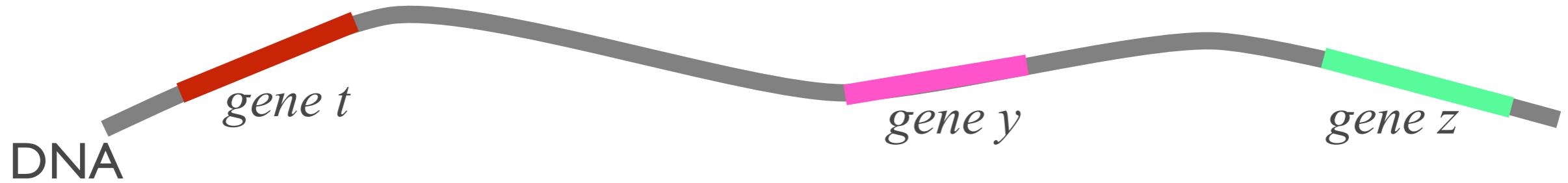


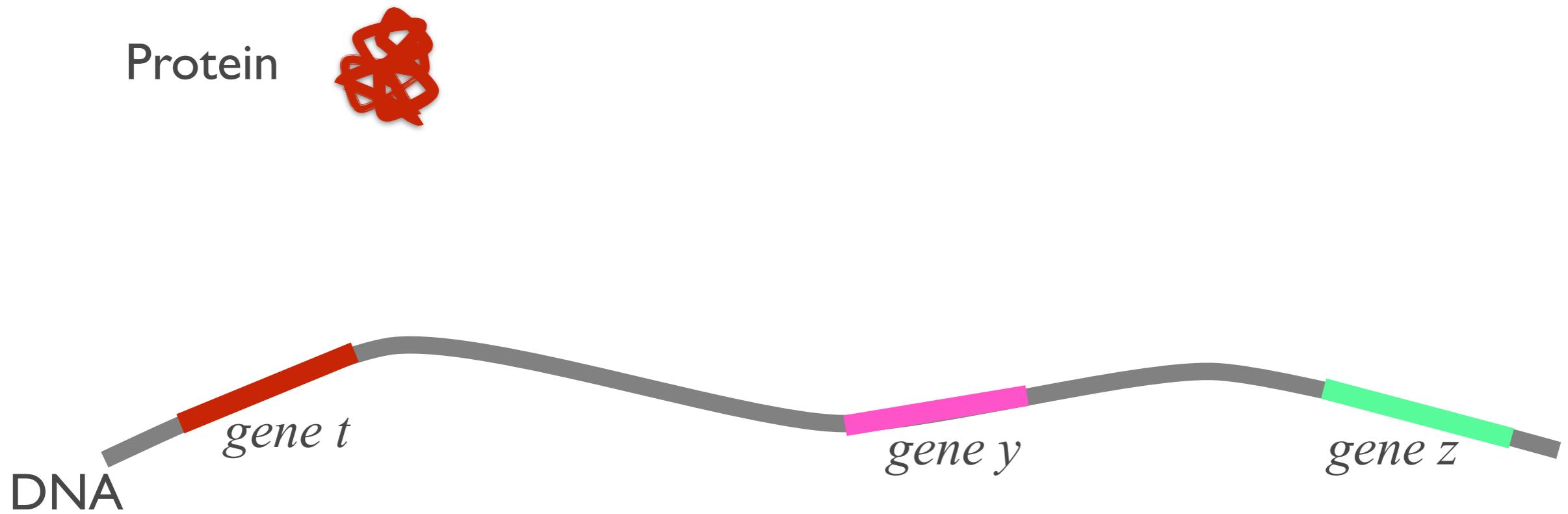
Basics on Metabolic network analysis

Jérémie Bourdon
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A simplistic illustration

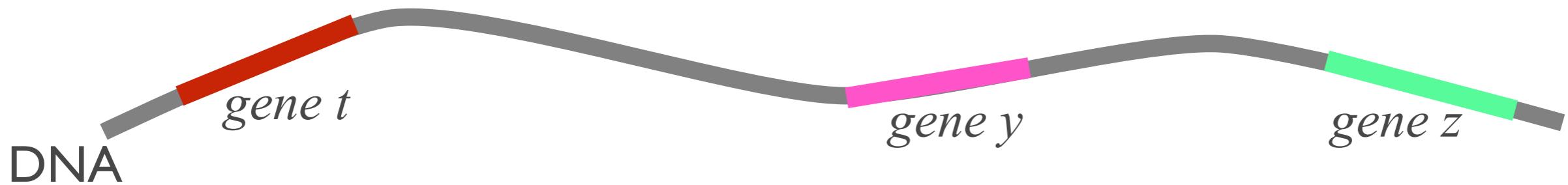
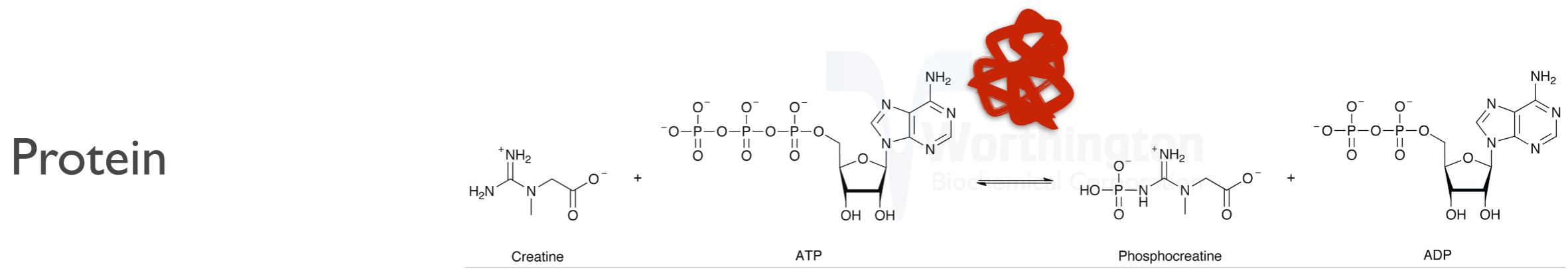


A simplistic illustration



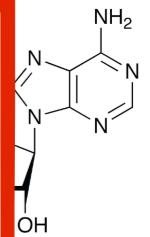
A simplistic illustration

Creatine kinase catabolizes the chemical reaction



A simplistic illustration

Creatine kinase catabolizes the chemical reaction



Major issue #3:

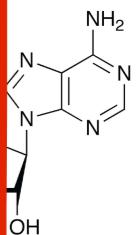
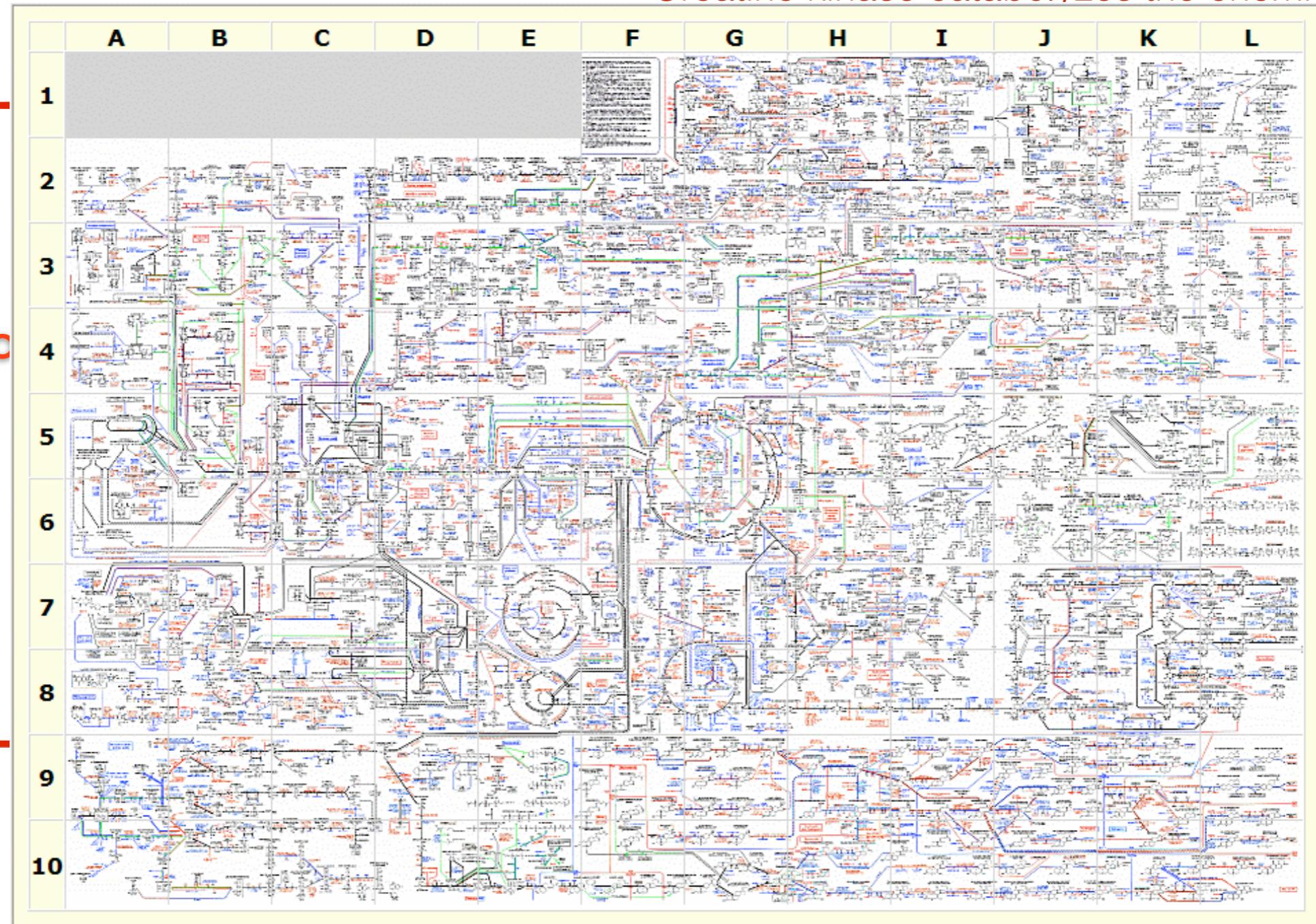
**How to understand/predict/enhance the biochemical functioning/
productions of a cell/tissue/organism ?**

