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# Autothermal recirculating reactor (ARR) with Cu-BN composite as a stable reactor material for sustainable hydrogen release from ammonia

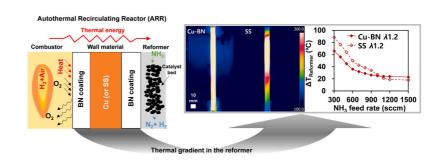
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#### HIGHLIGHTS

- Combustion of reformate H<sub>2</sub> was used as a heat source for NH<sub>3</sub> decomposition.
- Reforming efficiency of 70.95% was achieved with a lab-scale autothermal reformer.
- Fuel cell equivalent power of 84 W was obtained with CO<sub>X</sub>-free operation.
- Cu-BN was proposed as a viable reactor material for temperature-sensitive reactions.
- Autothermal NH<sub>3</sub> decomposition can be envisaged for onboard power generation.

## G R A P H I C A L A B S T R A C T



### ARTICLE INFO

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### ABSTRACT

Ammonia (NH<sub>3</sub>) has been proposed as a viable hydrogen (H<sub>2</sub>) carrier, but high reaction temperature and endothermic nature of NH<sub>3</sub> decomposition require an efficient reaction system to maximize useable energy from NH<sub>3</sub>. Adoption of carbon-free heat sources and efficient heat transfer to the reaction bed are crucial for sustainable H<sub>2</sub> release. Herein, the autothermal recirculating reactor (ARR) concept with the fractional utilization of the reformate H<sub>2</sub> as a clean combustion fuel is proposed and experimentally investigated. Additionally, BN-coated Cu as a composite reactor material is developed for heat transfer enhancement of high-temperature H<sub>2</sub> release reaction in a thermally-coupled NH<sub>3</sub> decomposition and H<sub>2</sub> combustion system. Coating performance against chemical degradation of Cu has been tested and confirmed. High NH<sub>3</sub> conversion of >99.6% and reforming efficiency of 70.95%, even with high fraction of heat loss owing to small scale validation, show feasibility of the as-proposed reformer. Operation of the suggested system is envisaged with self-sustained heat

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