



## Supporting Information

### Modeling historical budget for $\beta$ -Hexachlorocyclohexane (HCH) in the Arctic Ocean: A contrast to $\alpha$ -HCH

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## Table

**Table S1** Parameter  $A_{AW}$ ,  $B_{AW}$ ,  $A_{OA}$ ,  $B_{OA}$ , and half-lives for  $\alpha$ -HCH and  $\beta$ -HCH

Chemical	$A_{AW}^a$	$B_{AW}^a$	$A_{OA}^b$	$B_{OA}^b$	$t_{1/2air}^c$	$t_{1/2water}^c$	$t_{1/2soil}^c$
$\alpha$ -HCH	10.2	-3990	-3.38	3230	1418	3500	14000
$\beta$ -HCH	8.58	-3950	-5.99	4390	2127	5250	21000

**Note:**

- a.  $A_{AW}$  and  $B_{AW}$  are used for temperature-dependent ( $T$ ) air-water partition coefficient calculation ( $K_{AW}$ ) as

$$\log K_{AW} = A_{AW} + B_{AW}/T$$

$A_{AW}$  and  $B_{AW}$  are calculated as:

$$B_{AW} = \Delta U_{AW}/\ln(10)R$$

$$A_{AW} = \log H_0/RT_0 - B_{AW}/T_0$$

where,  $\Delta U_{AW}$  is the internal energy of phase transfer between air and water in  $J\ mol^{-1}$ .

Abraham et al. estimated the value using the LEFRs method [1].  $R$  is gas constant,  $R = 8.314\ kg\ m^2\ s^{-2}\ K^{-1}\ mol^{-1}$ .  $H_0$  is Henry's law constant at a specific temperature [2],  $T_0$  is the specific temperature,  $T_0 = 273.15K$ .

- b.  $A_{OA}$  and  $B_{OA}$  are used for temperature-dependent ( $T$ ) octanol-air partition coefficient calculation ( $K_{OA}$ ) as

$$\log K_{OA} = A_{OA} + B_{OA}/T$$

$A_{OA}$  and  $B_{OA}$  are calculated as [3]:

$$B_{OA} = \Delta U_{OA}/\ln(10)R$$

$$A_{OA} = \log K_0 - B_{OA}/T_0$$

where,  $\Delta U_{OA}$  is the internal energy of phase transfer between octanol and air in  $J\ mol^{-1}$ .  $K_{OA0}$  is octanol-air partition coefficient at a specific temperature.

- c.  $t_{1/2air}$ ,  $t_{1/2water}$ ,  $t_{1/2soil}$  are half-lives for  $\alpha$ -HCH and  $\beta$ -HCH in air, water, and soil (h). considering half-life data of  $\beta$ -HCH is lacking. The common treatment is to relate the half-life of  $\beta$ -HCH to  $\alpha$ -HCH with a factor [4]. According to the measure of Bidleman et al. [5], the half-life of  $\beta$ -HCH is three times larger than  $\alpha$ -HCH's in lakes. However, the natural observed half-lives is not only related to the persistence to degradation in a phase but also related to the distribution among media because half-lives are different among media and chemical in different media can exchange with each other. For most chemicals (including HCHs), the half-life in water is longer than that in air. Because of their difference in  $K_{AW}$ , most of  $\alpha$ -HCH distribute in air but  $\beta$ -HCH in water [2, 6]. For

HCHs, half-life in water is more than two times larger than that in air [4]. Thus, we roughly assumed the factor between half-lives of  $\alpha$ -HCH and  $\beta$ -HCH is 1.5 ( $t_{1/2\_Air}=2127$  h,  $t_{1/2\_Ocean}=5250$  h,  $t_{1/2\_Soil}=21000$  h). The corresponding half-life of  $\alpha$ -HCH is collected from Wöhrnschimmel et al [4]. Degradation in snowpack and ice would not be considered in this model.

**Table S2** Parameters of Mass transfer coefficients (MTCs)

Description	Symbol	Value	Unit
MTC for air's transfer to stratosphere	$MTC_{Air\_Strat}$	0.064 [7]	m/h
Air side air-water MTC	$MTC_{AW}$	30 [7]	m/h
Water side air-water MTC	$MTC_{WA}$	0.03 [7]	m/h
Soil side air-soil MTC	$MTC_{AS}$	1 [7]	m/h
Soil side air phase air-soil MTC	$MTC_{SA1}$	0.04 [7]	m/h
Soil side water phase air-soil MTC	$MTC_{SA2}$	1E-5 [7]	m/h
Soil side solid phase air-soil MTC	$MTC_{SA3}$	4.54E-5 [7]	m/h
Air side air-snow MTC	$MTC_{Air-Snow}$	5 [8]	m/h
Air side air-ice MTC	$MTC_{Air-Ice}$	5 [8]	m/h
Snow side air-snow MTC	$MTC_{Snow-Air}$	50 <sup>a</sup>	m/h
Ice side air-ice MTC	$MTC_{Ice-Air}$	0.5 <sup>a</sup>	m/h
Rain rate in Source region	$MTC_{Rain\_Source}$	9.7E-5 [9]	m/h
Rain rate in Arctic region	$MTC_{Rain\_Arctic}$	1.25E-5 [10]	m/h
Snow rate in Arctic region	$MTC_{Snow\_Arctic}$	1.25E-5 [10]	m/h

**Note:**

- a.  $MTC_{Snow-Air}$  and  $MTC_{Ice-Air}$  are given by Darcy's law:

$$MTC_{S\&I\_Air} = -\frac{P}{\mu} \frac{p_0}{\ln 2 z_0} \times 3600$$

where  $P$  is snow and ice permeability,  $P_{Snowpack} = 5 \times 10^{-9} \text{ m}^2$  and  $P_{Ice} = 1 \times 10^{-9} \text{ m}^2$  [8].  $\mu$  is air viscosity. In the Arctic,  $\mu = 1.58 \times 10^{-5} \text{ Pa} \cdot \text{s}$  [8].  $z_0$  is roughness length.  $z_0 = A_{sr}/30$ . While  $A_{sr}$  is surface roughness amplitude, set to 0.1 m [8].  $p_0$  the amplitude of the pressure, Pa.

$$p_0 = \frac{3\rho_{Air}v^2A_{sr}}{\lambda}$$

where  $\rho_{Air}$  is the density of air,  $\rho_{Air} = 1.3 \text{ kg/m}^3$ .  $\lambda$  is the wavelength of the surface roughness,  $\lambda = 5 \text{ m}$  [8].

