

JURIS

RA Number & Name		Date	Signed by:
RA 3720	Food, Drug, Cosmetics and Devices Act	June 22, 1963	Ferdinand Marcos
RA 5921	Philippine Pharmacy Law	June 23, 1969	Ferdinand Marcos
RA 6425	Dangerous Drugs Act of 1972	March 30, 1972	
RA 6675	Generics Act on 1988	September 13, 1988	Fidel Ramos
RA 7432	Senior Citizens Act	February 7, 1992	Corazon Aquino
RA 7394	Consumer Act of the Philippines	April 13, 1992	
RA 7581	Price Act	May 27, 1992	
RA 7876	Senior Citizen Center Act of the Philippines	February 14, 1995	
RA 8172	Salt Iodization Nationwide Act	December 29, 1995	Fidel Ramos
RA 8203	Special Law on Counterfeit Drugs	September 4, 1996	Fidel Ramos
RA 8293	Intellectual Property Code		
RA 8424	Tax Reform Act of 1997		
RA 8432	Traditional and Alternative Medicine Act	December 9, 1997	Fidel Ramos
RA 8981	PRC Modernization Act of 2000	December 5, 2000	
RA 9165	Comprehensive Dangerous Drugs Act of 2002	June 7, 2002	Gloria Arroyo
RA 9257	Expanded Senior Citizen's Act of 2003	February 26, 2003	Gloria Arroyo
RA 9211	Tobacco Regulation Act of 2003	June 23, 2003	Gloria Arroyo
RA 9334	Increase tax on alcohol	July 26, 2004	
RA 9502	Universally Accesible Cheaper and Quality Medicines Act of 2008	June 6, 2008	Gloria Arroyo
RA 9711	Food & Drug Administration Act of 2009	August 18, 2009	Gloria Arroyo
RA 9994	Expanded Senior Citizen's Act of 2010	February 15, 2010	
RA 10351	Restructing Excise Tax on Alcohol & Tobacco Products	July 23, 2012	
RA 10623	Amended RA 7581 (Price Act)	July 23, 2012	
RA 10640	Anti-Drug Campaign	July 22, 2013	
RA 10645	PhilHealth Benefits for Senior Citizens	July 28, 2014	
RA 10912	Continuing Professional Development Act of 2016	July 27, 2015	
RA 10918	Philippine Pharmacy Act	July 21, 2016	Benigno Aquino

PHYSICAL PHARMACY

Hydrogen-Lipophilic Balance		$HLB = \frac{H}{L}$		Standard Conditions		Ideal Gas Law			$PV = nRT$	
Anti-foaming		1-3		Temp	Kelvin (K) 273 + °C	P	Pressure		atm	
Water in oil		4-6		Pressure	ATM (atmosphere)	V	Volume		L	
Wetting agents		7-9		1 atm	760 mmHg/ (torr) /76 cm Hg	n	Moles		$\frac{grams}{MW}$	
Oil in water		8-18			1.01325 X 10 ⁶ dynes/cm ²	R	Universal Gas Constant		$0.08205 \frac{L \cdot atm}{n \cdot K}$	
Detergents		13-16			1.01325 X 10 ⁵ N/m ² or Pascal	T	Temperature		K	
Solubilizer		5-20			1.01325 bar	Density		$\frac{mass}{volume}$	$\frac{g}{mL}$	$\frac{g}{L}$
		Constant	Formula		Non-Ideal Gas Law a = correction for pressure b = correction for volume			$\left[P - \frac{n2a}{V2}\right] [V - nb] = nRT$		
Boyle's Law		Temperature	P ₁ V ₁ = P ₂ V ₂							
Charles' Law		Pressure	$\frac{T1}{T2} = \frac{V1}{V2}$	$\frac{T1}{T2} = \frac{V1}{V2}$	Clausius-Clapeyron Equation			$log \left(\frac{P2}{P1}\right) = \frac{\Delta Hr (T2 - T1)}{2.303 RT1T2}$		
Gay-Lusaac's		Volume	$\frac{P1}{T1} = \frac{P2}{T2}$	$\frac{P1}{T1} = \frac{P2}{T2}$	P	Pressure		Atm / mmHg		
Avogadro's Law			$\frac{V1}{V2} = \frac{N1}{N2}$		T	Temperature		K		
Combined Gas Law			$\frac{P1V1}{T1} = \frac{P2V2}{T2}$		R	Gas Constant		SI 8.314 Joules/n-K		
								Non-SI 1.987 Cal/n-K		
Dalton's Law of Partial Pressure			P _T = P ₁ + P ₂ +P ₃ P _(n)		ΔHr	Latent Heat of Vaporization		J/n		
Raoult's Law	ΔVp = P _I ^o - P _I	ΔVp = X _{solute} ▪ P _I ^o		P _I = X _{solvent} ▪ P _I ^o		Specific gravity			$\frac{\rho \text{ substance}}{\rho \text{ standard}}$	

