



## Short communication

# Kinetic properties of human placental glucose-6-phosphate dehydrogenase

Nazmi Özer \*, Yasemin Aksoy, I. Hamdi Ögüs

*Department of Biochemistry, Faculty of Medicine, Hacettepe University, Ankara 06100, Turkey*

Received 2 November 2000; received in revised form 8 January 2001; accepted 8 January 2001

---

**Abstract**

The kinetic properties of placental glucose-6-phosphate dehydrogenase were studied, since this enzyme is expected to be an important component of the placental protection system. In this capacity it is also very important for the health of the fetus. The placental enzyme obeyed “Rapid Equilibrium Ordered Bi Bi” sequential kinetics with  $K_m$  values of  $40 \pm 8 \mu\text{M}$  for glucose-6-phosphate and  $20 \pm 10 \mu\text{M}$  for NADP. Glucose-6-phosphate, 2-deoxyglucose-6-phosphate and galactose-6-phosphate were used with catalytic efficiencies ( $k_{\text{cat}}/K_m$ ) of  $7.4 \times 10^6$ ,  $4.89 \times 10^4$  and  $1.57 \times 10^4 \text{ M}^{-1}\text{s}^{-1}$ , respectively. The  $K_{m,\text{app}}$  values for galactose-6-phosphate and for 2-deoxyglucose-6-phosphate were  $10 \pm 2$  and  $0.87 \pm 0.06 \text{ mM}$ . With galactose-6-phosphate as substrate, the same  $K_m$  value for NADP as glucose-6-phosphate was obtained and it was independent of galactose-6-phosphate concentration. On the other hand, when 2-deoxyglucose-6-phosphate used as substrate, the  $K_m$  for NADP decreased from  $30 \pm 6$  to  $10 \pm 2 \mu\text{M}$  as the substrate concentration was increased from 0.3 to 1.5 mM. Deamino-NADP, but not NAD, was a coenzyme for placental glucose-6-phosphate dehydrogenase. The catalytic efficiencies of NADP and deamino-NADP (glucose-6-phosphate as substrate) were  $1.48 \times 10^7$  and  $4.80 \times 10^6 \text{ M}^{-1}\text{s}^{-1}$ , respectively. With both coenzymes, a hyperbolic saturation and an inhibition above 300  $\mu\text{M}$  coenzyme concentration, was observed. Human placental glucose-6-phosphate dehydrogenase was inhibited competitively by 2,3-diphosphoglycerate ( $K_i = 15 \pm 3 \text{ mM}$ ) and NADPH ( $K_i = 17.1 \pm 3.2 \mu\text{M}$ ). The small dissociation constant for the G6PD:NADPH complex pointed to tight enzyme:NADPH binding and the important role of NADPH in the regulation of the pentose phosphate pathway. © 2001 Elsevier Science Ltd. All rights reserved.

**Keywords:** Human placenta; Glucose-6-phosphate dehydrogenase; Substrate and coenzyme specificity; Inhibited by 2,3DPGA and NADPH

---

**Abbreviations:** Deamino-NADP, deamino nicotinamide adenine dinucleotide phosphate (oxidized form); 2-dG6P, 2-deoxyglucose-6-phosphate; 2,3-DPGA, 2,3-diphosphoglyceric acid; Gal6P, galactose-6-phosphate; G6PD, glucose-6-phosphate dehydrogenase; G6P, glucose-6-phosphate; GSH, glutathione (reduced form); GSSG, glutathione disulfide (oxidized form); GSSGGR, glutathione disulfide reductase; NAD, nicotinamide adenine dinucleotide (oxidized form); NADP, nicotinamide adenine dinucleotide phosphate (oxidized form); NADPH, nicotinamide adenine dinucleotide phosphate (reduced form);  $n_H$ : Hill constant.

\* Corresponding author. Tel.: +90-312-3110588; fax: +90-312-3116616.

E-mail address: nozerl@superonline.com (N. Özer).

