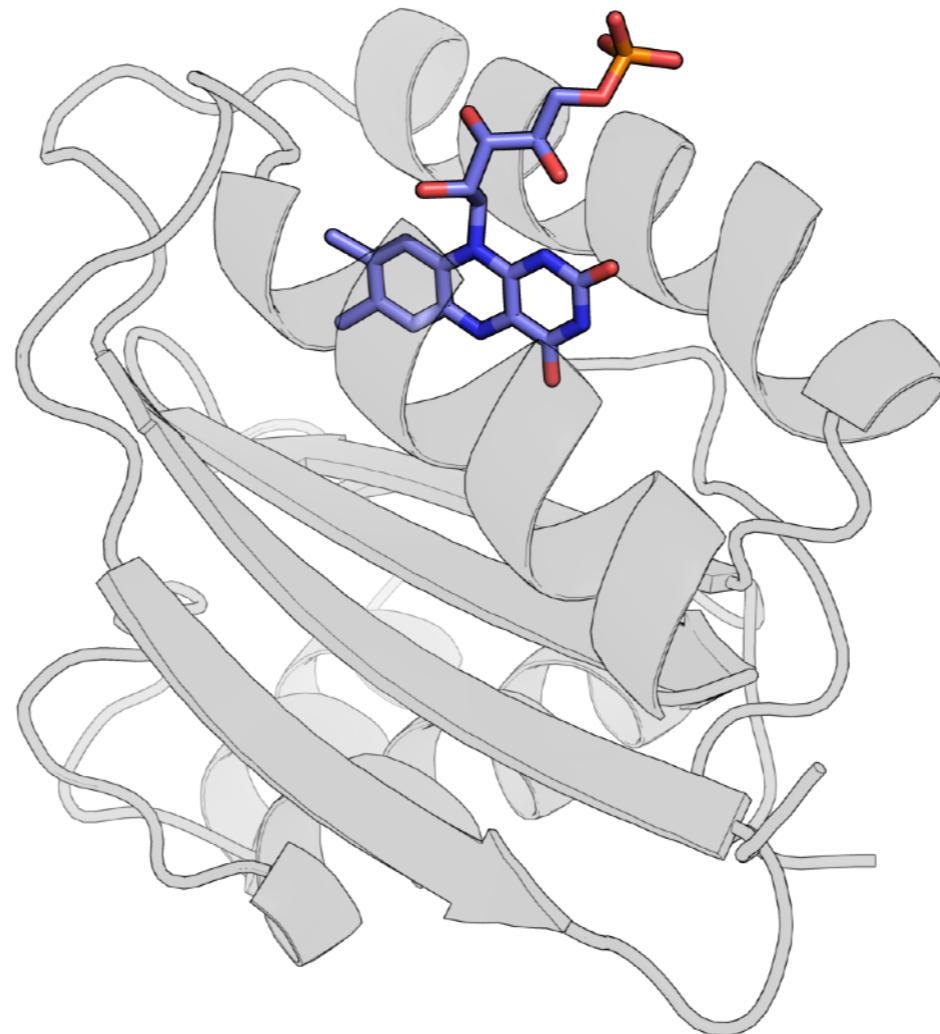


Impact of Protein Environment on Slr1694 BLUF Photoreceptor

**Joshua Goings
October 4, 1:00 pm**

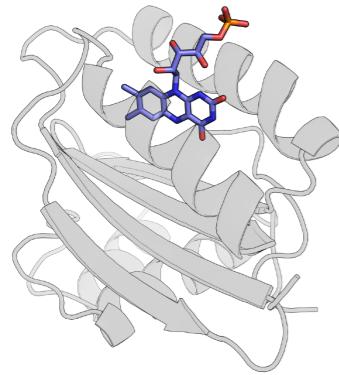
Photoreceptor domain BLUF



Blue Light Using Flavin

BLUF uses a flavin chromophore to convert a light signal into a biological response.

Photoreceptor domain BLUF



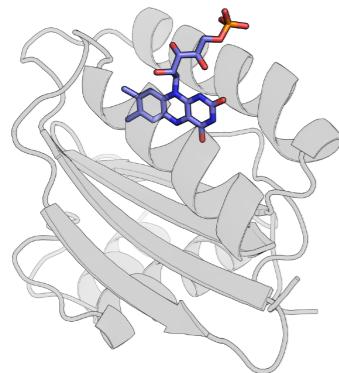
BLUF

Blue light using flavin

BLUF uses a flavin to convert a light signal into a biological response.

Found in photosynthetic bacteria, but also some single-celled eukaryotes.

Photoreceptor domain BLUF

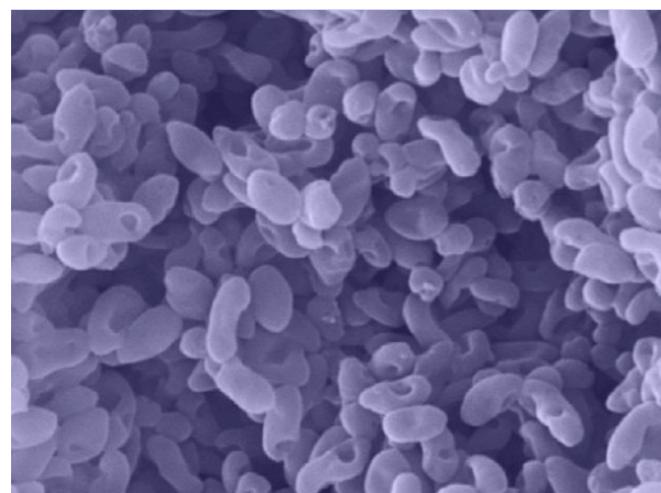


BLUF uses a flavin to convert a light signal into a biological response.

Found in photosynthetic bacteria, but also some single-celled eukaryotes.

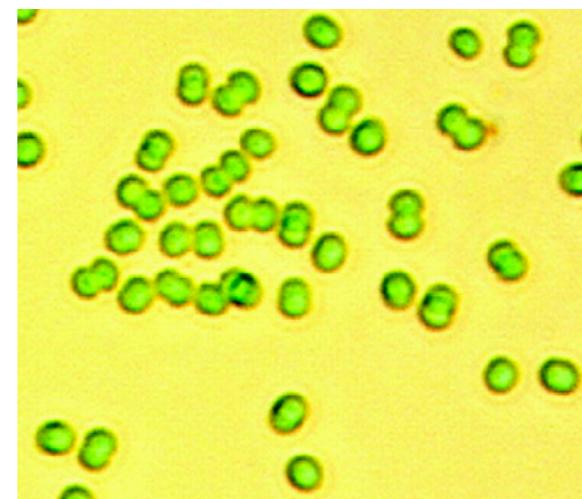
BLUF

Blue light using flavin



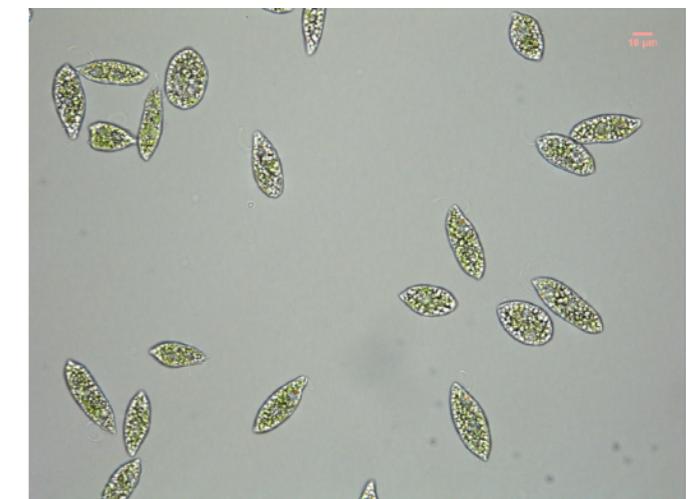
Purple Bacteria

R. sphaeroides



Cyanobacteria

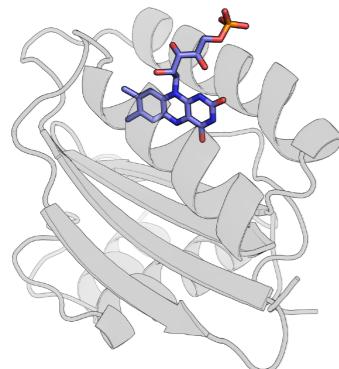
S. sp. PCC6803



Algae

E. gracilis

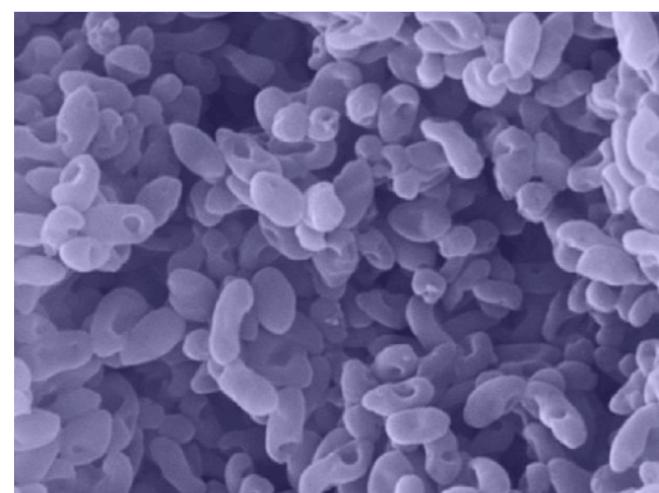
Photoreceptor domain BLUF



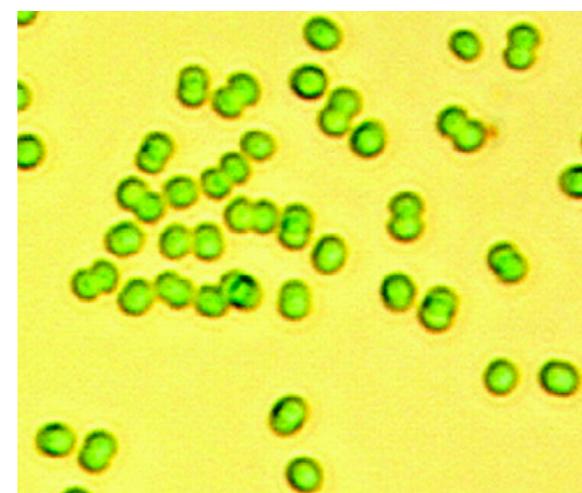
BLUF uses a flavin to convert a light signal into a biological response.

Found in photosynthetic bacteria, but also some single-celled eukaryotes.

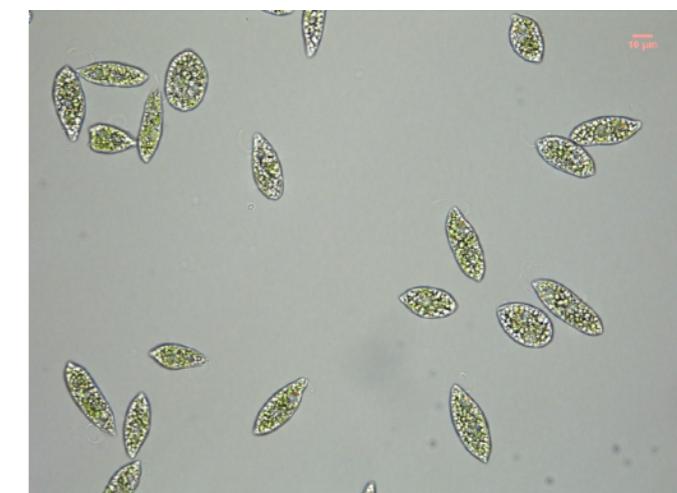
BLUF
Blue light using flavin



Purple Bacteria
R. sphaeroides



Cyanobacteria
S. sp. PCC6803



Algae
E. gracilis

AppA

Photosynthetic gene expression

Slr1694

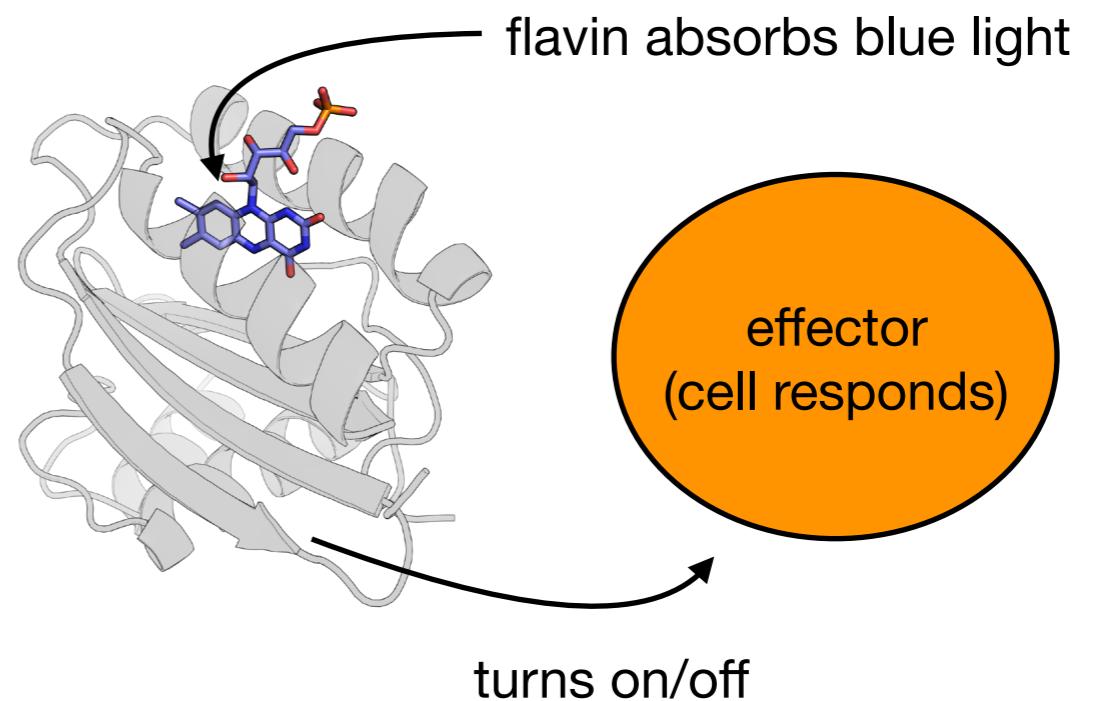
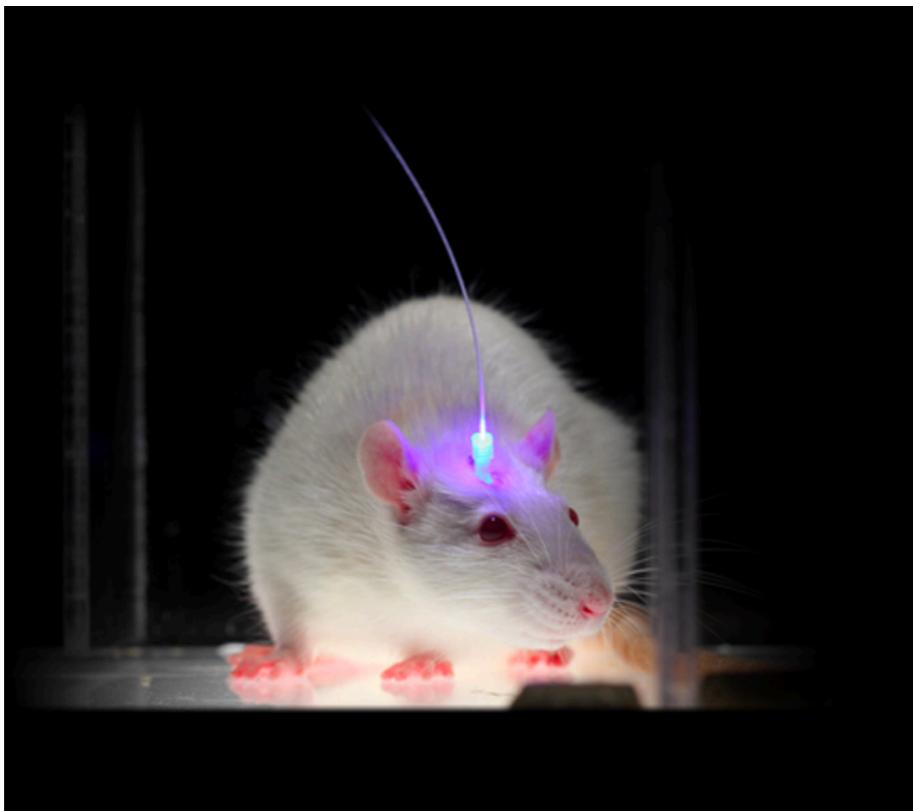
Positive phototaxis

PAC

Synthesize cAMP

Using BLUF for optogenetics

optogenetics: use light to control cells in living tissue

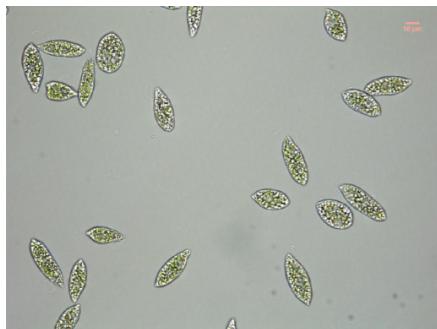


BLUF, coupled to an effector, causes changes under **blue light** (neuron fires, catalyze a reaction, etc.)

It's a **blue light** sensitive switch.

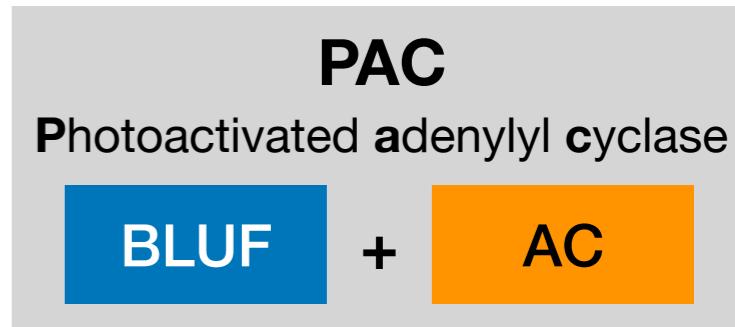
Source: Getty's Open Content Program (John B. Carnett / Getty Images)

Using BLUF for optogenetics

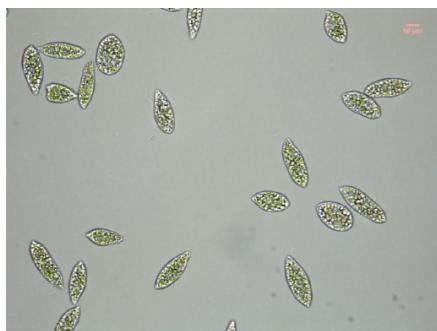


Algae
E. gracilis

Insert PAC into neurons of target organism
Blue-light illumination causes increase in cAMP



Using BLUF for optogenetics



Insert PAC into neurons of target organism
Blue-light illumination causes increase in cAMP

PAC

Photoactivated adenylyl cyclase

BLUF

+

AC

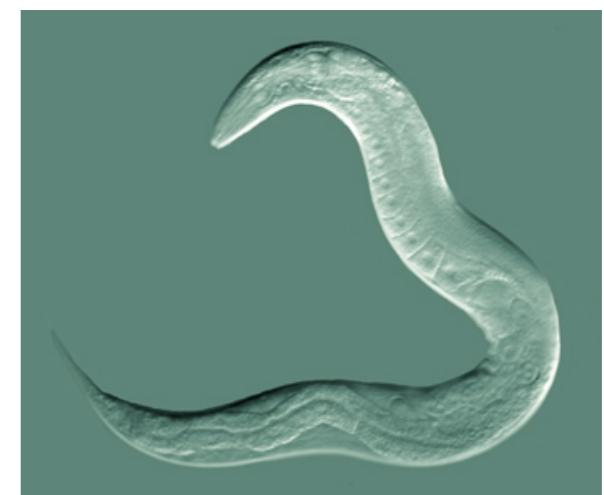
Algae
E. gracilis



Fruit fly
D. melanogaster
(Schroder-Lang, et al. 2007)

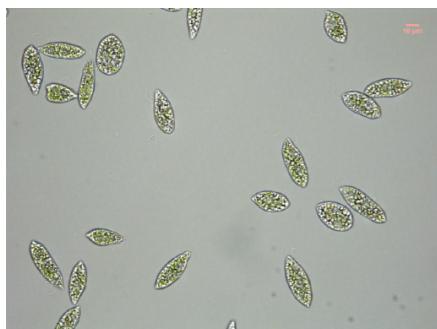


Sea slug
A. kurodai
(Nagahama, et al. 2007)



Roundworm
C. elegans
(Weissenberger, et al. 2011)

Using BLUF for optogenetics



Insert PAC into neurons of target organism
Blue-light illumination causes increase in cAMP

PAC

Photoactivated adenylyl cyclase

BLUF

+

AC

Algae
E. gracilis



Fruit fly
D. melanogaster
(Schroder-Lang, et al. 2007)



Sea slug
A. kurodai
(Nagahama, et al. 2007)



Roundworm
C. elegans
(Weissenberger, et al. 2011)

- induce hyperactivity
- decrease grooming behavior

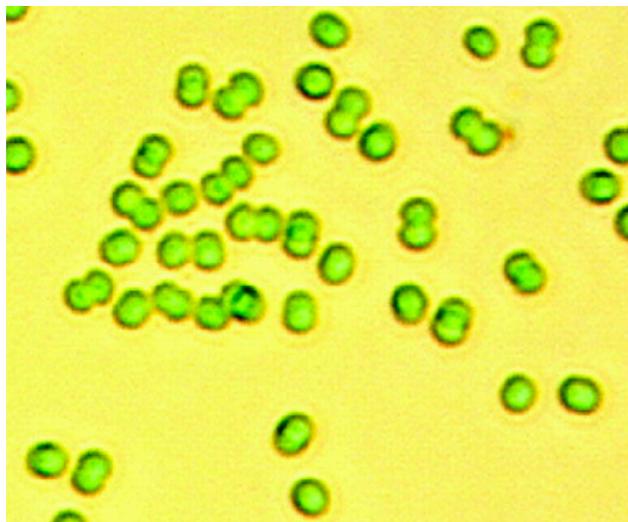
- changes neuron spike width and amplitude
- implies changes in synaptic strength (learning)

- increased swimming frequency
- increased rate of locomotion

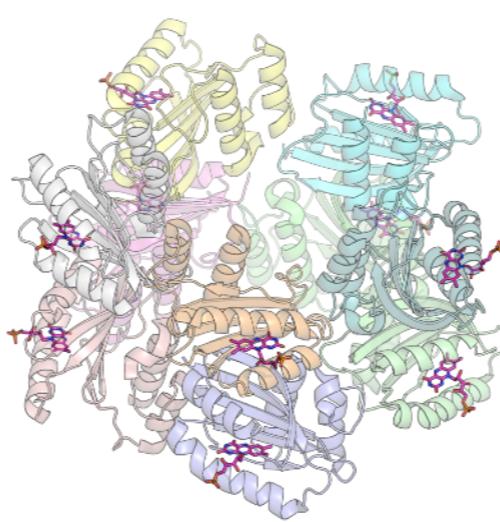
**If we want to engineer novel
optogenetic function...**

**...how does the BLUF
signaling mechanism work?**

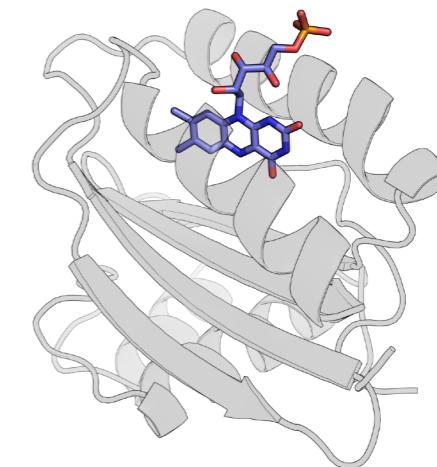
BLUF domain Slr1694 (PDB 2HFN)



Cyanobacteria
S. sp. PCC6803



2HFN decamer
1.8 Å resolution



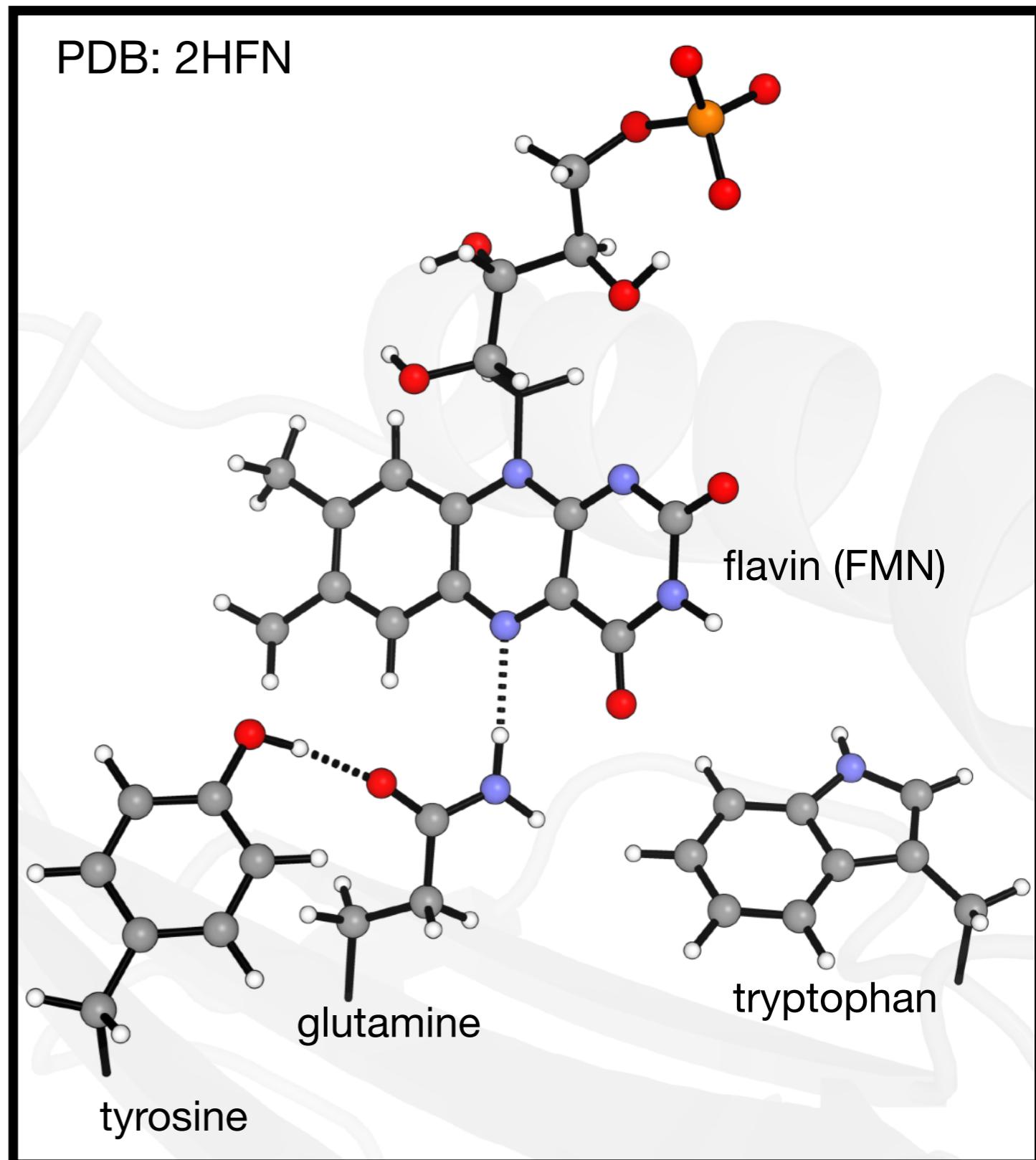
2HFN monomer
chain A

Interesting choice to study because:

1. *We have a crystal structure, and know the conserved residues*
2. *Spectroscopically well-characterized*
Photocycle progresses through metastable flavin radical intermediates.
Good for validation of any computational work (e.g. milestones)
3. *Fast formation of light-adapted (“on”) state (< 1 ns)*
Any computational dynamics will be much cheaper!