Stokes Law			VOLUME		$pKa = -logKa$	$pKa + pKb = 14$
V	Volume	$pKb = -logKb$		$Vg = Vp + \text{intraparticular space}$	$pKb = -logKb$	
d	diameter		True Volume (Vp)	$Vb = Vp + \text{Intra-} + \text{Inter-}$	Weak Acid	$pH = pKa + log \frac{[Sa]}{[Wa]}$
ρs	Density-solid			$Vb = Vg + \text{interparticular space}$	Weak Base	$pH = pKa + log \frac{[Ba]}{[Sa]}$
ρo	Density medium	$d = \sqrt{\frac{18n \cdot V}{(\rho s - \rho o)g}}$	Intraparticular porosity	$\frac{Vg - Vp}{Vg} \times 100$	Buffer Capacity / Index / Efficiency Value	
g	Gravity		Interparticular porosity	$\frac{Vb - Vg}{Vb} \times 100$	$\frac{\Delta N}{\Delta pH} = \frac{\text{Change in Normality}}{\text{Change in pH}}$	
n	viscosity		Total porosity	$\frac{Vb - Vp}{Vb} \times 100$	Newtons Law of Flow	
COMPRESSIBILITY			Fluidity		$N = \frac{F \rightarrow \text{shear stress}}{G \rightarrow \text{rate of shear}}$	
Carr's Index		$CI = \frac{Vo - Vf}{Vo} \times 100$	Solubility		Kinetic Viscosity	Unit: Stokes
0 = Vo = Vf		Not compressible	Molar solubility		$\frac{\text{absolute viscosity}}{\text{Density}}$	
1-99 = Vo > Vf		compressible	Molarity		Retrieve Viscosity	No unit
100 = Vf = O		Not possible	Molality		$\frac{\text{absolute viscosity}}{\text{viscosity of water}}$	
Hausner's Ratio		$HR = \frac{Vo}{Vf}$	Boiling Point Elevation		$Kb (H_2O) = 0.52^{\circ}C/\text{molal}$	
I = Vo = Vf		Not compressible	$\Delta Tb = Tbsoln - Tb^{\circ}$		$\Delta Tb = mKb$	$\Delta Tb = imKb - \text{electrolyte}$
> 1 = Vo>Vf		Compressible	Freezing Point Depression		$Kf (H_2O) = 1.86^{\circ}C/\text{molal}$	
<1 = Vo = Vf		Not possible	$\Delta Tf = Tf^{\circ} - soln$		$\Delta Tb = mKf$	$\Delta Kf = imKf - \text{electrolyte}$

