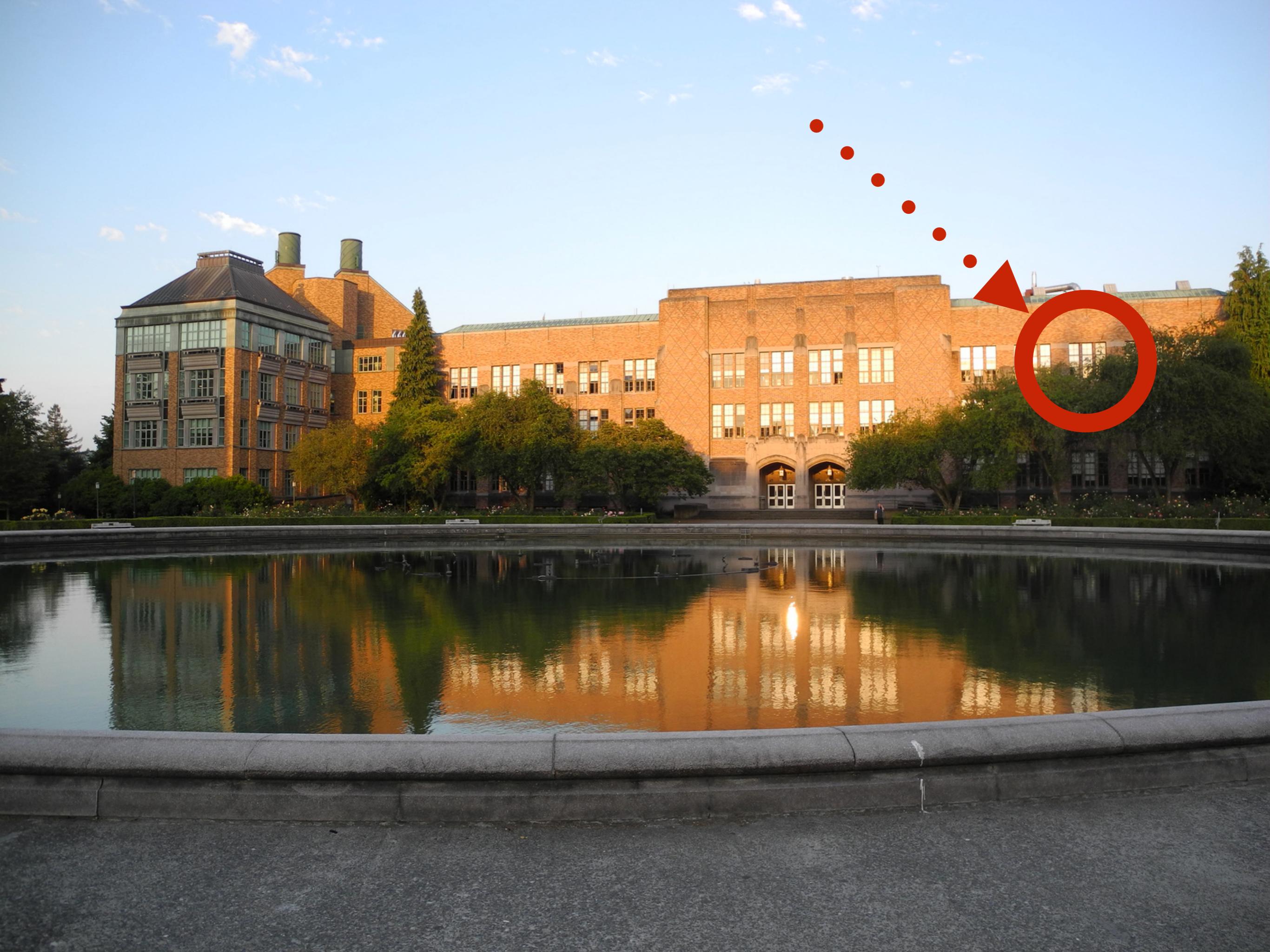


# Explorations into Molecular Magnetism

Joshua Goings  
University of Washington, Seattle, WA









**uw**

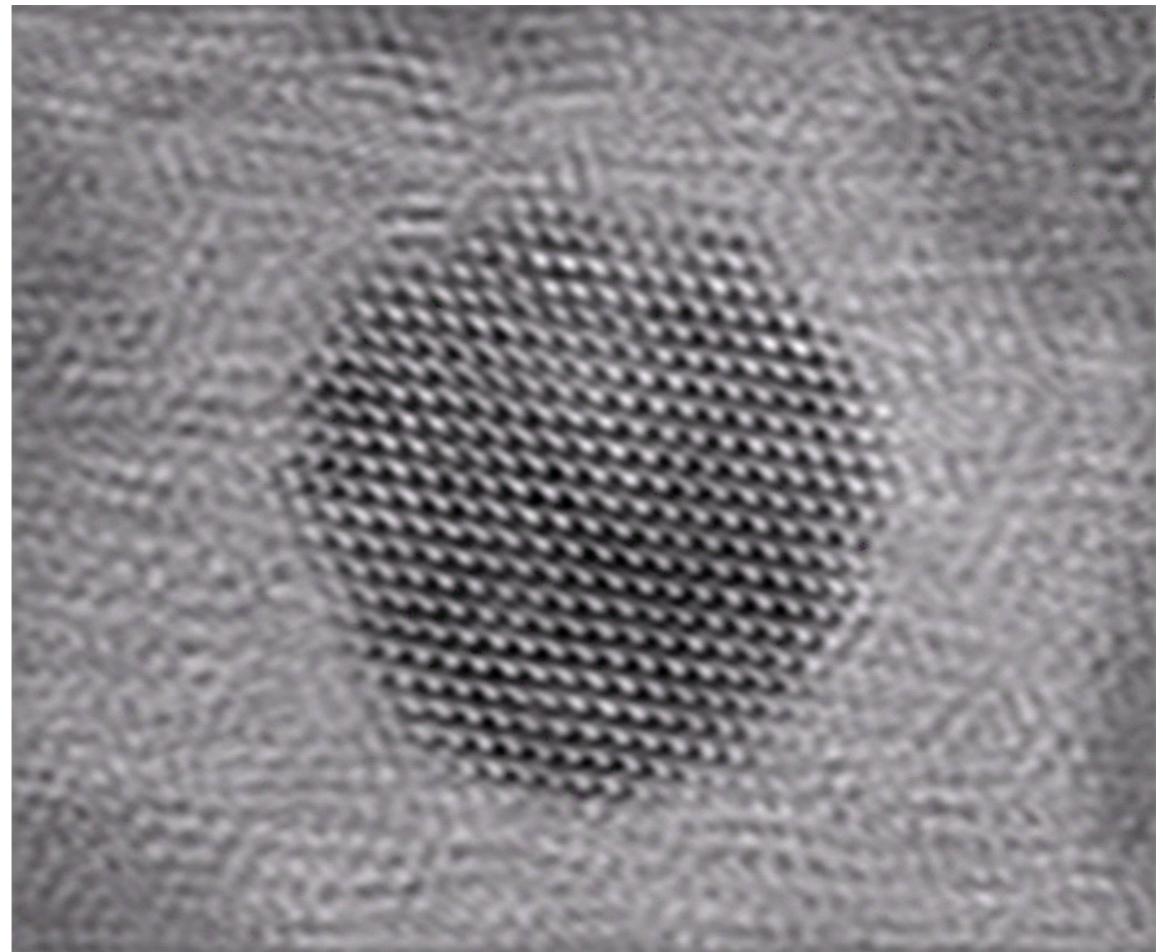
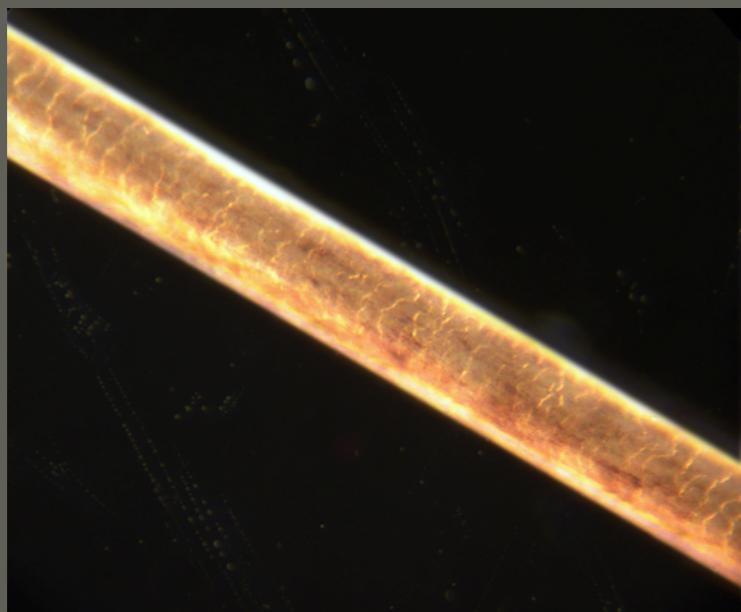
Li Research Group

**LABMEN**

Our big focus is using computers to predict properties of nanomaterials, e.g. quantum dots

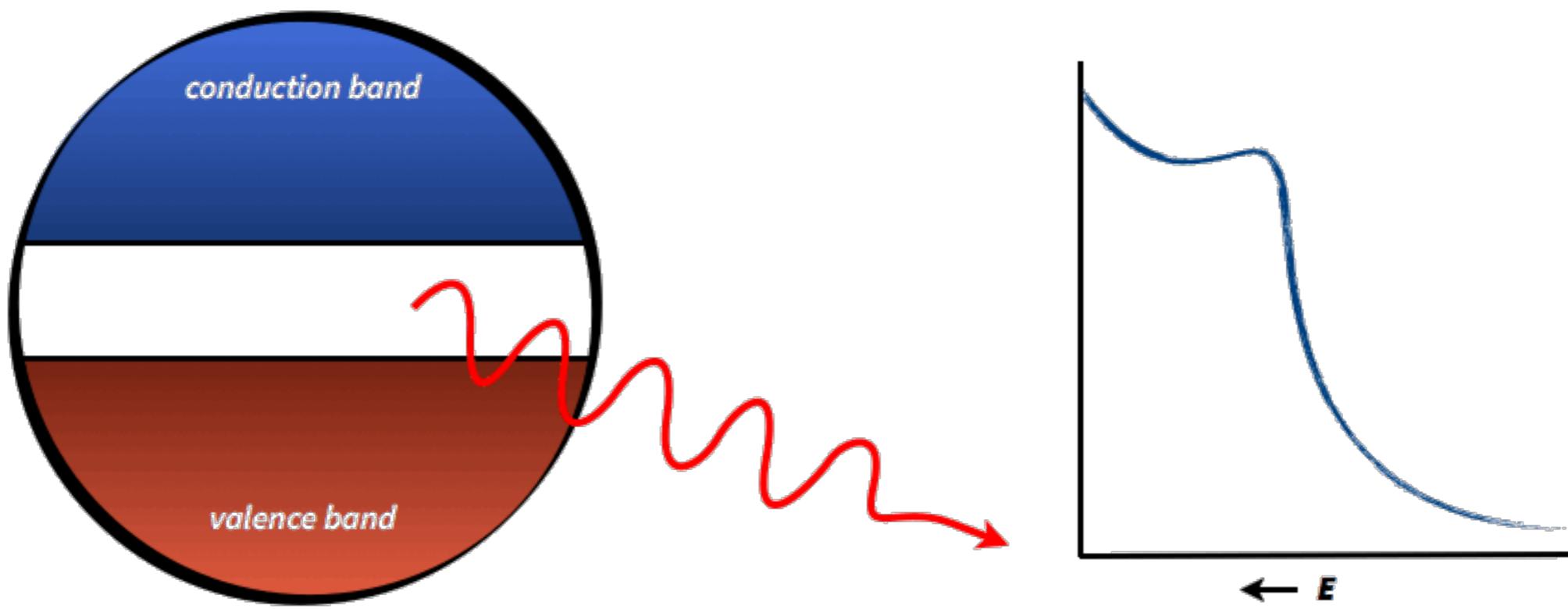
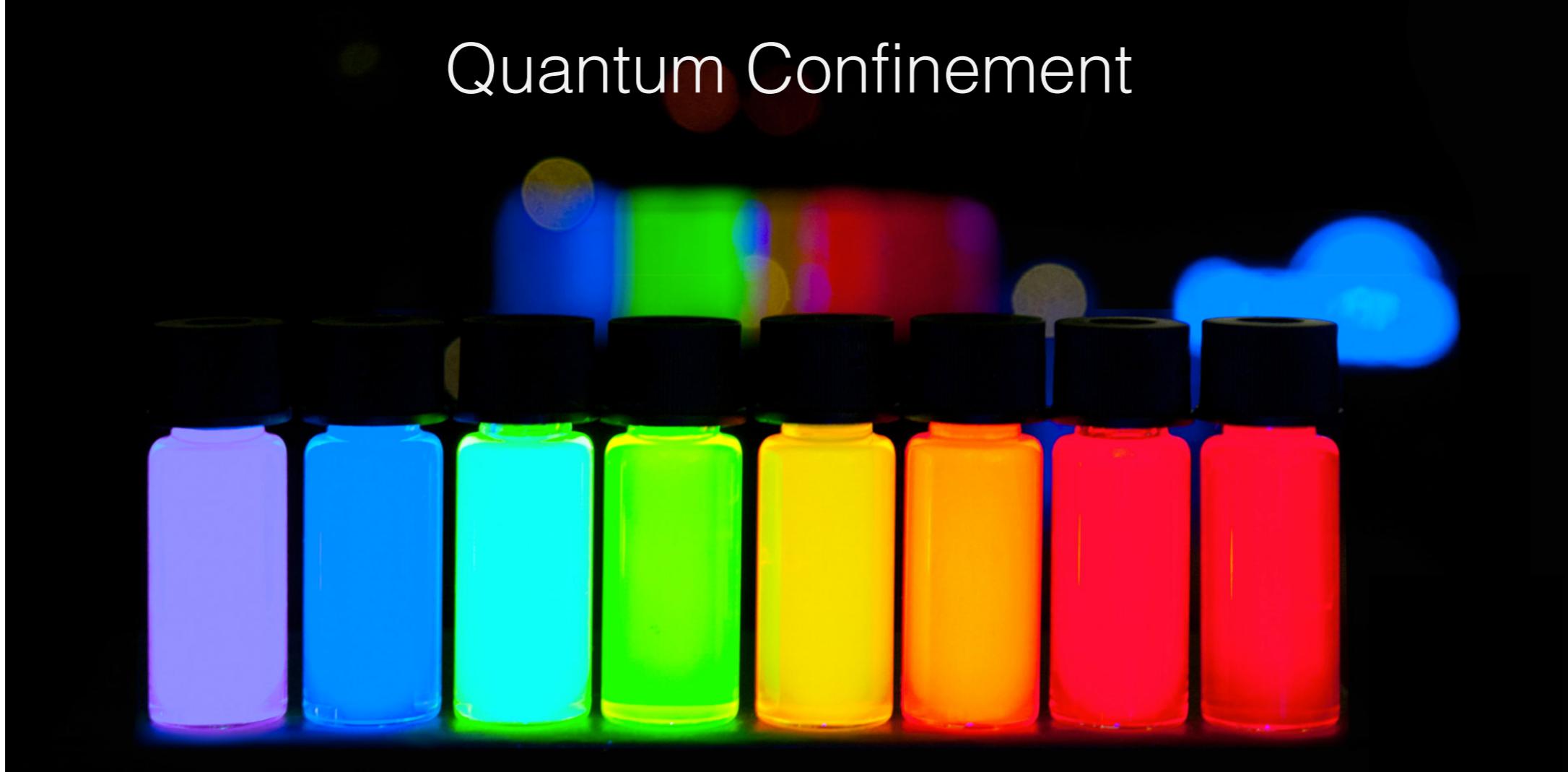


