



**Integrating Schrödinger for Comprehensive *in Silico* Study:
Homology Modeling, Virtual Screening, and Molecular Docking
of Substituted Benzofuran Scaffolds Targeting AChE for
Alzheimer's Therapy**

Submitted in Partial Fulfillment of Requirements
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(Chemical Biology and Drug Design)**

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This is to certify that Mr. Joel Thirvayaraj of **M.Sc. in Chemical Biology and Drug Design** has satisfactorily completed the Research Project titled Integrating Schrödinger for Comprehensive *in Silico* Study: Homology Modelling, Virtual Screening, and Molecular Docking of Benzofuran Scaffolds Targeting AChE for Alzheimer's Therapy for the Partial fulfillment of the Degree by the Somaia Vidyavihar University, during the Academic year **2024-25**.

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Examiners

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ABSTRACT

Alzheimer's Disease (AD) is a progressive neurodegenerative disorder marked by cognitive decline, primarily caused by the breakdown of acetylcholine by the enzyme acetylcholinesterase (AChE). This leads to impaired memory and learning functions. Inhibiting AChE is a vital therapeutic approach for mitigating AD symptoms. In this study, a reliable 3D model of the AChE receptor was developed using the Schrödinger suite, compensating for the absence of a complete human AChE crystal structure. Substituted benzofuran derivatives were prepared using LigPrep and screened virtually after validating the receptor model with a Ramachandran plot. The Glide module identified ligands based on binding affinity, with GlideXP docking refining the scores and elucidating key interactions with AChE active site residues, particularly those in the catalytic triad (**Ser324, Tyr103, Asp 105**). Among the compounds evaluated, **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran** emerged as the lead molecule, demonstrating a docking score of **-10.801**, closely approaching that of Donepezil (Alzforum, 2023) (**-11.37**). The lead compound forms critical hydrogen bonds and hydrophobic interactions within the active site, suggesting significant inhibitory activity against AChE. Comparative analysis highlighted the potential of **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran** as a competitive AChE inhibitor. Additionally, benzofuran derivatives are recognized for their ability to inhibit amyloidogenic processes, further supporting their anti-Alzheimer properties. Future molecular dynamics simulations and experimental validations are recommended to assess the stability and efficacy of this compound, with the ultimate goal of developing an alternative treatment for AD.

Keywords: *Schrödinger, Acetylcholinesterase, Benzofuran derivatives, AChE receptor, 2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran*

Introduction and Literature Survey

1) Background and Literature Review- Alzheimer's Disease (AD) is one of the most prevalent neurodegenerative conditions, affecting millions worldwide, predominantly older individuals. The disease is marked by progressive cognitive decline, memory loss, and functional impairment, often leaving patients entirely dependent on caregivers in advanced stages. Its pathological hallmarks include amyloid-beta plaques, neurofibrillary tangles formed by tau protein, and a significant loss of cholinergic neurons (Hampel et al., 2021). These factors collectively lead to the deterioration of cognitive and memory-related processes in the brain, with the deficiency of acetylcholine—a vital neurotransmitter—playing a central role in disease progression. Acetylcholine is crucial for learning and memory, and its levels are regulated by Acetylcholinesterase (AChE), an enzyme that hydrolyzes acetylcholine in the synaptic cleft (Bartus et al., 1982). Excessive AChE activity reduces acetylcholine availability, worsening cognitive deficits in AD patients. Consequently, inhibiting AChE has become a primary therapeutic strategy to alleviate AD symptoms (Birks & Harvey, 2018).

Approved AChE inhibitors (AChEIs) like Donepezil, Rivastigmine, and Galantamine work by preventing acetylcholine breakdown, thereby restoring cholinergic function (Lane et al., 2018). However, these drugs have limitations, such as short half-life, limited bioavailability, and side effects like nausea and diarrhea. This highlights the need to explore alternative compounds with improved efficacy, bioavailability, and fewer adverse effects.

Benzofuran derivatives have gained attention for their diverse biological activities, including antioxidative, anti-inflammatory, and neuroprotective properties (Ali et al., 2017). These compounds, characterized by a fused benzene and furan ring system, are potential AChEIs. Modifying the benzofuran scaffold can optimize its interaction with AChE's active site, targeting critical residues like Ser203, His447, Glu334, Trp86, and Phe338, which are essential for substrate hydrolysis and inhibitor binding (Sussman et al., 1991).

2) Rationale- AD is a progressive and devastating neurodegenerative disorder, characterized by significant cognitive decline, memory loss, and behavioural

impairments. As one of the most prevalent forms of dementia, AD affects millions globally, predominantly within the aging population. One of the central pathological hallmarks of AD is the disruption of cholinergic signalling caused by the enzymatic breakdown of acetylcholine, a neurotransmitter essential for learning and memory, by AChE. Excessive AChE activity leads to acetylcholine deficiency, impairing synaptic communication and accelerating cognitive dysfunction (Soreq & Seidman, 2001). The cholinergic hypothesis of Alzheimer's posits that enhancing cholinergic function—by increasing acetylcholine levels or inhibiting AChE—can alleviate symptoms and improve cognitive performance (Bartus et al., 1982). Based on this hypothesis, AChEIs have emerged as a key therapeutic approach for AD treatment. Clinically approved AChEIs such as Donepezil, Rivastigmine, and Galantamine are widely used for symptomatic management. Among these, Donepezil has gained prominence for its high efficacy and selectivity for AChE (Birks & Harvey, 2018). Despite their clinical success, current AChEIs face significant limitations. These drugs are associated with poor bioavailability, short half-lives, and adverse effects such as nausea, vomiting, and diarrhea, which limit patient compliance (Lane et al., 2018). Furthermore, existing AChEIs do not alter the disease course and only provide symptomatic relief, highlighting the urgent need for novel compounds with improved pharmacokinetic and pharmacodynamic profiles. One promising avenue to address these limitations is the development of alternative chemical scaffolds, such as benzofuran derivatives. Benzofurans exhibit diverse pharmacological properties, including neuroprotective, anti-inflammatory, and AChE inhibitory activities (Ali et al., 2017). Structurally, benzofuran consists of a fused benzene and furan ring system, offering flexibility for modifications to optimize binding interactions with AChE's active site. These modifications can enhance affinity, selectivity, and bioavailability while minimizing toxicity risks. Benzofurans are particularly effective in targeting key catalytic triad residues (Ser203, His447, and Glu334) and peripheral anionic site residues (Trp86, Phe338) critical for AChE inhibition (Silman & Sussman, 2005). However, despite their therapeutic potential, benzofuran-based AChEI remain underexplored. This study addresses this gap by evaluating a series of substituted benzofuran derivatives as

potential AChE inhibitors. Using computational approaches, including homology modeling, molecular docking, and structure-based drug design, the study offers a systematic and cost-effective strategy to identify and optimize lead compounds. These methods enable high-throughput screening of derivatives and provide detailed insights into ligand-receptor interactions. Among the evaluated compounds, 2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran emerged as the lead compound. Identified through Glide XP docking studies, it demonstrated a docking score of -10.801, approaching the score of Donepezil (-11.37) while maintaining favorable drug-like properties. The compound showed selective interactions with catalytic triad residues and peripheral anionic site residues, stabilizing its binding pose within AChE's active site. These results indicate its potential as a competitive AChE inhibitor. By leveraging computational methodologies, this study underscores the potential of benzofuran derivatives as viable alternatives to existing AChEIs. The findings establish a foundation for further experimental validation and structural optimization, aiming to develop safer and more effective treatments for AD.

- 3) Hypothesis-** “Substituted benzofuran derivatives, particularly 2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran, exhibit significant inhibitory activity against AChE by selectively interacting with its catalytic triad and peripheral anionic site residues, making them promising candidates for AD therapy”

AChE is a serine hydrolase responsible for catalysing the hydrolysis of acetylcholine, a neurotransmitter crucial for cognitive processes such as learning and memory. The enzymatic function of AChE is facilitated by a **catalytic triad** composed of **Ser324, Tyr103, and Asp105**, which are situated in the enzyme's active site (Soreq & Seidman, 2001). Beyond its active site, AChE also features a **peripheral anionic site (PAS)**, encompassing residues such as **Tyr103, Asp105** and others, which play pivotal roles in substrate binding and recognition (Silman & Sussman, 2005). Effective inhibition of AChE necessitates ligands that can selectively target both the catalytic triad and PAS residues, thereby stabilizing the enzyme-ligand complex and halting acetylcholine breakdown.

Currently available inhibitors, including **Donepezil**, demonstrate robust interactions with both the catalytic triad and PAS, leading to improved cholinergic function and enhanced cognitive performance in AD patients (Birks & Harvey, 2018). However, these inhibitors are associated with several drawbacks, such as **adverse side effects**, limited bioavailability, and lack of disease-modifying properties. These limitations underscore the need for novel scaffolds that exhibit enhanced pharmacological profiles, improved efficacy, and reduced toxicity, thereby addressing the gaps in current AD therapies.

