Department of Applied Sciences & Humanities Faculty of Engineering & Technology B. Tech. Semester-I End Semester Examination 2023-24 Paper Code: ASB-102

Maximum Marks: 45

Duration: 3 Hours

Instruction to the candidates

· Write your Roll No. on the top immediately on receipt of the question paper.

· Avoid circumlocution

Note: Attempt any two parts of each question. All questions carry equal marks

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	. No.	Questions	Marks	CO
1.	(a)	What is critical micelle concentration? Discuss the factors which	4.5	CO-1
		affect critical micelle concentration.		
	(b)	What are crystal defects? How crystal defects changes the	4.5	CO-1
		conductivity and density of the crystals.		
	(c)	Define atomic packing factor. How would you calculate it for	4.5	CO-1
		simple cubic and body centered unit cells?		
2.	(a)	Define principle of adsorption chromatography. Discuss its	4.5	CO-2
		various types and application.		
	(b)	Explain instrumentation and applications of UV-Vis spectroscopy.	4.5	CO-2
	(c)	What is atomic absorption spectroscopy used to determine?	4.5	CO-2
		Explain instrumentation and applications of atomic absorption		
		spectroscopy.		
3.	(a)	What are complexometric titrations? Discuss the complexometric	4.5	CO-3
		titrations used for water analysis.		
	(b)	What is phase rule? Draw phase diagram of water system and	4.5	CO-3
		discuss it in detail.		
	(c)	What are precipitation reactions? Discuss favourable conditions	4.5	CO-3
		for precipitations.		
4.	(a)	Discuss principle of galvanic cells with special reference to	4.5	CO-4
		Daniel's cell.		
	(b)	What are electrode potential and standard electrode potentials?	4.5	CO-4
		Explain E.M.F. series.		
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	(c)	What is cell constant? Discuss its measurement.	4.5	CO-4
5.	(a)	Discuss synthesis, properties and uses of any two the following	4.5	CO-5
		polymers: (i) Nylon, (ii) Teflon, (iii) Polyester		
	(b)	Define thermoplastics and thermosets. Explain the two with	4.5	CO-5
		examples.		
	(c)	What are conducting polymers? Discuss applications of	4.5	CO-5
		conducting polymers by citing suitable examples.		

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