No consensus on mechanism

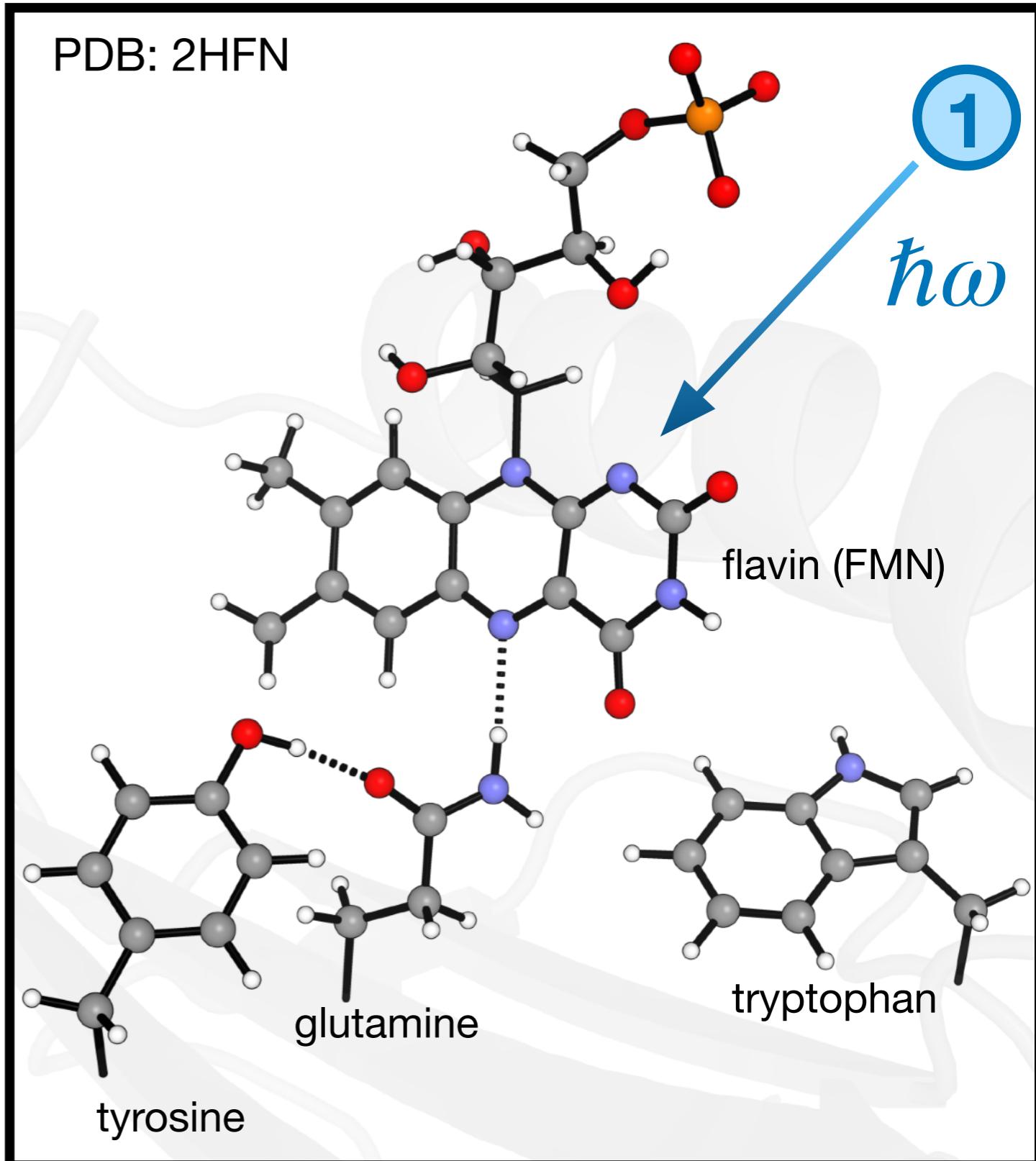
A rough outline of the Slr1694 photocycle



A rough outline of the Slr1694 photocycle

1

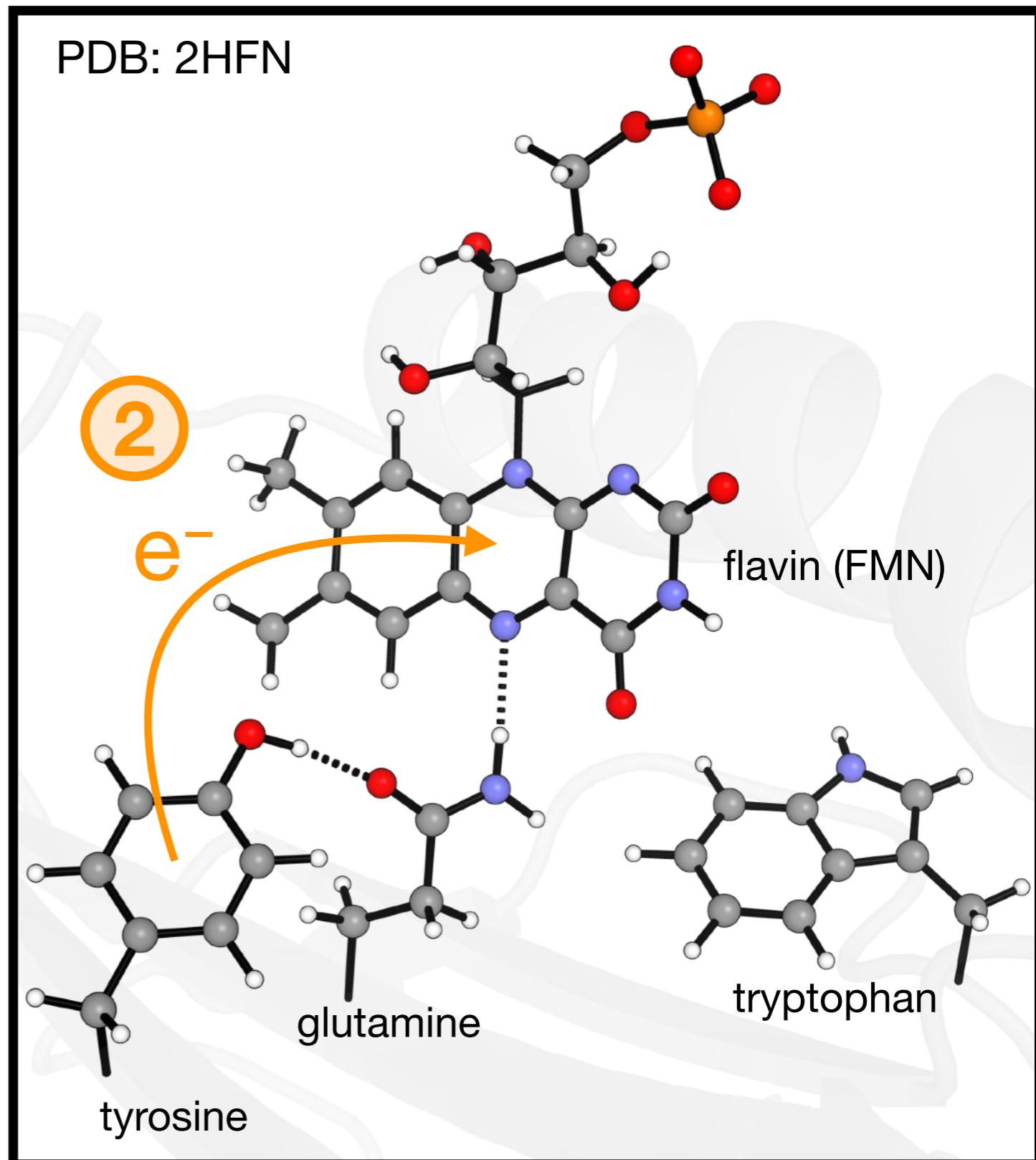
Flavin (FMN) absorbs
blue light, form FMN*



A rough outline of the Slr1694 photocycle

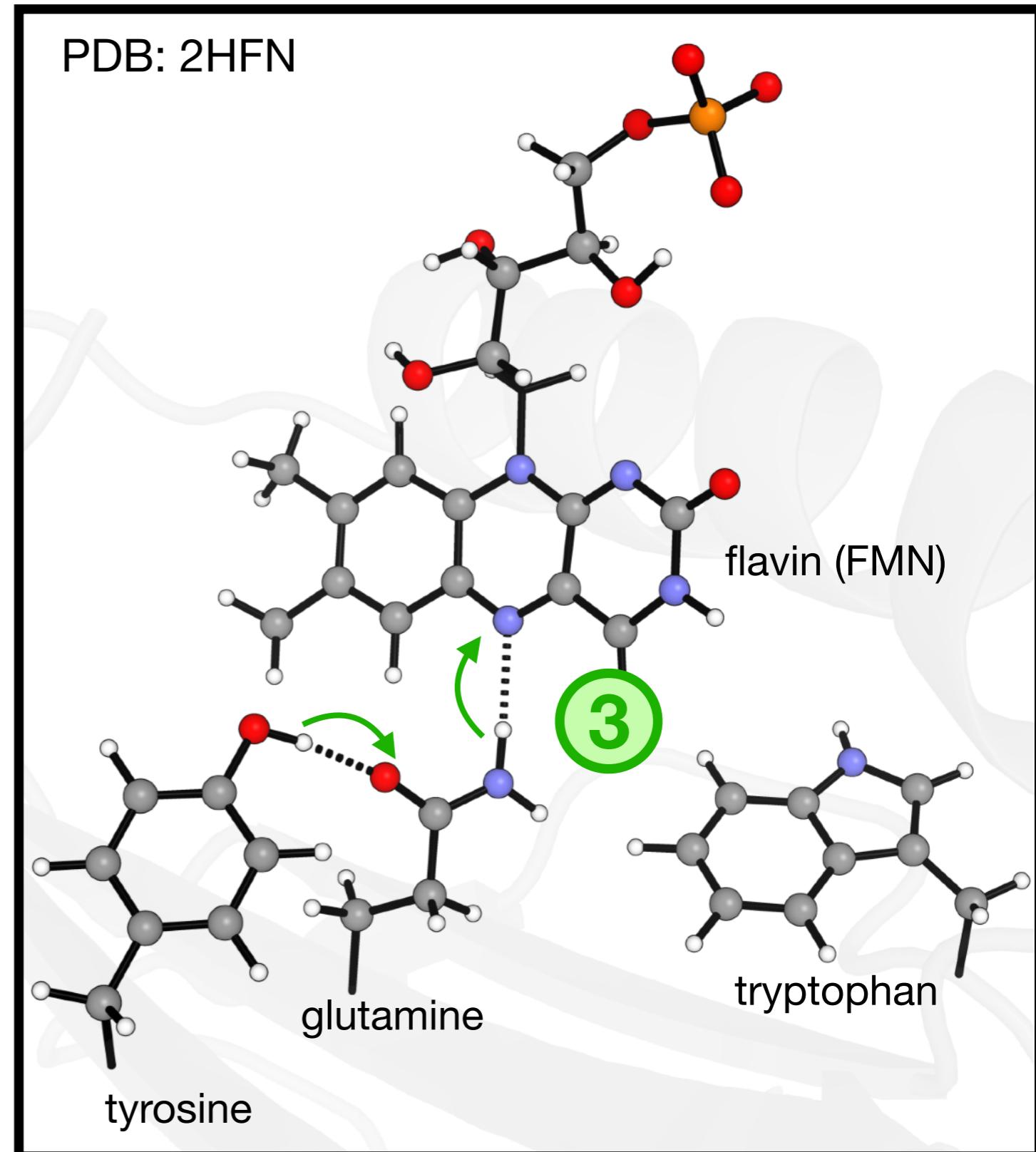
1 Flavin (FMN) absorbs blue light, form FMN^{*}

2 Electron transfer from Tyr to FMN^{*}, form FMN^{•-} (7 ps)



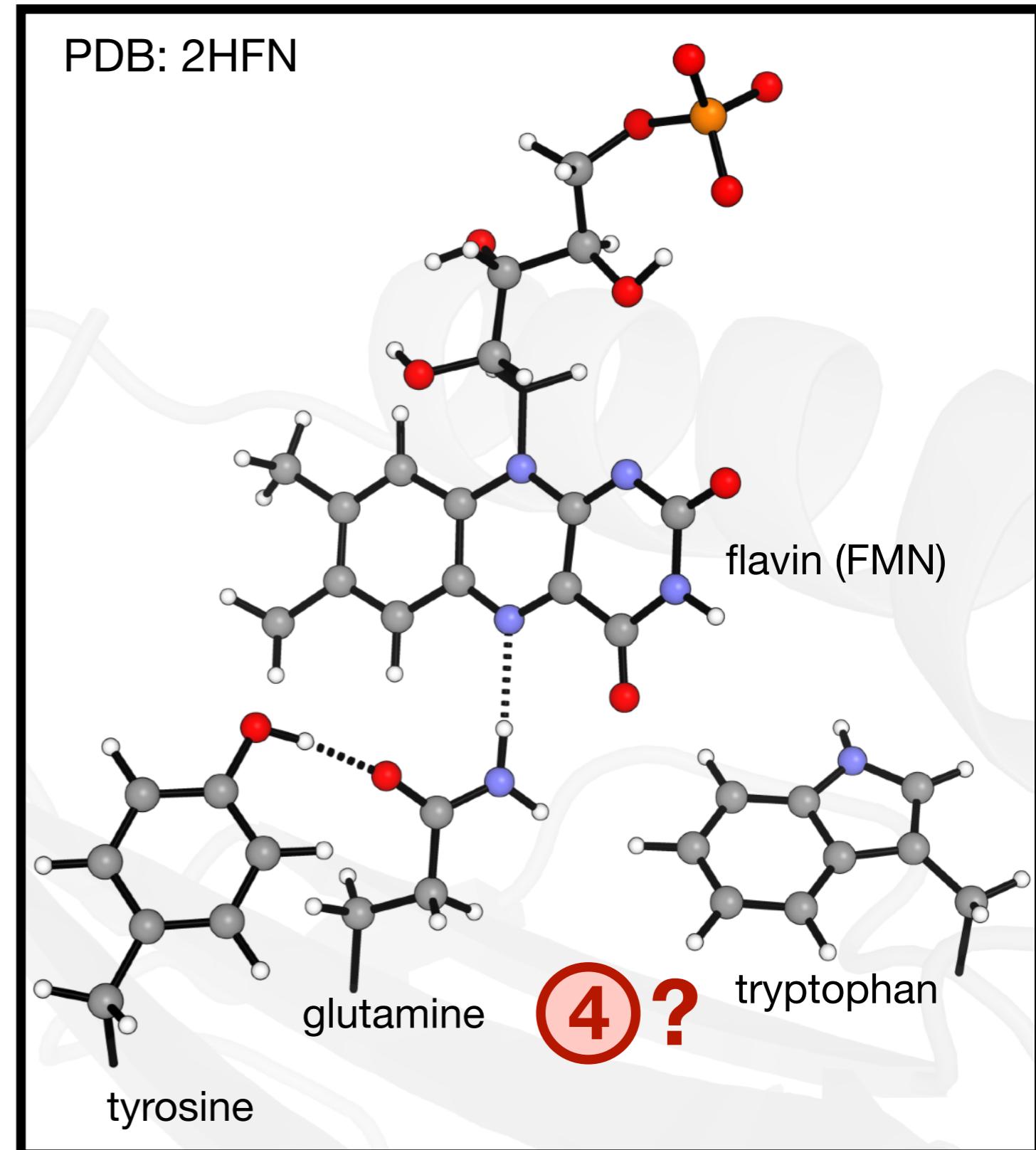
A rough outline of the Slr1694 photocycle

- 1 Flavin (FMN) absorbs blue light, form FMN^{*}
- 2 Electron transfer from Tyr to FMN^{*}, form FMN^{•-} (7 ps)
- 3 Proton transfer from Tyr to FMN^{•-}, form FMNH[•] (6 ps)



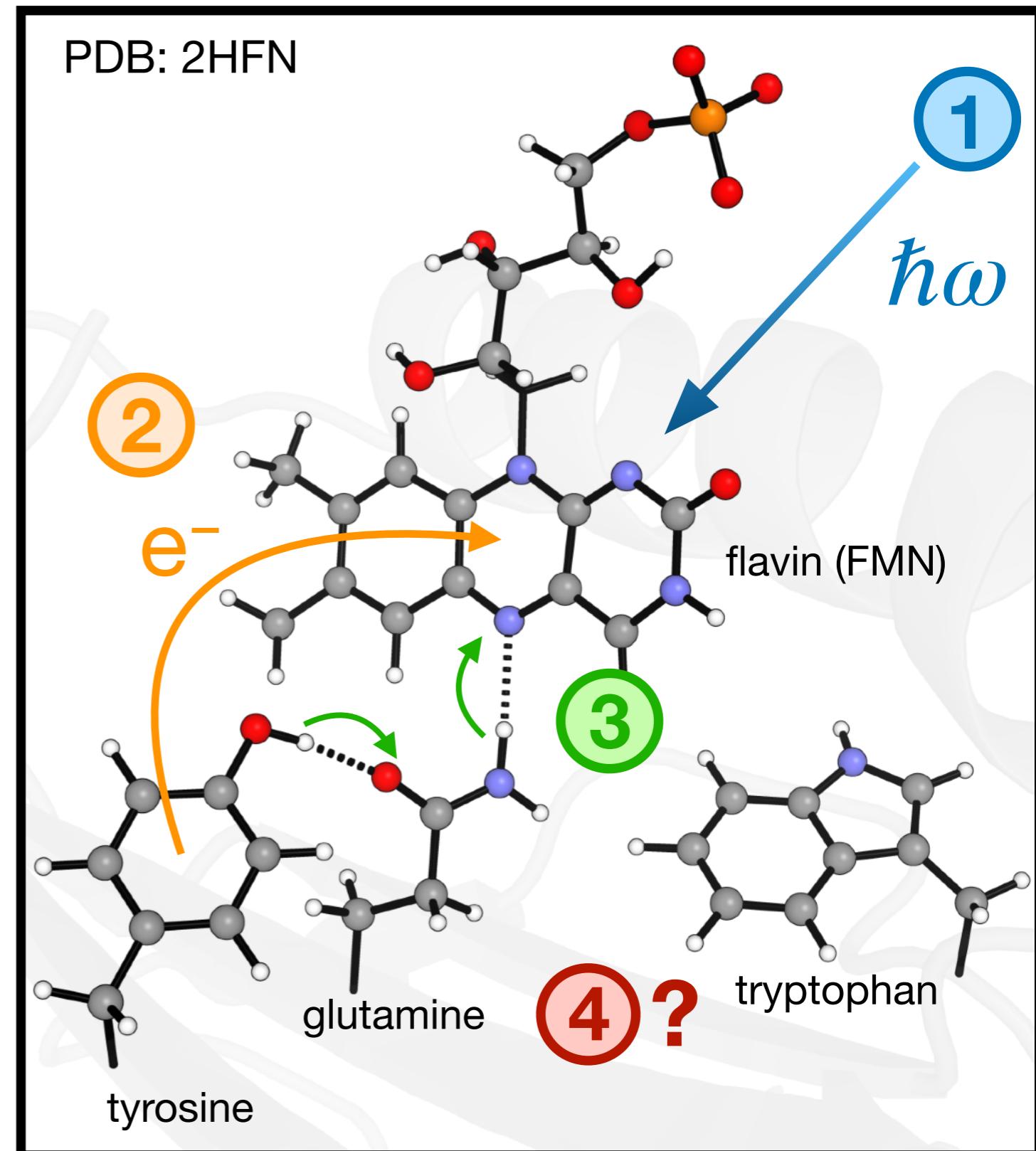
A rough outline of the Slr1694 photocycle

- 1 Flavin (FMN) absorbs blue light, form FMN^{*}
- 2 Electron transfer from Tyr to FMN^{*}, form FMN^{•-} (7 ps)
- 3 Proton transfer from Tyr to FMN^{•-}, form FMNH[•] (6 ps)
- 4 Relaxation to light-adapted state, back to FMN (65 ps)



A rough outline of the Slr1694 photocycle

- 1 Flavin (FMN) absorbs blue light, form FMN^{*}
- 2 Electron transfer from Tyr to FMN^{*}, form FMN^{•-} (7 ps)
- 3 Proton transfer from Tyr to FMN^{•-}, form FMNH[•] (6 ps)
- 4 Relaxation to light-adapted state, back to FMN (65 ps)

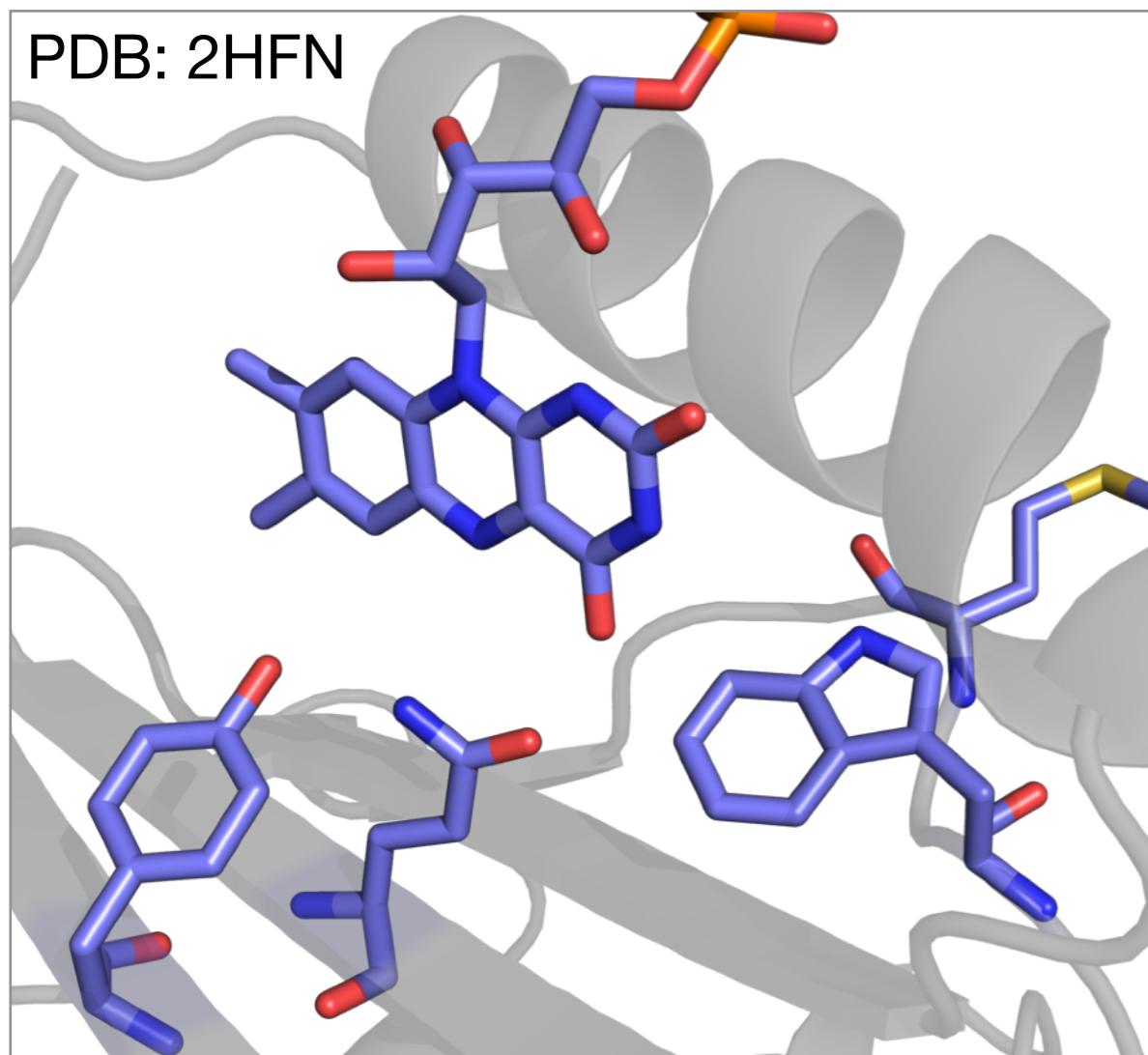


Molecular dynamics could help fill in the gaps in the photocycle...

