B_0}{k_B T}\right) / \sum_{m=-I}^{I} \left(1 + \frac{m\hbar\gamma B_0}{k_B T}\right)$$

$$\approx \left(1 + \frac{m\hbar\gamma B_0}{k_B T}\right) / (2I + 1),$$

- $-\exp(-E/k T) = 1 E/k T$
- denominator sums to = 2I + 1

The book gives an example for I=1/2

k: Boltzmann constant (1.3805 x 10⁻¹⁶ erg K ⁻¹)

T: temperature (K)

h: Plank constant (1.0546 x 10⁻²⁷erg S)

B₀: magnetic field strength

(Tesla, T; $100 \text{ MHz} \le 2.35 \text{ T}$)

γ: gyromagnetic ratio (10⁷ Hz T -1)

e.g.

At T = 300K and $B_0 = 5.875$ T (250 MHz),

¹**H**:

$$\frac{N_{\beta}}{N_{\alpha}} = 1 - \frac{26.7519 \times 10^{7} \times 1.0546 \times 10^{-27} \times 5.875}{1.3805 \times 10^{-16} \times 300}$$

$$N_{\beta} \approx (1 - 0.00004) N_{\alpha} = 0.99996 \bar{N}_{\alpha}$$

For $B_0 = 18.8 \text{ T} (800 \text{ MHz})$,

$$N_{\beta} \approx 0.99987 N_{\alpha}$$

In classical representation the nuclear dipoles precess about the magnetic field direction with Larmor Frequency:

$$\omega_{\text{\tiny 0}} = \gamma \, B_{\text{\tiny 0}} \quad (\text{Radians/sec})$$

$$\upsilon = \omega_{\text{\tiny 0}} / \, 2\pi = \gamma \, B_{\text{\tiny 0}} / \, 2\pi \quad (\text{Hertz or cycles/sec})$$

The z-component is given by the magnetic moment:

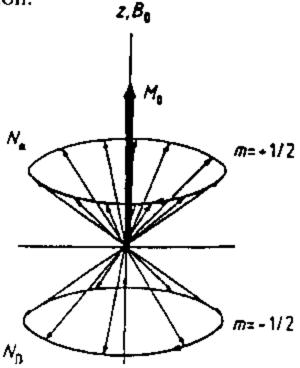
$$\mu_z = \gamma P_z = m\gamma h \qquad z, B_0$$

$$m = +1/2 \quad (\alpha)$$

$$m = -1/2 \quad (\beta)$$

When m = +/- 1/2; The nuclear dipoles precess around a double cone with a half-angle of the cone is 54.7° .

Due to the differences in populations of the energy states there is macroscopic magnetization along the magnetic field direction.



For a macroscopic sample,

The x and y components are random and sum to zero so:

$$M_0 \!\!= \gamma \!\!\! / \!\!\! h \sum_{m=-1}^I m N_m$$

but we know that:

$$N_m = N \exp(mh\gamma B_0/k_bT)/\sum \exp(mh\gamma B_0/k_bT)$$

Doing the math:

$$M_0 = N\gamma^2 h^2 B_0 I(I+1)/(3k_b T)$$

So the macroscopic magnetization (M₀) depends on:

- B₀

What happens in NMR:

To induce NMR transitions a radiofrequency pulse is applied to the sample. The pulse is generated by a linear alternating electromagnetic field (B_1) along a transverse axis. This results in M_0 being rotated into the XY plane.

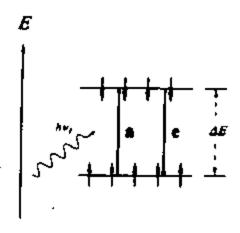
Resonance Condition:

$$h \upsilon_1 = \Delta E$$

Absorption: Lower energy ⇒ higher enrgy

Emission: Higher energy ⇒ Lower energy

Saturation: Equal number of emissions to absorption.



Flower 1-6.

