

# Nucleation and Growth During the Crystalization of Amorphous Te Thin Films

MSE 130: Experimental Materials Science and Design

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## Contents

# 1 Abstract

## 2 Introduction

novel electronic devices such as flexible displays will require the fabrication of semiconductors on noncrystalline substrates...the low melting points of such substrates (such as polymers etc) means that such processes must function on a low thermal budget...the semiconductor tellurium (Te) is an attractive candidate due to its relatively low melting point (449.5 degrees C)...this allows it to be deposited amorphously at low temperature, after which it may be crystallized near room temperature...additionally, the narrow band gap of bulk Te (0.32 eV) allows for straightforward bandgap engineering through electron confinement...depositing and crystallizing Te in a thin film allows its bandgap to be tuned to optimum values for a given application, such as FETs

the amorphous-deposition-crystallization manufacturing process may be understood in the context of classical Becker-Doering nucleation theory, which models the physical system as a distribution of atomic clusters. The free energy change ( $\Delta G$ ) of such a cluster is:

$$\Delta G(i) \approx -g_{c \rightarrow a}i + \gamma i^{2/3} \quad (1)$$

where  $i$  is the number of atoms in the cluster,  $-g_{c \rightarrow a}$  is the atomic free energy of the amorphous-to-crystalline transition, and  $\gamma$  is a factor convolving the cluster surface energy and the surface-area-to-volume ratio. Becker-Doering theory then uses steady-state and detailed-balance assumptions in order to derive the area nucleation rate  $\dot{N}$ :

$$\dot{N} \approx \dot{N}_0 \exp \left[ -\frac{\Delta E_{attach} + \Delta G(i_c)}{k_T T} \right] \quad (2)$$

where  $i_c$  is the critical cluster size and  $\Delta E_{attach}$  is the activation energy for an atom to jump across the amorphous-crystalline interface. The assumption of a spherical nucleus (Equation

1) motivates a critical nucleation energy of  $\Delta G(i_c) = \frac{4\gamma^3}{27g_{c \rightarrow a}^2}$ . The units of  $\dot{N}$  are  $m^{-2}s^{-1}$ .

The crystallization process may be additionally described through Johnson-Mehl-Avrami-Kolmogorov theory in the 2D limit, wherein a nucleus grows to span the thickness of the film faster than the next nucleus can appear. In such a limit, the area fraction transformed ( $\alpha$ ) is described by the JMAK equation:

$$\alpha = 1 - \exp \left[ -\frac{\pi}{3} \dot{N} v^2 t^3 \right] \quad (3)$$

where the growth velocity (in units of  $\frac{m}{s}$ ) follows an Arrhenius dependence:

$$v \approx v_0 \exp \left[ -\frac{\Delta E_{attach}}{k_B T} \right] \quad (4)$$

Finally, JMAK theory produces an expression for the average number of grains nucleated per unit area:

$$\rho = \left( \frac{3}{\pi} \right)^{\frac{1}{3}} \Gamma \left[ \frac{4}{3} \right] \left( \frac{\dot{N}}{v} \right)^{\frac{2}{3}} \quad (5)$$

where  $\Gamma \left[ \frac{4}{3} \right] \approx 0.8929795...$  is the Euler Gamma function.

Taken collectively, B-D theory and JMAK theory yield a straightforward method for calculating  $\dot{N}$ ,  $v$ , and their associated activation energies and prefactors. An amorphous Te thin film may be monitored during its anneal in order to measure the crystalline area fraction as a function of time, which may be fitted to Equation 3 to calculate  $\dot{N}v^2$ . Afterwards, the grain density may be assessed and Equation 5 may be employed to calculate  $\dot{N}/v$ . These values may be trivially solved for  $\dot{N}$  and  $v$ .

Repeating the procedure at different annealing temperatures allows the Arrhenius forms (Equations 4 and 2) to be fit, thus providing estimates of the activation energies governing nucleation and growth in the low-temperature Te manufacturing process.

The equations presented in this section represent the core results of B-D and JMAK theory. For a detailed discussion of their derivations and assumptions, as well as a demon-

stration of the validity of the 2D limit, see Reference [1].

### 3 Experimental Procedure

An Edwards 306 Thermal Evaporator was loaded with Te source pellets of 99.999% purity from Sigma-Aldrich. Te was deposited on a substrate at a deposition rate of approximately 10 Å/sec. Film thicknesses, as controlled by a quartz crystal microbalance, ranged from 4 to 20 nm. Since the substrate was maintained at -80°C, the deposited film was amorphous; this was verified by x-ray diffraction.

Six deposited films were selected for experimental analysis. Immediately after deposition, each film was heated to its target annealing temperature (10°C, 15°C, 20°C, 25°C, 30°C, or 35°C). Optical microscopy was used to observe the crystallization at an imaging rate of 2.87 frames/sec, with the intent that these data be used to assess the area fraction transformed (Equation 3).

After annealing, the samples were allowed to return to room temperature, then examined via transmission electron microscope. Bright field and dark field images of a 140 µm x 105 µm region were obtained, with the intent that they be used to determine the films' grain density (Equation 5). Due to their finer microstructure, additional images of a zoomed-in 140 µm x 105 µm region were collected for the 20°C, 25°C, 30°C, and 35°C samples.

