Preparation PCRS11-Red Sea Medium

The RCC Team

Abstract

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Protocol

Preparation of Hepes-NaOH 1M Stock Solution (skip if already available)

Step 1.

To 250mL of H_20 , add gradually 119.15g of Hepes. Adjust pH at 7.5 and complete the volume at 500mL. Store in refrigerator.

AMOUNT

500 ml Additional info: Water

AMOUNT

119.15 g Additional info: Hepes

Preparation of Na2-EDTA/FeCl3 Stock Solution (skip if already available)

Step 2.

- To 40mL of HCl 0.1N, add gradually 1.080g of FeCl₃
- To 40mL of NaOH 0.1N, add gradually 1.488g of Na₂-EDTA.

Now mix both solutions and fill up to the final volume of 2L with sterile water. Store in refrigerator

■ AMOUNT

1.08 g Additional info: FeCl3

AMOUNT

40 ml Additional info: HCl 0.1N

AMOUNT

40 ml Additional info: NaOH 0.1N

■ AMOUNT

1.448 g Additional info: Na2-EDTA

Preparation of Sodium Phosphate Stock Solution (skip if already available)

Step 3.

Prepare two solutions:

- Monosodium dihydrogen phosphate (NaH₂PO₄) at 50mM (6g in 1L)
- Disodium hydrogen phosphate (Na₂HPO₄) at 50mM (3.55g in 500mL)

Make an equimolar mixture of this two solutions and adjust the pH at 7.5.

■ CONCENTRATION

0.05 Molarity (M) Additional info: NaH2PO4

■ CONCENTRATION

0.05 Molarity (M) Additional info: Na2HPO4

Preparation of Trace metals "Gaffron+Se" Stock Solution (skip if already available)

Step 4.

