# HOW TO DEVELOP MODIFIED EMBEDDED ATOM METHOD POTENTIALS

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### **OUTLINE**

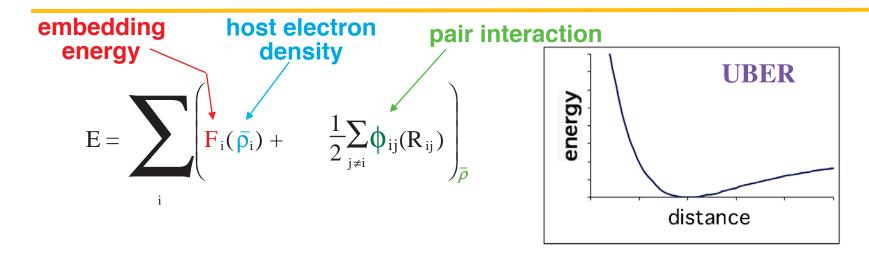
- Modified EAM (MEAM)
- Single Element
- Alloys







## THE EMBEDDED ATOM METHOD IS SEMI-EMPIRICAL



 $\rho$  is obtained from a linear superposition of atomic densities F and  $\varphi$  are obtained by fitting to the following properties: Universal Binding Energy Relationship (UBER)

(lattice constant, bulk modulus, cohesive energy)

Shear moduli

**Vacancy formation energy** 

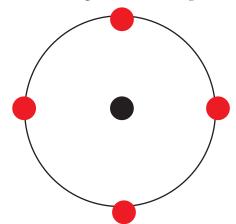
Los Alamos Structural energy differences (hcp/fcc, bcc/fcc) Jacobs

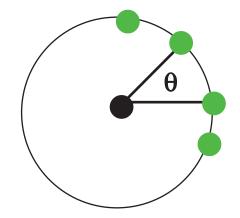




## COMPLEX MATERIALS REQUIRE THE ADDITION OF ANGULAR FORCES

- EAM uses a linear superposition of spherically averaged electron densities
- MEAM allows the background electron density to depend on the local symmetry







$$\rho_k^{(l)^2} = \sum_i \rho_l(R_{ik}) \sum_j \rho_l(R_{kj}) P_l^0 \left( \cos(\theta_{ikj}) \right)$$





# MODIFIED EMBEDDED ATOM METHOD (MEAM)

### Universal Binding Energy Relationship UBER

$$E^{u}(R) = -\frac{E_{c}}{1 + a^{*}} + \delta a^{*3} \frac{r_{e}}{R} e^{-a^{*}}$$

$$a^{*} = \alpha \left(\frac{R}{r_{e}} - 1\right) \qquad \alpha^{2} = \frac{9\Omega B}{E_{c}}$$

#### **Embedding Function**

$$F(\rho) = AE_c \rho \ln \rho$$

#### **Background Electron Density**

$$\rho = \rho^{(0)} \sqrt{1 + \Gamma}$$

$$\Gamma = \sum_{l=1}^{3} t^{(l)} \left( \rho^{(l)} / \rho^{(0)} \right)^{2}$$

$$\rho_{k}^{(l)^{2}} = \sum_{l} \rho_{l}(R_{ik}) \sum_{j} \rho_{l}(R_{kj}) P_{l}^{0} \left( \cos(\theta_{ikj}) \right)$$

$$\rho_{l} = e^{-b^{*}} S$$

$$b^{*=} \beta^{(l)} \left[ \frac{K}{r_{e}} - 1 \right]$$

#### **Pair Potential**

$$\phi(R) = \frac{2}{Z} \left\{ E^{u}(R) - F(\overline{\rho}^{0}(R)) \right\} S$$

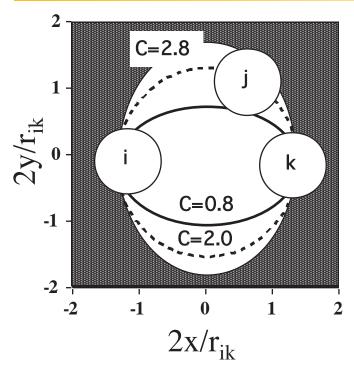


12 parameters + angular screening for the pair potential and electron densities





### **CONCEPT OF THE SCREENING ELLIPSE LEADS TO A SIMPLE SCREENING MODEL**



#### screening ellipse defined by C

$$x^{2} + \frac{1}{C} y^{2} = \left(\frac{1}{2} r_{ik}\right)^{2}$$

$$X_{ij} = \left(\frac{r_{ij}}{r_{ik}}\right)^{2} \qquad C = \frac{2(X_{ij} + X_{jk}) - (X_{ij} - X_{jk})^{2} - 1}{1 - (X_{ij} - X_{jk})^{2}}$$