Class I – add solute					Theories	Acid	Base		
D Method		E Method			Arrhenius *hydrogen	H+ donor	H+ acceptor		
$\frac{\frac{MW}{10}}{-1.86(i)} = \frac{\%}{-0.52}$		Old	$Evalue = 17 \frac{Liso\ Drug}{MW\ Drug}$		Bronsted-Lowry *proton	Proton donor	Proton acceptor		
Strong electrolyte	$i = 2$	New	$Evalue = \frac{MW\ NaCl}{iNaCl} \times \frac{Liso\ Drug}{MW\ Drug}$		Lewis *electron	Electron acceptor	Electron donor		
Weak electrolyte				Liso	<div>Goodluck! Kaya mo yan v Tiwala lang =) hindi ka ootot dito kung hindi mo kaya. Kanti na kung, ngayon ka pa ka sumoto?! Magpalunga kung napapagod. Wag panghinaan, hindi ka mag-isa sa laban na ito.</div> <div>- Gapang Pharma :*</div>				
# ions	i (80%)	Non-elect	Sucrose, Urea, Glycerin					1.9	
2	1.8	Weak elect	H ₃ BO ₃ , Cocaine, Phenobarbital					2.0	
3	2.6	Divalent	MgSO ₄ , ZnSO ₄					2.0	
4	3.4	Uni-uni	NaCl, KCl, Pilocarpine NO ₂ , Ephedrine SO ₄					3.4	
5	4.2	Uni-di	Na ₂ SO ₄ , Atropine SO ₄					4.3	
Non electrolyte	$i = 1$	Di-uni	ZnCl ₂ , CaBr ₂		4.8	REACTION KINETICS			
Class II – add solvent		Mono-tri	Na Citrate, K Citrate		5.2		Zero	First	Second
White Vincent Method		Tri-mono	AlCl ₃ , FeCl ₃		6	K (reaction Constant)	$\frac{Co - Ct}{t}$	$ln(\frac{Co}{Ct})(\frac{l}{t})$	$\frac{Co - Ct}{Co \cdot Ct} (\frac{l}{t})$
$Vol\ of\ H_2O = wt \times E \times 111.1$									
Sprowl's Method		Newtons Law of Flow			$t_{1/2}$	$\frac{0.5\ Co}{Ko}$	$\frac{0.693}{K1}$	$\frac{1}{Co \cdot K2}$	
$Vvalue = 0.3g \times E \times 111.1$		$N = \frac{F \rightarrow shear\ stress}{G \rightarrow rate\ of\ shear}$			t_{90}	$\frac{0.1\ Co}{Ko}$	$\frac{0.105}{K1}$		
					Unit of K	$\frac{Conc.}{Time}$	$\frac{1}{Time}$	$\frac{1}{Conc. \cdot Time}$	

MANUF

DEPARTMENTS OF MANUFACTURING PLANT			TYPES OF PACKAGING		POLYMERS OF PLASTIC					
1. Research & Development			Primary	Immediate container		1. PET (Polyethylene Terephthalate)	beverages			
2. Production Department			Secondary	Outer Packaging		2. HDPE (High-Density Polyethylene)	Thermoset for solid dosage forms			
3. Warehouse Deparment			Tertiary	Corrugated box						
3. Warehouse Department			CLASSIFICATION OF CONTAINERS				3. PVC (Polyvinyl Chloride)	Less resistant to permeation		
4. Quality Assurance Department			A. Protection Ability							
5. Quality Control Department			1. Well-Closed		4. Light-resistant		4. LDPE (Low-Density Polyethylene)	Thermoplastic for squeeze bottles & medicine droppers		
6. Marketing Department			2. Tight		5. Child-resistant					
7. Regulatory Department			3. Hermetic		6. Tamper-resistant					
8. Engineering Department			B. Quantity Held				5. Polypropylene	High temp, resistance		
9. Medical Department			1.Single Unit		2. Multiple Unit		6. Polystyrene			
STAGES OF DRUG DEVELOPMENT			C. Materials Used				7. Others			
1. Discovery & Development			1. Glass		4. Rubber		LABELLING			
2. Pre-clinical Research			2. Plastic		5. Metal		1. Name of the Product	4. Rx Symbol		
3. Clinical Research			3. Foils, Films & Laminates				2. Dosage form & strength	5. Name & Complete address of Manufacturer, trader or distributor		
4. FDA Review			TYPES OF GLASSES							
Phase			I	Highly resistant Borosilicate glass				3. Pharmacologic category	6. Net Content	
1	Healthy human	Safety	II	Treated Soda Lime Glass						
2	Subj w/ dse. (Small)	Efficacy	III	Soda Lime Glass						
3	Subj w/ dse. (Larger)	Safety & Efficacy	IV	General Purpose				STORAGE CONDITIONS		9C = 5F - 160
			TYPES OF PLASTIC				Cold	NMT 8°C Freezer → -20 to -10 °C	Refrigerator → 2-8 °C	
4	Post Marketing Surv.		1. Thermoplastic		2. Thermoset					
TABLET COMPONENTS					Cool		8-15 °C			
1. APIs			7. Colorants				Room Temp	Temp. prevailing in the place		
2. Diluents/Fillers		LaSuSDiAMiMa	8. Flavors				Controlled RT	20-25 °C		
3. Binder		StAT GeSuCeP	9. Sweeteners				Warm	30-40 °C		
4. Disintegrant		SCC					Excessive Heat	>40 °C		
5. Super disintegrants										
6. Antifrictionals		LAG								