## 1. Introduction

Glucose-6-phosphate dehydrogenase (G6PD, EC 1.1.1.49,  $\beta$ -D-glucose-6-phosphate; NADP<sup>+</sup> oxidoreductase) is an enzyme that catalyzes the first step of the pentose phosphate pathway (PPP). PPP has three important functions: (i) production of reducing equivalents in the form of NADPH; (ii) production of pentose phosphates necessary for nucleotide biosynthesis and (iii) to serve as the route of entry of pentoses to the glycolytic pathway. NADPH serves as a hydrogen and electron donor for a variety of reductive reactions, including fatty acid and cholesterol biosynthesis. NADPH also has very important functions in the protection of the cell against oxidative agents by transferring its reductive

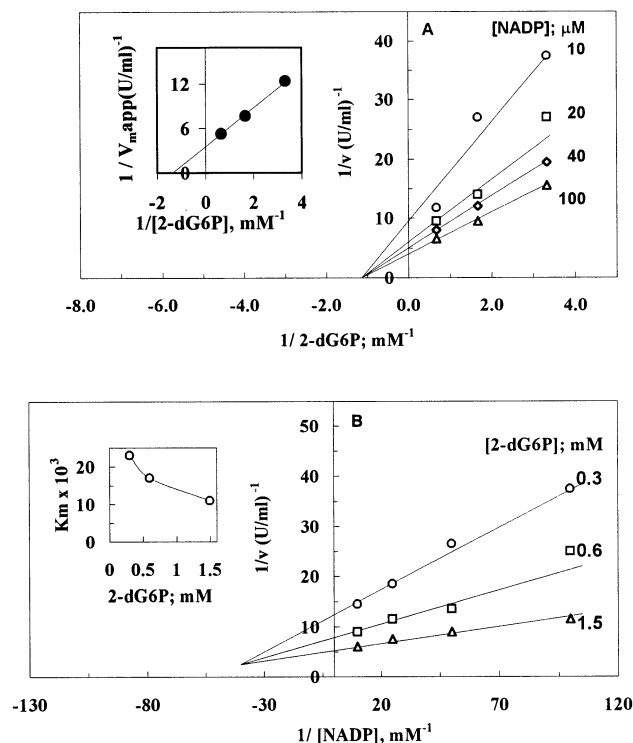


Fig. 1. Lineweaver–Burk plots for 2-dG6P and NADP at constant [NADP]: (–○–) 10, (–□–) 20, (–◇–) 40, and (–△–) 100 μM, (A) and at constant [2-dG6P]: (–○–) 0.3, (–□–) 0.6 and (–△–) 1.5 mM, (B). Double-reciprocal plots of  $V_m$  and  $K_m$  as a function of 2-dG6P inset to (A) and (B), respectively.

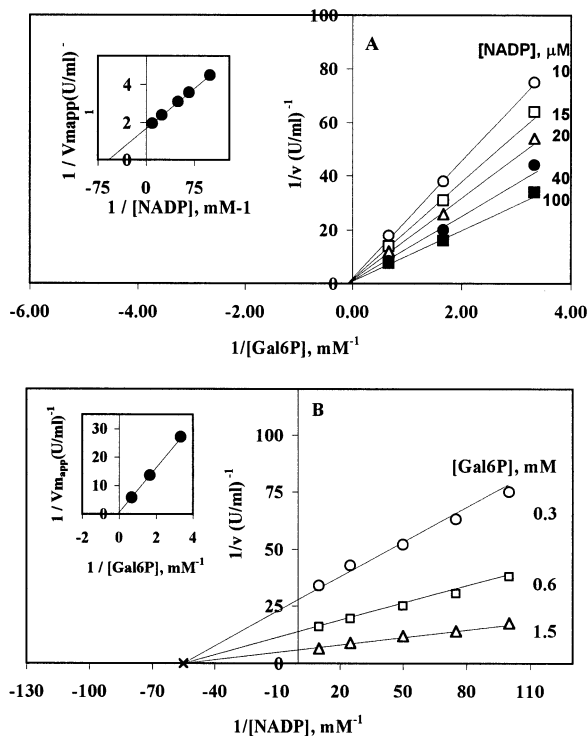


Fig. 2. Lineweaver–Burk plots for Gal6P and NADP at constant [NADP]: (–○–) 10, (–□–) 15, (–△–) 20, (–●–) 40, and (–■–) 100 μM, (A) and at constant [Gal6P]: (–○–) 0.3, (–□–) 0.6 and (–△–) 1.5 mM, (B). Double-reciprocal plots of  $V_m$  as a function of NADP (inset to A) and 2-dG6P (inset to B).

power to oxidized glutathione (GSSG) via glutathione disulfide reductase (GSSGGR) [1],[2]. G6PD was discovered by Warburg and Christian in 1931 and was isolated in crystalline form by Noltmann et al. from brewer's yeast [3,4].

