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Working

UC Davis - Glucose Protocol [↗](#)

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[dx.doi.org/10.17504/protocols.io.yj9fur6](https://doi.org/10.17504/protocols.io.yj9fur6)

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ABSTRACT

Summary:

Glucose is oxidized by glucose oxidase to gluconic acid and hydrogen peroxide. The hydrogen peroxide reacts in the presence of peroxidase with HBA and 4-aminoantipyrine forming a red quinoneimine dye. The intensity of the color formed is proportional to the glucose concentration and can be measured photometrically between 460 and 560 nm.

EXTERNAL LINK

<https://mmpc.org/shared/document.aspx?id=88&docType=Protocol>

MATERIALS

NAME	CATALOG #	VENDOR	CAS NUMBER	RRID
Calibrator	TR1591-030	Fisher Diagnostics		
Reagents	TR15103	Fisher Diagnostics		
PBS				
Microplate				
Platereader				

MATERIALS TEXT

Reagent Preparation:

PBS – ready to use

Reagent – reconstitute with distilled water to make a 2X solution

- 1 Reconstitute powdered reagent with only 25 ml of distilled water to make a 2X solution.
- 2 Add 3 µl of calibrator and sample to each well.
- 3 Add 150 µl of PBS to each well. Read at 540 nm.

IMPORTANT: Make sure not to add any bubbles to the wells when dispensing reagents, this will interfere with reading in the platereader.

- 4 Add 150 µl of 2X reagent to each well. Incubate at 37°C for 10 minutes. Read at 540 nm.
- 5 Subtract blank readings from final readings. The assay will be linear so the unknown samples can be calculated as $(\text{Sample Absorbance} \div \text{Calibrator Absorbance}) \times \text{Calibrator Concentration}$.



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