10,000 times  
smaller than  
diameter of  
a human hair



You could fit  
3,000,000  
end-to-end across  
the width of  
your thumb

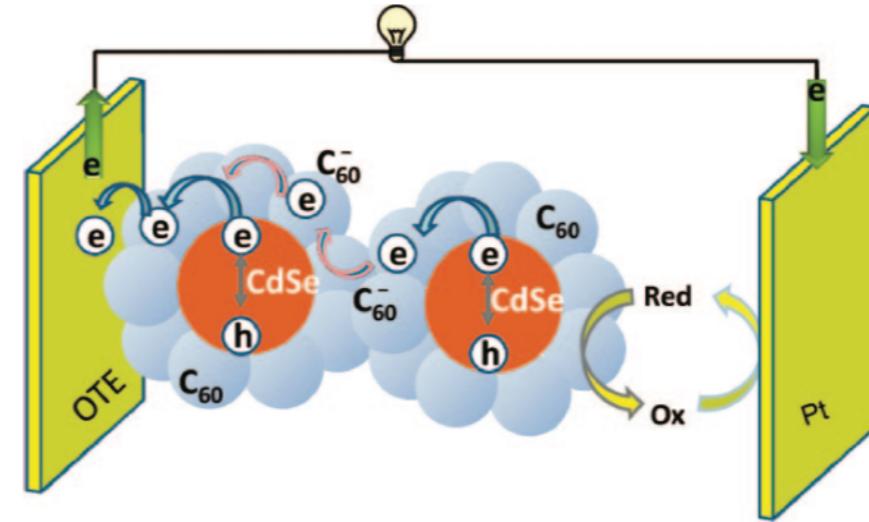
# Quantum Confinement



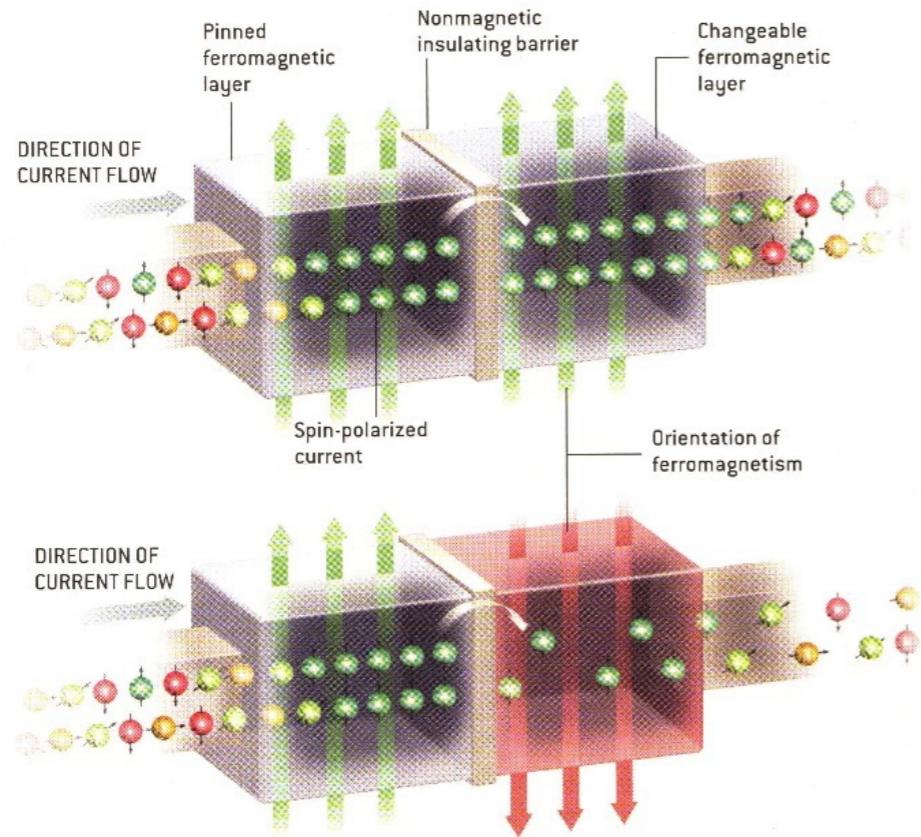
# Photovoltaics and Spintronics



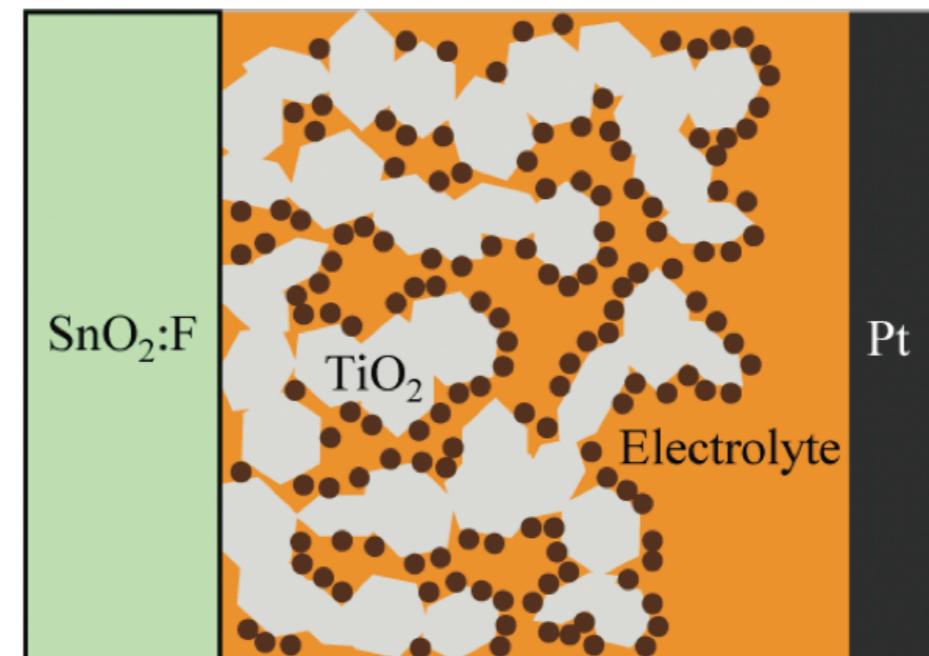
Ginger Research Lab, UW Chemistry



Kamat, et al. JACS, 2008

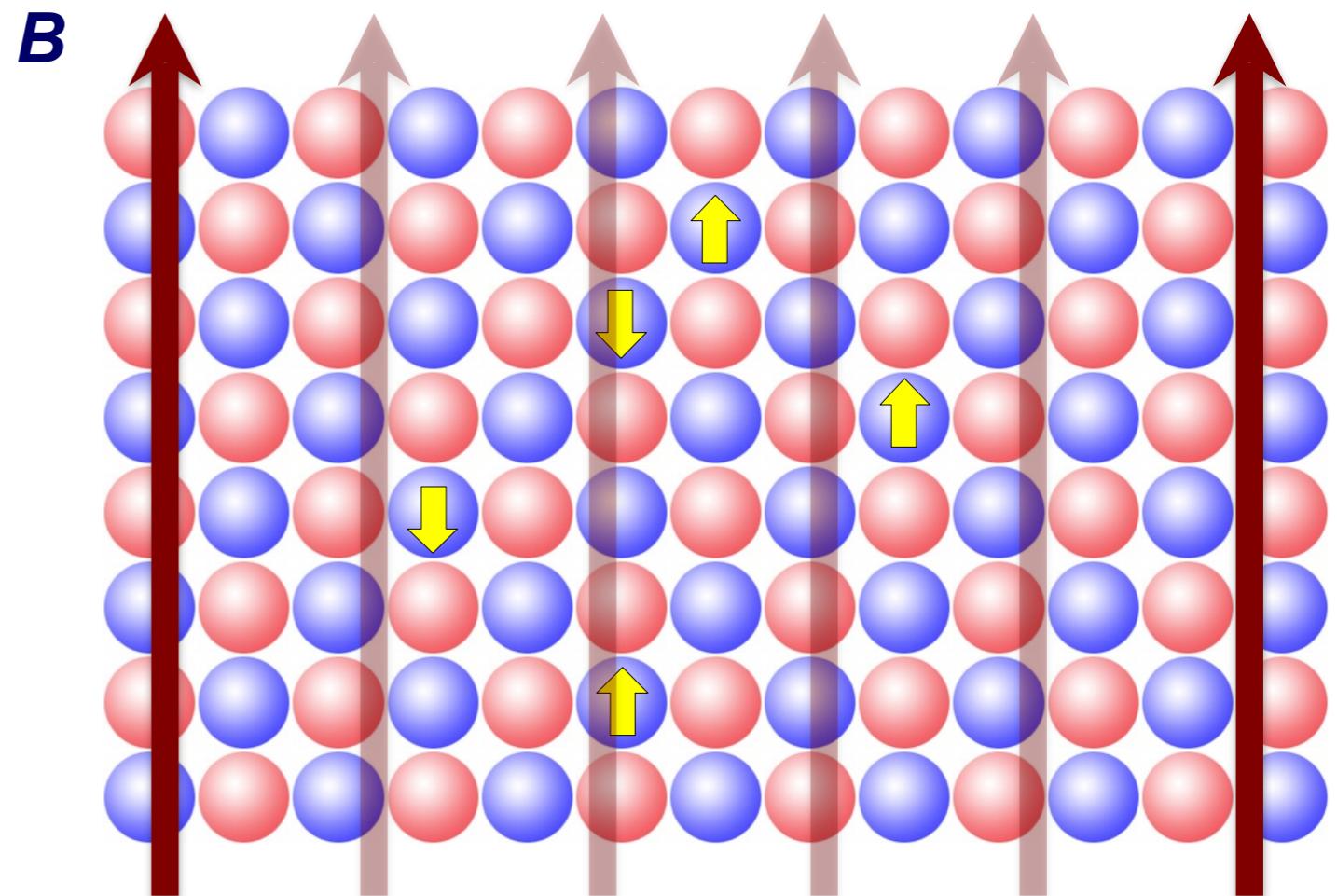


Scientific American, June 2002



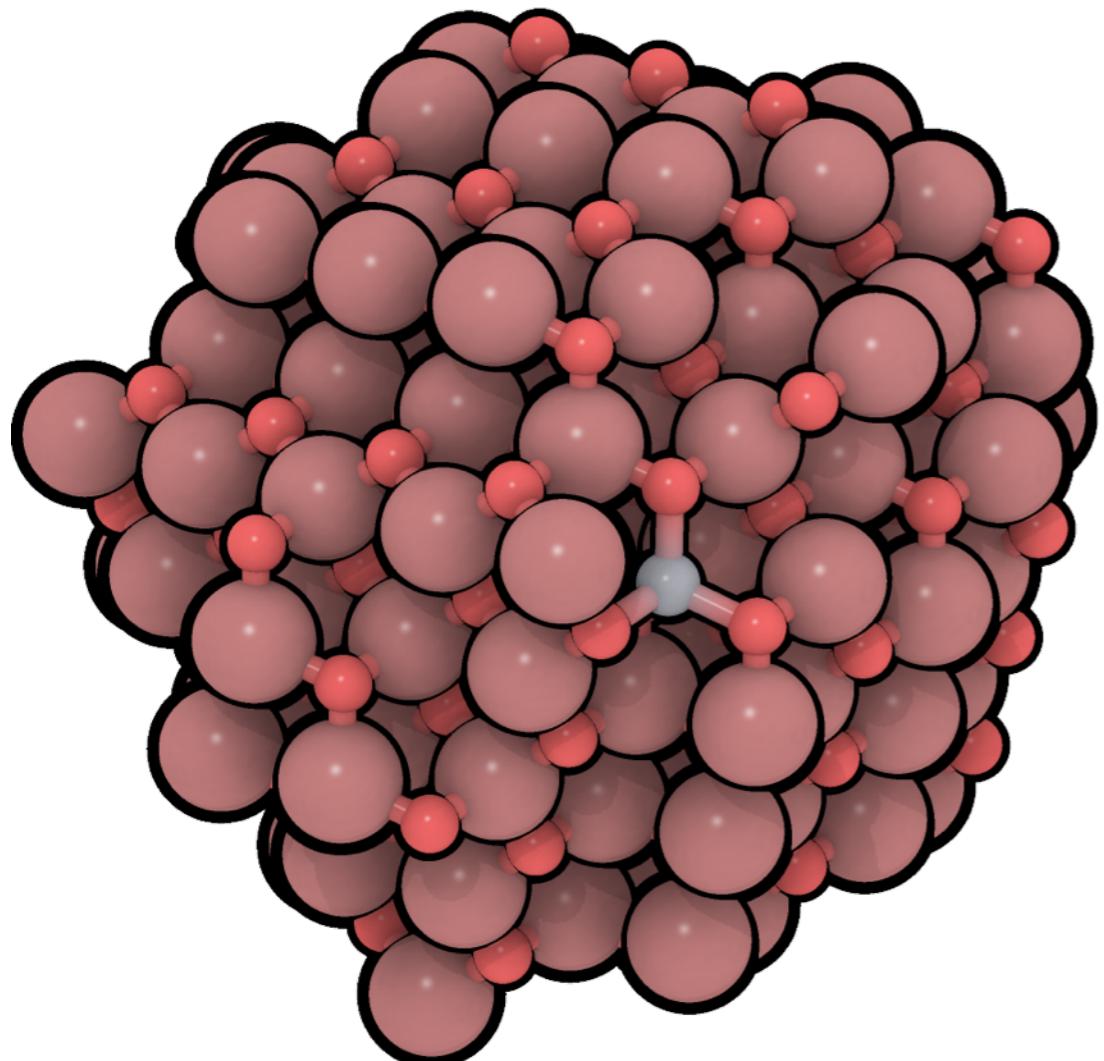
Sargent, et al. ACS Nano, 2010

# Diluted Magnetic Semiconductors (DMS)



- ✓ magnetization
- ✓ additional absorption features

These magnetic dopants interact strongly with certain other dopants that may lead to



New superconductors

High-temp ferromagnets

**Incredibly small hard drives**

**Super-fast processors**

The problem is no one really knows how this works.

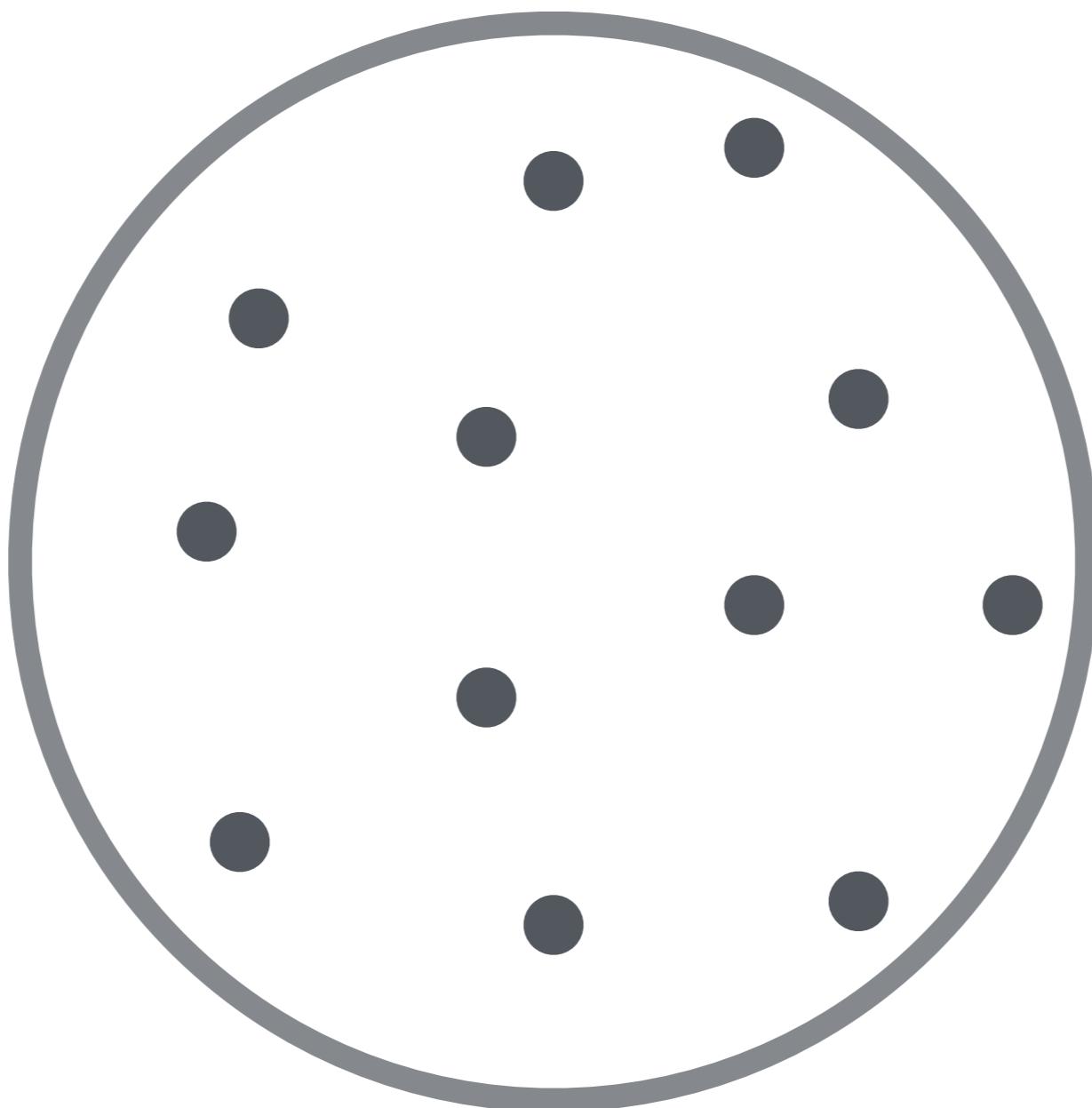
I use **quantum chemistry** to simulate these materials on supercomputing clusters

Hartree-Fock is a computational method  
for studying electrons in molecules

Instead of studying many electrons at once,  
we look at each electron individually moving  
in the field of the other electrons.

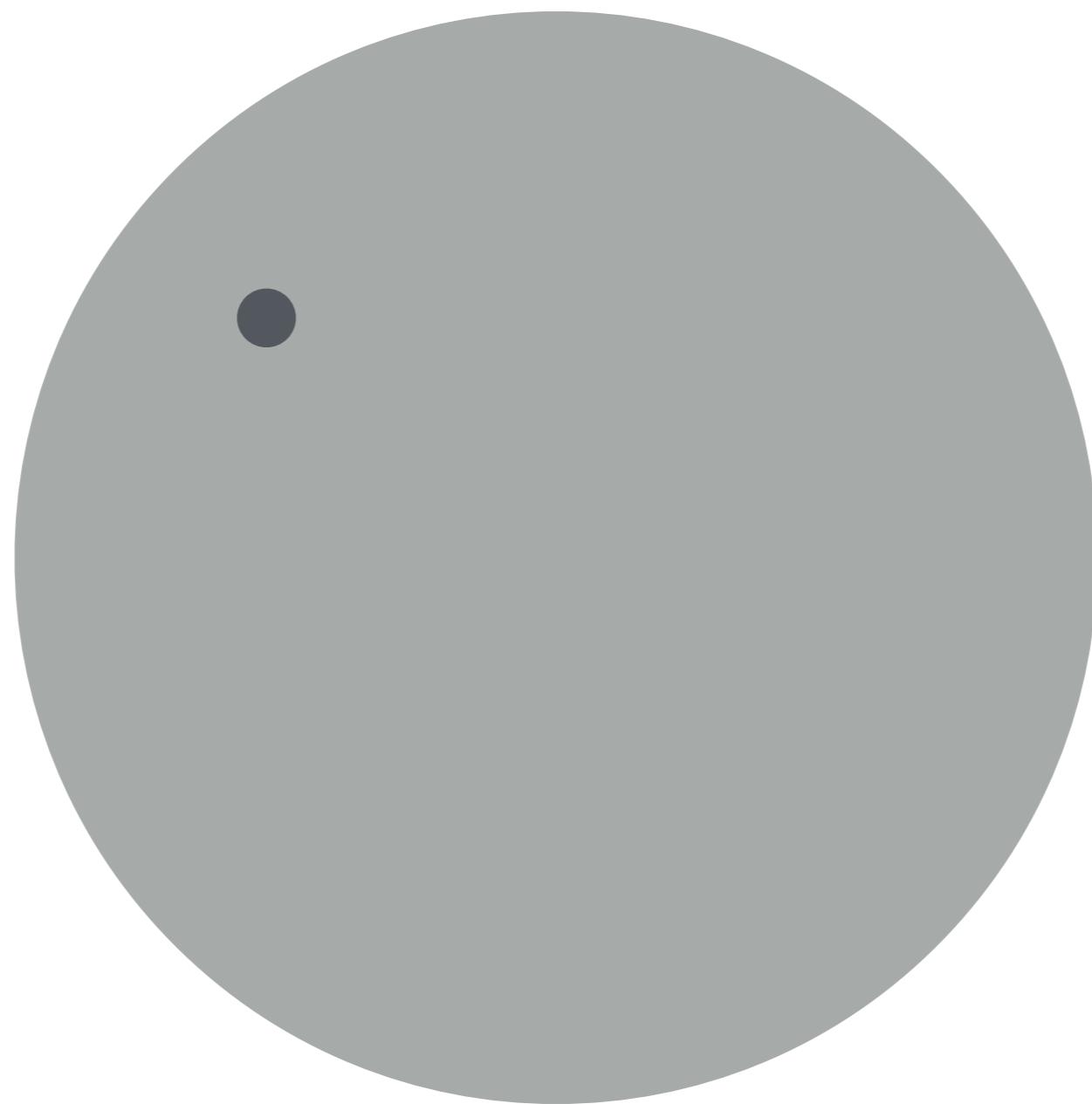
# “Molecule”