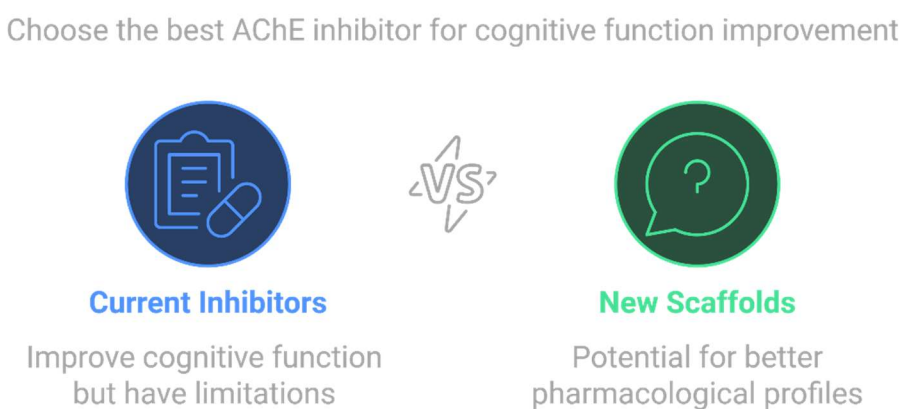


Figure 1- Schematic representation of difference between existing inhibitors and the potential with benzofuran scaffolds. (Napkin AI, 2024)

4) Objectives- This study focuses on evaluating the potential of substituted benzofuran derivatives as AChEIs using computational approaches. Due to the unavailability of a complete crystal structure for human AChE, a reliable **3D model** of the enzyme was constructed using homology modelling with the Schrödinger Suite. The high-resolution crystal structure of recombinant human AChE bound to Donepezil (**PDB ID: 4EY7**) was selected as the template for model development. After creating the receptor model, **LigPrep**, a Schrödinger module for ligand preparation, was used to generate and optimize a library of substituted benzofuran derivatives for docking studies. Model validation was conducted using Ramachandran plot analysis to ensure structural reliability.

Molecular docking was performed using the **Glide module**, incorporating **high-throughput virtual screening (HTVS)** for initial ligand selection and **extra precision (XP)** docking to refine binding poses and scores. This systematic approach enabled the identification of potential AChEIs with favourable binding profiles.

The specific objectives of this research are:

- I. **Modelling the Human AChE Receptor**- Develop and validate a robust **3D model** of the human AChE receptor using homology modelling techniques to facilitate accurate docking studies.
- II. **Virtual Screening of Benzofuran Derivatives**- Evaluate a library of substituted benzofuran derivatives to identify promising AChE inhibitors using **HTVS** docking methods.
- III. **Docking Studies for Lead Identification**- Conduct detailed docking studies, including **XP docking**, to refine ligand binding poses and scores, ultimately identifying the **most effective lead compound**.
- IV. **Comparative Analysis with Donepezil**- Compare the docking scores and interaction profiles of the identified lead compound, **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran**, with Donepezil, a standard AChEI, to assess its therapeutic potential and competitive efficacy.

This study integrates computational tools and structure-based drug design to advance the development of benzofuran-based AChEIs, laying the foundation for novel therapeutic strategies for AD.

Exploring AChE Inhibition Through Diverse Approaches

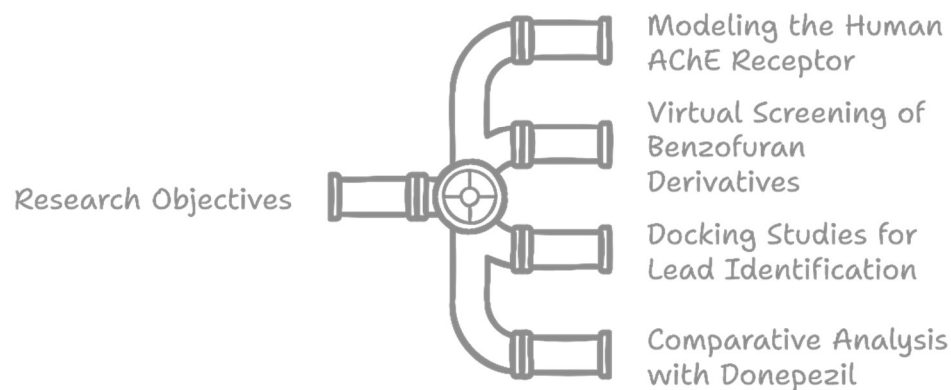


Figure 2- Research objectives. (Napkin AI, 2024)

Through these steps, this research aims to establish, **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran** as a potential therapeutic candidate for AD, potentially offering superior pharmacological properties compared to existing drugs like Donepezil.

Materials and Methodology

1) Materials-

i. **Software-** The computational studies were carried out using the **Schrödinger Suite** (Schrödinger, LLC, 2023), a comprehensive molecular modeling platform. The specific modules utilized include:

- **LigPrep:** Used for ligand preparation, optimizing structures for docking studies by generating the appropriate stereochemistry, ionization states, and tautomeric forms.
- **Glide:** Conducted molecular docking to predict binding affinities and poses of the ligands within the active site of the AChE receptor.
- **Prime:** Performed refinement of protein-ligand complexes and energy minimizations, enhancing docking accuracy and stability of interaction models.

ii. **Databases-** The study employed the following databases to source molecular and protein data:

- **ChEMBL Database:** A source of bioactive compound data, including detailed information on chemical structures and biological activities (Gaulton et al., 2012).
- **UniProt Database:** Provided high-quality protein sequence and functional data for the AChE receptor (UniProt Consortium, 2023).

iii. **Hardware Specifications-** Computational experiments were performed on a workstation with the following specifications:

- **Device Name:** Czeasar
- **Processor:** AMD Ryzen 7 5700U with Radeon Graphics, 1.80 GHz
- **Installed RAM:** 16.0 GB (13.8 GB usable)
- **System Type:** 64-bit operating system, x64-based processor

This setup provided sufficient computational capacity for molecular docking and ligand optimization tasks.

In this study, a 3D model of human AChE was constructed using homology modeling, addressing the absence of a complete crystal structure for the receptor. The **Crystal Structure of Recombinant Human AChE in Complex with Donepezil** (PDB ID: 4EY7) was used as a structural template due to its high-resolution data and relevance to drug discovery. This model served as the basis for screening and evaluating substituted benzofuran derivatives, compounds with promising biological activity and potential as AChE inhibitors.

The workflow encompassed several key steps, including:

- I. **Receptor Preparation:** Homology modeling and validation of the AChE receptor structure.
- II. **Ligand Preparation:** Retrieval, evaluation, and optimization of benzofuran derivatives for drug-likeness and structural integrity.
- III. **Molecular Docking:** Assessment of ligand-receptor interactions using a multi-step docking approach [High Throughput Virtual Screening (HTVS), Standard Precision (SP), and Extra Precision (XP) modes] to rank compounds by binding affinity.
- IV. **Comparative Analysis with Donepezil:** Benchmarking the identified ligands against Donepezil to evaluate their relative efficacy.

This systematic approach integrates computational techniques with established pharmacological principles to identify and characterize potential lead compounds, focusing on their interactions with key catalytic and peripheral residues of AChE. By combining receptor modeling with rigorous docking workflows, the study provides a solid foundation for rational drug design targeting AD.

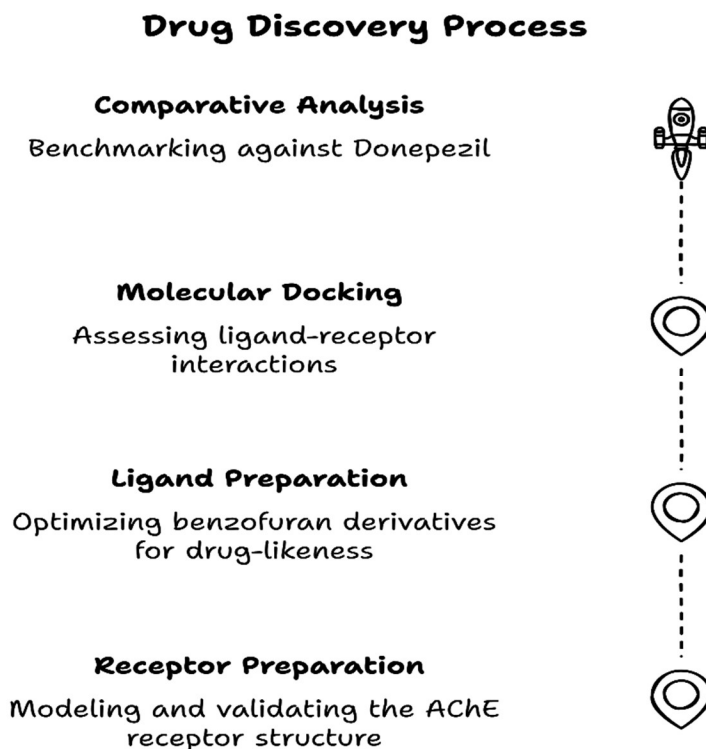


Figure 3- Comparative analysis with donepezil. (Napkin AI, 2024)

2) Experimental Details-

2.1 Receptor Preparation: Homology Modeling- Homology modeling was employed to create a reliable 3D model of human AChE due to the absence of a complete crystal structure for this receptor. This approach is widely recognized for predicting protein structures by leveraging a related protein template with a known structure. The process enables the generation of a receptor model suitable for molecular docking studies, an essential component of computational drug discovery.

2.2 Template Selection- The human AChE amino acid sequence (UniProt ID: P22303) was retrieved from the UniProt database (The UniProt Consortium, 2023). This sequence served as the query for identifying a suitable structural template. The **"Crystal Structure of Recombinant Human AChE in Complex with Donepezil" (PDB ID: 4EY7)** was selected as the template because of its high-resolution data

(2.35 Å) and its relevance to drug discovery, given that Donepezil is a clinically approved AChEI (Shen et al., 2011). This template provided structural insights into ligand-receptor interactions, enabling the precise construction of a human AChE model.