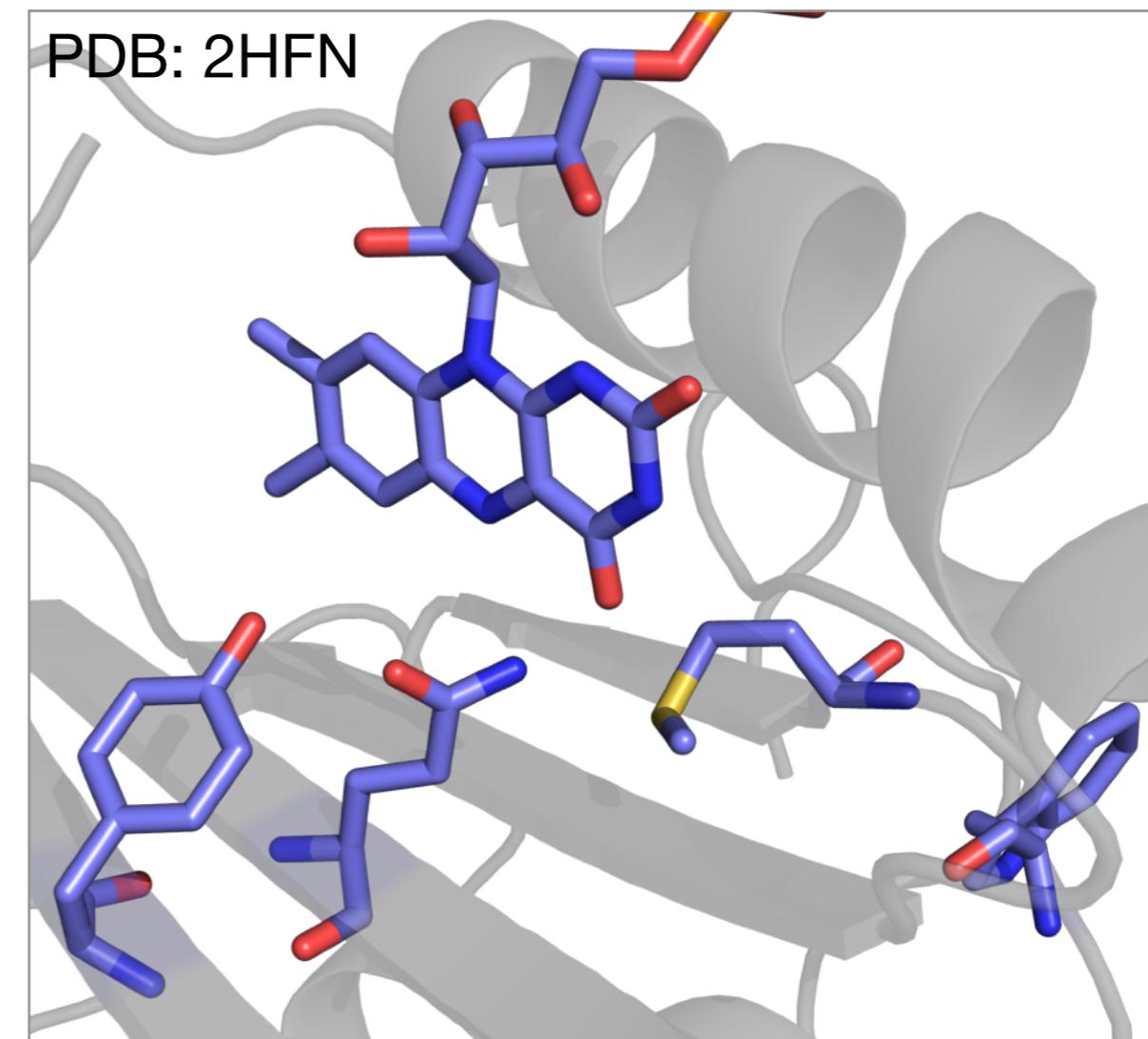
**...but the crystal structure doesn't give
unambiguous dark-adapted state.**

Crystal structure shows ambiguity

What is the biologically relevant configuration of the active site?



chain D

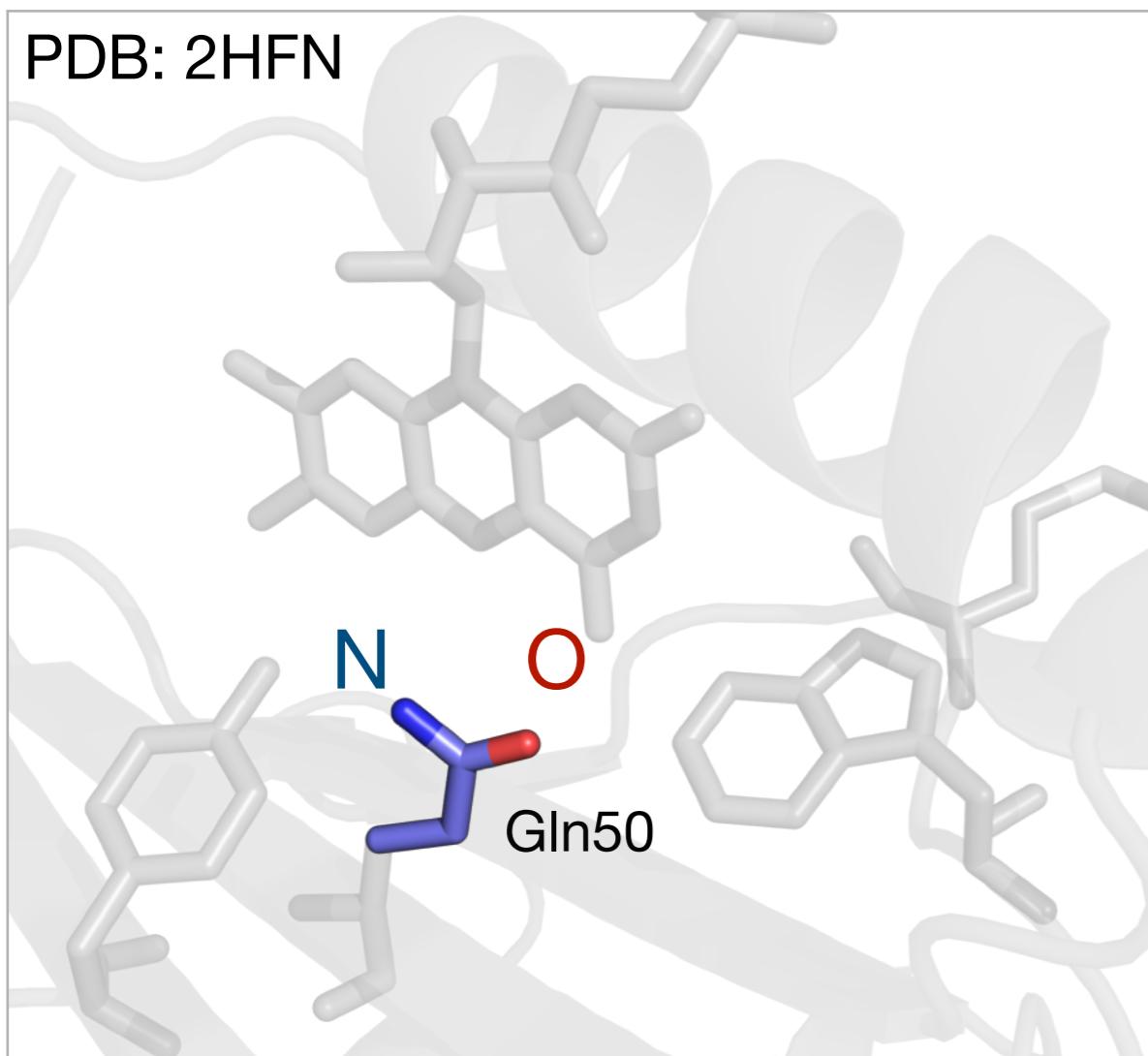


chain A

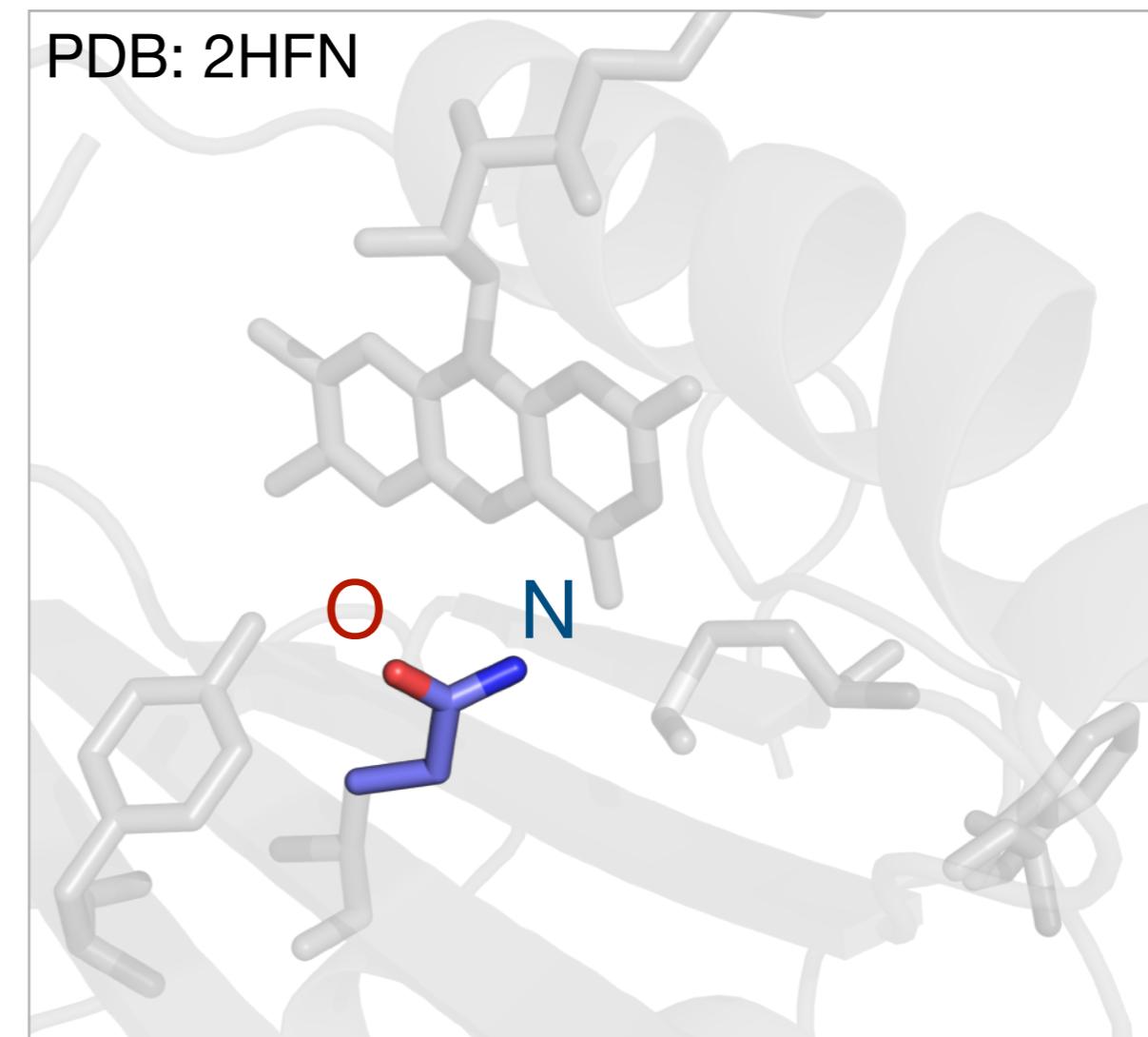
Same crystal structure, monomers differ on key residue location.

Crystal structure shows ambiguity

What is the preferred orientation of Gln50?



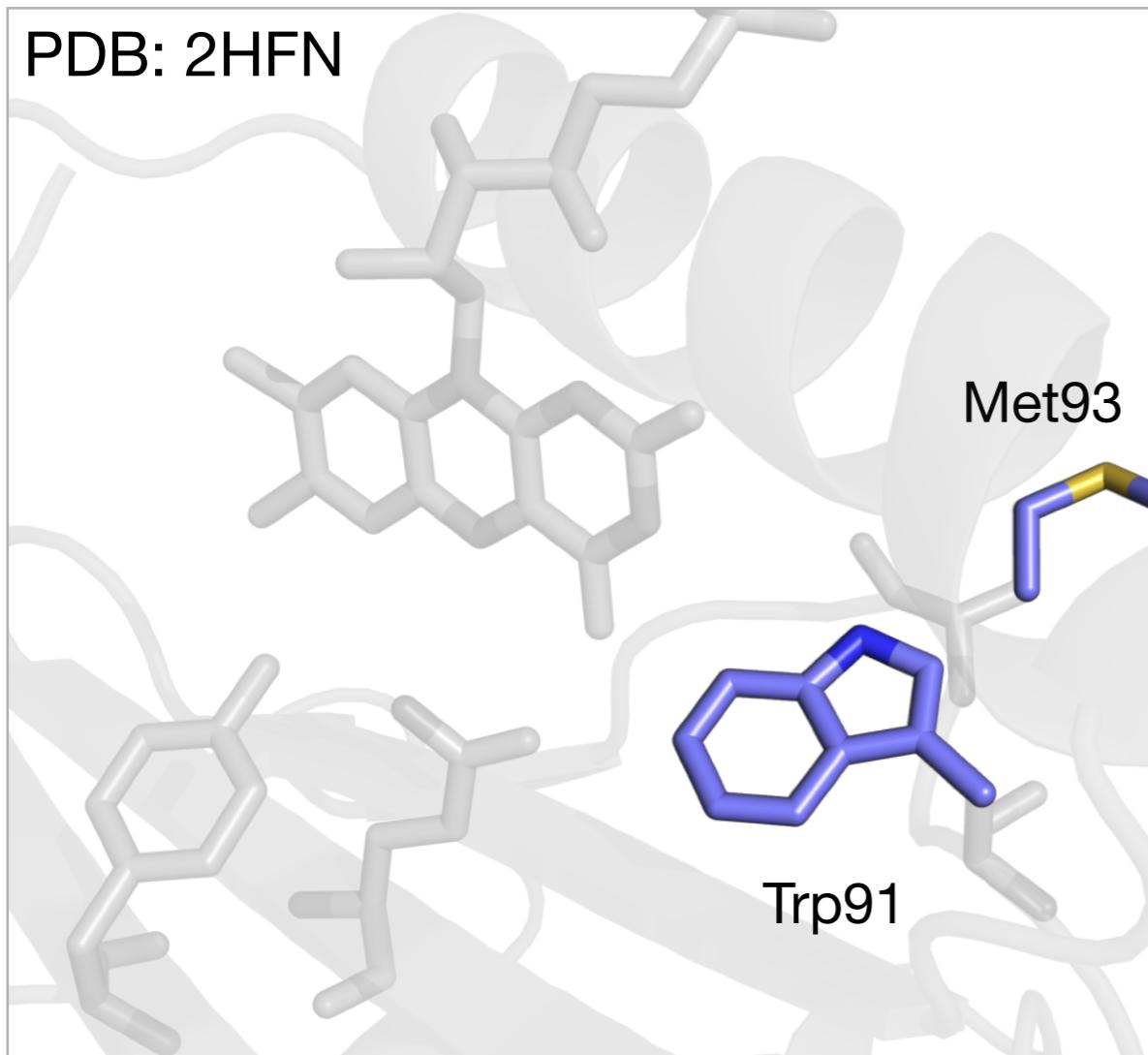
chain D



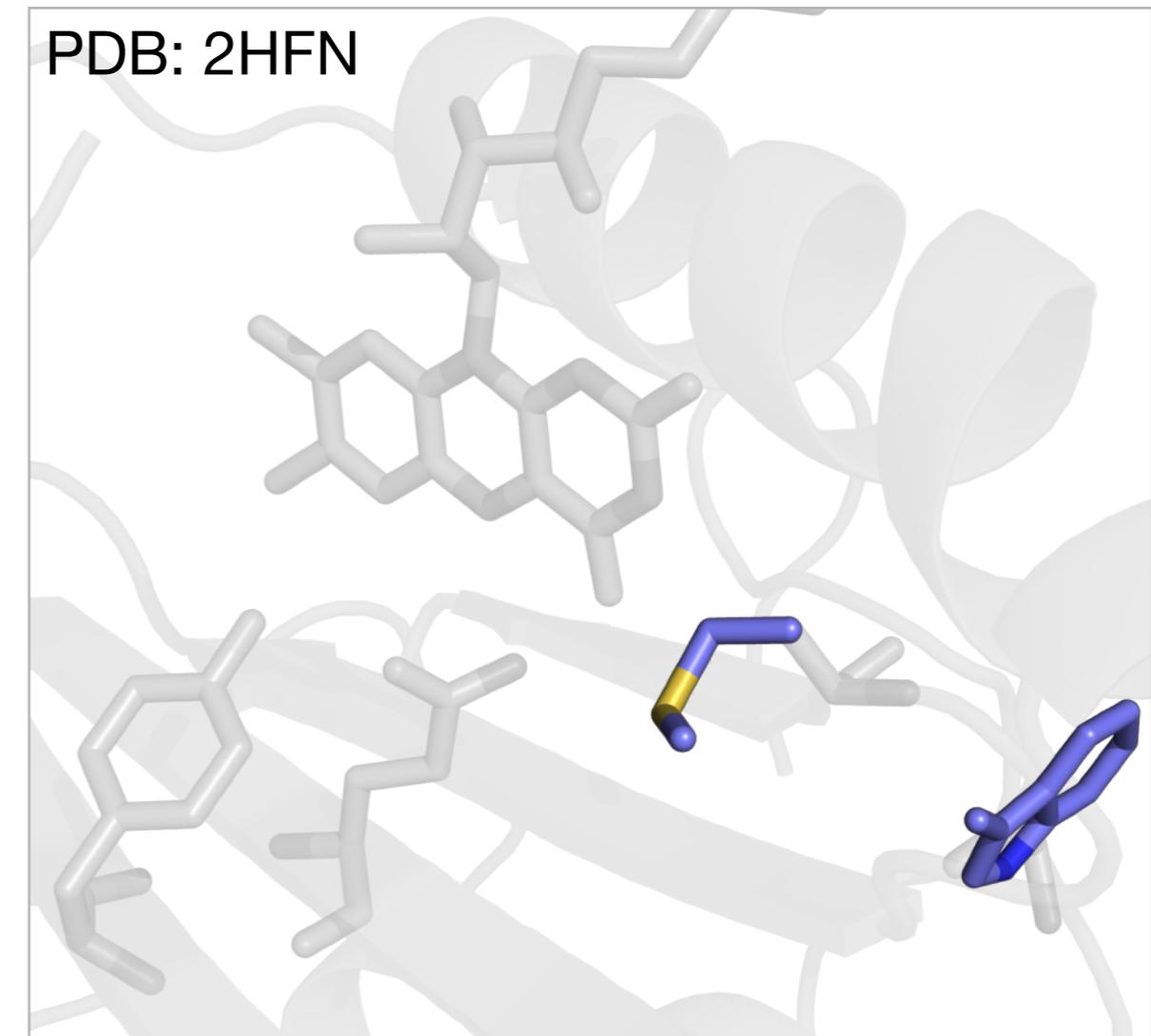
chain A

Crystal structure shows ambiguity

What is the preferred orientation of Trp91 (and Met93)?



chain D
“Trp_{in}”

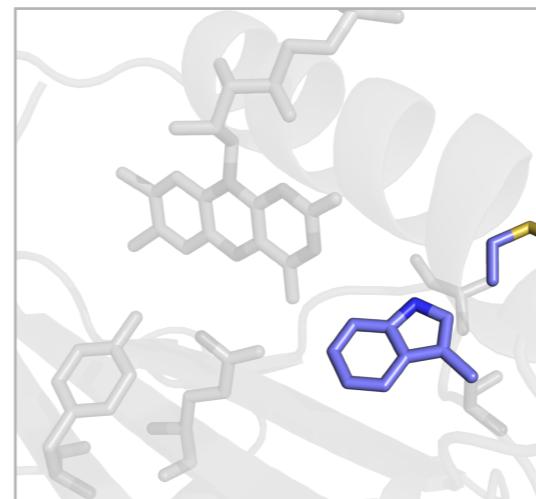
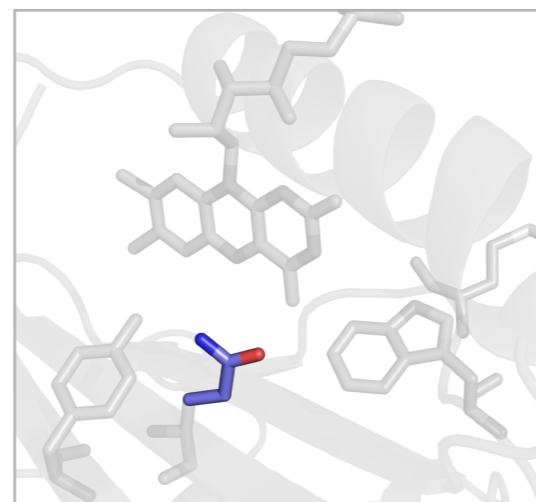


chain A
“Trp_{out}”

Crystal structure shows ambiguity

To understand the photocycle, we must understand the nature of the dark-adapted state.