### 4 Results

obtaining and fitting fraction transformed curves

assessing grain density from microstructural images

calculating parameters and estimating free energies/critical nuclei

#### 4.1 JMAK Equation Fitting to Determine $\dot{N}v^2$

For each annealing temperature, a series of optical microscopy images were collected...it was observed that crystalline regions corresponded to pixels of greater intensity...One method considered for determining a crystalline fraction from a given image would be to binarize the pixels according to a determined intensity threshold, then determine the area fraction of pixels above the threshold...This was, however, sensitive to any debris on the lens...the decision was therefore made to consider the crystalline fraction as a function of the average intensity, and to normalize the curve to its maximum and minimum values afterwards to produce the  $\alpha(t)$  curve...upon doing so, there were several jumps in average intensity, corresponding to automatic intensity rescaling done by the imaging system itself...rather than try to vertically shift the regions of the graph, the equation was only fit to the transition region of each curve...the actual fit employed a modified version of the JMAK equation:

$$\alpha = \begin{cases} 0 & t < t_0 \\ 1 - \exp \left[ -\frac{\pi}{3} \dot{N}v^2 (t - t_0)^3 \right] & t_0 \leq t \end{cases} \quad (6)$$

...nonlinear fitting to  $Nv^2$  and  $t_0$  was performed with the Python package `lmfit`..., which also estimated the standard error associated with each of the fitting parameters...this standard error is propagated through using the well-known result:

use equation here!

the code is contained in Appendix ??...the results for each fitting are summarized in Table ??...the curve fits are shown in Figure ??.

NEED table summarizing results ( $Nv^2$ , se- $Nv^2$ ,  $t_0$ )

NEED figure containing all six datasets on their own axes, as well the actual curve fits

NEED appendix figure containing the same information, but with all data points (highlighting those that were used for the fit) and the zero-value assigned point

## 4.2 Microstructural Image Processing to Determine $\dot{N}/v$

following the procedures of (cite papers here), a rudimentary algorithm was developed (largely ythrough trial and error) in order to process the microstructural images and count grains in a reliable manner...only the 140-micron images processed...the algorithm does the following using the skimage package:

1. merge the bright and dark field images, since it was observed that the bright field and dark field images highlighted different boundaries (boundary not visible in one might be visible in the other); dark field was simply converted to greyscale, while the bright field was converted to greyscale and the sobel gradient algorithm applied to locate boundaries...to "normalize" and to maximize contrast, the two images were then independently adjusted so that the middle 95% of pixel intensities were stretched across the range  $[0, 1]$ ...the maps were then averaged and the result inverted to create a map of grains, where dark regions are boundaries

2. so as to remove some noise and sharpen grain boundaries, gaussian smoothing was applied over cross-shaped pixel elements, followed by gaussian sharpening over a disk of radius 5 pixels

3. multiple otsu thresholding was used to sort the pixels into four different intensity bins; this was done so that small-angle grain boundaries (which were not as dark in the image) were caught by one of the intermediate bins, and the assumed grains corresponded only to the highest bin.

4. a "weighted opening" was executed, wherein the grains were eroded by a specified input radius, then dilated by half of that radius...the intermediate results were stacked, thereby creating a heatmap of the number of cycles required to eliminate any given pixel...essentially a smoothed version of calculating the distance to the nearest grain boundary

5. local maxima (of a radius specified by the minimum grain size) were counted as the locations of grains

6. for illustrative/diagnostic purposes, the watershed algorithm was used with the local

maxima as markers and the heatmap as elevations in order to produce a grain map and assess the quality of the algorithm

input parameter (min grain size) as well as number of grains, grain densities are presented in TABLE ??, along with NV results of inputting into the result from JMAK theory

figure illustrating the process (bright field, dark field, merged map, multiple otsu, heatmap, grain map) for one of the samples, with all such diagrams contained in Appendix B

discussion of the quality of the algorithm deferred for the Discussion, particularly with respect to the decision not to quantify error

### 4.3 Calculation of Crystallization Parameters

From  $Nv^2$  and  $N/v$ ,  $N$  and  $v$  were obtained for each temperature. (Results summarized in TABLE ??) These values were then fit in  $1/T-\ln$  coordinates so as to estimate the prefactor and the activation energies; the results of which are contained in Figures ?? and ?. All standard errors were propagated according to equation ?.

TABLE containing results for  $T$ ,  $N$ ,  $v$ , standard errors

TABLE summarizing results of the fit ( $v_0$ ,  $N_0$ ,  $\Delta G + E_{attach}$ ,  $E_{attach}$ ) and their uncertainties

FIGURE showing the fit results both in realspace and logspace

subtracting the activation energies yields  $\Delta G$  along with its standard error...of particular interest is the estimation of the critical cluster size and the free energy of crystallization  $g(c; a)$ ; however, as evidenced by the  $\Delta G$  equation, this requires knowledge of the gamma factor. A loose estimate may be obtained by the Materials Project in terms of average surface free energy; one would expect the corresponding free energy to be somewhere between this and zero...A spherical cluster is assumed, for which the nucleation barrier is typically calculated through EQUATION...recognizing that  $V = \omega \cdot i_c$ , the critical radius is  $(r_c)$ , the critical area is  $(A_c)$ , we can rewrite this in the form EQUATION...gamma can therefore be calculated as EQUATION and, from the formula for critical nucleation energy of a spheri-



cal nucleus,  $g(c-i)a$  can be calculated as ??...ic then can be calculated as EQUATION. A plot of the relationship between the (unknown) surface energy of the crystalline-amorphous interface and the resulting critical nucleus size is shown in figure ???

FIGURE showing relationship between gamma and ic (atoms)

## 5 Discussion

discussion of validity of free energy and critical nuclei result

discussion of validity of grain counting algorithm

discussion of improvements to the estimation procedure and attempt to derive gamma

## 6 Conclusions

## 7 Acknowledgments

## 8 References

<sup>1</sup>D. C. Chrzan, *MSE 130 Laboratory: Nucleation and Growth During the Crystallization of Amorphous Te Thin Films*, tech. rep. (Department of Materials Science and Engineering, University of California, Berkeley, 2020).