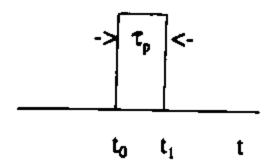
Larmor frequency: $v_1 = v_L = |\gamma| 2\pi |B_0|$

At 11.7T: for ${}^{1}\text{H} \gamma = 2.675 \times 10^{8} \ \upsilon_{0} = 500 \text{ MHz}$ For ${}^{13}\text{C} \gamma = 0.6726 \times 10^{8} \ \upsilon_{0} = 125 \text{ MHz}$

<i>B</i> ₁ [T]	Resonance frequencies (MHz)	,
	чн	¹³ C
1.41	60	15.1
1.88	80	20.1
2.11	90	22.63
2.35	100	25.15
4,70	200	50.3
5.87	250	62.9
7.05	300	75.4
9.40	400	100.6
11.74	500	125.7
14.09	600	150.9
17.62	750	188.2
18.79	800	201.2

Pulsed NMR:

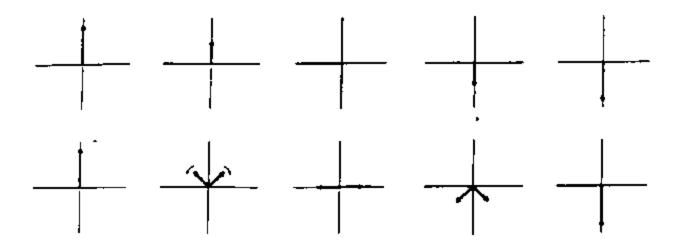
All nuclei are excited simultaneously by a radiofrequency pulse.



Pulse angle is proportional to the pulse width (duration) and pulse power (Magnitude of B₁ field).

The coil is arranged so that this field is perpendicular to the applied field (in the x-y plane)

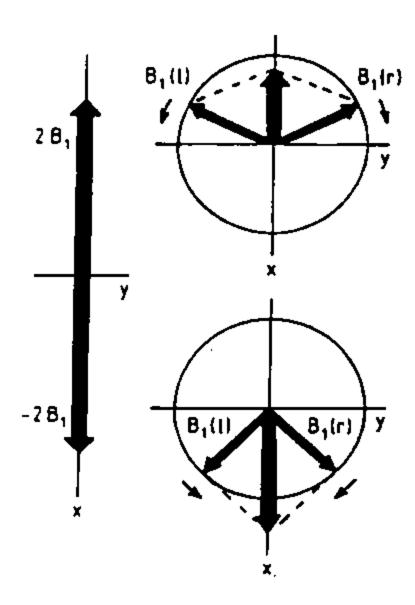
The oscillating magnetic field is equivalent to two counterrotating magnetization vectors.



We can use this pair of counter-rotating vectors as an equivalent representation of the rf signal.

We will represent the rf along the x-direction as two vectors with the same magnitude (B,)

- one vector rotates clockwise B₁ (r)
- one vector rotates c-clockwise B₁ (l)



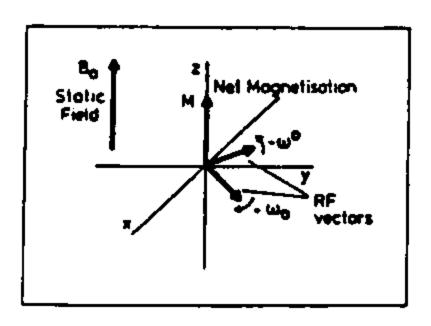
Only one of the two components is capable of interacting with the precessing nuclear dipoles (B₁).

The sample magnetization is static along the z-axis.

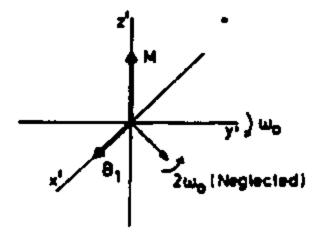
What happens when the rf magnetization (which is moving) interacts with the sample magnetization?

The Rotating Frame:

- Experimentally, detected signal is subtracted from the carrier frequency. We chose a set of coordinates that rotate along with the nuclear precession.



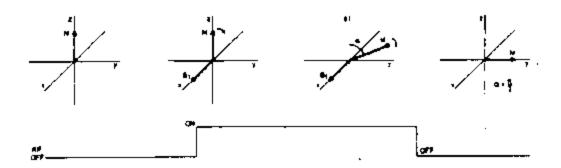
LAB FRAME



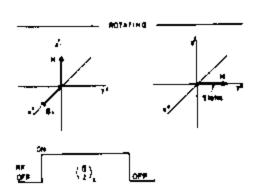
ROTATING FRAME

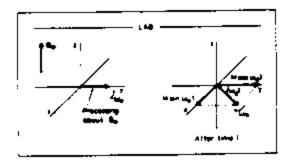
How can we use this to look at rf pulses(B_1)?

