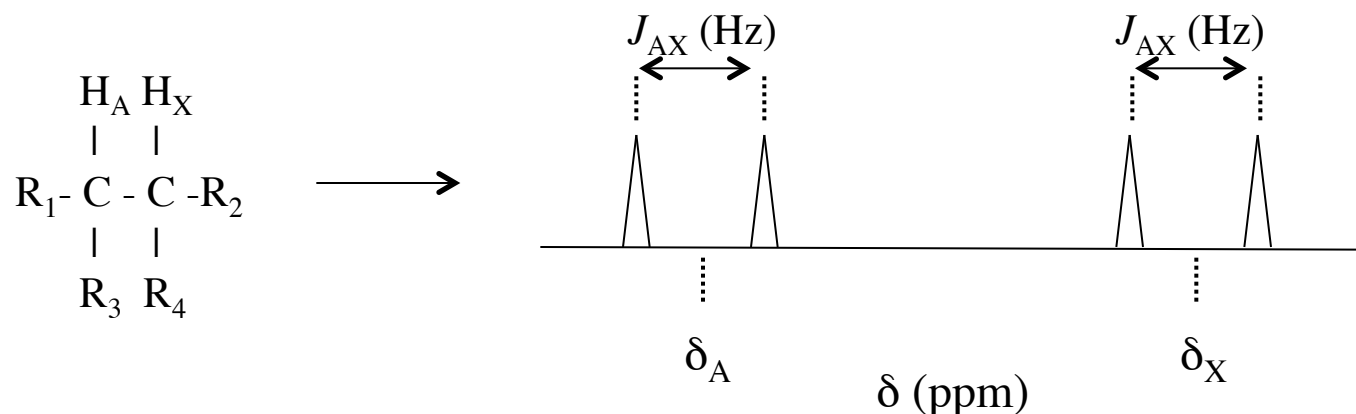


COUPLING

THROUGH-BOND COUPLING

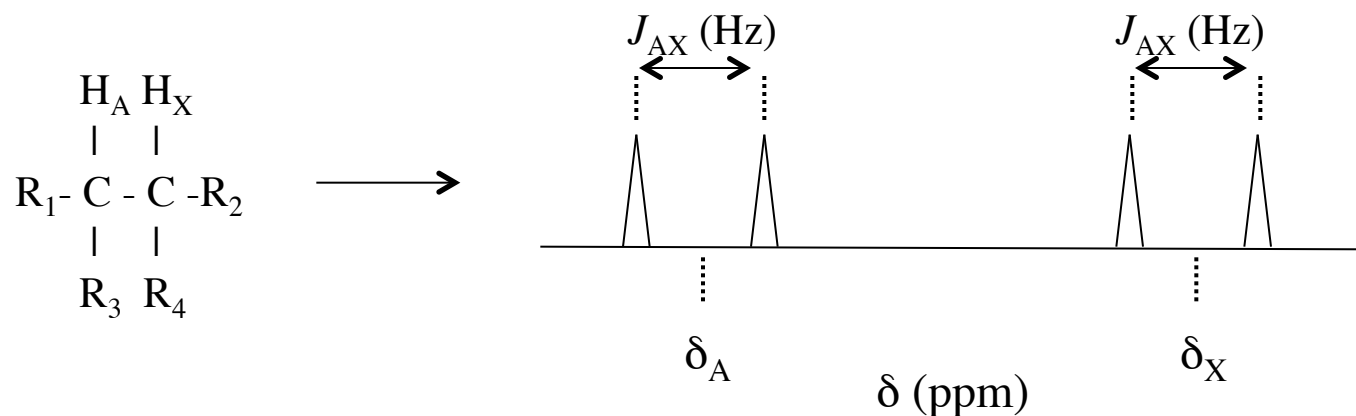
- Frequency-domain NMR *signals* are often split into two or more component *peaks* due to the effect of directly bonded nuclei or neighboring nuclei 2 or 3 bonds away



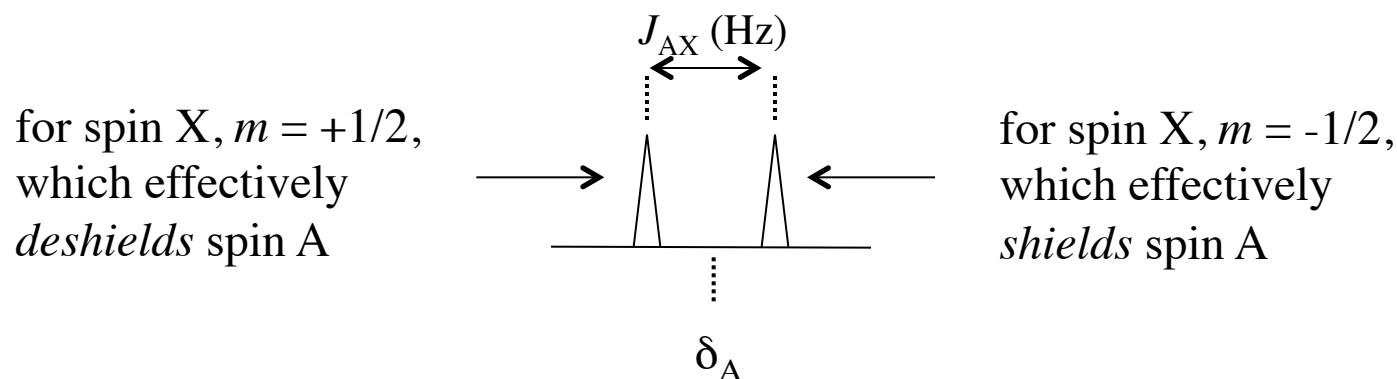
- This is due to *through-bond coupling*, also known as *scalar coupling*, *indirect coupling*, *spin-spin coupling*, or *J coupling*.....
- In the above example, the *signal* from H_A is split into two *peaks* (called a doublet), and the *signal* from H_X is, likewise, split into two *peaks* (a doublet)
- A signal split into two peaks is a *doublet*, a signal split into three peaks is a *triplet*, and so on (*quartet*, *quintet*, *sextet*, *septet*)