- electron



# “Molecule”

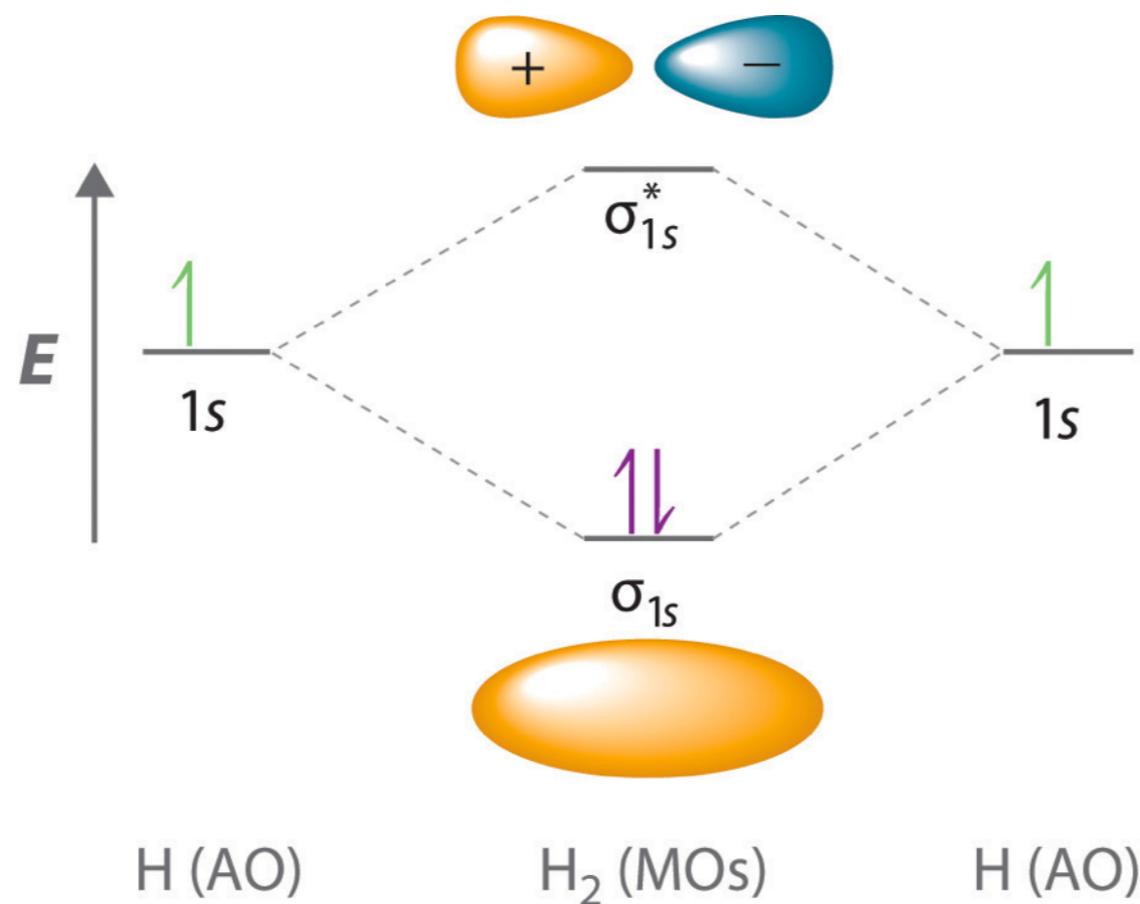
- electron



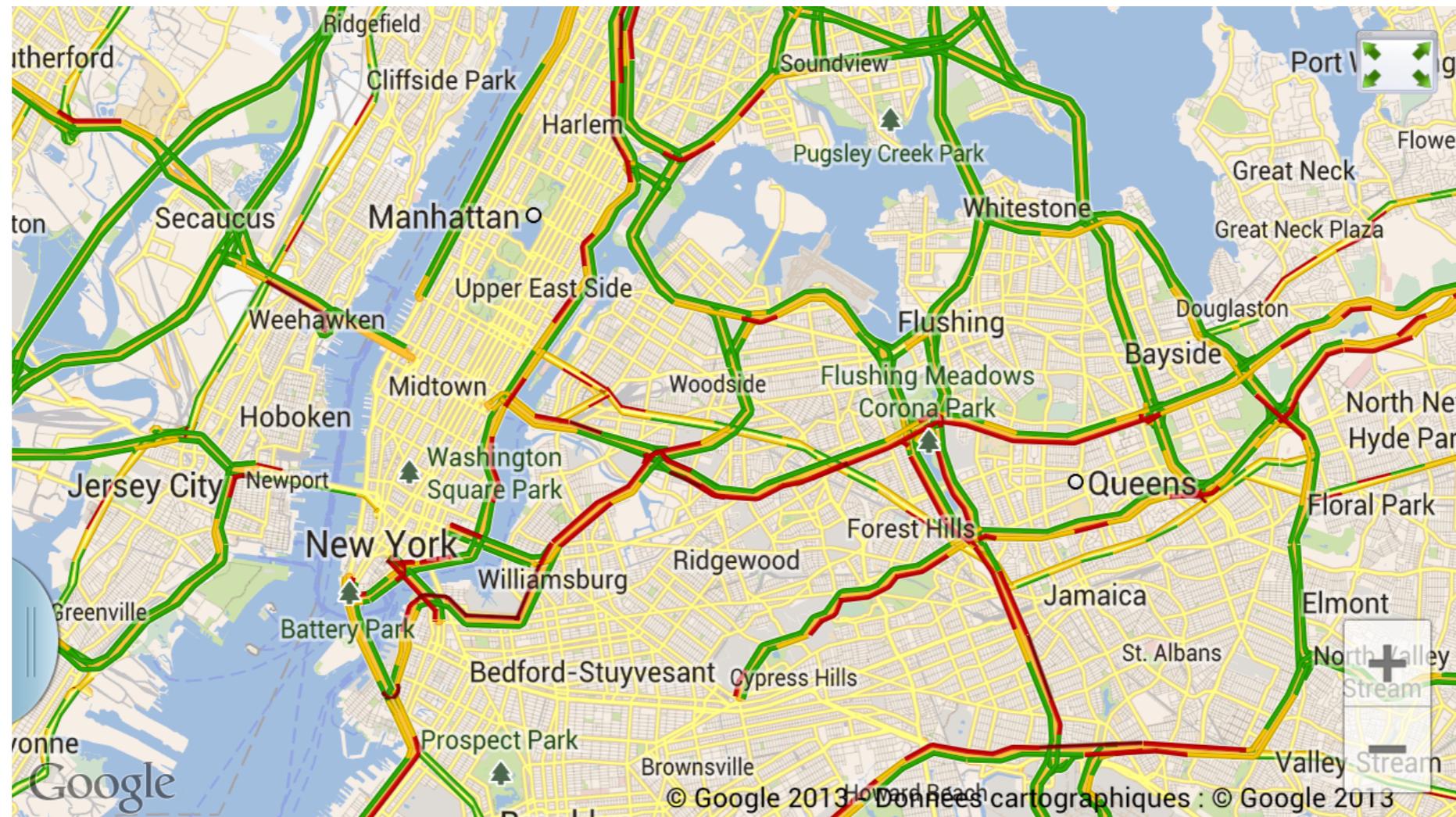
- “smeared”  
electron cloud

The function that describes how the electron moves in the “mean field” of the other electrons is its orbital.

## Hartree Fock theory is the basis of Molecular Orbital theory!



Less formally, Hartree Fock is the Google Maps of chemistry



All the cars are smeared out into “traffic density”

All the electrons are smeared out into “electron density”

But what sort of interactions should we consider?

Quantum chemists try to solve

$$\hat{H}\Psi = E\Psi$$

$\hat{H}$

the Hamiltonian describes  
interactions between electrons

$\Psi$

the wavefunction describes the molecule

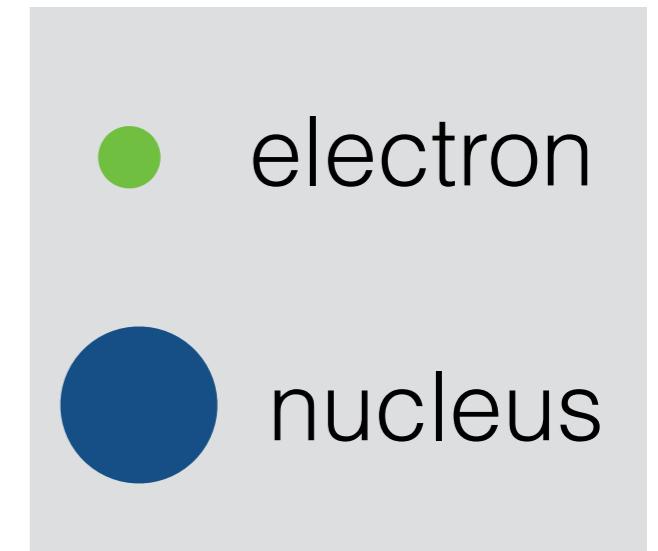
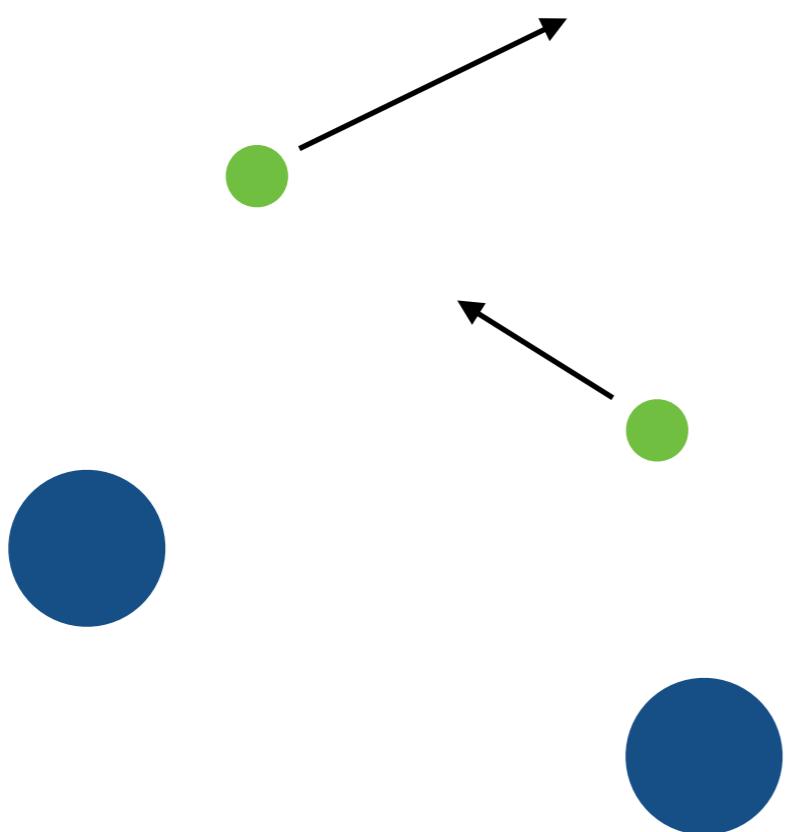
$E$

is the energy of the system

This is the **Schrodinger Equation**  
(and is in general impossible to solve)

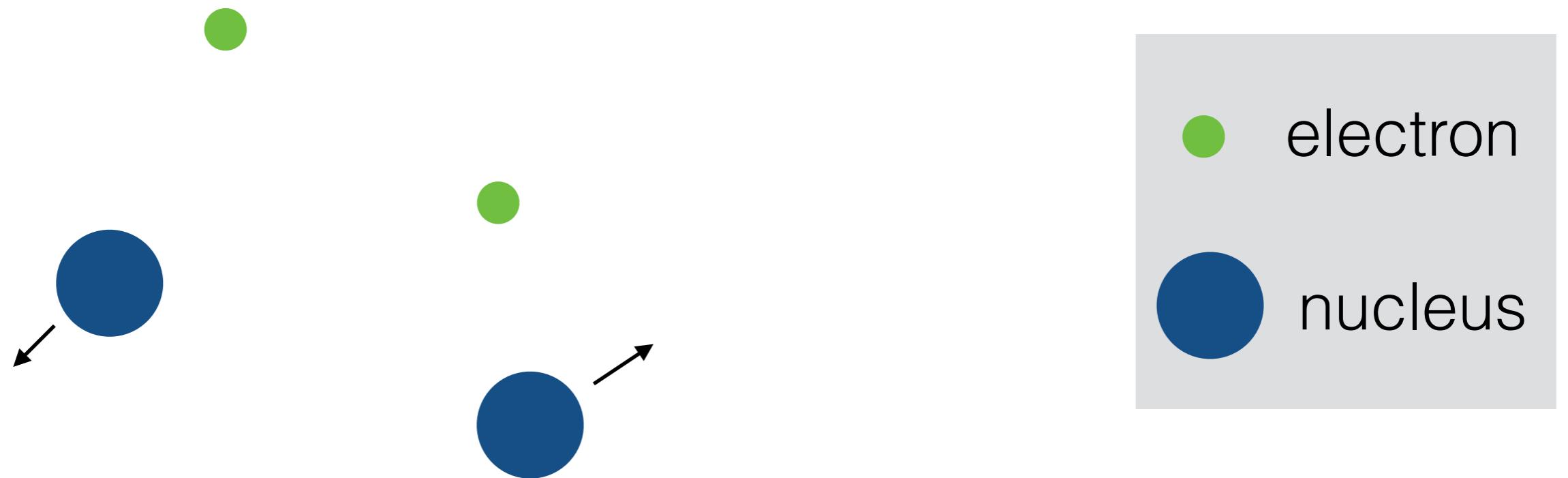
For molecules, the Hamiltonian describes five interactions

$$\hat{H} = \text{electron kinetic energy}$$



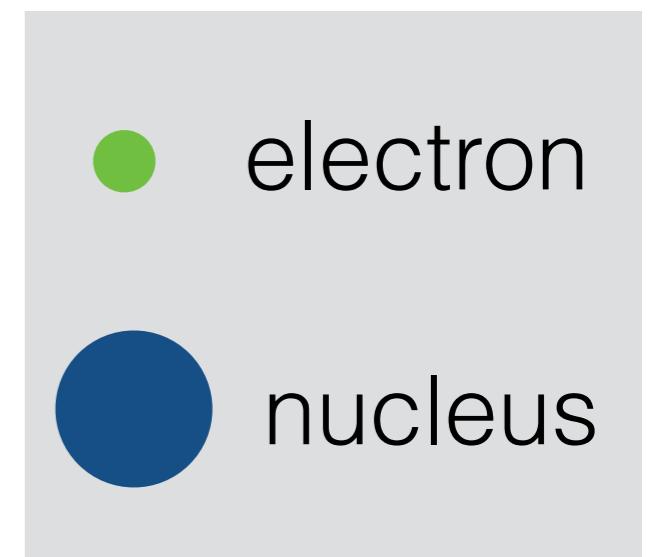
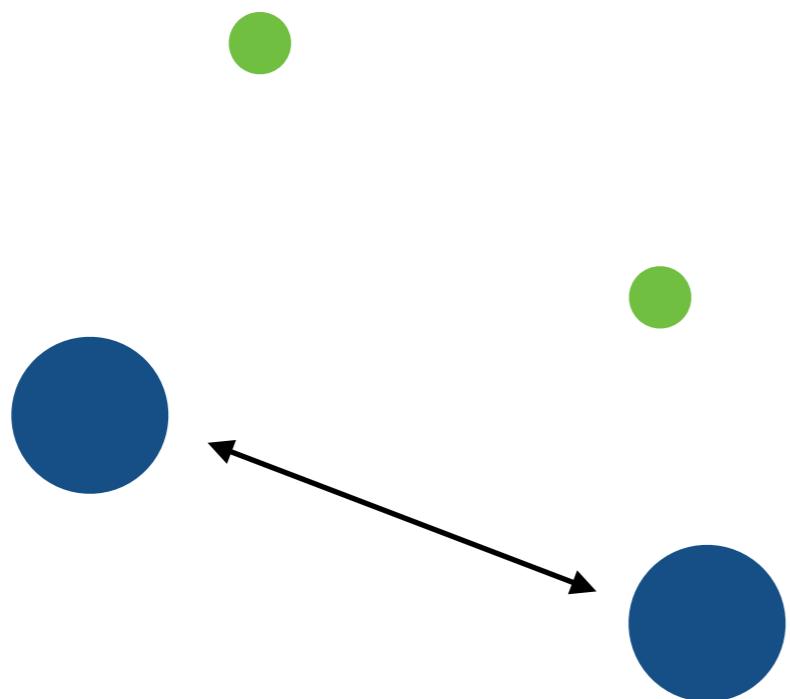
For molecules, the Hamiltonian describes five interactions

$$\hat{H} = \text{electron kinetic energy} + \text{nuclei kinetic energy}$$



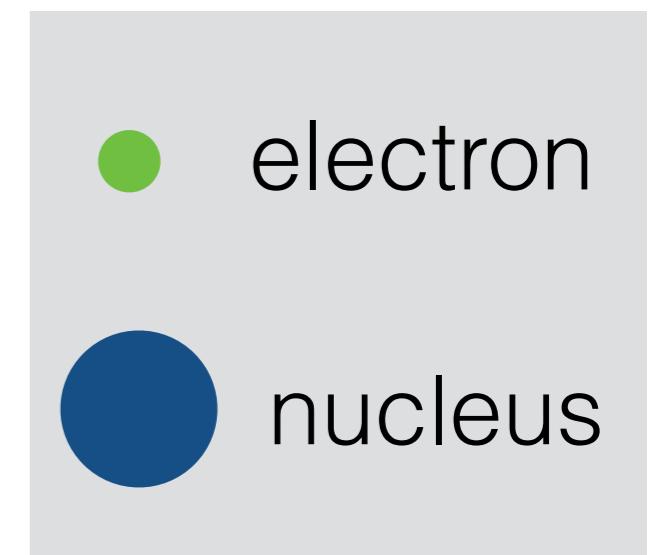
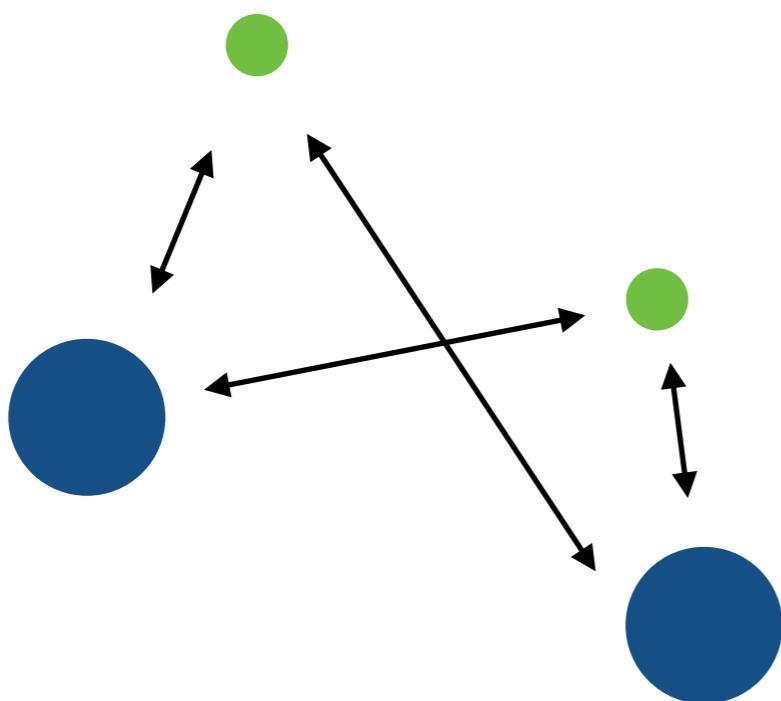
For molecules, the Hamiltonian describes five interactions

$$\hat{H} = \text{electron kinetic energy} + \text{nuclei kinetic energy} + \text{nuclear-repulsion}$$



