

Video: https://youtu.be/5Rkt-0MDFUk

普物期末程式專題 布朗運動

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前言

1.程式簡潔易懂,vpython 獨力編寫

2. 秉持模擬精神,不要導果為因

3.以物理觀念探討其他學術領域

4.從錯誤中學習、檢討



3D動 畫模擬 分子位 移分布 與機率

幾何布 朗運動



3D動畫模擬

簡介

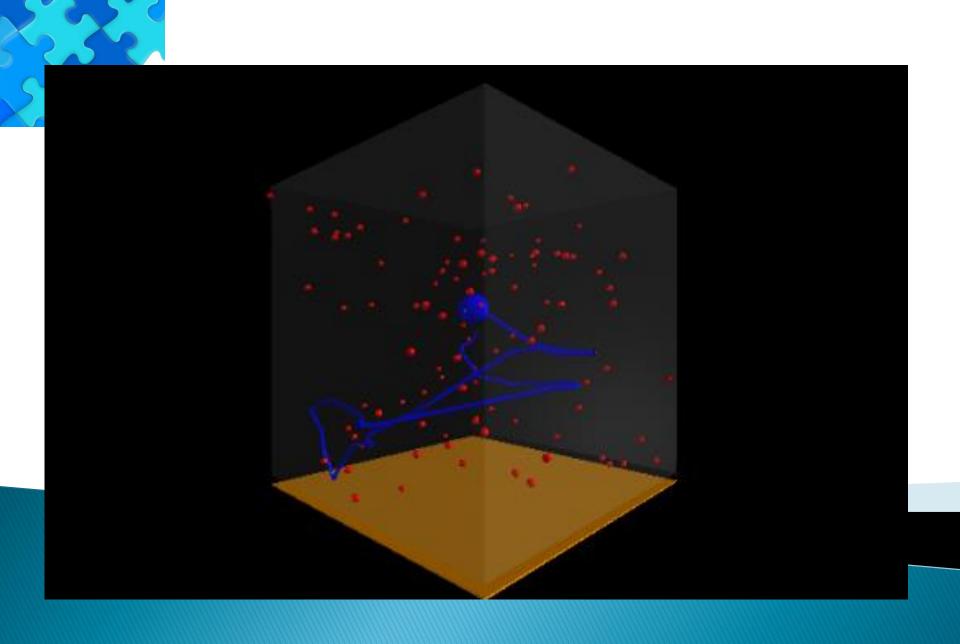
1.液體分子的熱擾動繼續,懸浮粒子便會繼鎖顯出隨機運動,即布朗運動

2.原子分子的存在乃氣體運動的基礎,而有期運動則是證明分子存在的一種現象

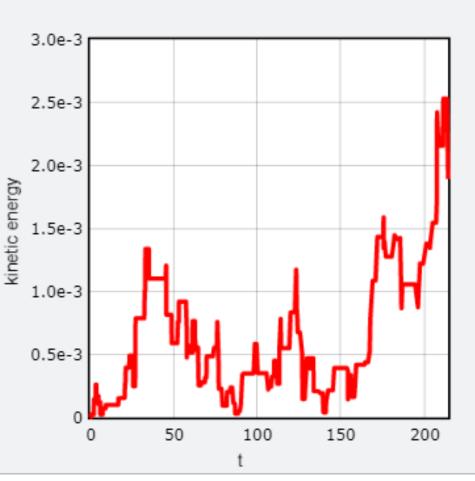
員的

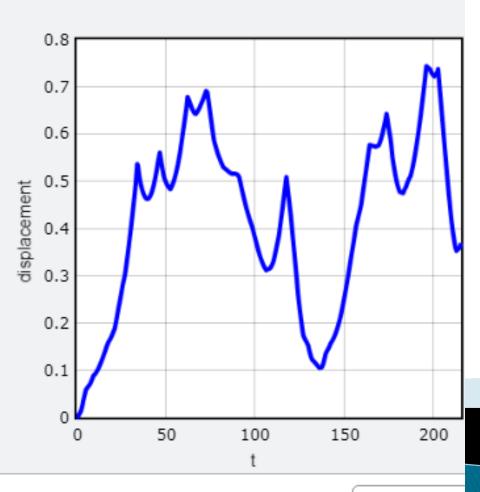
1.3D布朗運動模擬(boundary condition): #體積一單位的箱子 #一個大分子(r=0.05, m=1,原點,v0=0) #100個小分子(r=0.01, m=0.01)