**Table S3** K values used in the AMBBM 2.0 model

Symbol	Formula
$K_{AW\_Media\_Region\_Season}^a$	$10^{A_{AW}+B_{AW}/T_{Media\_Region\_Season}}$
$K_{OA\_Media\_Region\_Season}^a$	$10^{A_{OA}+B_{OA}/T_{Media\_Region\_Season}}$
$K_{GP}^b$	$10^{-2.91} \rho_{Particle} f_{OM\_Particle} K_{OA\_Air}$
$K_{SA}^c$	$f_{OM\_Soil} K_{OA\_Air}$
$K_{Snowpack\_Air}^d$	$SA_{Snowpack} \rho_{Snowpack} K_{Ice\_Air\_dim}$
$K_{Snowflake\_Air}^d$	$SA_{Snowflake} \rho_{Snowflake} K_{Ice\_Air\_dim}$
$K_{Ice\_Air}^d$	$SA_{Ice} \rho_{Ice} K_{Ice\_Air\_dim}$

**Note:**

- Temperature-dependent air-water partition coefficient ( $K_{AW}$ ) and octanol-air partition coefficient ( $K_{OA}$ ) are the most basic and important.  $K_{AW\_Media\_Region\_Season}$  and  $K_{OA\_Media\_Region\_Season}$  are  $K_{AW}$  and  $K_{OA}$  values in the specific media (air or ocean) at specific region (Source or Arctic) in specific season (warm season or cold season if the region is the Arctic). In the following equations, placeholders like “Media”, “Region”, and “Season” will be omitted if no specific value is set.  $A_{AW}$ ,  $B_{AW}$ ,  $A_{OA}$ , and  $B_{OA}$  are unique parameters for each chemical’s  $K$  values calculation, given in Table S1.  $T$  refers to the temperature in the specific environment.
- Harner-Bidleman equation [11] was used to describe gas/particle partition coefficient (dimensionless).  $\rho_{Particle}$  is the density of particle, equals to 1500 kg/m<sup>3</sup>.  $f_{OM\_Particle}$  is fraction of the organic matter in the particle in volume, was set to 0.2.
- $K_{SA}$  is the partition coefficient between soil and gas [12].  $f_{OM\_Soil}$  is the fraction of organic matter in soil,  $f_{OM\_Soil} = 0.04$ .
- This is a highlight to include snow and ice into Arcite’s mutimedia model. The calculation process of snowflake-air, snowpack-air and ice-air partition coefficient is a bit more complicated. According to the work by Steinlin et al. [13]. These coefficients are calculated as the following:

$$K_{Snow\&Ice\_Air} = SA_{Snow\&Ice} \rho_{Snow\&Ice} K_{Ice\_Air\_dim}$$

$K_{Snow\&Ice\_Air}$  eliminate dimension of  $K_{Ice\_Air\_dim}$  through multiply by specific surface area and density of ice or snow. The  $SA_{S\&I}$  is the specific surface area of snowflake, snowpack, and sea ice. Based on their different structure,  $SA_{Snowflake} = 96 \text{ m}^2/\text{kg}$ ,  $SA_{Snowpack} = 46 \text{ m}^2/\text{kg}$ , and  $SA_{Sea\_ice} = 10 \text{ m}^2/\text{kg}$  [13, 14]. Be relevant to the  $SA$ , their densities are also different.  $\rho_{Snowflake} = 150 \text{ kg/m}^3$ ,  $\rho_{Snowpack} = 400 \text{ kg/m}^3$ , and  $\rho_{Sea\_Ice} = 900 \text{ kg/m}^3$ . The

$K_{\text{Ice\_Air\_dim}}$  is calculated as [15]:

$$K_{\text{Ice\_Air\_dim}} = K_{\text{Ice\_Air\_dim\_c}} \exp \left[ \frac{-\Delta H + RT_{\text{av}}}{R} \left( \frac{1}{T_{\text{Air\_Arctic\_Cold}}} - \frac{1}{T_{\text{cc}}} \right) \right]$$

where,  $K_{\text{S\&I\_Air\_dim\_c}}$  is  $K_{\text{S\&I\_Air\_dim}}$  at  $T_{\text{cc}}$  (-6.8°C) [15].  $T_{\text{av}}$  is the average temperature in the considered temperature range, which equals to  $T_{\text{Air\_Arctic\_Cold}}$ .  $R$  is gas constant,  $R = 8.314 \text{ kg m}^2 \text{ s}^{-2} \text{ K}^{-1} \text{ mol}^{-1}$ .  $\Delta H$  the enthalpy of sorption, which is also a function of  $K_{\text{S\&I\_Air\_dim\_c}}$  [15].  $K_{\text{S\&I\_Air\_dim\_c}}$  can be calculated through pp-LEFR as [16]:

$$K_{\text{S\&I\_Air\_dim\_c}} = 10^{0.639L + 3.53A + 3.38B - 6.85}$$

And,

$$\Delta H = -4.43 \ln K_{\text{Ice\_Air\_dim\_c}} - 92.4$$

where  $L$  is logarithm of hexadecane–air partition coefficient,  $A$  is Electron acceptor property, and  $B$  is electron donor property.

**Table S4**  $Z$  values used in the AMBBM 2.0 model

Symbol	Formula
$Z_{\text{Air}}$	$1/RT_{\text{Air}}$
$Z_{\text{Particle}}$	$K_{\text{GP}}Z_{\text{Air}}$
$Z_{\text{Water}}$	$Z_{\text{Air}}/K_{\text{AW\_Water}}$
$Z_{\text{Snowflake}}$	$Z_{\text{Air\_Arctic\_Cold}}K_{\text{Snowflake\_Air}}$
$Z_{\text{Droplets}}$	$Z_{\text{Air}}/K_{\text{AW\_Air}}$
$Z_{\text{Soil}}$	$Z_{\text{Air}}K_{\text{SA}}$
$Z_{\text{Snowflake}}$	$Z_{\text{Air\_Arctic\_Cold}}K_{\text{Snowflake\_Air}}$
$Z_{\text{Snowpack}}$	$Z_{\text{Air\_Arctic\_Cold}}K_{\text{Snowpack\_Air}}$
$Z_{\text{Sea\_Ice}}$	$Z_{\text{Air\_Arctic\_Cold}}K_{\text{Ice\_Air}}$