We use **enhanced sampling molecular dynamics** methods to compute free energy profiles between conformations.



chain D



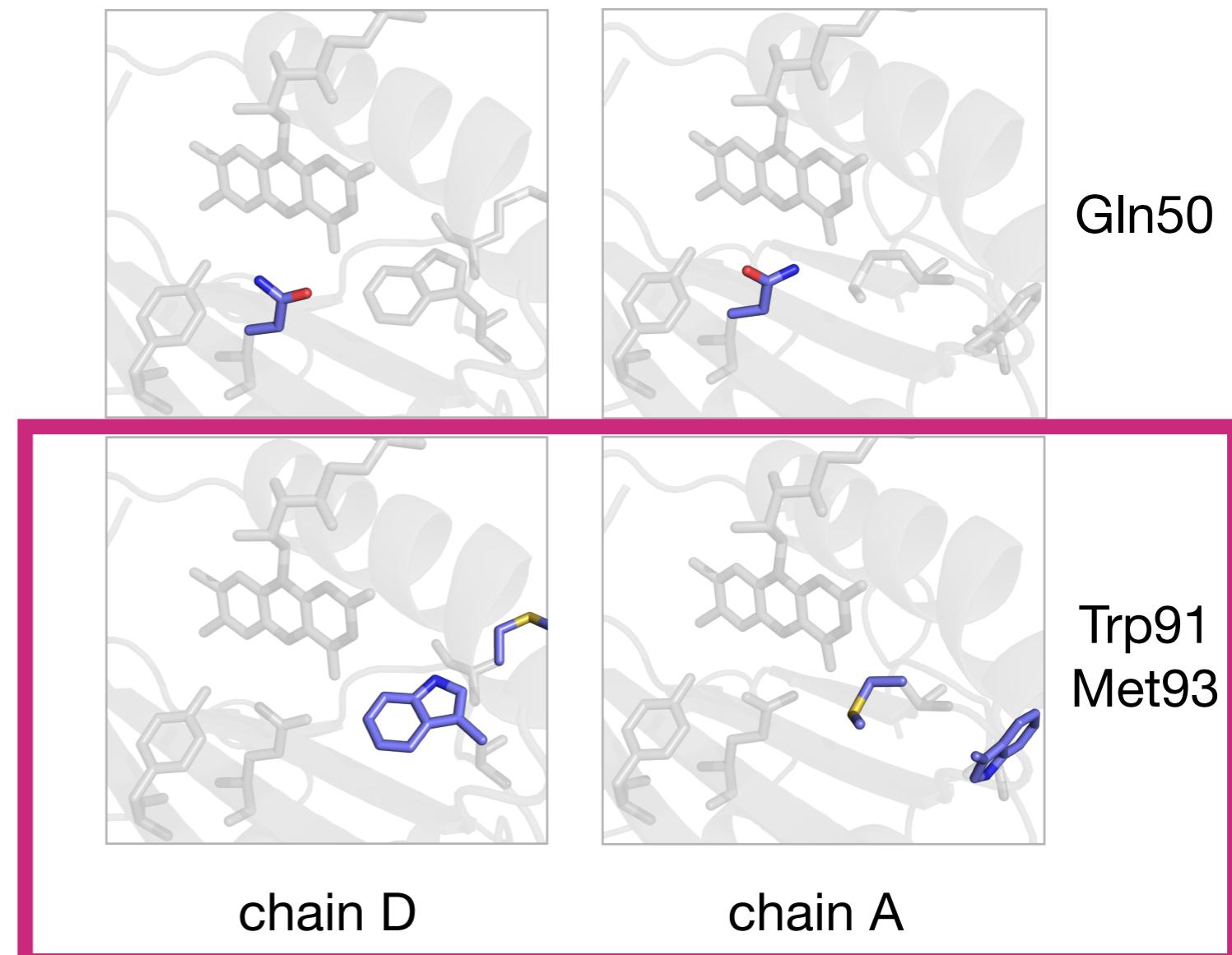
chain A

Gln50

Trp91
Met93

Crystal structure shows ambiguity

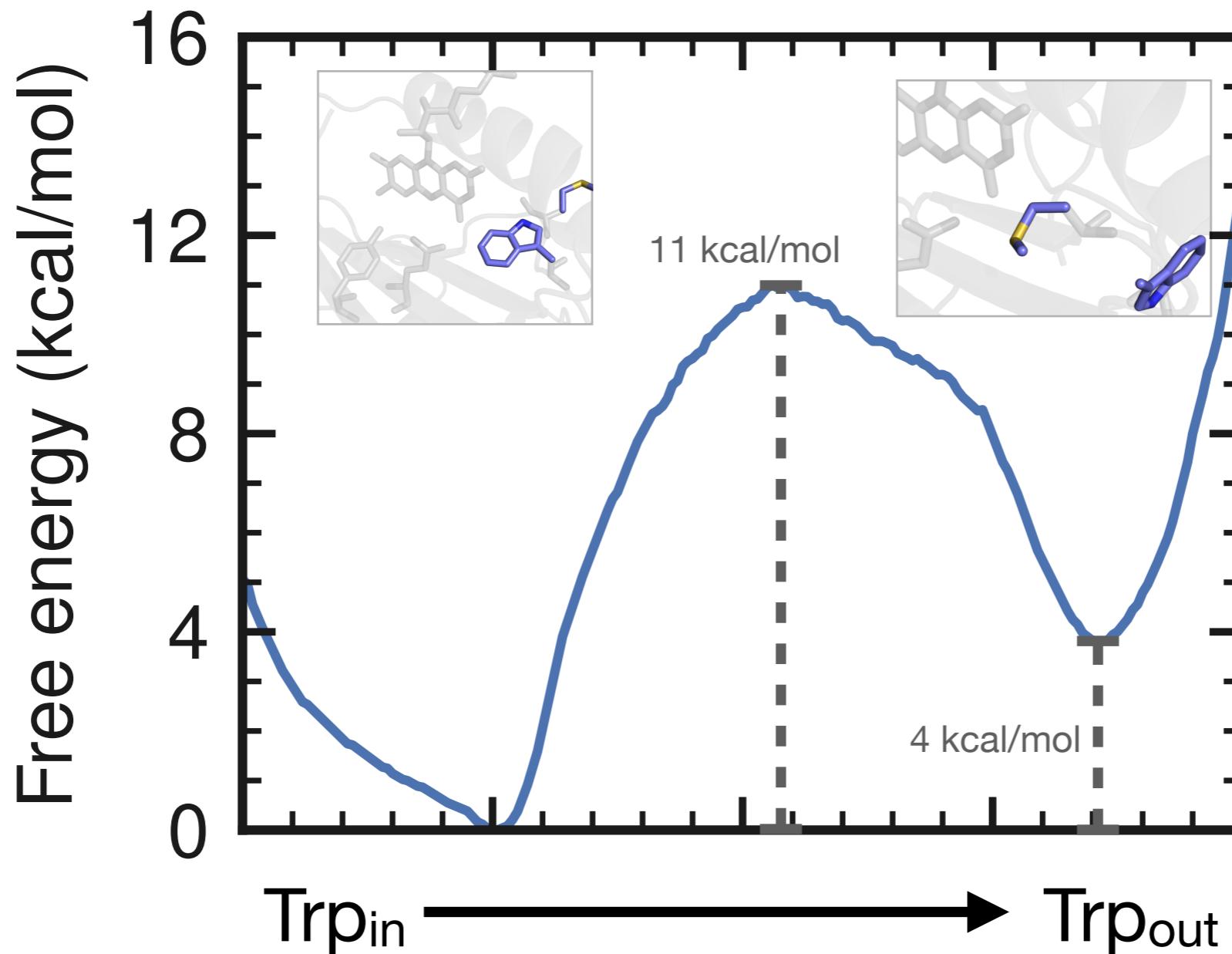
To understand the photocycle, we must understand the nature of the dark-adapted state.



We use **enhanced sampling molecular dynamics** methods to compute free energy profiles between conformations.

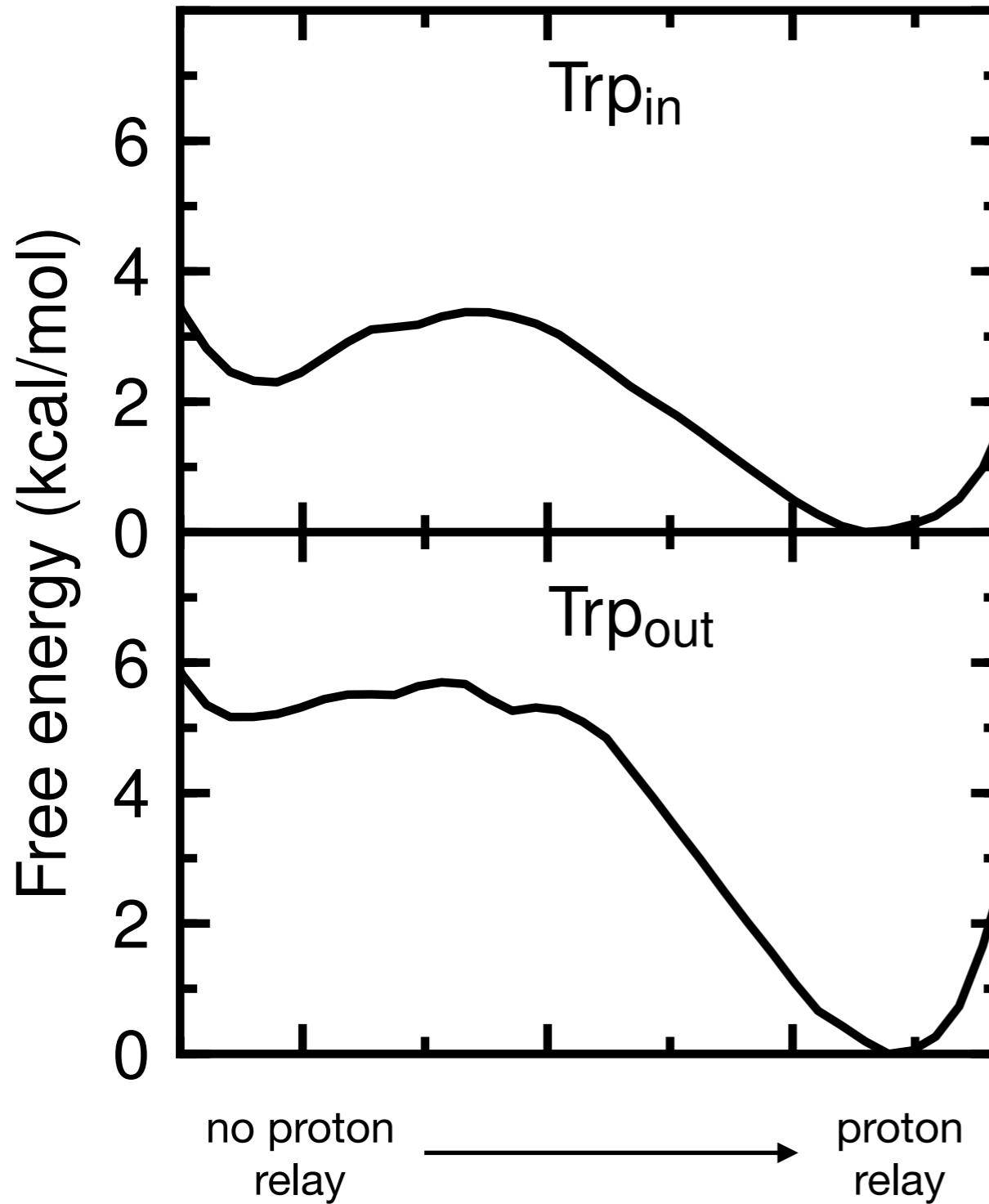
We will look at Trp/Met conformation first

Trp91 predominately in the active site

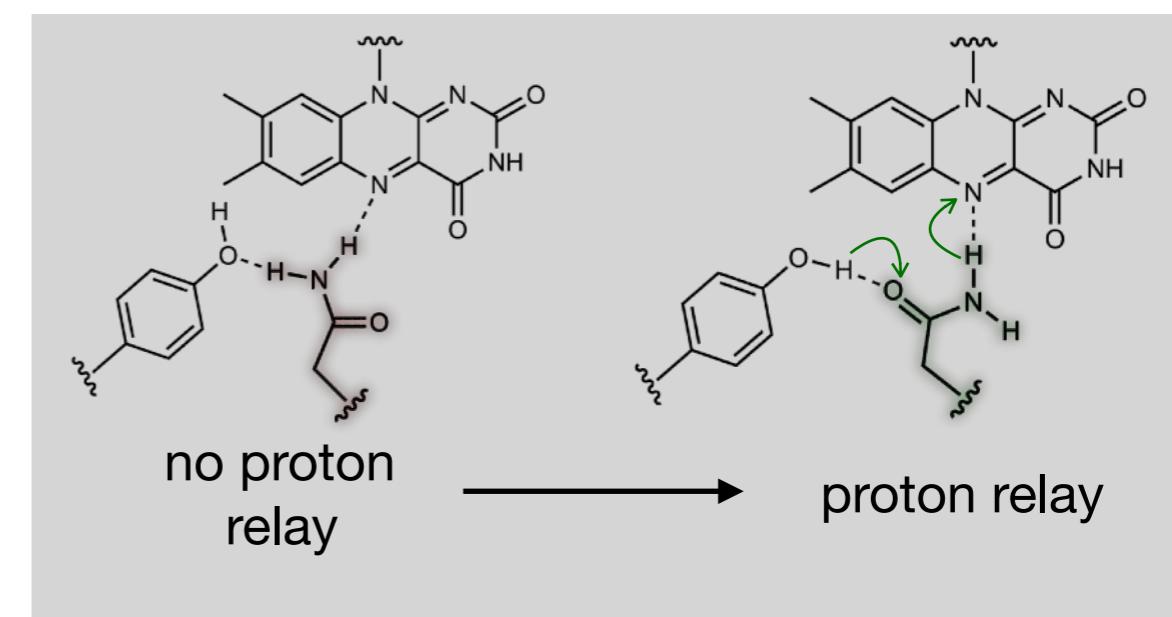


- Slr1694 can be in both Trp_{in} and Trp_{out} in crystal structures
- Adaptively-biased path optimization with CHARMM36 force field on S_0
- On S_0 , BLUF can interconvert Trp_{in} and Trp_{out} on $\sim 100 \mu\text{s}$ timescale
- Trp_{in} favored by 4 kcal/mol

Gln50: proton relay favored in Slr1694



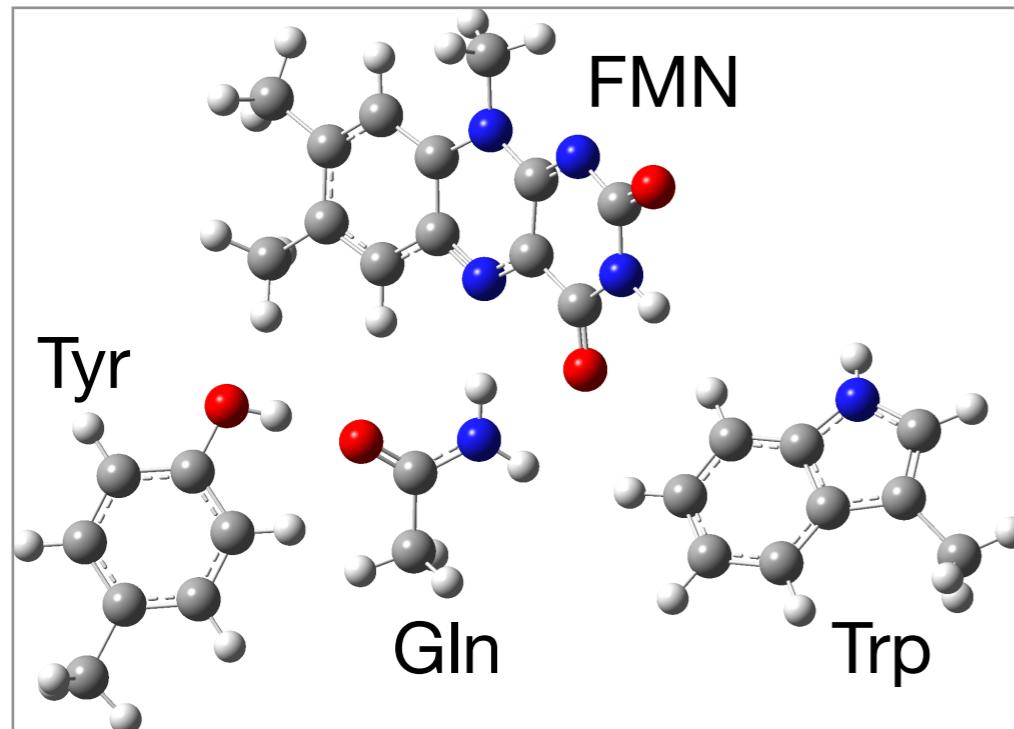
- Crystal structure does not provide unambiguous assignment of Gln orientation
- Umbrella sampling used to explore free energy changes along Gln rotation
- Proton relay always favored regardless of Trp_{in} or Trp_{out}



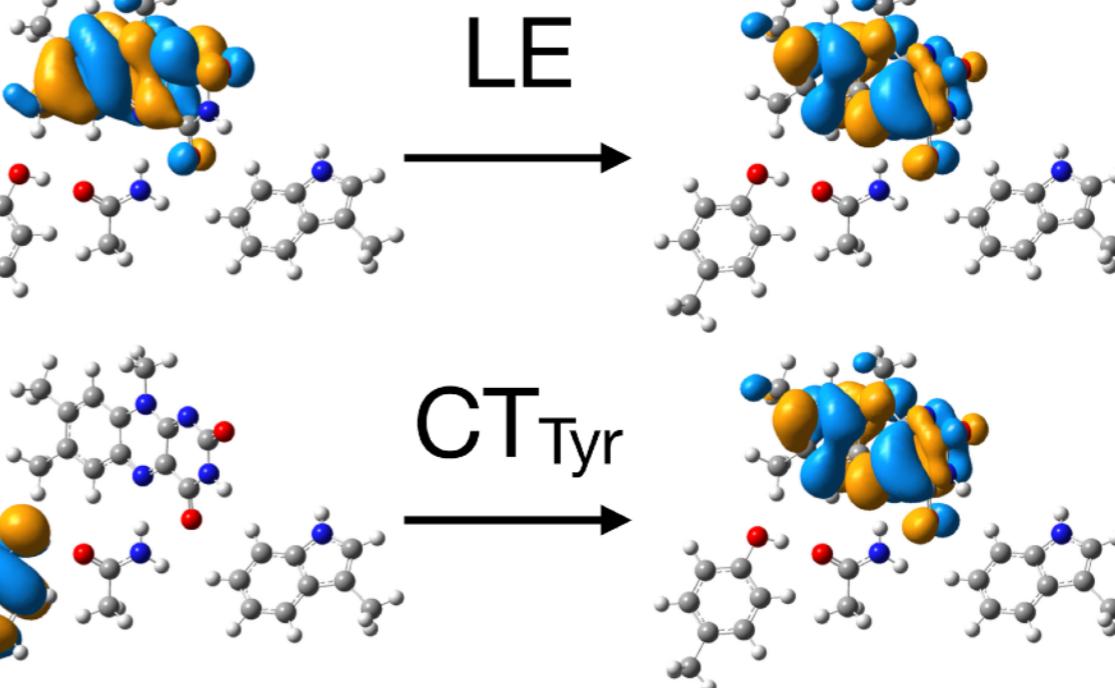
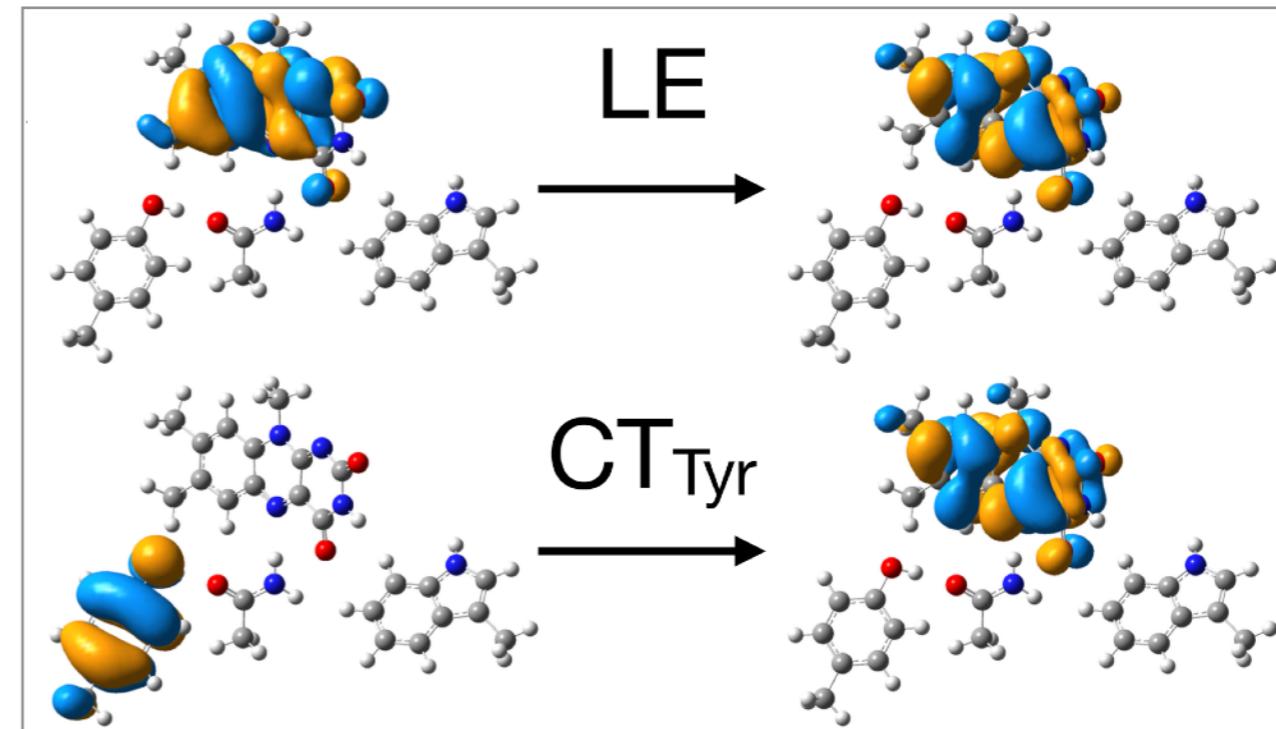
**Conformational free energies are useful,
but how do they impact excited states?**

TDDFT of active site in protein environment

In photocycle, LE state must cross with CT state to drive electron transfer.
How do conformational changes impact the electronic energy level ordering?



QM active site model

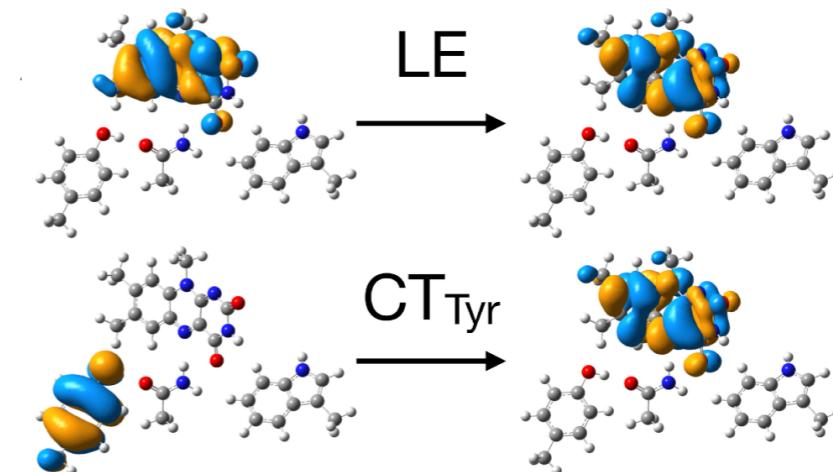


- Protein MM optimization after 20 ns NPT (300K, 1atm) molecular dynamics
- TDDFT/CAM-B3LYP/6-31+G** with electrostatic embedding of QM active site within protein and aqueous environment

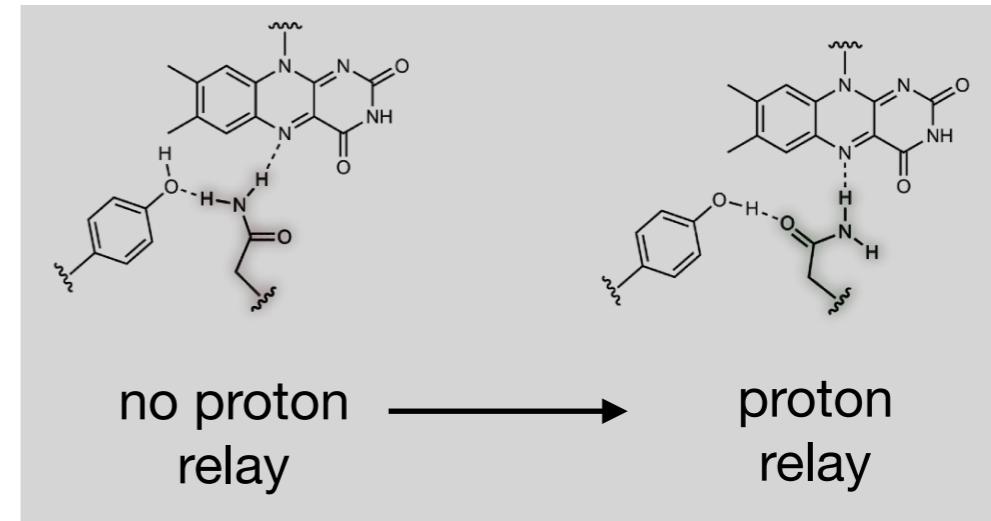
Proton relay facilitates charge transfer to flavin

Excitation energies (eV) of active site embedded in protein and aqueous environment obtained from ground state trajectories

| | no proton relay | proton relay |
|----------------|-----------------|--------------|
| LE | 3.00 | 2.99 |
| CT | 4.42 | 3.68 |
| CT–LE | | |
| gap ΔE | 1.42 | 0.69 |



