# **Complete Electroplating Process Guide for 3D Printed Parts**

# **Project Overview**

This guide details the process for copper electroplating 3D printed parts to enhance their mechanical properties. The process is designed for parts with varying geometric complexity, with no part exceeding  $2\times2\times2$  inches<sup>3</sup> in volume (ensuring plating time  $\leq$  30 minutes).

# **Safety Notice**

△ CRITICAL: Steps 3, 4, and 5 MUST be performed in the University Teaching Lab (UTL) with full PPE and supervision.

# **Step 1: 3D Print and Surface Preparation**

### **Required Materials:**

- 3D printed part (PLA, ABS, or PETG recommended)
- Sandpaper set (400, 800, 1200, 1600, 2000 grit)
- Steel wool (#0000 grade)
- Isopropyl alcohol (IPA) 99%
- Clean, lint-free cloths
- Compressed air (optional)

#### **Detailed Process:**

#### 1a. Print Model

- Print Settings:
  - Layer height: 0.1-0.2mm for smoother finish
  - Infill: minimum 30% for structural integrity
  - Support material: remove completely before proceeding

## Post-print checks:

- Ensure no warping or layer separation
- Verify dimensions are within 2×2×2 inches³
- Allow part to fully cool and stabilize

#### 1b. Progressive Sanding (Total time: 30-45 minutes)

### 1. 400 Grit (10 minutes)

- Purpose: Remove visible layer lines
- Technique: Circular motions with moderate pressure
- Focus on: High points and rough areas
- Check progress frequently under bright light

## 2. **800 Grit (10 minutes)**

- Purpose: Smooth scratches from 400 grit
- Technique: Figure-8 pattern
- Pressure: Medium-light
- Surface should start feeling smooth

### 3. 1200 Grit (10 minutes)

- Purpose: Further refinement
- Technique: Random orbital pattern
- Pressure: Light
- Surface should be uniformly smooth

### 4. 1600 Grit (5-10 minutes)

- Purpose: Pre-polish preparation
- Technique: Very light circular motions
- Check for any remaining imperfections

# 5. **2000 Grit (5-10 minutes)**

- Purpose: Final smoothing
- Technique: Gentle polishing motions
- Result: Satin-smooth finish

# 1c. Polish (Optional but recommended)

- Use #0000 steel wool
- Apply in one direction for uniform finish
- Duration: 5-10 minutes
- Remove all steel wool particles with compressed air

### 1d. Clean

### 1. Initial cleaning:

- Blow off all dust with compressed air
- Pay attention to crevices and complex geometries

#### 2. IPA cleaning:

- Soak lint-free cloth with 99% IPA
- Wipe entire surface thoroughly
- Use fresh cloth sections for final wipe

### 3. Drying:

- Air dry for 15-20 minutes
- Ensure no IPA residue remains
- Part must be completely dry before next step

# **Step 2: Apply Conductive Layer**

## **Required Materials:**

- Conductive copper spray paint
- Masking tape (if selective plating desired)
- Spray booth or well-ventilated area
- Digital multimeter
- Wire or holder for suspending part
- Disposable gloves

#### **Detailed Process:**

### 2a. Apply Copper Spray Paint

#### **Preparation:**

- 1. Set up in well-ventilated area or spray booth
- 2. Temperature should be 65-75°F (18-24°C)
- 3. Humidity below 50% for best results
- 4. Shake spray can vigorously for 2 full minutes
- 5. Test spray on scrap material

#### **Application Technique:**

1. First Coat:

- Distance: 8-10 inches from part
- Motion: Smooth, sweeping passes
- Overlap: 50% between passes
- Coverage goal: 60-70% opacity
- Dry time: 10 minutes

#### 2. Second Coat:

- Apply perpendicular to first coat
- Same distance and technique
- Coverage goal: 90-95% opacity
- Dry time: 15 minutes

### 3. Third Coat (if needed):

- Light application only
- Target any thin areas
- Final coverage: 100% opacity
- Dry time: 20 minutes