**Table S5**  $D$  values used in the AMBBM 2.0 model

Process	Symbol	Formula	Units
<i>Advection</i>			
Atmospheric inflow <sup>a</sup>	$D_{\text{Tran\_Air\_SA}}$	$\frac{v}{\sqrt{A_{\text{Arctic}}\pi/4}} V_{\text{Arctic\_Air}} Z_{\text{Air\_Arctic}} \cdot 24$	mol/(d·Pa)
Atmospheric outflow <sup>a</sup>	$D_{\text{Tran\_Air\_AS}}$	$\frac{A_{\text{Arctic}}}{A_{\text{Source}}} \frac{v}{\sqrt{A_{\text{Arctic}}\pi/4}} V_{\text{Source\_Air}} Z_{\text{Air}} \cdot 24$	mol/(d·Pa)
Current inflow <sup>b</sup>	$D_{\text{Tran\_Current\_SA}}$	$Flow_{\text{Current\_In}} Z_{\text{Water\_Source}} \cdot 24$	mol/(d·Pa)
Current outflow <sup>b</sup>	$D_{\text{Tran\_Current\_AS}}$	$Flow_{\text{Current\_Out}} Z_{\text{Water\_Arctic}} \cdot 24$	mol/(d·Pa)
River <sup>c</sup>	$D_{\text{Tran\_River\_SA}}$	$f_{\text{Runoff\_SA}} k_{\text{Soil\_Runoff\_Source}} V_{\text{Soil\_Source}} Z_{\text{Soil\_Source}} \cdot 24$	mol/(d·Pa)
Soil runoff <sup>c</sup>	$D_{\text{Runoff}}$	$k_{\text{Soil\_Runoff}} V_{\text{Soil}} Z_{\text{Soil}} \cdot 24$	mol/(d·Pa)
<i>Degradation and physical loss</i>			
Degradation in each phase <sup>d</sup>	$D_{\text{Deg}}$	$k_{\text{Deg}} V Z \cdot 24$	mol/(d·Pa)
Air transfer to the Stratosphere <sup>e</sup>	$D_{\text{Air\_Strat}}$	$A MTC_{\text{Air\_Strat}} Z_{\text{Air}} \cdot 24$	mol/(d·Pa)
Seawater mixing to deep ocean <sup>f</sup>	$D_{\text{Seawater\_sink}}$	$Flow_{\text{Sink}} Z_{\text{Water}} \cdot 24$	mol/(d·Pa)
<i>Gas deposition</i>			
Gas exchange with ocean <sup>g</sup>	$D_{\text{Air\_Water}}$	$\frac{f_{\text{Seawater}} A_{\text{Ocean}}}{1/(MTC_{\text{AW}} Z_{\text{Air}}) + 1/(MTC_{\text{WA}} Z_{\text{Water}})} \cdot 24$	mol/(d·Pa)
Gas exchange with soil <sup>g</sup>	$D_{\text{Air\_Soil}}$	$\frac{A_{\text{Land}}}{1/(MTC_{\text{AS}} Z_{\text{Air}}) + 1/(MTC_{\text{SA1}} Z_{\text{Air}} + MTC_{\text{SA2}} Z_{\text{Water}} + MTC_{\text{SA3}} Z_{\text{Soil}})} \cdot 24$	mol/(d·Pa)
Gas exchange with snowpack <sup>g</sup>	$D_{\text{Air\_Snowpack}}$	$\frac{A_{\text{Arctic\_Land}}}{1/(MTC_{\text{Air\_Snow}} Z_{\text{Air\_Arctic\_Cold}}) + 1/(MTC_{\text{Snow\_Air}} Z_{\text{Air\_Arctic\_Cold}})} \cdot 24$	mol/(d·Pa)



Gas exchange with sea ice <sup>g</sup>	$D_{\text{Air\_Sea\_Ice}}$	$\frac{(1 - f_{\text{Seawater}})A_{\text{Arctic\_Sea\_Ice}}}{1/(MTC_{\text{Air\_Ice}}Z_{\text{Air\_Arctic\_Cold}}) + 1/(MTC_{\text{Sea\_Air}}Z_{\text{Air\_Arctic\_Cold}})} \cdot 24$	mol/(d·Pa)
Rain scavenging	$D_{\text{Rain}}$	$MTC_{\text{Rain}}A Z_{\text{Droplets}} \cdot 24$	mol/(d·Pa)
Snow scavenging	$D_{\text{Snow}}$	$MTC_{\text{Snow}}A Z_{\text{Snowflake}} \cdot 24$	mol/(d·Pa)
<i>Particle deposition</i>			
Dry deposition <sup>h</sup>	$D_{\text{DD}}$	$\alpha MTC_{\text{Qd}} A \varphi_{\text{Particle\_Air}} Z_{\text{Particle}} \cdot 24$	mol/(d·Pa)
Wet deposition with rain <sup>h</sup>	$D_{\text{WD\_Rain}}$	$\alpha MTC_{\text{Rain}} A Q_{\text{Rain}} \varphi_{\text{Particle\_Air}} Z_{\text{Particle}} \cdot 24$	mol/(d·Pa)
Wet deposition with snow <sup>h</sup>	$D_{\text{WD\_Snow}}$	$\alpha MTC_{\text{Snow}} A Q_{\text{Snow}} \varphi_{\text{Particle\_Air}} Z_{\text{Particle}} \cdot 24$	mol/(d·Pa)

**Note:**

- a. According to the residence time calculation equation of air used in the Simplebox 4.0 model [17], the  $D$  values are calculated.  $A_{\text{Arctic}}$  and  $A_{\text{Source}}$  are area of the Arctic and the source area,  $A_{\text{Arctic}} = 2.13 \times 10^{13} \text{ m}^2$ ,  $A_{\text{Source}} = 2.40 \times 10^{14} \text{ m}^2$ .  $V_{\text{Arctic\_Air}}$  and  $V_{\text{Source\_Air}}$  are air volumn in the Arctic and source zone, where the height of the atmopshere is set to 6000 m.  $v$  is the wind speed of atmospheric transport, set to 8 m/s [8].
- b. Ocean currents' transport has been discussed by Macdonald et al. [18]. The inflow discharge of surface ocean is  $4.86 \times 10^{13} \text{ km}^3/\text{y}$ , outflow flux is  $2.86 \times 10^{13} \text{ km}^3/\text{y}$ . The rest sinks to deeper water.
- c. Rivers' budget comes from soil runoff in source area and the Arctic area, respectively.  $D_{\text{Tran\_River\_MA}}$  refers to river inflow from the source area.  $k_{\text{Soil\_Runoff\_Source}}$  is soil runoff rate in the source, which was set to  $2 \times 10^{-5} \text{ h}^{-1}$  [17].  $f_{\text{Runoff\_MA}}$  is the fraction of soil runoff in source to the Arctic. According to Li et al. [19], we assumed 3 % of soil runoff in the source finally enter the AO and the rest flows into oceans in the source zone. However, runoff process only happened in the warm season in the Arctic, as in the cold season, soil is reckoned to be frozen. And  $k_{\text{Soil\_Runoff\_Arctic}}$  is much smaller, which was modified by rain rate in regions ( $MTC_{\text{Rain\_Arctic}}$  and  $MTC_{\text{Rain\_Source}}$ ) as:

$$k_{\text{Soil\_Runoff\_Arctic}} = k_{\text{Soil\_Runoff\_Arctic}} \cdot \frac{MTC_{\text{Rain\_Arctic}}}{MTC_{\text{Rain\_Source}}}$$

Rain rate in the source zone and the Arctic zone are  $9.7 \times 10^{-5} \text{ m/h}$  [9] and  $1.2 \times 10^{-5} \text{ m/h}$  [10].