- In the rotating frame B₁ is at right angles to M.
- The net result is to produce a torque acting around B₁ at a speed depending on the field strength.

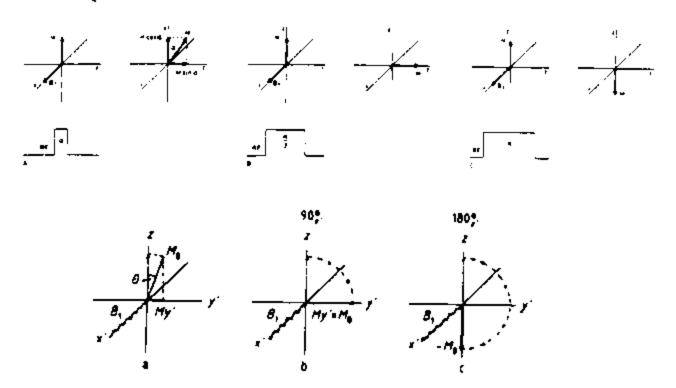


How does this look in our two frames of reference?

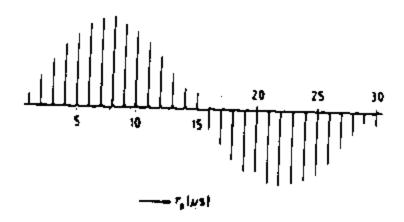




What happens to M as we increase the length of the pulse?

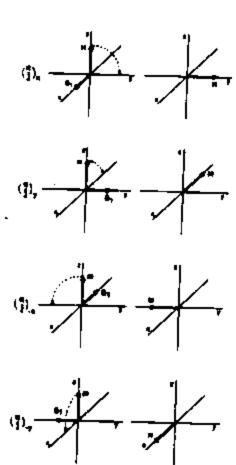


How does this effect the NMR signal?

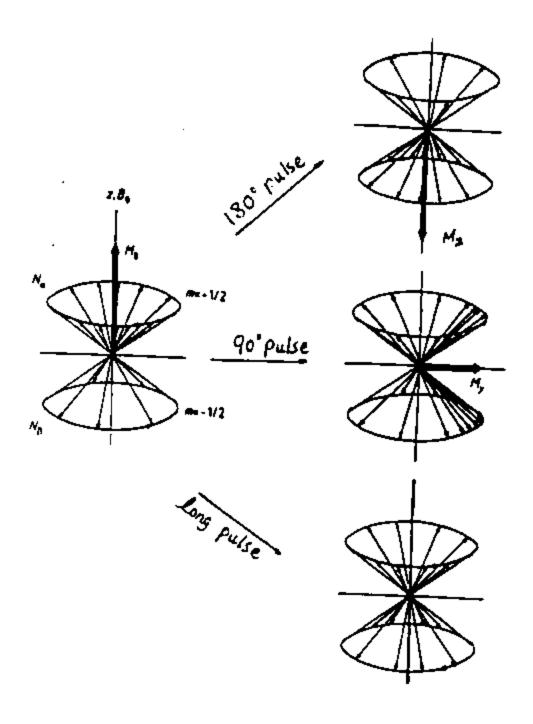


Our receiver is oriented with its axis along the ydirection. θ = pulse angle or pulse flip-angle.

Four common pulses and their phases (φ):



Pictorial representation of phase coherence:



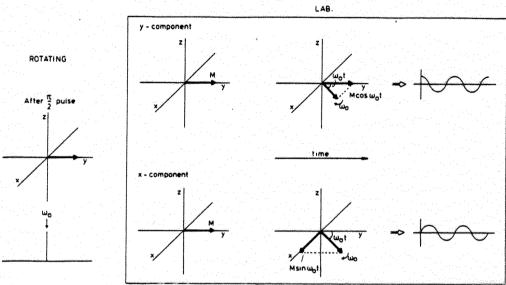


Figure 4.13 A closer look at the sample magnetisation in the lab. frame after a $(\pi/2)_x$ pulse reveals *two rf* signals, differing only in phase (by 90°).