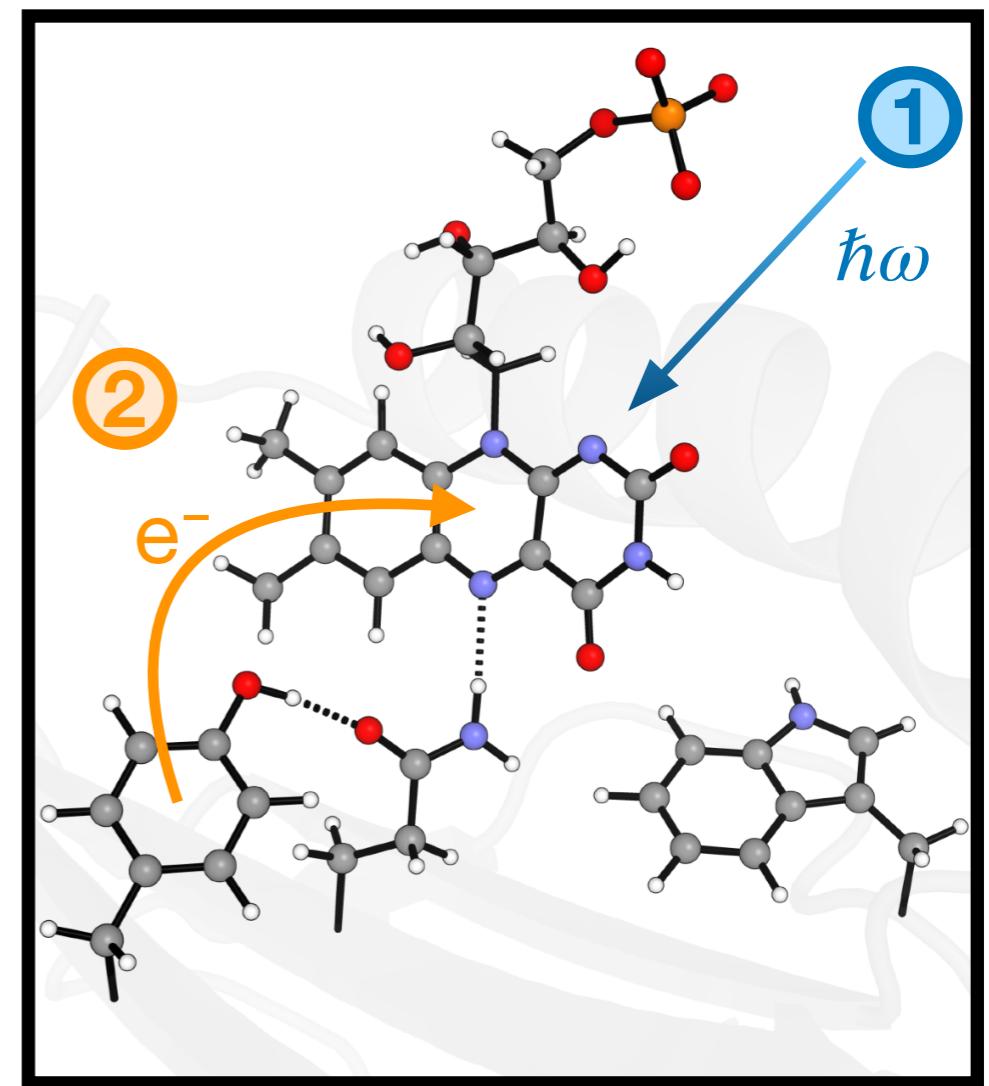
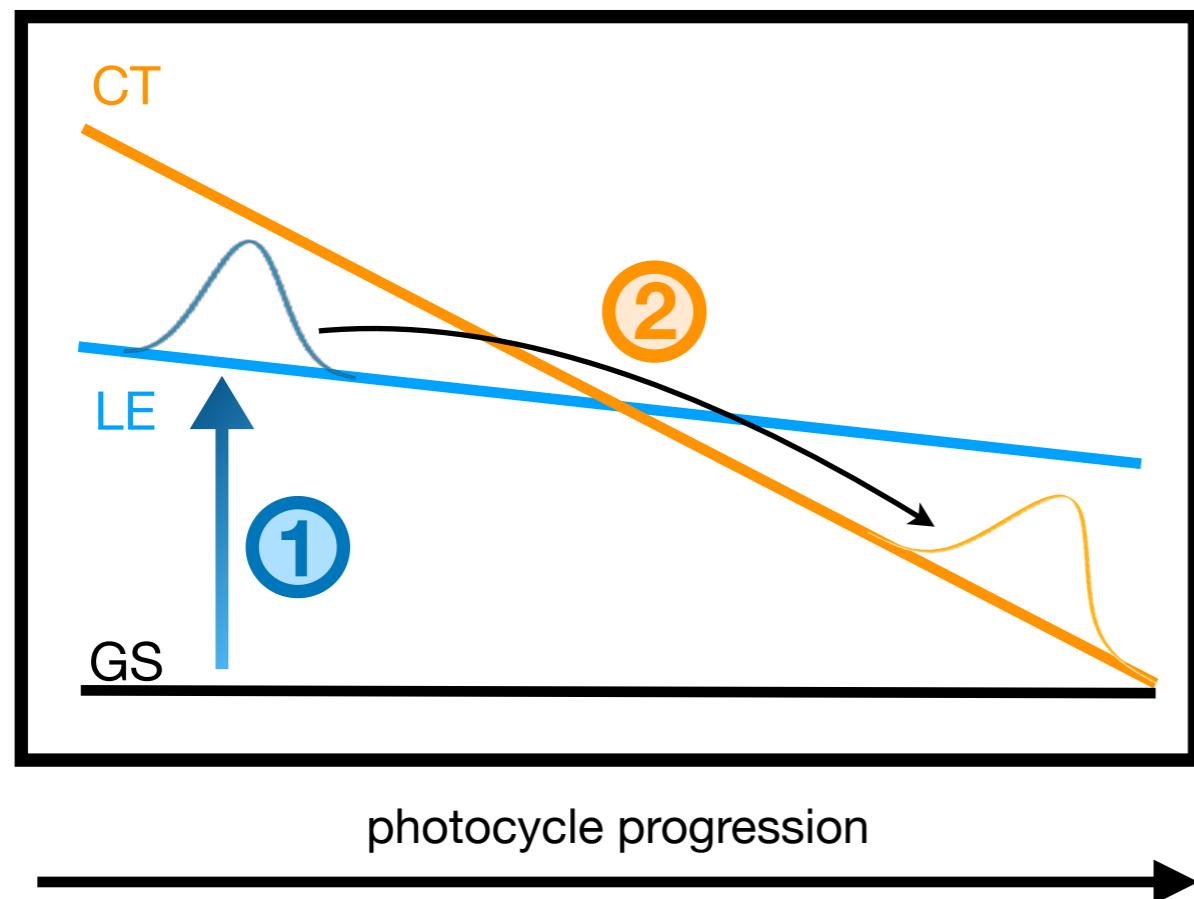
- Initial photoexcitation to LE state (only state with non-zero oscillator strength)
- Excitation energies for CT states depend strongly on active site Gln conformation
- LE states relatively insensitive to conformational changes



LE states still always lower energy than CT states – how do they cross?

Problem: CT states still above LE states

For the photocycle to progress, we must cross to the CT state – how?

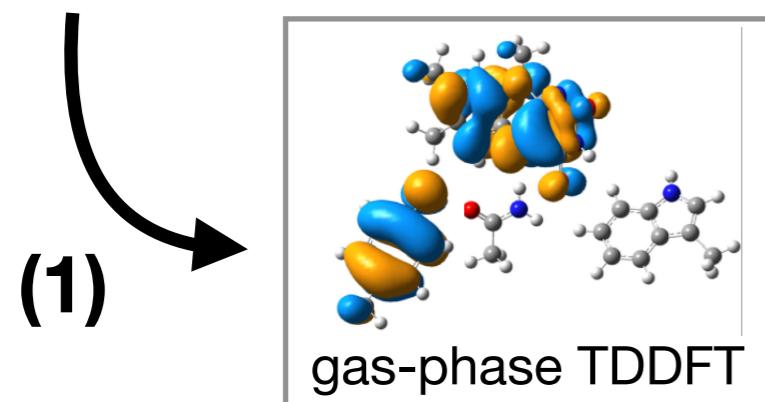
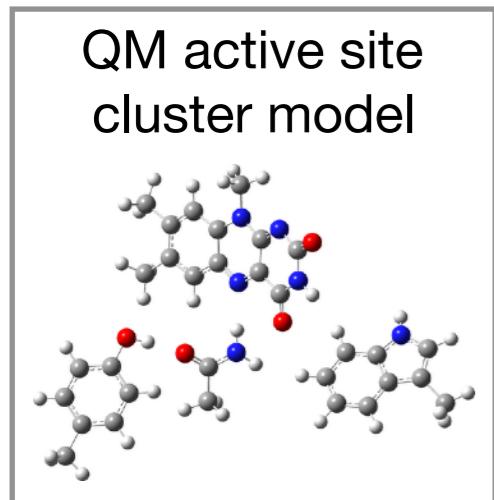


- Obviously a good question for nonadiabatic dynamics methods, but...
- Would be very interesting to find a BLUF conformation with the energy order swapped.
Hint at possible mechanism?

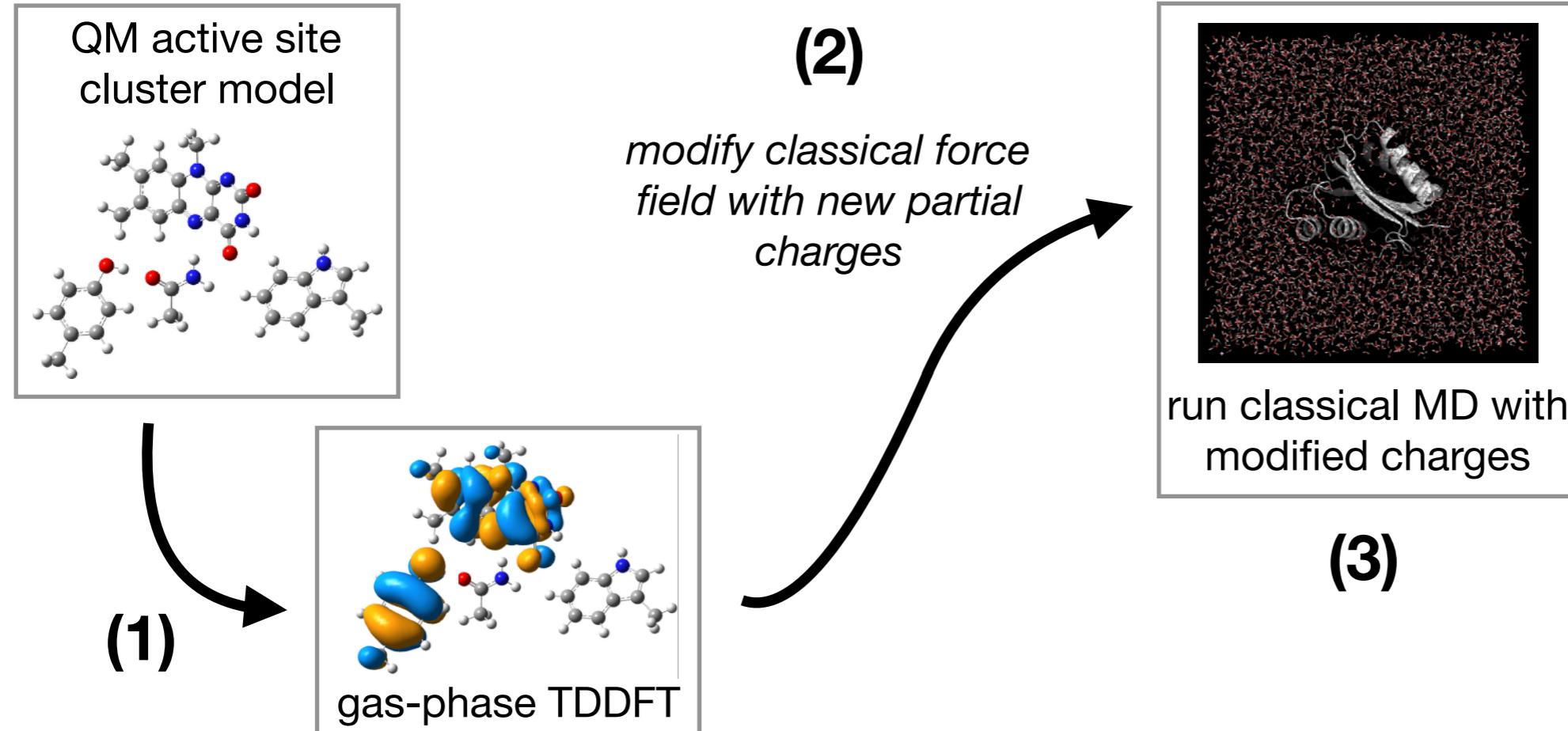
| | no proton relay | proton relay |
|-------------------------|-----------------|--------------|
| LE | 3.00 | 2.99 |
| CT | 4.42 | 3.68 |
| CT–LE gap ΔE | 1.42 | 0.69 |

(Excitation energies in eV)

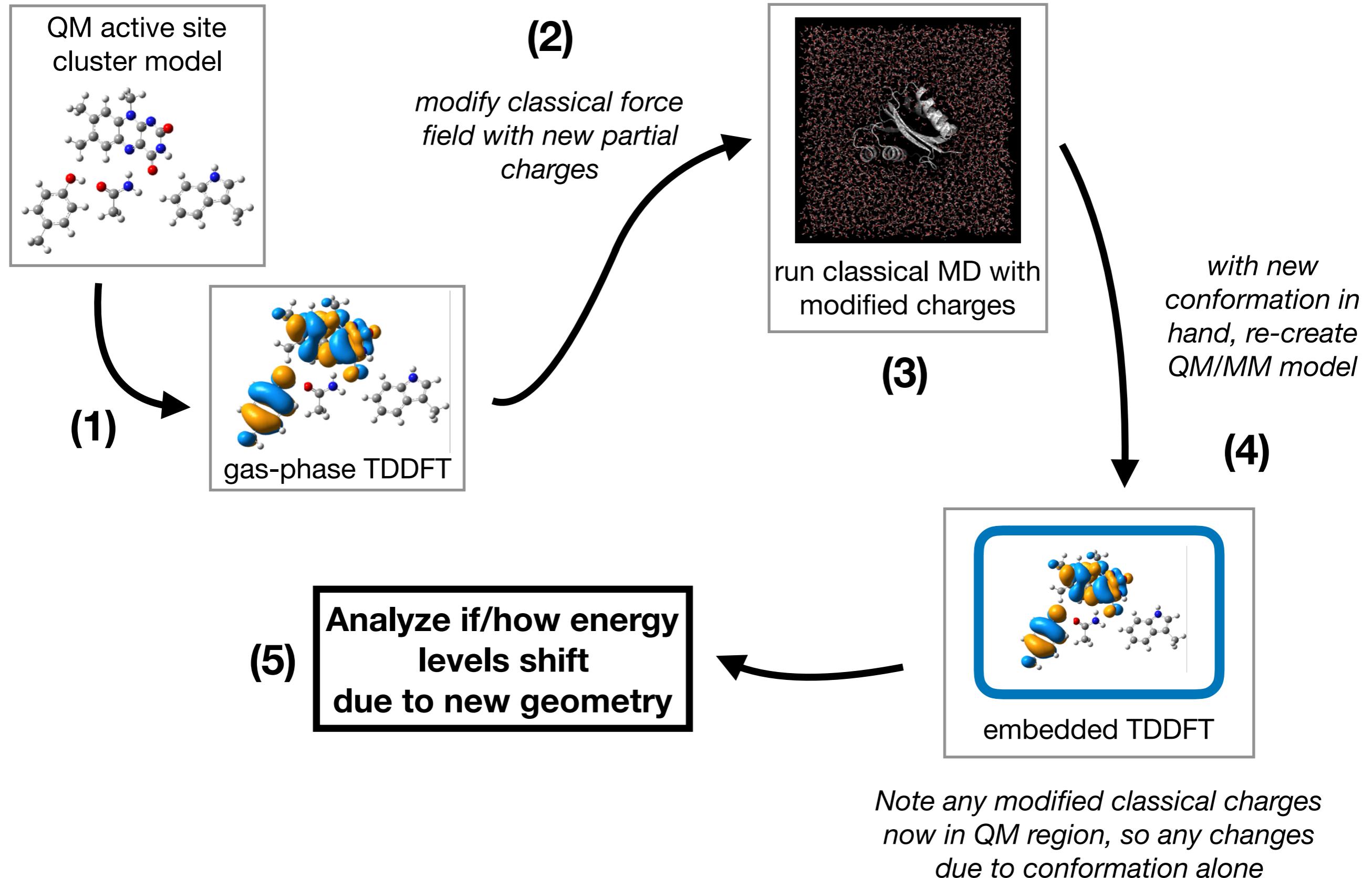
Finding conformations that stabilize CT



Finding conformations that stabilize CT



Finding conformations that stabilize CT



Protein reorganization can induce charge transfer

Use protein environments equilibrated on CT states to find conformations that stabilize the CT state.

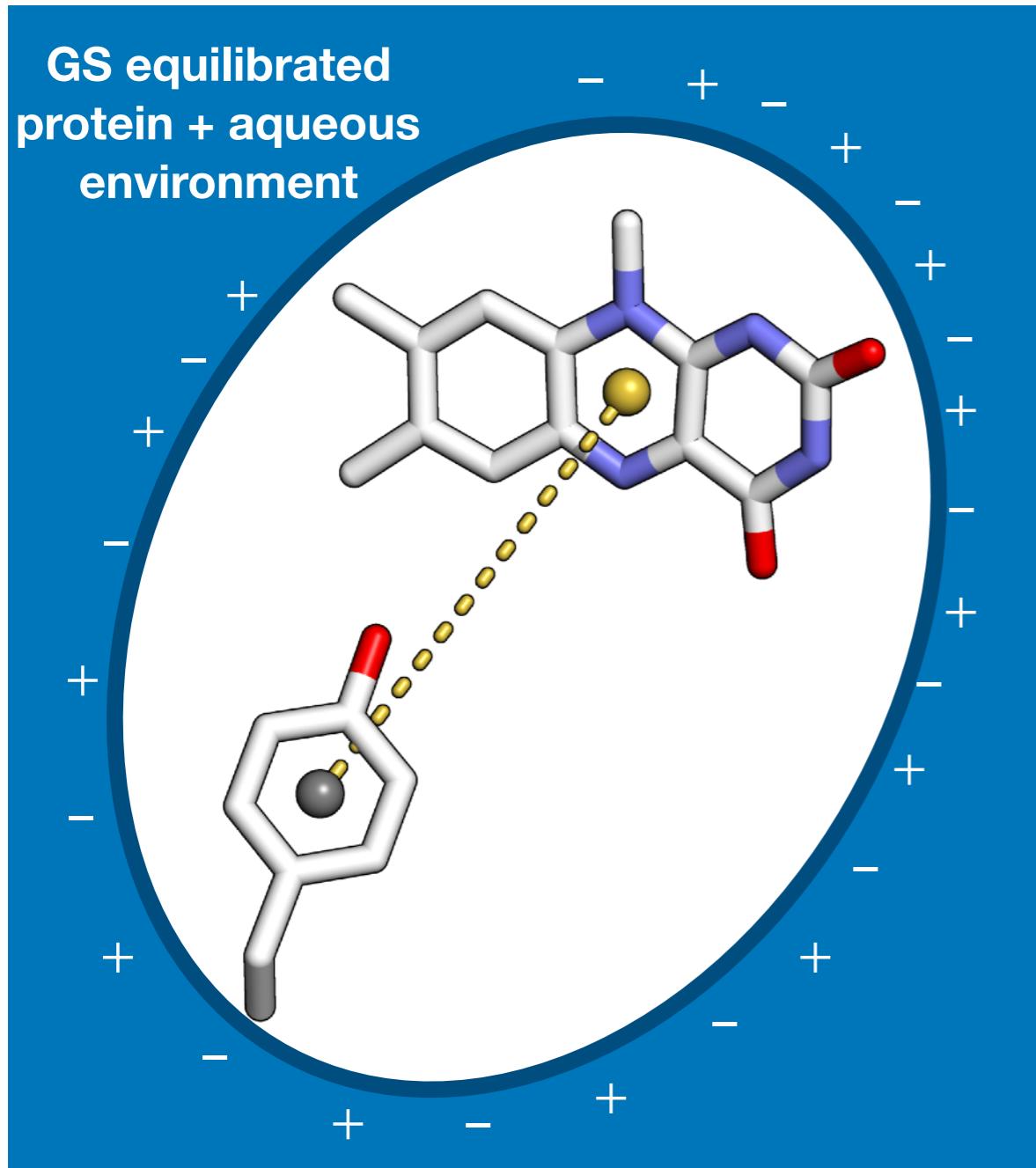
| <i>Excitation energies (eV) of active site embedded in protein and aqueous environment obtained from ground and excited state trajectories</i> | | | |
|------------------------------------------------------------------------------------------------------------------------------------------------|----------|----------|----------|
| | GS conf. | LE conf. | CT conf. |
| LE | 2.99 | 2.95 | 2.87 |
| CT | 3.68 | 3.77 | 1.71 |
| CT–LE gap ΔE | 0.69 | 0.82 | -1.16 |

- As expected, the conformation obtained from the CT state trajectory appears to markedly stabilize CT state. We can replicate for other independent conformations.
- CT stabilization goes away if we exclude electrostatic environment (e.g. gas phase)

What conformational change is leading to the stabilized CT state?

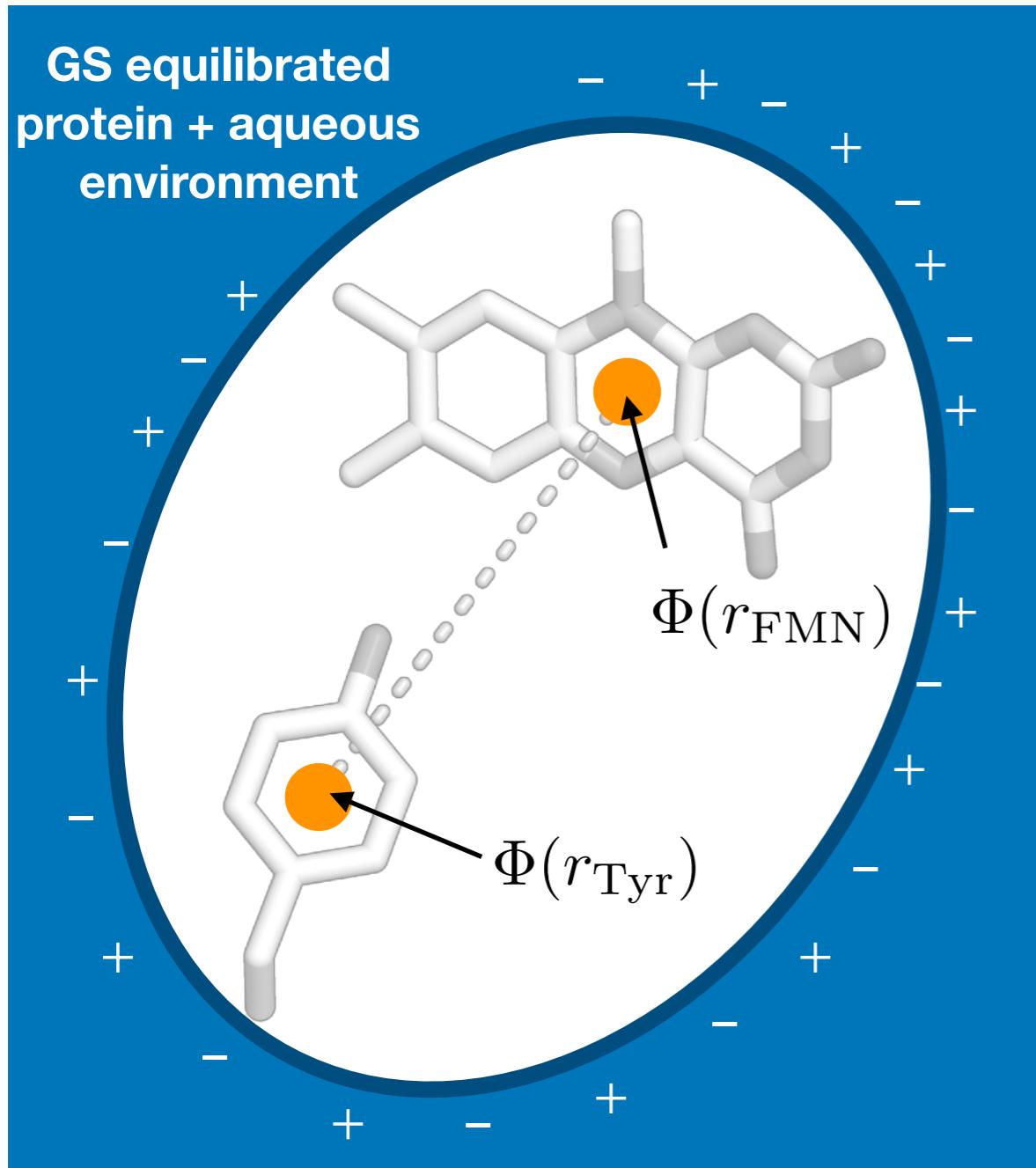
Protein reorganization can induce charge transfer

How does protein environment stabilize CT?



Protein reorganization can induce charge transfer

How does protein environment stabilize CT?