This cytosolic enzyme is widely distributed among microorganisms, plants and in different animal tissues [1–5]. Numerous reports are available regarding the reaction mechanism of G6PD but they have revealed inconsistencies and disagreements in the type of observed kinetics [5,6]. Some studies report classical Michaelis–Menten kinetics, others report sigmoidal kinetics for NADP<sup>+</sup> binding [3–6]. It is well known that in the presence of NADP the enzyme exists in the form of active dimers, trimers, tetramers and hexamers [7–10]. Kinetic properties of G6PD purified

from different sources such as erythrocytes, liver, kidney, brain have been published [5–20].

We have limited information about the biochemistry and physiology of human placenta. Elu-

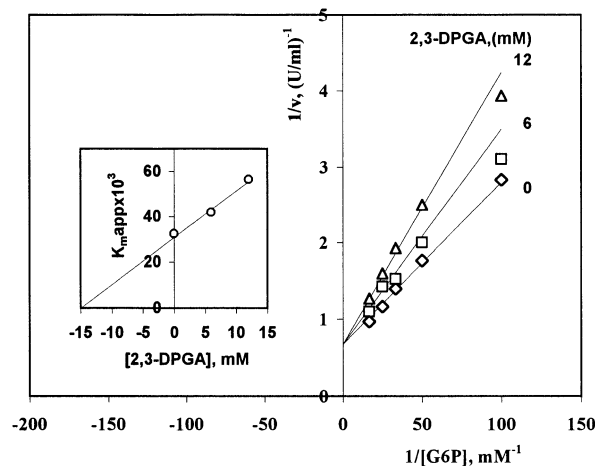


Fig. 3. Lineweaver–Burk plot for 2,3-DPGA inhibition. [2,3-DPGA], (–◇–) 0, (–□–) 6 and (–○–) 12 mM. Inset:  $K_{m,app}$  vs. [2,3-DPGA].

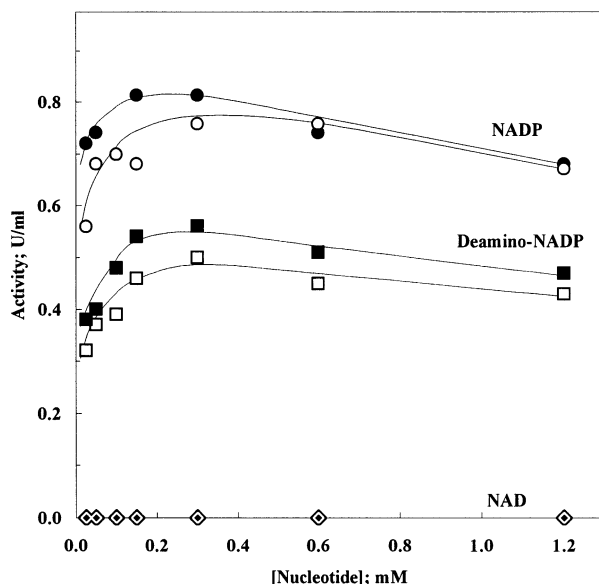


Fig. 4. Michaelis–Menten plots for nucleotides at constant [G6P]. [nucleotides], 40–1200  $\mu$ M; empty symbols 0.3 mM [G6P]; filled symbols 0.6 mM [G6P]. NAD<sup>+</sup>, (–◇–), (–◆–); deamino-NADP (–□–), (–■–); NADP, (–○–), (–●–).

