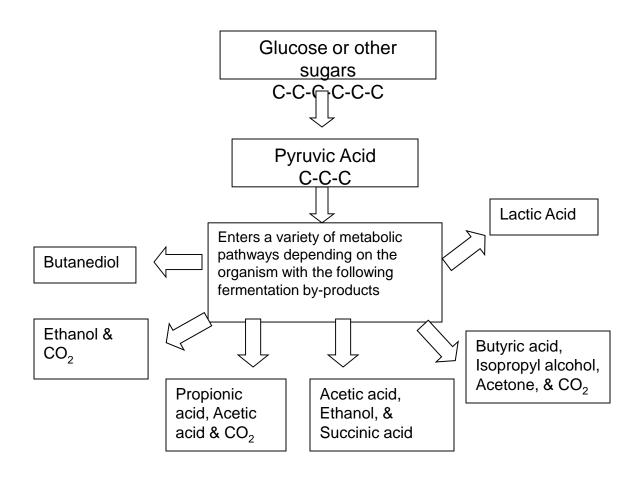
# **Fermentation Lab**

### Fermentation



### Media Used For Determination

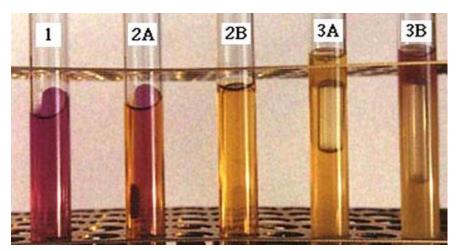
#### Fermentation Broth

- Fermentation of carbohydrates results in the abundant production of acidic end products, the presence of which can be detected by the pH indicator in the medium.
- Many organisms produce gas either CO<sub>2</sub> alone or a mixture of H<sub>2</sub> and CO<sub>2</sub>. H<sub>2</sub> is insoluble and is detected by bubble formation in a Durham tube placed in the medium.

### **Durham Tube**

- A Durham fermentation tube is a test tube that has a second inverted tube inside used to capture fermented gas byproducts of bacteria.
- Typically the tube has a pH indicator in it, like phenol red, that will change color when acid is present (in this case to yellow).
- Gases are trapped inside the Durham tube, indicating the production of gas.

### Possible Results



**Tube 1: No fermentation**. The pH indicator remains red. **There can still be growth** due to the use of amino acids as sources of energy (usually by respiration).

Tubes 2A and 2B: Fermentation with the production of acid (yellow color) but no gas. A slight amount of acid is seen in tube 2A, but fermentation is still recorded for this tube.

Tubes 3A and 3B: Fermentation with the production of acid (yellow color) and insoluble gas (bubble in Durham tube). Tube 3B shows an alkaline reaction on top; this is simply due to deamination of amino acids whose alkaline reaction has not been overneutralized by the acid diffusing through the tube from fermentation.

## Your Experiment

 You will use a loop to inoculate each of five different fermentation tubes with your unknown bacterium.

- Dextrose (D-glucose)
- Lactose
- Sucrose
- Maltose
- Mannose

## Cont'd

The tubes will incubate at 30 degrees Celsius.

- Tubes will be scored next week
  - Growth +/- (Turbidity?!)
  - Acid +/- (Yellow vs. red)
  - Gas +/- (Bubble in Durham tube)