## An Intro to Hydrostatic Equilibrium

## A Hydrostatic Haiku:

Gravity Pulls Down Down, Inexorably Down P briefly impedes

Consider the Earth's atmosphere by assuming the constituent particles comprise an ideal gas, such that  $P = nk_{\rm B}T$ , where n is the number density of particles (with units  $cm^{-3}$ ),  $k = 1.4 \times 10^{-16}$  erg K<sup>-1</sup> is the Boltzmann constant. We'll use this ideal gas law in just a bit, but first

1. Think of a small, cylindrical parcel of gas, with the axis running vertically in the Earth's atmosphere. The parcel sits a distance r from the Earth's center, and the parcel's size is defined by a height  $\Delta r \ll r$  and a circular cross-sectional area A (it's okay to use r here, because it is an intrinsic property of the atmosphere). The parcel will feel pressure pushing up from gas below  $(P_{\rm up} = P(r))$  and down from above  $(P_{\rm down} = P(r + \Delta r))$ .

Make a drawing of this, and discuss the situation and the various physical parameters with your group.

- 2. What other force will the parcel feel, assuming it has a density  $\rho(r)$  and the Earth has a mass  $M_{\oplus}$ ?
- 3. If the parcel is not moving, give a mathematical expression relating the various forces, remembering that force is a vector and pressure is a force per unit area.
- 4. Give an expression for the gravitational acceleration, g, at at a distance r above the Earth's center in terms of the physical variables of this situation.
- 5. Show that

$$\frac{dP(r)}{dr} = -g\rho(r) \tag{1}$$

This is the equation of hydrostatic equilibrium.

- 6. Now go back to the ideal gas law described above. Derive an expression describing how the density of the Earth's atmosphere varies with height,  $\rho(r)$ ? (HINT: It may be useful to recall that  $dx/x = d \ln x$ .)
- 7. Show that the height, H, over which the density falls off by an factor of 1/e is given by

$$H = \frac{kT}{\bar{m}q} \tag{2}$$

where  $\bar{m}$  is the mean (average) mass of a gas particle. This is the "scale height." First, check the units. Then do the math. Then make sure it makes physical sense, e.g. what do you think *should* happen when you increase  $\bar{m}$ ? Finally, pat yourselves on the back for solving a first-order differential equation and finding a key physical result!

8. What is the Earth's scale height,  $H_{\oplus}$ ? The mass of a proton is  $1.7 \times 10^{-24}$  g, and the Earth's atmosphere is mostly molecular nitrogen,  $N_2$ , where *atomic* nitrogen has 7 protons, 7 neutrons.