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Day 9

Ideal Gas

Start with a big box with 1 atom in it. What are the possible eigenvalues of energy?

$$E = \frac{\hbar^2 k^2}{2m}$$

Our wavefunction only has kinetic energy and there are three degrees of freedom i.e. p_x, p_y, p_z . You get a choice of boundary conditions — if we stick this in a box we now have:

$$E_{n_x,n_y,n_z} = \frac{\hbar^2 \pi^2 (n_x^2 + n_y^2 + n_z^2)}{2mL^2}$$

Now we want to work out the partition function:

$$Z = \sum_{n_x}^{\infty} \sum_{n_y}^{\infty} \sum_{n_z}^{\infty} e^{-\beta \frac{\hbar^2 \pi^2 (n_x^2 + n_y^2 + n_z^2)}{2mL^2}}$$

This sucks... So lets take a statistical approximation for a REALLY big box. First though let's simplify... We got this by separation of variables in quantum mechanics. Let's try and do that again.

$$\frac{\hbar^2 \pi^2 (n_x^2 + n_y^2 + n_z^2)}{2mL^2} \to \sum_{n_x}^{\infty} e^{-\beta \frac{\hbar^2 \pi^2 n_x^2}{2mL^2}} \sum_{n_y}^{\infty} e^{-\beta \frac{\hbar^2 \pi^2 n_y^2}{2mL^2}} \sum_{n_z}^{\infty} e^{-\beta \frac{\hbar^2 \pi^2 n_z^2}{2mL^2}}$$

$$= \left(\sum_{n}^{\infty} e^{-\beta \frac{\hbar^2 \pi^2 n_z^2}{2mL^2}}\right)^3 \quad \text{same in each dimension}$$

take: $\frac{\beta \pi^2}{2mL^2} \ll 1$ as classical approximation

So in this limit we expect that we can happily turn this sum into an integral for the limit given.

$$\begin{split} &\approx \int\limits_0^\infty e^{-\frac{\beta\hbar^2\pi^2}{2mL^2}n^2} dn \\ &\xi = \sqrt{\frac{\beta\hbar^2\pi^2}{2mL^2}n} \\ &\Rightarrow = \Big(\frac{2mL^2}{\beta\hbar^2\pi^2}\Big)^3 \Big(\int\limits_0^\infty e^{-\xi^2} d\xi\Big)^{3/2} = \Big(\frac{mL^2}{\beta\hbar^22\pi}\Big)^{3/2} \quad \text{handy integration trick for Gaussians} \end{split}$$

Thus we conclude that:

$$Z = \left(\frac{mkT}{2\hbar^2\pi}\right)^{3/2}V = n_QV$$
 quantum density

Question: What are S, U, p?

$$F = -kT \ln Z = -kT \ln \left(\left(\frac{mkT}{2\hbar^2 \pi} \right)^{3/2} V \right)$$

$$p = -\frac{\partial F}{\partial V} = \frac{kT}{V}$$

$$S = -\frac{\partial F}{\partial T} = k \ln(n_Q V) + \frac{3}{2} k$$

$$U = F + TS = \frac{3}{2} kT$$