2.旁邊顯示大分子隨時間變化的動能和位移









程式碼

```
make_molecules(n_molecules, radius, mass):
    molecules = [Dust(
        pos=vector(0, 0, 0),
        vel=vector(0, 0, 0),
        col=color.blue.
        rad=0.05.
        mass=1
    for i in range(n_molecules):
        molecules.append(Particle(
            pos=vector(r() - 0.5, r() - 0.5, r() - 0.5),
            vel=vector(r() - 0.5, r() - 0.5, r() - 0.5),
            col=color.red.
            rad=radius,
            mass=mass
    return molecules
def update_velocity(molecules, dt):
    for m in molecules:
        calc wall collision(m)
    calc part collision(molecules[0], molecules)
def update_position(molecules, dt):
    for m in molecules:
```

m.pos += m.vel * dt

```
# Stable (no oscillating velocity) because return to inside bounding box
# is guaranteed
def calc_wall_collision(particle):
    box bounds = 0.5
    if abs(particle.pos.x) >= box bounds:
        particle.vel.x *= -1
    if abs(particle.pos.y) >= box_bounds:
        particle.vel.v *= -1
    if abs(particle.pos.z) >= box_bounds:
        particle.vel.z *= -1
def calc_part_collision(p, molecules):
    for m in molecules:
        # Avoid collision with itself
        if p is m:
            continue
        # If collision detected, perform elastic momentum transfer
        # (Python simultaneous assignment ftw!)
        if mag(p.pos - m.pos) <= p.radius + m.radius:
            # If masses are equal, velocities are swapped
            if abs(p.mass - m.mass) < le-3:
                p.vel, m.vel = m.vel, p.vel
            else:
                p.vel. m.vel = \
                    (p.vel * (p.mass - m.mass) + 2 * m.mass * m.vel) / (p.mass + m.mass), \
                    (m.vel * (m.mass - p.mass) + 2 * p.mass * p.vel) / (p.mass + m.mass)
            # Process only one collision per frame
            break
```

