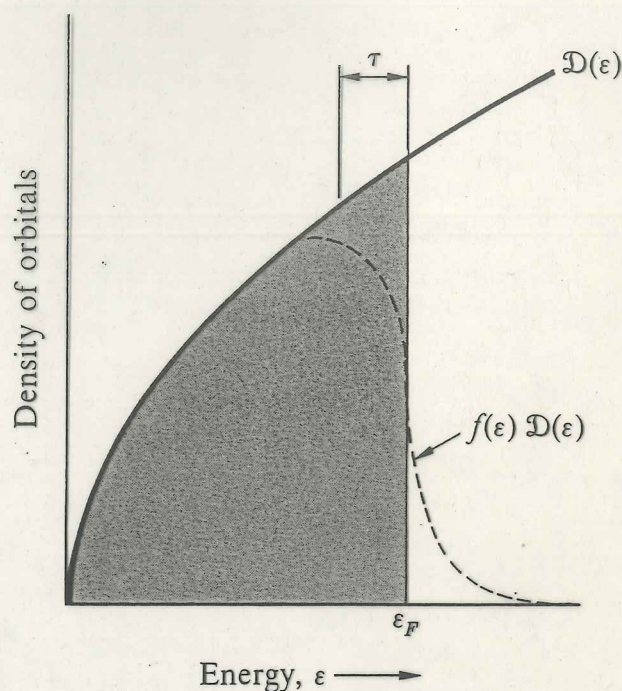


Figure 7.3 Density of orbitals as a function of energy, for a free electron gas in three dimensions. The dashed curve represents the density $f(\epsilon)\mathcal{D}(\epsilon)$ of occupied orbitals at a finite temperature, but such that τ is small in comparison with ϵ_F . The shaded area represents the occupied orbitals at absolute zero.



When multiplied by the distribution function (Figure 6.3), the density of orbitals $\mathcal{D}(\epsilon)$ becomes $\mathcal{D}(\epsilon)f(\epsilon)$, the density of occupied orbitals (Figure 7.3). The total number of electrons in a system may now be written as

$$N = \int_0^{\infty} d\epsilon \mathcal{D}(\epsilon) f(\epsilon, \tau, \mu), \quad (20)$$

where $f(\epsilon)$ is the Fermi-Dirac distribution function described in Chapter 6. In problems where we know the total number of particles, we determine μ by requiring that the total number of particles calculated from (20) be equal to the correct value. The total kinetic energy of the electrons is

$$U = \int_0^{\infty} d\epsilon \epsilon \mathcal{D}(\epsilon) f(\epsilon, \tau, \mu). \quad (21)$$

If the system is in the ground state, all orbitals are filled up to the energy ϵ_F , above which they are vacant. The number of electrons is equal to

$$N = \int_0^{\epsilon_F} d\epsilon \mathcal{D}(\epsilon), \quad (22)$$

and the energy is

$$U_0 = \int_0^{\epsilon_F} d\epsilon \epsilon \mathcal{D}(\epsilon). \quad (23)$$