

Molecular Dynamics Driven Screening of Wettability for Passive Cooling

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1 Introduction

Project Context

Electricity use for space-conditioning is expected to triple by 2050^[1]

⚡ Near-Zero-Carbon, All-Climate-Adaptive Air-Conditioning (NZC-AC-AC) Project with the University of Bath and the University of Hull.

- **Goal:** Replace vapour-compression with capillary-driven evaporative cooling
- **Impact:** 80-90 % energy bill savings, near-zero carbon emissions
- **Scope:** **Molecular Dynamics (MD)** Screening stage of Phase 1

Need: Fast method to assess wettability of materials

NZC-AC-AC Phase 1 of 5

Candidate Materials

MD-screening & Simulation

Refinement

Composition & Combination Methods Development

Characterisation & Selection

Experimental contact-angle (θ) tests are slow and error-prone (surface heterogeneity, operator bias)^[2]

Solution: Fast, Reproducible MD-Based Screening

2 Aims & Objectives

- Develop a foundation for a reusable, high-throughput MD framework for rapid contact-angle prediction
- Validate a water model with VLE to be used in the MD pipeline
- Assess how surface-atom density, wall chemistry, temperature, and droplet size jointly influence wettability on **implicit** walls.
- Validate MD-predicted qualitative trends against established trends and experimental data.
- Setup diffusion and explicit wall simulations for future work

