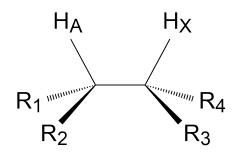
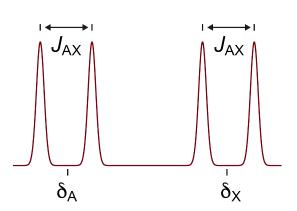
Origin of Scalar Couplings

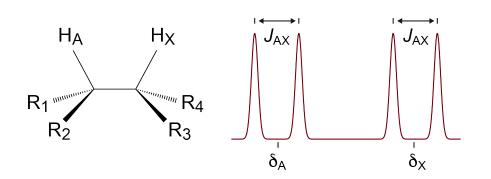
BCMB/CHEM 8190

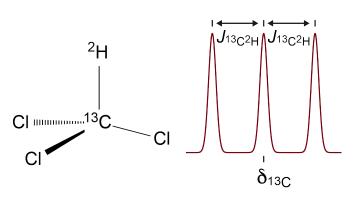
- Splitting of NMR signals due to through-bond interactions between nuclei is called scalar coupling (or J coupling or through-bond coupling)
- Scalar couplings are important:
 - the magnitudes of the splittings and splitting patterns contain useful structural information
 - they are central to magnetization transfer (COSY, HSQC)
 - readily measured (peak separation, 2D *J* resolved, intensity)
- Below, scalar coupling between H_A and H_X results in splitting of the signal from each into two peaks (doublets)



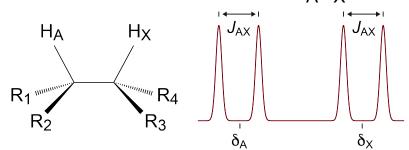


- Scalar coupling is mediated by bonding electrons
- The energy of a nucleus coupled to another will change slightly depending on the spin state of the coupled nucleus
- The energy difference results in small frequency changes
- Below (left), each ¹H nucleus ($I = \frac{1}{2}$) is split into two peaks, depending on whether the coupled ¹H nucleus is in the $m=+\frac{1}{2}$ (α) or $m=-\frac{1}{2}$ (β) state
- Below (right), the 13 C nucleus is split into three peaks, depending on whether the coupled 2 H nucleus (I = 1) is in the m=1, m=0, or m=-1 state





- The magnitude of the splitting is called the coupling constant, and is typically symbolized by J
- *J* is measured in Hz, and is magnetic field strength independent
- J measured from the splitting of the signal from one of the coupled nuclei is the same as the value measured at the signal from the other coupled nucleus
- The chemical shift of the nucleus is the center of the multiplet
- J is often given a superscript and subscript, the former indicating the number of bonds separating the coupled nuclei, the latter designating the atoms involved (i.e. ${}^3J_{\rm AX}$, or ${}^3J_{\rm HAH_X}$)
- The magnitude of J depends on structural factors and the number of intervening chemical bonds:



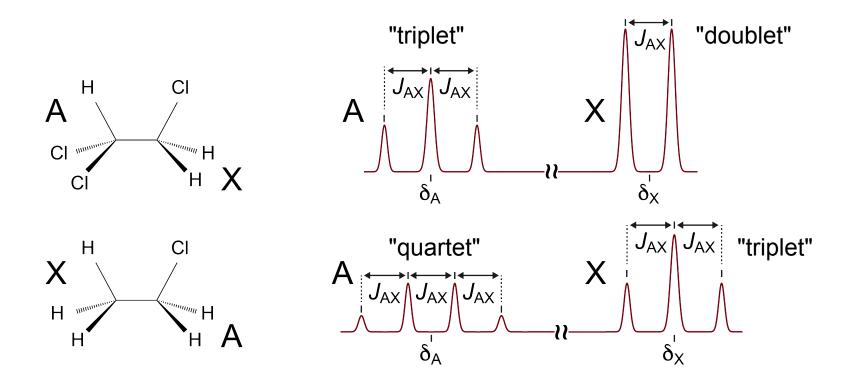
$$^{1}J_{\rm H,H}$$
≈ 275 Hz $^{2}J_{\rm H,H}$ ≈ 5-10 Hz typically

³J_{H H} ≈ 5-20 Hz typically

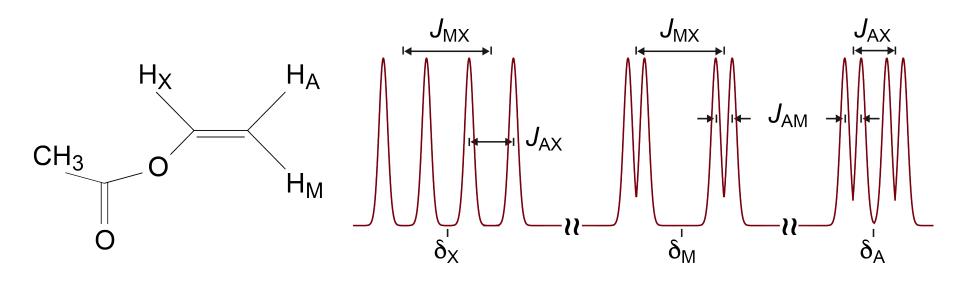
 $^{^4}J_{\rm H,H}$ often unobservable, but as large as 7 Hz