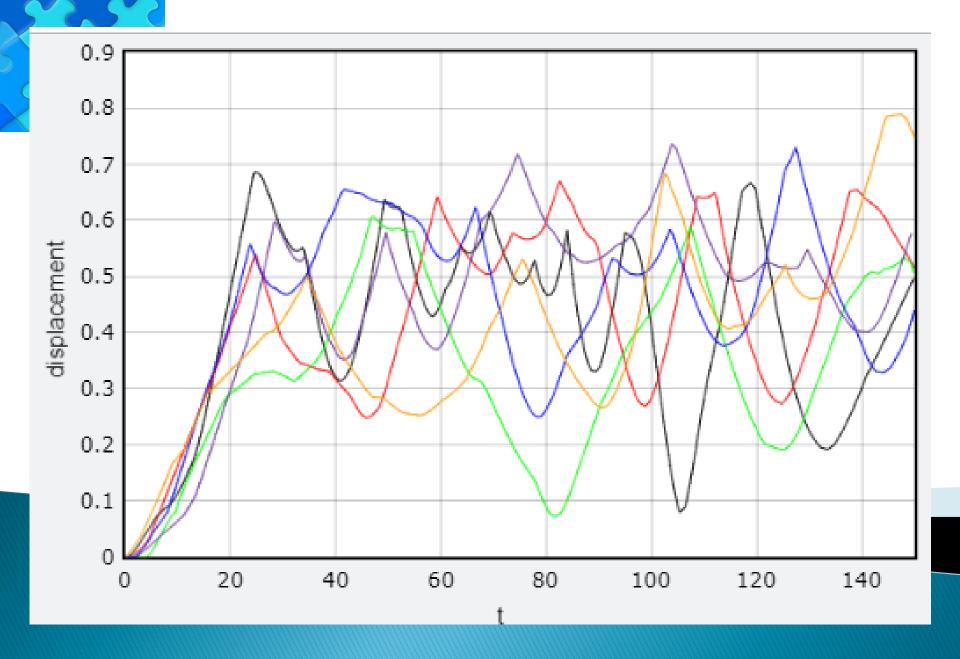


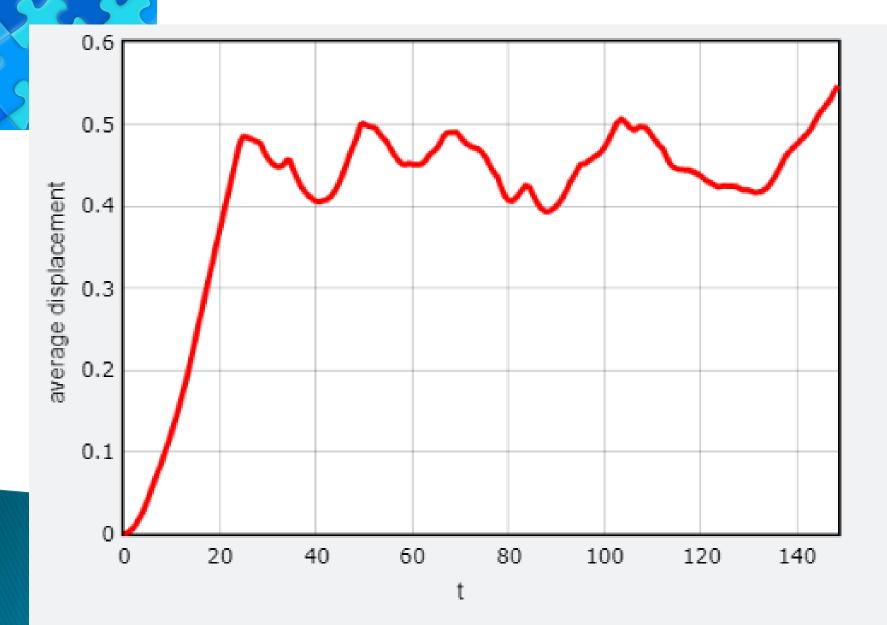
分子位移分布與機率

目的

1.進行多個樣本模擬(預設6個),求出平均位移量,隨時間變化顯示

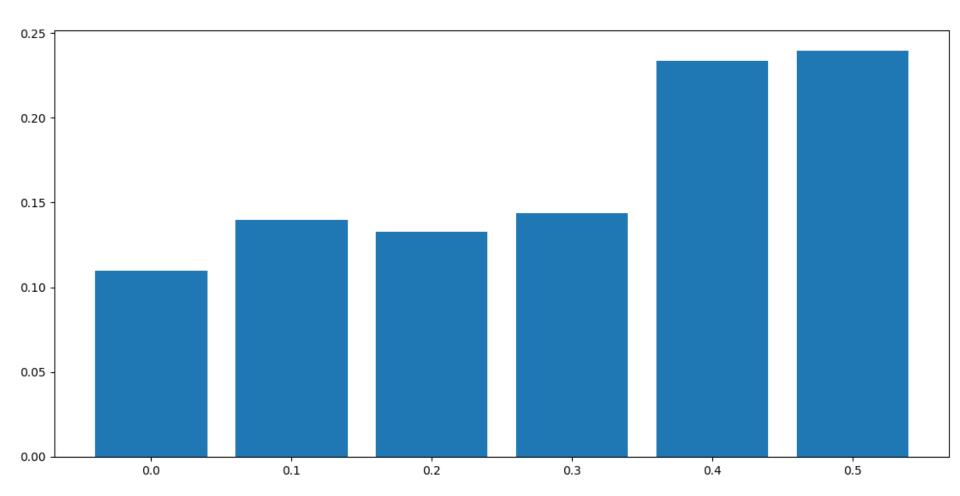
2.比較不同時間後位移量的機率分布,預測其符合的數學分布





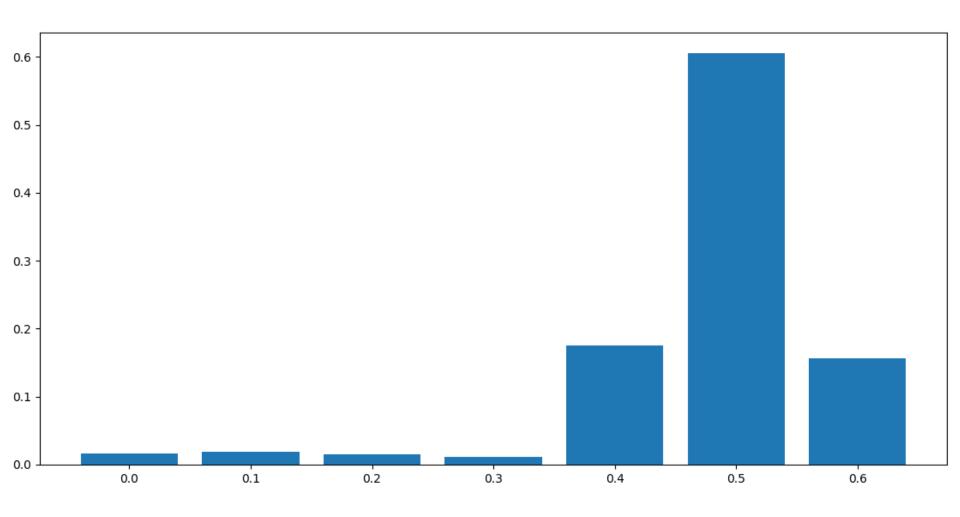


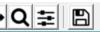
t = 100





t = 250



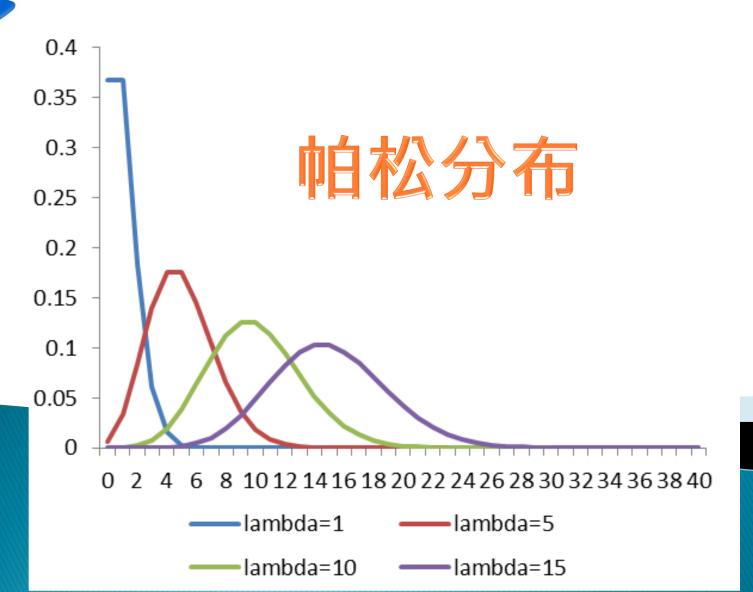


程式碼

```
class Dust:
    def __init__(self,pos,vel,rad,mass):
        self.pos = pos
        self.vel = vel
        self.rad = rad
        self.mass = mass
class Particle:
    def init (self, pos, vel, rad, mass):
        self.pos = pos
        self.vel = vel
        self.rad = rad
        self.mass = mass
class many molecules: #all molecules in one sample
    def init (self, n molecules, radius, mass):
        self.num = []
        self.num.append(Dust(vector(0, 0, 0), vector(0, 0, 0), 0.05, 1))
        for i in range(n molecules):
             self.num.append(Particle(vector(\mathbf{r}()) - 0.5, \mathbf{r}()) - 0.5, \mathbf{r}() - 0.5), vector(\mathbf{r}()) -
class collection: #collection of all samples
    def __init__(self,num):
        self.samples = []
        for i in range(num):
             self.samples.append(many_molecules(n_molecules=100, radius=0.01, mass=0.01).r
```

```
if t \ge 250:
        break
prob_dict = \{\}
for sample in prob_collection:
    if sample not in prob_dict:
        prob_dict[sample]=1
    else:
        prob dict[sample]+=1
all samples = sum(prob dict.values())
for sample in prob dict:
    prob dict[sample] = float(prob dict[sample]/all samples)
D = \{ \}
for key in prob_dict:
    D[str(key)] = prob dict[key]
plt.bar(range(len(D)), list(D.values()), align='center')
plt.xticks(range(len(D)), list(D.keys()))
plt.show()
```