THROUGH-BOND COUPLING

- Coupling is mediated by bonding electrons and results in small resonance frequency changes depending on the spin state(s) of neighboring nuclei

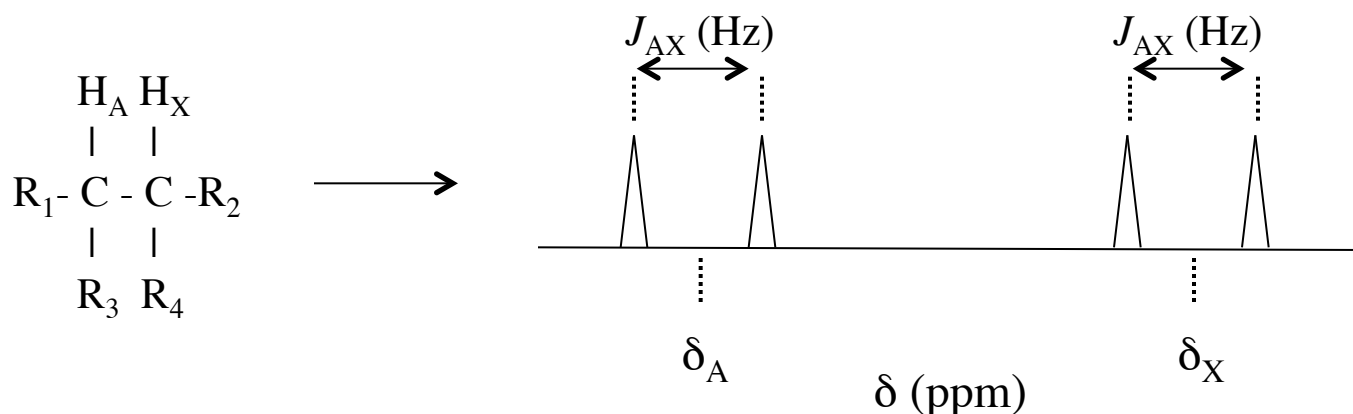


- In the example above, if we consider spin A (for instance) in an ensemble of molecules, in some of the molecules spin X will be in the α ($m = +1/2$) state, and in others spin X will be in the β ($m = -1/2$) state
- The result is two slightly different resonance frequencies for those A nuclei with X in the α state compared to those A nuclei with X in the β state



THROUGH-BOND COUPLING

- The magnitude of the splitting between the component peaks of the signal is called the *coupling constant*, or J



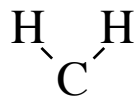
- J is usually given a superscript and subscript, the former describing the number of bonds separating the coupled nuclei, the latter designating the two atoms involved
-for the example above, $^3J_{AX}$ or $^3J_{HA,HX}$
- J is *always* measured in Hz
- J (in Hz) is *independent* of the magnitude of B_0
- The value of J measured from the splitting of the signal from one of the coupled nuclei is always the same as the value measured at the signal from the other coupled nucleus
- The chemical shift of the nucleus is the center of the multiplet (δ_A and δ_X)

MAGNITUDES OF THROUGH-BOND COUPLING CONSTANTS

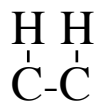
- The magnitude of the coupling constant, J , depends on the number of intervening bonds, as well as other structural factors
- The 1-bond H,H coupling constant is very large
- Typical 2-bond (***geminal***) H,H coupling constants are small (5-10 Hz), but occasionally can be much larger (for CH₂O, $^2J_{\text{H,H}} \approx 40$ Hz)
- Typical 3-bond (***vicinal***) H,H coupling tend to be somewhat larger than geminal coupling constants, and have a strong dependence on the torsional (dihedral) angle between the two C-H bonds (***Karplus relationship***)
- Typically, 4-bond couplings are too small to be observed, but, in special cases are large enough to be measured
- Normaly, 5-, 6-, etc.- bond couplings are too small to be observed



$$^1J_{\text{H,H}} \approx 275 \text{ Hz}$$



$$^2J_{\text{H,H}} \approx 5-10 \text{ Hz typically}$$



$$^3J_{\text{H,H}} \approx 5-20 \text{ Hz typically}$$



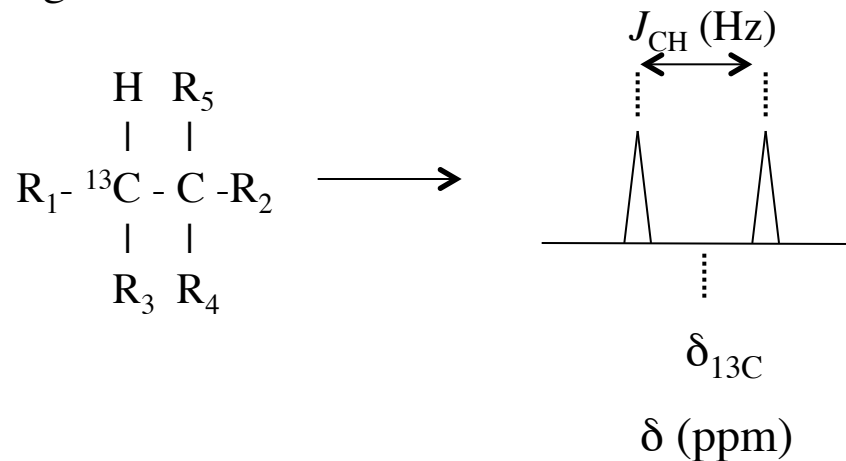
$$^4J_{\text{H,H}} \approx \text{usually small (unobservable), but can be as large as 7 Hz}$$



$$^5J_{\text{H,H}} \approx \text{usually too small to be observed}$$

^{13}C - ^1H THROUGH-BOND COUPLING

- In ^{13}C NMR spectra, through-bond coupling between ^1H and ^{13}C nuclei also results in splitting of ^{13}C NMR signals



