

# Preliminary Oral Defense: Designing Preorganized Chelators and Understanding Approximation Error Propagation

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# Designing Preorganized Chelators for Nuclear Fission Product Separations

## Why do we care?

Being able to separate out nuclear fission products allows for characterization of these species that allows for forensic information to be obtained. This can be done in several ways, but the only way that I can think of right now is that the concentrations of the nuclear fission products are statistically related to the source nuclear material that is used. Being able to identify the source material allows for better bracktracking to the source. Other reasons for separating out radioactive products include removal from patients within the medical field. If these radioactive species are not removed from the patient, they can be deposited into bones as a calcium replacement, causing further physiological complications.

Okay so lets do a bit of backtracking. We know that we're trying to chelate nuclear fission products, but is there a way to know what these products are first? The answer is yes. If we take a look at the NucDat3 interactive isotope table, it provides data for the Independent Fission Yield of 3 isotopes: Uranium 235, Plutonium 239, and Californium 252. Why are these important? With the exception of Californium 252, these isotopes are used as fissile material for nuclear reactors as well as nuclear and thermonuclear warheads. While it is interesting to consider the pathways to obtaining these isotopes, most lead to the production of U235 or Pu239.



Current separation procedures for nuclear fission products include resin based separations, and others that are being used within the medical field.

What makes chelating resins a good choice for nuclear fission products?

What can we do with nuclear fission products once they are separated?



What does it mean to make preorganized chelators?

Why does preorganizing these chelators make them better? What properties and components are involved? How can we quantify these properties in terms of computational results? What tools can we use to show the importance of certain properties when it comes to creating coordination complexes? How has this been done before? What do we have available now to better characterize and improve these properties that those who came before didn't have?

What are some of the nuances that you need to take into consideration when it comes to developing these preorganized chelators? Huh?

## Application: What has your work involved?

Starting functional group: 1,2-HOPO (1,2-Hydroxypyridinone), part of a class of hydroximates, which are commonly used by siderophores to chelate and ingest Fe from the surroundings. Basically, we copied nature a bit and are trying to make it better for our use case.

Crystal Structure Database: Finding examples of compounds with HOPO chelating them, if they've been made then that's good! We can use them as precursors for possible new chelators.

HostDesigner: using the crystal structure data, we can generate a bunch of different compounds that replace the carbon linkers in the system, thus increasing our pool of possible new chelators.

CMI Model: this model was trained on the IUPAC Stability Constant Database and by using a SMILES string we can generate predicted LogK1 values for the systems. This essentially allows us to see the separation factors between the possible chelators.

Quantum Mechanical Modeling: Based on the insights that can be gained from the CMI Model, we can both verify the results we get from the model, and optimize toward the model by identifying properties that are improving the log k values of the system.

# Current Work

# NWChemEx

# Not Another QM Package! Why!?

Has you or a loved one tried to develop quantum chemistry models only to find that you need to know how to write Fortran, compile it against the original project, and ensure that it works in the end, only to have the entire project not work after removing your new code and left at square 0 with no quantum chemistry package to use?

That's basically a story I have heard from every physical chemist that I have talked to. It is already hard to find and use the tools for computational chemistry. To develop new tools with zero computer science and programming background? Nearly impossible (Ask me how I know).



Therein lies the problem: Every time a chemist needs to

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