預測數學分布





適合於描述單位時間內隨機 事件發生的次數的機率分佈。

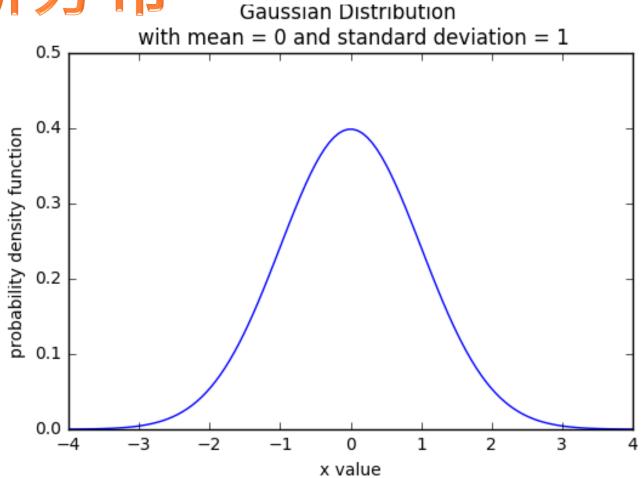
#參數入是單位時間(或單位面積)內隨機事件的平均發生率

$$P(X=k)=rac{e^{-\lambda}\lambda^{\kappa}}{k!}$$



真實數學分布

高斯分布



考慮三維情形,
$$\langle R^2 \rangle = \langle x^2 \rangle + \langle y^2 \rangle + \langle z^2 \rangle$$

$$\underline{\mathbb{H}}\left\langle x^{2}\right\rangle =\left\langle y^{2}\right\rangle =\left\langle z^{2}\right\rangle$$

$$\Rightarrow \langle R^2 \rangle = 3\langle x^2 \rangle$$

$$\therefore \left\langle R^2 \right\rangle = \frac{6KT \cdot t}{\mu}$$

$$P_{n(x)} = \frac{2}{(2\pi n l^2)^{\frac{1}{2}}} \cdot e^{-\frac{x^2}{2l^2 n}}$$

TA 其中
reference
$$\sigma^2 = nl^2 = \alpha t$$
= 0



推斷可能錯誤原因

- 1.規模小、粒子數不足
 - 2.未考慮溫度
 - 3.未考慮液體黏性
- 4.只有衡量位移量(純量)
- 5.Boundary condition 大分子會與容器壁相撞