- d. Besides atmospheric circulation and ocean current, chemicals are mostly removed through degradation in phases, air's exchange with the stratosphere, and ocean's exchange with deep ocean. Degradation rate ( $k_{\text{Deg}}$ ) was calculated from half-lives in different phases and modefied by temperature.

$$k_{\text{Deg}} = \frac{\ln 2}{t_{1/2}} e^{E \left( \frac{1}{298.15 R} - \frac{1}{T} \right)}$$

$t_{1/2}$  is half-live in different media at 298.15K, given in Table S1,  $E$  is activation energy for degradation process in different phases. Wöhrnschimmel et al [4] estimated  $E$  value of  $\alpha$ -HCH and  $\beta$ -HCH and accepted in this work ( $E_{\text{Air}} = 1.12 \times 10^4 \text{ J}$ ,  $E_{\text{Ocean}} = 1.073 \times 10^5 \text{ J}$ ,  $E_{\text{Soil}} = 3 \times 10^4 \text{ J}$ ).

- e. Air will transport to upper air above 6000 m. By habit, it was named transfer to stratosphere.
- f. Surface sea water also can mix with deep water. According to the Macdonald et al.[18], the sinking flux is  $2 \times 10^{13} \text{ km}^3/\text{y}$ .

- g. Gas exchange between air with soil, sea water, snowpack, and sea ice are calculated by the two-film theory widely used in fugacity method[9].  $f_{\text{Seawater}}$  is the bare sea water area fraction in the AO and in source there is no sea ice.  $f_{\text{Seawater}} = 0.6$  in Arctic warm season and  $f_{\text{Seawater}} = 0.7$  in Arctic cold season.  $MTC$  value are given in Table S2.
- h. Li et al. proposed the steady-state equation for G/P partition calculation [20]. This method has been proved to be more accurate for OCPs' stimulation [21] and successfully used in models [22, 23]. Based on the equilibrium gas/particle partition theory, a parameter  $\alpha$  was added to correct environmental process about the particle in the air.  $\alpha$  is calculated as [20]:

$$\alpha = 1 / (1 + 4.18 \times 10^{-11} f_{\text{OM\_Particle\_Air}} K_{\text{OA}})$$

$f_{\text{OM\_Particle}}$  is the fraction of organic matter in particle, which was set to 0.2 [24].

$\phi_{\text{Particle\_Air}}$  is the volumn fraction of particle in the air. It can be calculated by total suspend particle (TSP,  $\mu\text{g}/\text{m}^3$ ).

$$TSP = 10^{-9} \frac{TSP}{\rho_{\text{Particle}}}$$

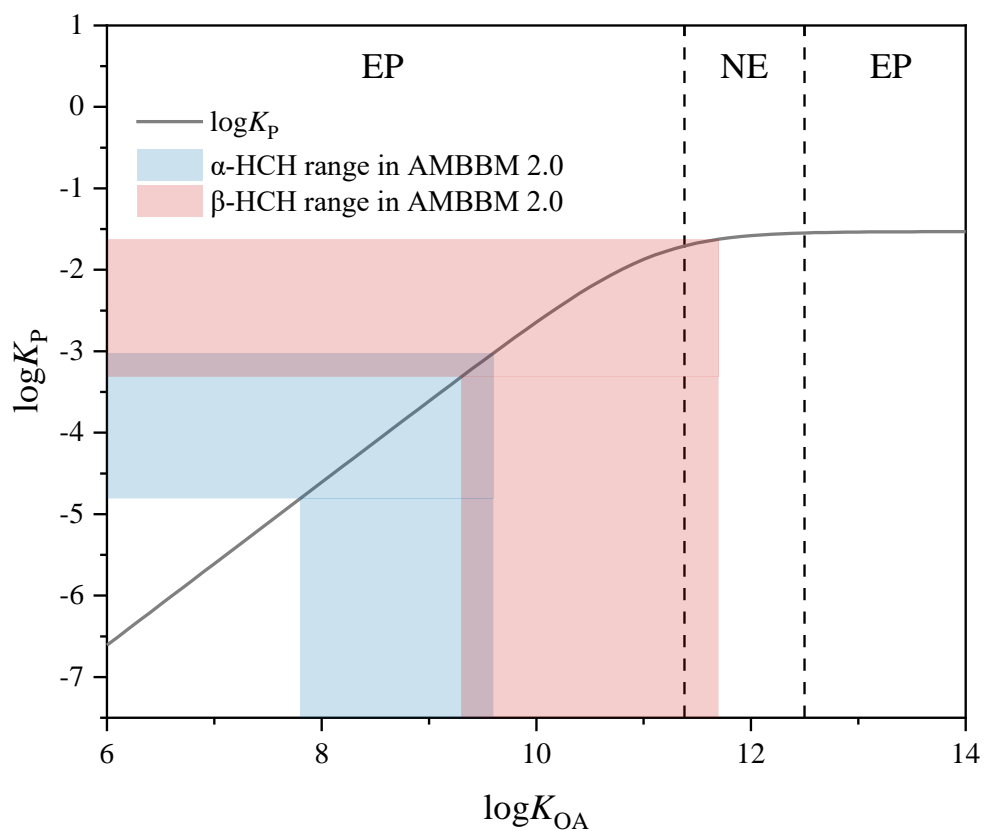
The atmosphere in the Arcitc is clear, the total suspend particle in the Arctic ( $TSP_{\text{Arctic}}$ ) is set to 1  $\mu\text{g}/\text{m}$  [25]. Correspondingly,  $TSP_{\text{Source}}$  is set to 30  $\mu\text{g}/\text{m}^3$ .  $Q_{\text{Rain}}$  and  $Q_{\text{Snow}}$  are scavenging ratio of rain and snow,  $Q_{\text{Rain}} = 200000$  [9] and  $Q_{\text{Snow}} = 320000$  [26].