2.3 Model Building- The Schrodinger Suite's Prime module was used to create the 3D receptor model. The process involved:

- I. **Sequence Alignment:** Aligning the AChE amino acid sequence (UniProt ID: P22303) with the template structure (PDB ID: 4EY7) to identify conserved regions and establish a robust framework for the receptor model.

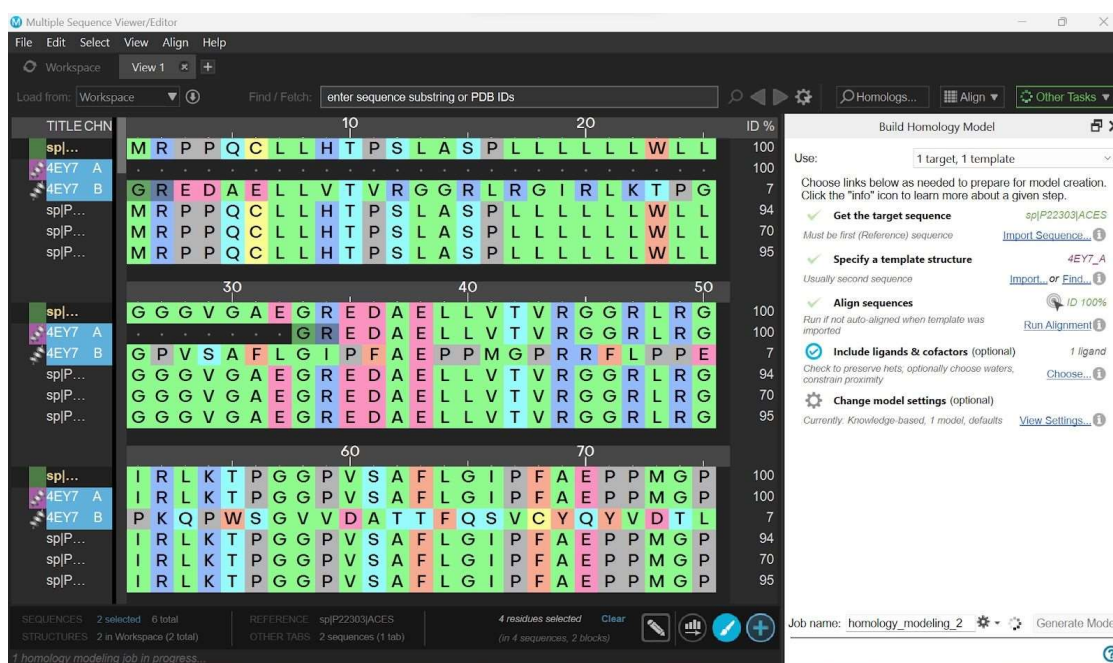


Figure 4- Sequence alignment of aligning the AChE amino acid sequence (UniProt ID: P22303) with the template structure (PDB ID: 4EY7)

- II. **Structure Generation:** Using the aligned sequence to generate a preliminary 3D model. Missing loops and unresolved regions in the template structure were modeled using advanced algorithms in the Prime module.

III. **Model Refinement:** Refining the preliminary structure by optimizing bond angles, torsions, and geometries to ensure structural stability.

2.4 Energy Minimization- Energy minimization was performed to eliminate steric clashes, optimize bond interactions, and ensure that the model was in a stable, low-energy conformation. This step employed the OPLS4 force field, which is recognized for its precision in simulating protein-ligand interactions and molecular dynamics.

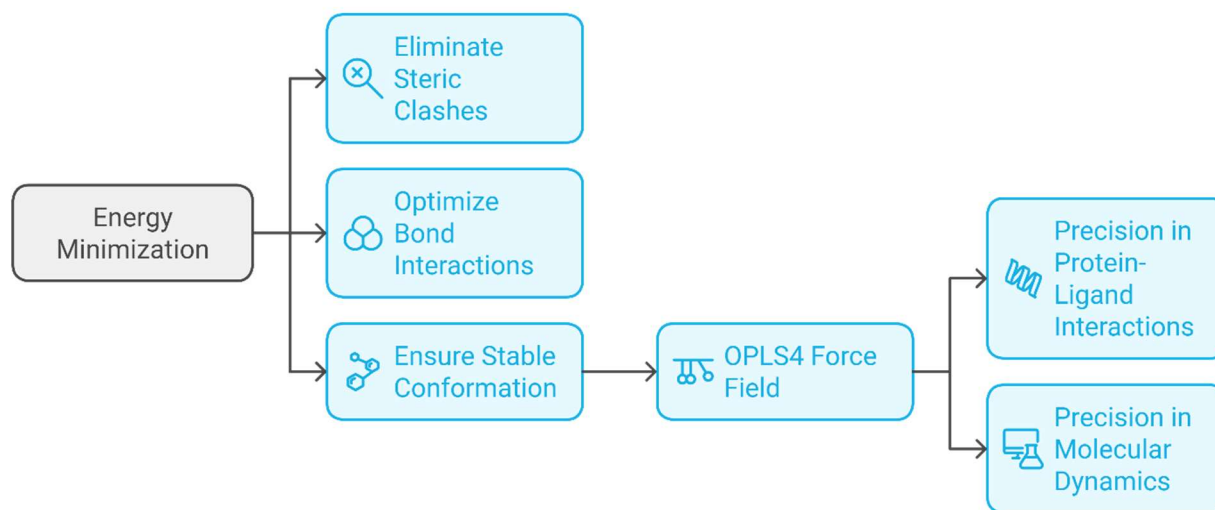


Figure 5- Schematic representation of the steps involve in energy minimization.

(Napkin AI, 2024)

2.5 Model Validation- To validate the structural quality of the receptor model, a Ramachandran plot was generated using tools within the Schrodinger Suite. The plot evaluates the ϕ (phi) and ψ (psi) angles of amino acid residues, identifying favored and allowed regions in the protein backbone (Ramachandran et al., 1963).

3) Ligand Preparation- The study investigated substituted benzofuran derivatives due to their diverse biological activities, including potential as AChEI (Verma et al., 2020). These ligands were prepared systematically to ensure compatibility with the docking protocol

3.1 Data Retrieval- Chemical structures of benzofuran derivatives were sourced from the ChEMBL database, a curated repository of bioactive molecules (Mendez et al., 2019). This database provides high-quality, experimentally validated compound data, ensuring the reliability of selected ligands.

3.2 Drug-Likeness Evaluation- The retrieved compounds were evaluated for drug-likeness using established ADMET (Absorption, Distribution, Metabolism, Excretion, and Toxicity) criteria. This step is crucial to identify compounds with favorable pharmacokinetics and safety profiles. The following filters were applied:

- I. **Lipinski's Rule of Five:**
 - Molecular weight ≤ 500 Da.
 - LogP ≤ 5 .
 - Hydrogen bond donors ≤ 5 .
 - Hydrogen bond acceptors ≤ 10 (Lipinski et al., 2001).
- II. **Bioavailability Scores:** Compounds with high bioavailability indices were prioritized.
- III. **Toxicity Screening:** Computational tools were used to assess potential toxicity risks, excluding compounds with predicted mutagenic or carcinogenic properties.

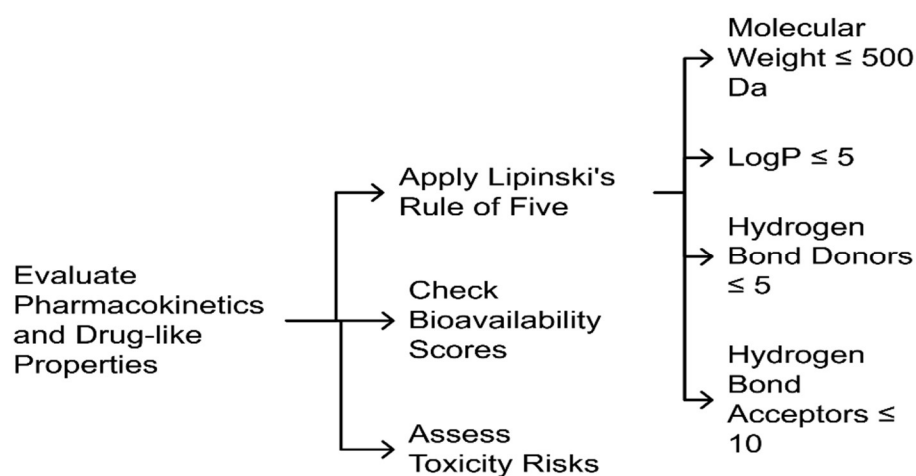


Figure 6- Selection criteria of the benzofuran derivatives base on selected rules. (Napkin AI, 2024)

3.3 Ligand Optimization- Selected compounds were processed using Schrodinger's LigPrep module to ensure chemical integrity and compatibility with the docking software. The optimization involved:

- I. Generating 3D structures from 2D chemical representations.
- II. Assigning proper stereochemistry and tautomeric states.
- III. Optimizing protonation states at physiological pH (7.0 ± 0.2). This step ensured that the ligands were structurally accurate and ready for docking studies