For molecules, the Hamiltonian describes five interactions

$$\hat{H} = \text{electron kinetic energy} + \text{nuclei kinetic energy} \\ + \text{nuclear-repulsion} + \text{electron-nuclear attraction}$$



For molecules, the Hamiltonian describes five interactions

$$\hat{H} = \text{electron kinetic energy} + \text{nuclei kinetic energy} \\ + \text{nuclear-repulsion} + \text{electron-nuclear attraction} \\ + \text{electron-electron repulsion}$$

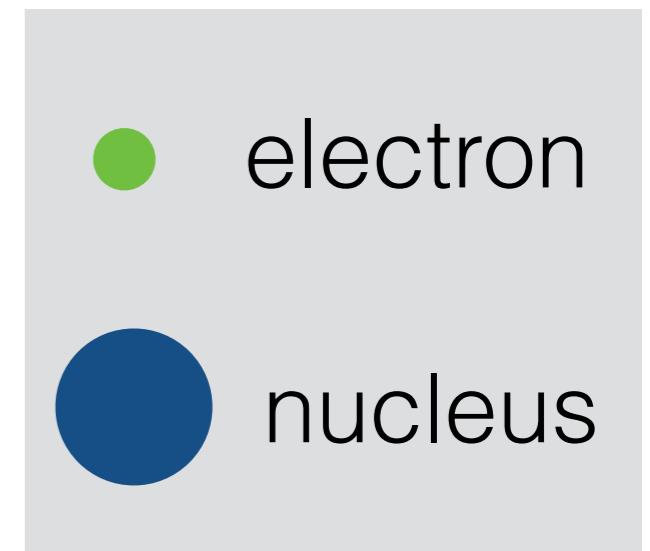
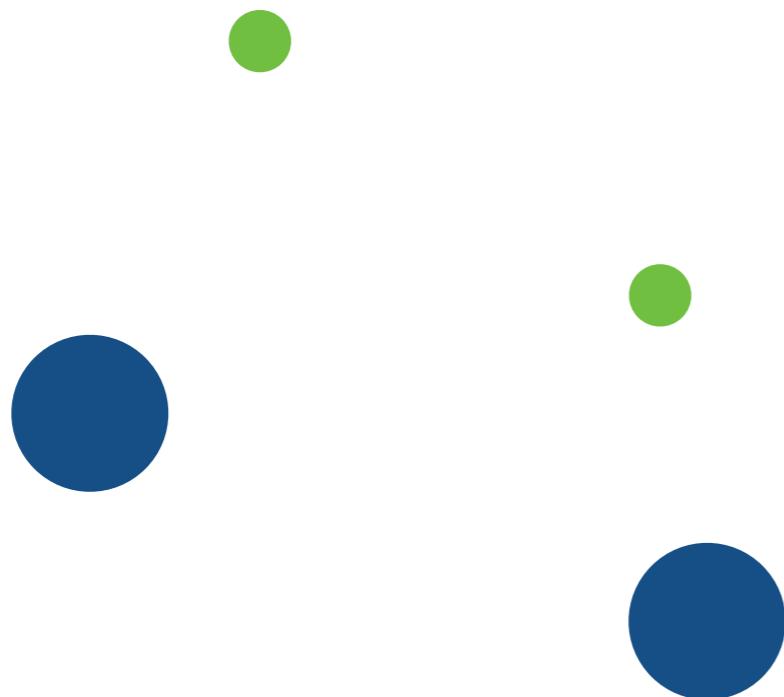


## What's missing?

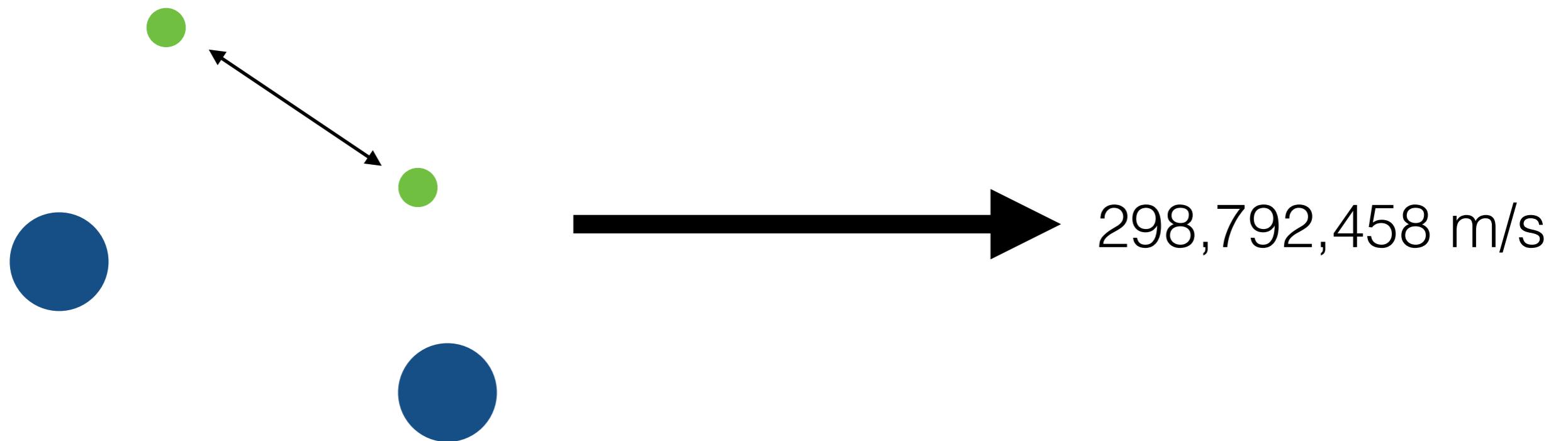
Spin doesn't show up in our Hamiltonian.

However, we need spin to describe magnetism.

**The origin of spin is actually really weird.**

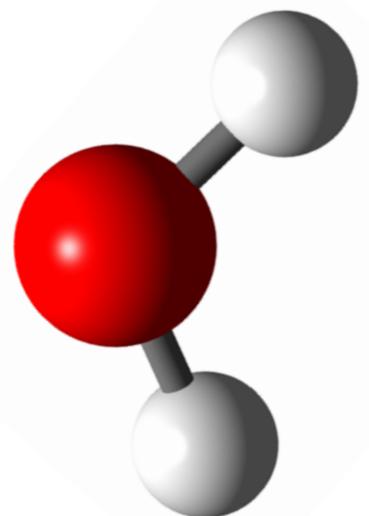


If we want to include spin, we actually have to think about how our equations change as we approach the **speed of light**.

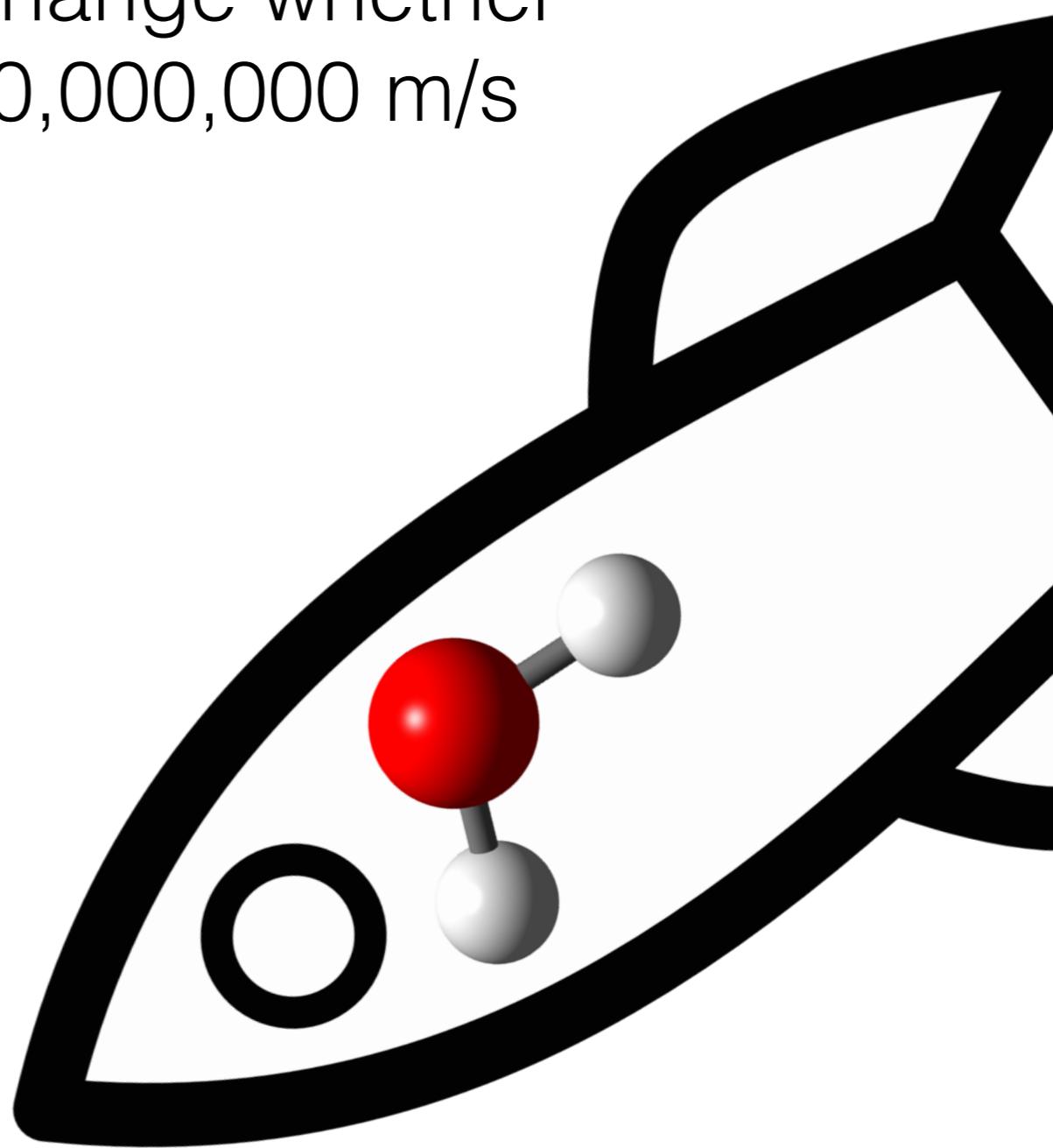


Einstein says our equations shouldn't change when we approach the speed of light

**Chemistry** shouldn't change whether we do it here or at 290,000,000 m/s



=



Mathematically, if require chemistry to be the same regardless of how fast we are moving...

**Spin** shows up in our Hamiltonian

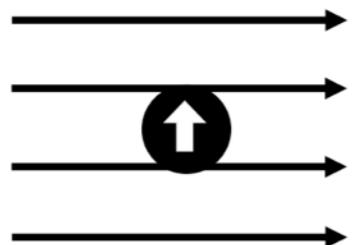
Interactions between spins show up

We get the proper description for  
**magnetism in molecules.**

Actually doing this, however, is a very hard problem.

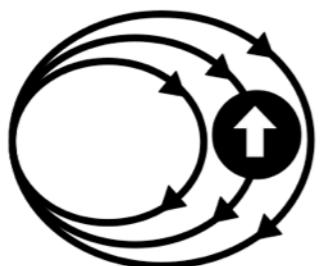
There are many types of spin interactions

***new physics***



*electric field*

+



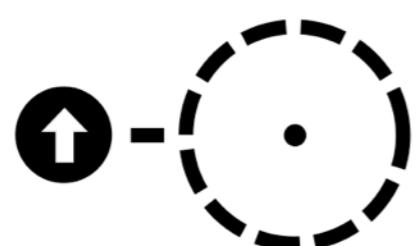
*magnetic field*

+



*spin-orbit*

+



*spin-other-orbit*

+



*spin-spin*

**limit**

non-relativistic



fully relativistic

# A table from the back of “Theoretical Foundations of Electron Spin Resonance” (John Harriman)

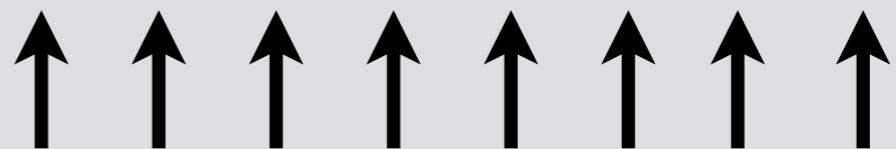
These are just some of the spin interactions we might consider

Term	Description	Operator	Atomic units
1.3	Electron-electron Darwin term	$\sum_{j < k} (-\pi \delta(\mathbf{r}_{jk}))$	$\alpha^2 \epsilon_0 a_0^3$
1.4	Electron-electron orbital interaction	$\sum_{j < k} \left[ +\frac{1}{2} \left( \frac{\nabla_j \cdot \nabla_k}{r_{jk}} - \frac{(\mathbf{r}_{jk} \cdot \nabla_j)(\mathbf{r}_{jk} \cdot \nabla_k)}{r_{jk}^3} \right) \right]$	$\alpha^2 \epsilon_0 a_0^3$
1.5	Electron-electron orbital interaction	$\sum_{j < k} \left[ +\frac{1}{4} \frac{1}{r_{jk}^3} (\mathbf{r}_{jk} \cdot \nabla_k - \mathbf{r}_{jk} \cdot \nabla_j) \right]$	$\alpha^2 \epsilon_0 a_0^3$
1.6	Electron-electron orbital interaction	$\sum_{j < k} [\pi \delta(\mathbf{r}_{jk})(\mathbf{r}_{jk} \cdot \nabla_j - \mathbf{r}_{jk} \cdot \nabla_k)]$	$\alpha^2 \epsilon_0 a_0^3$
Terms depending on spin but not field (Electron spin)			
2.1	Spin-orbit interaction	$\sum_{j, v} \left( -\frac{g'}{4} Z_v i \frac{k_0^2(\mathbf{r}_{jv})}{r_{jk}^3} \mathbf{S}_j \cdot (\mathbf{r}_{jv} \times \nabla_j) \right)$	$\alpha^2 \epsilon_0 a_0^3$
2.2	Spin-other-orbit interaction	$\sum_{j, k} \frac{i}{r_{jk}^3} \mathbf{S}_j \cdot [(\mathbf{r}_{kj}) \times \nabla_k]$	$\alpha^2 \epsilon_0 a_0^3$
2.3	Electron-electron spin-orbit interaction	$\sum_{j, k} \left( \frac{g'}{4} \frac{i}{r_{jk}^3} \mathbf{S}_j \cdot \mathbf{r}_{jk} \times \nabla_j \right)$	$\alpha^2 \epsilon_0 a_0^3$
2.4	Spin-spin dipolar interaction	$\sum_{j, k} \left( \frac{(\mathbf{S}_j \cdot \mathbf{S}_k)}{r_{jk}^3} - \frac{(\mathbf{S}_j \cdot \mathbf{r}_{jk})(\mathbf{S}_k \cdot \mathbf{r}_{jk})}{r_{jk}^5} \right)$	$\alpha^2 \epsilon_0 a_0^3$
Terms depending on spin and field (Nuclear spin)			
2.5	Spin-spin contact interaction	$\sum_{j < k} \left( -\frac{8\pi}{3} \mathbf{S}_j \cdot \mathbf{S}_k \delta(\mathbf{r}_{jk}) \right)$	$\alpha^2 \epsilon_0 a_0^3$
2.6	Nuclear dipole-dipole interaction	$\sum_{v < v'} \frac{g_v g_{v'}}{4} \left[ \frac{\mathbf{I}_v \cdot \mathbf{I}_{v'}}{R_{vv'}^3} - 3 \frac{(\mathbf{I}_v \cdot \mathbf{R}_{vv})(\mathbf{I}_{v'} \cdot \mathbf{R}_{vv'})}{R_{vv'}^5} \right]$	$\left( \frac{m}{M_p} \right)^2 \alpha^2 \epsilon_0 a_0^3$
2.7	Nuclear quadrupole interaction	$\sum_{j, v} \left[ \frac{Q_v}{2I_v(2I_v - 1)} \right] \left[ \frac{I_v^2}{r_{jv}^3} - 3 \frac{(\mathbf{r}_{jv} \cdot \mathbf{I}_v)^2}{r_{jv}^5} \right]$	$\epsilon a_0$
Orbital hyperfine interaction			
	Orbital hyperfine correction	$\sum_{j, v} \frac{-g_v}{2} k_0(r_{jv}) \frac{i}{r_{jv}^3} (\mathbf{r}_{jv} \times \nabla_j) \cdot \mathbf{I}_v$	$\left( \frac{m}{M_p} \right) \alpha^2 \epsilon_0 a_0^3$
Electron coupled nuclear spin-spin interaction			
	Dipolar hyperfine interaction	$\sum_{j, v, v'} \frac{g_v g_{v'}}{8} \frac{k_0(r_{vj})}{r_{jv}^3 r_{jv'}^3} \times [(\mathbf{I}_v \cdot \mathbf{I}_{v'})(\mathbf{r}_{jv} \cdot \mathbf{r}_{jv'}) - (\mathbf{I}_v \cdot \mathbf{r}_{jv})(\mathbf{I}_{v'} \cdot \mathbf{r}_{jv'})]$	$\left( \frac{m}{M_p} \right)^2 \alpha^4 \epsilon a_0^4$
and nuclear spins)			
Fermi) contact hyperfine interaction			
		$\sum_{j, v} \left( -\frac{gg_v}{4} k_0(r_{jv}) \left[ \frac{\mathbf{S}_{jv} \cdot \mathbf{I}_v}{r_{jv}^3} - 3 \frac{(\mathbf{S}_{jv} \cdot \mathbf{r}_{jv})(\mathbf{I}_v \cdot \mathbf{r}_{jv})}{r_{jv}^5} \right] \right)$	$\left( \frac{m}{M_p} \right) \alpha^2 \epsilon$
		$\sum_{j, v} \frac{2\pi}{3} g' g_v \mathbf{S}_j \cdot \mathbf{I}_v \delta(\mathbf{r}_{jv})$	$\left( \frac{m}{M_p} \right) \alpha^2$

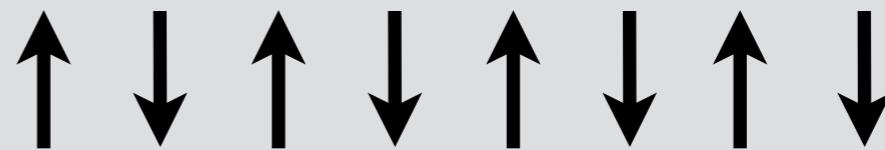
All magnetic materials have unpaired electrons.