### C<sub>min</sub> and C<sub>max</sub> set limits of screening

$$S_{ijk} = \begin{cases} 0 & C \le C_{\min} \\ f(\frac{C - C_{\min}}{C_{\max} - C_{\min}}) & C_{\min} < C < C_{\max} \\ 1 & C \ge C_{\max} \end{cases}$$

$$S_{ik} = \prod_{j \neq i,k} S_{ijk}$$

 $S_{ik} = \prod S_{ijk}$  f(x) goes from 0 to 1 smoothly





### RECIPE FOR SINGLE ELEMENT

- Choose Reference Structure
- Assemble Data Base
- Associate/Fit Parameters with Data Base
- Validate
- Iterate







### CHOOSE REFERENCE STRUCTURE

- Simple Crystal Structure
  - fcc
  - bcc
  - hcp
  - diamond cubic
- Usually Equilibrium Ground State
- Must Have Data
- Relevant to Simulations







### **ASSEMBLE DATA BASE**

- Experiment (reference structure)
  - Cohesive energy
  - Lattice constant
  - Elastic constants
  - Vacancy formation energy
  - Stacking fault energy
  - Thermal expansion
  - Other phases
- First Principles (reference structure)
  - Energy vs. volume
  - Other phase relative energies
  - Transformation path





calibration

validation

# ASSOCIATE/FIT PARAMETERS WITH DATA BASE (I)

#### UEOS

- Cohesive energy (E<sub>c</sub>)
- Lattice constant (r<sub>e</sub>)
- Bulk modulus (α)
- Thermal expansion (δ)
- Partial Electron Density Weights
  - Vacancy formation energy (t<sub>1</sub>)
  - Shear elastic constants (t<sub>2</sub>)
  - Stacking fault energy (t<sub>3</sub>)







# ASSOCIATE/FIT PARAMETERS WITH DATA BASE (II)

- Embedding Energy Strength (A)
  - Energy of other phases
- Atomic Electron Density Decay
  - Energy of other phases ( $\beta_0$ )
  - Surface relaxation (β₁)
  - Shear elastic constants ( $\beta_2$ )
  - $c/a (\beta_3)$
- Angular Screening
  - Energy of other phases
  - Shear elastic constants







### **VALIDATE (I – Reference Phase)**

- Thermal properties
  - Expansion
  - Specific heat
- Surfaces
  - Energy
  - Relaxation
  - Reconstruction
- Point defects
  - Vacancy mobility
  - Interstitial
    - Formation energy
    - Geometry
    - Migration energy







### **VALIDATE (II – Other Phases)**

#### Liquid

- Heat of fusion
- Density
- Melting point
- Solid
  - Relative stability
  - Lattice constants
  - Internal relaxation
  - Elastic constants
  - Transformation paths







### **EXPERIMENTAL DATA BASE FOR FE**

Cohesive Energy (eV)	4.29	
α Lattice Constant (300K) (Å)	2.866	
Bulk Modulus (GPa)	167	
C <sub>44</sub> (GPa) 117		
C' (GPa)	47.5	
$\alpha \rightarrow \gamma$ Transformation Temperature (P=0) (K)	1185	
$\gamma \rightarrow \delta$ TransformationTemperature (P=0) (K)	1657	
α →ε Transformation Pressure (0K) (GPa)	6-15	
Vacancy Formation Energy (eV)	1.7	
Thermal Expansion (500K) µm·m <sup>-1</sup> ·K <sup>-1</sup>	14.4  → UCSD School of Engineering	



### MEAM PARAMETERS ARE CORRELATED WITH PHYSICAL PROPERTIES

Physical Property	<b>MEAM Parameter</b>
Cohesive Energy	E <sub>c</sub>
lpha Lattice Constant	${\sf r_e}$
Bulk Modulus	α
Thermal Expansion (500K)	δ
C <sub>44</sub>	t <sub>2</sub> , β <sub>2</sub>
C'	t <sub>2</sub> , β <sub>2</sub>
$\alpha \rightarrow \gamma$ Transformation Temperature (P=0)	A, $\beta_0$ , $C_{min}$
$\gamma \rightarrow \delta$ TransformationTemperature (P=0)	A, $\beta_0$ , $C_{min}$
$\alpha \rightarrow \epsilon$ Transformation Pressure (0K)	t <sub>3</sub> , β <sub>3</sub>
Vacancy Formation Energy	t <sub>1</sub> , β <sub>1</sub>
SAL LABORATORY EST.1943	<b>₹</b> UCSD <b>School of Engineering</b>