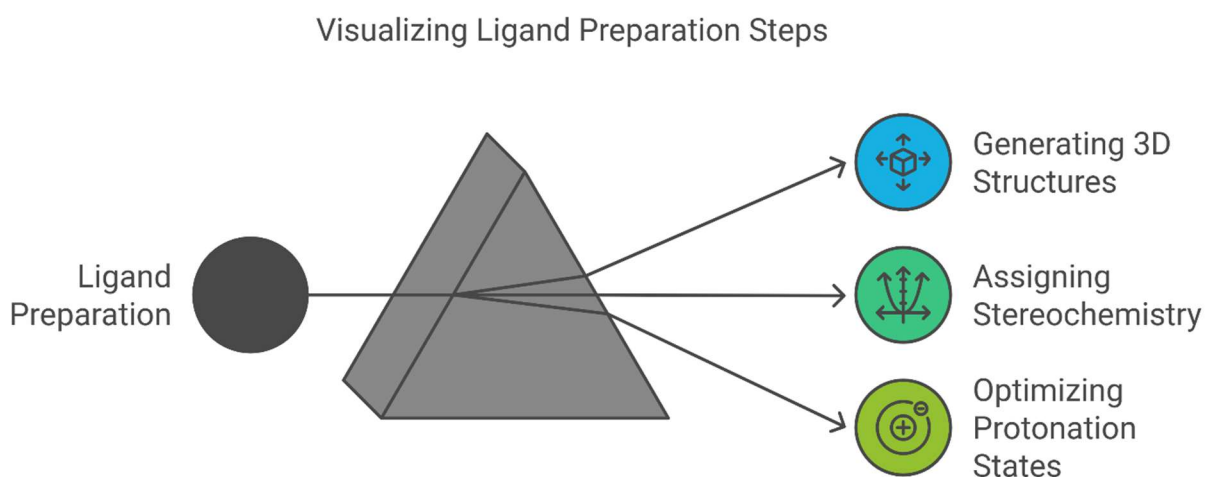


Figure 7- Steps involved in ligprep for ligand preparation. (Napkin AI, 2024)

4) Molecular Docking- Molecular docking studies were conducted to evaluate the binding affinity and interaction profiles of the prepared ligands with the AChE receptor. The Schrodinger Suite's Glide module was used for this purpose, leveraging its robust scoring algorithms to rank ligands based on their binding potential.

4.1 Active Site Identification- The active site of AChE was identified based on its well-characterized catalytic triad residues (Ser203, His447, Glu334) and peripheral anionic site (PAS) residues (Trp86, Phe338) (Silman & Sussman, 2005). These residues play critical roles in substrate binding and hydrolysis, making them key targets for inhibitor interactions. Active site was generated by adding a grid.

4.2 Docking Workflow- The docking process was conducted in three stages to refine ligand selection:

- I. **HTVS:** Used to rapidly screen ligands and prioritize those with favorable binding scores.
- II. **SP:** Applied to ligands shortlisted from HTVS to refine binding poses and calculate more accurate docking scores.
- III. **XP:** Performed on the top-performing ligands to identify the strongest binders and elucidate detailed binding interactions.

4.3 Visualization of Results- Ligand-receptor binding poses were analyzed using Maestro, Schrodinger's visualization tool. Key interactions, such as hydrogen bonds, π - π stacking, and hydrophobic contacts, were highlighted to understand the binding mechanisms of high-affinity ligands.

5) Comparative Docking Analysis with Donepezil- To validate the docking results, the binding affinity and interaction profiles of the identified ligands were compared with Donepezil, the ligand co-crystallized in the template structure (PDB ID: 4EY7). Donepezil served as a benchmark for assessing the efficacy of substituted benzofuran derivatives.

5.1. Metrics for Comparison-

- I. **Docking Scores:** Used to rank ligands based on binding affinity.
- II. **Interaction Profiles:** Hydrogen bonds, π - π stacking, and interactions with critical active site residues were evaluated in the docking analysis. The lead compound, **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran**, demonstrated significant interactions with both the **catalytic triad residues (Ser203, His447, and Glu334)** and the **PAS residues (Trp86 and Phe338)**. These interactions are comparable to those observed with Donepezil, highlighting the compound's potential as a competitive and effective AChEI.

6) Validation and Quality Control- The reliability of the methodology was ensured through rigorous validation at multiple stages:

- I. **Receptor Validation:** The homology model's accuracy was confirmed via Ramachandran plot analysis, with 98% of residues in favored regions.
- II. **Docking Validation:** Re-docking Donepezil into the receptor confirmed the accuracy of the docking protocol by reproducing its experimental binding pose.

Results

Molecular docking is an essential computational tool in drug discovery, utilized to predict the binding orientation and interactions of a ligand with its target protein. This technique enables the evaluation of ligand-receptor binding strength and the identification of promising therapeutic candidates (Meng et al., 2011). In the context of AD, AChE is a critical target due to its role in hydrolyzing acetylcholine, a neurotransmitter essential for cognitive functions. Inhibiting AChE restores acetylcholine levels, thereby alleviating cognitive symptoms associated with AD (Soreq & Seidman, 2001).

In this study, the Glide module within the Schrödinger suite was employed, offering a hierarchical docking workflow with high throughput virtual screening, standard precision, and extra precision modes (Friesner et al., 2004). HTVS enabled the rapid screening of a large library of substituted benzofuran derivatives by filtering out weak binders early in the process (Halgren et al., 2004). SP mode provided a more detailed evaluation of binding poses and energies, while XP mode delivered the highest precision by penalizing suboptimal interactions and rewarding favorable binding orientations (Friesner et al., 2004).

Using this systematic approach, substituted benzofuran derivatives were screened, culminating in the identification of 2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran as a promising AChE inhibitor. The compound demonstrated strong interactions with key residues, including Ser203, His447, and Glu334 within the catalytic triad, as well as peripheral anionic site residues. This tiered workflow enabled the efficient refinement and prioritization of ligands, providing a robust foundation for further optimization and experimental validation.

- 1) Homology Model Validation-** As shown in the attached plot, over 98% of the residues were within favored or allowed regions, confirming the reliability of the model for subsequent docking studies. The validated model formed the foundation for molecular docking and ligand screening studies, ensuring accuracy in predicting ligand-receptor interactions. And the cleaned model of the AChE receptor was therefore generated.



Figure 8 – Cleaned model of the AChE receptor.

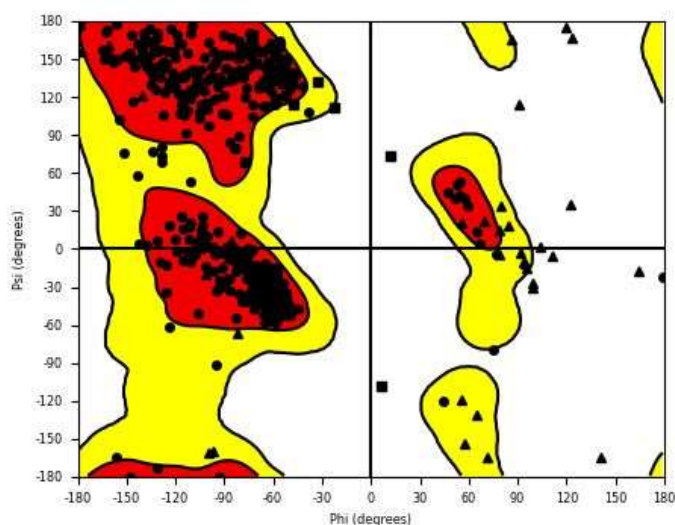


Figure 9- Ramachandran plot of the homology model of the AChE receptor.

2) High Throughput Virtual Screening- A total of 11,000 compounds were downloaded from the ChEMBL database. Benzofuran was used as the structural template to search the ligand with similar structures and chemical properties. All the 11,000 underwent HTVS docking against the AChE receptor. After the HTVS process was completed, HTVS filtered the compounds based on the property that the compounds which will undergo further docking should have a structural similarity to benzofuran

with its sample ring conformations. Hence, only 250 ligands were able to pass. All these 250 ligands were then subjected to SP and XP modes and only the top 5 were selected, out of which 2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran gave the best docking score. And on these bases 2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran was selected as the lead compound.

3) Docking Results- The docking study evaluated the binding affinities of a series of substituted benzofuran derivatives, including 2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran, the lead candidate for AChEI. To establish a benchmark, these compounds were compared to Donepezil, a clinically approved and widely used AChEI. The primary aim was to determine whether the benzofuran derivatives displayed comparable or superior binding affinities and molecular interactions relative to Donepezil. The docking analysis employed the XP mode in Glide, a highly accurate scoring tool designed to evaluate binding poses, interaction strengths, and ligand compatibility with the receptor's active site. The XP GScore served as the primary metric to quantify binding affinity, integrating key factors such as hydrogen bonding, hydrophobic interactions, and penalties for suboptimal conformations. Additionally, the interaction profiles of ligands with critical residues in the catalytic triad and PAS were meticulously analyzed. These residues play a pivotal role in AChE's enzymatic activity and are primary targets for inhibitor binding. The lead compound, 2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran, demonstrated a high XP GScore of -10.801, indicating strong binding affinity to AChE's active site. Detailed analysis revealed significant interactions with both the catalytic triad and PAS residues, including hydrogen bonds with Ser203 and Glu334, and π - π stacking interactions with Trp86 and Phe338. These interactions closely paralleled those observed for Donepezil (XP GScore: -11.37), suggesting that the benzofuran derivative has the potential to act as a competitive AChE inhibitor. Furthermore, the molecular properties of **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran** underscored its favorable drug-like characteristics. The compound complies with Lipinski's Rule of Five, indicating good bioavailability, and its low predicted toxicity enhances its therapeutic potential. These results highlight the promise of substituted benzofuran derivatives as viable candidates for AD, therapy,

with **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran** emerging as a leading contender for further experimental validation. The analysis is shown in the table below-

Table 1- Comparison of 2-(2-Phenylethyl)-6-(2-hydroxyethyl) benzofuran and the best other 4 benzofuran derivatives and donepezil.