- As with the H,H coupling constant, the magnitude of the ^{13}C ,H coupling constant depends on the number of intervening bonds, and other factors as well
- The 1-bond ^{13}C ,H coupling constant ($^1J_{\text{CH}}$) is very large
- Typical 2-bond ^{13}C ,H coupling constants range up to ~ 20 Hz, and often are not observed
- Normally 3-, 4-, 5-, etc.-bond couplings are too small to be observed

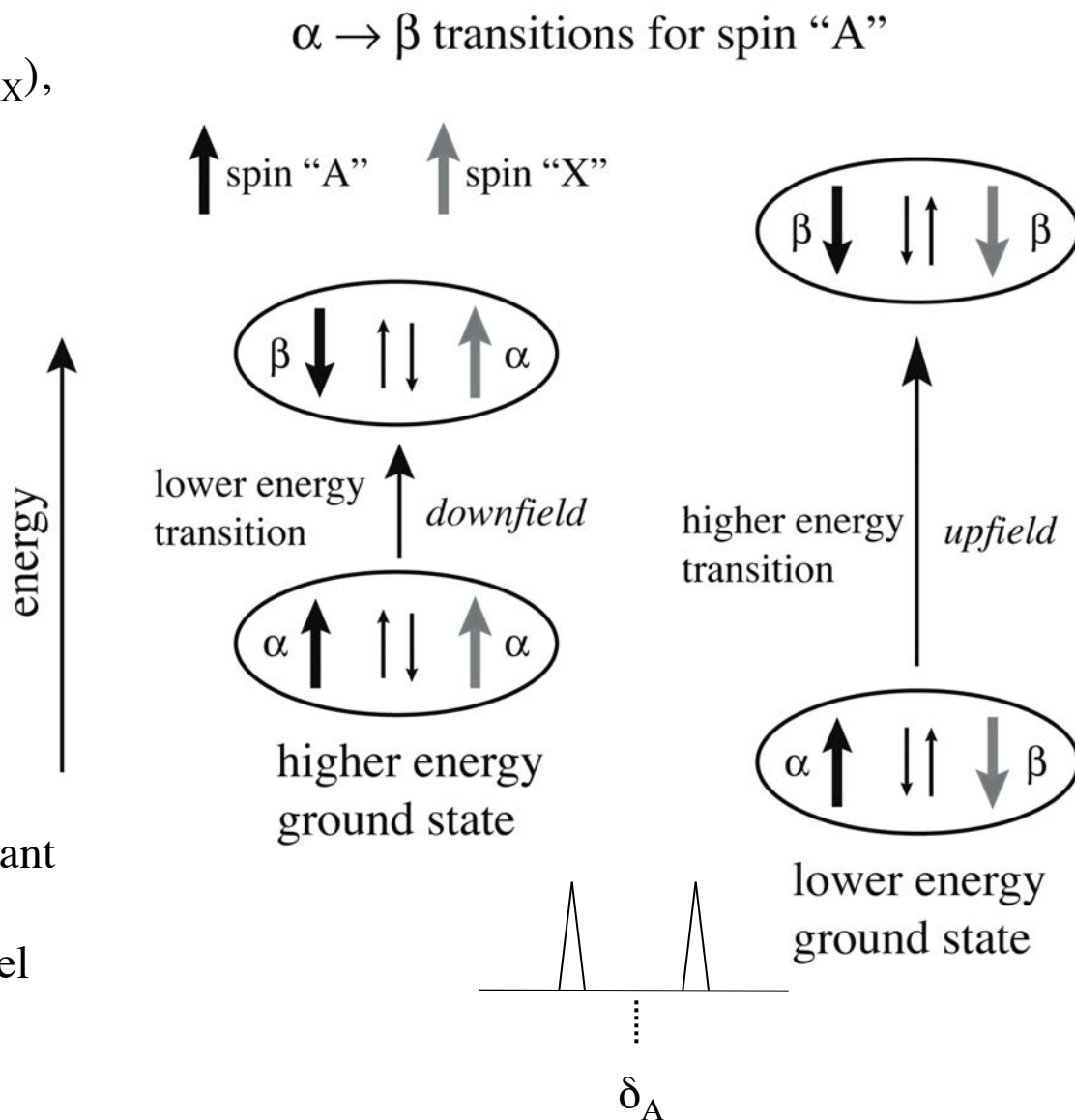
^{13}C -H $^1J_{\text{C,H}} \approx 100\text{-}250$ Hz (typical values are 120-150 Hz)

^{13}C -C-H $^2J_{\text{C,H}} \approx 1\text{-}20$ Hz typically

MECHANISM OF THROUGH-BOND COUPLING: DIRAC MODEL

- Coupling occurs through bonds, i.e. via interactions with bonding electron spins
- Consider the one-bond coupling ($^1J_{AX}$), for instance ^1H - ^{13}C

- Energetically preferred states are normally those where nuclear and electron spins are *antiparallel*
- Electron spins of bonding pair are antiparallel (*Pauli exclusion principle*)
- Thus, for low energy states, if nuclear spins are *antiparallel*, then couplings are said to be **positive**
- In other words, the coupling constant is positive if it stabilizes the state where nuclear spins are antiparallel
- Normally, for $\gamma > 0$, 1J is positive



MECHANISM OF THROUGH-BOND COUPLING: DIRAC MODEL

- Coupling occurs through bonds, i.e. via interactions with bonding electron spins

- Consider the two-bond coupling ($^2J_{AX}$),
for instance ^1H - ^{12}C - ^1H (geminal ^1H - ^1H coupling)

-Energetically preferred states are normally those where nuclear and electron spins are *antiparallel*