**Table S6** Dispersion factors of properties

Properties	dispersion factors
$K_{AW}$	5
$K_{OA}$	5
$HL_{Air}$	10
$HL_{Water}$	10
$HL_{Soil}$	10

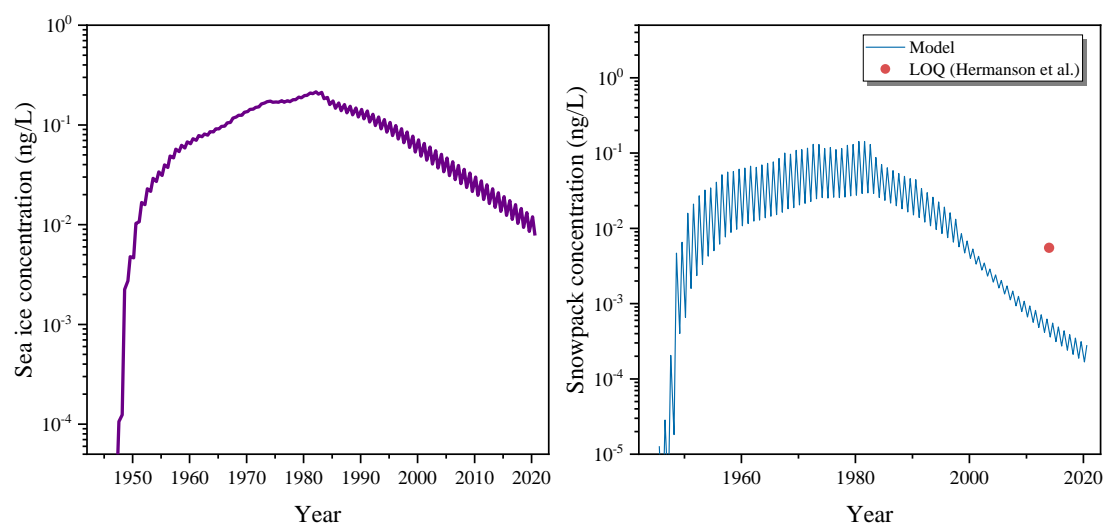
## Figure

**Figure S1** Gas/partition of  $\alpha$ -HCH and  $\beta$ -HCH in AMBBM 2.0



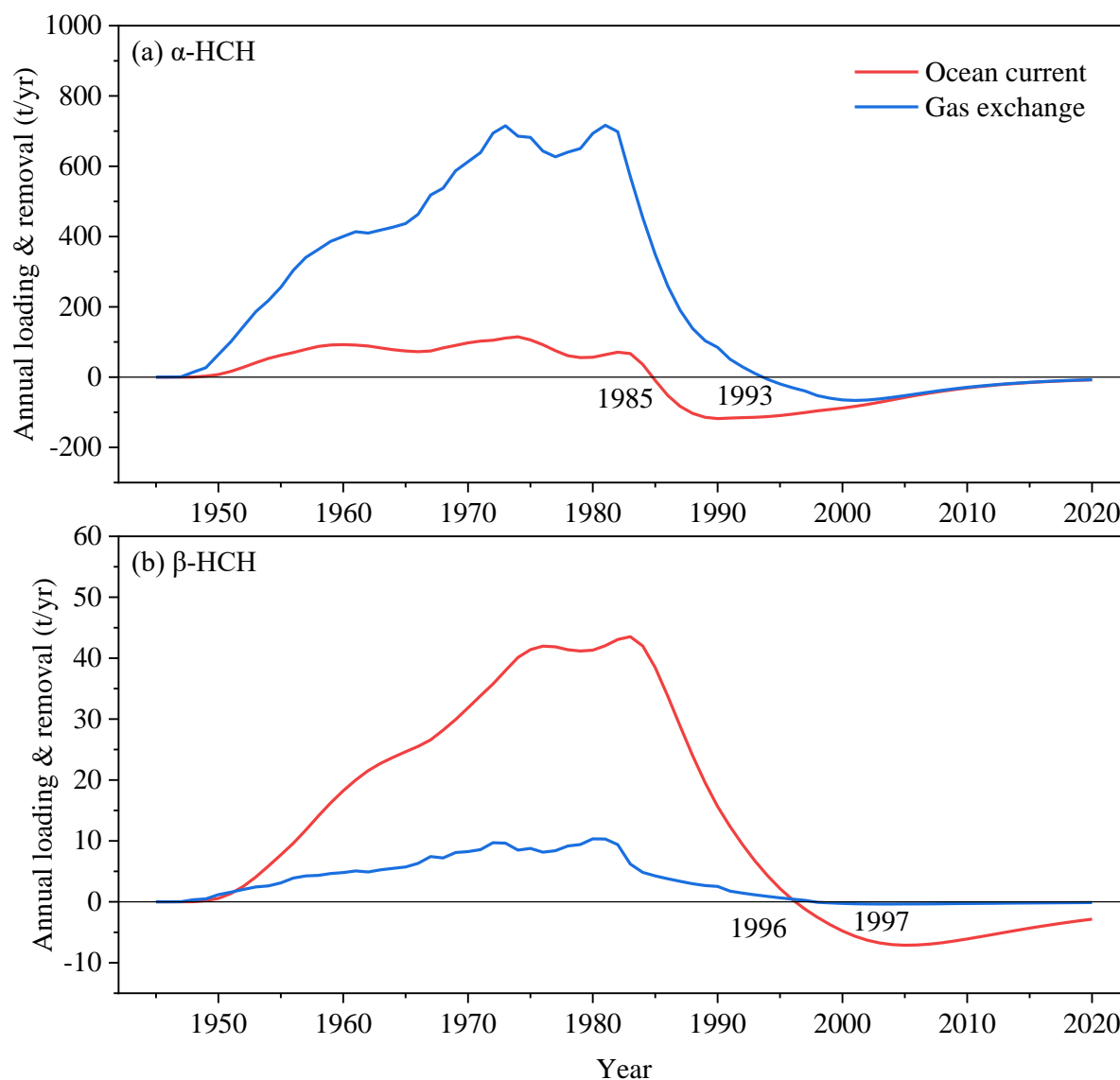
This figure depicts the possible  $\log K_{OA}$  and  $\log K_p$  range of  $\alpha$ -HCH and  $\beta$ -HCH in the model, which indicates that the gas/particle partition of  $\alpha$ -HCH and  $\beta$ -HCH are basically both failed in the EP domain [24]. Thus, whether equilibrium or steady state assumption used in the model won't raise a huge influence on the result.

