

Project 1

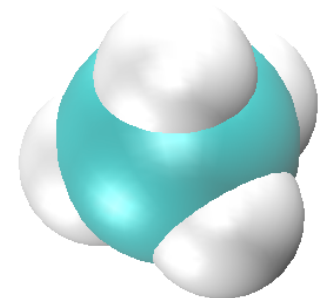
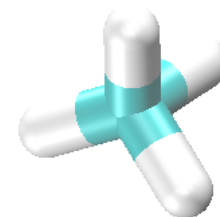
**MD simulations and calculations
of kinetic properties**

Assignments



- Compute the Einstein's diffusion constant
- Compute the velocity autocorrelation function
- Validate the Einstein relation of Langevin dynamics (the fluctuation-dissipation theorem)

- MD trajectories are available at [box.com](https://uwmadison.box.com/s/wkijstaz6h3ehtkkmk2pbwrh7xw7to2t7)



Random walk of a methane molecule in water: <https://uwmadison.box.com/s/wkijstaz6h3ehtkkmk2pbwrh7xw7to2t7>

Notes

- Einstein's diffusion coefficient:

$$\overline{\Delta \mathbf{x}(t)^2} = 2Dt$$

$$\overline{\Delta \mathbf{x}(t)^2} \equiv \overline{|\mathbf{x}(t) - \mathbf{x}(0)|^2}$$

$$= \sum_{t_0} |\mathbf{x}(t_0 + t) - \mathbf{x}(t_0)|^2 / \sum_{t_0} 1$$

- The velocity autocorrelation function:

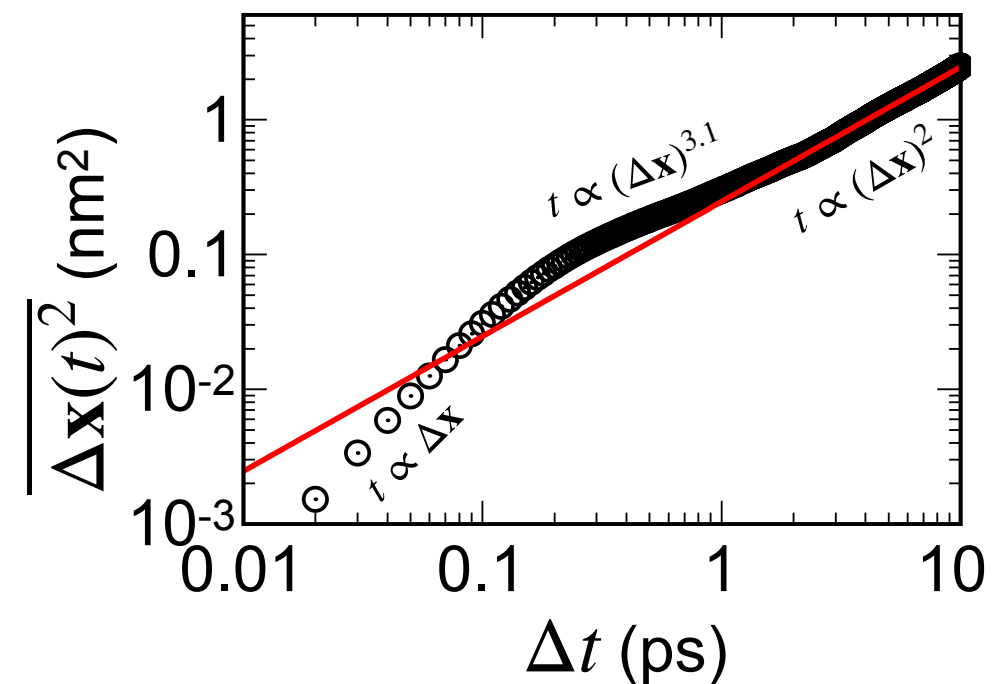
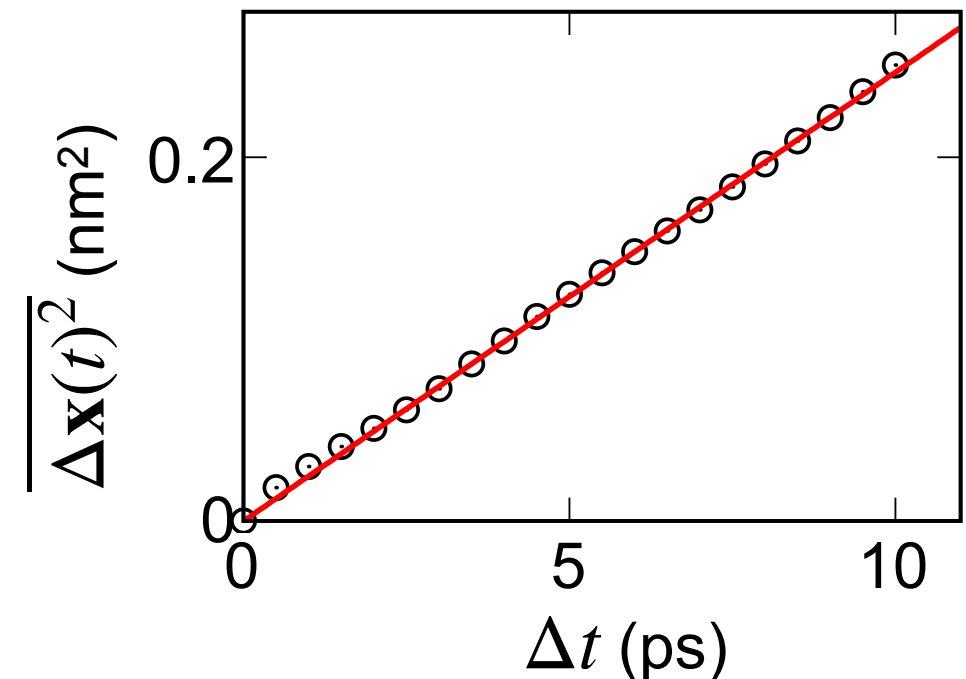
$$C_{vv}(t) = \overline{\mathbf{v}(t) \cdot \mathbf{v}(0)} = \sum_{t_0} \mathbf{v}(t_0 + t) \cdot \mathbf{v}(t_0) / \sum_{t_0} 1$$