4 Results

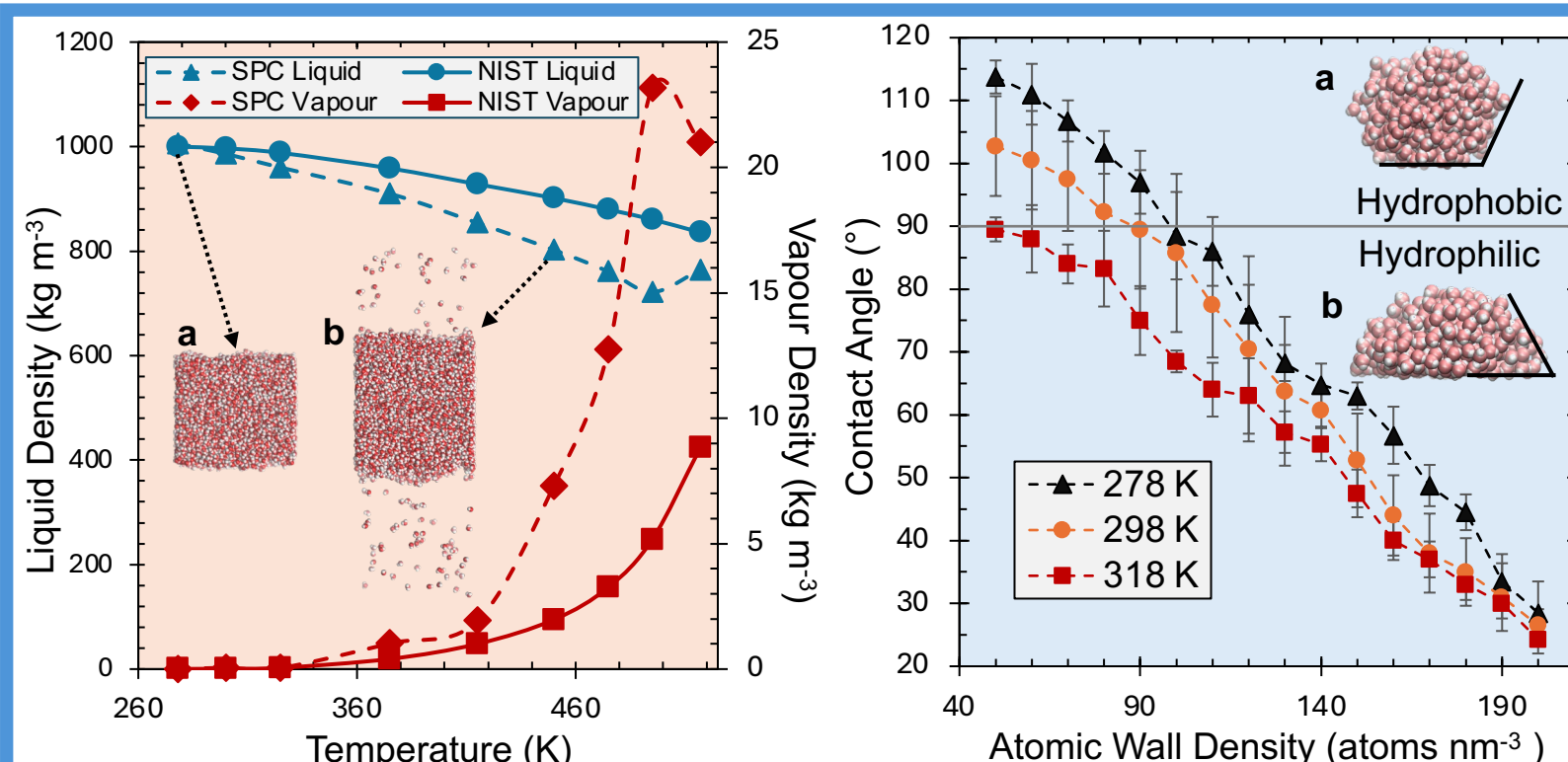


Figure 2. VLE curves 278-517 K for SPC water model compared against NIST data. Inset a) SPC at 278 K, Inset b) SPC at 450 K.

