Thermal Runaway in an Exothermic Reaction

The temperature above ambient U(x, t) at position x and time t in a body undergoing an exothermic reaction satisfies the boundary value problem

$$\frac{\partial U}{\partial t} = k \frac{\partial^2 U}{\partial x^2} + Q e^{bU} \quad \text{for } x, t > 0$$

Here *k* is the thermal diffusivity and *Q*, *b* control the rate at which the reaction is producing heat.

If it is assumed the solution of this equation either tends to infinity (thermal runaway) or reaches a steady state (equilibrium), then the existence of a steady state guarantees the temperature U(x,t) stays finite. At a steady state U does not depend on t so the above equation becomes

$$kU'' + Qe^{bU} = 0$$

to be solved for U(x). Multiplying by U' and integrating with respect to x gives

$$\frac{1}{2}kU'^2 + \frac{Q}{h}e^{bU} = C$$

where C is a constant. Find the solution of this first order separable equation.