⁵J_{H.H}≈ usually unobservable

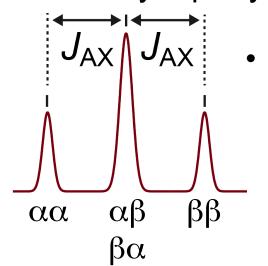
- The multiplet splitting pattern of a signal split by another nucleus (or group of equivalent nuclei) depends on n, the number of nuclei doing the splitting, and l
- The multiplicity (number of peaks in the multiplet signal) is equal to 2nI+1. For spin $\frac{1}{2}$, $I=\frac{1}{2}$, and this reduces to n+1.



- Split signals can be split again by coupling to additional nuclei
- In vinyl acetate (below), the signal from "X" is split into a doublet by "M", and this doublet is split into a doublet of doublets by "A"
- Likewise, "M" is split by "A" and "X", and "A" is split by "M" and "X"



- For a doublet , the two peaks signify that the coupled nucleus ($I=\frac{1}{2}$) is in either the $m=+\frac{1}{2}$ (α) or $m=-\frac{1}{2}$ (β) state
- The intensities of the peaks are the same, because the numbers of coupled nuclei in the α and β states are virtually
- For coupling to a spin 1 (*I* = 1) nucleus, the intensities of the three peaks are the same, because the *m*=1, *m*=0, and *m*=-1 states are virtually equally populated



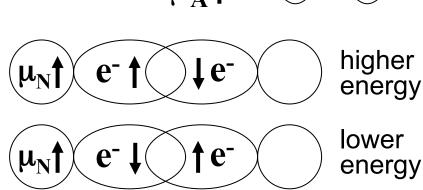
For coupling to equivalent nuclei (I = ½) intensities are not equal: peaks from states with nearly identical energies (αβ, βα) overlap (¹H next to methylene, triplet with 1:2:1 intensities)

*J*13_C2H **∖***J*13_C2H

 δ_{13C}

The Origin of Scalar Couplings

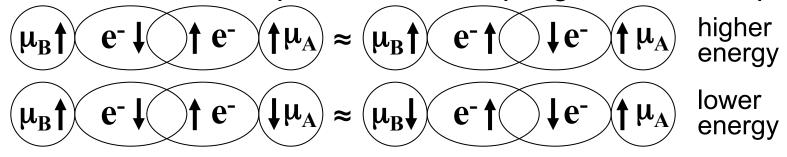
- Dipolar coupling model (E ∝ (3cos²θ 1) / r³) cannot account for scalar coupling (r is fixed, molecule tumbles in space, result is average over all values of θ, result = 0)
- Scalar coupling is due to a Fermi Contact mechanism (interactions between nuclear and electron magnetic moments)
- Electron moments in bond are antiparallel (Pauli). Lower energy state is when nuclear moment is antiparallel to adjacent electron moment



 $\mathbf{B_0}$

H

 When coupling lowers the energy of the state where nuclear moments are antiparallel, the coupling is said to be positive



Fermi-Contact Expression for Scalar Coupling

• In the limit of weak coupling $(v_i - v_j >> J_{ij})$ the scalar coupling term in the Hamiltonian reduces to products of I_z (simple α and β spin functions, ignore x- and y-components of the dot product)

$$E_J = \sum_{i \neq j} J_{ij} I_i \cdot I_j \approx \sum_{i \neq j} J_{ij} I_{zi} I_{zj}$$

 Expression for Fermi contact energy comes from second order perturbation treatment of spin contributions to electronic energy

$$E = (4g\beta\gamma_{N}h/3)^{2} \sum_{i,k} \langle \psi_{0} | \delta(\mathbf{r}_{kA})\delta(\mathbf{r}_{kX})S_{jz}S_{kz} | \psi_{0} \rangle I_{zA}I_{zX} \qquad \text{(homonuclear case)}$$

- $\gamma_A \gamma_X$ (i.e. γ_N^2) and $I_{ZA} I_{ZX}$ dependence as expected
- Dirac delta functions $(\delta(r=0)=1, \delta(r\neq 0)=0)$ at the positions of the nuclei: probability of finding electron k at nucleus A $(\delta(r_{kA}))$ and nucleus X $(\delta(r_{kX}))$ (only s orbitals are non-zero at the nucleus)
- If the electrons are not correlated (i.e. there is no bond between the nuclei), the expression for the average of the $S_{iz}S_{kz} = 0$
- So, there has to be a bond and the magnitude of the coupling constant depends on the 's' character of the bond