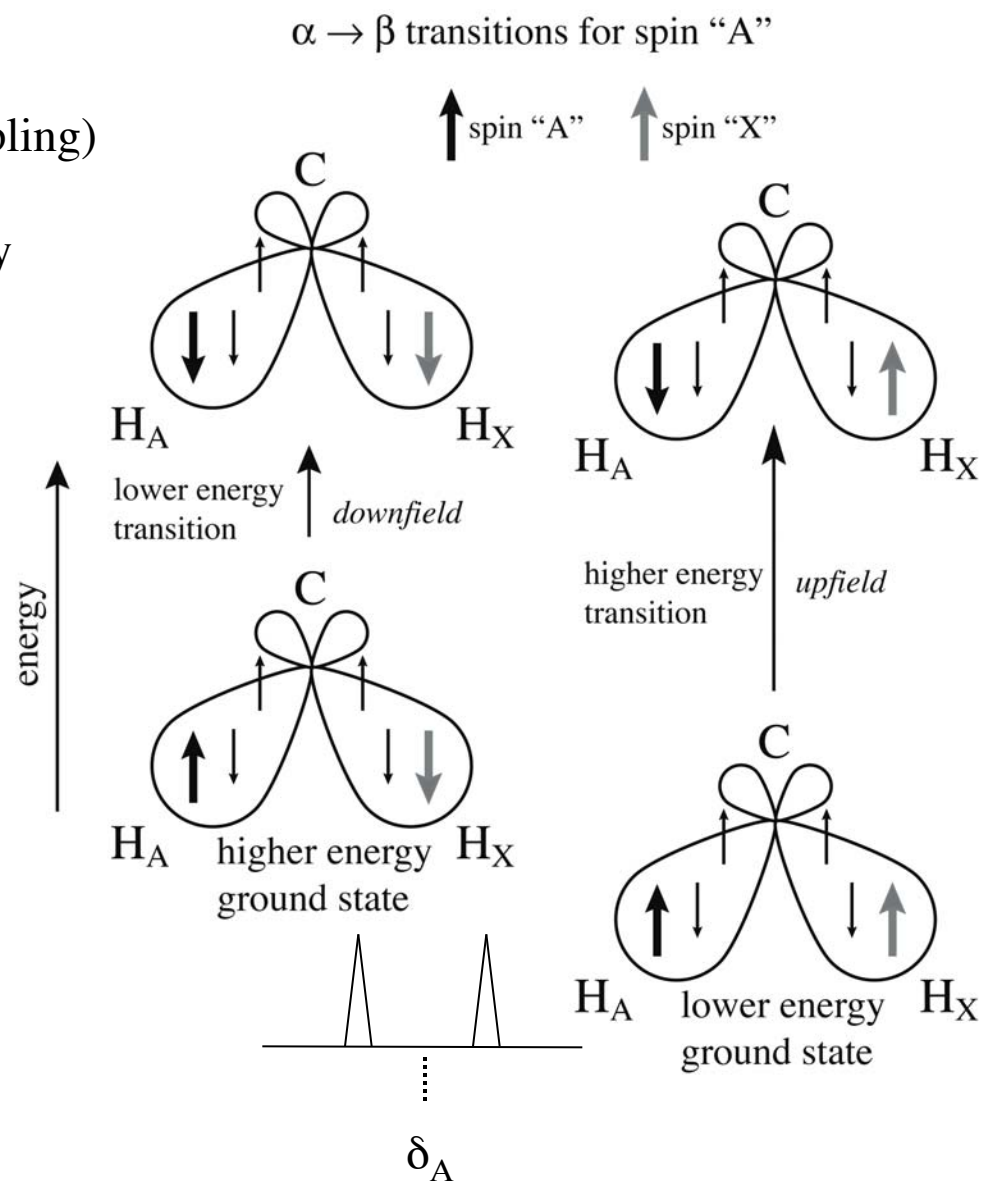
-Electron spins of bonding pair are antiparallel (*Pauli exclusion principle*)

-The energetically preferred state for the bonding electrons on the carbon atom is when these are parallel (*Hund's rule*)

-Thus, for low energy states, if nuclear spins are *parallel*, then couplings are said to be **negative**

-In other words, the coupling constant is negative if coupling stabilizes the state where nuclear spins are parallel