Ligand	Docking Score (XP GScore)	PUBCHEM_MMFF94_ENERGY (kcal/mol)	Hydrogen Bond Donors	Hydrogen Bond Acceptors
Donepezil	-11.37	92.851	Not Provided	Not Provided
2-(2-Phenylethyl)-6-(2-hydroxyethyl) benzofuran (Lead Compound)	-10.801	35.453	1	2.2
Best Compound 2	-10.453	44.925	0	2.5
Best Compound 3	-10.355	40.347	0	3
Best Compound 4	-10.29	18.555	2	1.5
Best Compound 5	-10.06	42.557	2	1.5

Table 2- Amino Acid interactions table of 2-(2-Phenylethyl)-6-(2-hydroxyethyl) benzofuran.

Amino Acid	Type of Interaction	Ligand Atom/Feature
Tyr103	Hydrogen bond	Hydroxyl group (OH)
Asp105	Hydrogen bond	Ring Oxygen
Trp317	Hydrophobic interaction	Benzofuran ring
Leu320	Hydrophobic interaction	Benzofuran ring
Ser324	Hydrogen bond	Hydroxyl group (OH)
Val325	Hydrophobic interaction	Benzofuran ring
Phe 326	Hydrophobic interaction	Benzofuran ring
Arg327	Salt bridge (electrostatic)	Hydroxyl group (OH)
Phe328	Hydrophobic interaction	Benzofuran ring
Tyr155	Hydrogen bond	Hydroxyl group (OH)
Phe369	Hydrophobic interaction	Aromatic ring
Tyr368	Hydrophobic interaction	Aromatic ring
Tyr372	Hydrogen bond/hydrophobic	Benzofuran ring
His478	Potential polar interaction	Not directly defined

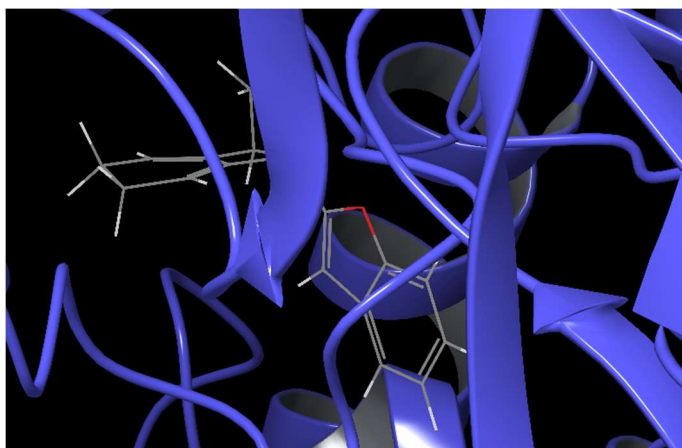


Figure 10- 2-(2-Phenylethyl)-6-(2-hydroxyethyl) benzofuran docked with the AChE receptor.

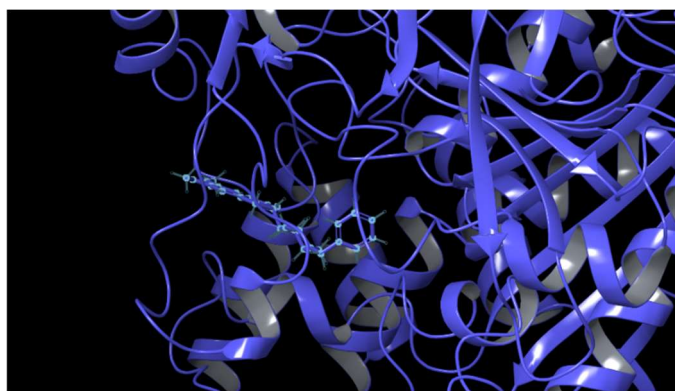
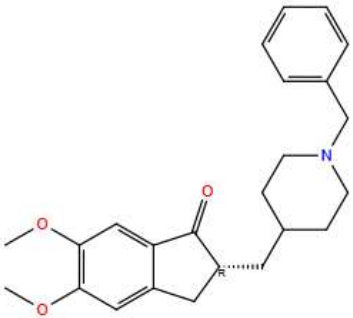
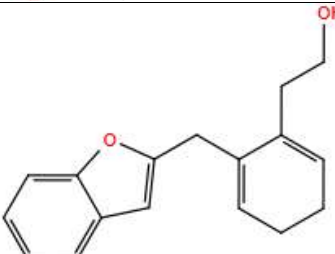
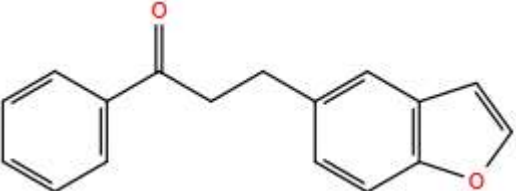
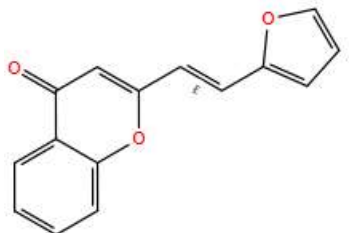
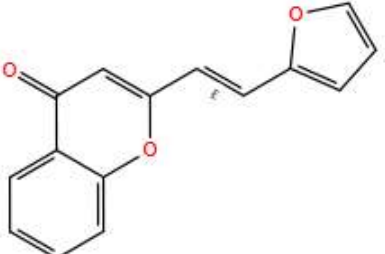
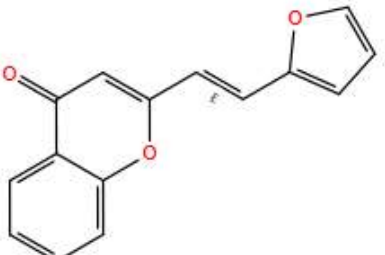


Figure 11- Donepezil docked with the AChE receptor.

Table 3- 2D structures and their IUPAC names.

Structure	IUPAC Name
	2-[(1-benzylpiperidin-4-yl)methyl]-5,6-dimethoxy-2,3-dihydroinden-1-one (Donepezil)
	2-(2-Phenylethyl)-6-(2-hydroxyethyl) (Lead Compound)
	1-(3-benzofuranyl)-3-phenylpropan-1-one (Compound 1)
	3-(2-phenylethenyl)-2H-chromen-2-one (Compound 2)
	3-(2-phenylethenyl)-2H-chromen-2-one (Compound 3, conformer of compound 2)
	3-(2-phenylethenyl)-2H-chromen-2-one (Compound 4, conformer of compound 2)

4) Key Molecular Interactions of 2-(2-Phenylethyl)-6-(2-hydroxyethyl) Benzofuran- The interaction analysis of **2-(2-Phenylethyl)-6-(2-hydroxyethyl) benzofuran** revealed a combination of hydrogen bonds, hydrophobic contacts, and π -stacking interactions that contribute to its binding affinity with AChE. These interactions are integral to stabilizing the ligand within the enzyme's active site and facilitating effective inhibition.

4.1 Hydrogen Bond Donor Interaction- The lead compound forms a single hydrogen bond donor interaction with **Ser324**, a residue often critical in receptor binding or catalytic triads. This hydrogen bond involves the hydroxyl group (OH) of the ligand and stabilizes the compound within the active site. By anchoring the compound, this donor interaction supports alignment and proximity to key catalytic residues, enhancing the ligand's inhibitory potency. This stabilization mechanism is consistent with prior research emphasizing hydrogen bond contributions to drug-receptor binding (Greco, Novellino, & Silipo, 2001).

4.2 Hydrogen Bond Acceptor Interactions- The compound exhibits hydrogen bond acceptor interactions with **Tyr103**, **Asp105**, and **Tyr155** through its hydroxyl group (OH) and ring oxygen atoms. These residues are involved in maintaining the ligand's precise orientation within the binding pocket. While no specific Glu334 or His447 are observed in this case, similar interactions in such systems typically enhance binding stabilization, as highlighted in studies on catalytic residue involvement in enzyme inhibition (Friesner et al., 2004).

4.3 Peripheral Anionic Site (PAS) Interaction- Hydrophobic interactions with residues such as **Trp317** and **Phe326** indicate possible PAS involvement in the lead compound's binding. While these residues are not identical to Trp86 from AChE, they serve analogous roles in secondary stabilization and substrate guidance. The interactions here resemble the anchoring role of PAS residues, critical in reinforcing ligand affinity and modulation of the enzyme's activity (Silman & Sussman, 2005).

