Stochastic thermodynamics for the study of life



Living Systems Institute (Exeter)

Jesús Rubio

Department of Physics & Astronomy





LSI seminar 2021, 27th January



University of Bristol

An interdisciplinary approach to the physics of life







Jesús Rubio Janet Anders (quantum & stochastic thermo. + quantum info. & estimation)

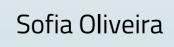


Frank Vollmer Littlechild

(nano & quantum biosensing + biochemistry)









Adrian Mulholland

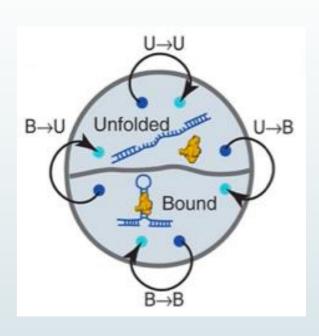
(molecular dynamics simulations in chemistry)

- Interdisciplinary team with more than 20 scientists
- https://www.exeter.ac.uk/livingsystems/research/mmi/

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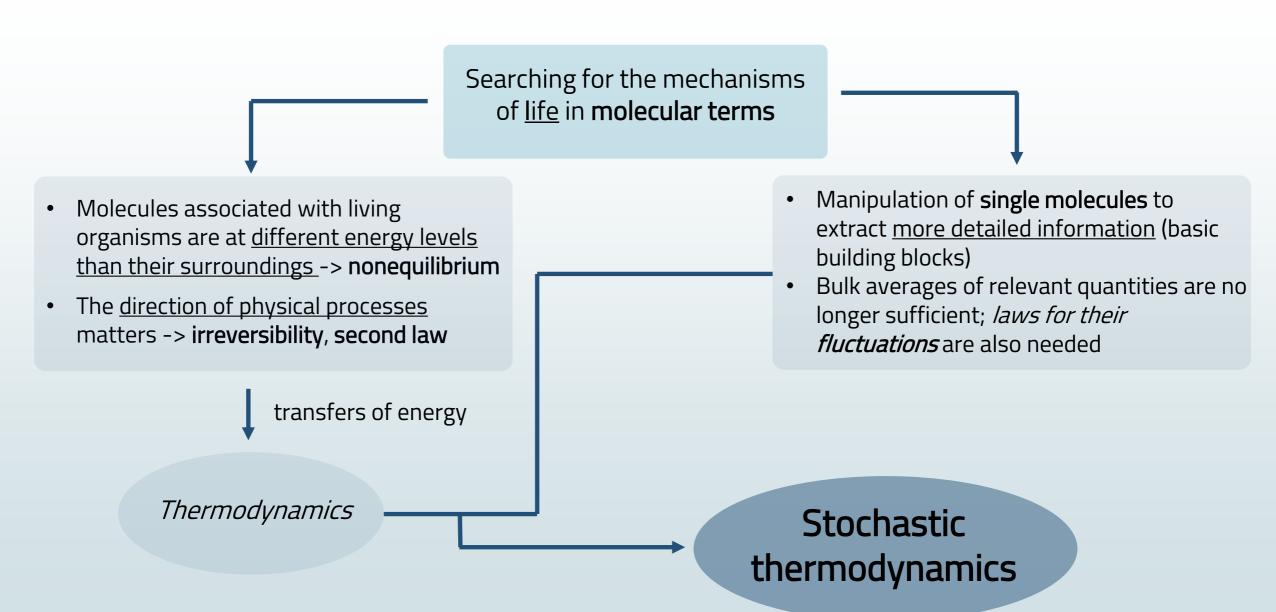


- I. Nonequilibrium physics and life
- II. Stochastic thermodynamics and the second law
 - a. A finite-information approach to the estimation of free energy differences
- III. Three case studies:
 - a. Single-molecule folding and unfolding
 - b. <u>Instantaneous frequency shift in a harmonic oscillator</u>
 - c. Forwards and backwards redox reactions



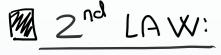
Life and the thermodynamics of small systems

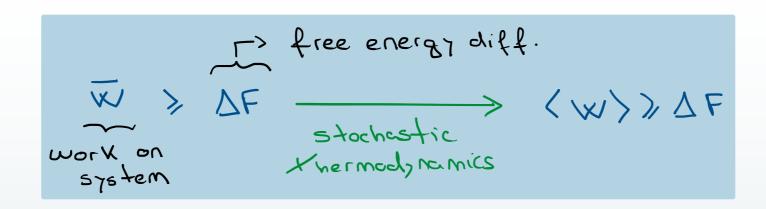


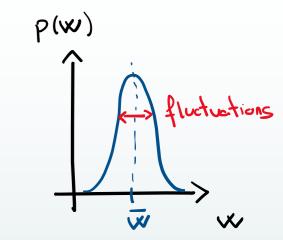


Stochastic thermodynamics and the second law











MY THE POWER OF STOCHASTIC THERMODYNAMICS:

$$\frac{P(W|\Delta F, \Lambda)}{P(-W|\Delta F, \Lambda)} = e \longrightarrow (e) = e \longrightarrow (w) \rightarrow \Delta F$$

$$\boxed{Crooks relation}$$

$$\boxed{2} \quad \Delta arzynski \quad equality \quad statement$$

S. Vinjanampathy and J. Anders 2016 Contemporary Physics, 57, 545 U. Seifert 2012 Rep. Prog. Phys. 75 126001

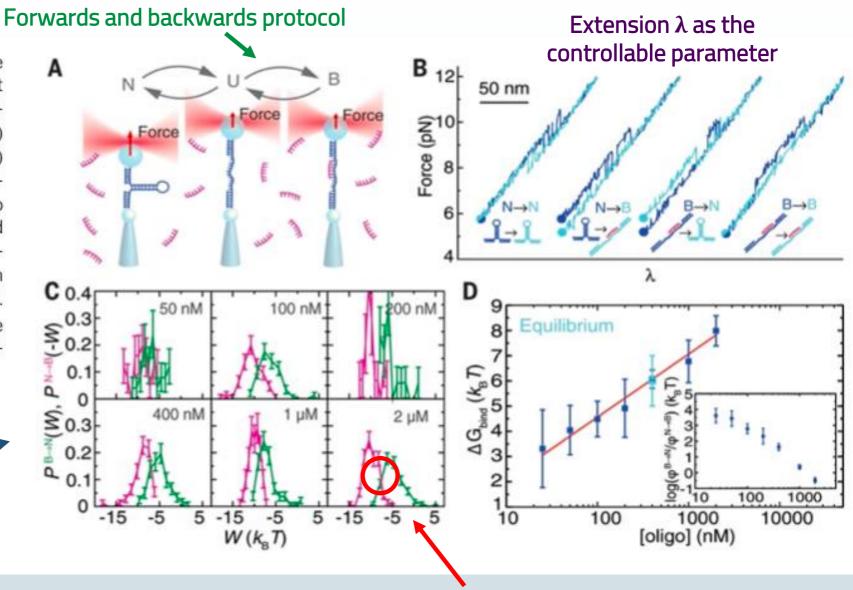


Free energy differences in single-molecule folding experiments



Fig. 2. Oligonucleotide binding to DNA. (A) Scheme of N, U, and B states. (B) Cyclic pulling curves that start and end at low forces (\sim 6 pN) classified according to their initial (blue dot) and final (cyan dot) state. (C) Partial work distributions of B \rightarrow N (green) and N \rightarrow B (magenta) transitions. (D) Binding energy of the 10-base oligonucleotide (blue) and fit to the law of mass action (red line). The value obtained from hopping equilibrium experiments at [oligonucleotide] = 400 nM is shown in cyan (see section S1.7 of the supplementary materials and methods). (D, inset) Contribution of the ratio $\phi^{N\to B}/\phi^{B\to N}$ to the binding energy. Error bars were obtained from bootstrap as described in Fig. 1.

Forwards and backwards histograms



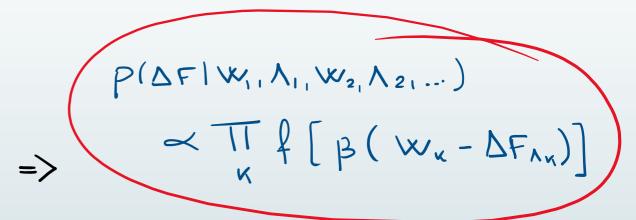
Crooks relation + Bayes theorem: a <u>finite-information approach</u> to the estimation of free energy differences



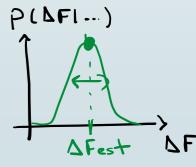
* Available information:

3
$$P(W, \Lambda | \Delta F) + P(-W, \hat{\Lambda} | \Delta F) \sim 1$$

$$\left[\frac{4}{(w_1, \lambda_1)}, (w_2, \lambda_2), \dots\right]$$



with



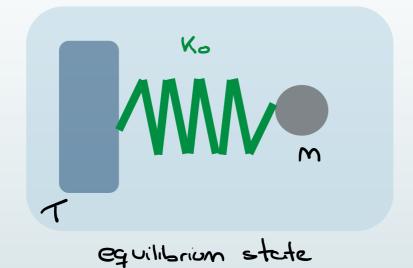
P. Maragakis et al. 2008 J. Chem. Phys. 129, 024102