Systems Biology : Metabolic network analysis

A simplistic illustration

Creatine kinase catabolizes the chemical reaction

Holding on to the
ring/



Mathematical modeling a very simple task !

Hypothesis 0: from now, we consider the metabolic network composed by (and only by) the possible biochemical reactions (the enzyme is present or it is not required).



.....

	R1	R2	R3
GLC	-1	0	-1
ATP	-1	-1	-1
G6P	+1	-1	-1
G3P	0	+2	0
lactose	0	0	+1

Stoichiometric matrix S

Mathematical modeling a very simple task !



.....

	R1	R2	R3
GLC	-1	0	-1
ATP	-1	-1	-1
G6P	+1	-1	-1
G3P	0	+2	0
lactose	0	0	+1

Stoichiometric matrix S

The evolution of G6P concentration can be deduced by the speeds of reactions (i.e., fluxes) R1, R2 and R3

$$d[\text{G6P}](t)/dt = +1 * V_{R1} - 1 * V_{R2} - 1 * V_{R3}$$

Hypothesis 1: The metabolites can't accumulate in the cell.

$$0 = d[\text{G6P}](t)/dt = +1 * V_{R1} - 1 * V_{R2} - 1 * V_{R3}$$

$$\text{i.e.,: } V_{R1} = V_{R2} + V_{R3}$$

Mathematical modeling a very simple task !



.....

	R1	R2	R3
GLC	-1	0	-1
ATP	-1	-1	-1
G6P	+1	-1	-1
G3P	0	+2	0
lactose	0	0	+1

Stoichiometric matrix S

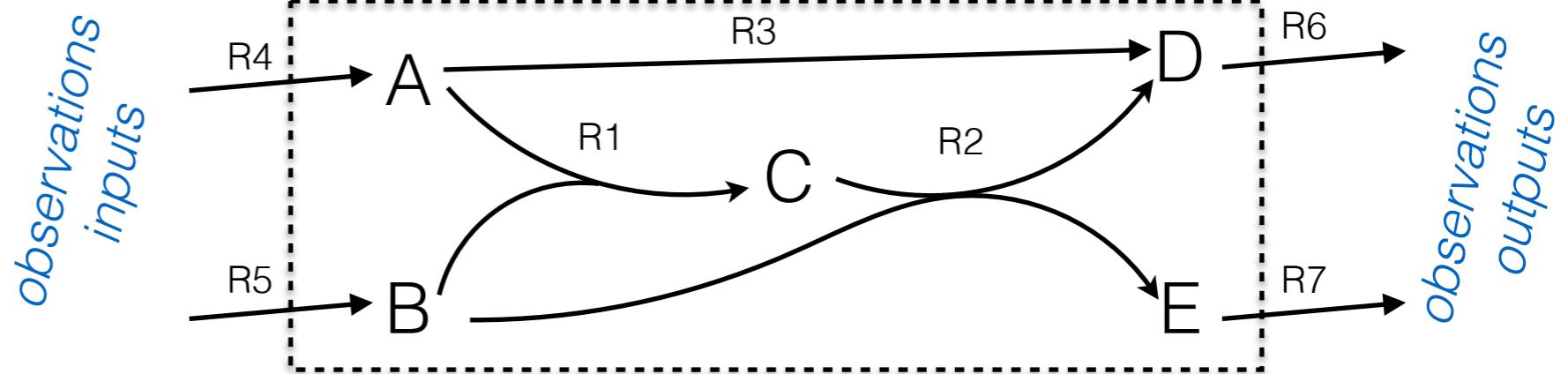
The evolution of metabolites depends on the fluxes of the reactions.
Finding appropriate fluxes \approx solving systems of linear equations !!!
 \rightarrow quite easy !

Hypothesis 1: The metabolites can't accumulate in the cell.

$$0 = d[\text{G6P}](t)/dt = +1 * V_{R1} - 1 * V_{R2} - 1 * V_{R3}$$
$$\text{i.e.,: } V_{R1} = V_{R2} + V_{R3}$$

Example

R1 : $2A + B \rightarrow C$
R2 : $B + C \rightarrow D + 2E$
R3 : $4A \rightarrow D$
R4 : $\rightarrow A$
R5 : $\rightarrow B$
R6 : $D \rightarrow$
R7 : $E \rightarrow$



Question 1 : Write the stoichiometric matrix

Question 2 : Write the linear system that has to be solved

Question 3 : Suppose that we observe $V_4 = 1$ and $V_7 = 1$ and solve the system

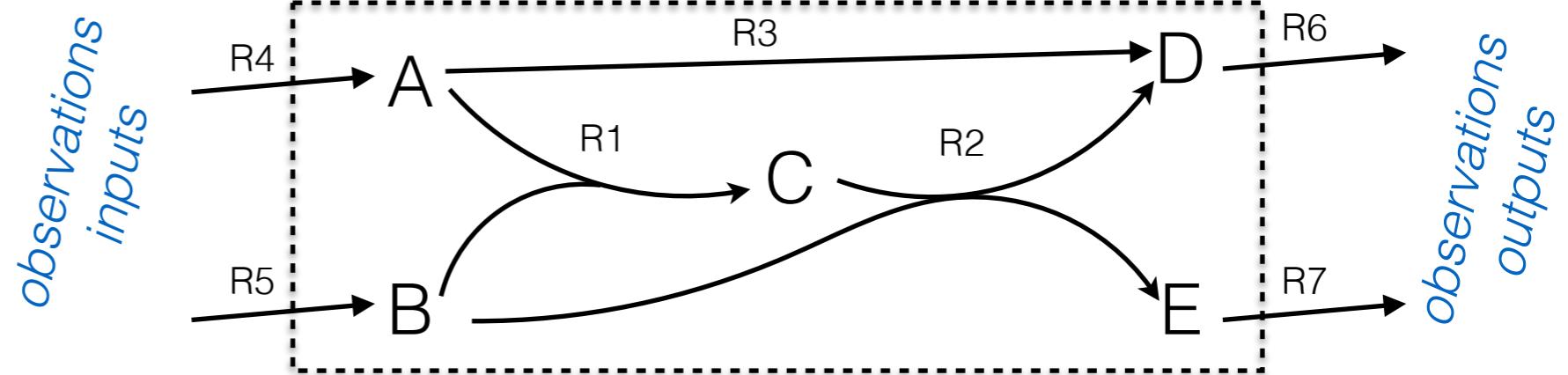
Question 4 : Suppose that we observe $V_4 = 1$ and $V_7 = 2$ and solve the system

Question 5 : All the reactions are irreversible. Plot the solution set. Suppose now that $V_4 > 1$ and $V_7 > 1$.

Question 6 : One has $V_4 < 6$ and $V_7 < 4$. What is the solution that maximizes $V_4 + V_7$?