-Normally, for $\gamma > 0$, 2J is negative



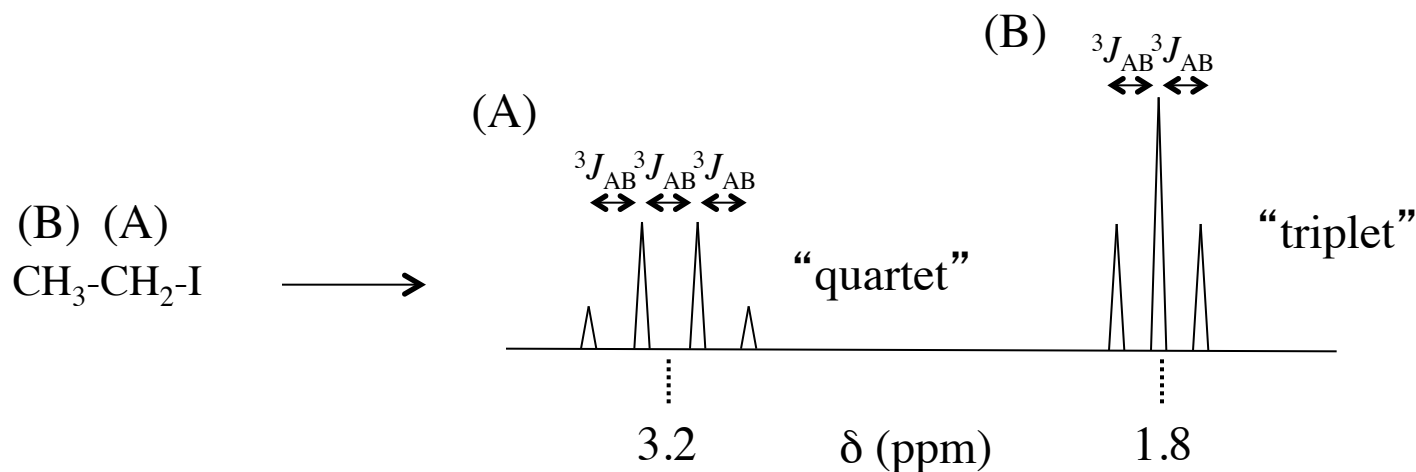
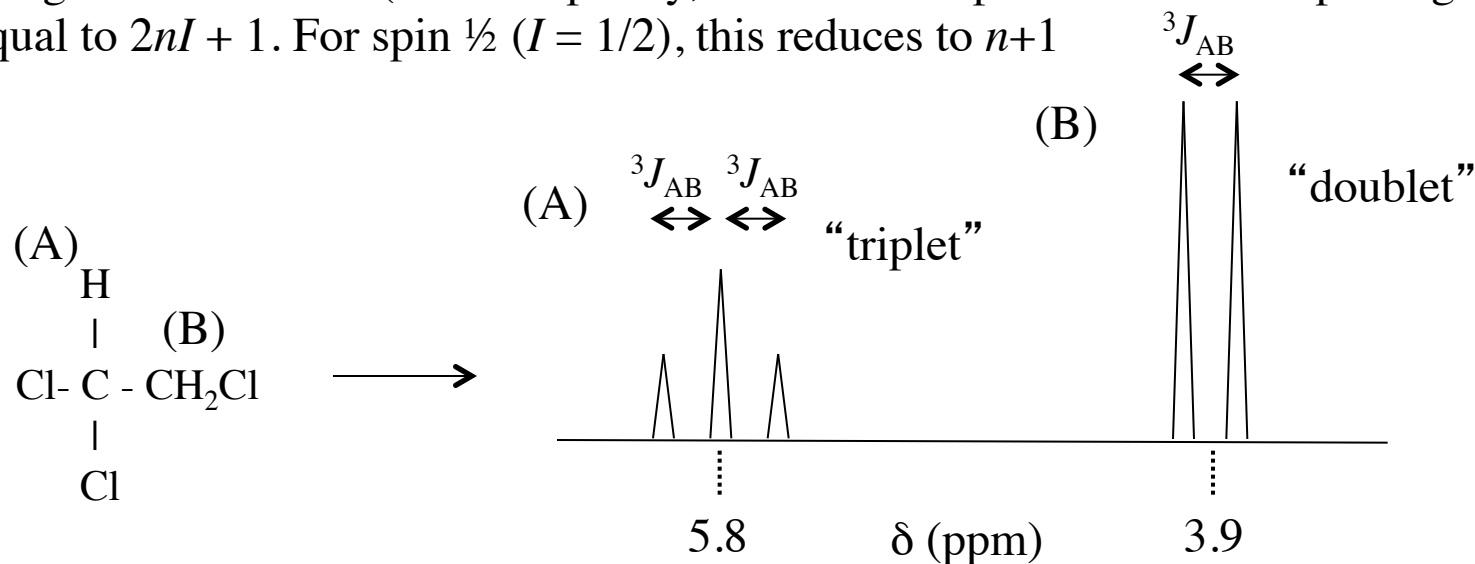
MECHANISM OF THROUGH-BOND COUPLING

- Fermi contact: interaction of nuclear spins via bonding electrons
 - depends on electron density at the pair of nuclei
 - depends on the “*s*” electrons, or *s* character of the bond(s) between the nuclei
 - electron spins must be correlated, i.e. there must be a bond
- Dependence on “*s*” character
 - consider one bond ^{13}C - ^1H couplings:

$\text{CH}_3\text{-CH}_3$	s, sp^3	$1 \times 1/4 = 1/4$	125 Hz
$\text{CH}_2\text{=CH}_2$	s, sp^2	$1 \times 1/3 = 1/3$	156 Hz
C_6H_6	s, sp^2	$1 \times 1/3 = 1/3$	158 Hz
$\text{HC}\equiv\text{CH}$	s, sp	$1 \times 1/2 = 1/2$	249 Hz