- The equal-partition theorem:

$$\overline{\frac{1}{2} m \mathbf{v} \cdot \mathbf{v}} = \frac{3}{2} kT$$

$$C_{vv}(0) = \frac{3kT}{2m}$$

$$\frac{0.5 \times 16 \text{g/mol} \times 0.4608 \text{nm}^2/\text{ps}^2}{8.3145 \text{J/mol/K} \times 295 \text{K}} = 1.5$$



Notes

- The Langevin's equation and velocity autocorrelation function

$$m \frac{d}{dt} \mathbf{v}(t) = -m\gamma \mathbf{v}(t) + \mathbf{F}(t) \qquad m \frac{d}{dt} \mathbf{v}(t) = -\vec{\nabla} U(x) - \gamma \mathbf{v}(t) + \mathbf{F}(t)$$

$$\mathbf{v}(t) = \int_{-\infty}^t \frac{F(t')}{m} e^{-\gamma(t-t')} dt'$$

$$\overline{\mathbf{v}(t) \cdot \mathbf{v}(0)} = \int_{-\infty}^t dt_1 \int_{-\infty}^0 dt_2 \frac{\overline{\mathbf{F}(t_1) \cdot \mathbf{F}(t_2)}}{m^2} e^{-\gamma(t-t_1-t_2)}$$

$$= \int_{-\infty}^t dt_1 \int_{-\infty}^0 dt_2 \frac{a_F \delta(t_1 - t_2)}{m^2} e^{-\gamma(t-t_1-t_2)}$$

$$\begin{aligned} \overline{\mathbf{F}(t_1) \cdot \mathbf{F}(t_2)} &= a_F \delta(t_1 - t_2) \\ a_F &= \overline{\mathbf{F}(t)^2} \end{aligned}$$

$$C_{vv}(t) = \overline{\mathbf{v}(t) \cdot \mathbf{v}(0)} = \frac{a_F}{2m^2\gamma} e^{-\gamma t} = C_{vv}(0) e^{-\gamma t}$$