### 2b. Allow to Dry

- Total drying time: 30-45 minutes minimum
- Test dryness: Light touch with gloved finger
- Surface should be completely tack-free
- Avoid handling during this period

#### **2c. Test Conductivity**

#### 1. Multimeter setup:

- Set to resistance mode (Ω)
- Use lowest range (200Ω)

#### 2. Testing procedure:

- Place probes 1 inch apart
- Test at least 5 different locations
- Record readings

#### 3. Acceptable values:

<50Ω between any two points: Excellent</li>

- 50-100Ω: Acceptable
- 100Ω: Apply additional coat

# **Step 3: Chemical Cleaning Process (UTL REQUIRED - Full PPE)**

## **Required Safety Equipment:**

- Chemical-resistant goggles (not safety glasses)
- Nitrile gloves (double layer recommended)
- Lab coat or chemical apron
- Closed-toe shoes
- Fume hood access

## **Required Materials:**

- Sodium hydroxide (NaOH) 50g
- Sodium carbonate (Na₂CO₃) 50g
- Hydrochloric acid (HCl) 35% 100ml
- Distilled water 3 liters
- Glass beakers (1L) 3
- Glass stirring rods 3
- Plastic tongs (acid-resistant)
- Timer
- pH test strips

# **Solution Preparation:**

### **Prepare in Fume Hood:**

### 5% NaOH Solution (Degreasing)

- 1. Measure 800ml distilled water in 1L beaker
- 2. Slowly add 50g NaOH pellets while stirring
- 3. **CAUTION:** Solution will heat to ~60°C
- 4. Stir until completely dissolved
- 5. Add distilled water to reach 1L mark
- 6. Cool to room temperature before use

7. Label: "5% NaOH - CAUSTIC"

#### 5% Na<sub>2</sub>CO<sub>3</sub> Solution (Neutralizing)

- 1. Measure 800ml distilled water in 1L beaker
- 2. Add 50g Na<sub>2</sub>CO<sub>3</sub> while stirring
- 3. Continue stirring until dissolved
- 4. Add distilled water to reach 1L mark
- 5. Label: "5% Na<sub>2</sub>CO<sub>3</sub> Neutralizer"

#### 10% HCl Solution (Activation)

- 1. **CRITICAL:** Always add acid to water
- 2. Measure 700ml distilled water in 1L beaker
- 3. SLOWLY add 100ml concentrated HCl
- 4. Stir gently with glass rod
- 5. Add distilled water to reach 1L mark
- 6. Label: "10% HCI ACID"

## **Cleaning Process:**

#### 3a. Soak in 5% NaOH Solution

- 1. Using tongs, fully submerge part
- 2. Duration: 5 minutes
- 3. Agitate gently every minute
- 4. Purpose: Removes organic contaminants and oils

#### 3b. Rinse in 5% Na₂CO₃ Solution

- 1. Remove from NaOH with tongs
- 2. Rinse in running distilled water for 30 seconds
- 3. Transfer to Na<sub>2</sub>CO<sub>3</sub> solution
- 4. Soak for 2 minutes
- 5. Purpose: Neutralizes residual base

### 3c. Dip in 10% HCl Solution (Brief)

1. Remove from Na<sub>2</sub>CO<sub>3</sub> with clean tongs

- 2. Quick rinse in distilled water
- 3. Dip in HCl for exactly 10 seconds
- 4. Purpose: Activates copper surface
- 5. Immediately rinse in distilled water

### **3d. Dry Between Steps**

- Use compressed air (preferred) or lint-free cloth
- Ensure complete drying between solutions
- Work quickly after HCl dip to prevent oxidation
- Part should proceed to plating within 5 minutes

# **Step 4: Setup Electroplating Bath (UTL REQUIRED - Full PPE)**

## **Required Equipment:**

- PEAK Tech 30A DC Power Supply
- Glass or plastic container (2L capacity)
- Copper anodes (4 oz total)
- Anode bags
- Copper wire (16 AWG)
- Alligator clips (6)
- Hot plate with magnetic stirrer
- Digital thermometer
- Glass stirring rod