Example

$R1 : 2A + B \rightarrow C$
 $R2 : B + C \rightarrow D + 2E$
 $R3 : 4A \rightarrow D$
 $R4 : \rightarrow A$
 $R5 : \rightarrow B$
 $R6 : D \rightarrow$
 $R7 : E \rightarrow$



Question 1 : Write the stoichiometric matrix

	R1	R2	R3	R4	R5	R6	R7
A	-2	0	-4	1	0	0	0
B	-1	-1	0	0	1	0	0
C	+1	-1	0	0	0	0	0
D	0	+1	1	0	0	-1	0
E	0	+2	0	0	0	0	-1

	R1	R2	R3	R4	R5	R6	R7
A	-2	0	-4	1	0	0	0
B	-1	-1	0	0	1	0	0
C	+1	-1	0	0	0	0	0
D	0	+1	1	0	0	-1	0
E	0	+2	0	0	0	0	-1

R6 : D \longrightarrow
 R7 : E \longrightarrow

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	R1	R2	R3	R4	R5	R6	R7
A	-2	0	-4	1	0	0	0
B	-1	-1	0	0	1	0	0
C	+1	-1	0	0	0	0	0
D	0	+1	1	0	0	-1	0
E	0	+2	0	0	0	0	-1

R6 : D → V ↗ B ↗ E ↗

observations
outputs

for A : $V_4 = 2 V_1 + 4 V_3$

for B : $V_5 = V_1 + V_2$

for C : $V_1 = V_2$

for D : $V_2 + V_3 = V_6$

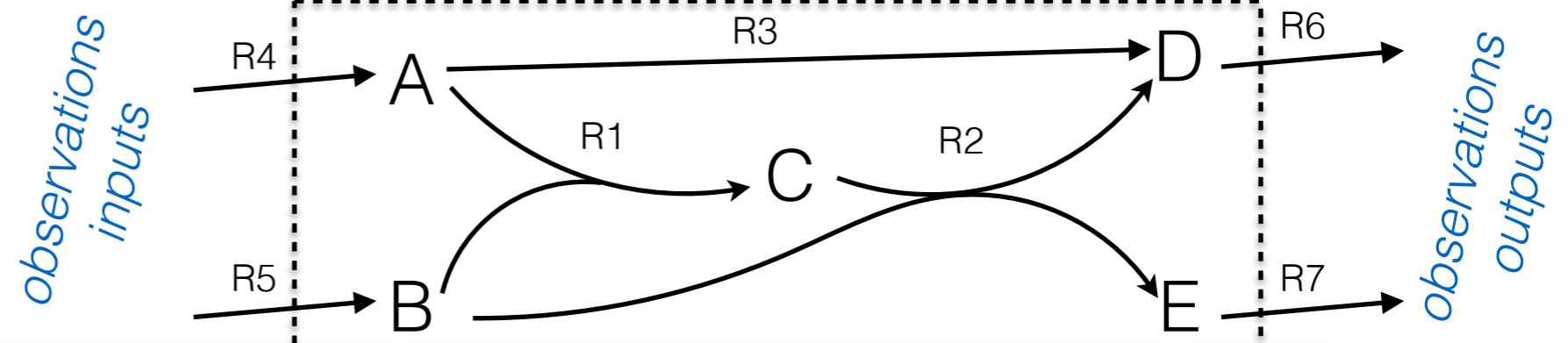
for E : $2 V_2 = V_7$

$V_1 = V_2, V_5 = V_7 = 2 V_2, V_6 = V_2 + V_3$ and $V_4 = 2 V_2 + 4 V_3$

only 2 unknowns but we must observe one of V_4 or V_6 and one of V_5 or V_7 .

Example

$R1 : 2A + B \rightarrow C$
 $R2 : B + C \rightarrow D + 2E$
 $R3 : 4A \rightarrow D$
 $R4 : \rightarrow A$
 $R5 : \rightarrow B$
 $R6 : D \rightarrow$



for A : $V4 = 2 V1 + 4 V3$
 for B : $V5 = V1 + V2$
 for C : $V1 = V2$
 for D : $V2 + V3 = V6$
 for E : $2 V2 = V7$

$V1 = V2, V5 = V7 = 2 V2, V6 = V2 + V3$ and $V4 = 2 V2 + 4 V3$

only 2 unknowns but we must observe one of $V4$ or $V6$ and one of $V5$ or $V7$.

for A : $V_4 = 2V_1 + 4V_3$
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for D : $V_2 + V_3 = V_6$
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Question 1 : Write the stoichiometric matrix

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$V_1 = V_2 = 0.5, V_3 = 0, V_4 = 1, V_5 = 1, V_6 = 0.5, V_7 = 1$

for A : $V_4 = 2V_1 + 4V_3$
for B : $V_5 = V_1 + V_2$
for C : $V_1 = V_2$
for D : $V_2 + V_3 = V_6$
for E : $2V_2 = V_7$

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Question 6 : One has $V_4 < 6$ and $V_7 < 4$. What is the solution that maximizes $V_4 + V_7$?

$$V_1 = V_2 = 0.5, V_3 = 0, V_4 = 1, V_5 = 1, V_6 = 0.5, V_7 = 1$$

$$V_1 = V_2 = 1, V_3 = -0.25, V_4 = 1, V_5 = 2, V_6 = 0.75, V_7 = 1$$

The flux of an irreversible reaction is necessarily positive !!

$$V_1 = V_2, V_5 = V_7 = 2 V_2, V_6 = V_2 + V_3 \text{ and } V_4 = 2 V_2 + 4 V_3$$

↓ expresses by means of V_4 and V_7

$$V_1 = V_2 = V_7/2, V_3 = (V_4 - V_7)/4, V_5 = V_7 \text{ and } V_6 = (V_4 + V_7)/4$$

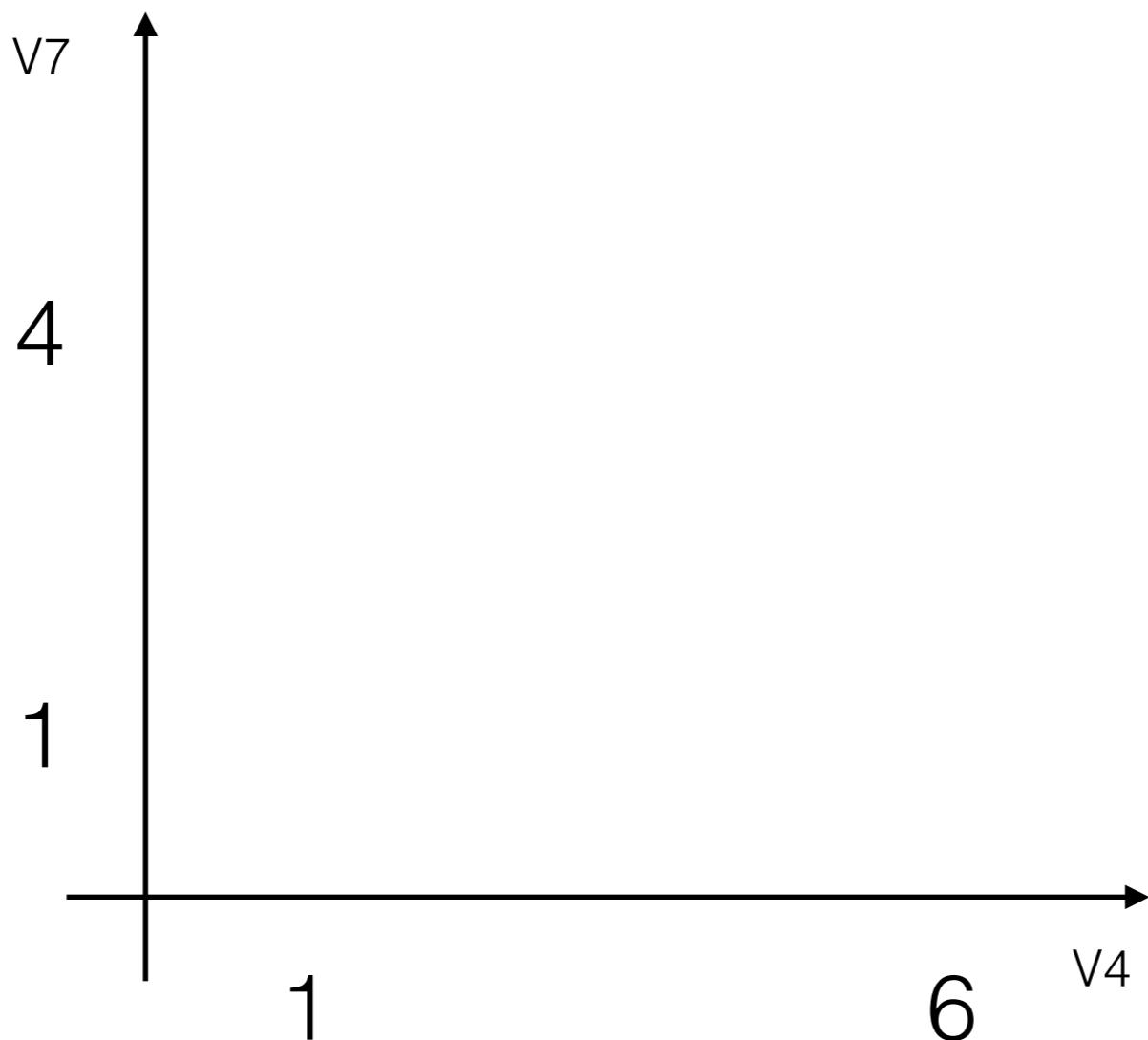
$$1 < V_4 < 6, 1 < V_7 < 4 \text{ and } V_1, \dots, V_6 > 0$$