**Figure S2** Concentration of  $\beta$ -HCH in snowpack and sea ice



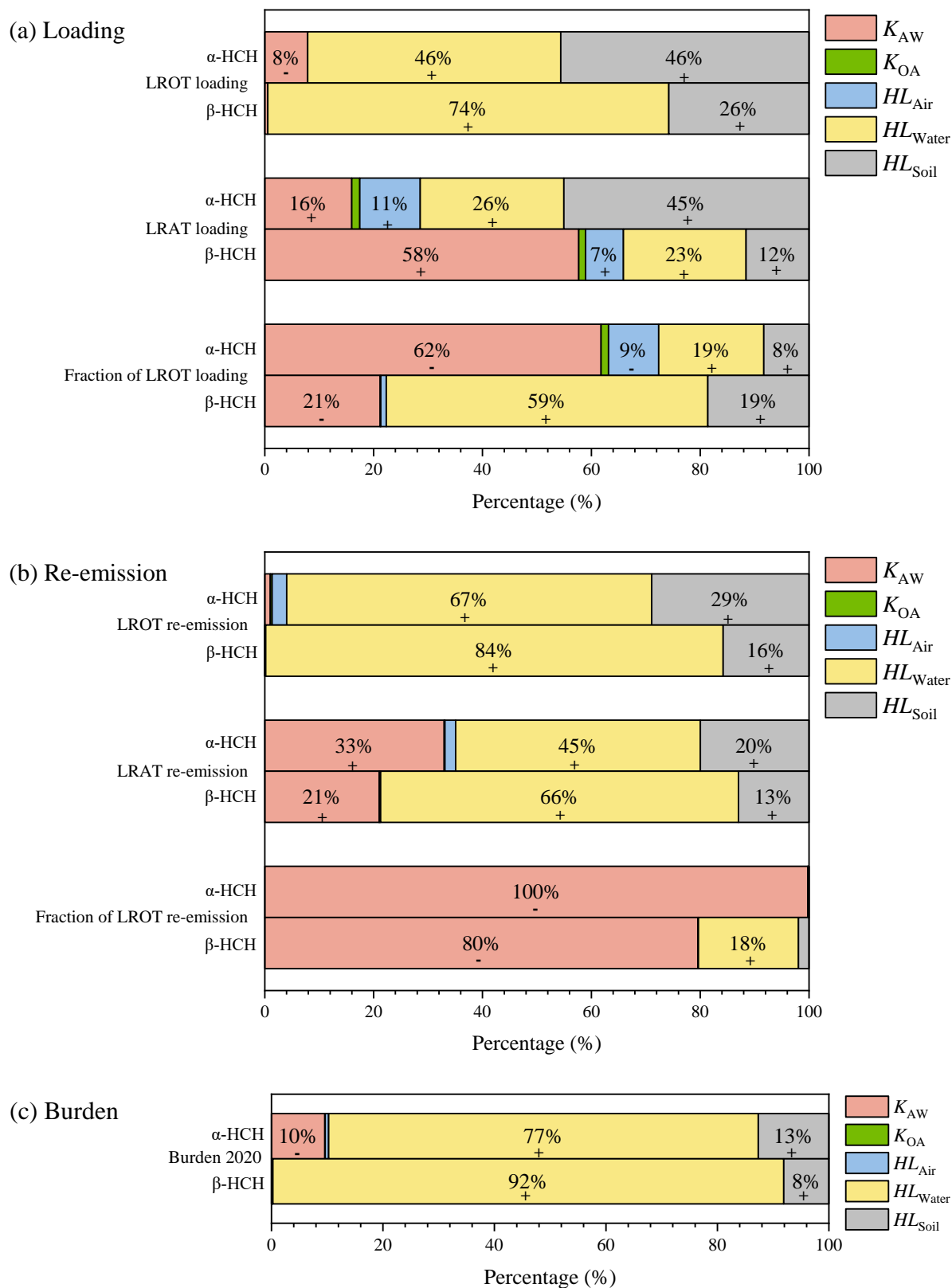
This figure depicts modeling results of sea ice and snowpack. The content of the figure is discussed in the section 4.1 in the manuscript.

**Figure S3** “Switching point” of  $\alpha$ -HCH and  $\beta$ -HCH



This figure depicts the yearly flux of ocean current and gas exchange. The top figure is  $\alpha$ -HCH and the bottom figure is  $\beta$ -HCH. The red lines represent ocean current and the blue lines represent gas exchange.  $\alpha$ -HCH’s “switching point” of ocean current and gas-water exchange happened in 1985 and 1993. But  $\beta$ -HCH’s “switching point” of them happened in 1996 and 1997.

**Figure S4** Result of sensitivity analysis



This figure depicts the variance contribution of physical-chemical properties to HCHs' behaviors. The bars are depicted paired. The upper bar is α-HCH and the lower bar is β-HCH.



Positive or negative correlation is also marked under the percentage value as “+” or “-”. But properties whose contribution less than 5% won’t be marked.

(a) Loading

From top to bottom are the sum of HCHs entering the Arctic Ocean through LROT, LRAT, and the fraction of direct pathway.

(b) Re-emission

From top to bottom are the sum of HCHs’ re-emission in the Arctic Ocean through LROT, LRAT, and the fraction of direct pathway.

(c) Burden

From top to bottom are chemical’s burden in 2020.

# Text

## Text S1 Mass balance equations of this model

### Source Zone

$$\begin{aligned}\frac{dM_{Air\_S}}{dt} &= - \left( \frac{D_{Tran\_Air\_SA} + D_{Deg\_Air\_S} + D_{Air\_Seawater\_S}}{+D_{Air\_Soil\_S} + D_{Rain\_S} + D_{DD\_S} + D_{WD\_S} + D_{Air\_Strat\_S}} \right) \frac{M_{Air\_S}}{V_{Air\_S}Z_{Air\_S}} \\ &\quad + D_{Air\_Water} \frac{M_{Seawater\_S}}{V_{Seawater\_S}Z_{Seawater\_S}} + D_{Air\_Soil} \frac{M_{Soil\_S}}{V_{Soil\_S}Z_{Soil\_S}} + E_{Air} \\ \frac{dM_{Seawater\_S}}{dt} &= - \left( \frac{D_{Tran\_Seawater\_SA} + D_{Deg\_Seawater\_S}}{+D_{Seawater\_Sink\_S} + D_{Air\_Water\_S}} \right) \frac{M_{Seawater\_S}}{V_{Seawater\_S}Z_{Seawater\_S}} \\ &\quad + \left( \frac{D_{Rain\_Seawater\_S} + D_{DD\_Seawater\_S}}{+D_{Air\_Water\_S} + D_{WD\_Seawater\_S}} \right) \frac{M_{Air\_S}}{V_{Air\_S}Z_{Air\_S}} \\ &\quad + k_{Runoff} D_{Runoff\_S} \frac{M_{Soil\_S}}{V_{Soil\_S}Z_{Soil\_S}} + E_{Seawater} \\ \frac{dM_{Soil\_S}}{dt} &= -(D_{Deg\_Soil\_S} + D_{Runoff\_S} + D_{Air\_Soil\_S}) \frac{M_{Soil\_S}}{V_{Soil\_S}Z_{Soil\_S}} \\ &\quad + (D_{Rain\_Soil\_S} + D_{DD\_Soil\_S} + D_{Air\_Soil\_S} + D_{WD\_Soil\_S}) \frac{M_{Air\_S}}{V_{Air\_S}Z_{Air\_S}} + E_{Soil}\end{aligned}$$