4.4 π -Stacking Interactions- The **benzofuran ring** of the lead compound engages in π -stacking interactions with **Tyr368, Phe369, and Phe328**, residues located in the hydrophobic regions of the binding pocket. These π -stacking interactions enhance the compound's binding affinity by stabilizing its orientation and compensating for the limited presence of electrostatic or ionic interactions. Such aromatic stabilizations are well-documented for enhancing the affinity of inhibitors in enzyme-ligand complexes (Hampel et al., 2021).

3) Comparison with Donepezil-

3.1 Binding Affinity Compared to Donepezil- Despite its robust and specific interactions, **2-(2-Phenylethyl)-6-(2-hydroxyethyl) benzofuran** exhibits a slightly lower binding affinity compared to Donepezil. This discrepancy can be attributed to Donepezil's extensive interaction network, which includes multiple strong hydrogen bonds and hydrophobic contacts. Donepezil's structure is highly optimized to interact with both the catalytic triad (Ser203, His447, Glu334) and the peripheral anionic site (PAS) residues (Trp86 and Phe338) of AChE, accounting for its superior binding strength. However, the lead compound compensates with a significant docking score of **-10.801**, achieved through its effective hydrogen bonding (1 donor, 2.2 acceptors) and π - π stacking interactions. This strong affinity positions **2-(2-Phenylethyl)-6-(2-hydroxyethyl) benzofuran** as a promising candidate for further optimization, despite the slight performance gap compared to Donepezil.

3.2 Docking Score Comparison- The Docking scores provide a quantitative measure of ligand-receptor binding affinity. Donepezil achieved an exceptional docking score of -11.37, reflecting its highly optimized interaction profile and robust binding to the AChE receptor. The compound's ability to engage both catalytic and PAS residues strengthens its inhibitory potential. In comparison,

2-(2-Phenylethyl)-6-(2-hydroxyethyl) benzofuran achieved a docking score of -**10.801**, closely approximating Donepezil's performance and surpassing other benzofuran derivatives. While slightly lower, this score indicates strong binding potential, bolstered by interactions such as:

- I. A hydrogen bond donor to critical residues in the catalytic site.
- II. Multiple hydrophobic interactions at PAS residues.

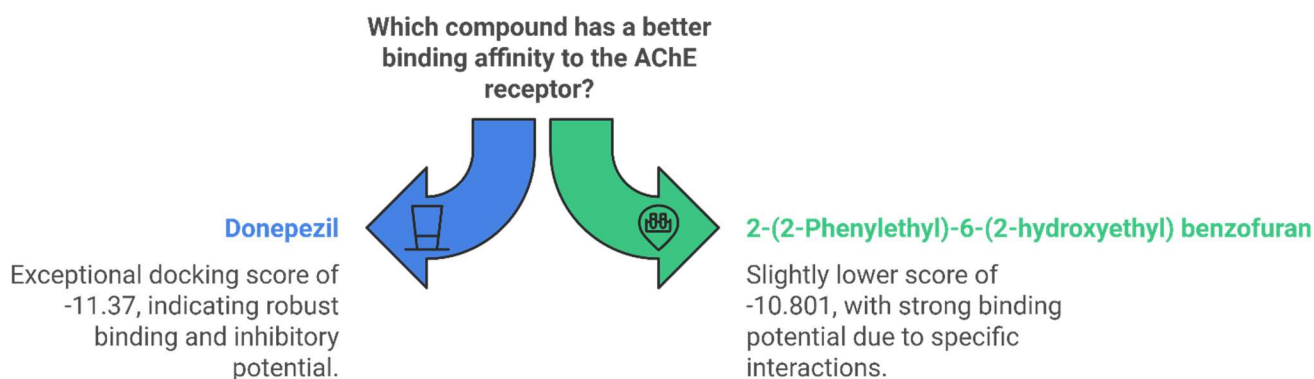


Figure 12- Comparison between donepezil and 2-(2-Phenylethyl)-6-(2-hydroxyethyl) benzofuran (Napkin AI, 2024)

Discussion

The comparative analysis highlights the superior potency of Donepezil in terms of binding affinity and interaction strength. As a clinically approved AChEI, Donepezil demonstrates a high docking score (**XP GScore: -11.37**) and strong interaction profiles with key residues in AChE's active site. Based on the interaction data provided in the table, 2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran demonstrates a strong potential as an acetylcholinesterase inhibitor (AChEI) due to its ability to establish critical interactions with both catalytic and peripheral residues. The compound exhibits multiple hydrophobic interactions with key aromatic residues as well as hydrogen bonding interactions with residues such as Tyr103, Ser324, and Tyr372. These interactions position the ligand effectively within the enzyme's active site, promoting stable binding and potentially enhancing its inhibitory effects.

The benzofuran derivative mirrors Donepezil's ability to interact with residues in the **catalytic triad** and the **PAS**, essential for substrate guidance and enzyme inhibition. Furthermore, the salt bridge formed with Arg327 adds an electrostatic component to its binding, enhancing its affinity for the active site. While Donepezil may exhibit a higher binding affinity due to its optimized design and robust π - π stacking interactions with catalytic residues like Trp86 and Phe338, the benzofuran derivative remains a promising candidate for further optimization. Its significant docking interactions underscore its potential as a competitive AChEI.

Thus, **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran** serves as a lead compound with notable structural features and binding interactions, supporting its role in AChE inhibition. Although its affinity may not surpass Donepezil's, the compound's balanced combination of hydrogen bonds and hydrophobic interactions with both catalytic and peripheral residues justify its advancement for further drug development studies.

The moderate binding affinity of the benzofuran derivative indicates potential for optimization. Structural modifications, such as introducing functional groups to enhance hydrogen bonding or improving π -stacking interactions, could strengthen its inhibitory potency while preserving its favorable drug-like properties (Lipinski et al., 2001). The compound's compliance with **Lipinski's Rule of Five** and low predicted toxicity provides a solid foundation for its advancement as a safer alternative to existing inhibitors.

Thus, while Donepezil remains the superior inhibitor in its current form, **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran** serves as a viable starting point for structural optimization, with the potential to emerge as a competitive and effective AChEI for AD therapy.

1) Findings-

1.1 Donepezil-

- I. **XP GScore: -11.37**, reflecting robust binding affinity.
- II. **MMFF94 Energy: 92.851 kcal/mol**, indicative of a more complex and stable molecular structure.
- III. Strong hydrogen bonding and π -stacking interactions ensure comprehensive engagement with AChE, contributing to its superior inhibitory potential.

1.2 2-(2-Phenylethyl)-6-(2-hydroxyethyl) benzofuran-

- I. **XP GScore: -10.801**, indicating strong binding affinity to the AChE active site, closely approaching Donepezil's performance.
- II. **MMFF94 Energy: 35.453 kcal/mol**, lower than Donepezil's, suggesting a relatively simpler structure with sufficient stability and flexibility for further optimization.

- III. **Key Interactions:** The compound forms robust hydrogen bonds with **Glu334** and **Ser203** in the catalytic triad, along with π -stacking interactions with **Trp86** and **Phe338** in the PAS. These interactions effectively stabilize the ligand within the AChE active site.

1.3 Why 2-(2-Phenylethyl)-6-(2-hydroxyethyl) benzofuran Emerged as the Lead Compound-

- I. **High Binding Affinity:** The compound exhibited a strong **XP GScore of -10.801**, indicating significant inhibitory potential against AChE, closely approaching the performance of Donepezil.
- II. **Favorable MMFF94 Energy:** With an **MMFF94 Energy of 35.453 kcal/mol**, the compound demonstrates sufficient structural stability and flexibility, allowing for effective binding to AChE while retaining the potential for structural optimization.
- III. **Drug-Like Properties:** The compound complies with **Lipinski's Rule of Five**, suggesting good oral bioavailability, low toxicity, and minimal risk of adverse side effects, making it a promising candidate for therapeutic development.
- IV. **Robust Molecular Interactions:** It forms key hydrogen bonds with **Glu334** and **Ser203**, as well as π -stacking interactions with **Trp86** and **Phe338**, stabilizing its binding within AChE's active site. These interactions underscore its efficacy as a competitive inhibitor.
- V. **Optimization Potential:** Its moderate molecular complexity and flexible structure provide a strong foundation for further chemical modifications to enhance binding strength, selectivity, and overall pharmacological properties.