$V_1 = V_2, V_5 = V_7 = 2V_2, V_6 = V_2 + V_3$ and $V_4 = 2V_2 + 4V_3$

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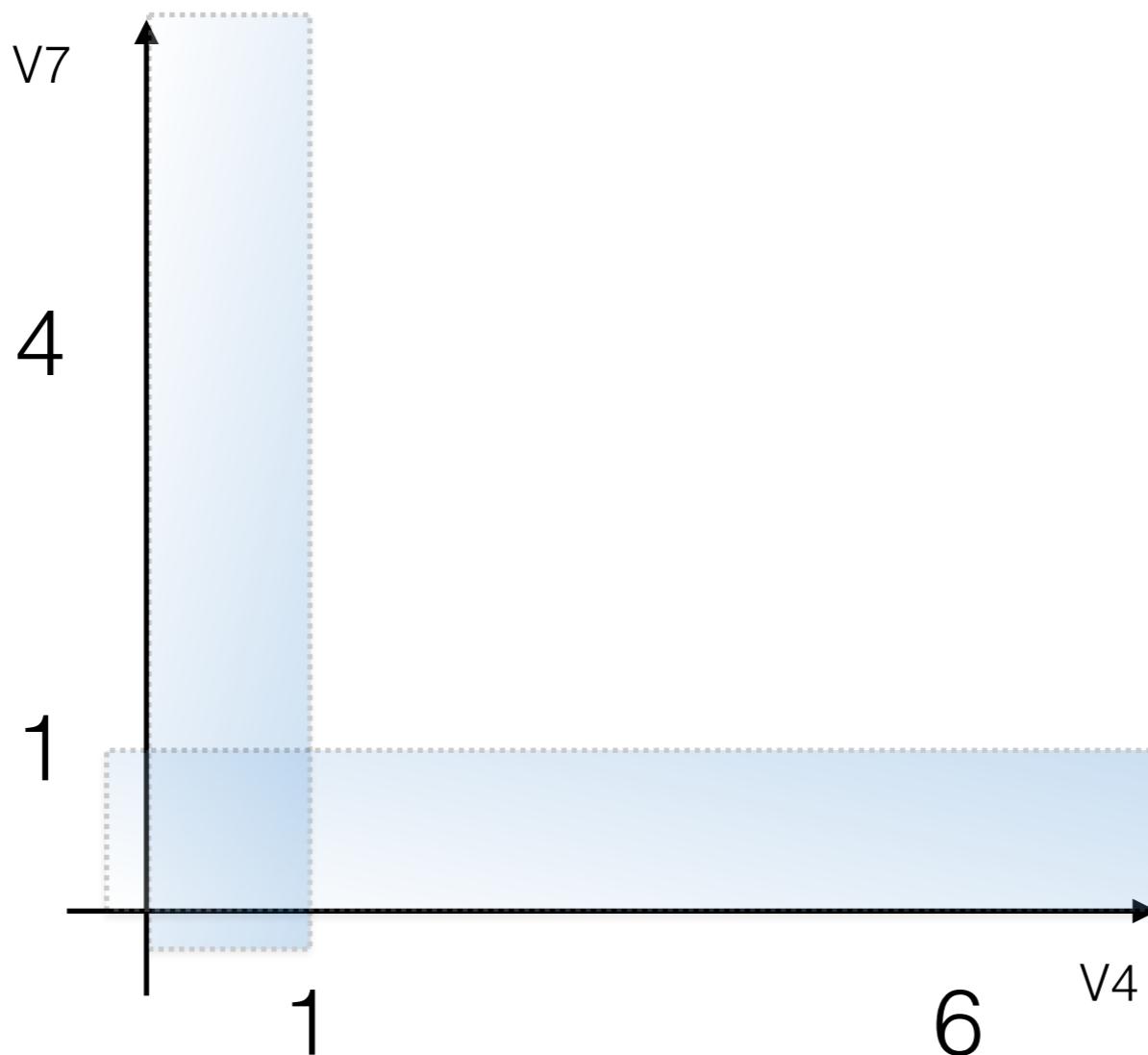


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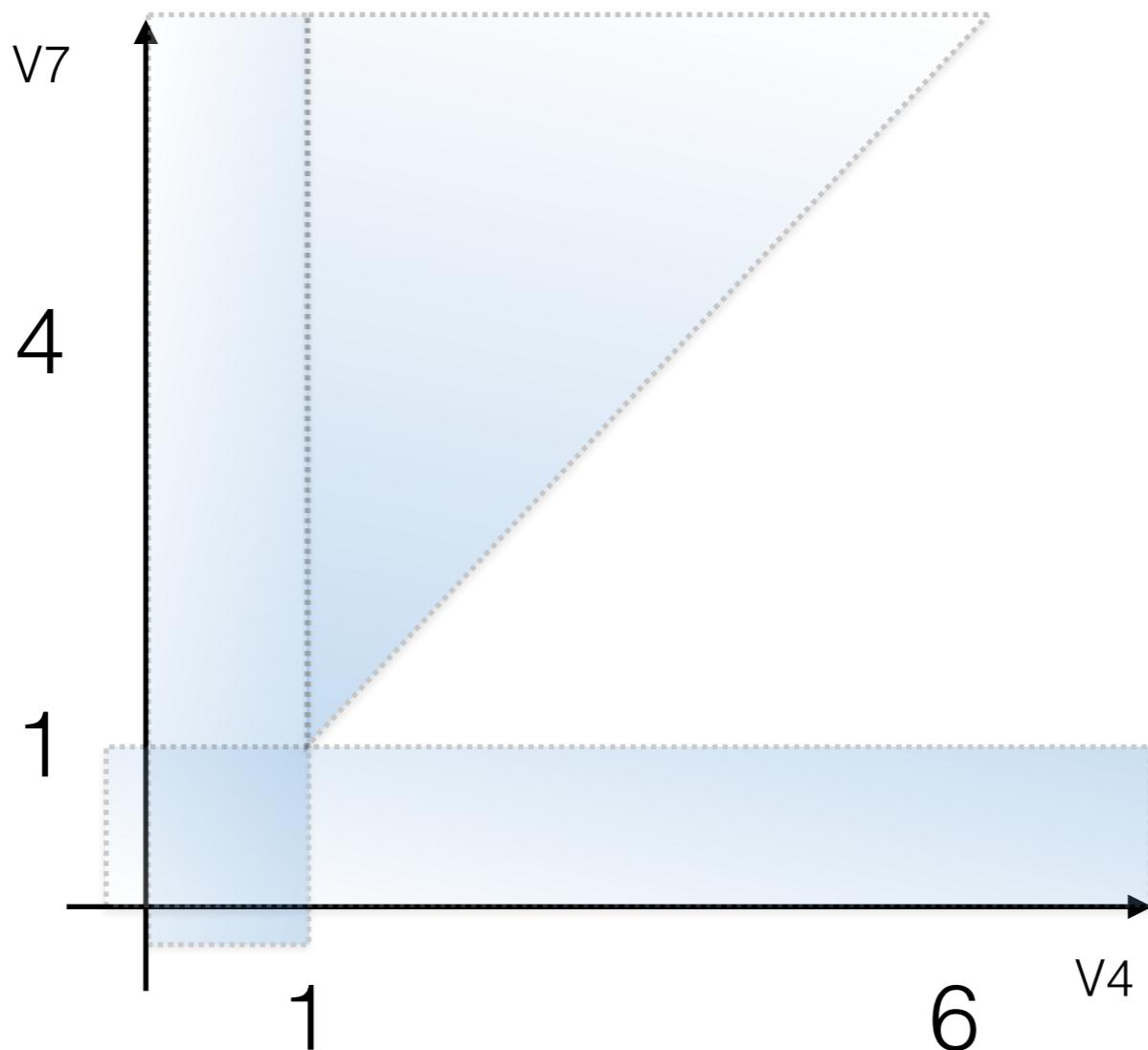