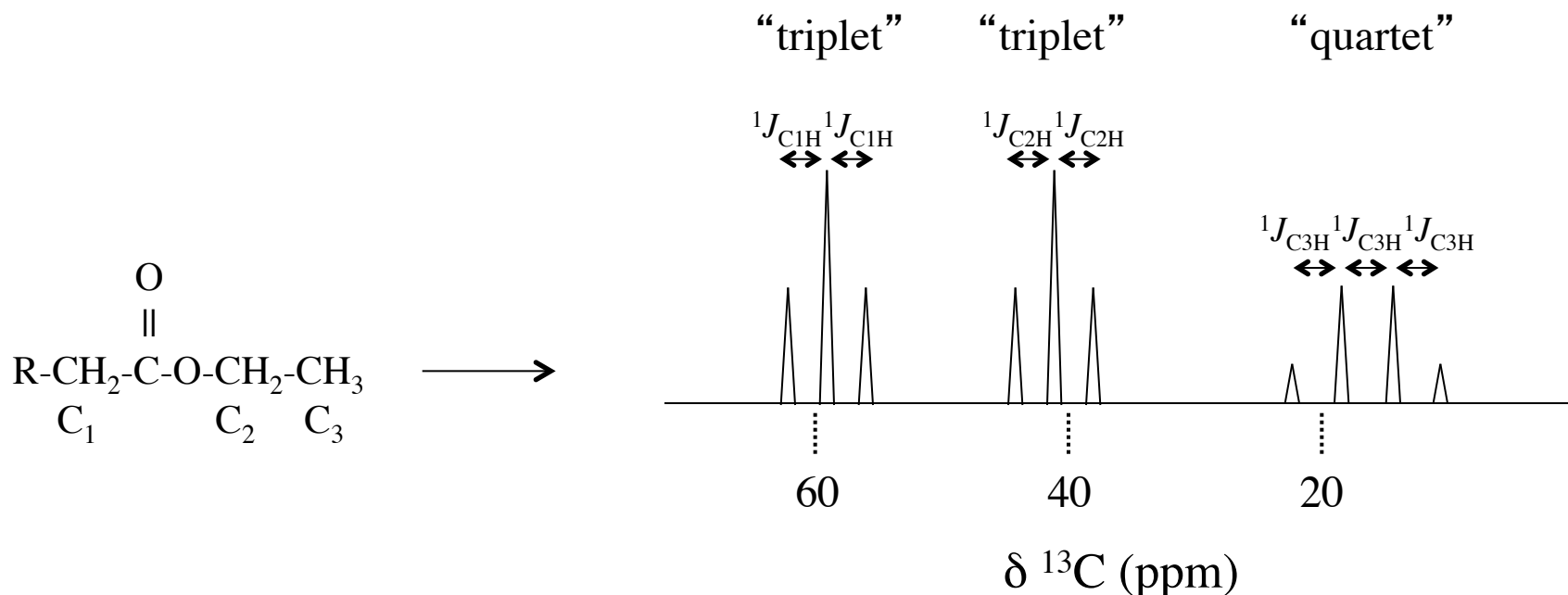
SPIN-SPIN SPLITTING: $N+1$ RULE

- Multiplet splitting patterns are determined by the number of nuclei (n) doing the splitting and the spin angular momentum quantum number (I)
- The general rule is M (the multiplicity, or number of peaks in the multiplet signal) is equal to $2nI + 1$. For spin $\frac{1}{2}$ ($I = 1/2$), this reduces to $n+1$



SPIN-SPIN SPLITTING: $n+1$ RULE

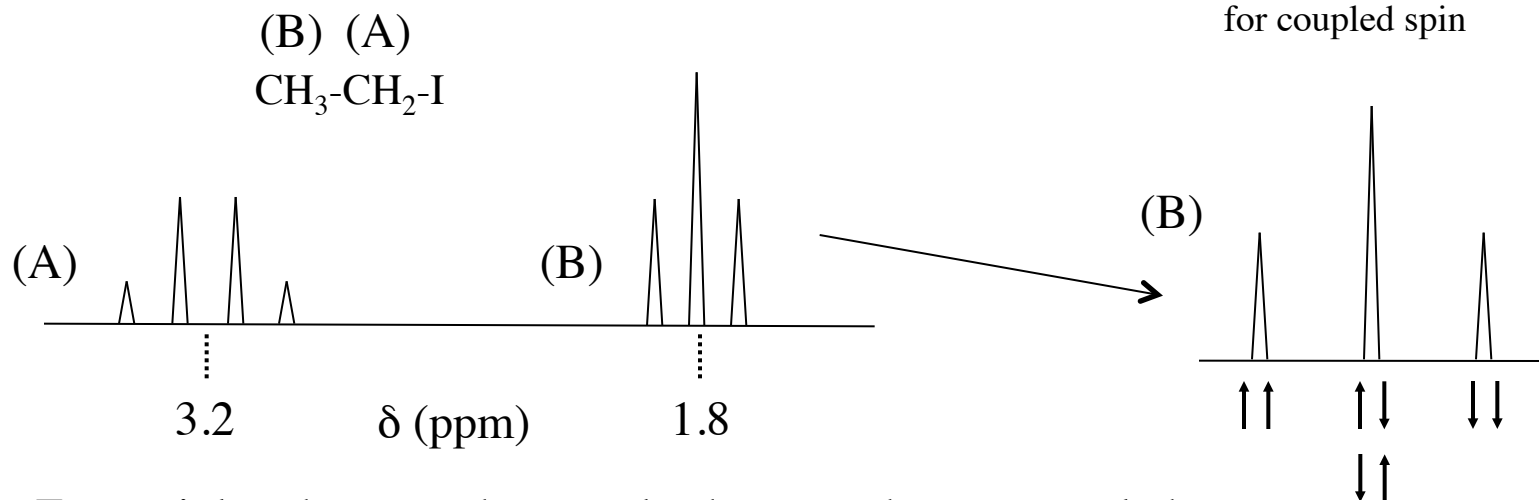
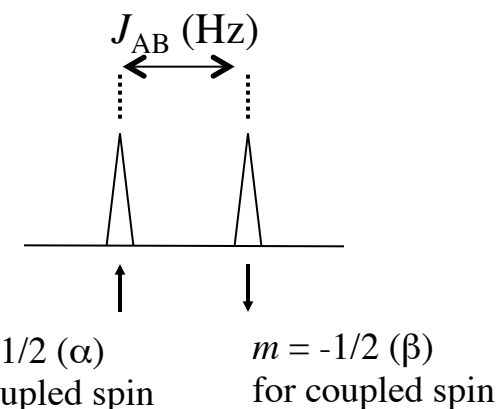
- In ^{13}C NMR spectra, normally only the 1-bond ^{13}C -H couplings are observed
- Both ^1H and ^{13}C are spin $\frac{1}{2}$ ($I = 1/2$), so the $n+1$ rule applies



- ^{13}C nuclei are also coupled to directly bonded ^{13}C
- However, because ^{13}C comprises only about 1% of all C at natural abundance, the chance that a ^{13}C nucleus is next to another ^{13}C nucleus is very small, so no signal splitting normally results from ^{13}C - ^{13}C coupling
- Signal splitting resulting from ^{13}C - ^{13}C coupling will be observed in compounds synthesized with excess ^{13}C

SPIN-SPIN SPLITTING: INTENSITIES AND PASCAL'S TRIANGLE

- For a doublet, we saw that there are two peaks, because the coupled spin can be in either the α or β state
- The *peak heights are equal* because, for the coupled spin, there are essentially equal numbers of spins in α and β states