How the spins order determines the type of magnet (at 0K).

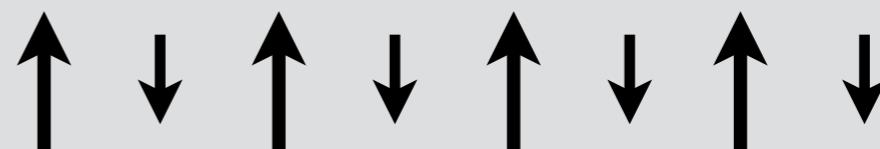
Ferromagnetic



Antiferromagnetic



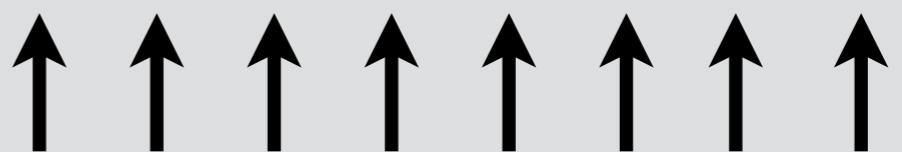
Ferrimagnetic



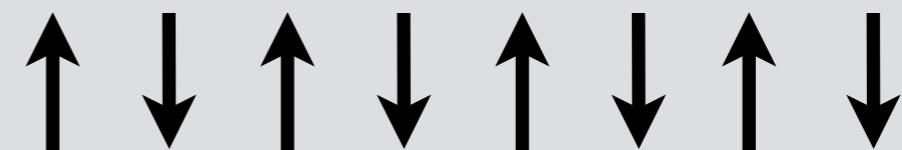
All these states are collinear.

Spins point either up or down.

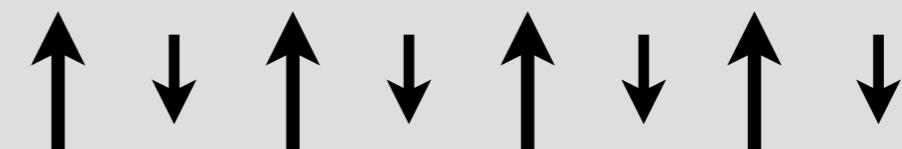
Ferromagnetic



Antiferromagnetic



Ferrimagnetic



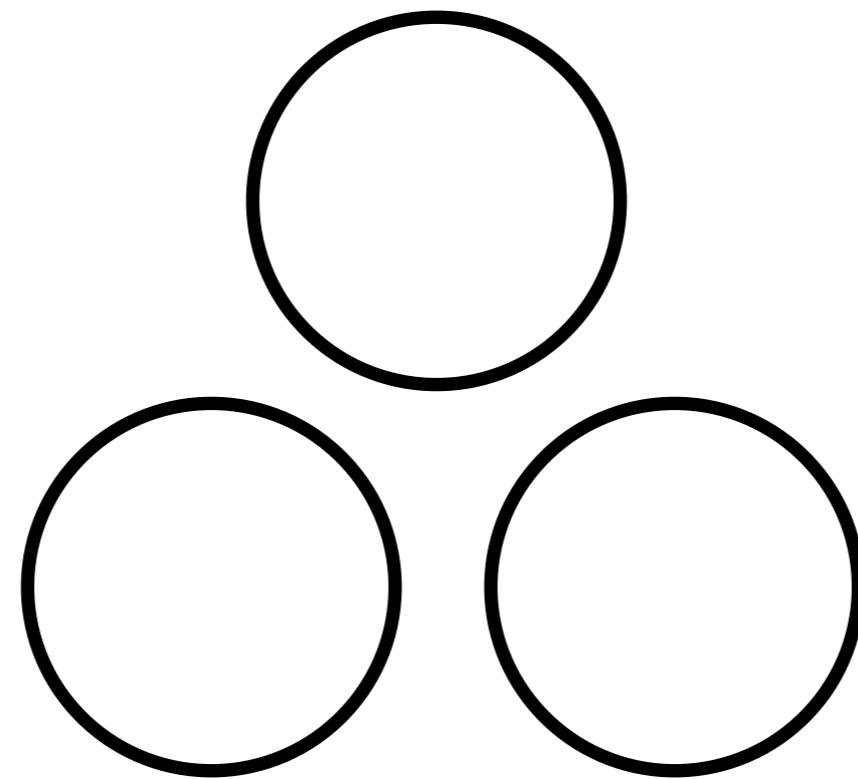
Heating up a magnet past its critical temperature gives us a disordered state

Paramagnetic



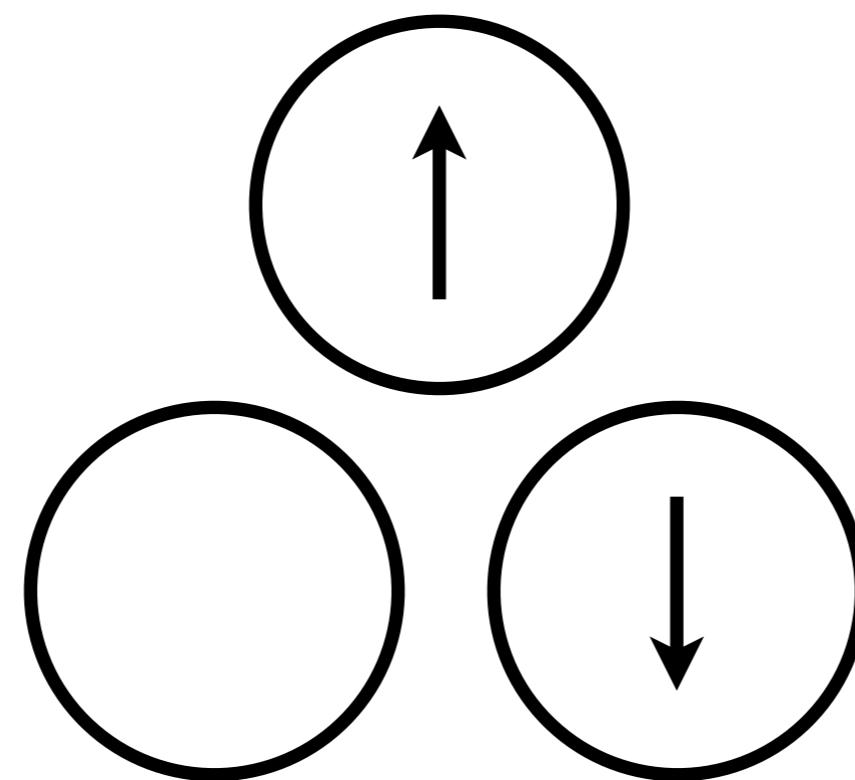
Is it possible to obtain a disordered state at absolute zero?

Spin frustration leads to non-collinear magnetism



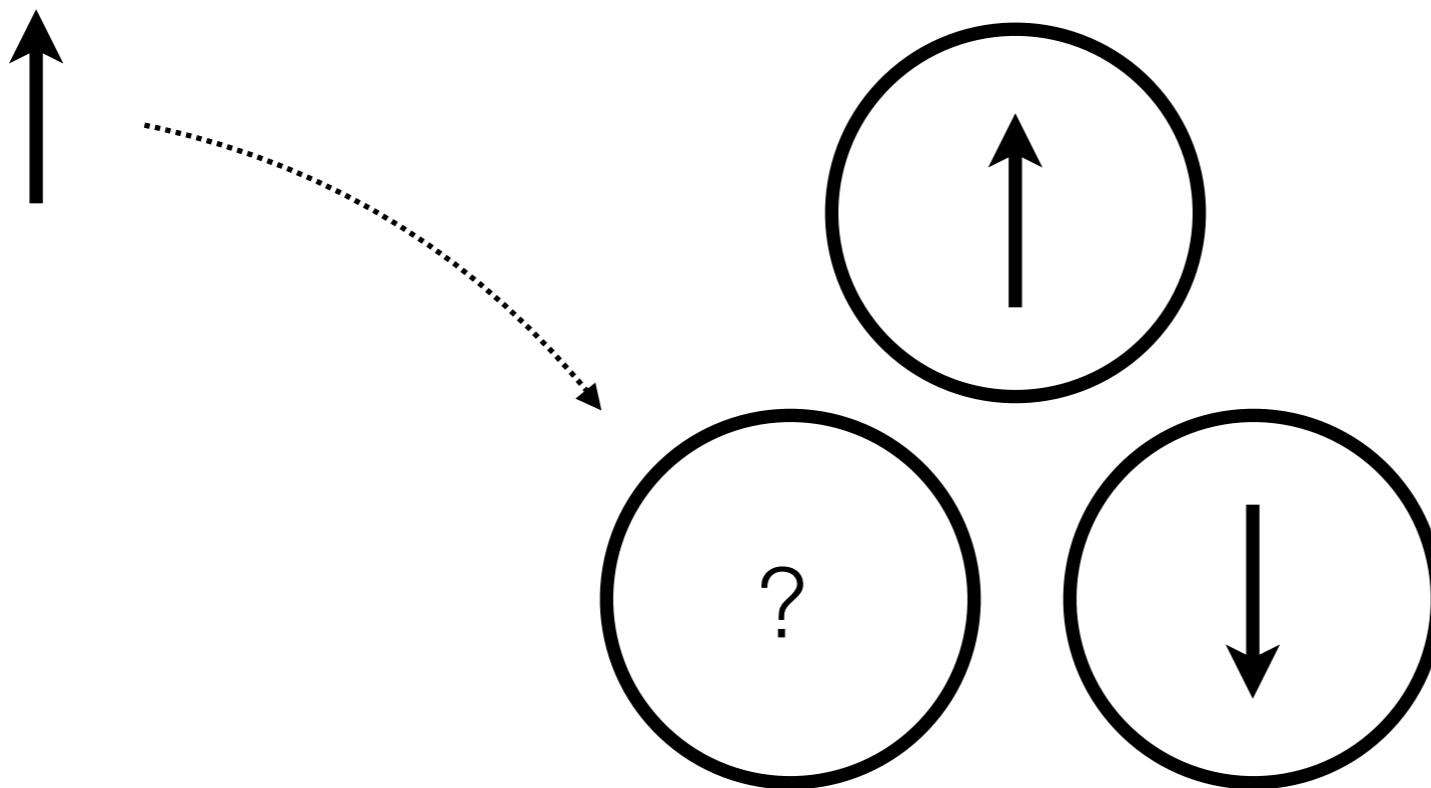
Take a three site lattice

Spin frustration leads to non-collinear magnetism



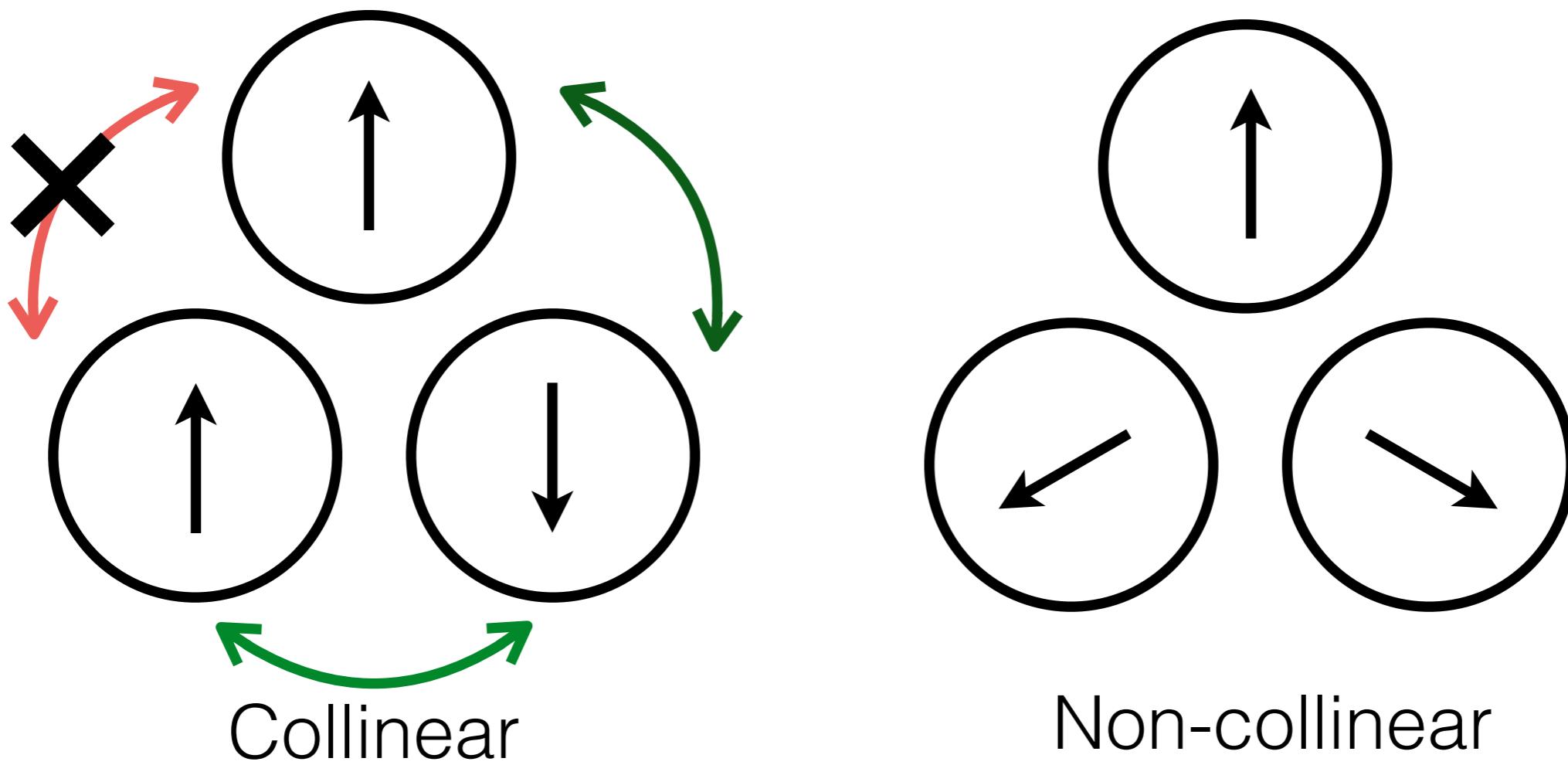
Add two electrons.  
(Assume anti-ferromagnetism favored).

Spin frustration leads to non-collinear magnetism



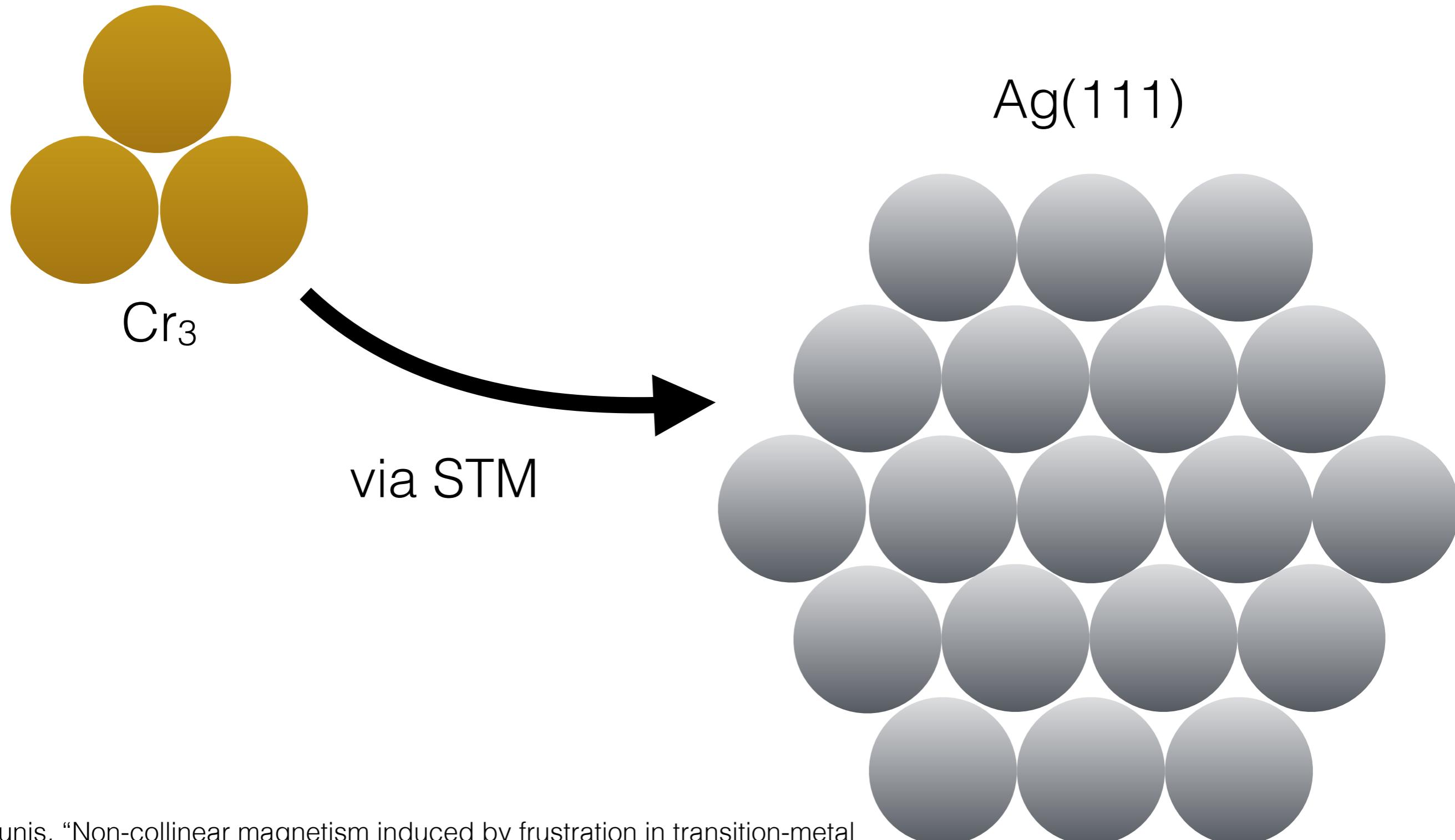
Now, add the third electron.  
No spin orientation simultaneously favors  
all anti-ferromagnetic exchange interactions