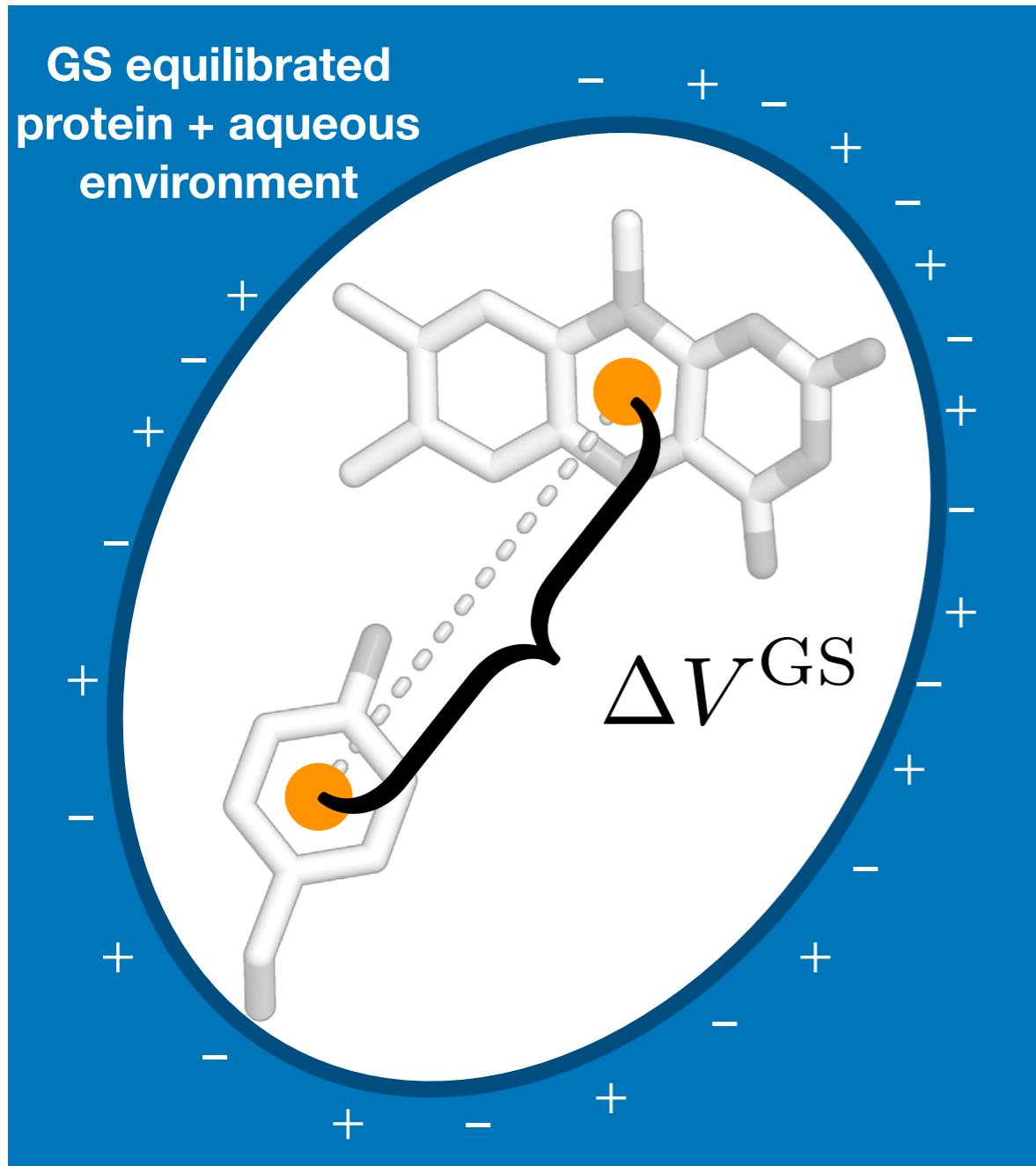


1. Measure electrostatic potential due to **environment** at Tyr and FMN

$$\Phi(r_i) = \frac{1}{4\pi\epsilon_0} \sum_j \frac{q_j}{r_{ij}}$$

Protein reorganization can induce charge transfer

How does protein environment stabilize CT?



1. Measure electrostatic potential due to **environment** at Tyr and FMN

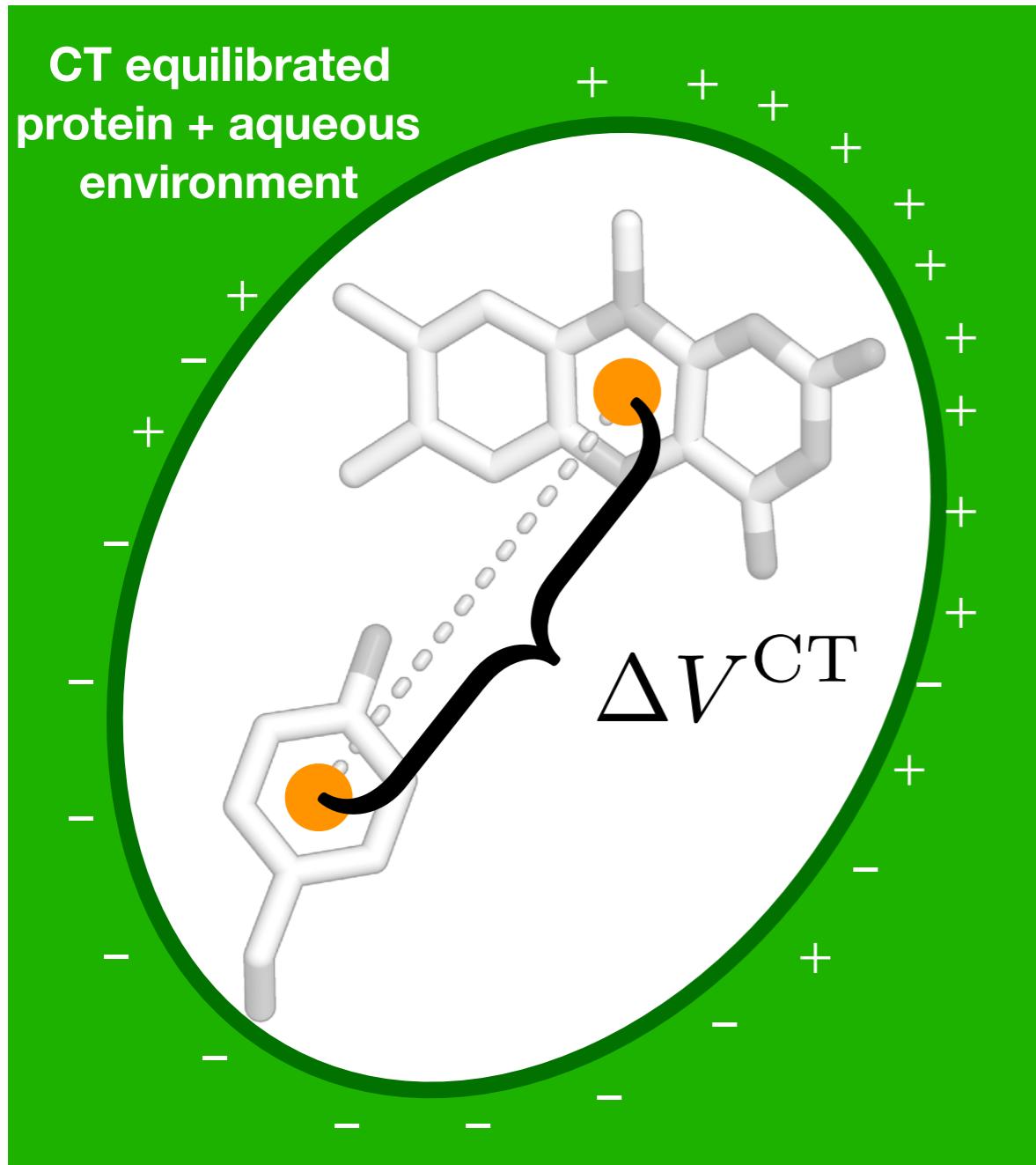
$$\Phi(r_i) = \frac{1}{4\pi\epsilon_0} \sum_j \frac{q_j}{r_{ij}}$$

2. Compute potential difference between FMN and Tyr

$$\Delta V = \Phi(r_{\text{FMN}}) - \Phi(r_{\text{Tyr}})$$

Protein reorganization can induce charge transfer

How does protein environment stabilize CT?



1. Measure electrostatic potential due to **environment** at Tyr and FMN

$$\Phi(r_i) = \frac{1}{4\pi\epsilon_0} \sum_j \frac{q_j}{r_{ij}}$$

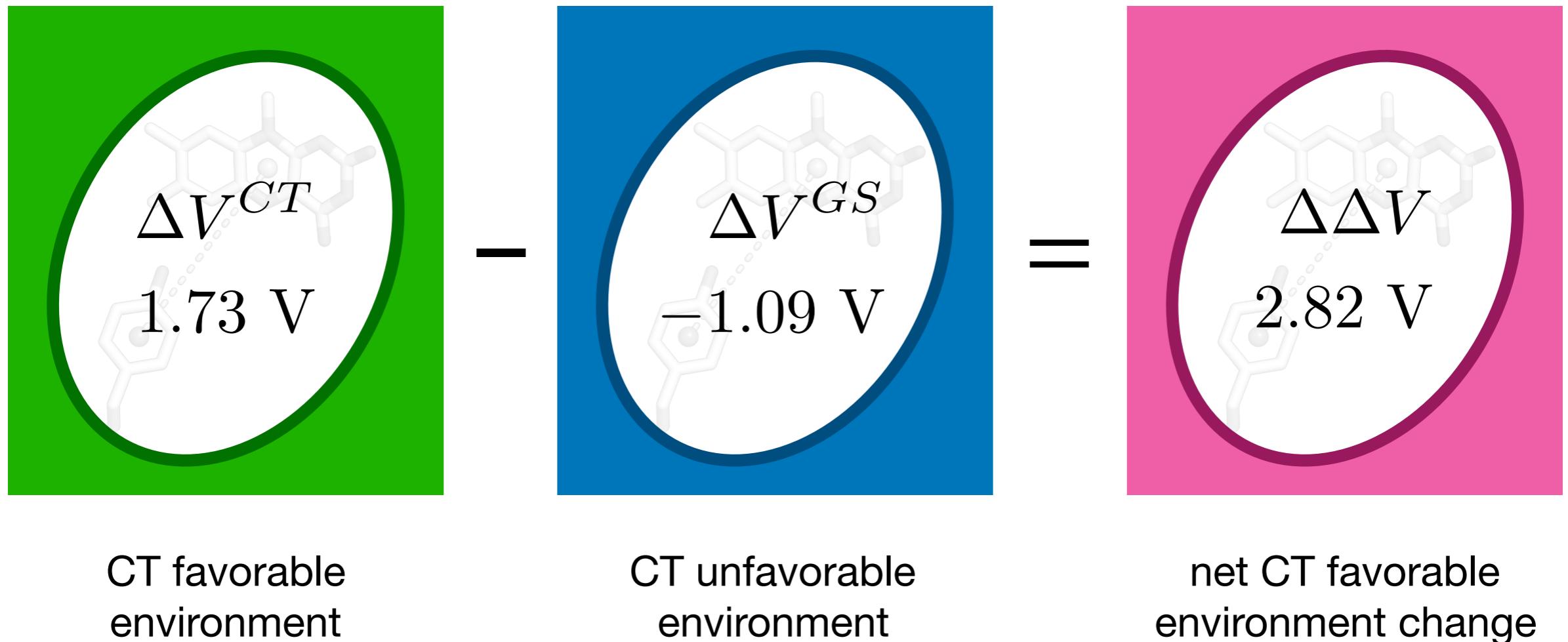
2. Compute potential difference between FMN and Tyr

$$\Delta V = \Phi(r_{\text{FMN}}) - \Phi(r_{\text{Tyr}})$$

3. Perform for both CT and GS. The difference tells how much CT environment stabilizes CT state over GS environment.

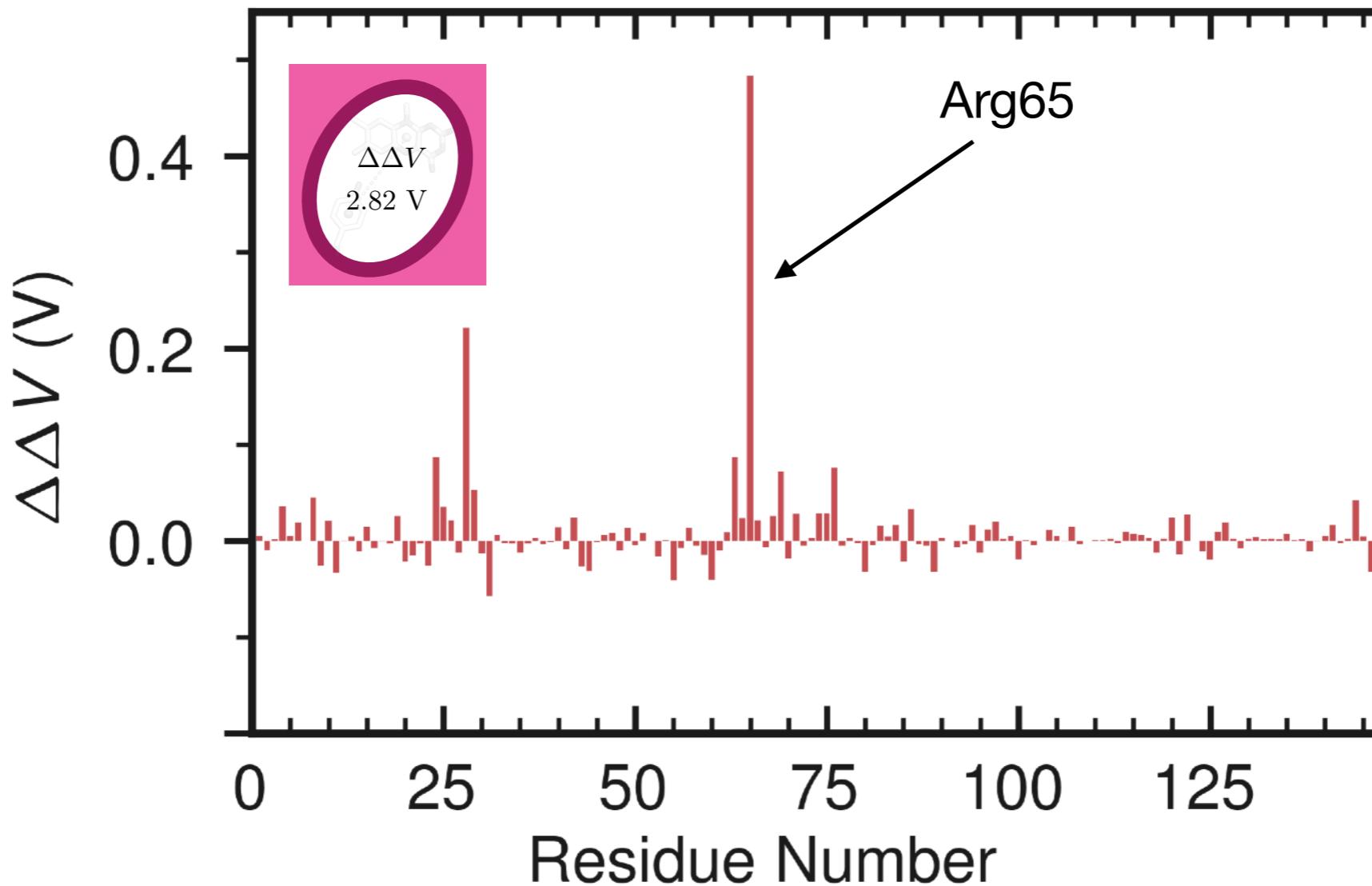
$$\Delta\Delta V = \Delta V^{\text{CT}} - \Delta V^{\text{GS}}$$

Protein reorganization can induce charge transfer



Many ways the protein can reorganize. What is going on in this case?

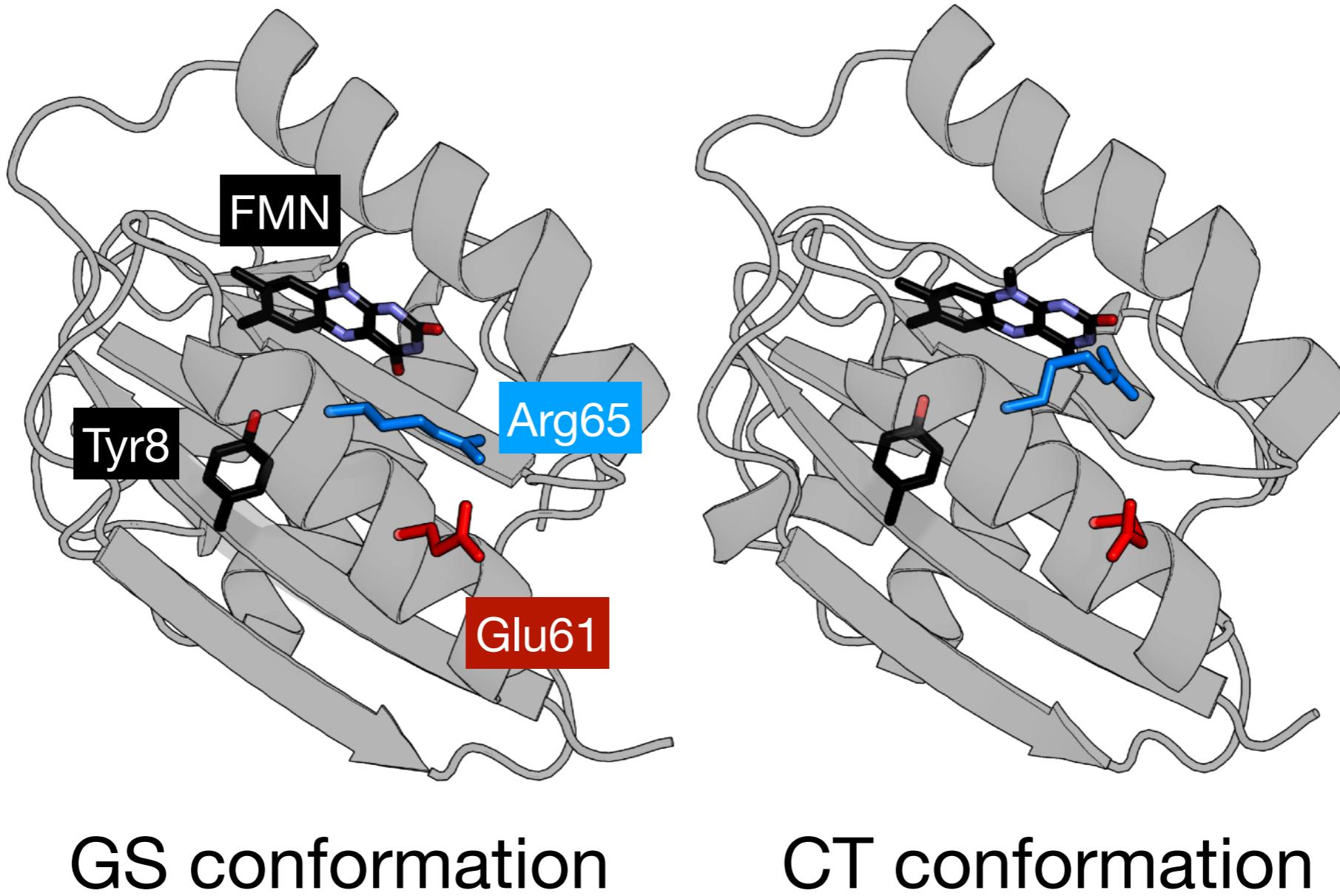
Individual residue contribution to CT stabilization



Potential difference is additive, so we can decompose into contributions by individual residue

Here we will look at the contribution by Arg65

Arg65 contribution to CT stabilization



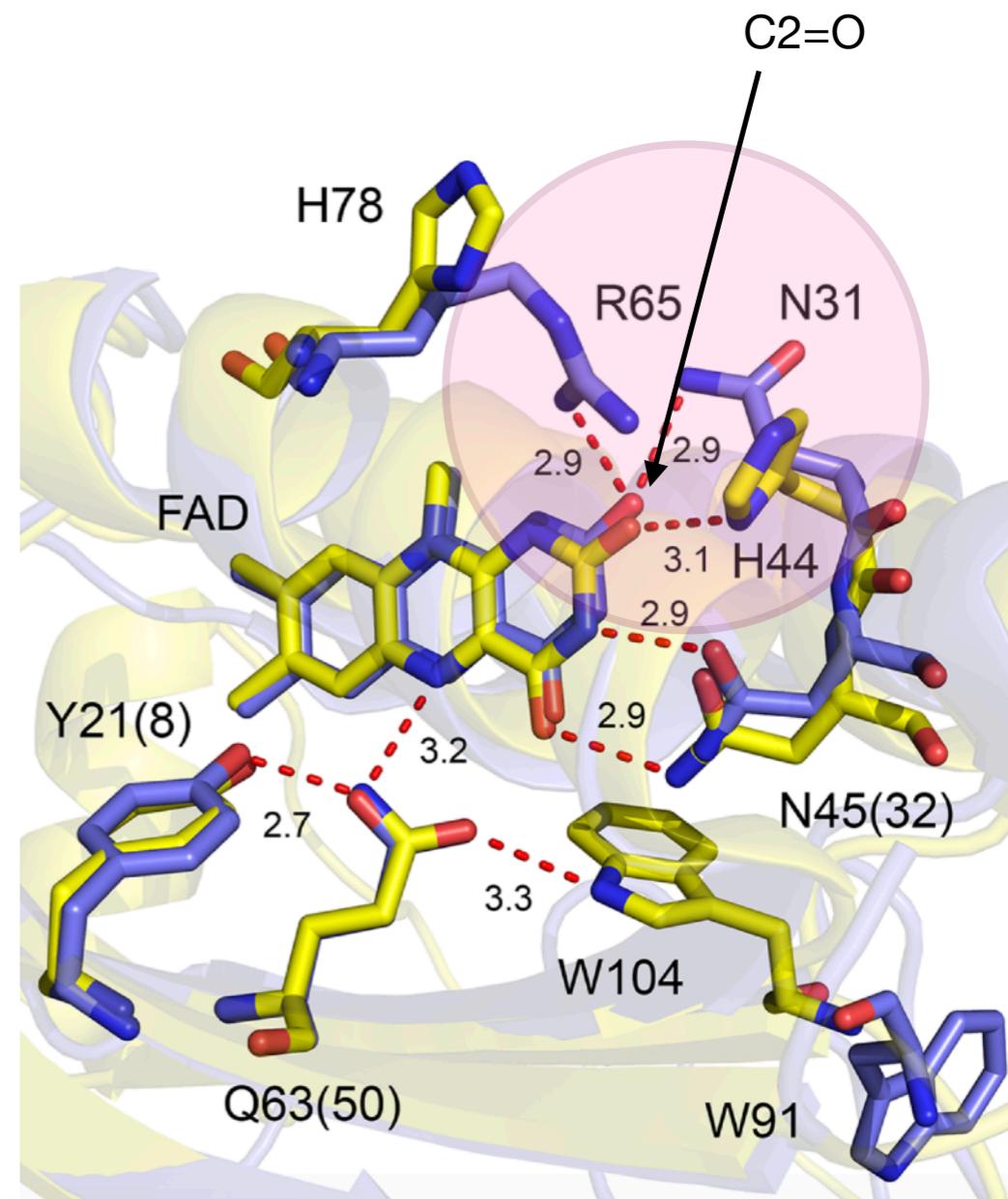
Arg65 ([positively charged](#)) swings toward flavin,
stabilizing potential negative charge buildup on the flavin

Interesting connection to experiment

Gil, et al. point out that BLUF proteins with two hydrogen bonds to C2=O correlate with radical formation.

We see Arg65 facilitating charge transfer to form flavin radical intermediates

Pursuing the impact of changes in the flavin binding pocket may be a fruitful line of research



AppA (yellow) and Slr1694 (blue)
crystal structures

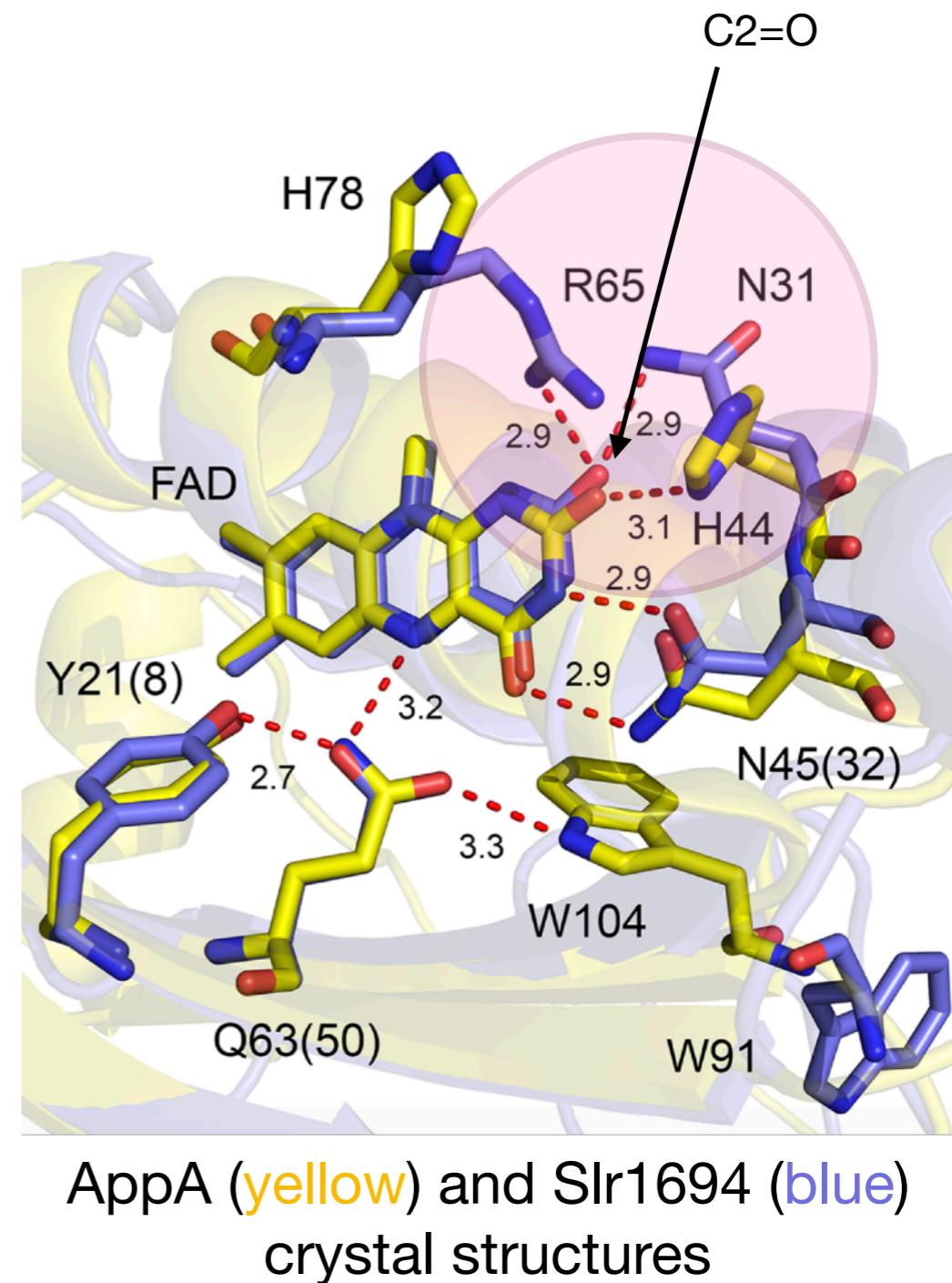
Interesting connection to experiment

Gil, et al. point out that BLUF proteins with two hydrogen bonds to C2=O correlate with radical formation.