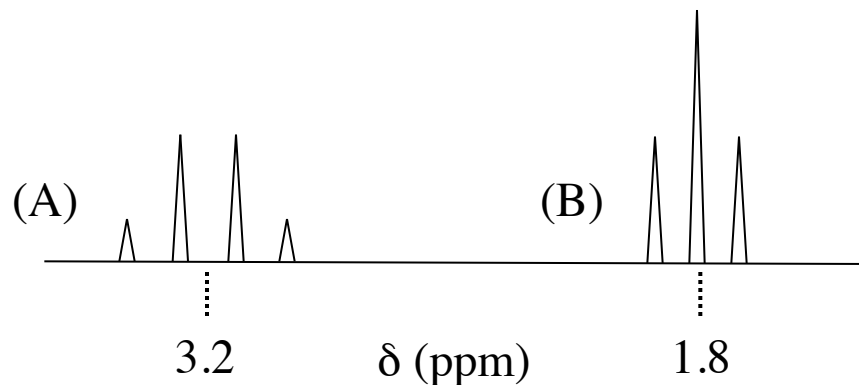
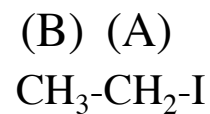
- For a triplet, there are three peaks, because the two coupled spins can both be in the α state, they can both be in the β state, or one can be α and the other β
- The *relative peak heights are 1:2:1* because there are 2 ways that one spin can be α and the other β , and only one way both can be α and only one way both can be β

SPIN-SPIN SPLITTING: INTENSITIES AND PASCAL'S TRIANGLE













- The relative intensities of the multiplet components in a signal split by spin-spin coupling can be obtained from Pascal's triangle

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relative intensities (in the spectrum)	1 3 3 1	3 6 3
relative intensities (in the multiplet)	1 3 3 1	1 2 1



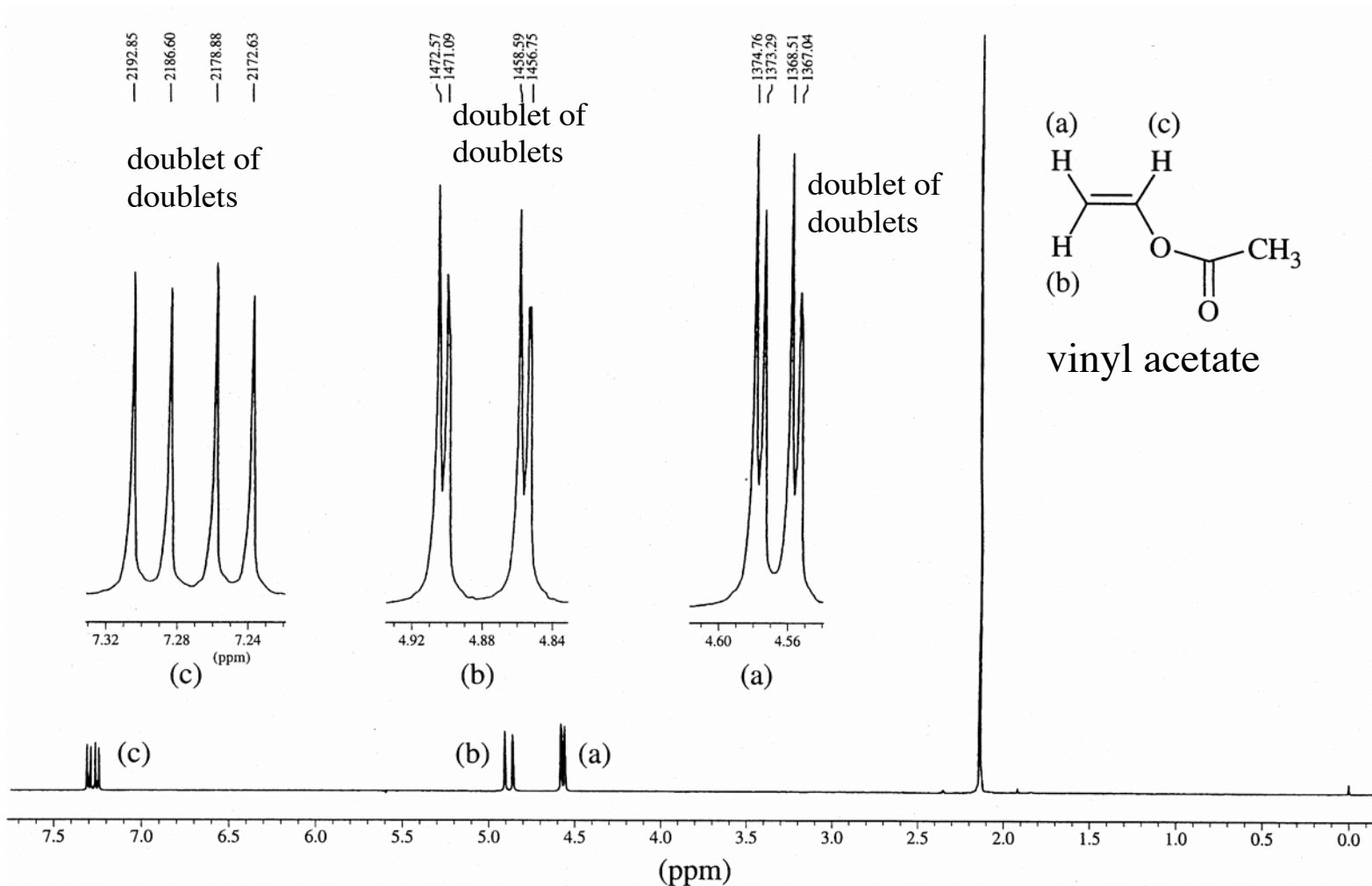
COMMON SPLITTING PATTERNS

	$\begin{array}{c} \quad \\ \text{X}-\text{CH}-\text{CH}-\text{Y} \\ (\text{X} \neq \text{Y}) \end{array}$	
	$\begin{array}{c} \\ -\text{CH}_2-\text{CH} \\ \end{array}$	
	$\begin{array}{c} \text{X}-\text{CH}_2-\text{CH}_2-\text{Y} \\ (\text{X} \neq \text{Y}) \end{array}$	
	$\begin{array}{c} \\ \text{CH}_3-\text{CH} \\ \end{array}$	
	CH_3-CH_2-	
	$\left\{ \begin{array}{l} \text{CH}_3 \\ \text{CH}_3 \end{array} \right\} \text{CH}-$	