DILUENTS/FILLERS		SUPER DISINTEGRANTS				ANTIFRICTIONALS			
1. Lactose	1. Sodium Starch Glycolate		Explotab [®] , Primogel [®]		3 Roles: Lubricant, Anti-adherent, Glidant				
2. Sucrose	2. Croscarmellose Na		Cross polymers		Stearates Mg,Ca,Na		3 roles		
3. Starch	3. Crospovidone				Purified Talc		LA		
4. Dibasic CaHPO ₄	ANTIFRICTIONALS				Colloidal Talc		Glidant	Cab-o-sil [®]	
5. Anhydrous Lactose	Blue No.1	Brilliant blue	Red No.40	Allura Red	Colloidal SiO ₂				
6. Microcrystalline Cellulose	Blue No.2	Indigotine	Yellow No.5	Tartrazine	Silicates Ca,Mg		Glidant		
7. Mannitol & Xylitol	Green No.3	Fast Green	Yellow No.6	Sunset Yellow	PEG & SLS				
BINDER		Red No.3	Erythrosine		SWEETENERS				
1. Starch	FLAVORS				Sucralose	1000x Splenda	Acesulfame K	180-200x	
2. Acacia	Salty	Cinnamon, Orange, Cherry, Butterscoth		SCCOB	Saccharin	500x	Aspartame	180-200x	
3. Tragacanth					Na Saccharin	300x	Na cyclamate	30x magic sugar	
4. Gelatins	Bitter	Chocolate, Cherry, Raspberry, Mint		Bitter C CRaM	Equal = Acesulfame K + Aspartame				
5. Sucrose									
6. Cellulose	Sour	Raspberry, Lemon, Fruity		SouRaLF					
7. PVP	Oily	Mint, Lemon, Orange		OiLeM					
	Unpleasant Sweet	Vanilla, Fruity		UnVF					
PROCESS IN TABLET MANUF									
Dispensing → Milling → Mixing → Granulation → Tableting → Coating									
MILLING EQUIPMENTS		MIXING EQUIPMENTS		GRANULATION			TABLETING		
1. Cutter Mill	1. Batch Type Mixer		Methods:			Parts:			
2. Edge Runner Mill	a. Rotating Shell Mixer		1. Wet Granulation		2. Dry Granulation	1. Hopper	4. Punches		
3. Hammer Mill	▪ Drum-type blender		Old Process		a. Slugging	2. Feed Shoe	5. Cam tracks		
4. Fluid Energy Mill	▪ Slanted Position of the Cylindrical Drum		1. Blending of Dry Ingredients		b. Roller Compaction	3. Die			
5. Roller Mill			2. Addition of Liquid Blender			Types:			
6. Ball Mill	b. Fixed Shell Mixer		3. Screening of Damp Mass			Single Station	Multiple Station		
	▪ Ribbon Blender		4. Drying the Granulation			Requirements:			
	▪ Sigma Blade Mixer		5. Screening the dry granules			1. Flowability		(2)	
	▪ Planetary Mixer		New Process			2. Compressiblity			
	▪ Vertical Impeller		Fluid Bed Granulator						