Spin frustration leads to non-collinear magnetism

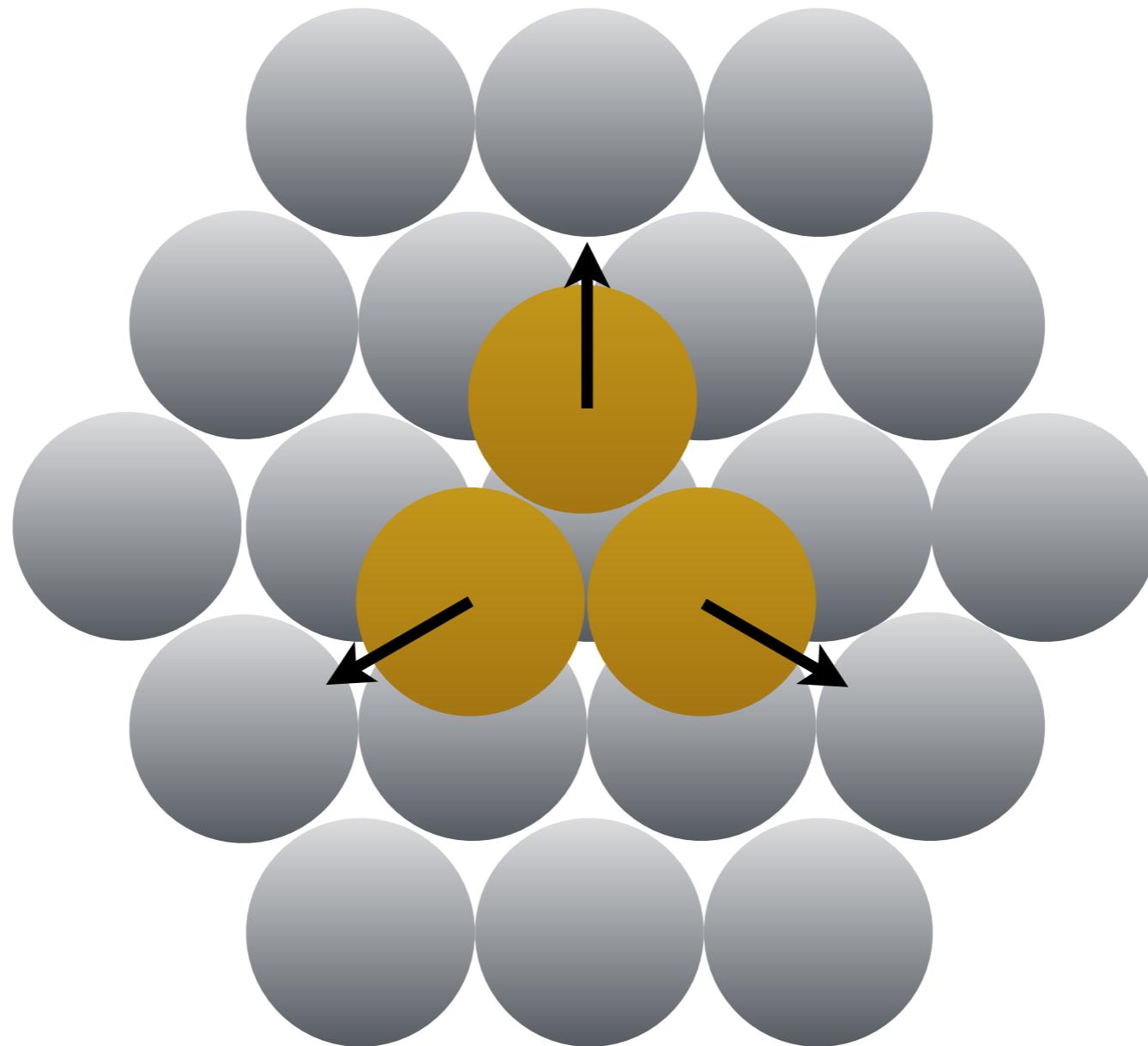


Most electronic structure methods  
cannot capture non-collinearity!

# Generating non-collinear magnetism by spin frustration with transition metals



# Generating non-collinear magnetism by spin frustration with transition metals



Lock metal trimer into triangle

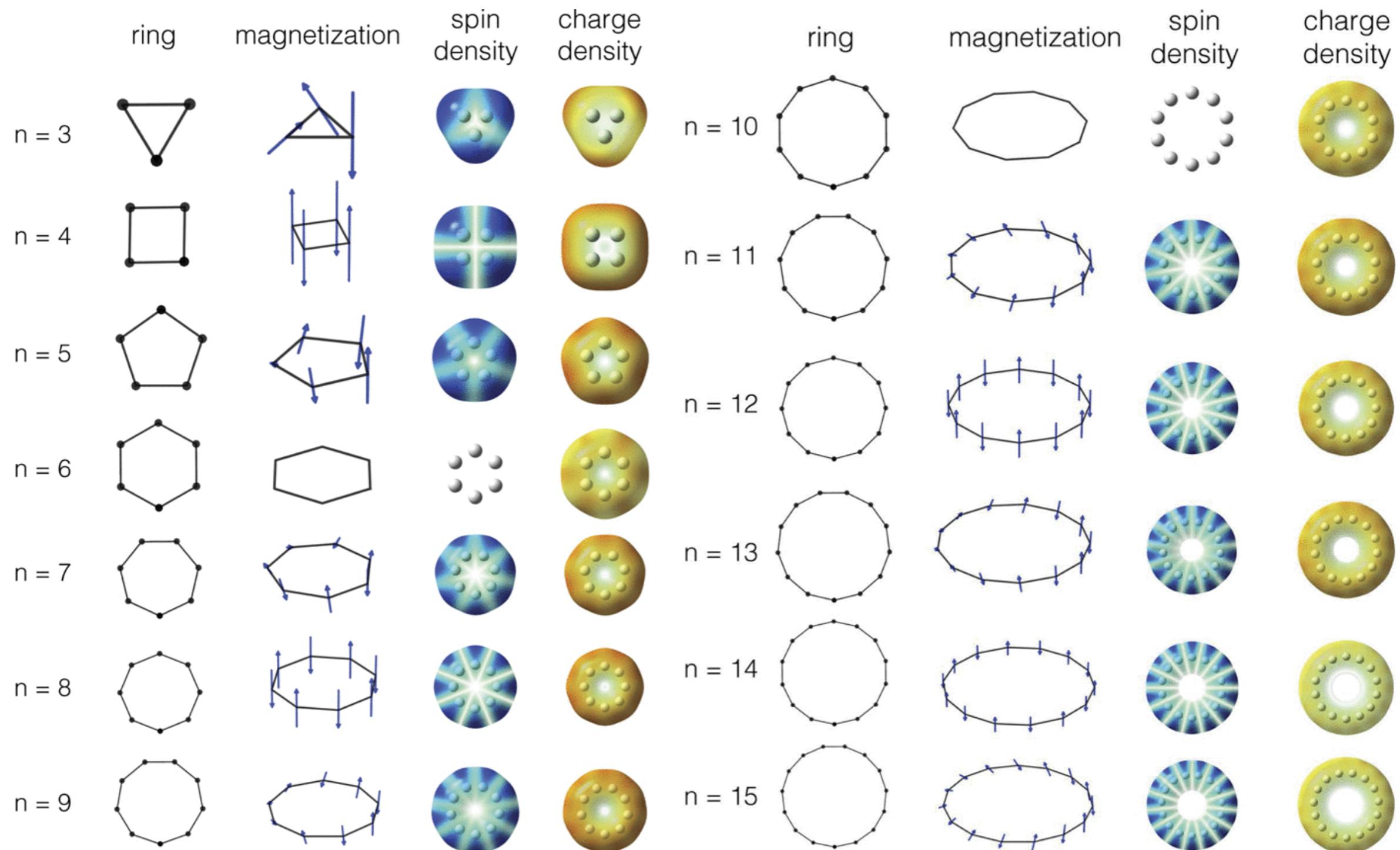
# Stability of the complex generalized Hartree-Fock equations

Joshua J. Goings,<sup>1</sup> Feizhi Ding,<sup>1</sup> Michael J. Frisch,<sup>2</sup> and Xaosong Li<sup>1,a)</sup>

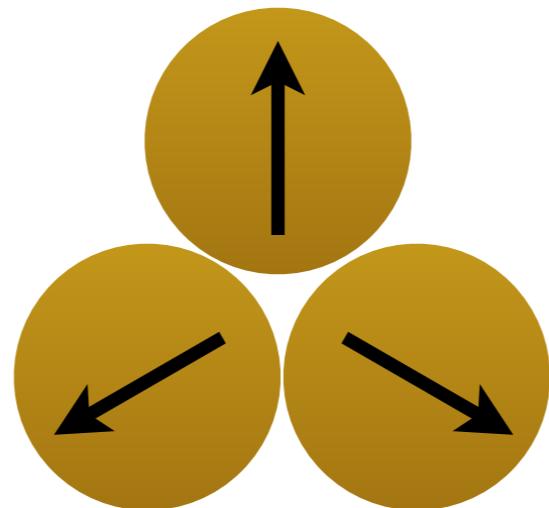
<sup>1</sup>Department of Chemistry, University of Washington, Seattle, Washington 98195, USA

<sup>2</sup>Gaussian, Inc., 340 Quinnipiac St., Bldg 40, Wallingford, Connecticut 06492, USA

(Received 19 December 2014; accepted 2 April 2015; published online 16 April 2015)



These molecules are great for testing out ideas about magnetism because they have many competing magnetic interactions



This understanding helps us understand more complicated interactions in quantum dots.

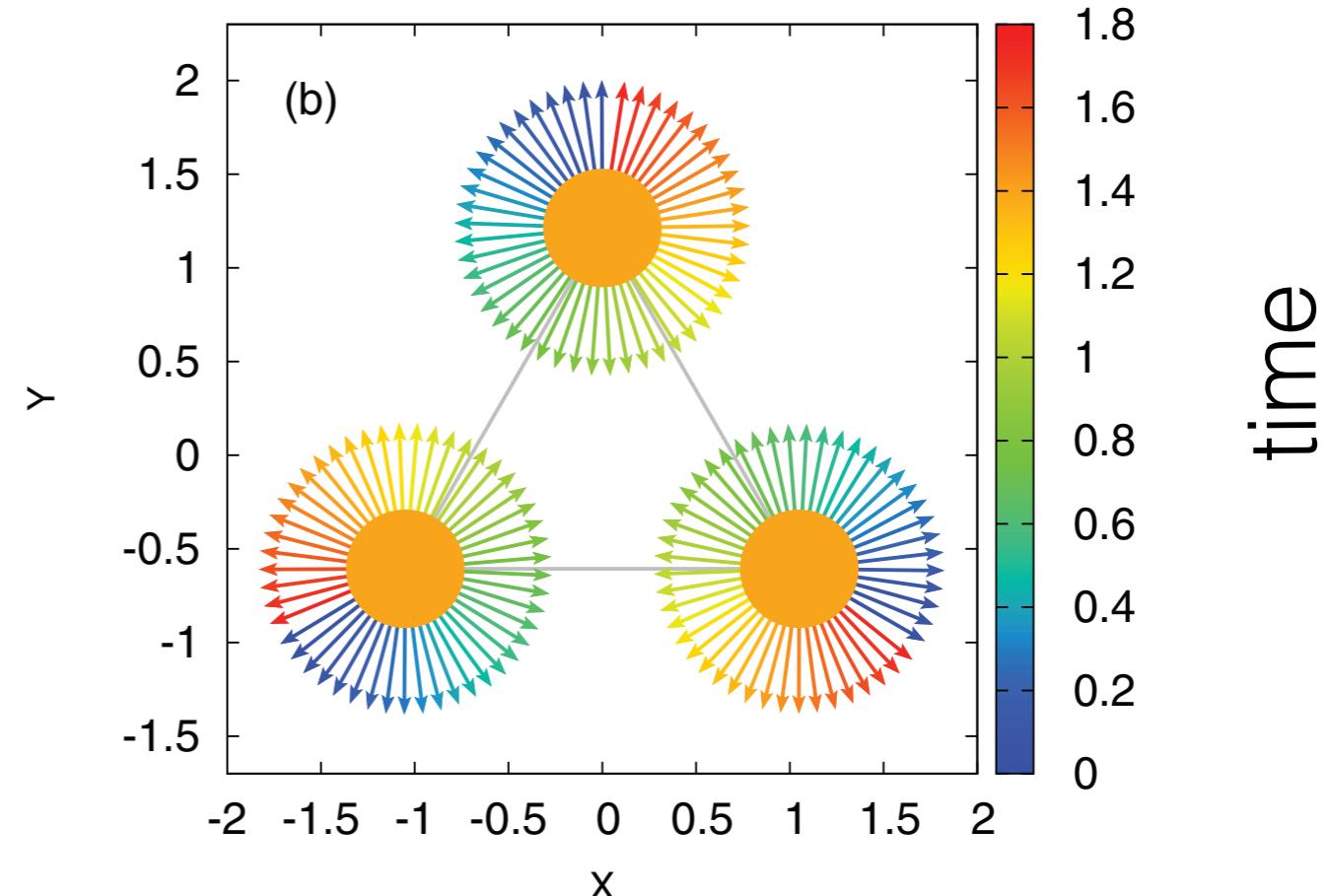
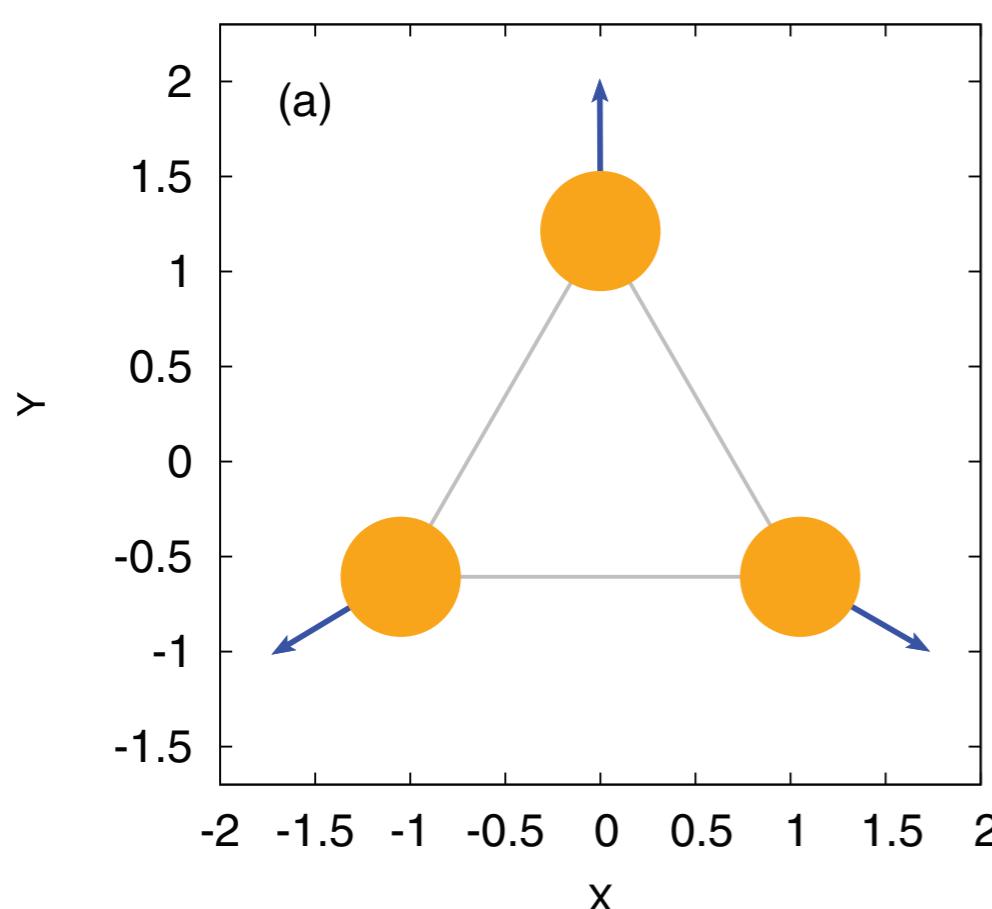
## Ab initio non-relativistic spin dynamics

Feizhi Ding,<sup>1</sup> Joshua J. Goings,<sup>1</sup> Michael J. Frisch,<sup>2</sup> and Xaosong Li<sup>1,a)</sup>

<sup>1</sup>*Department of Chemistry, University of Washington, Seattle, Washington 98195, USA*