# **Required Chemicals:**

- Copper sulfate pentahydrate (CuSO₄·5H₂O) 250g
- Sulfuric acid (H₂SO₄) battery acid 50ml
- Distilled water 1L

## **Bath Setup Process:**

### 4a. Prepare Container

- 1. Clean container with distilled water
- 2. Ensure no contamination

- 3. Place on magnetic stirrer/hot plate
- 4. Insert clean stir bar

### 4b. Prepare Electrolyte Solution

## **Copper Sulfate Bath Recipe:**

#### 1. Heat water:

- Add 800ml distilled water to container
- Heat to 50-60°C with stirring

### 2. Dissolve copper sulfate:

- Slowly add 250g CuSO<sub>4</sub>·5H<sub>2</sub>O while stirring
- Maintain temperature at 50-60°C
- Stir until completely dissolved (10-15 minutes)
- Solution will be bright blue

#### 3. Cool and add acid:

- Cool solution to 40°C
- In fume hood, slowly add 50ml H₂SO₄
- NEVER add water to acid
- Stir gently for 5 minutes

#### 4. Final adjustment:

- Add distilled water to reach 1L total
- Mix thoroughly
- Cool to room temperature (20-25°C)

#### **Final Bath Parameters:**

- Copper sulfate: 200-250 g/L
- Sulfuric acid: 50-75 g/L
- pH: 0.5-1.0 (verify with pH strips)
- Temperature: 20-25°C for operation

#### 4c. Place Anodes

#### 1. Anode preparation:

• Use 2-4 copper anodes (depending on part size)

- Place each anode in an anode bag
- Bags prevent copper particles from contaminating bath

#### 2. Positioning:

- Space anodes evenly around container
- Anodes should be 3-4 inches from center
- Submerge anodes 80% into solution
- Leave top portion dry for connections

### 4d. Suspend 3D Part

#### 1. Create hanging system:

- Attach copper wire to conductive paint area
- Use minimal contact point
- Create hook or cradle to suspend part

## 2. Positioning requirements:

- Center part in bath
- Minimum 2 inches from nearest anode
- Part fully submerged except wire connection
- No contact with container bottom or sides

#### 4e. Ensure Electrical Connections

#### 1. Anode connections (Positive):

- Connect all anodes together with copper wire
- Use alligator clips for secure connection
- Connect to positive (+) red terminal of PSU

#### 2. Cathode connection (Negative):

- Connect part wire to negative (-) black terminal
- Ensure solid connection at conductive paint

#### 3. Verification:

- Double-check polarity (anodes positive, part negative)
- Ensure all connections are secure
- No exposed wire in solution except at connection points

# **Step 5: Electroplating Process (UTL REQUIRED - Full PPE)**

## **Plating Parameters:**

• Current density: 20-30 ASF (Amps per Square Foot)

• Recommended: 25 ASF for optimal results

Voltage: 1-3V (will auto-adjust on constant current)

• Bath temperature: 20-25°C

· Agitation: Gentle magnetic stirring

## **Pre-Plating Calculations:**

### 5a. Calculate Required Current

#### 1. Measure surface area:

- Calculate total surface area in square inches
- For complex shapes, estimate or use CAD data

### 2. Convert to square feet:

• Area (ft²) = Area (in²) ÷ 144

#### 3. Calculate current:

- Current (A) = Area (ft²) × 25 ASF
- Example:  $6 \text{ in}^2 = 0.042 \text{ ft}^2 \times 25 = 1.05 \text{A}$

# **5b. Calculate Plating Time**

#### For desired thickness:

- Use online calculator (link in original procedure)
- Or use formula:

```
Time (min) = (Thickness (inches) \times 2.95 \times Area (in<sup>2</sup>)) / Current (A)
```