### **MEAM PARAMETERS FOR FE**

E <sub>c</sub> (eV)	4.29	α	5.0729	
r <sub>e</sub> (Å)	2.469	δ	0.3	
A	0.6	$\beta_0$	4.045	
t <sub>1</sub>	-1.6	β <sub>1</sub>	2	
t <sub>2</sub>	12	$\beta_2$	0.8	
<b>t</b> <sub>3</sub>	-0.05	$\beta_3$	1	
C <sub>max</sub>	1.9	C <sub>min</sub>	0.7	







### MEAM REPRODUCES DATA BASE EXTREMELY WELL FOR IRON

Physical Property	Experiment	MEAM
Cohesive Energy (eV)	4.29	4.29
lpha Lattice Constant (300K) (Å)	2.866	2.866
Bulk Modulus (GPa)	167	170
C <sub>44</sub> (GPa)	117	116
C' (GPa)	47.5	49
$\alpha \rightarrow \gamma$ Transformation Temperature (P=0) (K)	1185	1175
$\gamma \rightarrow \delta$ TransformationTemperature (P=0) (K)	1657	1750
$\alpha \rightarrow$ ε Transformation Pressure (0K) (GPa)	6-15	11
Vacancy Formation Energy (eV)	1.7	1.7
Thermal Expansion (500K) µm⋅m <sup>-1</sup> ⋅K <sup>-1</sup>	14.4	14.3
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### RECIPE FOR A BINARY SYSTEM

- Choose Reference Structure
- Assemble Data Base
- Associate/Fit Parameters with Data Base
- Validate
- Iterate







### CHOOSE REFERENCE STRUCTURE

- Simple Crystal Structure
  - Preferably with 1NN of opposite type
  - B1 (rock salt –NaCl)
- Equilibrium Ground State if Simple
- Must Have Data
- Relevant to Simulations
  - stoichiometry
  - Structure
- Derive Analytic Expression for Cross Pair Potential





### **ASSEMBLE DATA BASE**

- Experiment or First Principles (reference structure)
  - Cohesive energy
  - Lattice constant
  - Elastic constants
  - Thermal expansion
- Other Phases
  - Cohesive energy







# ASSOCIATE/FIT PARAMETERS WITH DATA BASE

- UEOS (reference structure)
  - Cohesive energy (E<sub>c</sub>)
  - Lattice constant (r<sub>e</sub>)
  - Bulk modulus (α)
  - Thermal expansion (δ)
- Electron Density Scaling
  - Elastic constants (ρ<sup>a</sup><sub>0</sub>)
  - Other cohesive energies (ρ<sup>a</sup><sub>0</sub>, C<sub>min</sub>, C<sub>max</sub>)







### **ELASTIC CONSTANT RELATIONSHIPS**

From continuum anisotropic elasticity theory we know

$$U == \frac{1}{2} \int C_{ijkl} \varepsilon_{ij} \varepsilon_{kl} dV = \frac{1}{2} C_{ijkl} \varepsilon_{ij} \varepsilon_{kl} V$$

$$u = \frac{U}{N} = \frac{1}{2} C_{ijkl} \varepsilon_{ij} \varepsilon_{kl} \frac{V}{N} = \frac{1}{2} C_{ijkl} \varepsilon_{ij} \varepsilon_{kl} \Omega \Rightarrow C_{ijkl} = \frac{\partial^{2} u}{\partial \varepsilon_{ij} \partial \varepsilon_{kl}} \frac{2}{\Omega}$$

$$\sigma_{ij} = C_{ijkl} \varepsilon_{kl} \Rightarrow C_{ijkl} = \frac{\partial \sigma_{ij}}{\partial \varepsilon_{kl}}$$







### HOW TO CALCULATE ELASTIC CONSTANTS

- Consider these states of strain (ε=0.001) applied to the equilibrium structure
  - for B  $\varepsilon_x = \varepsilon_y = \varepsilon_z = \varepsilon$ ,  $\varepsilon_{xy} = \varepsilon_{yz} = \varepsilon_{yz} = 0$ , e.g., scale=1± $\varepsilon$ , 1± $\varepsilon$
  - for C'  $\varepsilon_x = -\varepsilon_v = \varepsilon_z = \varepsilon_{xv} = \varepsilon_{xz} = \varepsilon_{vz} = 0$ , e.g., scale=1± $\varepsilon_x$ 1-(± $\varepsilon_x$ ),1
  - for  $C_{44} \epsilon_{xz} = \epsilon_{z} \epsilon_{x} = \epsilon_{z} = \epsilon_{zz} = \epsilon_{zz} = 0$ , e.g., shear=± $\epsilon_{zz}$
- We can calculate the elastic constants in each state