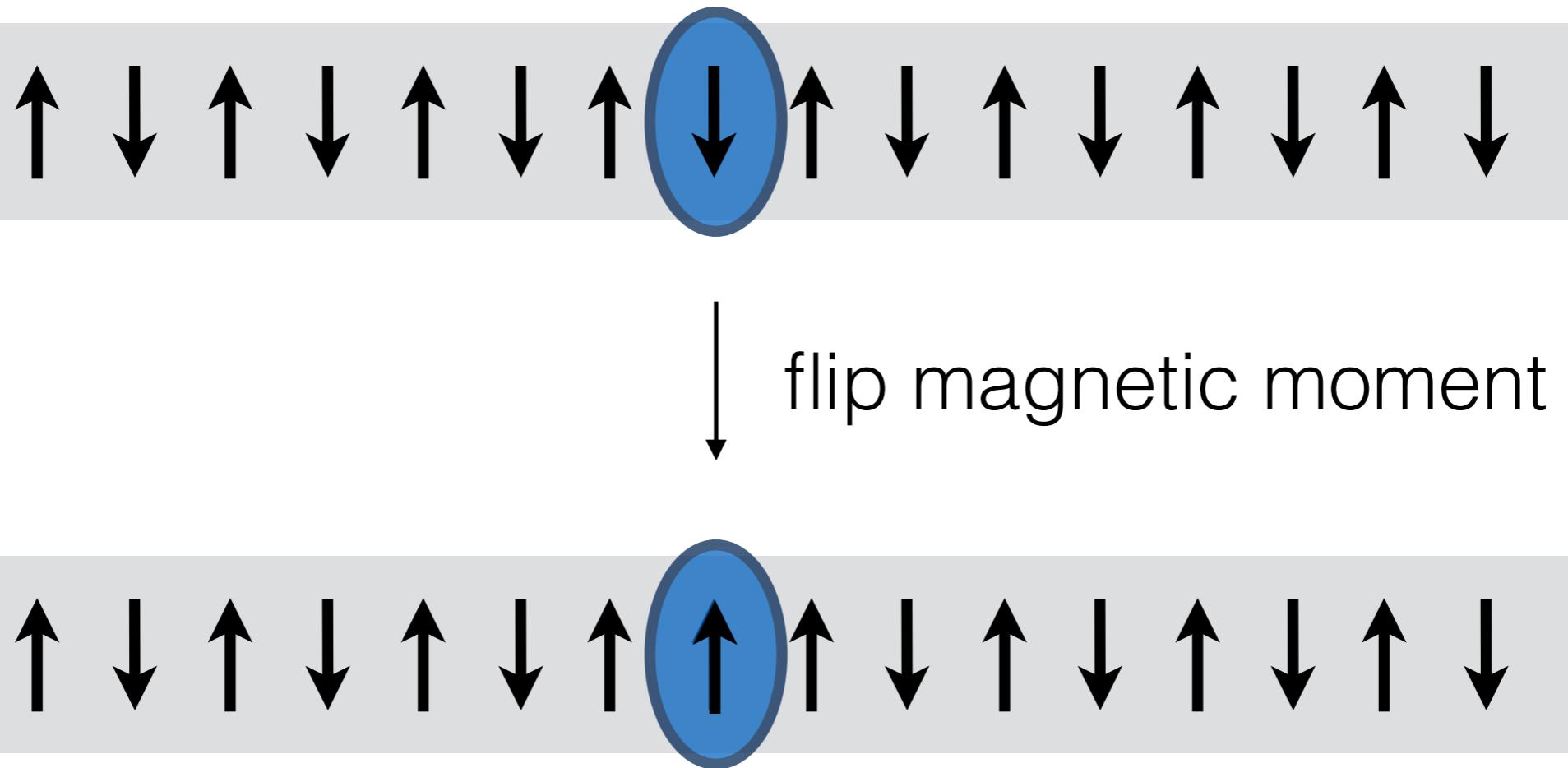
<sup>2</sup>*Gaussian, Inc., 340 Quinnipiac St, Bldg 40, Wallingford, Connecticut 06492, USA*

(Received 29 August 2014; accepted 17 November 2014; published online 4 December 2014)



Applying a magnetic field (into page)  
causes spins to precess

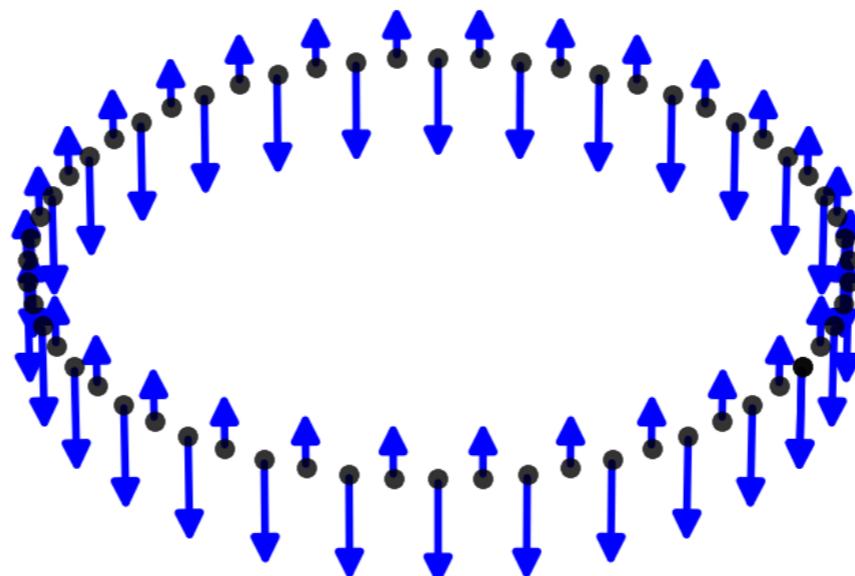
Here we tried to model spin waves.



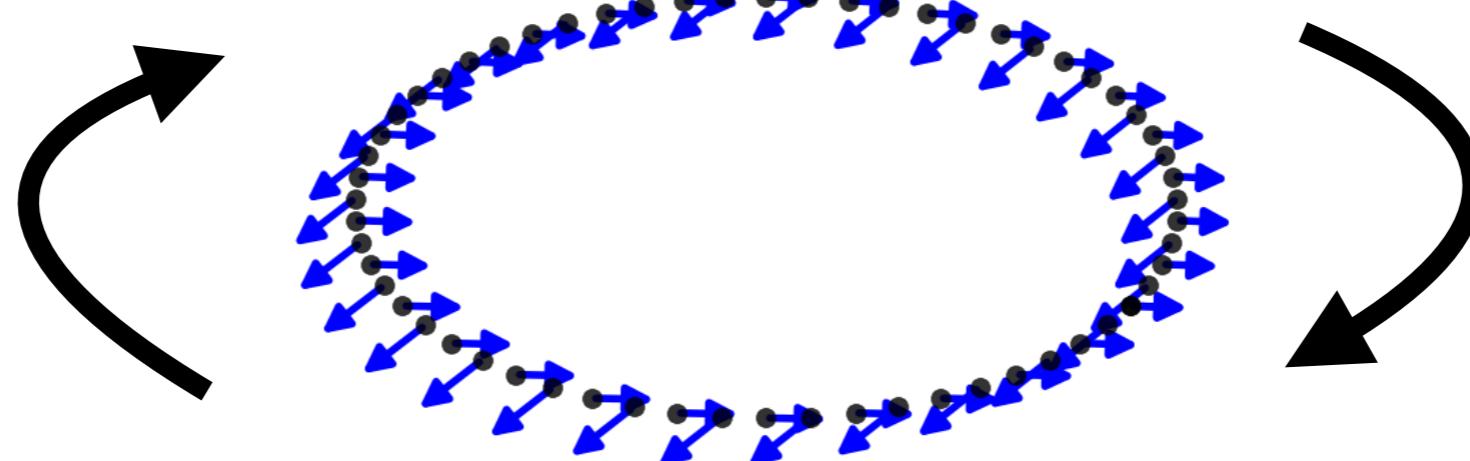
...and watch it go!

(movie)

# Initial Spin-Flipped State

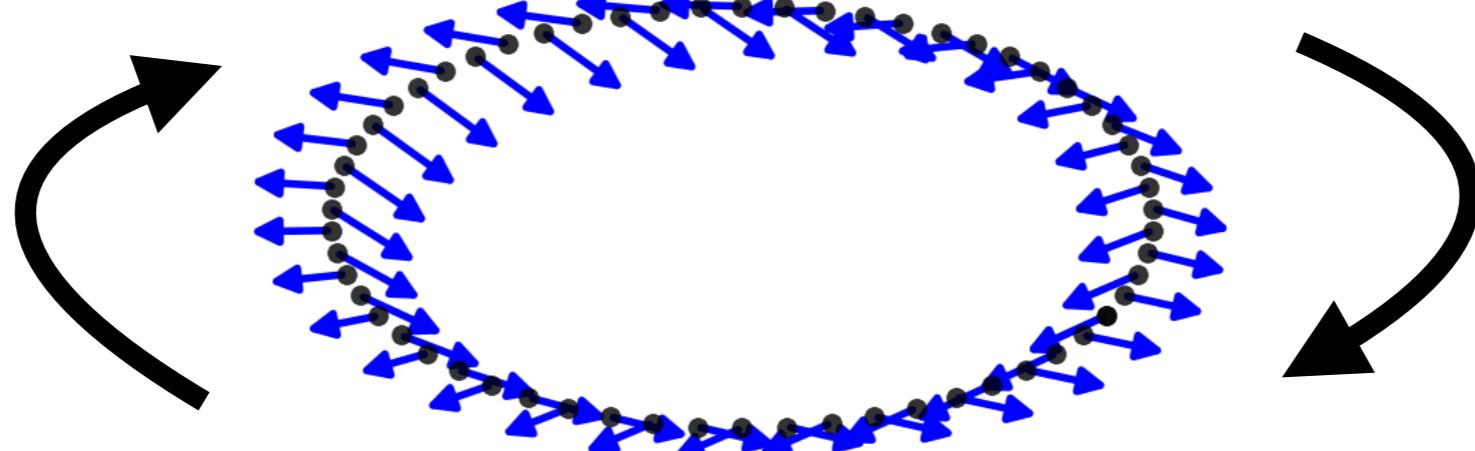


$t \sim 20 \text{ fs}$

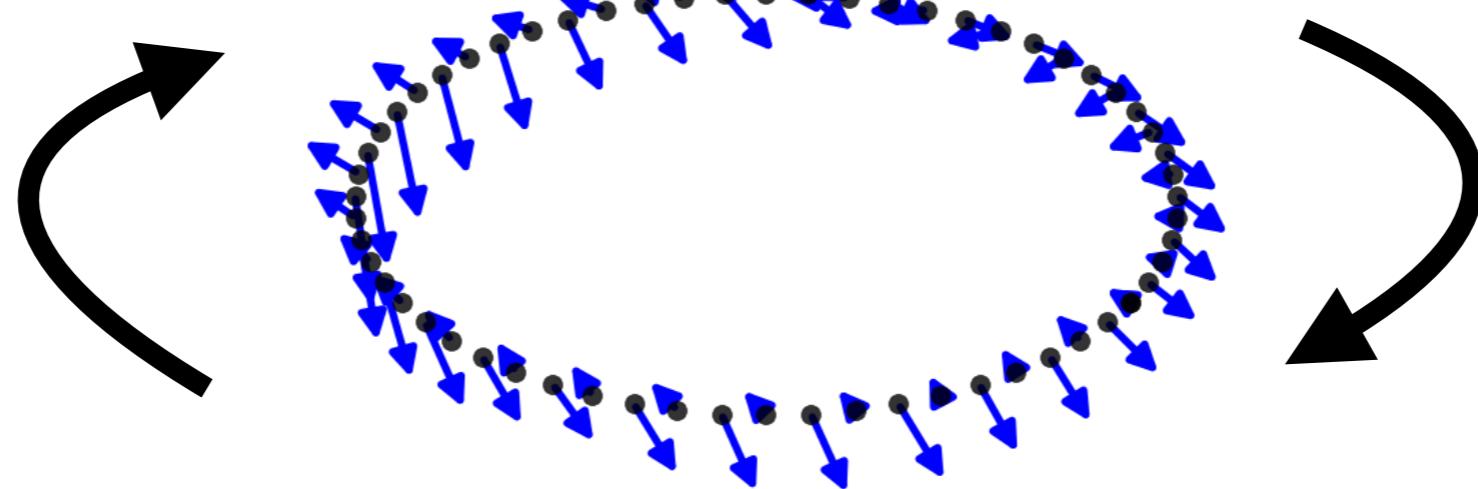


**20 fs** is to **1 s**  
as  
**1 s** is to **one million years**

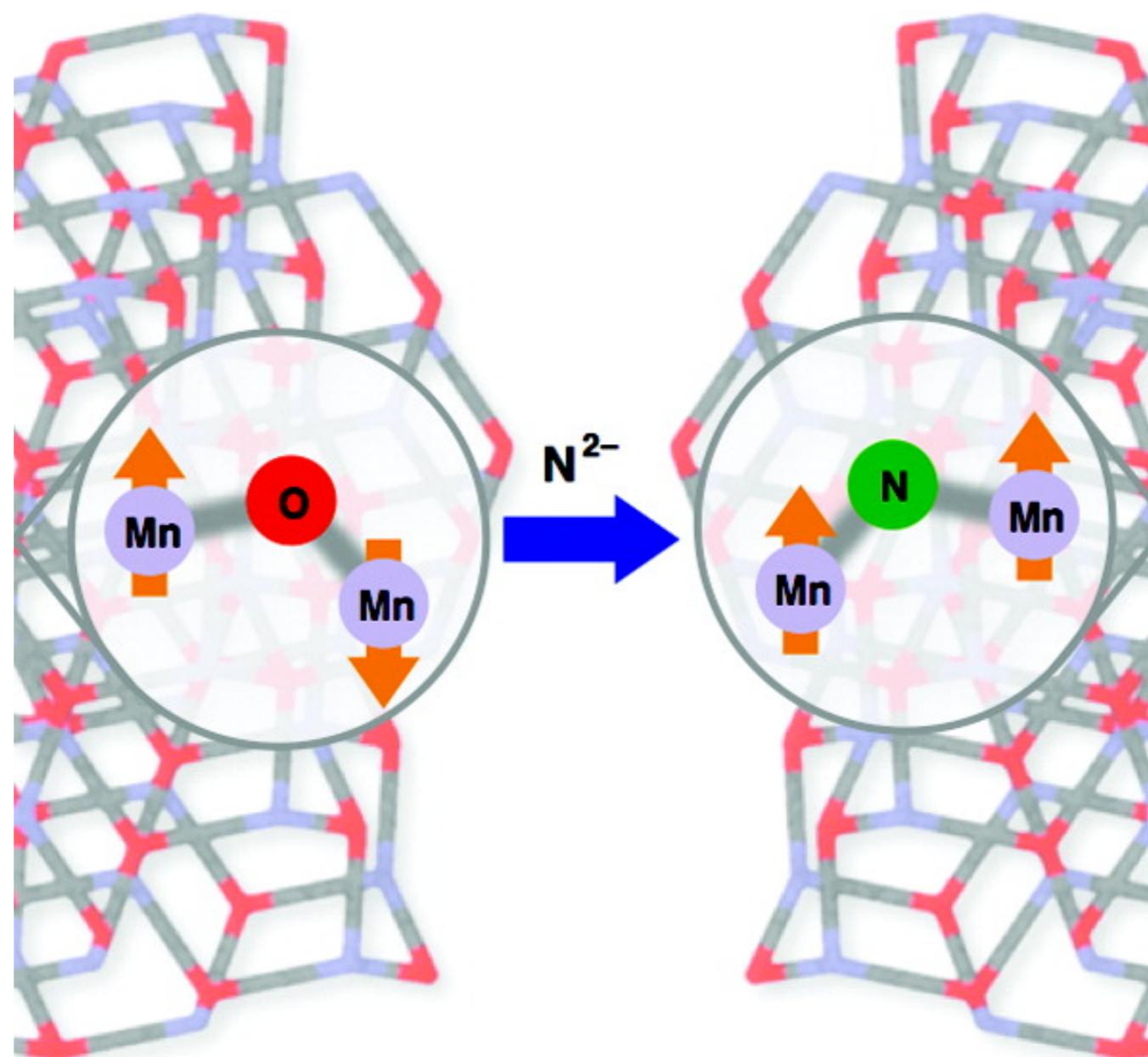
$t \sim 40 \text{ fs}$



$t \sim 60 \text{ fs}$



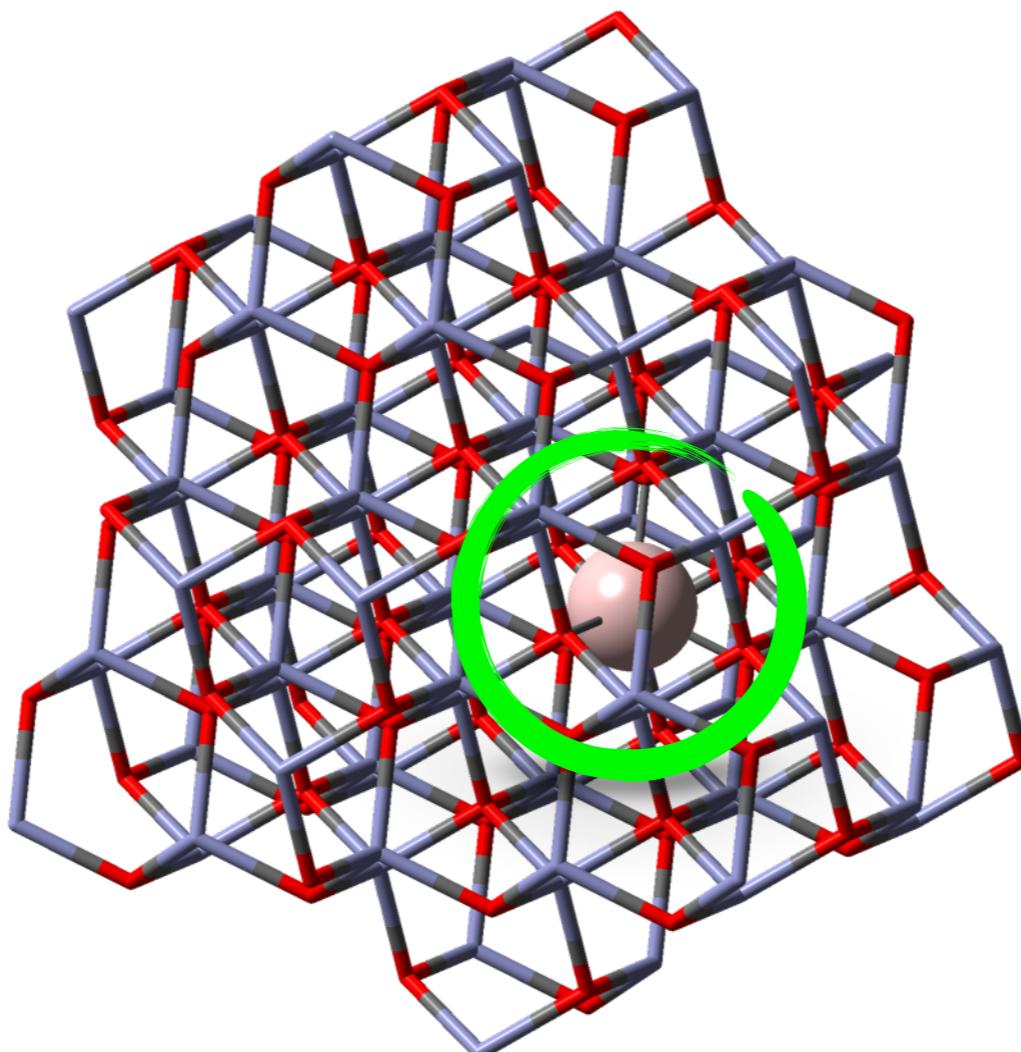
All these tools give us complementary information to interpret and explain what we see in the chemistry lab.