Some predictions for one-bond couplings

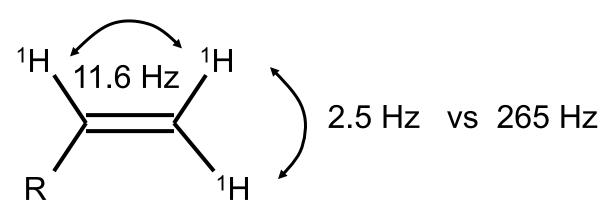
Couplings are short range and depend on electron densities at the coupled nuclei (ie, the product of "s characters") as well as the values of γ for the nuclei.

$$13C-13C$$
 CH_3-CH_3 $35 Hz sp^3 - sp^3$ $1/4 x 1/4 = 1/16$ $CH_2=CH_2$ $67 Hz sp^2 - sp^2$ $1/3 x 1/3 = 1/9$ C_6H_6 $57 Hz sp^2 - sp^2$ $1/3 x 1/3 = 1/9$ $HC=CH$ $171 Hz sp-sp$ $1/2 x 1/2 = 1/4$ $\frac{13C-1H}{CH_3-CH_3}$ $125 Hz s-sp3$ $1 x 1/4 = 1/4$ $CH_2=CH_2$ $156 Hz s-sp2$ $1 x 1/3 = 1/3$ C_6H_6 $158 Hz s-sp2$ $1 x 1/3 = 1/3$ $HC=CH$ $249 Hz s-sp$ $1 x 1/2 = 1/2$

 Knowledge of C-C and C-H couplings important for magnetization transfers in heteronuclear experiments

Predictions about Magnitudes of Scalar Couplings

- Wave functions (electronic) for ¹H-¹H can be calculated at a high level of theory – even at VB *prediction* of 200 Hz is good.
- Experimental ¹H-²H is 44 Hz; correcting for γs , $J_{H-H} = \gamma_{1H}/\gamma_{2H} \times 44 = 6.5 \times 44 = 265$ Hz
- Directly bonded pairs should have much larger couplings:



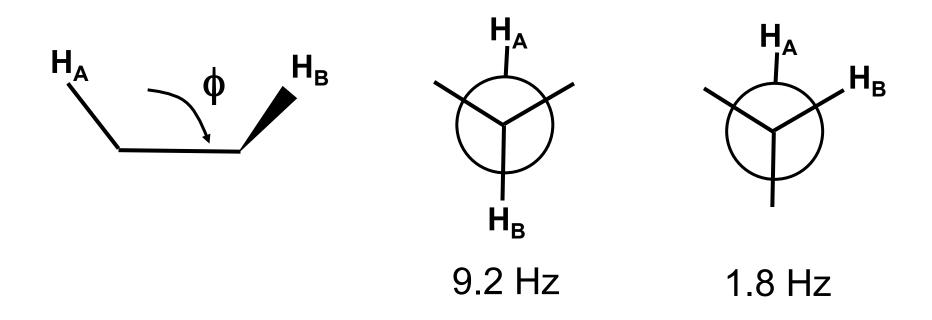
A two-bond coupling is small and sometimes negative – why?

- Electron moments in bonds are antiparallel (Pauli). Moments of bonding electrons on central carbon are parallel (Hund).
 Low energy state is now when nuclear moments are parallel
- Thus, the coupling would be negative and could have a large magnitude

(definition of negative coupling)

- Consider the possibility of correlated electron density directly between the two hydrogens (i.e. not via C)
 This coupling would be positive, and high energy
- Addition of the negative and positive contributions to the coupling constant would then tend to make the absolute value of the magnitude small

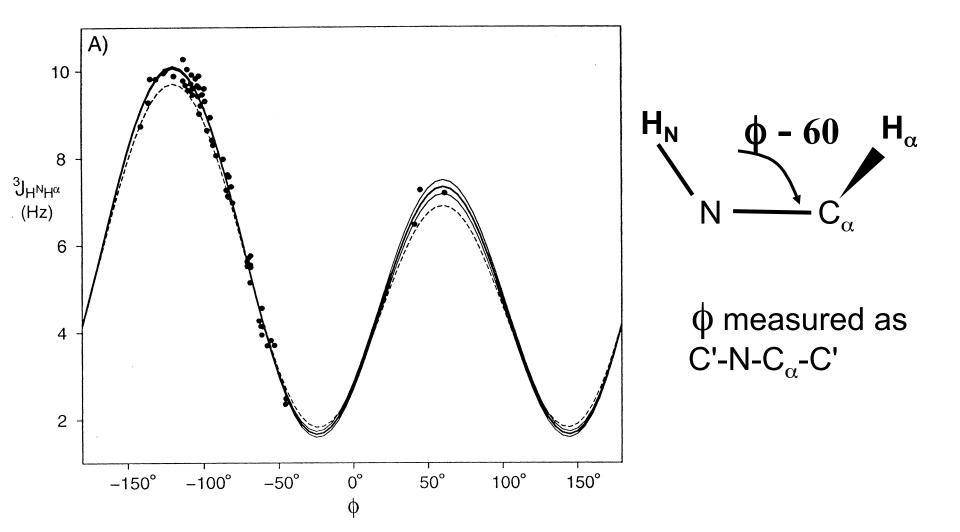
Vicinal Couplings (³J_{HH})