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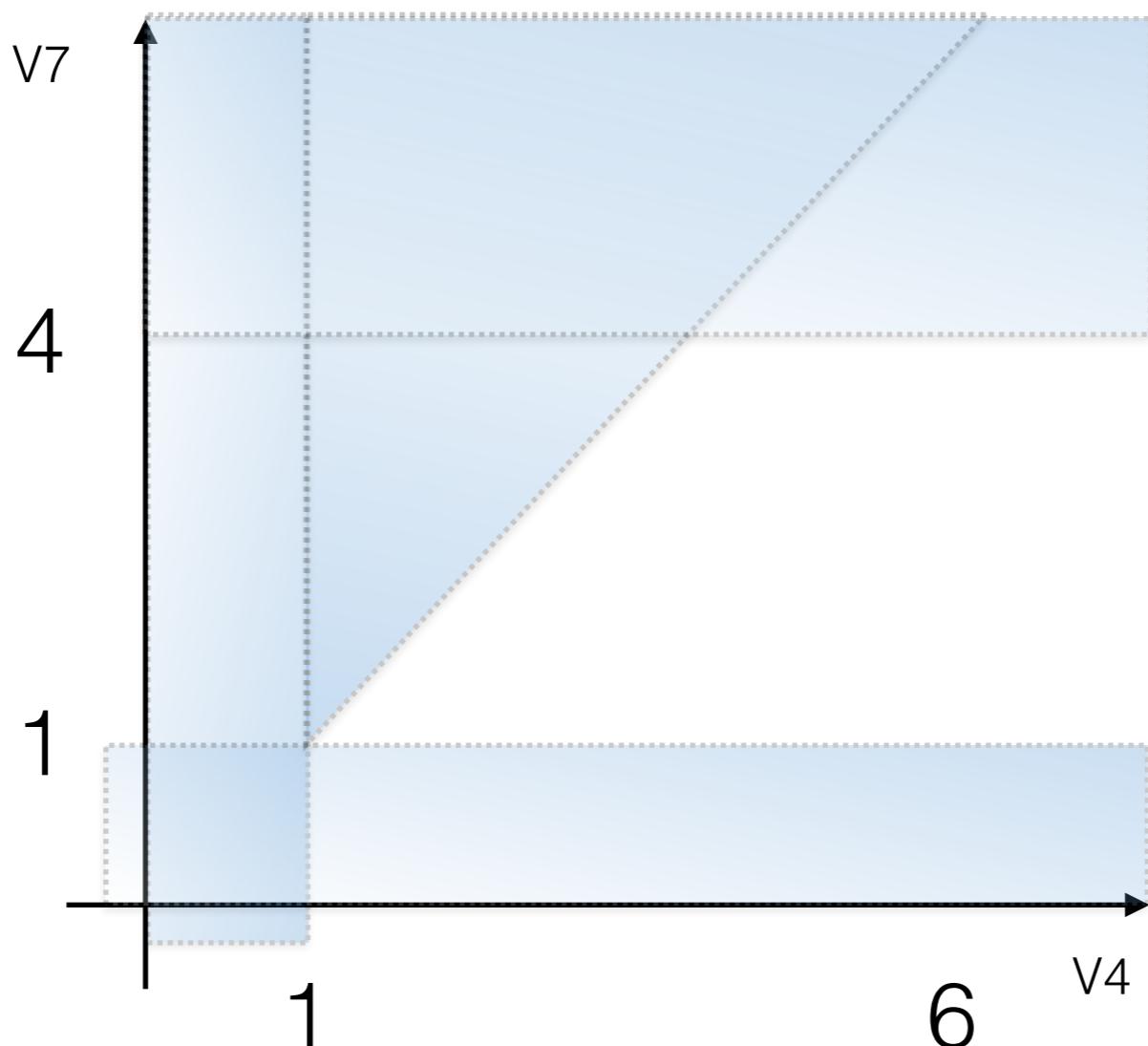


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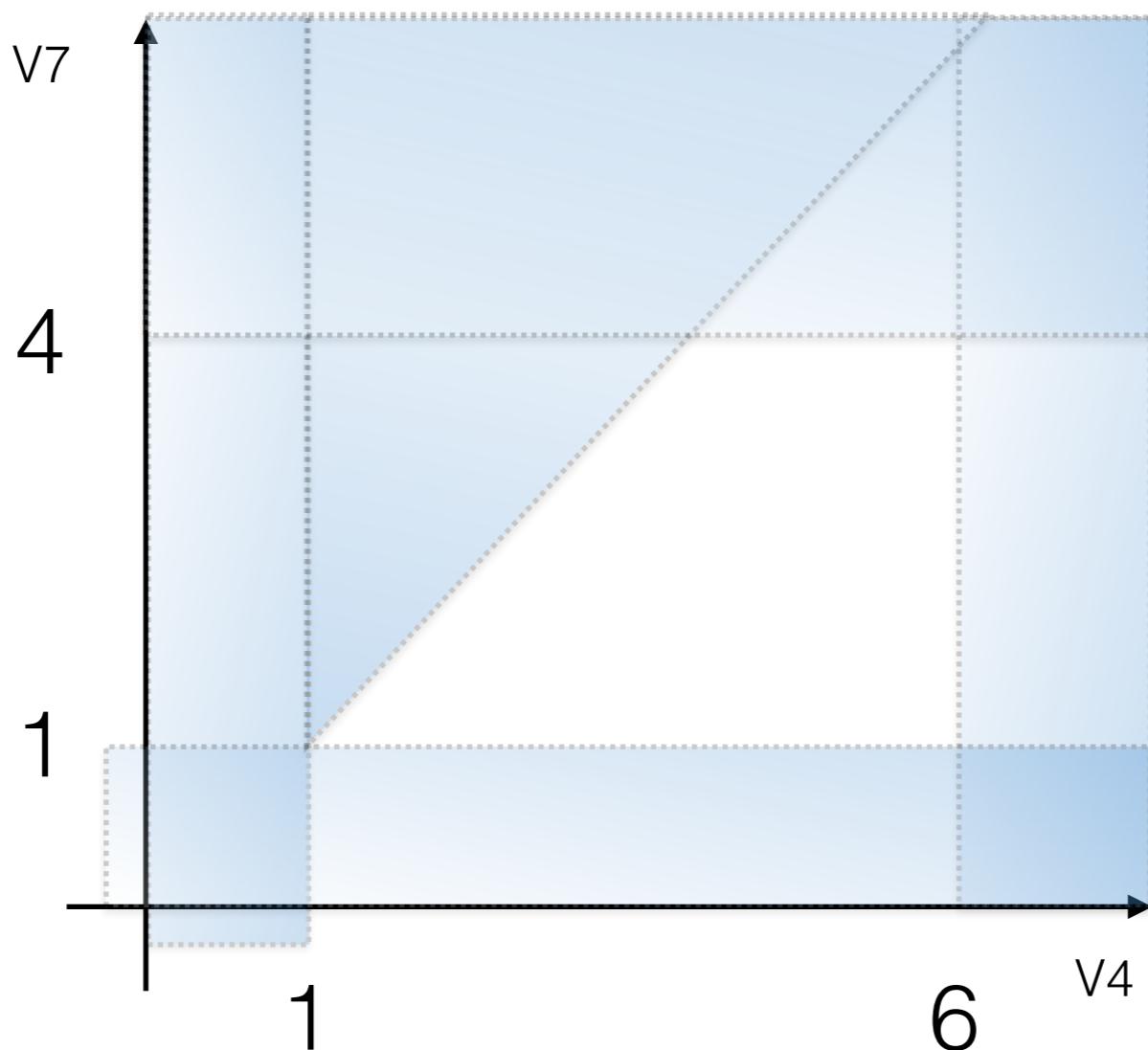