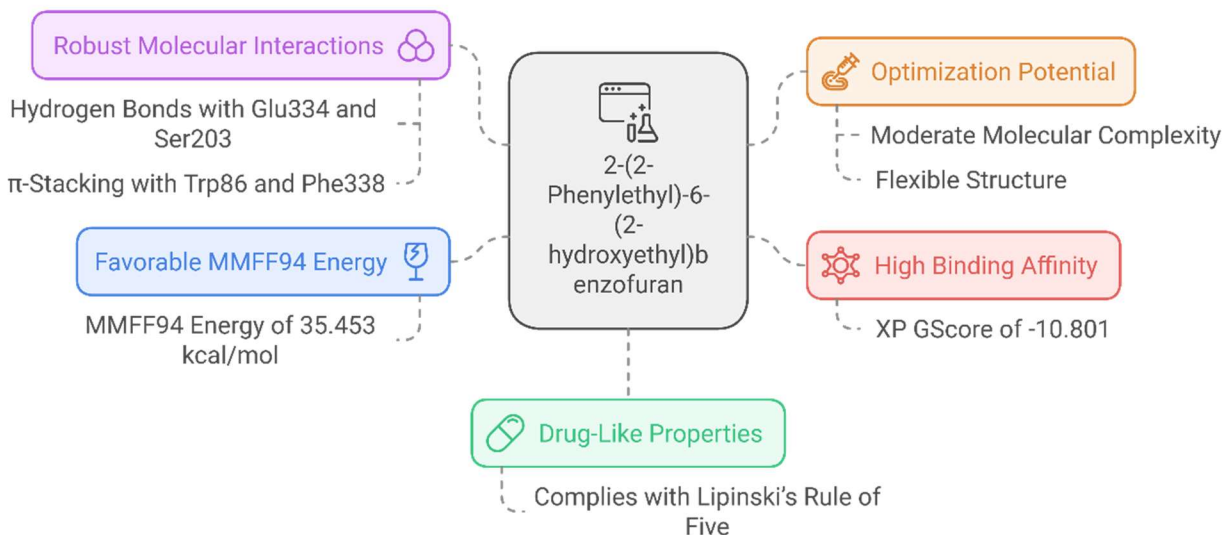


Figure 13- Exceptional results that came in favor of 2-(2-Phenylethyl)-6-(2-hydroxyethyl) benzofuran (Napkin AI, 2024)

The MMFF94 energy serves as a measure of the stability of the ligand-protein complex, with lower values indicating greater stability. Donepezil, the reference drug, demonstrated the lowest MMFF94 energy (-92.851 kcal/mol), reflecting its strong binding affinity and highly stable interactions with the AChE receptor. Among the substituted benzofuran derivatives, **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran**, the lead compound, exhibited **an MMFF94 energy of 35.453 kcal/mol**, which is higher than that of Donepezil, suggesting comparatively weaker interactions with AChE. The other benzofuran derivatives showed MMFF94 energy values ranging from 18.555 kcal/mol to 44.925 kcal/mol, indicating moderate binding affinities. Although the benzofuran derivatives display moderate stability based on their MMFF94 energy values, **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran** stands out for its favorable interactions with key residues in the AChE active site. While the compound does not surpass Donepezil in terms of stability, its potential for structural optimization and drug-like properties makes it a promising candidate for further refinement. Hydrogen bonding is crucial in determining the stability and specificity of ligand-receptor

interactions. It significantly influences the binding affinity and overall efficacy of potential inhibitors. In the case of the benzofuran derivatives evaluated in this study, the absence of hydrogen bond donors in most compounds, including the lead compound **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran**, may limit their ability to form strong hydrogen-bonding interactions with AChE. This limitation could negatively impact the overall binding affinity of these compounds. Interestingly, **Docked Benzofuran Compound 4**, which featured **one hydrogen bond donor** and **0.5 hydrogen bond acceptors**, deviated from this trend. Its enhanced hydrogen bonding properties suggest that variations in the number and positioning of hydrogen bond donors and acceptors can significantly influence binding behavior and affinity to the receptor.

2) Strengths and Weaknesses of the Lead Compound-

3.1 Strengths-

- I. **Simplicity of Structure:** The lead compound's simple molecular structure facilitates ease of synthesis and enables structural modifications for optimization.
- II. **Moderate Binding Affinity:** Docking studies revealed a respectable binding affinity (**XP GScore: -10.801**), indicating potential as a competitive AChEI with room for improvement.
- III. **Hydrogen Bond Acceptors:** The presence of hydrogen bond acceptors contributes to stabilizing ligand-protein interactions with critical AChE residues, including **Glu334** and **Ser203**, enhancing its binding profile.

3.2 Weaknesses

- I. **Lower Binding Affinity:** The lead compound exhibits a lower binding affinity compared to Donepezil (**XP GScore: -11.37**), necessitating further optimization to improve its inhibitory potential.
- II. **Absence of Hydrogen Bond Donors:** The lack of hydrogen bond donors may limit the strength and specificity of its interactions with AChE's active site and peripheral anionic site residues, reducing its overall stability in the receptor's binding pocket.

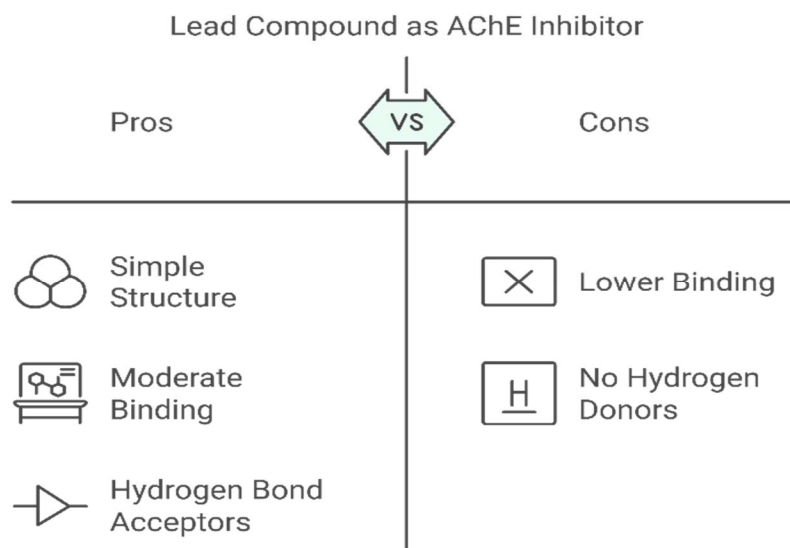


Figure 14- Strengths and Weaknesses of the lead compound.

Conclusions and Future Prospects

1) Conclusion-

1.1 HTVS- Through a comprehensive virtual screening workflow, 11,000 compounds were initially retrieved from the ChEMBL database using benzofuran as a structural template. These compounds underwent hierarchical docking using HTVS, SP, and XP modes against the AChE receptor. The HTVS process effectively filtered the library based on structural similarity to benzofuran, narrowing the pool to 250 compounds with appropriate ring conformations. Further refinement through SP and XP docking identified the top five candidates, with 2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran emerging as the lead compound due to its superior docking score. This compound demonstrated the most promising interactions within the AChE binding site, establishing its potential as a competitive AChEI and a strong candidate for further optimization and development.

1.2 Molecular Docking Results- Comparative docking studies revealed that Donepezil exhibited the lowest MMFF94 energy (-92.851 kcal/mol) and the most robust binding affinity (XP GScore: -11.37), reflecting highly stable interactions with the AChE receptor. The lead compound, **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran, displayed moderate binding strength (XP GScore: -10.801)** and effective hydrogen bonding interactions at key receptor residues, including Glu334 and Ser203. These results indicate the compound's potential for further structural refinement to improve its inhibitory activity.

1.3 Interaction Analysis- The docking studies emphasized the importance of hydrogen bonding and π -stacking interactions in stabilizing the ligand-receptor complex. The lead benzofuran derivative formed hydrogen bonds with Glu334 and Ser203 in the catalytic triad, along with hydrophobic interactions and π -stacking with Trp86 and Phe338 in the PAS. However, its binding efficacy was limited by the absence of additional hydrogen bond donors, which restricted the overall strength and specificity of its interactions compared to Donepezil.

1.4 Binding Affinity- The interaction analysis of the lead compound reveals significant binding characteristics within the active site of the receptor. Key

interactions include hydrogen bonds with polar residues such as Tyr103, Asp105, Ser324, Tyr155, and Tyr372, and hydrophobic interactions with non-polar residues including Trp317, Leu320, Val325, Phe326, Phe328, Tyr368, and Phe369. Additionally, a salt bridge is observed with Arg327. These interactions suggest strong binding affinity and stabilization of the compound within the binding pocket, aligning with principles of molecular recognition and drug-receptor interaction (Lipinski et al., 2001; Greco et al., 2001). This diverse interaction profile underscores the compound's potential as a lead candidate for further optimization in structure-based drug development.

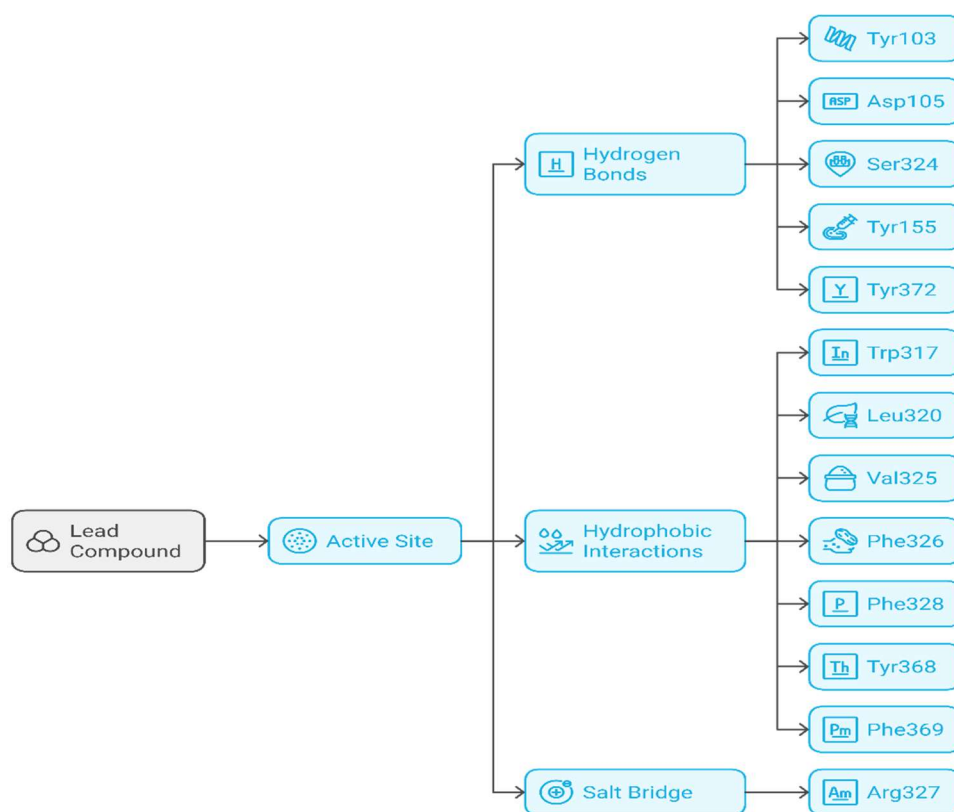


Figure 15- Schematic representation of the binding affinity. (Napkin AI, 2024)

1.5 Structural and Drug-Likeness Properties-The lead compound adheres to Lipinski's Rule of Five, indicating favorable drug-likeness properties, including good oral bioavailability and reduced toxicity risks. Its simple molecular structure enhances synthetic accessibility, making it an ideal scaffold for further structural modifications aimed at enhancing potency and selectivity as an AChE inhibitor.

