

## Homework Set #5

**Due Date:** Before class Friday March 17th

### 1) Reading

(2 points)

Read chapter 3.

### 2) Discovery of the electron

(10 points)

The following table gives the Electronic and Magnetic fields and displacements measured by Thompson in his experiment on cathode rays. The E&M field where applied at right angles to the cathode ray motion over a region of XX. The rays where then allowed to drift for X before striking end of the tube where the displacements were measured.

numbers from weinberg.

- calculate  $q/m$  for each row
- calculate the particle  $v$  for each row ? Is it OK to use newtonian physics or are relativistic corrections required ?

### 3) Black body radiation

(30 points)

- a) Find and enumerate the different possible standing waves on a 1D string of length  $L$  with the condition that the amplitude goes to zero at the boundies. What are the allowed values of  $k = \frac{2\pi}{\lambda}$  ?
- b) Find and enumerate the different possible standing waves in a 3D box with sides of length  $L$  with the condition that the amplitude goes to zero at the boundies. What are the allowed values of  $k_x, k_y,$  and  $k_z$  ?
- c) Show that number of modes for a given  $k = |\vec{k}|$  is  $L^3 \frac{k^2}{\pi^2} dk$ .
- d) In the previus step you derived  $g_k(k)$ , the number of modes per  $k$  per volume, find  $g_\lambda(\lambda)$  from the constraint  $g_\lambda(\lambda)d\lambda = g_k(k)dk$ .
- e) Derive Rayleigh-Jeans equation (ie: the classical prediction for the energy density of black-body radiation) assuming that each mode has average energy  $kT$ .
- f) Using Planks assumption of quantized energies, derive Plank's law  $u(\lambda) = \frac{8\pi hc \lambda^{-5}}{e^{\frac{hc}{\lambda kT}} - 1}$ . Sketch this on top of the result from classical physics as a function of  $\lambda$ . *Hints:*  $\sum_0^\infty x^n = \frac{1}{1-x}$  for  $x < 1$  and  $\sum_0^\infty n e^{-nC} = -\frac{d}{dC} \sum_0^\infty e^{-nC}$
- g) What experimental features of the black-body radiation can be explained by classical physics? Which cannot?
- h) Show that Plank's law predicts the Stefan-Boltzmann law i.e. that the total energy density is proportional to  $T^4$ .

- i) Show that Planck's law predicts Wein's law i.e. that  $\lambda_{max} \sim \frac{1}{T}$ .

**4) Photoelectric Effect.**

(17 points)

- a) How is the result that the maximum photoelectric current is proportional to the intensity explained in the photon model of light?
- b) What experimental features of the photoelectric effect can be explained by classical physics? Which cannot?
- c) This problem is one of *estimating* the time lag (expected classically, but not observed) for the photoelectric effect. Assume that a point light source emits  $1 \text{ W} = 1 \text{ J/s}$  of light energy. Assuming uniform radiation in all directions, find the light intensity in  $\text{eV}/(\text{s} \cdot \text{m}^2)$  at a distance of 1 m from the light source. Assuming some reasonable size for an atom, find the energy per unit time incident on the atom for this intensity. If the work function (see book for work function definition) is 2 eV, how long does it take for this much energy to be absorbed, assuming that all of the energy hitting the atom is absorbed? **Estimate! Perfection not required.**

**5) Photons.**

(15 points)

- a) The wavelengths of visible light range from about 380 nm to about 750 nm. What is the range of photon energies (in eV) in visible light? A typical FM radio station's broadcast frequency is about 100 MHz. What is the energy of an FM photon of the frequency?
- b) The NaCl molecule has a bond energy of 4.26 eV; that is, this energy must be supplied in order to dissociate the molecule into neutral Na and Cl atoms. What are the minimum frequency and maximum wavelength of the photon necessary to dissociate the molecule? In what part of the electromagnetic spectrum is this photon?
- c) A 40-W incandescent bulb radiates from a tungsten filament operating at 3300 K. Assuming that the bulb radiates like a blackbody, what are the frequency and the wavelength at the maximum of the spectral distribution? If the maximum frequency is a good approximation of the average frequency of the photons emitted by the bulb, about how many photons is the bulb radiating per second? If you are looking at the bulb from 5 m away, how many photons enter your eye per second? (The diameter of your pupil is about 5.0 mm.)