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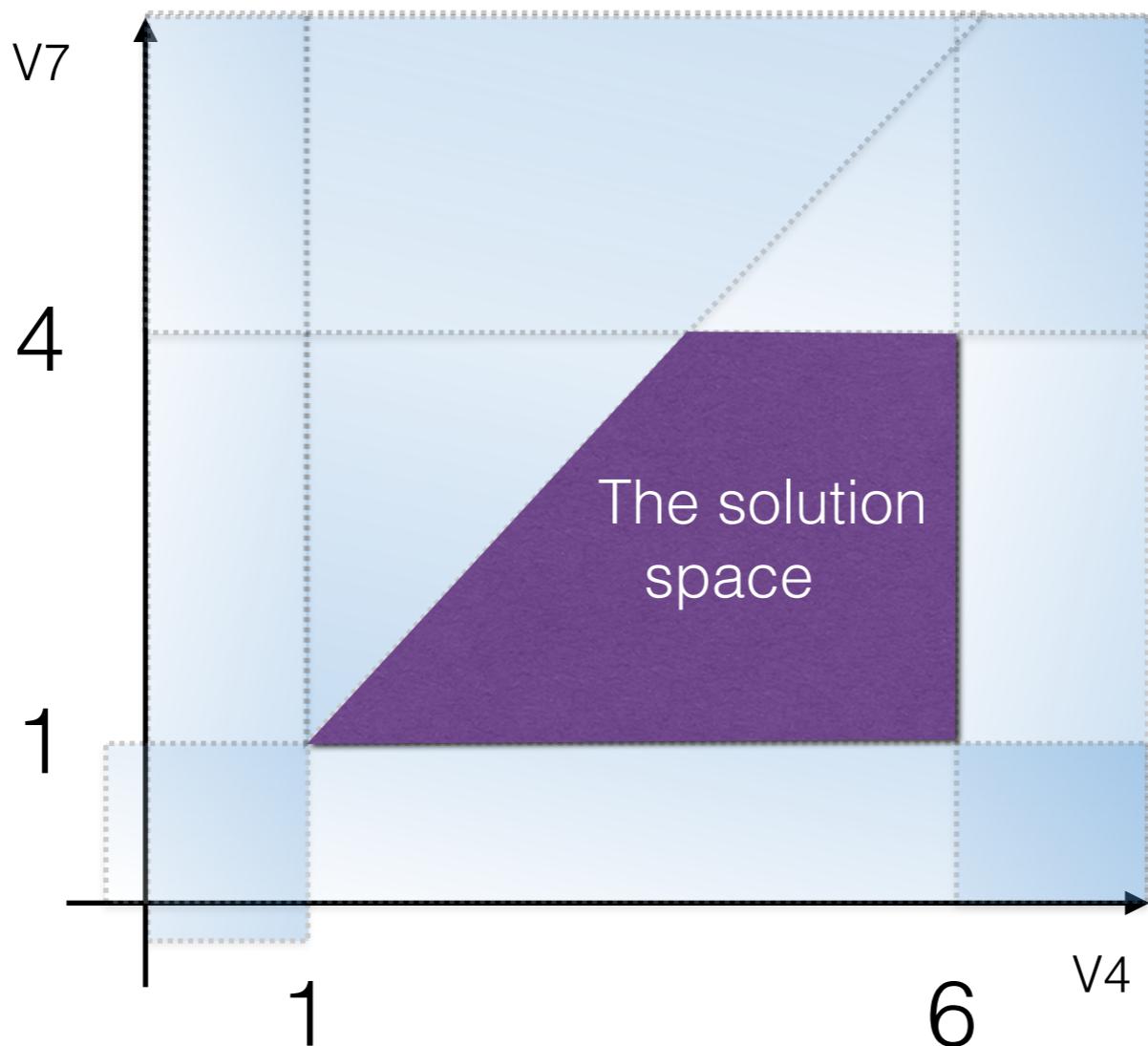


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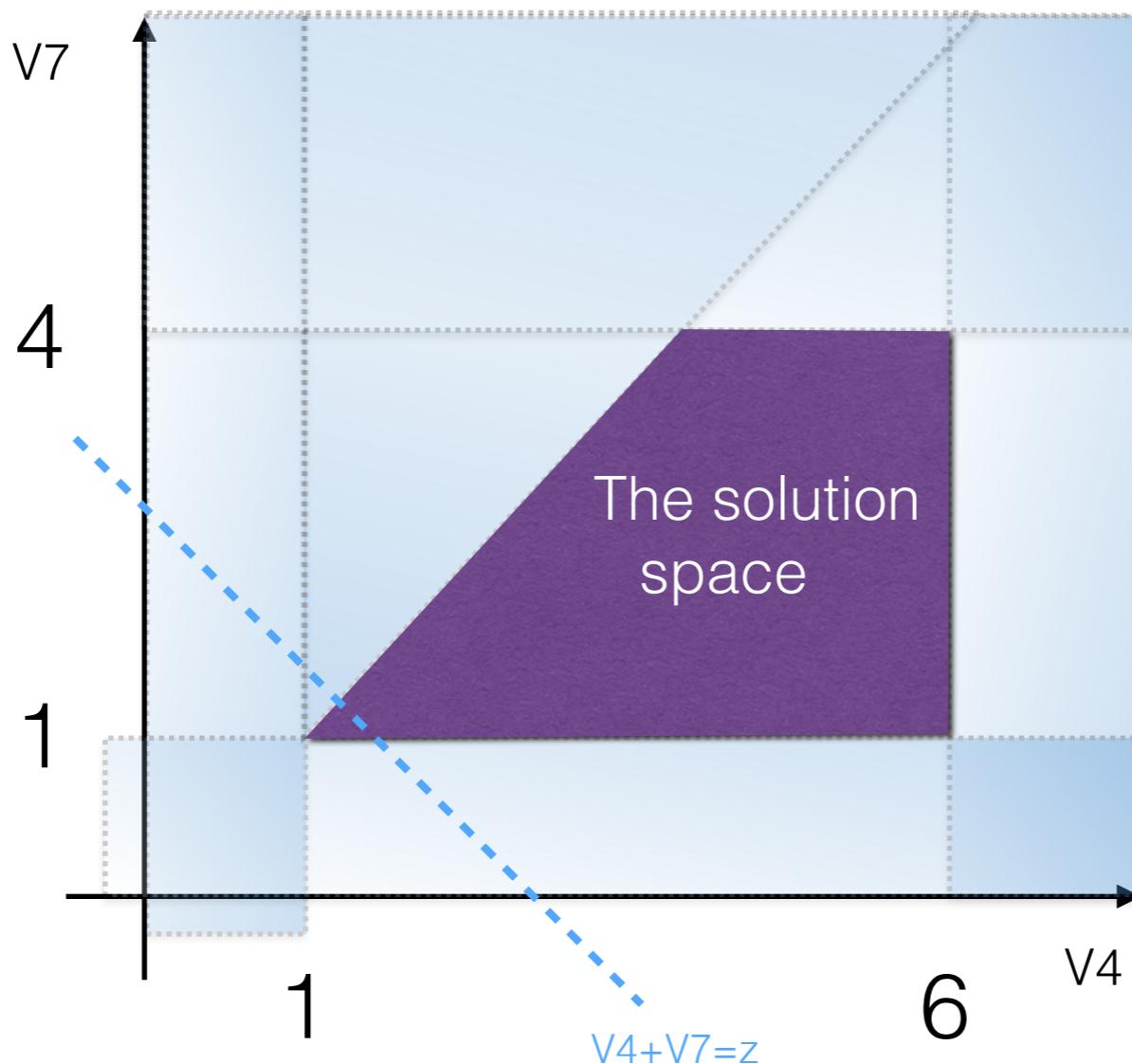