To 500mL of H₂0, add gradually the following nutrients

186 Boric acid (H₃BO₃) 150 101 Manganese (II) Sulfate Monohydrate (MnSO₄.H₂O) 30 1.98 Sodium Tungstate dihydrate (Na₂WO₄.2H₂O) 0.3 5.16 Ammonium molybdate tetrahydrate ((NH₄)₆MOγO₂₄.4H₂O) 1.45 7.14 Potassium bromide (KBr) 3 4.98 Potassium iodide (KI) 1.5 17.25 Zinc sulfate heptahydrate (ZnSO₄.7H₂O) 3 9.25 Cadium Nitrate (Cd(NO₃)₂.4H₂O) 1.5 8.76 Cobalt (II) Nitrate (Co(NO₃)₂.6H₂O) 1.5 7.5 Copper (II) Sulfate (CuSO₄.5H₂O) 1.5 7.1 Nickel Chloride (NiCl₂.6H₂O) 1.5 2.4 Chromium (III) Nitrate (Cr(NO₃)₃.9H₂O) 0.3 1.5 Vanadyl Sulfate Pentahydrate (VOSO₄.5H₂O) 0.3 28.4 Aluminium Potassium Sulfate (KAI(SO₄)₂.12H₂O) 3 3.3 Selenium (IV) Oxyde (SeO₂) 1.5	Quantity (mg/L)	Compound Final	Concentration in media (nM)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	186	Boric acid (H ₃ BO ₃)	150
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	101	Manganese (II) Sulfate Monohydrate (MnSO ₄ .H ₂ O) 30
7.14 Potassium bromide (KBr) 3 4.98 Potassium iodide (KI) 1.5 17.25 Zinc sulfate heptahydrate (ZnSO ₄ .7H ₂ O) 3 9.25 Cadium Nitrate (Cd(NO ₃) ₂ .4H ₂ O) 1.5 8.76 Cobalt (II) Nitrate (Co(NO ₃) ₂ .6H ₂ O) 1.5 7.5 Copper (II) Sulfate (CuSO ₄ .5H ₂ O) 1.5 7.1 Nickel Chloride (NiCl ₂ .6H ₂ O) 1.5 2.4 Chromium (III) Nitrate (Cr(NO ₃) ₃ .9H ₂ O) 0.3 1.5 Vanadyl Sulfate Pentahydrate (VOSO ₄ .5H ₂ O) 0.3 28.4 Aluminium Potassium Sulfate (KAI(SO ₄) ₂ .12H ₂ O) 3	1.98	Sodium Tungstate dihydrate (Na ₂ WO ₄ .2H ₂ 0)	0.3
4.98 Potassium iodide (KI) 1.5 17.25 Zinc sulfate heptahydrate (ZnSO ₄ .7H ₂ O) 3 9.25 Cadium Nitrate (Cd(NO ₃) ₂ .4H ₂ O) 1.5 8.76 Cobalt (II) Nitrate (Co(NO ₃) ₂ .6H ₂ O) 1.5 7.5 Copper (II) Sulfate (CuSO ₄ .5H ₂ O) 1.5 7.1 Nickel Chloride (NiCl ₂ .6H ₂ O) 1.5 2.4 Chromium (III) Nitrate (Cr(NO ₃) ₃ .9H ₂ O) 0.3 1.5 Vanadyl Sulfate Pentahydrate (VOSO ₄ .5H ₂ O) 0.3 28.4 Aluminium Potassium Sulfate (KAI(SO ₄) ₂ .12H ₂ O) 3	5.16	•	1.45
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	7.14	Potassium bromide (KBr)	3
9.25 Cadium Nitrate $(Cd(NO_3)_2.4H_2O)$ 1.5 8.76 Cobalt (II) Nitrate $(Co(NO_3)_2.6H_2O)$ 1.5 7.5 Copper (II) Sulfate $(CuSO_4.5H_2O)$ 1.5 7.1 Nickel Chloride $(NiCl_2.6H_2O)$ 1.5 2.4 Chromium (III) Nitrate $(Cr(NO_3)_3.9H_2O)$ 0.3 1.5 Vanadyl Sulfate Pentahydrate $(VOSO_4.5H_2O)$ 0.3 28.4 Aluminium Potassium Sulfate $(KAl(SO_4)_2.12H_2O)$ 3	4.98	Potassium iodide (KI)	1.5
8.76 Cobalt (II) Nitrate $(Co(NO_3)_2.6H_2O)$ 1.5 7.5 Copper (II) Sulfate $(CuSO_4.5H_2O)$ 1.5 7.1 Nickel Chloride $(NiCl_2.6H_2O)$ 1.5 2.4 Chromium (III) Nitrate $(Cr(NO_3)_3.9H_2O)$ 0.3 1.5 Vanadyl Sulfate Pentahydrate $(VOSO_4.5H_2O)$ 0.3 28.4 Aluminium Potassium Sulfate $(KAl(SO_4)_2.12H_2O)$ 3	17.25	Zinc sulfate heptahydrate (ZnSO ₄ .7H ₂ O)	3
7.5 Copper (II) Sulfate (CuSO ₄ .5H ₂ O) 1.5 7.1 Nickel Chloride (NiCl ₂ .6H ₂ O) 1.5 2.4 Chromium (III) Nitrate (Cr(NO ₃) ₃ .9H ₂ O) 0.3 1.5 Vanadyl Sulfate Pentahydrate (VOSO ₄ .5H ₂ O) 0.3 28.4 Aluminium Potassium Sulfate (KAI(SO ₄) ₂ .12H ₂ O) 3	9.25	Cadium Nitrate (Cd(NO ₃) ₂ .4H ₂ O)	1.5
7.1 Nickel Chloride (NiCl ₂ .6H ₂ O) 1.5 2.4 Chromium (III) Nitrate (Cr(NO ₃) ₃ .9H ₂ O) 0.3 1.5 Vanadyl Sulfate Pentahydrate (VOSO ₄ .5H ₂ O) 0.3 28.4 Aluminium Potassium Sulfate (KAl(SO ₄) ₂ .12H ₂ O) 3	8.76	Cobalt (II) Nitrate (Co(NO ₃) ₂ .6H ₂ O)	1.5
2.4 Chromium (III) Nitrate (Cr(NO ₃) ₃ .9H ₂ O) 0.3 1.5 Vanadyl Sulfate Pentahydrate (VOSO ₄ .5H ₂ O) 0.3 28.4 Aluminium Potassium Sulfate (KAl(SO ₄) ₂ .12H ₂ O) 3	7.5	Copper (II) Sulfate (CuSO ₄ .5H ₂ O)	1.5
1.5 Vanadyl Sulfate Pentahydrate (VOSO ₄ .5H ₂ O) 0.3 28.4 Aluminium Potassium Sulfate (KAl(SO ₄) ₂ .12H ₂ O) 3	7.1	Nickel Chloride (NiCl ₂ .6H ₂ O)	1.5
28.4 Aluminium Potassium Sulfate (KAl(SO ₄) ₂ .12H ₂ O) 3	2.4	Chromium (III) Nitrate (Cr(NO ₃) ₃ .9H ₂ O)	0.3
	1.5	Vanadyl Sulfate Pentahydrate (VOSO ₄ .5H ₂ O)	0.3
3.3 Selenium (IV) Oxyde (SeO ₂) 1.5	28.4	Aluminium Potassium Sulfate (KAl(SO ₄) ₂ .12H ₂ O)	3
	3.3	Selenium (IV) Oxyde (SeO ₂)	1.5

Complete the volume at 1L and store in refrigerator.

Preparation PCRS11-Red Sea medium

Step 5.

To 1L of H2O, add 33,33g of Red Sea Salt. Dissolve by shake (20min on agitator)

Preparation PCRS11-Red Sea medium

Step 6.

Heat seawater during 20min at 100°C.

▮ TEMPERATURE

100 °C Additional info:

Preparation PCRS11-Red Sea medium

Step 7.

Under the hood, to water, add these nutrients beforehand autoclaved (except vitamin):

Quantity	Compound	Final Concentration
1.0 mL	Hepes-NaOH 1M (pH 7.5) (See receipe below)	1mM
1.0 mL	Na ₂ -EDTA/FeCl ₃ (See recipe above)	8μΜ
1.0 mL	Sodium Phosphate (NaPO ₄) 50mM (pH 7,5) (See recipe above)	50 μΜ
1.0 mL	Ammonium Sulfate 400mM (NH4)2-SO4	400μΜ
0,1 mL	Trace metals "Gaffron+Se" (See recipe above)	
0.1 mL	Cyanocobalamin 10mg/L (Vit. B12)	1μg/L

Filter the medium on 0.2 microns.