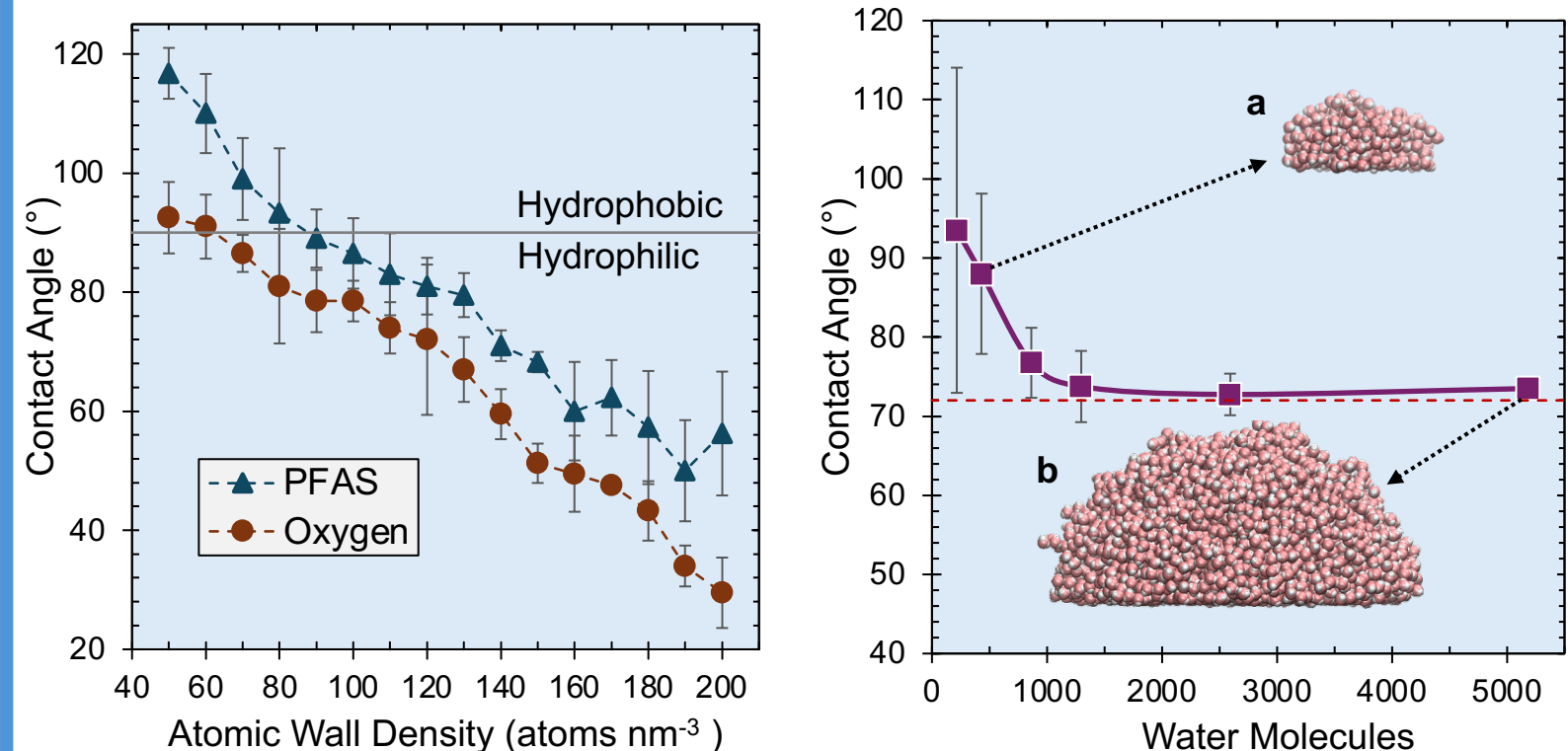


Figure 3. Contact angle θ vs. atomic wall density at 278 K (▲), 298 K (●), and 318 K (■). a) Hydrophobic droplet b) Hydrophilic droplet