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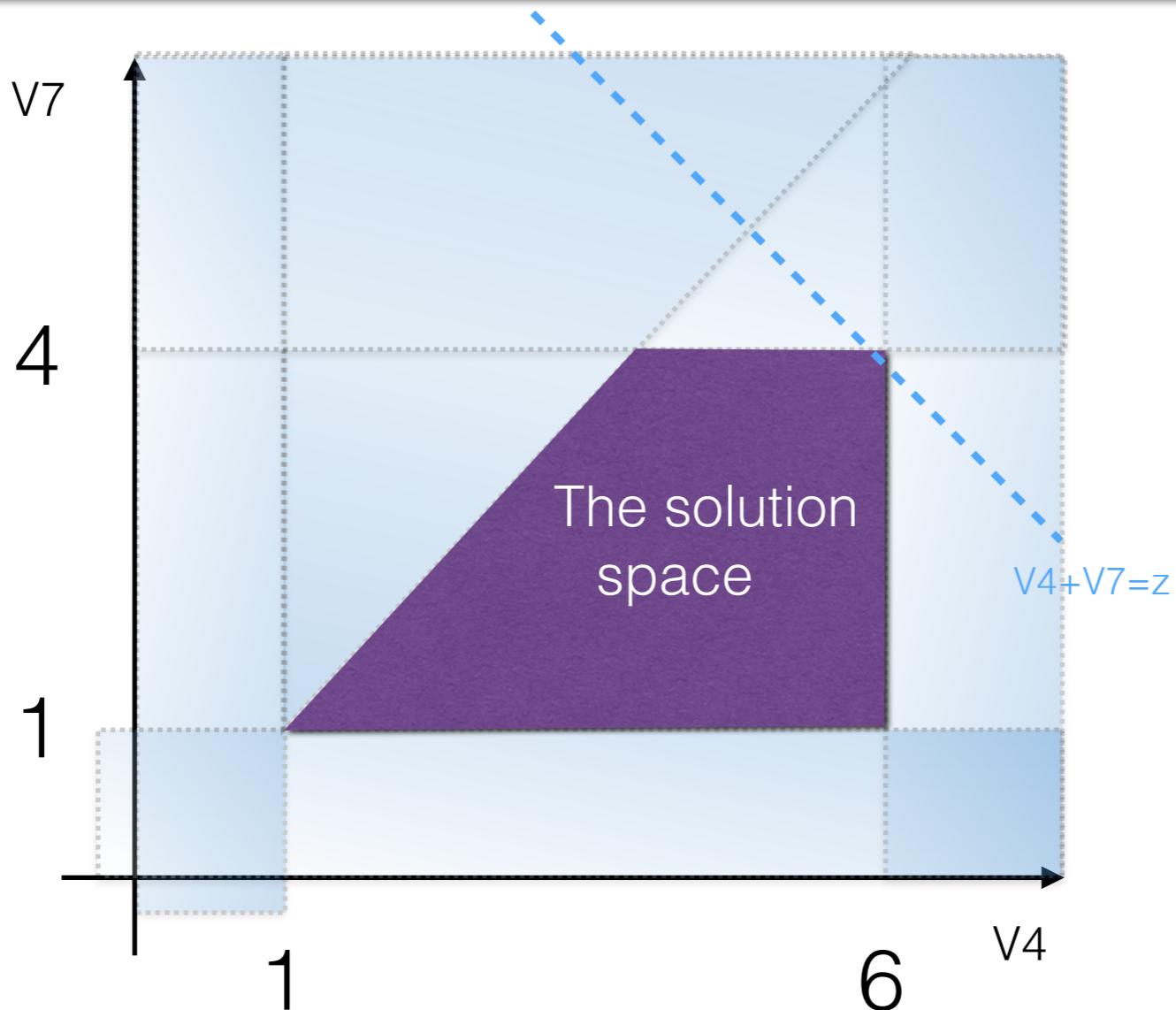


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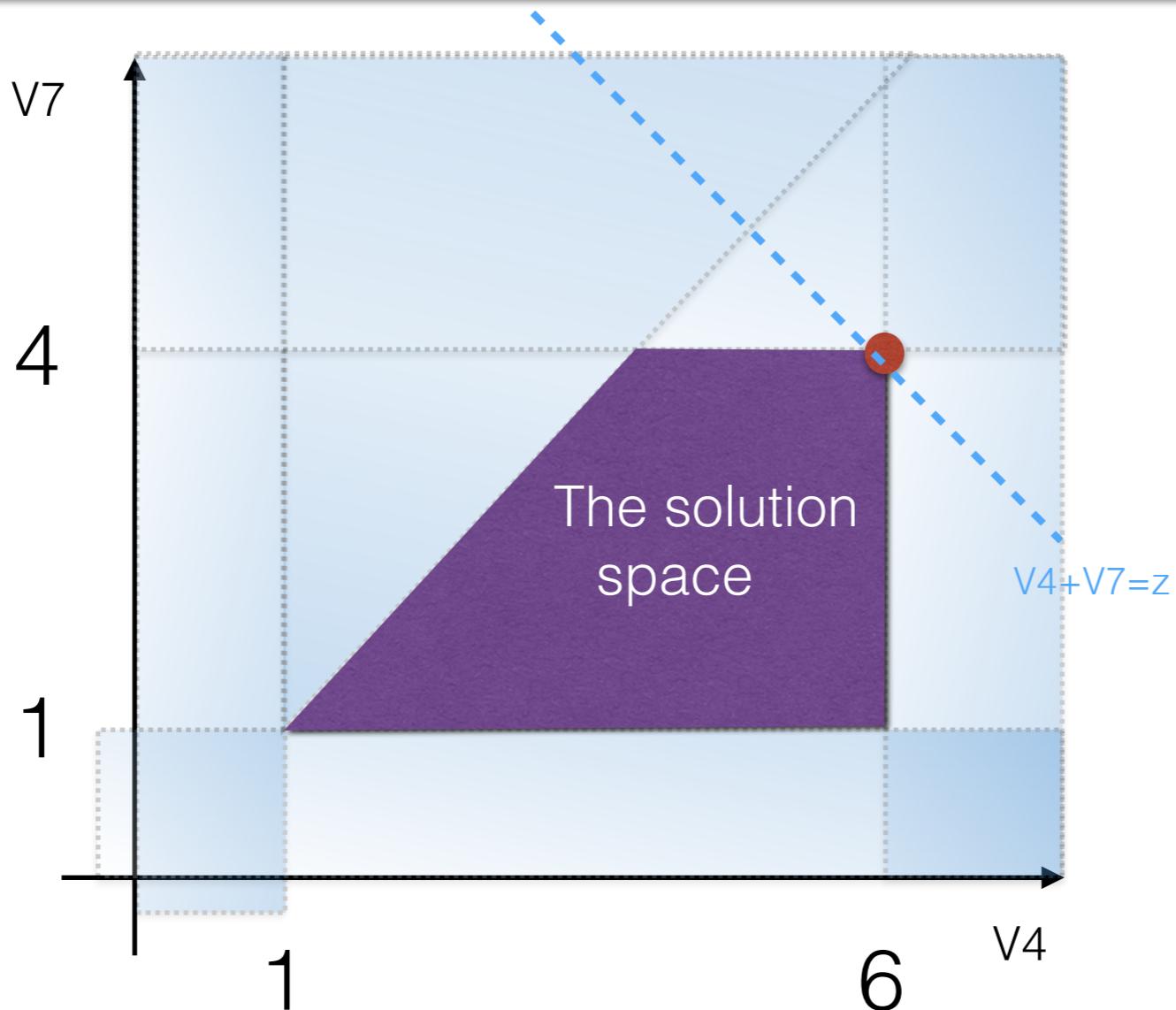


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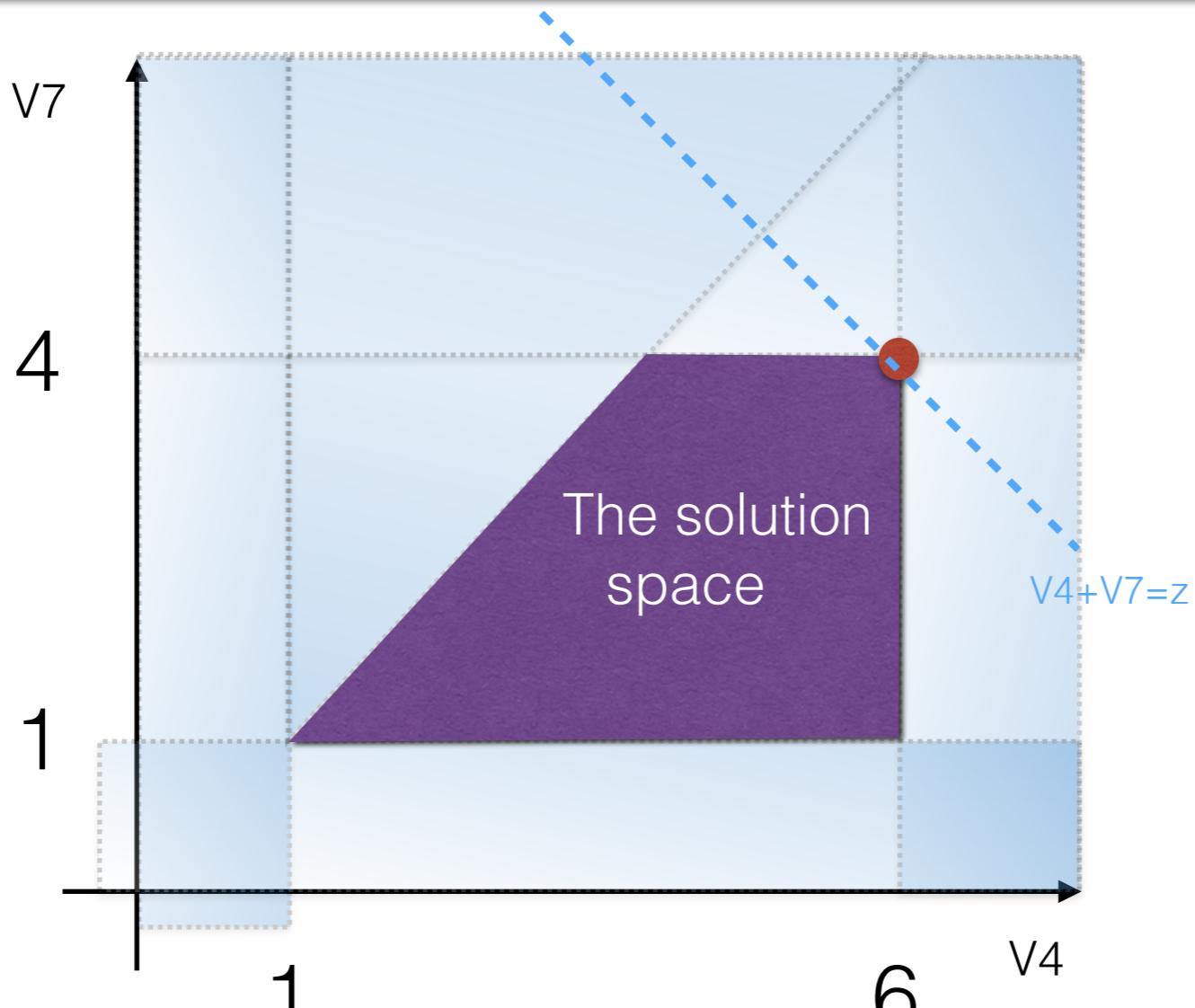