### Arctic Zone

$$\begin{aligned}\frac{dM_{Air\_A}}{dt} &= - \left( \frac{D_{Tran\_Air\_AS} + D_{Deg\_Air\_A} + D_{Air\_Strat\_A} + D_{Air\_Soil\_A}}{+D_{Air\_Snowpack\_A} + D_{Air\_Sea\_Ice\_A} + D_{Rain\_A}} \right) \frac{M_{Air\_A}}{V_{Air\_A}Z_{Air\_A}} \\ &\quad + D_{Tran\_Air\_SA} \frac{M_{Air\_S}}{V_{Air\_S}Z_{Air\_S}} + D_{Air\_Water} \frac{M_{Seawater\_A}}{V_{Seawater\_S}Z_{Seawater\_S}} + D_{Air\_Soil} \frac{M_{Soil\_A}}{V_{Soil\_S}Z_{Soil\_S}} \\ &\quad + D_{Air\_Snowpack} \frac{M_{Snowpack\_A}}{V_{Snowpack\_A}Z_{Snowpack\_A}} + D_{Air\_Sea\_Ice} \frac{M_{Sea\_Ice\_A}}{V_{Sea\_Ice\_A}Z_{Sea\_Ice\_A}} \\ \frac{dM_{Seawater\_A}}{dt} &= - \left( \frac{D_{Tran\_Seawater\_AS} + D_{Deg\_Seawater\_A}}{+D_{Seawater\_sink\_A} + D_{Air\_Water\_A}} \right) \frac{M_{Seawater\_A}}{V_{Seawater\_A}Z_{Seawater\_A}} \\ &\quad + \left( \frac{D_{Rain\_Seawater\_A} + D_{Snow\_Seawater\_A}}{+D_{DD\_Seawater\_A} + D_{Air\_Water\_A} + D_{WD\_Seawater\_A}} \right) \frac{M_{Air\_A}}{V_{Air\_A}Z_{Air\_A}} \\ &\quad + (1 - k_{Runoff}) D_{Runoff\_S} \frac{M_{Soil\_S}}{V_{Soil\_S}Z_{Soil\_S}} \\ &\quad + D_{Tran\_Seawater\_SA} \frac{M_{Seawater\_S}}{V_{Seawater\_S}Z_{Seawater\_S}} + D_{Runoff\_A} \frac{M_{Soil\_A}}{V_{Soil\_S}Z_{Soil\_A}} \\ \frac{dM_{Soil\_A}}{dt} &= -(D_{Deg\_Soil\_A} + D_{Runoff\_A} + D_{Air\_Soil\_A}) \frac{M_{Soil\_A}}{V_{Soil\_A}Z_{Soil\_A}} \\ &\quad + \left( \frac{D_{Rain\_Soil\_A} + D_{Snow\_Soil\_A}}{D_{Air\_Soil\_A} + D_{WD\_Soil\_A} + D_{DD\_Soil\_A}} \right) \frac{M_{Air\_A}}{V_{Air\_A}Z_{Air\_A}} \\ \frac{dM_{Sea\_Ice\_A}}{dt} &= -(D_{Deg\_Sea\_Ice\_A} + D_{Air\_Sea\_Ice\_A}) \frac{M_{Sea\_Ice\_A}}{V_{Sea\_Ice\_A}Z_{Sea\_Ice\_A}} \\ &\quad + \left( \frac{D_{Rain\_Sea\_Ice\_A} + D_{Sea\_Ice\_A}}{D_{Air\_Sea\_Ice\_A} + D_{WD\_Sea\_Ice\_A} + D_{DD\_Sea\_Ice\_A}} \right) \frac{M_{Air\_A}}{V_{Air\_A}Z_{Air\_A}} \\ \frac{dM_{Snowpack\_A}}{dt} &= -(D_{Deg\_Snowpack\_A} + D_{Air\_Snowpack\_A}) \frac{M_{Snowpack\_A}}{V_{Snowpack\_A}Z_{Snowpack\_A}} \\ &\quad + \left( \frac{D_{Rain\_Snowpack\_A} + D_{Snow\_Snowpack\_A}}{D_{Air\_Snowpack\_A} + D_{WD\_Snowpack\_A} + D_{DD\_Snowpack\_A}} \right) \frac{M_{Air\_A}}{V_{Air\_A}Z_{Air\_A}}\end{aligned}$$

This model is calculated by ordinary differential equations printed here. Each equation correspond to a compartment in a environment combines fugacity method and advection

method. As mentioned in the section 2.2 of the manuscript, there are 3 compartments in the Source Zone and 5 compartments the Arctic. From top to bottom are equations about air in the Source Zone, seawater in Source Zone, soil in the Source Zone, air in the Arctic Zone, seawater in Arctic Zone, soil in the Arctic Zone, sea ice in the Arctic Zone, and snowpack in Arctic Zone.