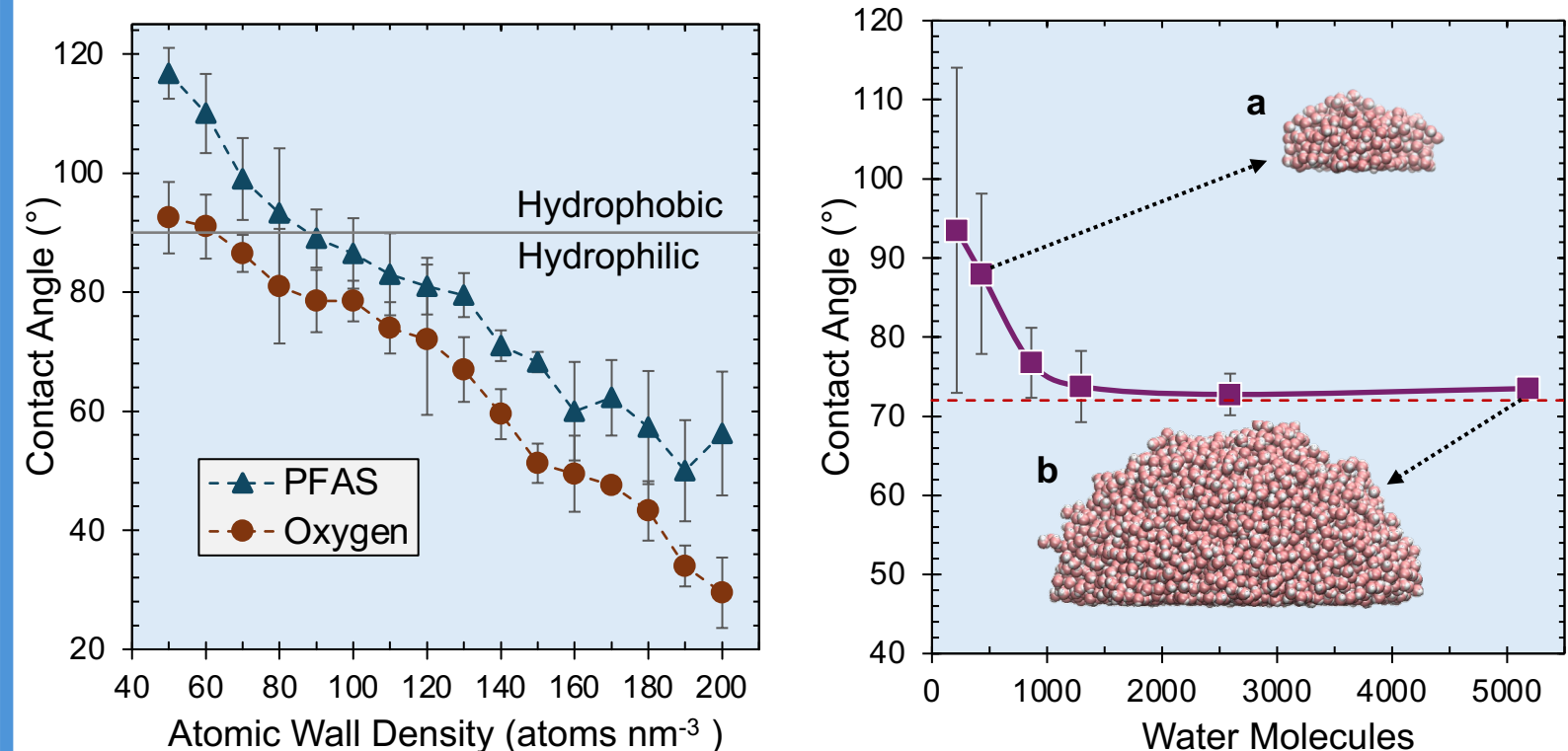


Figure 4. Contact angle θ vs. atomic wall density for PFAS (▲), Oxygen (●) like surfaces.

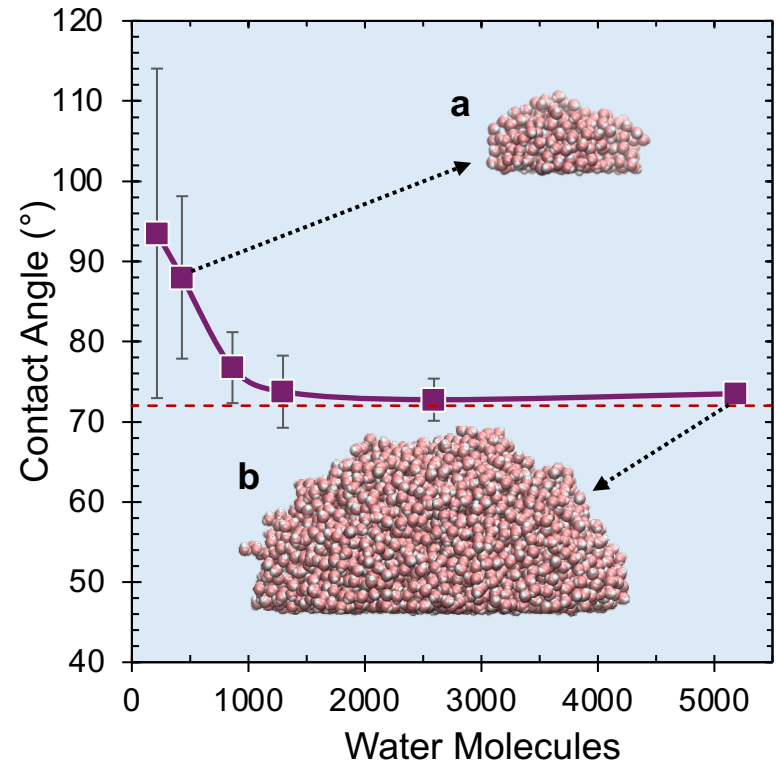


Figure 5. Contact angle θ vs. droplet size (216 – 5186 molecules). a) 416 molecules, b) 5186 molecules

6 Future Work

- Conduct humid-air simulations (Figure 6) to calculate multicomponent diffusion coefficients.
- Incorporate explicit atomistic walls (Figure 7) with mixed chemistries to validate and obtain quantitative predictions
- Couple atomistic insights with pore-scale performance metrics to guide sustainable cooling bed design.