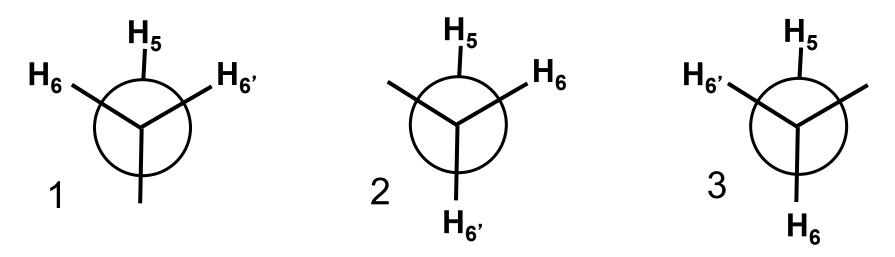
- Originally for H-C-C-H
- M. Karplus (1959) JCP 30 11 Karplus Equation
- $J_{HNHA} = 7.0 \cos^2(\phi) 1.4 \cos(\phi) + 1.7$ for peptides
- (Bax and Wang (1996) JACS 118 2492)

Wang-Bax Equation Compared to Ubiquitin Data

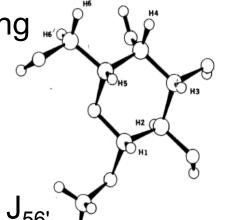
 Empirical parameterization of Karplus coefficients for proteins (Wang and Bax, 1996, J. Am. Chem. Soc. 118, 2483-2494)



Populations of Rotomeric States from Couplings An Application of the Karplus Equation



- Occurs in pyranosides, furanosides, and nucleosides
- Assume only minimum energy states populated
- Assume rapid equilibration averaging of coupling _{★♥}
- $J_{56} = 1.8 F_1 + 1.8 F_2 + 9.2 F_3$
- $J_{56}^{7} = 1.8 F_1 + 9.2 F_2 + 1.8 F_3$
- $F_1 + F_2 + F_3 = 1$
- 3 equations, 3 unknowns, can solve with J₅₆ and J₅₆



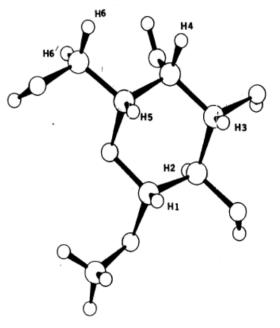
Application of Karplus Equations (cont)

Measure J_{65} and $J_{6'5}$ for β -Me-galactose

7.0 Hz, 3.0 Hz

Using averaging formula find:

0.15, 0.7, 0.15 for states 1, 2, and 3



Reference on scalar coupling in Levitt: 217-223

Scalar Couplings Can be Predicted Theoretically

- "Interpretation of chemical shifts and coupling constants in macromolecules" D.A. Case, Curr. Opin. Struc. Biol. 10, 197-203 (2000).
- "Density functional calculations on disaccharide minics: studies on molecular geometries and spin-couplings", F. Cloran, I Carmichael, A.S. Serianni, JACS, 121, 9843-9851 (1999).

Couplings And Hydrogen Bonds

- "Direct observation of hydrogen bonds in nucleic acid base pairs by internucleotide ²J(NN) couplings", A.J. Dingley, S. Grzesiek, JACS, 120, 8293-8297 (1998).
- "NMR scalar couplings across Watson-Crick base pair hydrogen bonds in DNA …", K. Pervushin, … K. Wuthrich, PNAS, 95, 14147-14151 (1998).

 $^{15}N-H$ ^{15}N or $^{15}N-H$ $O=^{13}C$

 "N-15-H-1 Scalar Coupling Perturbation: An Additional Probe for Measuring Structural Changes Due to Ligand Binding, Ma JH, Gruschus JM, Tjandra N, J. AM. CHEM. SOC., 131, 9884 (2009)

Chemical shifts and Coupling Constants: How Many Peaks can you Assign?