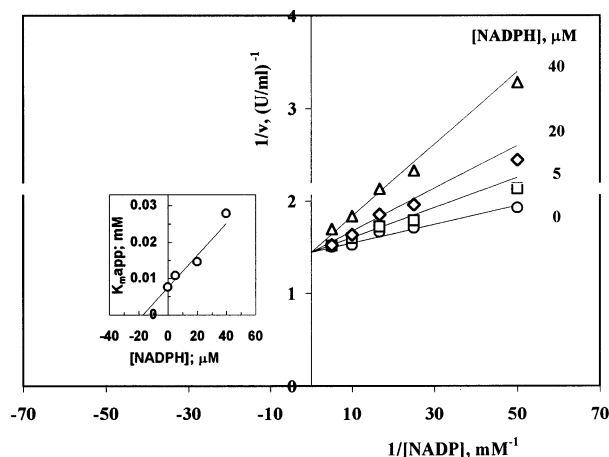


Fig. 5. Lineweaver–Burk Plot for NADPH inhibition. [NADPH], (–○–) 0, (–□–) 5, (–◇–) 20, and (–△–) 40  $\mu$ M. Inset:  $K_{m,app}$  vs. [NADPH].

citation of the kinetic mechanism of placental G6PD will increase our knowledge about the physiology of human placenta and its critical role in the health of the fetus. Glutathione (GSH) is very important for glutathione-S-transferase catalyzed detoxification reactions. In cells, GSH is strictly controlled by GSSGGR, at the expense of NADPH which is a product of G6PD activity. There are only a few reports on the purification of G6PD from human placenta and no report exists on its kinetics [21]. Studying the kinetic mechanism of human placental G6PD in detail will provide information about the control of the placental PPP, the biosynthesis of some physiologically important biomolecules and the status of placental detoxification in relation to the health of the fetus.

## 2. Materials and methods

### 2.1. Materials

DEAE-Sepharose (fast flow), 2',5'-ADP-Sepharose 4B, PBE 96 and PB 74 were from Pharmacia-LKB, Sweden; glycerol from Merck, Germany; 2-mercaptoethanol and Celite from BDH, UK; glucose-6-phosphate, galactose-6-phosphate

(Gal6P), 2-deoxyglucose-6-phosphate (2-dG6P), nicotinamide adenine dinucleotide phosphate (NADP<sup>+</sup> and NADPH), nicotinamide hypoxanthine dinucleotide phosphate (deamino-NADP),  $\epsilon$ -aminocaproic acid ( $\epsilon$ -ACA), nicotinamide adenine dinucleotide (NAD), 2,3-diphosphoglyceric acid (2,3-DPGA), ethylenediamine tetraacetic acid (EDTA) from Sigma, USA.

All other chemicals were standard products of Sigma or Aldrich, USA.

## 2.2. Placental G6PD

Placental G6PD was purified from full-term (40 weeks) human placenta using DEAE-Sepharose (fast flow), 2',5'-AD-Sepharose 4B chromatography, and chromatofocusing on PBE 96 with PB 74, with 62% yield and had a specific activity of 87 IU mg<sup>-1</sup> protein [22]. It was stored in 10 mM Tris-HCl, pH 7.4 containing 5 mM 2-mercaptoethanol, 2 mM EDTA, 1 mM  $\epsilon$ -ACA and 40% (v/v) glycerol at -10°C.

## 2.3. Protein determination

Protein concentration was determined by the micromethod of Bradford, using bovine serum albumin as standard [23].

## 2.4. Kinetics of the human placental G6PD

G6PD activity was determined [24] at 37°C in 125 mM Tris-HCl, pH 7.4, containing 10 mM MgCl<sub>2</sub>, and various concentrations of NADP, G6P, substrate or coenzyme analogues and inhibitors. The reactions were initiated by the addition of enzyme and the conversion of NADP to NADPH was followed by monitoring the change in absorbance at 340 nm, using a Milton-Roy 3000 spectrophotometer ( $\epsilon_{340} = 6.22$  mM<sup>-1</sup>cm<sup>-1</sup>). The activities were used in constructing Lineweaver-Burk and additional diagnostic plots to obtain  $K_m$  values for substrate, coenzyme, substrate and coenzyme analogues and  $K_i$  values for inhibitors [25].

## 3. Results and discussion

The aggregation state and the specific activity of G6PD depend on several factors such as enzyme, NADP, salt, RSH, Mg<sup>2+</sup> (or Mn<sup>2+</sup>) concentrations and pH [5–10,26]. The presence of RSH, high enzyme and Mg<sup>2+</sup> (or Mn<sup>2+</sup>) concentrations and low pH and ionic strength favour oligomeric, active, forms of G6PD (dimer, tetramer and hexamer) and the opposite conditions favour inactive monomers [5–20,26].