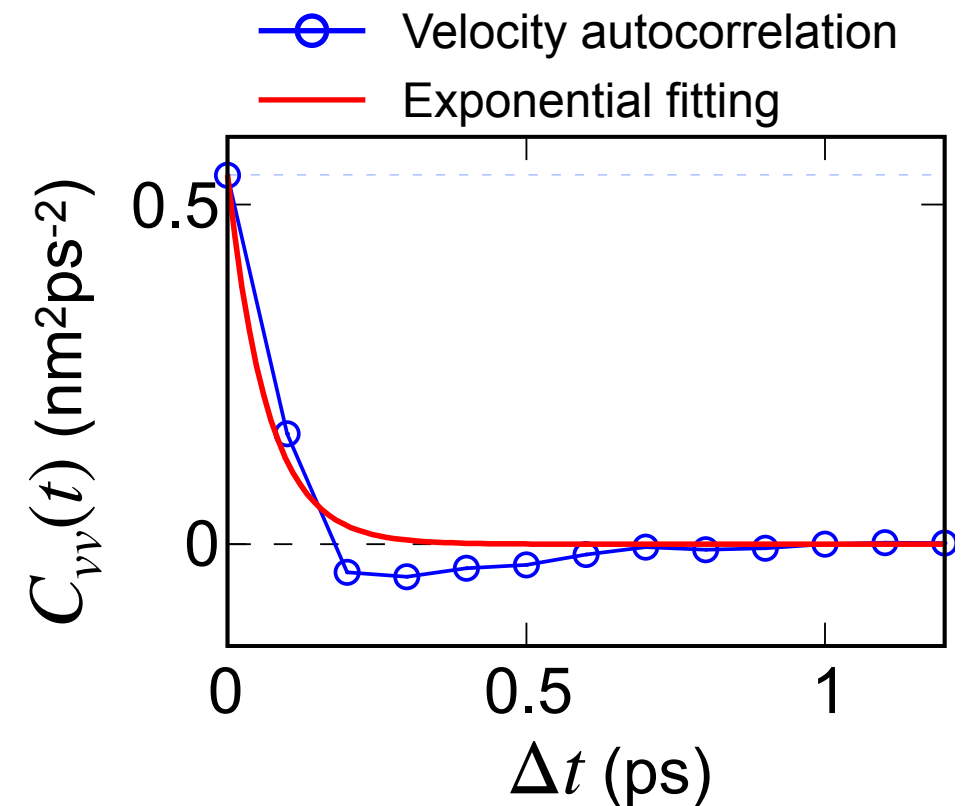
Notes

- The dissipation coefficient γ :

$$m \frac{d}{dt} \mathbf{v}(t) = -m\gamma \mathbf{v}(t) + \mathbf{F}(t)$$

$$C_{vv}(t) = \overline{\mathbf{v}(t) \cdot \mathbf{v}(0)} = C_{vv}(0)e^{-\gamma t}$$

Compute γ by exponential fitting.



- The Einstein relation of Langevin dynamics
(aka. the fluctuation-dissipation)

$$D = \frac{kT}{m\gamma} \quad \overline{\Delta \mathbf{x}^2} = 2Dt$$

Notes

- Compute γ with Boon-Yip's method:

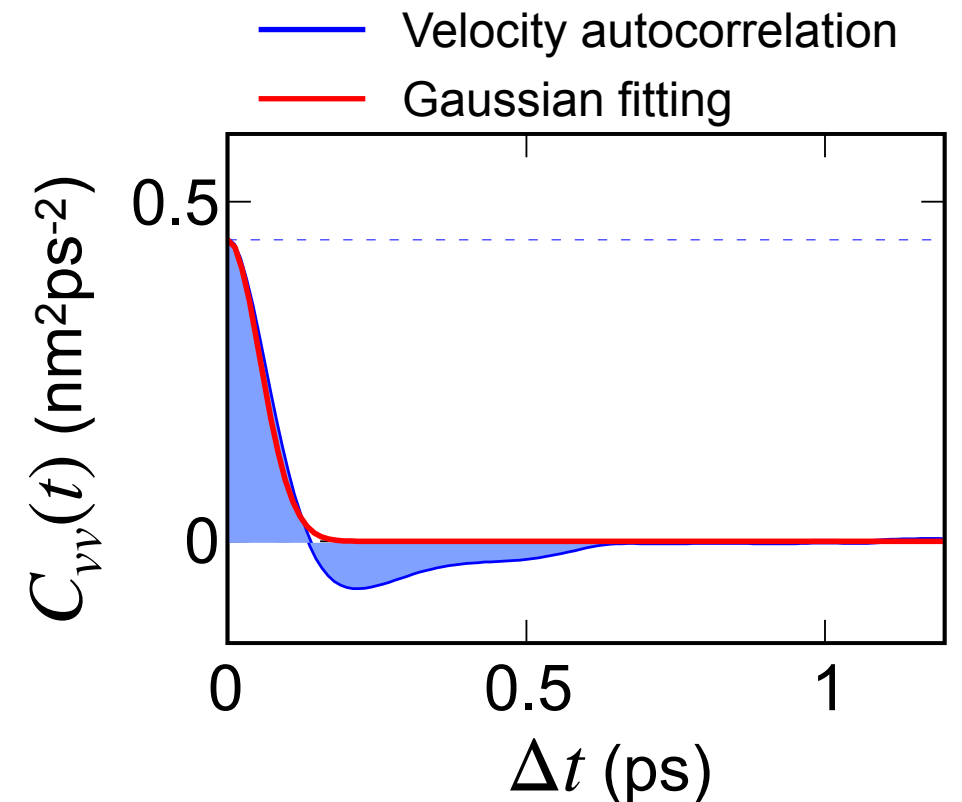
$$m \frac{d}{dt} \mathbf{v}(t) = -m\gamma \mathbf{v}(t) + \mathbf{F}(t)$$

$$C_{vv}(t) = \overline{\mathbf{v}(t) \cdot \mathbf{v}(0)} = C_{vv}(0)e^{-\gamma t}$$

$$\gamma = \int_0^\infty e^{-\gamma t'} dt' = C_{vv}(0)^{-1} \int_0^\infty C_{vv}(t) dt$$

- The Einstein relation of Langevin dynamics
(aka. the fluctuation-dissipation)

$$D = \frac{kT}{m\gamma} \quad \overline{\Delta \mathbf{x}^2} = 2Dt$$



Notes about trajectories

- Time unit: **picoseconds**; length unit: **nanometer**; velocity unit: **nm•ps⁻¹**
- File format of GRO:

Time stamp (unit: ps)

Velocities (vx, vy, vz)
unit: nm/ps

```
methane in water t= 0.00000 step= 0
5
1MOL      C1      1      1.084      0.171      1.599      -0.6902      -0.2804      0.1925
1MOL      H1      2      0.996      0.235      1.591      -0.3420      0.3593      1.2614
1MOL      H2      3      1.044      0.079      1.641      0.5320      -0.3079      1.3658
1MOL      H3      4      1.127      0.138      1.504      -1.2076      -0.2422      -0.0610
1MOL      H4      5      1.153      0.222      1.667      0.6313      -0.5384      -0.9108
2.99601      2.99601      2.99601
methane in water t= 0.01000 step= 5
5
1MOL      C1      1      1.076      0.167      1.602      -0.5842      -0.3240      0.3454
1MOL      H1      2      0.996      0.240      1.589      -0.1792      0.1755      0.5872
1MOL      H2      3      1.052      0.079      1.662      -0.0691      -0.0585      0.9648
1MOL      H3      4      1.114      0.151      1.501      -1.9272      0.2084      -0.2576
1MOL      H4      5      1.163      0.211      1.652      0.4377      -0.2529      -1.4121
2.99601      2.99601      2.99601
...
```

Dimensions of periodic box
unit: nm

Coordinates (x, y, z)
unit: nm

Notes about trajectories

- MD ensembles:
 - NPT: $T = 298\text{K}$, $p = 1\text{ bar}$, $V_{\text{box}} = 26.87 \pm 0.25\text{ nm}^3$
 - NVT: $T = 298\text{K}$, $V_{\text{box}} = 26.8924\text{ nm}^3$
 - NVE: $V_{\text{box}} = 26.8924\text{ nm}^3$
- Frames:
 - $dt = 10\text{fs}$, 20001 frames (in dt10fs/ folder)
 - $dt = 100\text{fs}$, 2001 frames (in dt100fs/ folder)