幾何布朗運動

簡介

1.連續時間情況下的隨機過程,其中隨機變量的對數遵循布朗運動,也稱維那過程

2.幾何布朗運動在金融數學中有所應用, 用來模仿股票價格

$$dS_t = \mu S_t dt + \sigma S_t dW_t$$

Wt是一個維那過程,也是布朗運動,而 μ ('百分比drift') 和 σ ('百分volatility') 則是常數。

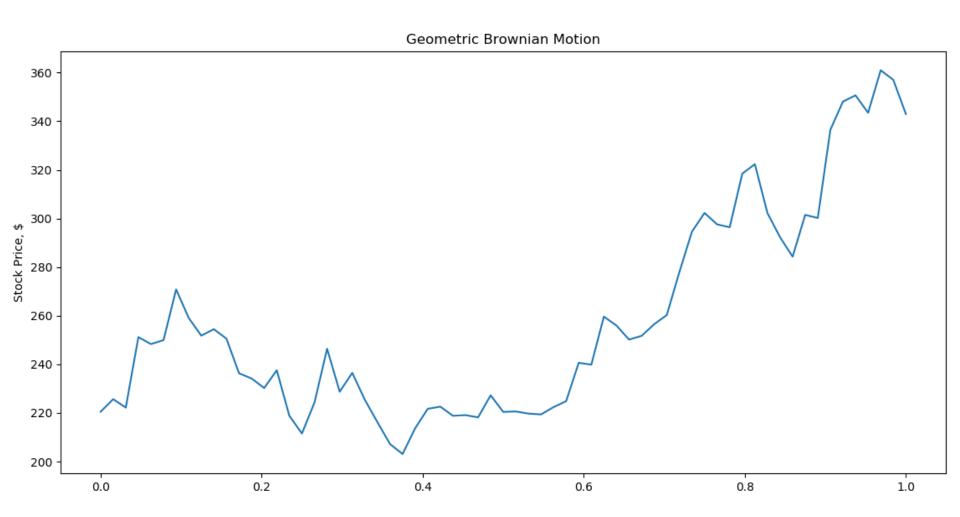


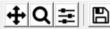
1.用SDEs 模擬股價的浮動 簡單版:許多變數因子未考慮

2.探討布朗運動和GBM的關係,把理論物理 學的觀念應用於其他領域



seed=5

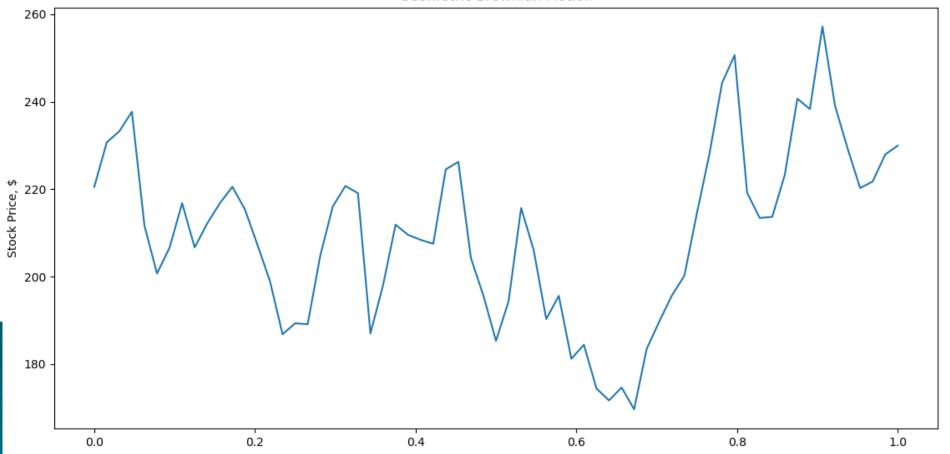






seed=20







程式碼

T = 1.

```
def Brownian(seed, N):
    np.random.seed(seed)
    dt = 1./N
                                                   # time step
    b = np.random.normal(0., 1., int(N))*np.sqrt(dt) # brownian increments
    W = np.cumsum(b)
                                                    # brownian path
    return W. b
def GBM(So, mu, sigma, W, T, N):
    t = np.linspace(0.,1.,N+1)
    S = []
    S.append(So)
    for i in range(1,int(N+1)):
        drift = (mu - 0.5 * sigma**2) * t[i]
        diffusion = sigma * W[i-1]
        S \text{ temp} = So*np.exp(drift + diffusion)
        S.append(S_temp)
    return S, t
So = 220.5 #tsmc
mu = 0.15
sigma = 0.4
seed = 20
W = Brownian(seed, N)[0]
```

soln = GBM(So, mu, sigma, W, T, N)[0] # Exact solution t = GBM(So, mu, sigma, W, T, N)[1] # time increments for plotting

可進步的空間

- 1.不同變因(溫度、黏滯係數)的布朗運動
- 2.驗證波茲曼常數
- 3.考慮阻滯力、推導擴散係數
- 4.愛因斯坦方程
- 5.讓程式有更大規模,不影響模擬速率



- 1.陳昱蓁、許耘慈兩位課程助教提供的額外參考資料
- 2.https://matplotlib.org/tutorials/introductory/pyplot.html
- 3. http://cs231n.github.io/python-numpy-tutorial/



4.https://www.itsfun.com.tw/%E5%B9%B E%E4%BD%95%E5%B8%83%E6%9C%97 %E9%81%8B%E5%8B%95/wiki-4755729-3108498 5.https://web.math.sinica.edu.tw/math_m edia/d164/16408.pdf



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