• For 0.001" (1 mil) on 6 in<sup>2</sup> at 1A  $\approx$  18 minutes

# **Plating Process:**

#### 5c. Begin Plating

#### 1. Final checks:

- Part is clean and dry
- All connections secure

• Bath temperature 20-25°C

#### 2. Start sequence:

- Turn on magnetic stirrer (low speed)
- Set PSU to constant current mode
- Set current limit to calculated value
- Turn on PSU with current at zero

## 3. **Ramp up:**

- Slowly increase current over 30 seconds
- Watch voltage (should stabilize at 1-3V)
- Look for uniform bubble formation

## **5d. Monitor During Plating**

### 1. Visual checks (every 5 minutes):

- Even bubble distribution
- No dark or burnt areas
- Uniform color development

## 2. Parameter monitoring:

- Current remains stable
- Voltage within 1-3V range
- Temperature stays 20-25°C

#### 3. Adjustments if needed:

- Reduce current if burning occurs
- Rotate part for even coverage
- Add distilled water if evaporation occurs

#### **5e. Complete Plating**

#### 1. Shutdown sequence:

- Reduce current to zero over 30 seconds
- Turn off PSU
- Turn off stirrer

#### 2. Part removal:

· Lift part slowly from bath

- Allow excess to drip back
- Immediately rinse in distilled water
- Rinse for minimum 1 minute

# **Troubleshooting Guide:**

Problem	Cause	Solution
Dull/matte finish	Low acid concentration	Add 10ml H₂SO₄ per liter
Dark/burnt areas	Current too high	Reduce by 20%
Poor adhesion	Inadequate surface prep	Restart from Step 1
Rough/nodular	Contaminated bath	Filter bath, clean anodes
Uneven thickness	Poor current distribution	Reposition anodes

# **Step 6: Part Finishing**

# **Required Materials:**

- Sandpaper (400-2000 grit)
- Steel wool (#0000)
- Metal polish (optional)
- Clean cloths
- Clear lacquer (optional)

# **Finishing Process:**

## 6a. Initial Sanding (if needed)

# 1. Identify imperfections:

- High spots
- Rough areas
- Nodules

## 2. Progressive sanding:

- Start with 400 grit for major issues
- Progress through grits as in Step 1
- Be gentle copper layer is 0.001-0.002" thick
- Focus on problem areas only

### 6b. Polish (Optional)

#### 1. Steel wool polishing:

- Use #0000 steel wool
- Light pressure, circular motions
- · Work small sections at a time

#### 2. Metal polish (for mirror finish):

- · Apply small amount to cloth
- Work in circular motions
- Buff with clean cloth
- · Repeat until desired shine achieved

## **Post-Process Requirements:**

## **Waste Disposal:**

#### 1. Chemical solutions:

- NEVER pour down drain
- Neutralize acids with Na₂CO₃ before disposal
- Follow institutional hazardous waste procedures

#### 2. Copper bath:

- Can be reused 10-20 times
- Store covered and labeled
- Filter before reuse

#### **Documentation:**

- Record all parameters used
- Note any issues encountered
- Photo document results
- Calculate actual thickness achieved

#### **Quality Testing:**

#### 1. Adhesion test:

- Tape test (ASTM D3359)
- No copper should remove with tape

#### 2. Thickness measurement:

- Use micrometer if available
- Compare weight before/after

#### 3. Conductivity:

• Should be <1Ω across surface

# **Safety Reminders**

### 1. Chemical safety:

- Always add acid to water
- Use fume hood for all chemical work
- Know location of safety shower/eyewash

### 2. Electrical safety:

- Never touch electrodes while power is on
- Keep electrical connections dry
- Maximum 30A current for safety

## 3. **PPE requirements:**

- Never work without full PPE in chemical steps
- Replace gloves if contaminated
- Wash hands thoroughly after work

## **Additional Resources**

- SDS sheets for all chemicals (see materials list)
- Online plating calculator (mentioned in original procedure)
- ASTM standards for plating quality
- University chemical waste disposal guidelines