

Molecular Weight Distributions in Step-Growth Polymerizations

Model 1: Probabilities of Forming Different Chain Lengths

Suppose we perform a step-growth polymerization of AB-type monomers and stop the polymerization at extent of reaction p (i.e. we stop the polymerization when the fraction of A groups that have reacted is equal to p).

Suppose we then select a single molecule from this reaction mixture. This molecule will have an unreacted A group on one end, and an unreacted B group on the other.

Consider the following argument:

1. The unreacted 'B' group was originally part of an 'AB' monomer. Of these AB monomers ($i = 1$),

- The fraction in which the A group monomer did *not* react, and the molecule remained an AB monomer, is $1 - p$.
- The fraction in which the A group *did* react, and the selected molecule is at least an AbaB dimer, is p .

2. Of the molecules that reacted to form AbaB dimers ($i = 2$),

- The fraction *of dimers* in which the A group did not react, and the molecule remained an AbaB dimer, is $1 - p$. The total fraction *of molecules* that are AbaB dimers is thus

$$\begin{aligned} & (\text{fraction of molecules that form dimers}) \cdot (\text{fraction of dimers that don't react further}) \\ & = p(1 - p) \end{aligned}$$

- The fraction *of dimers* in which the A group did react, and the selected molecule is at least an AbabaB trimer, is p . The total fraction *of molecules* that form at least an AbabaB trimer is thus

$$\begin{aligned} & (\text{fraction of molecules that form dimers}) \cdot (\text{fraction of dimers that react further}) \\ & = p \cdot p = p^2 \end{aligned}$$

3. Of the molecules that reacted to form AbabaB trimers ($i = 3$),

- The fraction *of trimers* in which the A group did not react, and the molecule remained an AbabaB trimer, is $1 - p$. The total fraction *of molecules* that are AbabaB trimers is thus

$$\begin{aligned} & (\text{fraction of molecules that form trimers}) \cdot (\text{fraction of trimers that don't react further}) \\ & = p^2(1 - p) \end{aligned}$$

- The fraction *of trimers* in which the A group did react, and the selected molecule is at least an AbababaB tetramer, is p . The total fraction *of molecules* that form at least an AbababaB tetramer ($i = 4$) is thus

$$\begin{aligned} & (\text{fraction of molecules that form trimers}) \cdot (\text{fraction of trimers that react further}) \\ & = p^2 \cdot p = p^3 \end{aligned}$$

Critical Thinking Questions:

1. Following this reasoning,

a) How would you calculate the fraction of *molecules* that *remain* as AbababaB tetramers ($i = 4$)? Write your answer in both words and symbols.

b) How would you calculate the fraction of molecules that react further to form at least an AbabababaB pentamer ($i = 5$)? Write your answer in both words and symbols.

2. Using the information in the model, and your answers to the previous question, fill in the following table:

i	Fraction of molecules that contain exactly i monomers
1	
2	
3	
4	
5	

3. What pattern do you notice in these values? Briefly describe your observations in 1-2 complete sentences.

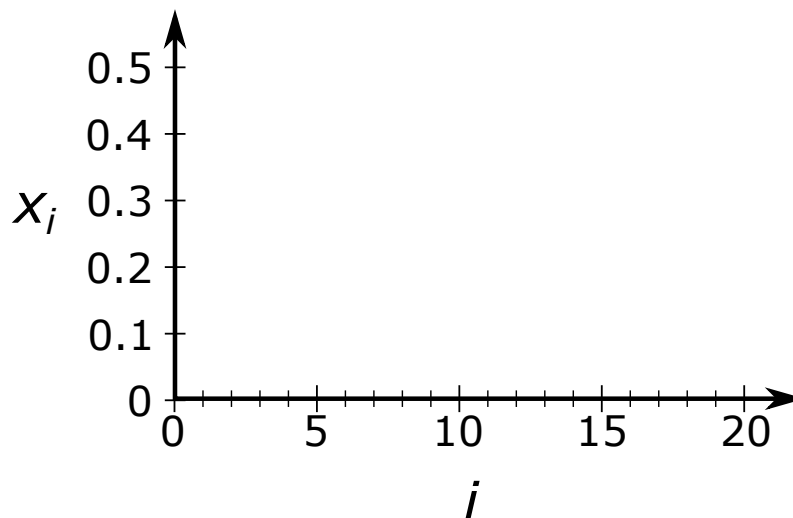
4. Complete the following statement:

“The fraction of molecules, x_i , that are composed of exactly i monomers is _____.”

5. Using this expression, calculate the fraction of molecules that have exactly length i for both $p = 0.5$ and $p = 0.9$ at the following values of i :

i	x_i when $p = 0.5$	x_i when $p = 0.9$
1		
2		
3		
5		
10		
15		
20		

6. Plot your results on the following axes. Make sure to use a different symbol for points corresponding to $p = 0.5$ than for the points corresponding to $p = 0.9$.



7. How are the plots for $p = 0.5$ and $p = 0.9$ similar, and how are they different? Briefly describe your observations in 2-3 complete sentences.

8. What is the *most probable* chain length for each value of p ?

9. Can the fraction of chains with length $i + 1$ ever be *greater* than the fraction of chains with length i ? Justify your answer in 1-2 complete sentences.

Model 2: M_n and M_w for Step-Growth Polymerizations

To calculate M_n and M_w , we need to know n_i , or the total number of chains with i monomers.

If we started with v_A^0 monomers, then when the extent of reaction is equal to p , there will be $(1-p)v_A^0$ unreacted A groups left. Recalling that the number of unreacted A groups is equal to the number of molecules in the reaction mixture, this lets us write

$$\begin{aligned} n_i &= (\text{fraction of molecules that have length } i) \times (\text{number of molecules in reaction mixture}) \\ &= (p^{i-1}(1-p)) ((1-p)v_A^0) \\ &= p^{i-1}(1-p)^2 v_A^0 \end{aligned}$$

If we plug this expression into our equation for M_n , we get

$$M_n = \frac{\sum_i n_i M_i}{\sum_i n_i} = M_0 \frac{\sum_i p^{i-1}(1-p)^2 i}{\sum_i p^{i-1}(1-p)^2}$$

where M_0 is the molecular weight of the monomer ($M_i = M_0 i$).

Evaluating these sums is a bit tedious, but if we do so, we obtain

$$M_n = \frac{M_0}{1-p} \quad \text{or} \quad N_n = \frac{M_n}{M_0} = \frac{1}{1-p}$$

which is exactly what we expected (whew - our math worked!).

Similarly, if we plug this expression into our equation for M_w and work through the sums, we get

$$M_w = \frac{\sum_i n_i M_i^2}{\sum_i n_i M_i} = M_0 \frac{1+p}{1-p} \quad \text{or} \quad N_w = \frac{M_w}{M_0} = \frac{1+p}{1-p}$$

Critical Thinking Questions:

- Calculate the dispersity for a step-growth reaction with extent of reaction p .

11. What is the value of the dispersity when $p = 0$? Briefly comment on whether or not this answer makes sense.

12. What is the value of the dispersity when $p = 1$?

13. Can the dispersity of a polymer produced by step-growth polymerization ever be greater than 2? Briefly defend your answer in 1-2 complete sentences.

Exercises:

1. Suppose you synthesized a polymer by step-growth polymerization and found that it had a dispersity of 1.86.
 - a) What must the extent of reaction have been in this polymerization?
 - b) What would you expect the number-average degree of polymerization of this polymer to be?