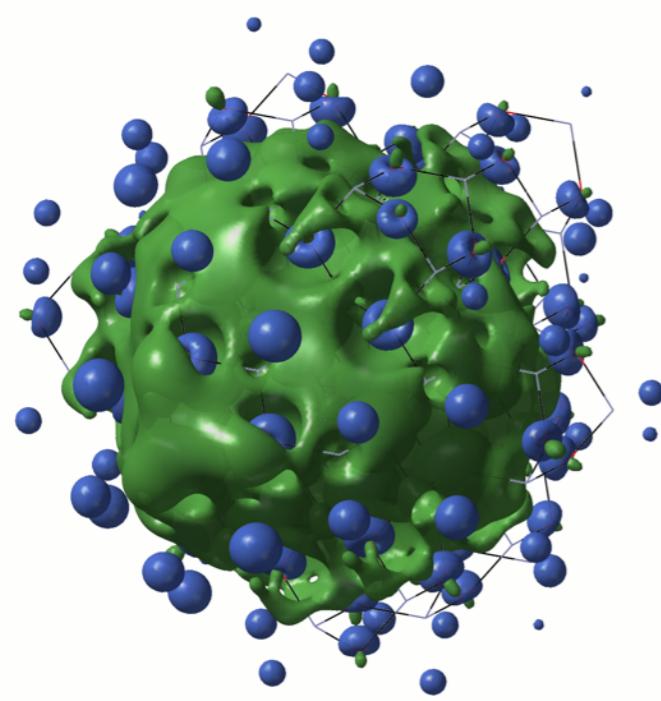
## Theoretical Characterization of Conduction-Band Electrons in Photodoped and Aluminum-Doped Zinc Oxide (AZO) Quantum Dots

Joshua J. Goings, Alina M. Schimpf, Joseph W. May, Robert W. Johns, Daniel R. Gamelin,\*  
and Xiaosong Li\*

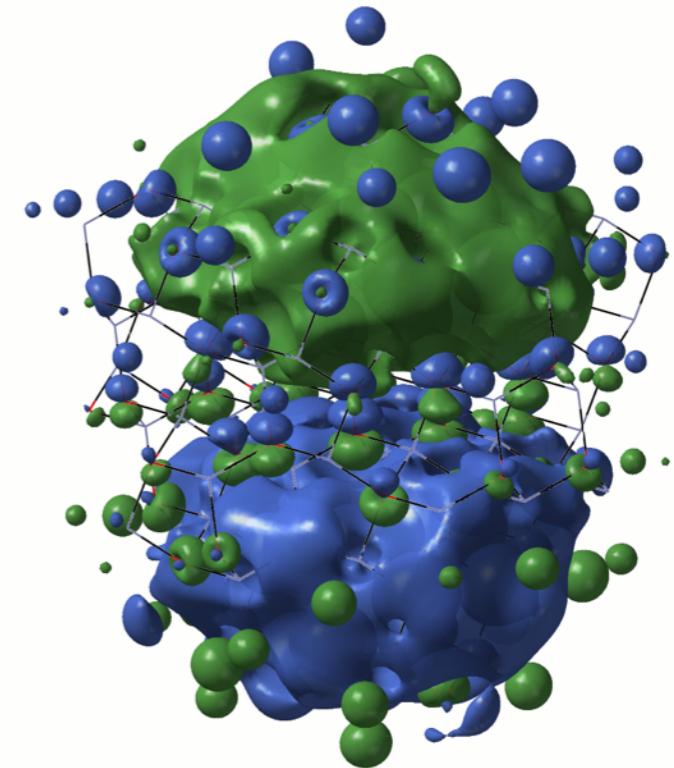
Department of Chemistry, University of Washington, Seattle, Washington 98195, United States



Give the quantum dots an extra electron by doping with aluminum...

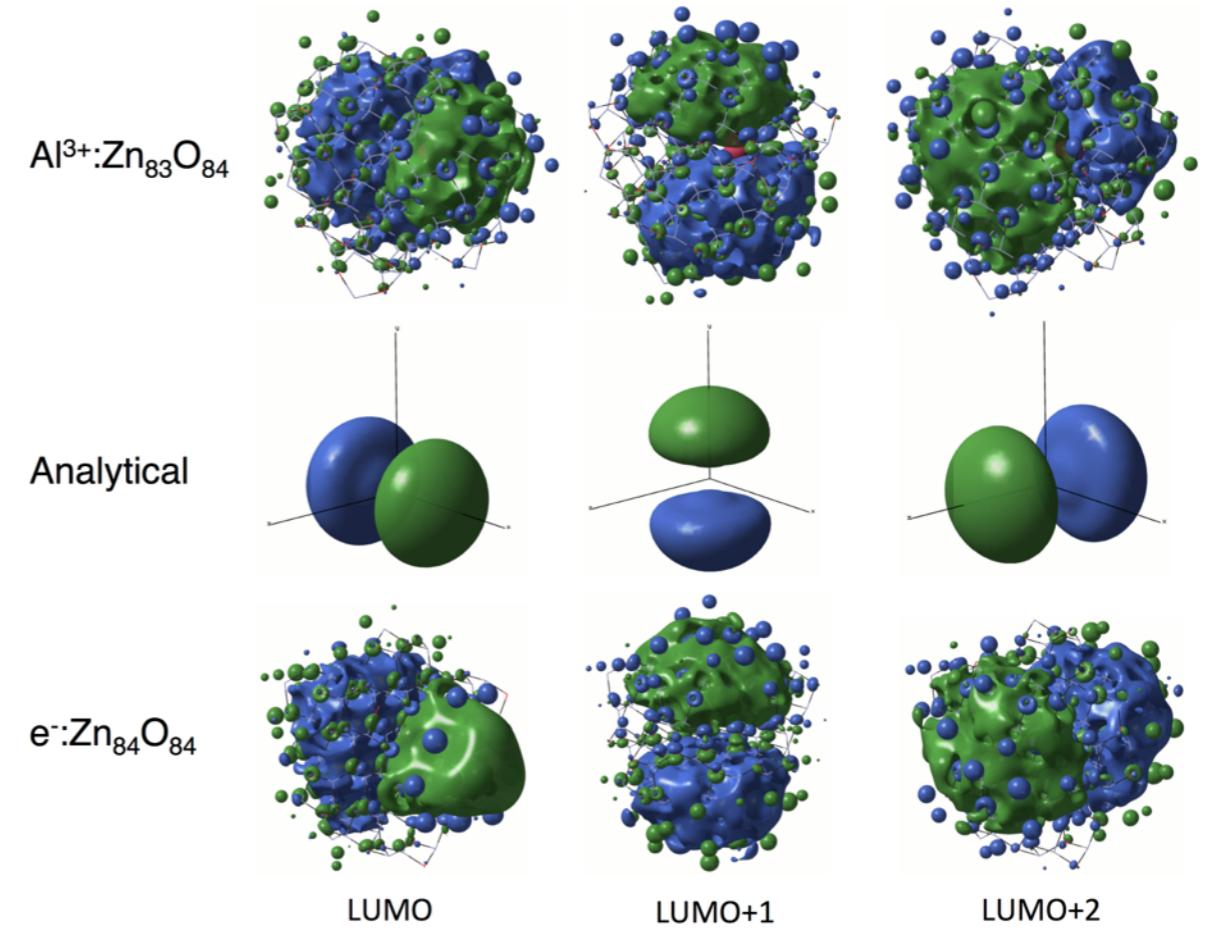
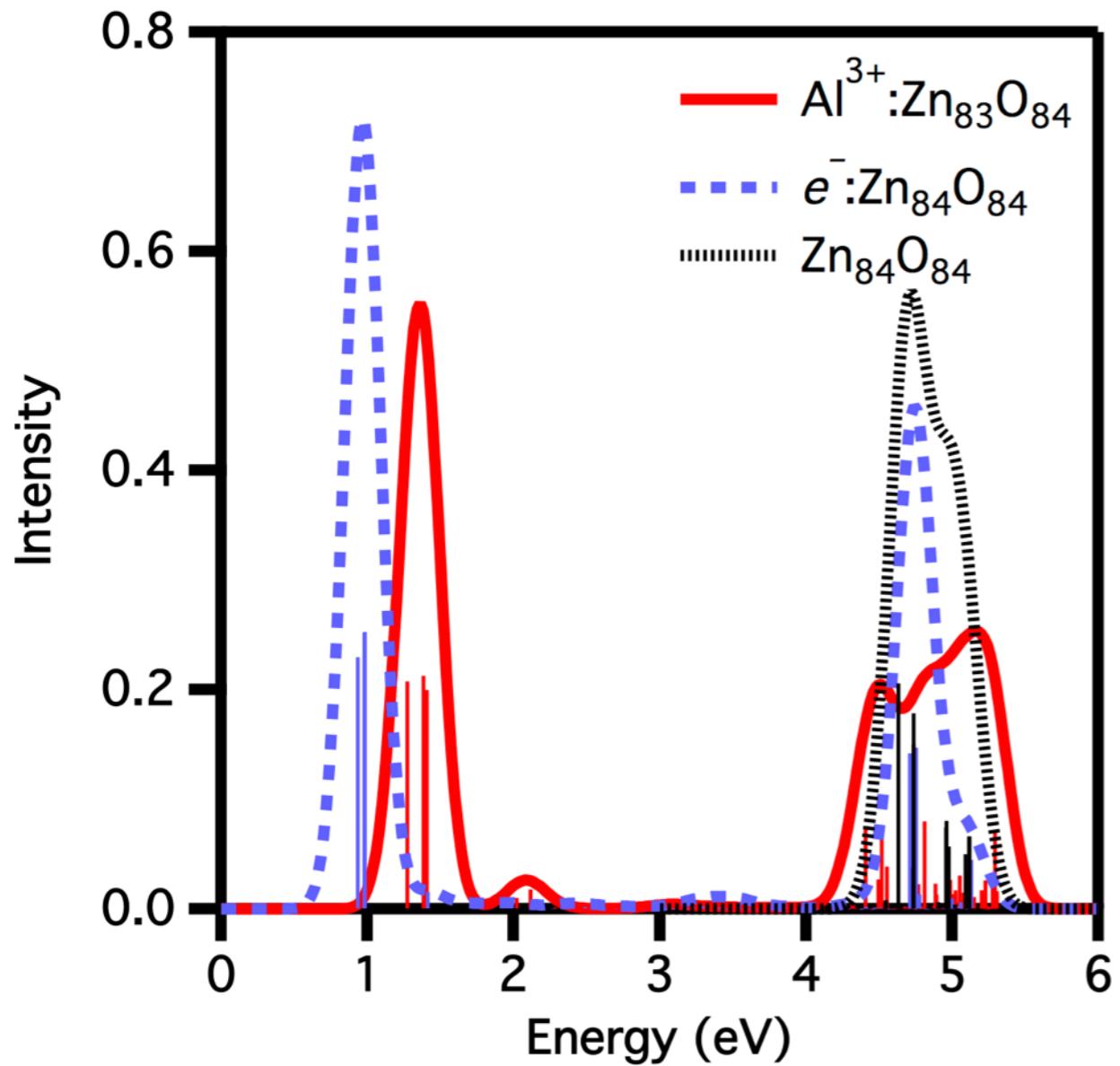


HOMO (s)



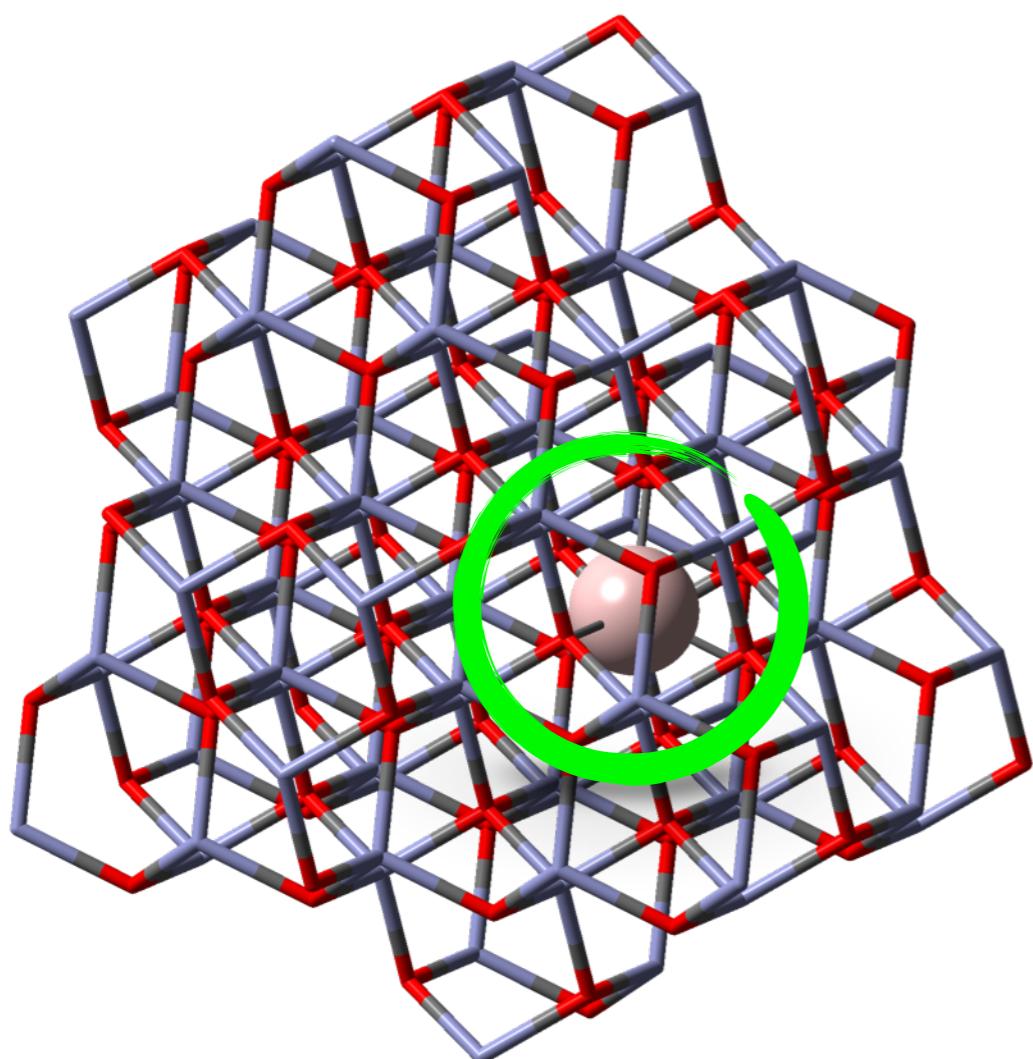
LUMO (p)

Quantum dot gets ‘atomic’ orbitals (“Super-orbitals”)



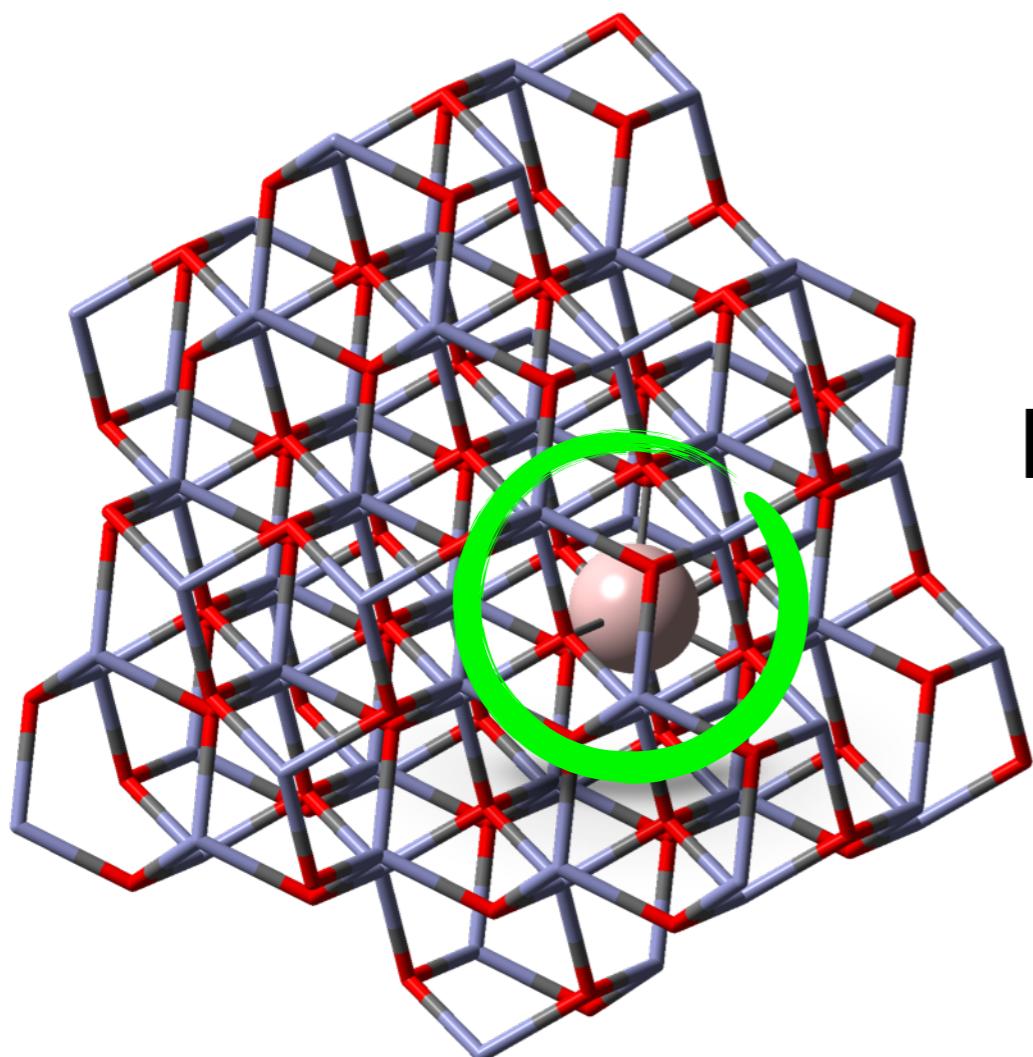
The “extra” electron carries a spin.  
It *should* interact with magnetic centers.

We think it may make a good room-temperature magnet



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**Part of the fun is seeing if it will!**

