

i) cont) cylinder. Also calculate the probability

P(Z) depsity P(z), defined such that

P(z) dz is the probability that the particle

lies between z and z+dz (z=0 being

the bottom of the cylinder). Does # P(z)

vanish or diverge for any value of z? If

so, why? (20pts)

2) Calculate S2(E), T(E), E(T), and (1+) for

**Single one dimensional harmonic ascillator:

a mass manua free to move in one divection only while connected to a spring of spring constant k. Also calculate P(x), the propability density defined so that

P(x)dx is the probability that the mass is

Dani) Le theren X and x+dx. Does Phaldwerge Détine x=0 as the equilibrium prosition of the spring - the point of which it exerts no Force). Does P(x) diverge for any x? If so, why? Where is P(x) of Minimum? Again, why? (20pts)

The same calculation for a particle of mass m moving in the gravitation (paint)
field of a much larger (mass M. Calculate, for 3 dimensional motion, the phase space volume S2(E), ECPSE T(E), E(T), and C(T). Also calculate the protostility (P(r) that the partiete TiPS Fretween va all anywhe

3) cont) P(r), defined as the pro

so that Plv) dv is the probability

That the particle of mass m lies

anywhere in the spherical shell between

v and v tdv centered on the big mass M.

Does S2(E) diverge at any E?

If so, why?

Does P(r) diverge or vanish at any
r? Again, why?

Define the gravitational potential energy

to be zero when the two masses are

infinitely far apart. Note that the

total energy may be negative in this case.

4) Consider a gas with known equation
of state P(V,T) contined to a cell
of area A
by a movable spring held piston
held is place by a spring of spring
constant h:

[&h]

Movable

Piston

P(V,T)

Z

Assume the spring is tensionless when z=0 (i.e., who the piston sits at the bottom of the cell).

Calculate the specific heat at the constant this situation (the "specific heat at constant spring constant", to coin an ungainly moniher)

in terms of the heat capacity at constant volume Cy and ther modynamic derivatives

that can be obtained from the equation of state.

(Hint: Use grometry and Hook's law to relate pressure and volume execute to Z;

Then eliminate z to obtain a relation between P, V, T, and the spring constant k.)

(20 pts)

Prove, without using Jacobians, that $C_{p} = C_{v} - \frac{T(\frac{\partial v}{\partial T})_{p}^{2}}{(\frac{\partial v}{\partial Y})_{p}}.$

Vse the fact that Palways

decreases upon isothermal expansion to

show that (p > (v) always. (10pts)

(a) Consider a system obeying the ideal gas equation of state with Wax

6) (ont) constant heat capacity at constant volume (v= XW kB), where N is the number of particles and 8 a dimensionless constant. Derive the relation between temperature and volume during an adiabatic process in terms of 8. Verify that your result reduces to that derived in class for a monoatomic glass ideal gas (Hint: what is 8 for such a gas?) 20 pts