- Some commonly observed splitting patterns

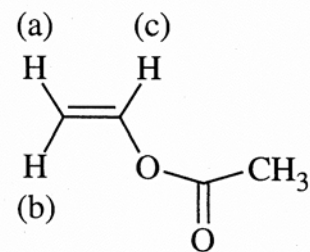
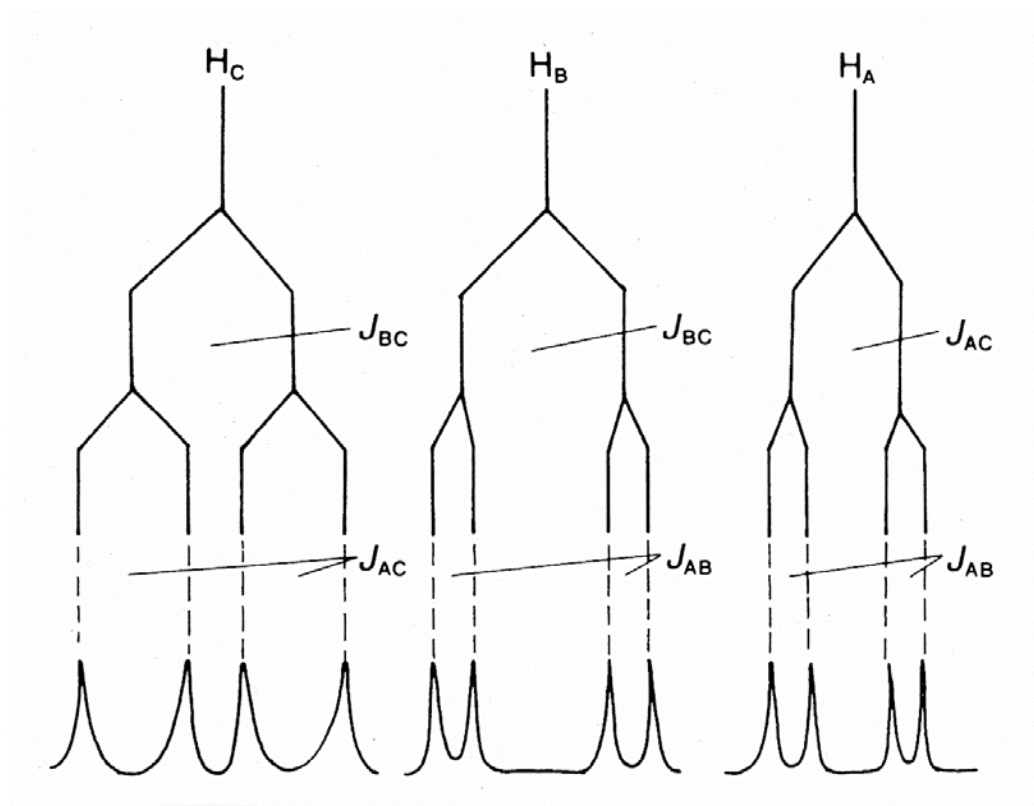
MULTIPLE/COMPLEX SPLITTING

- Split signals can be split again by coupling to additional spins
 - in vinyl acetate, for instance, the signal from “c” is split into a doublet by “b”, and this doublet is split into a doublet of doublets by “a”
 - likewise, “b” is split by “a” and “c”, and “a” is split by “b” and “c”



MULTIPLE/COMPLEX SPLITTING

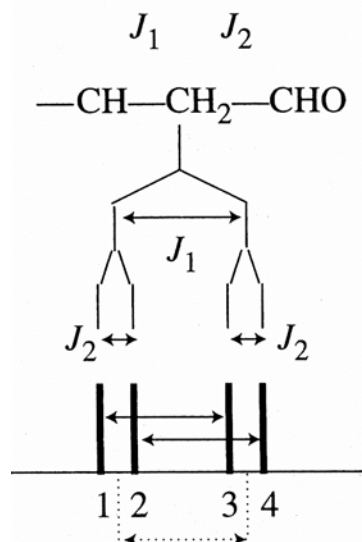
- The coupling constant between any two spins will be observed in the signal from each of the spins
- Observation of a coupling in more than one signal identifies coupled spins



vinyl acetate

EXTRACTING THE COUPLING CONSTANTS

- Analysis and determination of coupling constants

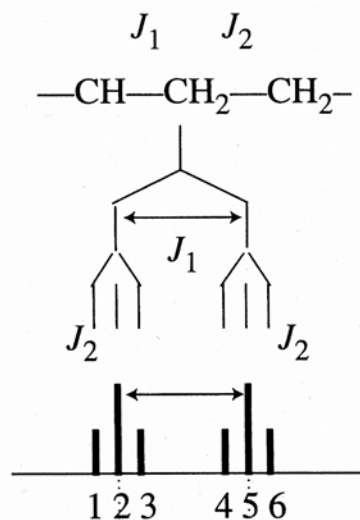


Doublet of Doublets (dd)

To obtain J_1 measure the difference between lines 1 and 3, or 2 and 4, in Hz.*

*Do not try to find the centers of the doublets!

J_2 is the spacing between lines 1 and 2, or 3 and 4



Doublet of Triplets (dt)

To obtain J_1 measure the difference between the most intense lines (2 and 5) in Hz

J_2 is the spacing between lines 1 and 2, or 2 and 3, or those in the other triplet.

SPIN DECOUPLING

- It is often advantageous to reverse or remove the splitting caused by spin-spin coupling
- This is called *spin decoupling*
- Spin decoupling (or just “decoupling”) can be used for several reasons
 - to simplify spectra
 - to assist in identification of coupling between nuclei
 - to improve signal-to-noise
- We’ ll discuss spin decoupling in the context of “double resonance” experiments