We see Arg65 facilitating charge transfer to form flavin radical intermediates

Pursuing the impact of changes in the flavin binding pocket may be a fruitful line of research

BLUF photocycle dynamics are strongly influenced by changes in the protein environment.



Thank you!

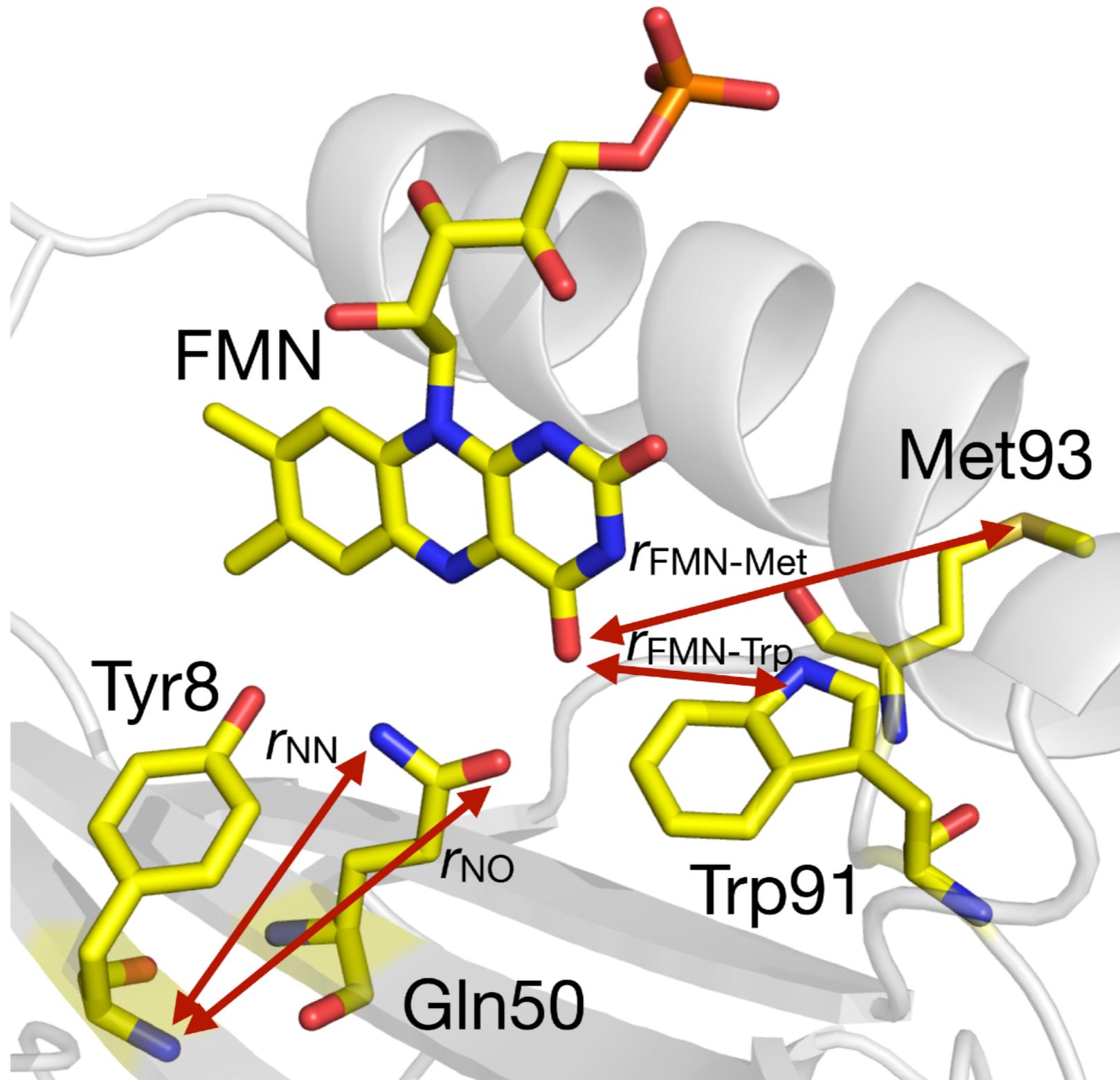
Acknowledgments

- Prof. Sharon Hammes-Schiffer
- Clorice Reinhardt
- Alexander Soudakov
- Puja Goyal
- Archit Vasan

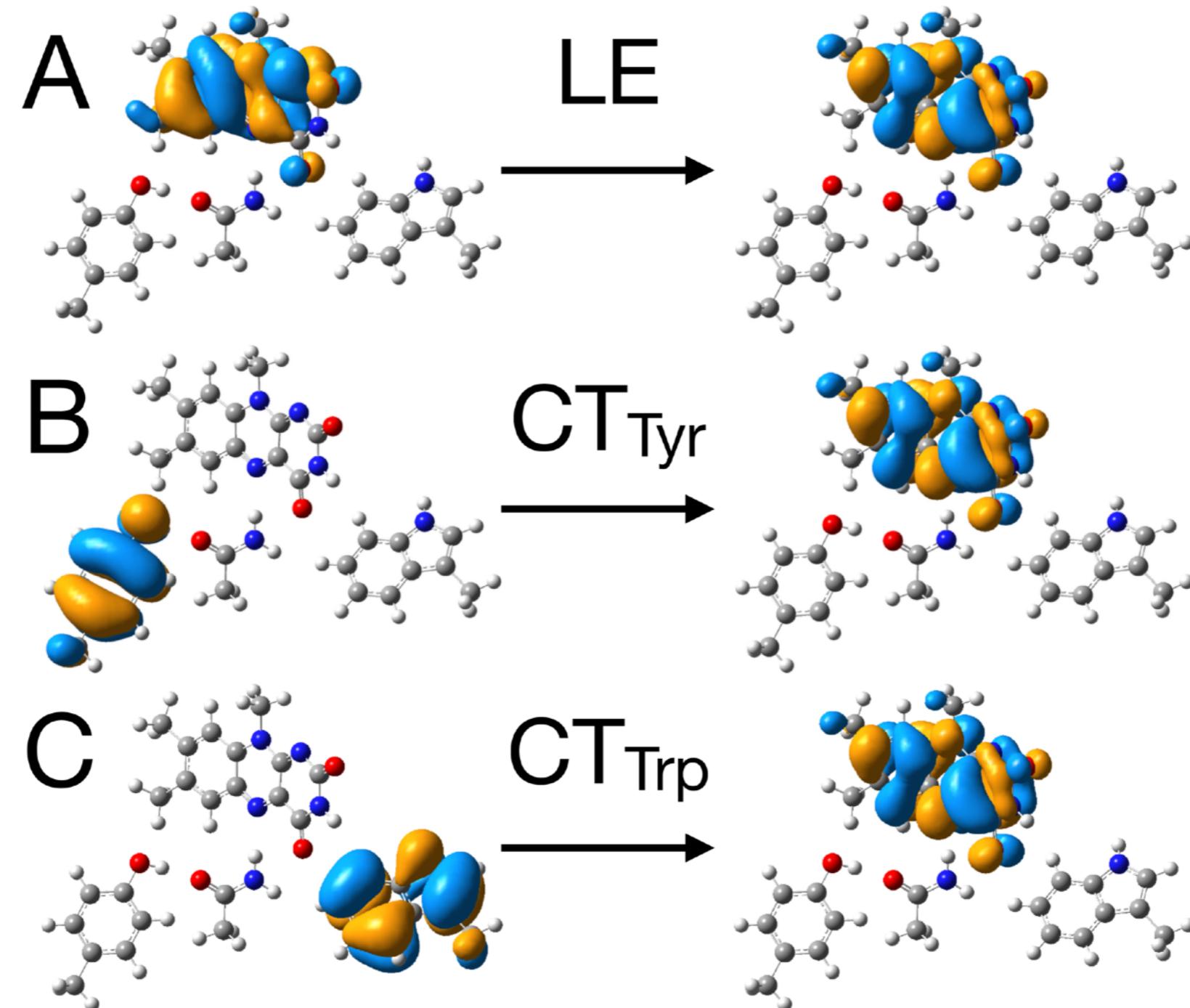


BLUE WATERS

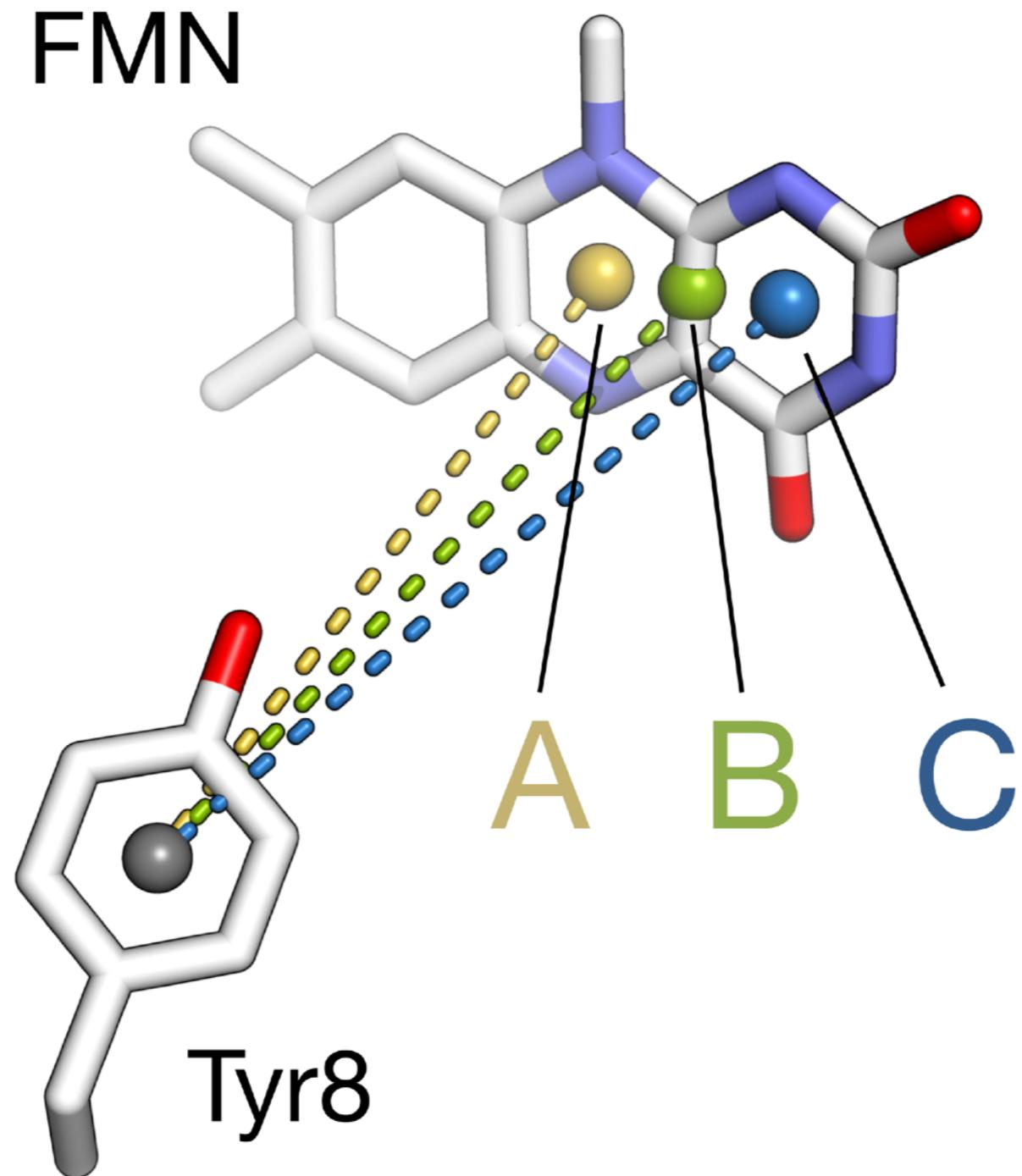
Free-energy reaction coordinates



Natural transition orbitals



Different definitions of $\Delta\Delta V$



Yields nearly identical results, regardless of location

Proton relay facilitates charge transfer to flavin

Table 1. Excitation Energies in eV from TDDFT/CAM-B3LYP/6-31+G** Calculations of Active Site of Slr1694 including Protein and Aqueous Environment Obtained from Ground State Trajectories^{a,b}

| | Trp _{in} (NH _{in}) | | Trp _{in} (NH _{out}) | |
|------------------------------|---------------------------------------|--------------|----------------------------------------|--------------|
| | no proton relay | proton relay | no proton relay | proton relay |
| LE | 3.05 (2.99) | 2.98 (2.95) | 3.00 (3.00) | 2.99 (2.93) |
| CT _{Tyr} | 4.09 (3.95) | 3.25 (3.36) | 4.42 (4.25) | 3.68 (3.17) |
| CT _{Trp} | 3.52 (2.15) | 3.43 (3.25) | 3.46 (2.80) | 3.58 (3.17) |
| CT _{Tyr} -LE gap ΔE | 1.04 (0.96) | 0.27 (0.41) | 1.42 (1.25) | 0.69 (0.24) |
| CT _{Trp} -LE gap ΔE | 0.47 (-0.84) | 0.45 (0.30) | 0.46 (-0.20) | 0.59 (0.24) |

^a The active site is shown in Figure 5, and the protein and aqueous environment was included with electrostatic embedding. The gas phase results are given in parentheses. The conformations were obtained from MD trajectories propagated in the GS, followed by energy minimization.

Protein reorganization can induce charge transfer

Table 2. Excitation Energies in eV from TDDFT/CAM-B3LYP/6-31+G** Calculations of Active Site of Slr1694 including Protein and Aqueous Environment Obtained from Excited State Trajectories^{a, b}

| | Trp _{in} (NH _{in}) / proton relay | | Trp _{in} (NH _{out}) / proton relay | |
|---------------------------------|------------------------------------------------------|--------------|-------------------------------------------------------|--------------|
| | LE conf. | CT conf. | LE conf. | CT conf. |
| LE | 2.90 (2.94) | 2.89 (2.92) | 2.95 (2.93) | 2.87 (2.91) |
| CT _{Tyr} | 3.33 (3.31) | 1.71 (3.10) | 3.77 (3.31) | 1.71 (3.15) |
| CT _{Trp} | 3.64 (3.35) | 3.16 (3.17) | 3.83 (3.48) | 2.15 (3.17) |
| CT _{Tyr} -LE gap ΔE | 0.43 (0.37) | -1.18 (0.18) | 0.82 (0.38) | -1.16 (0.24) |
| CT _{Trp} -LE gap ΔE | 0.74 (0.41) | 0.27 (0.25) | 0.88 (0.26) | -0.72 (0.26) |

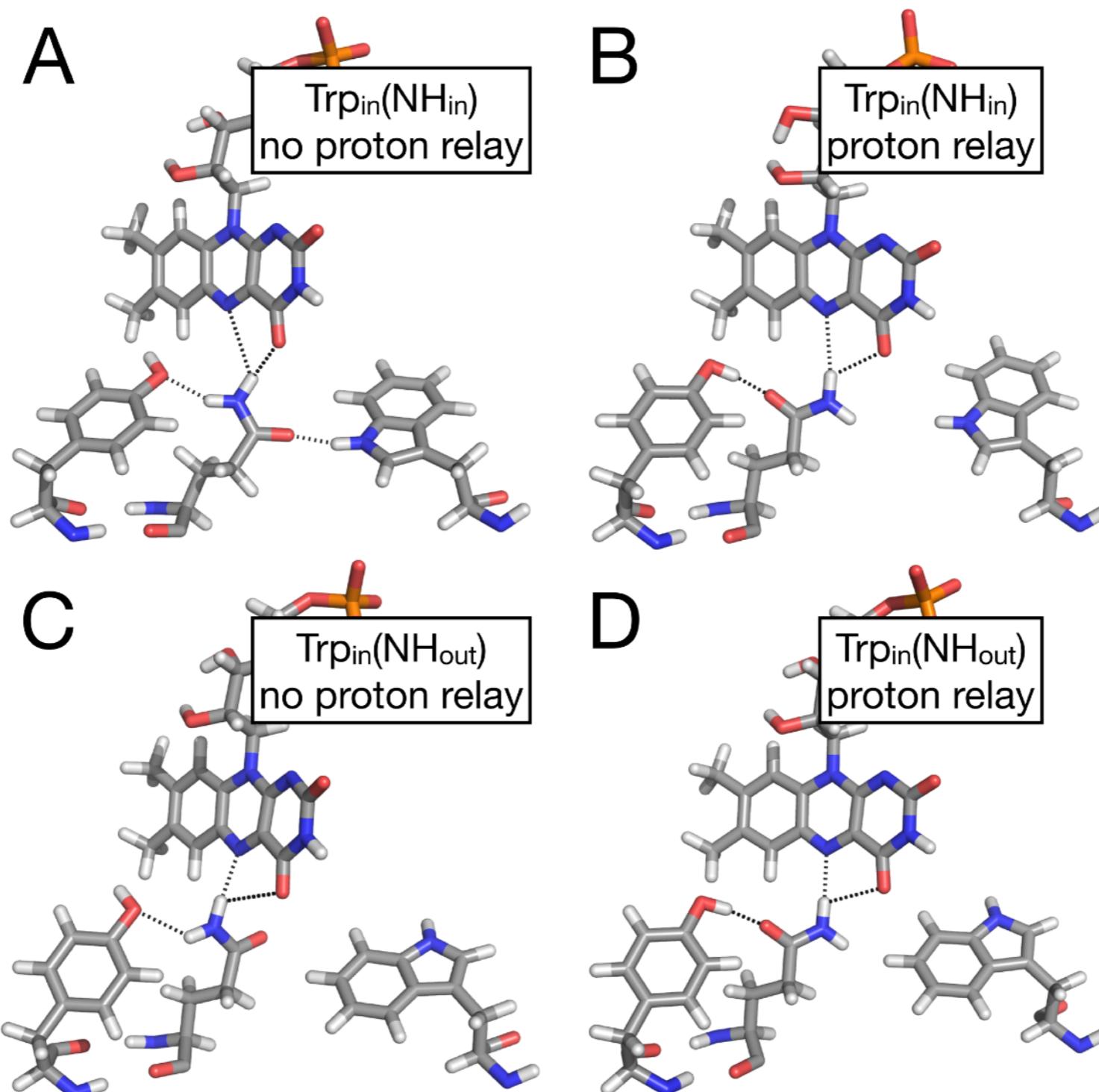
^a The active site is shown in Figure 5, and the protein and aqueous environment was included with electrostatic embedding. The gas phase results are given in parentheses. The conformations (conf.) were obtained from MD trajectories in the LE and CT_{Tyr} state, followed by energy minimization.

Potential differences for both Slr1694 conformations

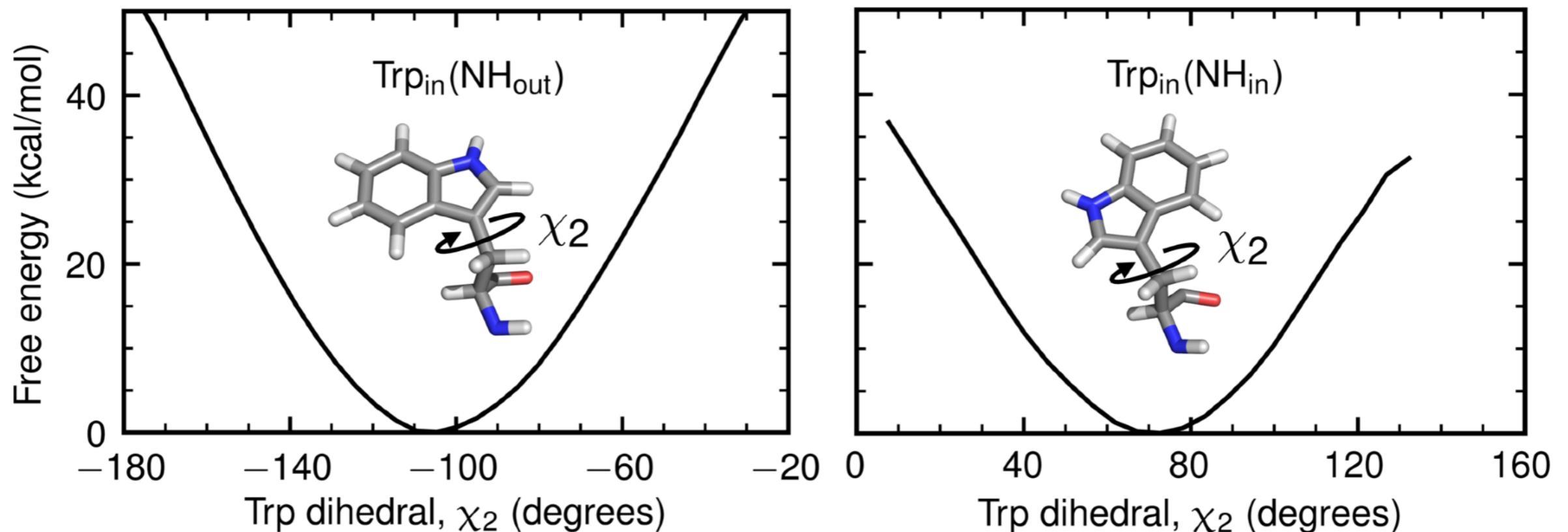
Table 3. Electrostatic Potential Differences in Volts
Between Tyr and FMN Due to Electrostatic
Environment

| | Trpin(NH _{in}) | Trpin(NH _{out}) |
|------------------------|--------------------------|---------------------------|
| ΔV^{GS} | 0.01 | -1.09 |
| ΔV^{CT} | 1.63 | 1.73 |
| $\Delta\Delta V$ | 1.62 | 2.82 |