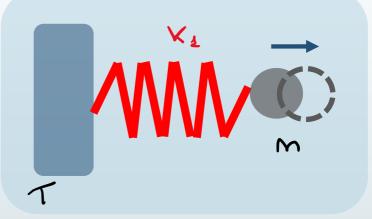
J. Rubio and J. Dunninham 2019 New J. Phys. 21 043037

Harmonic oscillator protocol



Energy:
$$H(q, p) = \frac{p^2}{zm} + \frac{m\omega^2q^2}{z}$$
, with $\omega = \sqrt{\frac{\kappa}{m}}$





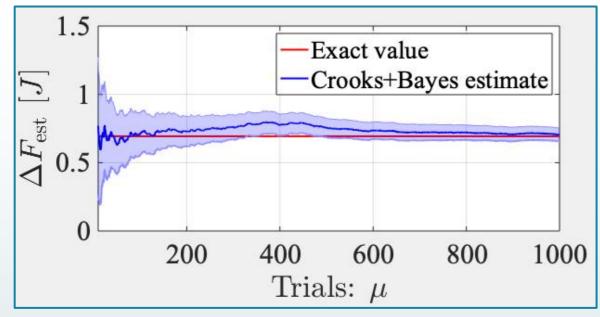
nonequilibrium state

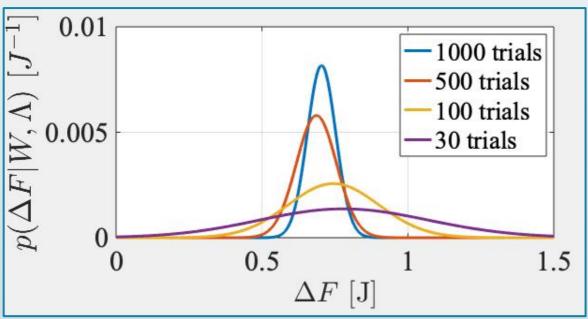
$$[W = H(q_i, P_i) - H(q_i, P_i) = \Delta H;$$
instantaneous
protocol

protocol

Harmonic oscillator protocol: free energy estimates





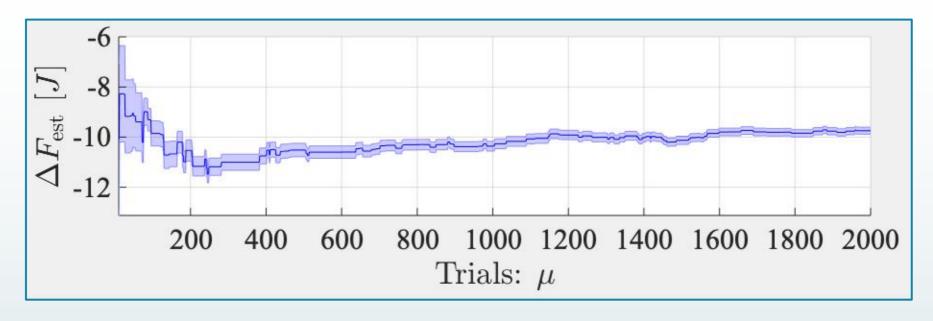


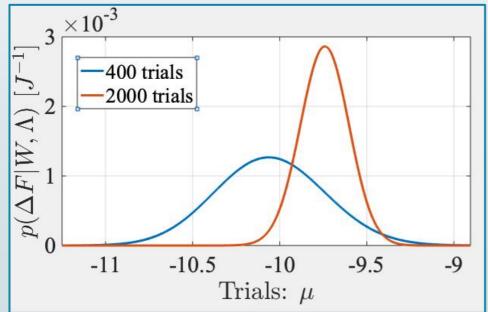
$\Delta F_{\rm exact} = \log 2 J \simeq$	0.69 <i>J</i>	

Trials	Estimate [J]	Error [J]	
30	0.8	0.3	
100	0.8	0.2	
500	0.69	0.07	
1000	0.70	0.05	

Single-molecule redox reaction as a stochastic-thermodynamical protocol EXETER







$$V_{\rm exp} = -11 \, \rm kJ/mol$$

Trials	Estimate [kJ/mol]	Error [kJ/mol]
400	-10.1	0.3
2000	-9.7	0.1

Conclusions and outlook



- Stochastic thermodynamics studies <u>transfers of energy and their fluctuations in small systems</u>. It thus enables the study of the nonequilibrium aspects of life.
- **Fluctuation theorems** enable the measurement of quantities such as <u>binding energies</u> or <u>redox potentials</u> in situations where traditional bulk measurements may be inaccurate or unfeasible.
- The MMI is implementing a **realistic, finite-information** estimation approach to the estimation of free energy differences in biological protocols.

Thank you for your attention!