Sigmoidal or hyperbolic kinetics have been reported for G6PD in the intracellular medium and this property is reportedly lost upon purification [11]. The kinetics have been shown to depend on the storage and assay conditions [5–13,26,27]. It has been proposed that the sigmoidal kinetics in G6PD arise from hexamer-tetramer-dimer transitions in the enzyme. Aggregation states higher than dimer are responsible for sigmoidicity [5–14].

Under the storage and assay conditions given above, more than 95% of G6PD in our preparation existed in dimeric form [22]. The enzyme obeyed "Rapid Equilibrium Ordered Bi Bi" sequential kinetic model in the range, G6P = 20–300  $\mu$ M and NADP = 10–200  $\mu$ M [22]. This kinetic behaviour was in accordance with most of the G6PD from other sources [1,5]. The  $K_m$  values for G6P and NADP, obtained from Lineweaver-Burk plots, were  $40 \pm 8$  and  $20 \pm 10$   $\mu$ M, respectively (Table 1) [22]. An intrinsic  $V_m$  value of 87 IU mg<sup>-1</sup> protein was obtained from secondary plots of  $1/V_m$  versus  $1/\text{G6P}$  (or  $1/\text{NADP}$ ) (Table 1) [22]. These values are comparable with most of the kinetic parameters reported for G6PD from other sources [5–20].

Kinetic studies with substrate analogues have shown that G6PDs obtained from different sources (*Bacillus stearothermophilis* [28], rat testis [29] and human erythrocyte [13]) use these analogues with different efficiencies. The catalytic efficiencies ( $k_{\text{cat}}/K_m$ ) of placental G6PD with G6P, 2-dG6P and Gal6P were  $7.4 \times 10^6$ ,  $4.89 \times 10^4$  and  $1.57 \times 10^4$  M<sup>-1</sup>.s<sup>-1</sup>, respectively (Table 1). These data show that G6P is 150 and 500 times better as substrate than 2-dG6P and Gal6P. Among the substrate analogues 2-dG6P was three times better

Table 1  
Kinetic parameter of G6PD from human placenta

Kinetic parameters: substrate, coenzyme, substrate and coenzyme analogues	$V_m$ (IU $\text{mg}^{-1}$ protein)	$K_m \pm \text{S.D.}$ (mM)	$k_{\text{cat}}$ ( $\text{s}^{-1}$ )	$K_{\text{cat}}/K_m$ ( $\text{M}^{-1}\text{s}^{-1}$ )	$K_i \pm \text{S.D.}$ (mM)	$n_H \pm \text{S.D.}$
Glucose-6-phosphate	87	$0.040 \pm 0.008$	296	$7.40 \times 10^6$	-	$0.84 \pm 0.15$
NADP (G6P as substrate)	87	$0.020 \pm 0.010$	296	$1.48 \times 10^7$	-	$0.86 \pm 0.14$
2-Deoxyglucose-6-phosphate	13	$0.87 \pm 0.06$	43	$4.89 \times 10^4$	-	-
Galactose-6-phosphate	46	$10 \pm 2$	157	$1.57 \times 10^4$	-	-
Deamino-NADP	29	$0.0205 \pm 0.0055$	98	$4.80 \times 10^6$	-	-
NADPH	-	-	-	-	$0.0171 \pm 0.0032$	-
2,3-Diphosphoglycerate	-	-	-	-	$15 \pm 3$	-

as substrate than Gal6P implicating the hydroxyl group on the fourth carbon is more critical than the hydroxyl group on the second carbon in determining selectivity. The  $K_m$  of  $0.87 \pm 0.06$  mM for 2-dG6P was not affected by a change in the concentration of NADP (Fig. 1A and inset) but the reverse was not true; the  $K_m$  value for NADP decreased from  $30 \pm 5$  to  $10 \pm 2$   $\mu\text{M}$  as the 2-dG6P concentration increased from 0.3 mM and approaching to a plateau at around 1.5 mM (Fig. 1B and inset). The  $K_m$  values for Gal6P and NADP were found to be  $10 \pm 2$  and  $20 \pm 10$   $\mu\text{M}$  (Fig. 2A, B and the secondary plots as insets to Fig. 2A and B).