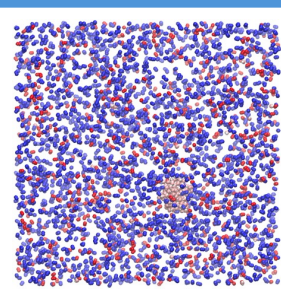


Figure 6. 63% RH, 293 K, 1 bar 18x18x18 nm simulation box with water oxygen and nitrogen.

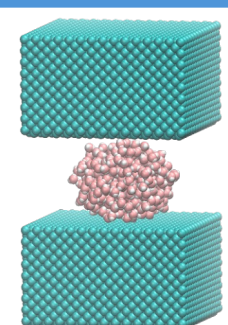
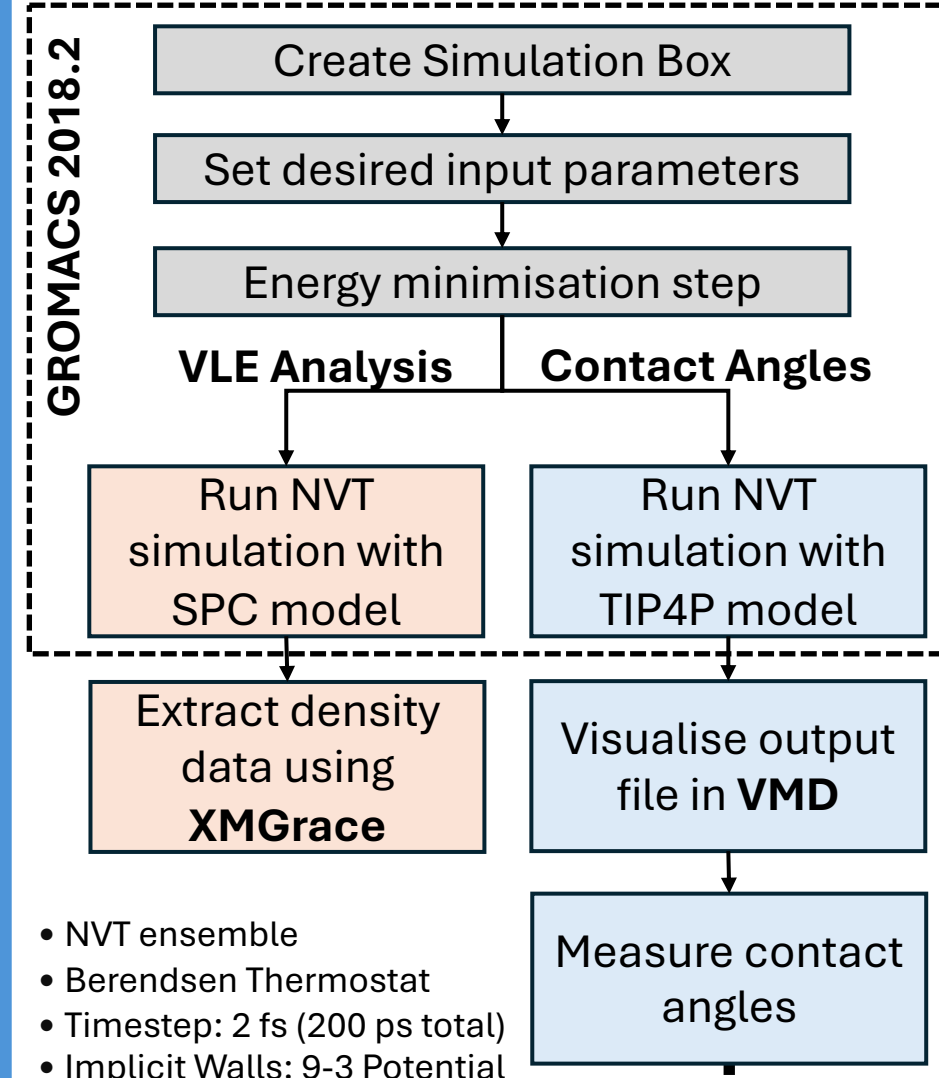


Figure 7. 5 x 5 x 5 nm simulation box of water in a pore with explicit carbon walls.

