Kolbe's Electrolysis

1 Reaction and Mechanism

Kolbe's Electrolysis

$$R \longrightarrow C \longrightarrow O^{-}Na^{+}$$

$$+ H_{2}O$$

KingDra

2 Reaction Observations

- i. C^{\bullet} obtained as intermediate.
- ii. pH of medium increases due to formation of NaOH.
- iii. CO_2 obtained at anode and H_2 at cathode.
- iv. CH_4 can never be obtained.