TABLET DEFECTS			COATING					
Process		Equipments:	Examples	FILM-COATING STEPS		Examples		
1. Capping	3. Cracking	1. Standards Coating Pan	Pellegrini Fan	1. Film-former				
2. Lamination		2. Perforated Coating Pan	Accela-Cota Pan, Glatt coater	NonCeMeP	Non-enteric	Cellulose, Methacylate, PVP		
Excipients					Enteric	Shellac, CAP, PVAP, salol		
1. Chipping	3. Picking	3. Fluid Bed Coater	Air suspension or Wurster process	2. Plasticizer	PlastiCErin	Castor oil, Glycerin		
2. Sticking				3. Surfactant		Polysorbate Tween®		
Machine		Types:		4. Alloying Substance		PEG		
1. Double impression		1. Sugar-coating	2. Film-coating	5. Glossant		Beeswax		
More than 1 factor		SUGAR-COATING STEPS			6. Volatile solvent/vehicle		Alcohol + Acetone	
1. Mottling		1. Sealing 2. Subcoating 3. Smoothing 4. Color-coating 5. Polishing	CAP Cellulose PVAP Polyvinyl	Acetate Pthalate	COATING DEFECTS			
						1. Mottling	6. Blistering	11. Film Cracking Type I
						2. Sweating	7. Blooming	12. Film Cracking Type II
						3. Bridging	8. Blushing	13.Delayed dissolution
						4. Erosion	9. Twinning	
						5. Cratering	10. Orange Peel	
MANUFACTURING OF CAPSULES				MANUFACTURING OF SEMI-SOLID DOSAGE FORMS				
A. Hard Gelatin Capsules		B. Soft Gelatin Capsule		A. Ointments				
Steps:	Special Technique	Method		Methods:	1. Incorporation method		2. Fusion	
1. Supply	1. Sealing	1. Plate Process		B. Gels				
2. Rectification	▪ Gelatin Bonding	2. Rotary Die Process		Alginic Acid	Natural	Colloidal SiO ₂	Cab-o-sil®	
3.Separation	▪ Heat welding	3. Reciprocating Die Process		Cellulose		Mg Al Silicate	Veegum®	
4. Filling	▪ Thermal coupling			Tragacanth		Carbomer	Carbopol®	
5. Joining/Closing	2. Imprinting							
6.Finishing	3. Coating							
MANUFACTURING OF LIQUID DOSAGE FORMS								
A. Equipments:			Components:			Stability Enhancers		
1. Mixing tanks	SS 304	SS316	1. APIs	4. Viscosity Enhancers		Preservatives		
2. Mixers	▪ Mechanical stirrer	▪ Homogenizer	2. Solvent or vehicle	5. Humectant		Antioxidants		
	▪ Colloidal Mill	▪ Ultrasonifier	3. Buffer	6. Stability Enhancers				