3 Methods

Simulation Workflow



- NVT ensemble
- Berendsen Thermostat
- Timestep: 2 fs (200 ps total)
- Implicit Walls: 9-3 Potential

VLE simulations: 278 – 517 K

Contact Angle Simulations:

- **Temperature:** 50 – 200 atoms nm⁻³ at 278, 298, and 318 K (Carbon wall)
- **Atom Type:** 50 – 200 atoms nm⁻³ with PFAS and Oxygen wall (at 298 K)
- **Droplet size:** 216 – 5186 molecules at 298 K (Carbon wall).

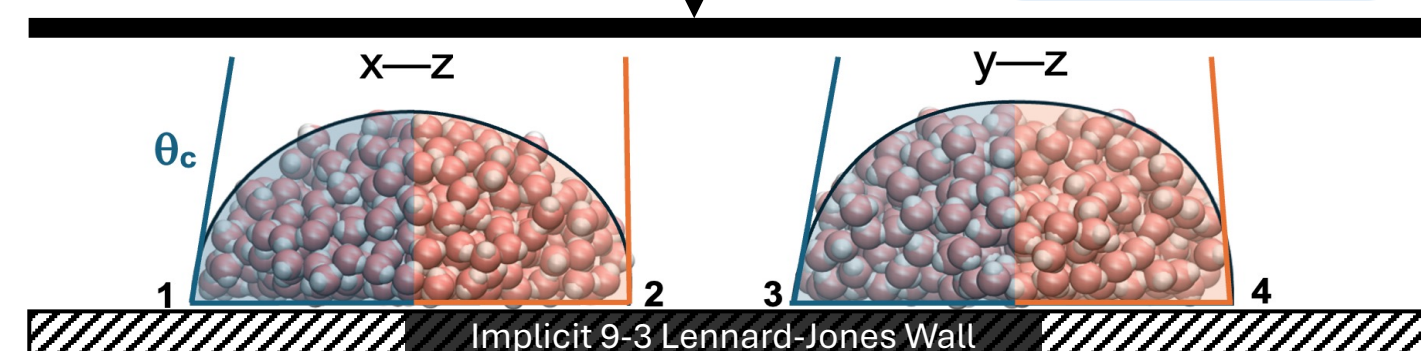


Figure 1. Four angles measured per droplet, two from each plane.

5 Conclusions

- **TIP4P water model** with implicit walls reproduce ρ_s - θ trends with $R^2 > 0.97$. The qualitative trends align with experimental literature for similar hydrophobic^[3] and hydrophilic surfaces^[4].
- **Structural lever** (ρ_s): $\Delta\theta/\Delta\rho_s = 0.45$ – 0.58° per atom-nm⁻³ (0.45° at 278 K, rising to 0.58° at 318 K), indicating greater sensitivity at lower temperatures.
- **Chemistry effect:** At identical densities, PFAS-like C remains $\sim 14^\circ \pm 5.2^\circ$ more hydrophobic than O walls
- **Size scaling:** Above ~ 6 nm droplets (2 000 H₂O), $\Delta\theta$ with size is $< \pm 3^\circ$. Convergence wasn't expected at such few molecules
- Initial **diffusion** and **explicit** wall simulation boxes were setup to help guide the future work.
- Importantly, we show that this MD framework can guide early material selection with fast simulations before employing more computation on explicit wall simulations.

Quantitative θ should be validated with explicit-wall MD or experiment; the trends in this study are robust for triage purposes.

Acknowledgements

We gratefully acknowledge the Department of Chemical Engineering providing resources to access HPC computing. I would also like to give thanks to Dr. Carmelo Herdes for his excellent supervision.

References

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3. T. Werder et al., J. Phys. Chem. B 107, 1345 (2003).
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