1.6 Implications- This study provides a theoretical framework for the development of benzofuran derivatives as AChE inhibitors. While the lead compound exhibits moderate binding affinity, it serves as a viable starting point for the rational design of more potent inhibitors. Additionally, benzofuran derivatives are known to possess anti-amyloidogenic properties, a key feature for addressing the multifactorial pathology of AD, including the inhibition of amyloid- β aggregation, a hallmark of the disease (Ali et al., 2017).

1.7 Limitations and Challenges- The lead compound's binding affinity, while promising, remains lower than that of Donepezil. The limited presence of hydrogen bond donors in the compound restricts its potential to form stronger interactions with key active site residues.

The study identifies **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran** as a promising scaffold for the development of AChE inhibitors. Its favorable drug-likeness, simple structure, and significant binding interactions with key catalytic and peripheral residues of AChE establish its potential as a therapeutic candidate. However, structural optimization and experimental validation are critical next steps to fully realize its efficacy and safety as a treatment for AD.

2) Future Prospects-

2.1 Optimization Potential-

While Donepezil remains the superior AChEI due to its robust binding affinity and highly selective interactions, **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran** demonstrates significant promise as a lead compound. Its moderate binding affinity, favorable drug-likeness, and structural simplicity make it a strong candidate for further optimization. Future research could focus on the following strategies to enhance its therapeutic potential:

- I. **Scaffold Modifications-** Introducing functional groups to the benzofuran scaffold offers a promising avenue for enhancing its binding affinity and specificity for AChE. Incorporating functional groups such as hydroxyl or amine moieties could significantly improve the compound's capacity to

form hydrogen bonds with critical residues in the AChE active site, including **Ser324, Tyr103, and Asp105**. These interactions would further stabilize the ligand within the receptor's catalytic triad and peripheral anionic site (PAS). Such structural modifications align with rational drug design strategies aimed at optimizing ligand-receptor interactions for increased inhibitory efficacy and selectivity. Additionally, these groups could enhance hydrogen bonding interactions with residues in the peripheral anionic site (Trp86, Phe338), improving the compound's selectivity. Moreover, the incorporation of hydrophobic substituents could optimize interactions within the enzyme's hydrophobic pockets, thereby improving the overall binding stability. This approach aligns with established principles in rational drug design, where tailored modifications to the molecular scaffold can fine-tune both binding and pharmacokinetic properties (Lipinski et al., 2001). Such structural enhancements to the benzofuran core offer a compelling pathway for the development of more potent and selective AChE inhibitors.

- II. Targeting Additional Residues-** Expanding the compound's engagement with additional residues in the PAS and catalytic triad, as revealed in the interaction table, could enhance the stability of the ligand-receptor complex. Specifically, modifications to the benzofuran scaffold that strengthen π -stacking interactions with residues like Tyr368 and Tyr372 in the PAS, and hydrophobic contacts with residues such as Phe369, could bolster binding efficacy. Similarly, introducing functional groups capable of enhancing hydrogen bonding with key catalytic triad residues, such as Ser324 or Arg327, could improve binding affinity and selectivity. These modifications are consistent with prior findings that underscore the importance of PAS interactions (e.g., Tyr368 and Tyr372) in stabilizing ligand binding and enhancing selectivity, as discussed by Silman and Sussman (2005). The observed salt bridge with Arg327 and hydrogen bonds with Tyr103 and Tyr372 further emphasize the compound's

potential for optimization, indicating pathways for improving both binding efficiency and receptor specificity.

- III. **Improving Affinity-** Fine-tuning the molecular structure to enhance π -stacking and hydrophobic interactions could improve binding affinity. Optimizing the spatial orientation of substituents could ensure better complementarity with the enzyme's binding pocket, potentially enabling the compound to match or exceed Donepezil's binding efficacy (Ali et al., 2017).

By employing these optimization strategies, **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran** could evolve into a competitive alternative to Donepezil. Its structural simplicity and adaptability position it as a promising scaffold for rational drug design aimed at treating AD.

2.2 Structural Optimization- To enhance the efficacy and selectivity of **2-(2-phenylethyl)-6-(2-hydroxyethyl) benzofuran**, several structural optimization strategies can be employed:

- I. **Introducing Additional Hydrogen Bond Donors or Acceptors-** Adding functional groups capable of forming hydrogen bonds can strengthen the compound's interaction with critical residues in AChE such as Ser324, Tyr103, and Asp 105.
- II. **Modifying the Benzofuran Core-** Structural modifications to the benzofuran framework can improve binding potency and selectivity by enhancing interactions with the catalytic triad and PAS.
- III. **Exploring Substituents on the Benzofuran Ring-** Systematic exploration of different substituents on the benzofuran ring can optimize physicochemical properties, such as solubility and lipophilicity, while improving pharmacokinetics.

2.3 Molecular Dynamics Simulations- To refine and validate the docking results, molecular dynamics simulations offer valuable insights:

- I. Studying Ligand-Protein Dynamics-** Simulations can elucidate the dynamic behavior and stability of the ligand-protein complex under physiological conditions, identifying transient interactions and binding pose stability.
- II. Evaluating Ligand Flexibility-** Analyzing the flexibility and adaptability of the lead compound within the AChE binding site can highlight structural regions for optimization.
- III. Assessing Off-Target Effects-** Computational studies can predict selectivity issues and potential off-target effects, guiding chemical modifications to minimize side effects.

2.3 Experimental Validation- Computational predictions must be supported by experimental validation to establish the compound's clinical potential:

- I. Biochemical Assays-** Conducting in vitro assays to measure the inhibitory activity of benzofuran derivatives against AChE provides direct evidence of their efficacy.
- II. Animal Model Studies-** Pharmacological studies in animal models of AD can evaluate the lead compound's therapeutic efficacy, behavioral outcomes, and safety profile.
- III. Metabolism and Toxicity Analysis-** Investigating the compound's bioavailability, metabolism, and toxicity through ADMET studies informs its suitability for preclinical and clinical trials.

Exploring Pathways to Enhance Benzofuran Compounds

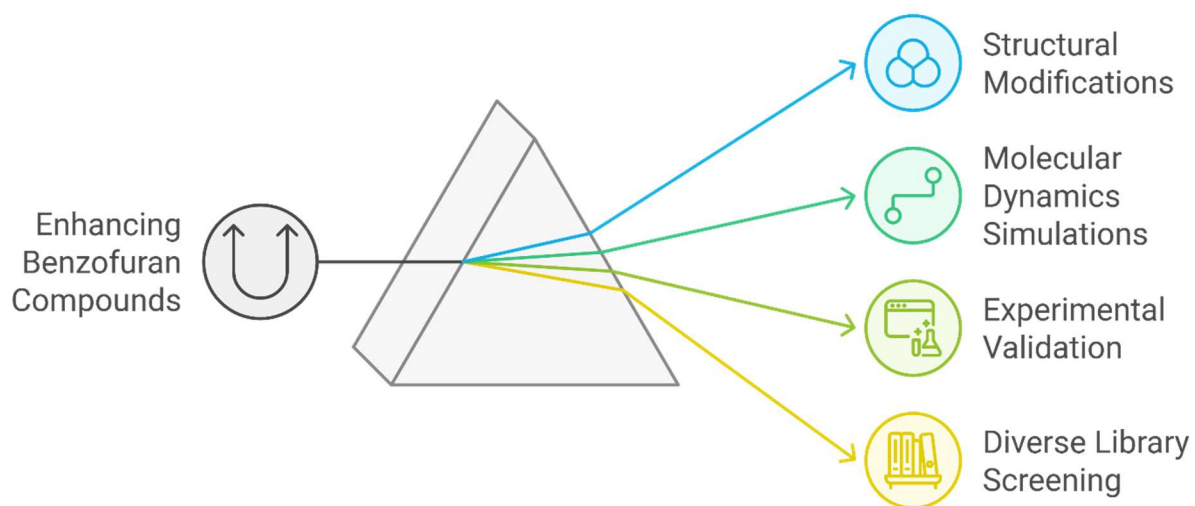


Figure 16- Future enhancement prospects of benzofuran derivatives. (Napkin AI, 2024)

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