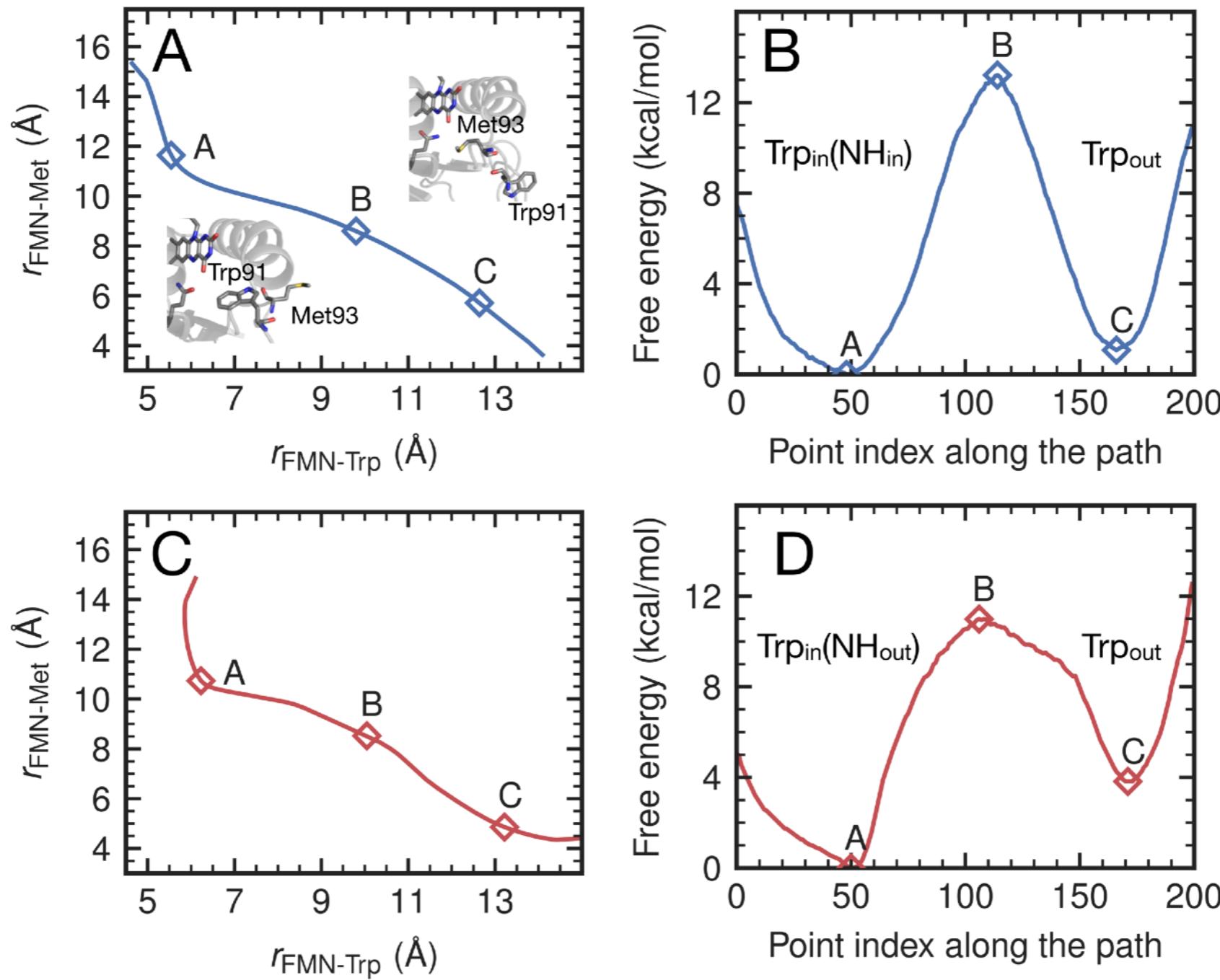
Slr1694 active site conformations



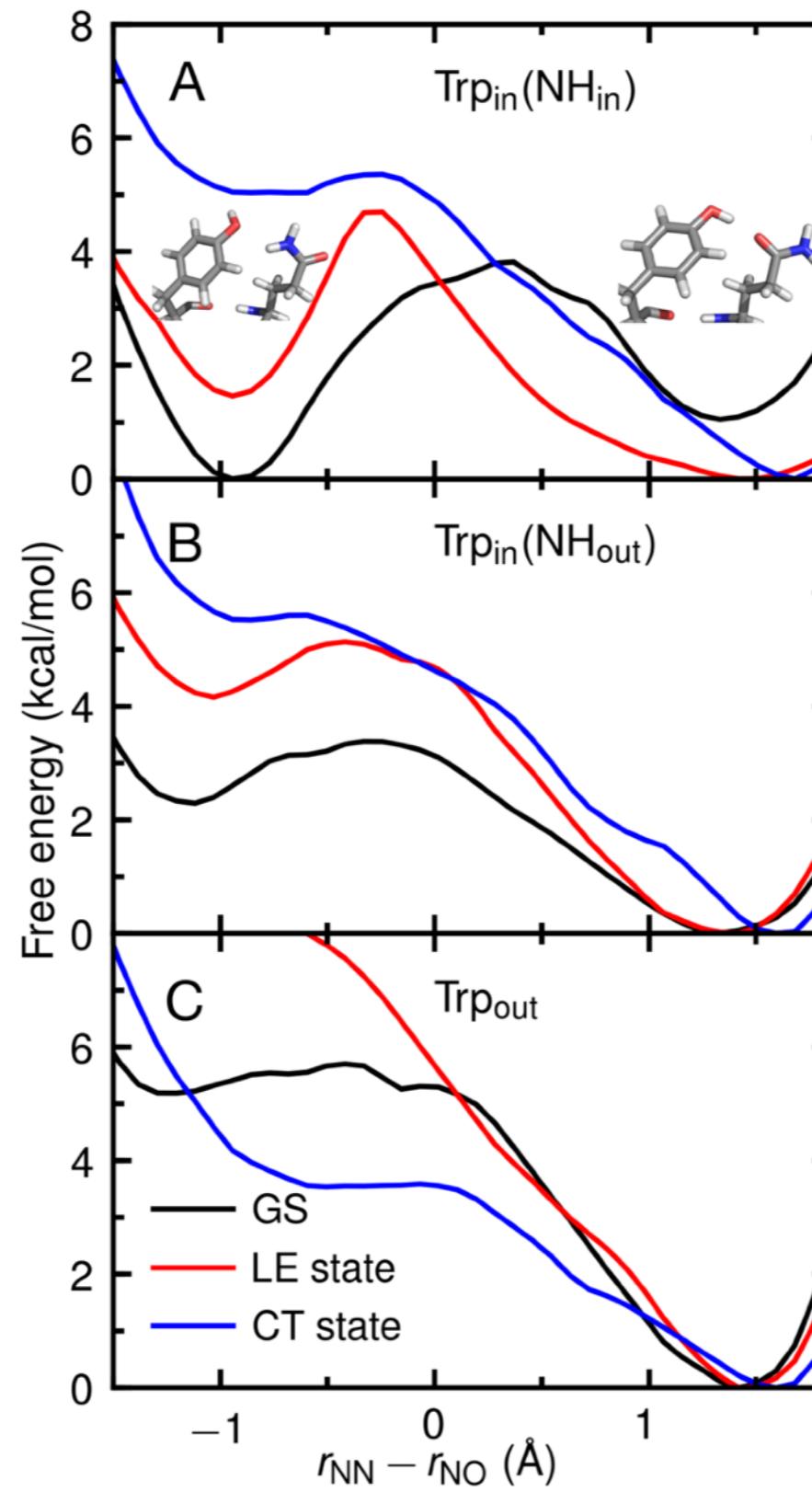
Trp_{in} freedom to rotate internally



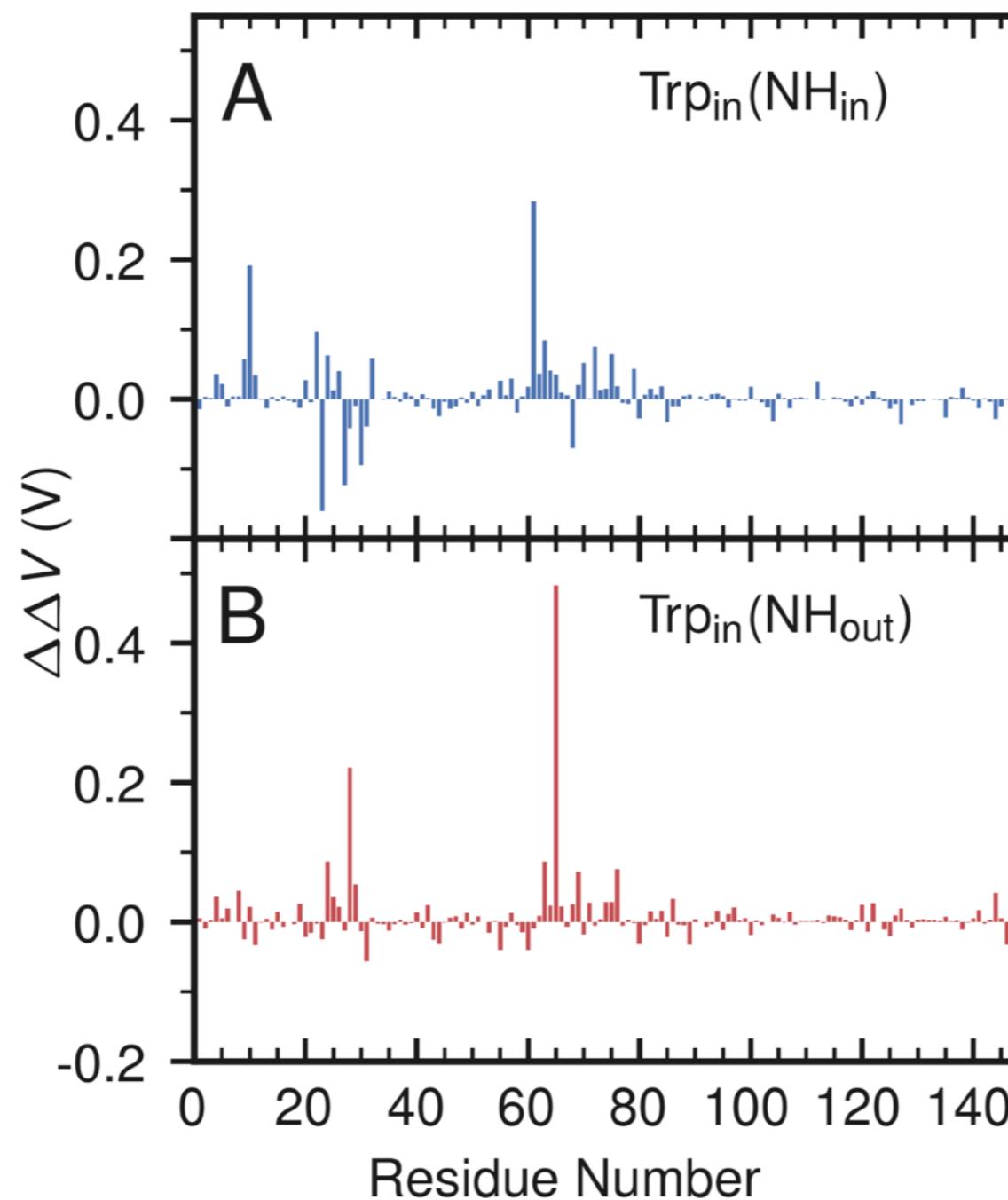
Trp/Met ABPO for two S_{Ir1694} Trp_{in} conformers



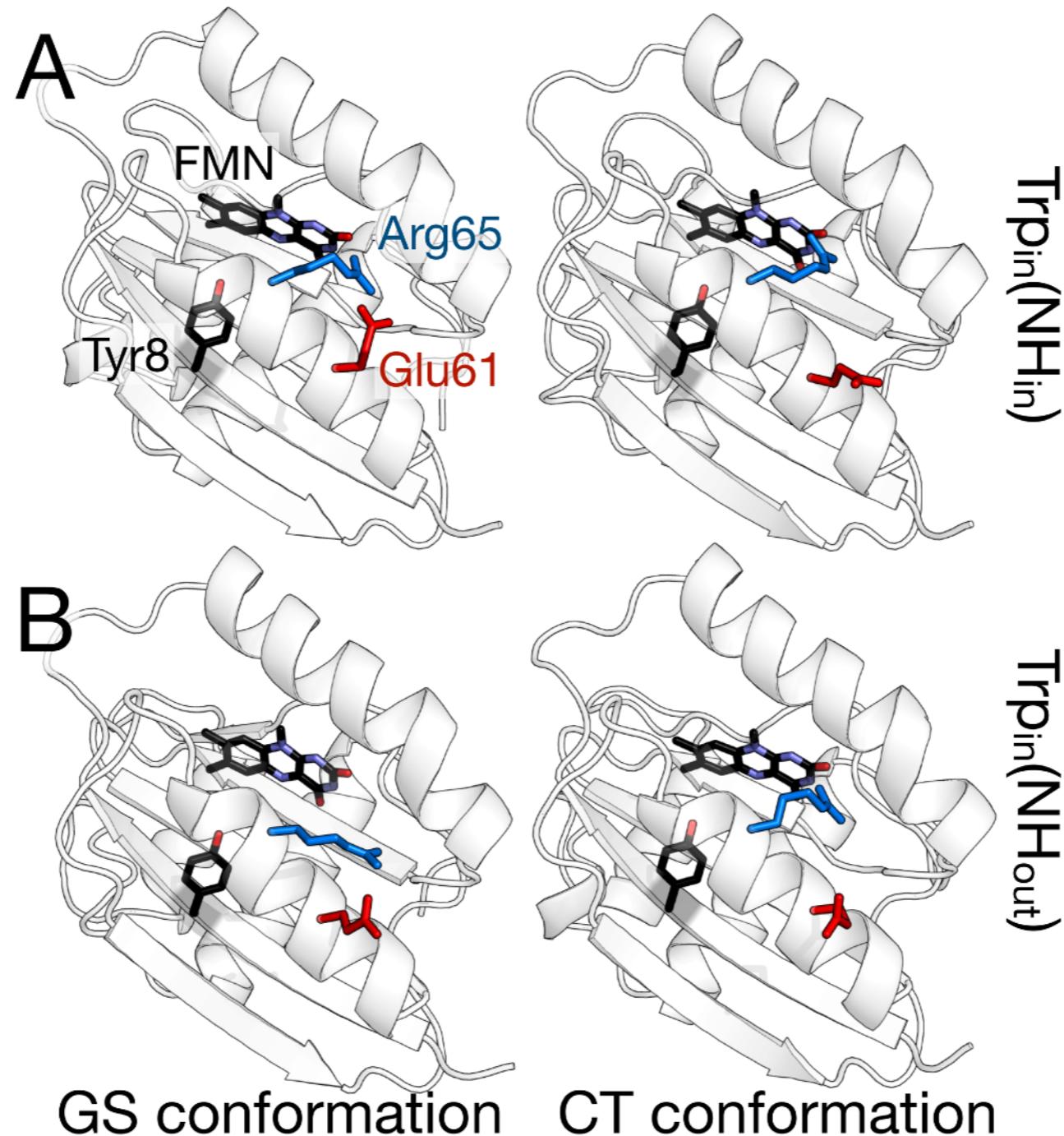
Proton relay for three S_{Ir1694} Trp_{in} conformers



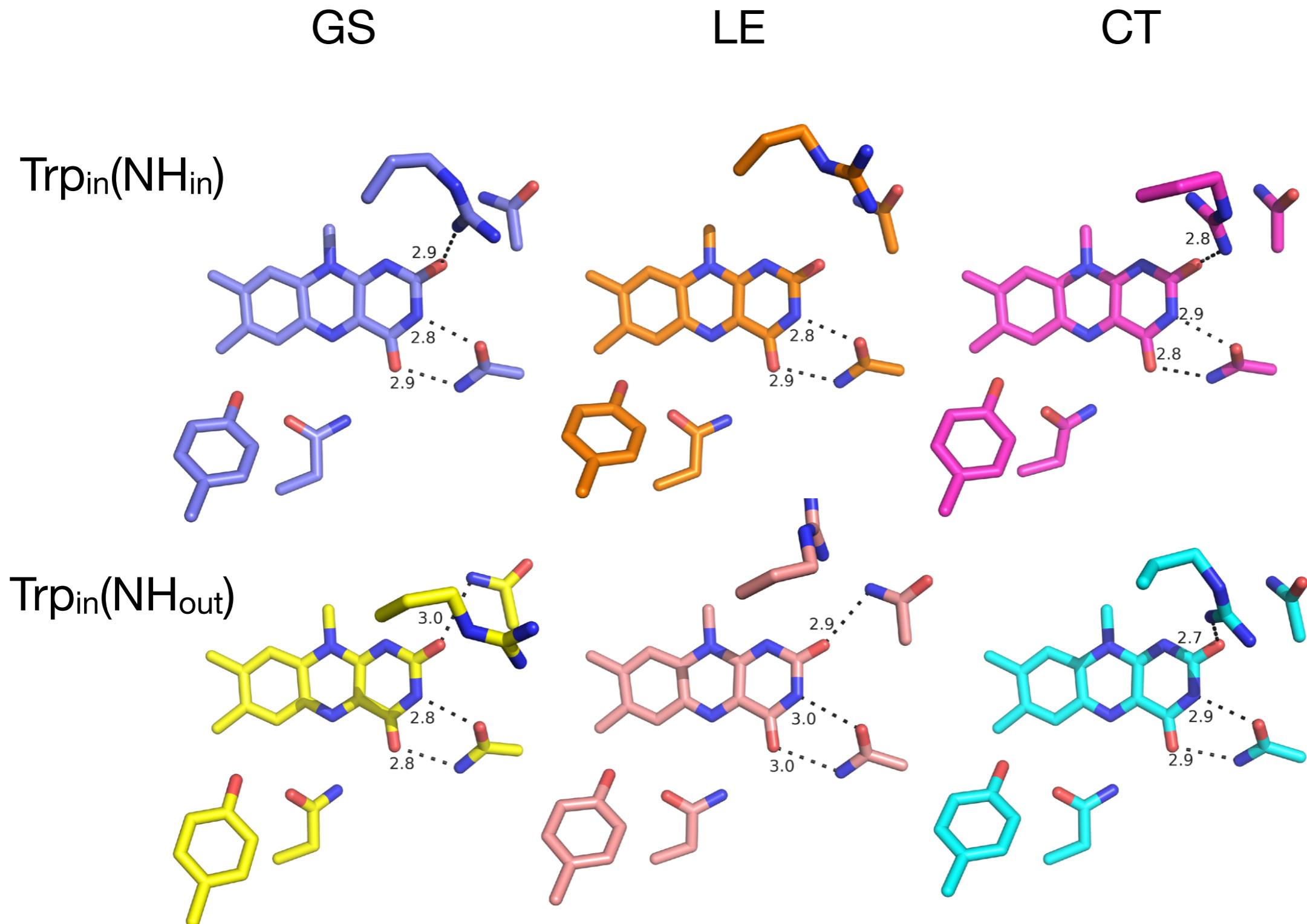
Individual residue contribution to CT stabilization for both Trp_{in} conformers



Arg65 and Glu61 contribution to CT stabilization



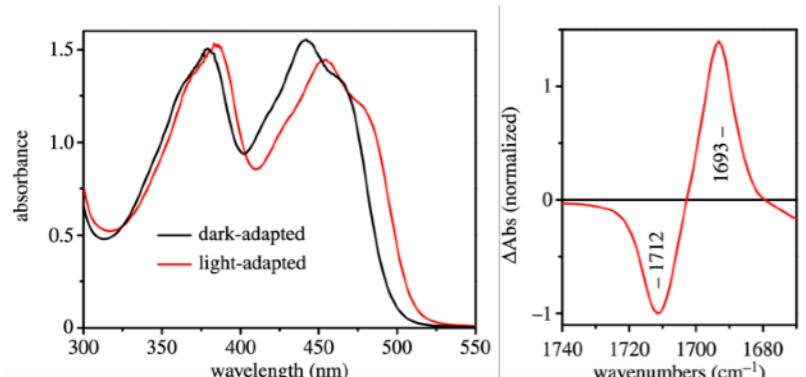
Binding pocket changes in snapshots



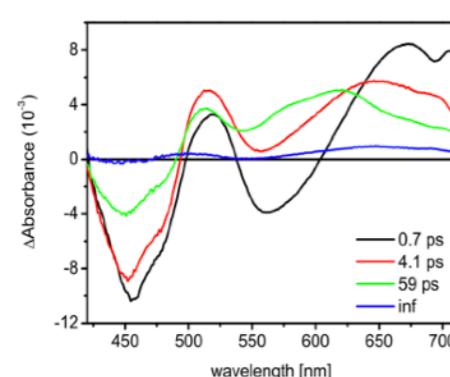
Focus on BLUF Slr1694: what we know

Fast formation of light-adapted state (< 1 ns) – good for dynamics!

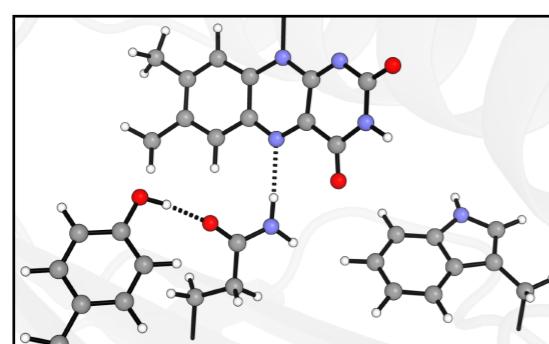
Radical intermediates well-characterized – good for validation!



1. Environment around flavin changes
dark-vs-light adapted states (“off-vs-on”)
10-15 nm red-shift in absorption spectra
20 cm⁻¹ red shift in C4=O stretch of flavin



2. Progresses via flavin redox intermediates
Time-resolved spectroscopy implicates a photoinduced proton-coupled electron-transfer (PI-PCET) mechanism



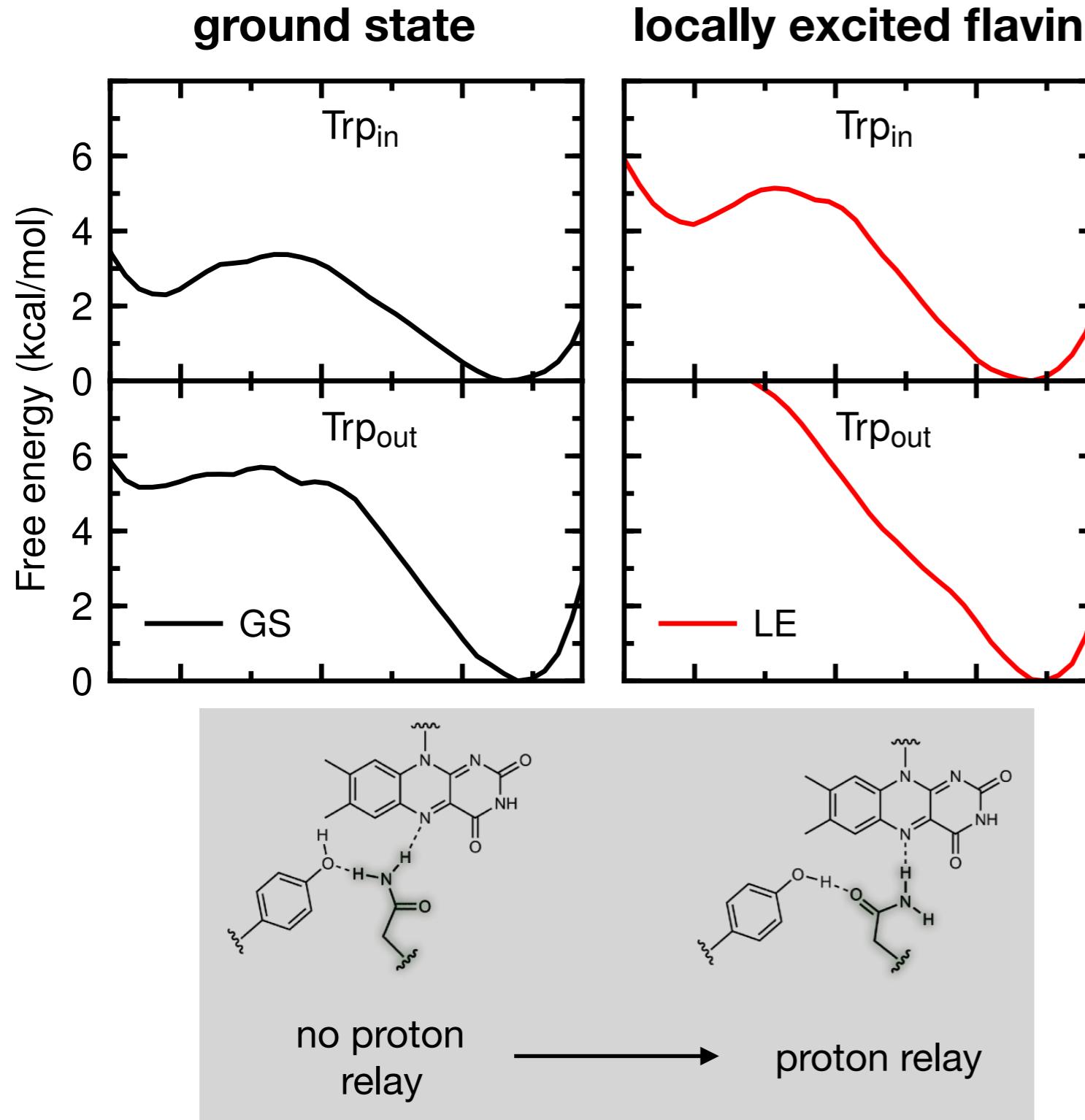
3. Several key residues required
Mutation studies implicate flavin, Tyr, Gln, Trp, Met (more on this next slide)

Currently no consensus on mechanism.

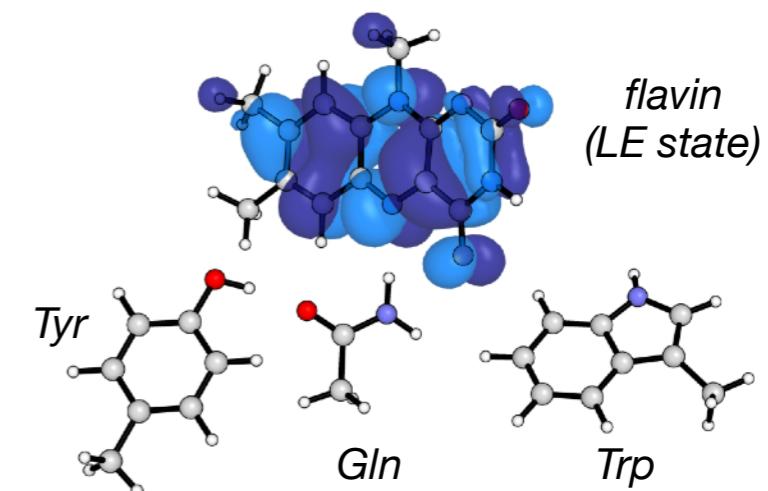
Source: Kennis, J. T. M.; Mathes, T. Molecular Eyes: Proteins That Transform Light into Biological Information. *Interface Focus* 2013, 3 (5), 20130005.

Source: Mathes, T. et al.. Hydrogen Bond Switching among Flavin and Amino Acids Determines the Nature of Proton-Coupled Electron Transfer in BLUF Photoreceptors. *JPCL* 2012, 3 (2), 203–208.

Flavin excitation promotes proton relay



Umbrella sampling with modified partial charges to reflect locally excited flavin



Modify force field with partial charges taken from cluster TDDFT/ CAM-B3LYP

Similar results for charge transfer state between Tyr and flavin (CT_{Tyr}) – proton relay strongly preferred