$$B = \frac{1}{3} \left( C_{11} + C_{12} \right) = \frac{1}{9\Omega} \frac{\partial^2 u}{\partial \varepsilon^2} = \frac{\sigma_{xx}}{3\varepsilon}$$

$$C' = \frac{1}{2} \left( C_{11} - C_{12} \right) = \frac{1}{4\Omega} \frac{\partial^2 u}{\partial \varepsilon^2} = \frac{\sigma_{xx}}{2\varepsilon}$$

$$C_{\text{LOS Alamos}} C_{44} = \frac{1}{4\Omega} \frac{\partial^2 u}{\partial \varepsilon^2} = \left| \frac{\sigma_{xz}}{2\varepsilon} \right|$$





### HOW TO CALCULATE ELASTIC CONSTANTS

#### Energy method

Calculate u" numerically using potential energies from eam88

$$\frac{\partial^2 u}{\partial \varepsilon^2} = \frac{u(1+\varepsilon) + u(1-\varepsilon) - 2u(1)}{\varepsilon^2}$$

- Stress method
  - The stress tensor  $\sigma$  may be obtained from d the stress tensor in eam88

$$\sigma = \frac{\dot{\Phi}}{N\Omega}$$
 total volume of computational cell

### **VALIDATE (I – Reference Phase)**

- Thermal properties
  - Expansion
  - Specific heat
- Surfaces
  - Energy
  - Relaxation
  - Reconstruction
  - Segregation







### **VALIDATE (II –Reference Phase)**

- Point defects
  - Vacancy (on A and B sub-lattices) mobility
  - Interstitial
    - Formation energy
    - Geometry
    - Migration energy







### **VALIDATE (III – Other Phases)**

#### Liquid

- Heat of fusion
- Density
- Melting point

#### Solid

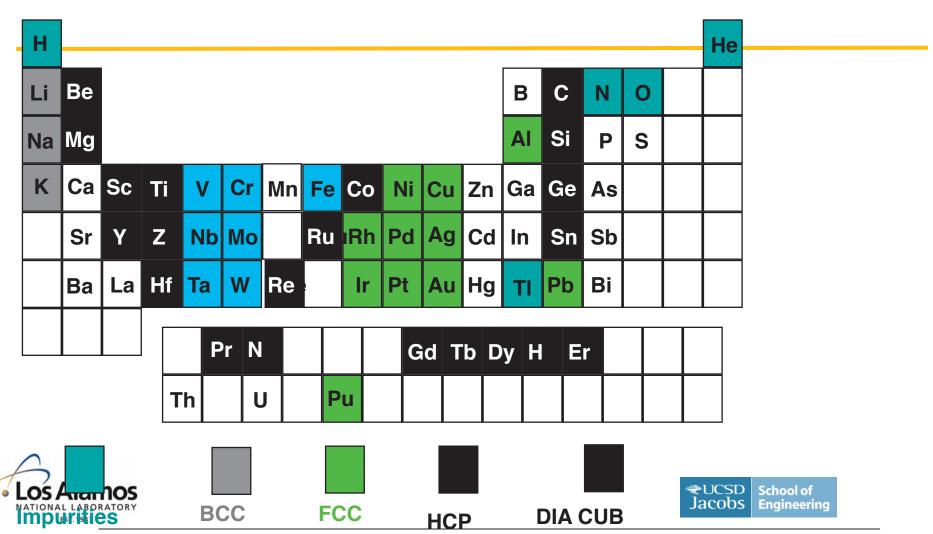
- Relative stability
- Lattice constants
- Internal relaxation
- Elastic constants
- Transformation paths
- Dilute heats of solution
- Short range order

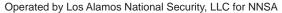






### CURRENTLY DEVELOPED MEAM FUNCTIONS COVER MOST OF THE PERIODIC TABLE







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# ISSUES FOR MULTI-COMPONENT SYSTEMS

- There are a limited number of free parameters for multi-components
  - For AB choose ρ<sup>aB</sup><sub>0</sub>
  - For AC choose ρ<sup>aC</sup><sub>0</sub>
  - There is no ρ to choose for BC
- There are C<sub>min</sub>(A,B,C) and C<sub>max</sub>(A,B,C) to choose
  - All combinations of A, B, C







### **TAKE AWAY**

- MEAM is Parameterized to Facilitate Relatively Easy Function Determination
- MEAM Can Quantitatively Reproduce the Fe Data Base Including Phase Transformations