PRESERVATIVES		ANTIOXIDANTS		SOLUTION			
Parabens (p-hydroxybenzoic acid)		▪ True Antioxidants		Dispensing → Mixing → Storage → Filtration → Filling & Aging			
MOA: taken by cell membrane of microorganisms & lysis the cell		1. Vit. E	2. Alkylgalates	Co-solvents: Alcohol : Glycerin			
		Butylated Hydroxy	3. Toluene BHT 4. Anisole BHA				
<u>Methylparaben</u>	molds	▪ Reducing Agents		Membrane Filter	0.22 μm	For sterile dosage from	
<u>Ethylparaben</u>					0.5 μm	Smallest bacteria	
<u>Propylparaben</u>	Yeasts & bacteria	1. Vit. C	3. Glutathione		QC: Bubble point Test		
<u>Benzoic acid</u>	Disrupting of Cell membrane	2. Sulfites		Filling	Gravimetric	Constant level	
<u>Sorbic acid</u>		▪ Antioxidant synergist		SUSPENSION			
<u>Benzyl alcohol</u>	Protein	1. EDTA	3. Tartaric acid	Formulation:			
<u>Chlorobutanol</u>	denaturation	2. Citric acid		Suspending Agents		Wetting agents	
<u>Benzalkonium Cl</u>	Disrupting of Cell membrane	Mnemonics		<u>Veegu</u>	<u>Bentonite magma</u>	<u>Glycerin</u>	<u>Syrup</u>
		<u>Suspending</u>	VeCeAT BeAGCa	<u>Cellulose</u>	<u>Gelatin</u>	<u>PEG</u>	<u>Surfactant</u>
<u>Thimerosal & Phenylmercuric NO₃</u>	Cause enzyme inhibition	<u>Wetting</u>	SurfGlyPPS	<u>Acacia</u>	<u>Agar</u>	<u>PPG</u>	
		<u>Flocculating</u>	Na/K Cl	<u>Tragacanth</u>	<u>Carageenan</u>		
				Flocculating agents	<u>NaCl</u>	<u>KCl</u>	
EMULSION							
Theories:	Key words	Carbohydrates		Finely Divided Solid		Instabilities:	
1. <u>Surface Tension</u>	“spherical”	<u>Acacia</u>	<u>Pectin</u>	<u>Bentonite</u>	<u>AlOH₂</u>	1. Sedimentation	
2. <u>Oriented Wedge</u>	“monomolecular”	<u>Agar</u>	<u>Xanthan</u>	<u>MgOH₂</u>		2. Creaming	
3. <u>Interfacial Film</u>	“Interface between the oil & water”	<u>Carageenan</u>	<u>Tragacanth</u>	Synthethic surfactants		3. Breaking/Cracking	
		Protein		a. Anionic		4. Phase inversion	
4. <u>Viscosity</u>	“viscosity”	<u>Gelatin</u>	<u>Casein</u>	<u>Soaps</u>			
Emulsifying agents		<u>Egg yolk</u>	<u>Casein</u>	<u>Sodium Lauryl</u>	<u>Sulfate SLS</u>		
1. Carbohydrates	4. Finely Divided Solid	HMW alcohol			<u>Ether SLE</u>		
2. Proteins	5. Synthetic Surfactants	<u>Stearyl alcohol</u>		b. Cationic	<u>Benzalkonium Cl</u>		
3. HMW Alcohol		<u>Cetyl alcohol</u>		c. Amphoteric	<u>Betaine (Cocamidopropyl betaine)</u>		
		<u>Glyceryl Monostearate</u>		d. Non-ionic	<u>Span® & Tween®</u>		
		<u>Cholesterol</u>		<u>Span®</u>	Sebo	<u>Sorbitan esters</u>	<u>Monostearate & Monoacetate</u>
				<u>Tween®</u>	Tubig	<u>Polyoxyethylene sorbitane</u>	

MANUFACTURING OF STERILE DOSAGE FORMS							
A. Sterilization method					B. Depyrogenation		
	Instrument	MOA	Condition	BI	Oven settings		
Moist heat	Autoclave	Protein coagulation	121°C, 15psi, 15mins.	Bacillus stearothermophilus	180°C for 4hrs.	600°C for 1hr.	
					250°C for 45mins.		
Dry heat	Oven	Oxidation	160-170 °C, 2-4hrs.	Bacillus subtilis	C. Sterile Production Area		
					Clean Room Parts:		
Membrane Filtration		Physical separation		Brevundimonas diminuta	1. Ante-room	ISO Class 100,000	
					2. Buffer Area	ISO Class 10,000	
Gas	Ethylene Oxide or β-popiolactone	Alkylation		Bacillus subtilis	3. Compounding Area		
					4. Aseptic Filling Area		
Ionizing radation	Gamma or cathode rays	DNA mutation		Bacillus pumilus	5. Quarantine Area		
					6. Finishing Area		
PROCEDURE							
1. Dispensing & Cleaning			4. Filling	Net weight = solids			
				Volumetric & Gravimetric for liquids			
2. Compounding	Spray drying & Freeze drying for sterile solids		5. Sealing	Vials: Siliconiation or Halogenization			
				Ampules:	Tip/Bead sealing		
Clarification		2-3 µm			Pull sealing		
3. Filtration	Cold filtration			0.2-2.3 µm			