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$$1 < V_4 < 6, 1 < V_7 < 4 \text{ and } V_1, \dots, V_6 > 0$$



z is maximal when

$$V_1 = V_2 = 2, V_3 = 0.5, V_4 = 4, V_5 = 4, V_6 = 2.5 \text{ and } V_7 = 6$$

$$V_1 = V_2, V_5 = V_7 = 2V_2, V_6 = V_2 + V_3 \text{ and } V_4 = 2V_2 + 4V_3$$

↓ expresses by means of V_4 and V_7

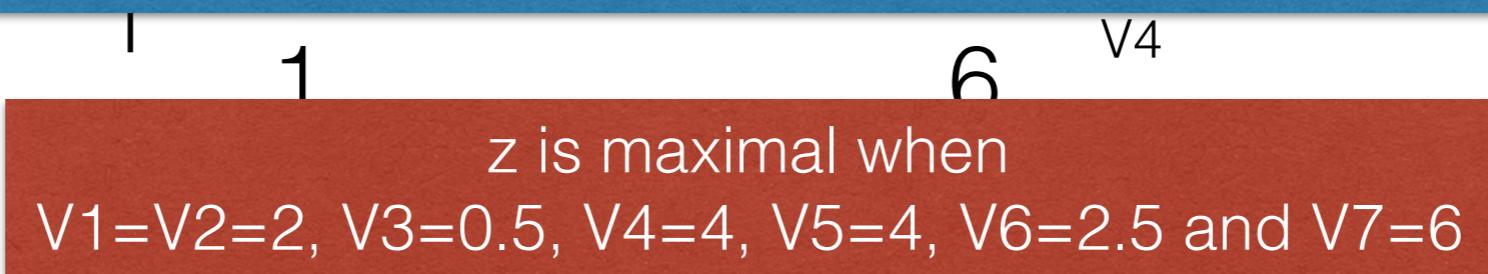
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$$1 < V_4 < 6, 1 < V_7 < 4 \text{ and } V_1, \dots, V_6 > 0$$



In higher dimensions,

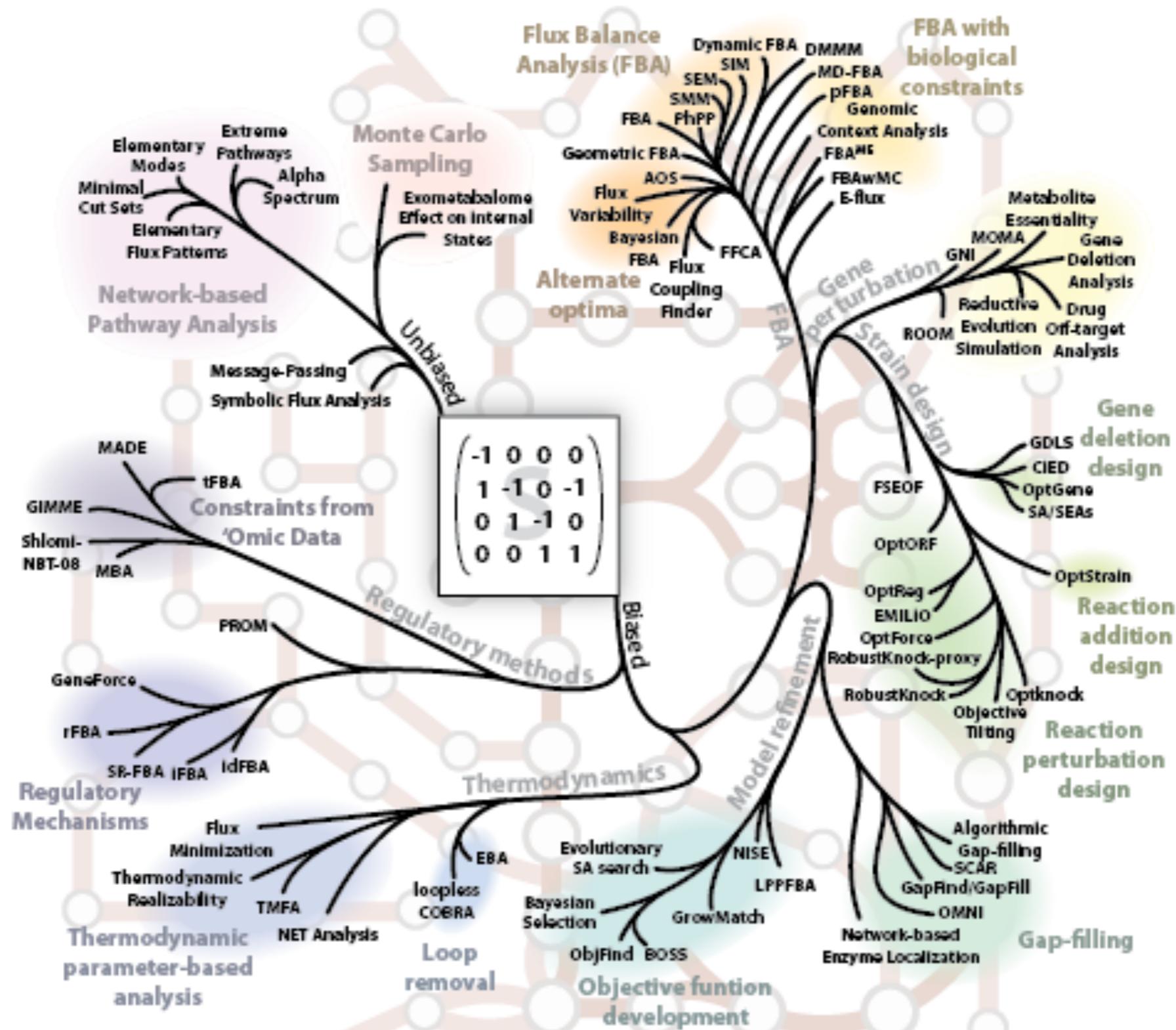
1. The figure defined by inequalities on linear expressions is always convex (it is a convex polyhedra also known as a simplex)
2. The optimal value of a linear expression is always on a vertex of the simplex (or on a face for very particular objectives)
3. Very efficient algorithms (such as the simplex algorithm) can find these optima



Why do we need a computer ?

- Current human metabolism has
 - 1,789 enzyme-encoding genes,
 - 7,440 reactions
 - 2,626 unique metabolites
- Study of the flux variations => $7440 * 2 * \text{FBA}$
- Study of single gene knock out => $1789 * \text{FBA}$
- Study of double gene knock out => $1789^2 * \text{FBA}$
- Many other kind of analysis => A. Bockmayr talk !

Many tools



Many tools

The screenshot shows the COBRApy software interface. At the top, there is a navigation bar with links to "Flux Balance Analysis (FBA)", "Dynamic FBA", "DMMM", "MD-FBA", and "FBA with biological constraints". Below the navigation bar, there is a sidebar with several icons and their corresponding names: "Min Cut", "Gene Deletion Analysis", "Drug Off-target Analysis", "GIMME", "Gene", "SOLs deletion", "IED design", "OptGene", "A/SEAs", "OptStrain", "Reaction addition design", "knock", "Reaction perturbation design", "gap-filling", "The para", and "development". In the center of the interface, there is a large image of a cobra with its hood spread, and the text "COBRApy" and "Constraints-based modeling of biological networks". Below this, there are three main sections: "Installation", "Documentation", and "Help". At the bottom of the interface, there are logos for "Utah State University", "UC San Diego", and "systems biology research group".

Flux Balance Analysis (FBA)

Dynamic FBA

DMMM

MD-FBA

FBA with biological constraints

Min Cut

Gene Deletion Analysis

Drug Off-target Analysis

GIMME

Gene

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Constraints-based modeling of biological networks

Installation

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Help

Utah State University

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systems biology research group