With G6P as the variable substrate, G6PD was inhibited competitively by 2,3 DPGA with a  $K_i$  of  $15 \pm 3$  mM (Fig. 3 and inset).

G6PDs isolated from different sources are reported to use NAD as coenzyme [18,19,28,29]. NAD was neither a coenzyme nor an inhibitor for human placental G6PD indicating that this enzyme belongs to Class II type G6PDs (Fig. 4) [1]. Human erythrocyte G6PD utilizes deamino-NADP as coenzyme with ca. 60% efficiency compared to NADP [30–32]. The  $k_{\text{cat}}/K_m$  value for deamino-NADP as coenzyme in the human placental G6PD system was  $4.80 \times 10^6 \text{ M}^{-1}\text{s}^{-1}$ , corresponding to 32.4% of the native activity with NADP (Fig. 4, Table 1). Both NADP and deamino-NADP showed hyperbolic saturation upto 300  $\mu\text{M}$  coenzyme concentration beyond which an inhibition was observed, regardless of the G6P concentration (Fig. 4).

All G6PDs isolated from different sources were inhibited by NADPH [1,5], with  $K_i$  values ranging from 6 (human platelet) to 200  $\mu\text{M}$  (spinach leaf) [1,5]. Human erythrocyte G6PD  $K_i$  was found to be 9.1  $\mu\text{M}$  [7]. Human placental G6PD was inhibited competitively by NADPH, with a  $K_i$  of  $17.1 \pm 3.2$   $\mu\text{M}$  (Fig. 5, inset), pointing to tight enzyme:NADPH binding and the important role of NADPH in the regulation of the pentose phosphate pathway which makes a critical contribution to many detoxification processes.

## References

- [1] H.R. Levy, Glucose-6-phosphate dehydrogenase, in: A.

