

A Mathematical Exploration of the Decay of Caffeine

A Mathematics Internal Assessment



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1 Introduction

Coffee has become an integral part of many lives, including mine. It's what wakes you up in the morning and sustains your energy-levels throughout the day. The leading chemical stimulant in coffee is called caffeine and it's classified as a "Psychoactive drug", meaning that it alters the chemical processes that goes on in your brain. The main effect of caffeine is to alter the brain's perception of when it's tired, and is the reason why I and many others admire this mild drug. And since the world average amount of sleep lies bellow the hourly optimum, it is no wonder why caffeine consumed to such extents as it is today.

Due to the psychoactive effects of caffeine on tiredness many advice to not drink caffeine-containing beverages 3 hours before bed, but how did they come to this conclusion?

The purpose of this math IA is therefore to explore the decay of caffeine with the aim of figuring out how early before bed you can dink your last cup of coffee without it affecting your sleep.

2 Data collection

Much like radioactive elements in physics, caffeine follows the same pattern that alpha decay does, meaning that it has a half-life. A Half-life in the realm of nuclear physics is defined as time taken for the radioactive activity of an element to halve itself. So if we were to have a sample of the element Plutonium-238 has an initial activity of 1 and a half life of 2 hours. Then if you then wait for 2 hours, the activity is now going to be $\frac{1}{2}$. I have therefore concluded to use computer simulations as my data collection, and since there are no available simulations specifically for caffeine I will collect data from a simple python program which will apply the principles of the half-life (Raw code is attached to the appendix).

To obtain the half-life of caffeine a search of scholarly articles have been conducted. A study preformed by the American Society of Health-System Pharmacists has

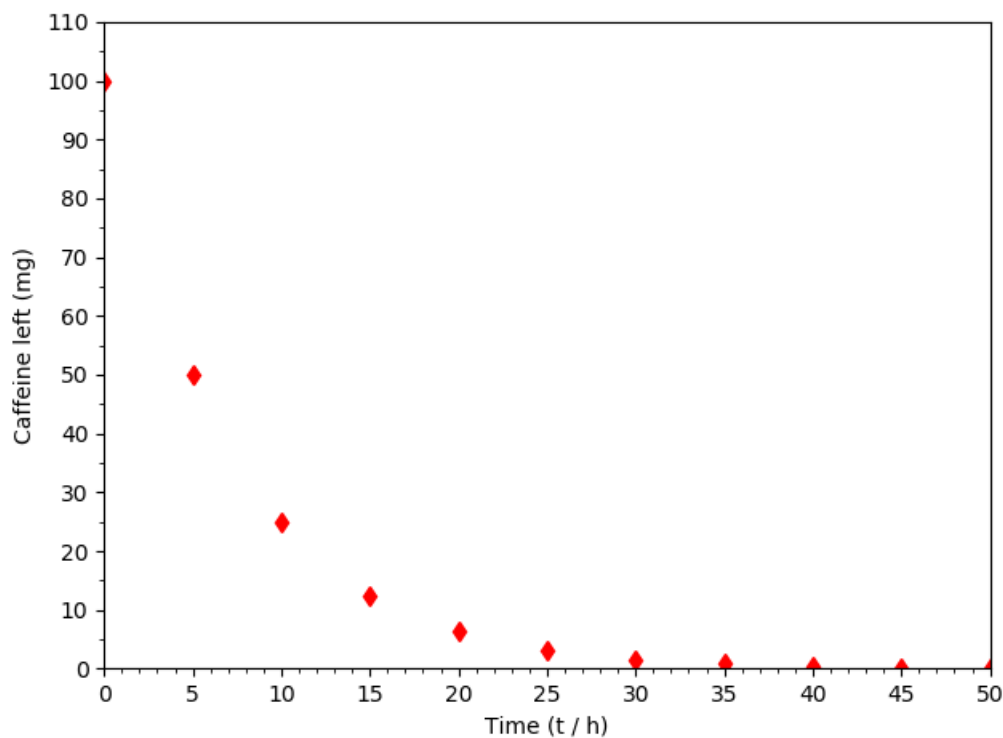
found that the half-life of caffeine lies varies between 3 to 5 hours in adults. This is subject to errors made by the original writer but for the sake of this exploration we will assume that his findings hold true.

Since the aim of this exploration is to determine how early before bed you can drink your last cup of coffee, we will take assume that the person drinking is at the lower echelon of caffeine decomposition, and caffeine will therefore have a half-life of 5 *hours*.

the table below shows the simulation output. A note to the data is that the simulation will not calculate for values down to the absolute 0, but rather down to a value of 0.01 (15 data points). This is due to the nature of Mathematics when continuously dividing. (the last output of the simulation before it returned 0.0 was 4.95×10^{-324})

Time (h)	mg of caffeine
0	100.0
1	50.0
2	25.0
3	12.5
4	6.25
5	3.13
6	1.56
7	0.781
8	0.391
9	0.195
10	0.0977
11	0.0488
12	0.0244
13	0.0122
14	0.00610

The data above can be plotted on a scatter-graph and is shown below.



But we'd like to draw a line of best fit on the graph, do do that we must use the points and data given to calculate. It's apparent that it's an exponential function due to its initial curve sloping down and getting closer and closer to zero.

3 Determining a function

As discussed in the previous section, it seems like the data follows an exponential path, and since it's a downward sloping curve it's fair to assume that its exponent is negative.

We can observe that the rate of decay is dependant on the amount of substance left. We can therefore assume that there's a proportionality between the initial amount of substance, N_0 , and the amount decayed, N .

We can express this Mathematically as follows:

$$\frac{\delta N}{\delta t} = -\lambda N_{\Theta}$$

Where $\frac{\delta N}{\delta t}$ is the change in the substance over a change in time, which is just another way of saying "How much decayed over X amount of time". This must be proportional to the initial amount of the substance, but for the proportionality to work out we need a *proportionality constant*, which in this case is λ (this is also known as the decay constant). We know all the values except the decay constant, λ . This can be mathematically solved.

By looking at our original equation we can see that it is possible to set it up as a *differential equation* and solve for