## Reference

- [1] M. H. Abraham, K. Enomoto, E. D. Clarke and G. Sexton. Hydrogen Bond Basicity of the Chlorogroup; Hexachlorocyclohexanes as Strong Hydrogen Bond Bases. *The Journal of Organic Chemistry* 67(14) (2002) 4782-4786.
- [2] H. Xiao, N. Q. Li and F. Wania. Compilation, evaluation, and selection of physical-chemical property data for alpha-, beta-, and gamma-hexachlorocyclohexane. *Journal of Chemical and Engineering Data* 49(2) (2004) 173-185.
- [3] M. Yang, Y.-F. Li, L.-N. Qiao and X.-M. Zhang. Estimating subcooled liquid vapor pressures and octanol-air partition coefficients of polybrominated diphenyl ethers and their temperature dependence. *Science of the Total Environment* 628-629 (2018) 329-337.
- [4] H. Wöhrnschimmel, P. Tay, H. von Waldow, H. Hung, Y.-F. Li, M. MacLeod and K. Hungerbühler. Comparative Assessment of the Global Fate of alpha- and beta-Hexachlorocyclohexane before and after Phase-Out. *Environmental Science & Technology* 46(4) (2012) 2047-2054.
- [5] T. F. Bidleman, S. Backus, A. Dove, R. Lohmann, D. Muir, C. Teixeira and L. Jantunen. Lake Superior Has Lost over 90% of Its Pesticide HCH Load since 1986. *Environmental Science & Technology* 55(14) (2021) 9518-9526.
- [6] Y. F. Li, R. W. Macdonald, L. M. M. Jantunen, T. Harner, T. F. Bidleman and W. M. J. Strachan. The transport of beta-hexachlorocyclohexane to the western Arctic Ocean: a contrast to alpha-HCH. *Science of the Total Environment* 291(1-3) (2002) 229-246.
- [7] F. Wegmann, L. Cavin, M. MacLeod, M. Scheringer and K. Hungerbühler. The OECD software tool for screening chemicals for persistence and long-range transport potential. *Environmental Modelling & Software* 24(2) (2009) 228-237.
- [8] L. J. Thibodeaux and D. Mackay, *Handbook of chemical mass transport in the environment*, CRC Press 2010.
- [9] J. M. Parnis and D. Mackay, *Multimedia environmental models: the fugacity approach*, CRC press 2020.
- [10] R. Bintanja. The impact of Arctic warming on increased rainfall. *Sci Rep* 8(1) (2018) 16001-16001.
- [11] T. Harner and T. F. Bidleman. Octanol–Air Partition Coefficient for Describing Particle/Gas Partitioning of Aromatic Compounds in Urban Air. *Environmental Science & Technology* 32(10) (1998) 1494-1502.
- [12] Y.-F. Li, T. Harner, L. Liu, Z. Zhang, N.-Q. Ren, H. Jia, J. Ma and E. Sverko. Polychlorinated Biphenyls in Global Air and Surface Soil: Distributions, Air–Soil Exchange, and Fractionation Effect. *Environmental Science & Technology* 44(8) (2010) 2784-2790.
- [13] C. Steinlin, C. Bogdal, P. A. Pavlova, M. Schwikowski, M. P. Lüthi, M. Scheringer, P. Schmid and K. Hungerbühler. Polychlorinated Biphenyls in a Temperate Alpine Glacier: 2. Model Results of Chemical Fate Processes. *Environmental Science & Technology* 49(24) (2015) 14092-14100.
- [14] F. Domine, A.-S. Taillandier and W. R. Simpson. A parameterization of the specific surface area of seasonal snow for field use and for models of snowpack evolution. *Journal of Geophysical Research* 112(F02031) (2007) 1 à 13 pages.
- [15] K.-U. Goss and R. P. Schwarzenbach. Empirical Prediction of Heats of Vaporization and Heats of Adsorption of Organic Compounds. *Environmental Science & Technology* 33(19) (1999) 3390-3393.
- [16] C. M. Roth, K.-U. Goss and R. P. Schwarzenbach. Sorption of Diverse Organic Vapors to Snow. *Environmental Science & Technology* 38(15) (2004) 4078-4084.
- [17] A. Hollander, M. Schoorl and D. van de Meent. SimpleBox 4.0: Improving the model while keeping it simple.... *Chemosphere* 148 (2016) 99-107.

- [18] R. W. Macdonald, L. A. Barrie, T. F. Bidleman, M. L. Diamond, D. J. Gregor, R. G. Semkin, W. M. J. Strachan, Y. F. Li, F. Wania, M. Alaee, L. B. Alexeeva, S. M. Backus, R. Bailey, J. M. Bewers, C. Gobeil, C. J. Halsall, T. Harner, J. T. Hoff, L. M. M. Jantunen, W. L. Lockhart, D. Mackay, D. C. G. Muir, J. Pudykiewicz, K. J. Reimer, J. N. Smith, G. A. Stern, W. H. Schroeder, R. Wagemann and M. B. Yunker. Contaminants in the Canadian Arctic: 5 years of progress in understanding sources, occurrence and pathways. *Science of The Total Environment* 254(2) (2000) 93-234.
- [19] Y. F. Li, R. W. Macdonald, J. M. Ma, H. Hung and S. Venkatesh. Historical alpha-HCH budget in the Arctic Ocean: the Arctic Mass Balance Box Model (AMBBM). *Science of the Total Environment* 324(1-3) (2004) 115-139.
- [20] Y. F. Li, W. L. Ma and M. Yang. Prediction of gas/particle partitioning of polybrominated diphenyl ethers (PBDEs) in global air: A theoretical study. *Atmos. Chem. Phys.* 15(4) (2015) 1669-1681.
- [21] L.-n. Qiao, Z.-f. Zhang, L.-y. Liu, W.-w. Song, W.-l. Ma, N.-z. Zhu and Y.-f. Li. Measurement and modeling the gas/particle partitioning of organochlorine pesticides (OCPs) in atmosphere at low temperatures. *Science of The Total Environment* 667 (2019) 318-324.
- [22] Y.-F. Li, M. Qin, P.-F. Yang, S. Hao and R. W. Macdonald. Particle/gas partitioning for semi-volatile organic compounds (SVOCs) in Level III multimedia fugacity models: Gaseous emissions. *Science of The Total Environment* 795 (2021) 148729.
- [23] M. Qin, P.-F. Yang, P.-T. Hu, S. Hao, R. W. Macdonald and Y.-F. Li. Particle/gas partitioning for semi-volatile organic compounds (SVOCs) in level III multimedia fugacity models: Both gaseous and particulate emissions. *Science of The Total Environment* 790 (2021) 148012.
- [24] Y.-F. Li, M. Qin, P.-F. Yang, L.-Y. Liu, L.-J. Zhou, J.-N. Liu, L.-L. Shi, L.-N. Qiao, P.-T. Hu, C.-G. Tian, A. Nikolaev and R. Macdonald. Treatment of particle/gas partitioning using level III fugacity models in a six-compartment system. *Chemosphere* 271 (2021) 129580.
- [25] Q. T. Nguyen, T. B. Kristensen, A. M. K. Hansen, H. Skov, R. Bossi, A. Massling, L. L. Sørensen, M. Bilde, M. Glasius and J. K. Nøjgaard. Characterization of humic-like substances in Arctic aerosols. *Journal of Geophysical Research: Atmospheres* 119(8) (2014) 5011-5027.
- [26] L. Zhang, I. Cheng, D. Muir and J. P. Charland. Scavenging ratios of polycyclic aromatic compounds in rain and snow in the Athabasca oil sands region. *Atmos. Chem. Phys.* 15(3) (2015) 1421-1434.