- Meister (Ed.), *Advances in Enzymology*, vol. 48, Wiley, New York, 1979, pp. 97–192.
- [2] L.V. Eggleston, H.A. Krebs, Regulation of the pentose phosphate cycle, *Biochem. J.* 138 (1974) 425–435.
- [3] E.A. Noltmann, C.J. Gubler, S.A. Kuby, Glucose-6-phosphate dehydrogenase (zwischenferment), *J. Biol. Chem.* 236 (1961) 1225–1230.
- [4] O. Warburg, W. Christian, Aktivierung von kohlehydrat in roten blut zellen, *Biochem. Z.* 238 (1931) 131–134.
- [5] M.A. Rosemeyer, The biochemistry of glucose-6-phosphate dehydrogenase, 6-phosphogluconate dehydrogenase and glutathione reductase, *Cell. Biochem. Func.* 5 (1987) 79–95.
- [6] S.A. Adediran, Sigmoidal kinetics of human erythrocyte glucose-6-phosphate dehydrogenase, *Arch. Biochem. Biophys.* 262 (1988) 354–359.
- [7] A. Yoshida, Hemolytic anemia and G6PD deficiency, *Science* 179 (1973) 532–537.
- [8] P. Cohen, M.A. Rosemeyer, Glucose-6-phosphate dehydrogenase from human erythrocytes, *Methods Enzymol.* 61 (1975) 208–214.
- [9] S.A. Kuby, J.T. Wu, R.N. Roy, Glucose-6-phosphate dehydrogenase from brewers' yeast (zwischenferment), *Arch. Biochem. Biophys.* 165 (1986) 153–178.
- [10] A. Bonsignore, A. De Flora, Glucose-6-phosphate dehydrogenase, in: B.L. Horecker, E.R. Stadman (Eds.), *Current Topics in Cellular Regulation*, vol. 6, Academic Press, New York and London, 1972, pp. 21–62.
- [11] H.N. Kirkman, G.F. Gaetani, Regulation of glucose-6-phosphate dehydrogenase from human erythrocytes, *J. Biol. Chem.* 261 (1986) 4033–4038.
- [12] S.J. Soldin, D. Balinsky, The kinetic properties of human erythrocyte glucose-6-phosphate dehydrogenase, *Biochemistry* 7 (1968) 1077–1081.
- [13] A. Yoshida, Glucose-6-phosphate dehydrogenase of human erythrocytes, *J. Biol. Chem.* 241 (1966) 4966–4976.
- [14] S.A. Adediran, Kinetic properties of normal human erythrocyte glucose-6-phosphate dehydrogenase dimers, *Biochimie* 73 (1991) 1211–1218.
- [15] A. Yoshida, M. Lin, Regulation of glucose-6-phosphate dehydrogenase activity in red blood cells from hemolytic and nonhemolytic variant subjects, *Blood* 41 (1973) 877–890.
- [16] F.J. Corpas, L. García-Salguero, J. Peragón, J.A. Lupiáñez, Kinetic properties of hexose-monophosphate dehydrogenases. I. Isolation and partial purification of glucose-6-phosphate dehydrogenase from rat liver and kidney cortex, *Life Sci.* 56 (1995) 179–189.
- [17] M.A. Askar, K. Sumathy, N.Z. Baquer, Regulation and properties of purified glucose-6-phosphate dehydrogenase from rat brain, *Indian J. Biochem. Biophys.* 33 (1996) 512–518.
- [18] H.R. Levy, R.R. Raineri, B.H. Nevaldine, On the structure and catalytic function of mammary glucose-6-phosphate dehydrogenases, *J. Biol. Chem.* 241 (1966) 2181–2187.
- [19] C. Olive, M.E. Geroch, H.R. Levy, Glucose-6-phosphate dehydrogenase from *Leconostoc mesenteroides* kinetic studies, *J. Biol. Chem.* 246 (1971) 2047–2057.
- [20] H.R. Levy, C. Cook, Purification and properties of NADP-linked glucose-6-phosphate dehydrogenase from *Acetobacter hansenii* (*Acetobacter xylinum*), *Arch. Biochem. Biophys.* 291 (1991) 161–167.
- [21] G.W. Oertel, P. Benes, The effects of steroids on glucose-6-phosphate dehydrogenase, *J. Steroid Biochem.* 3 (1972) 493–496.
- [22] Y. Aksoy, I.H. Ögüs, N. Özer, Purification and some properties of human placental glucose-6-phosphate dehydrogenase, *Prot. Exp. Purif.* 21 (2001) 000–000.
- [23] M.M. Bradford, A rapid and sensitive method for the quantitation of microgram quantities of protein utilizing the principle of protein-dye binding, *Anal. Biochem.* 72 (1976) 248–254.
- [24] E. Beutler, *Red Cell Metabolism*, Grune & Stratton, New York and London, 1971, pp. 62–64.
- [25] I.H. Segel, *Enzyme Kinetics*, Wiley, New York, 1975.
- [26] H.N. Kirkman, E.M. Hendrikson, Glucose-6-phosphate dehydrogenase from human erythrocytes, *J. Biol. Chem.* 237 (1962) 2371–2376.
- [27] A.E. Chung, R.G. Langdon, Human erythrocyte glucose-6-phosphate dehydrogenase. II. Enzyme–coenzyme interrelationship, *J. Biol. Chem.* 238 (1963) 2317–2324.
- [28] H. Okuno, K. Nagata, H. Nakajima, Purification and properties of glucose-6-phosphate dehydrogenase from *Bacillus stearothermophilus*, *J. Appl. Biochem.* 7 (1985) 192–201.
- [29] C. Lee, J.H. Yuan, D. Moser, J.M. Kramer, Purification and characterization of mouse glucose-6-phosphate dehydrogenase, *Mol. Cell. Biochem.* 24 (1979) 67–73.
- [30] A. Pekrun, S.W. Eber, W. Schröter, G6PD avenges and G6PD moosburg: biochemical and erythrocyte membrane characterization, *Blut* 58 (1989) 11–14.
- [31] H. Ogura, T. Morisaki, K. Tani, H. Kanno, H. Tsutsumi, K. Takahashi, T. Miyamori, H. Fujii, S. Miwa, A new glucose-6-phosphate dehydrogenase variant (G6PD Tsukui) associated with congenital hemolytic anemia, *Hum. Genet.* 78 (1988) 369–371.
- [32] J.T. Prchal, K. Hall, M. Csepregy, M. Lilly, R. Berkow, C.W. Scott, Two apparent glucose-6-phosphate dehydrogenase variants in normal XY males: G6PD Alabama, *Am. J. Med.* 84 (1988) 517–522.