

Lecture 7 — Probability, Normalization & Wavefunction Constraints

Reading: Engel 4th ed., Chapter 2 (Sections 2.5–2.7)

Learning Objectives

- State and apply the Born interpretation of the wavefunction
 - Normalize wavefunctions in one and three dimensions
 - List the requirements for a physically acceptable wavefunction
 - Calculate probabilities from a wavefunction using the particle in a box as a model
 - Interpret the wavefunction for a free particle using wave packets
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1. The Born Interpretation (1926)

Max Born proposed that the wavefunction $\Psi(x, t)$ does not have direct physical meaning, but $|\Psi(x, t)|^2$ does:

$$P(x, t) dx = |\Psi(x, t)|^2 dx = \Psi^*(x, t) \Psi(x, t) dx$$

This is the **probability** of finding the particle between x and $x + dx$ at time t .

- $|\Psi|^2$ is the **probability density** (probability per unit length)
- Ψ itself is the **probability amplitude**

Probability in a Region

The probability of finding the particle between x_1 and x_2 :

$$P(x_1 \leq x \leq x_2) = \int_{x_1}^{x_2} |\Psi(x, t)|^2 dx$$

PIB Example: Where Is the Particle?

For the ground state of the particle in a box, $\psi_1(x) = \sqrt{2/a} \sin(\pi x/a)$, the probability of finding the particle in the left half of the box is:

$$P\left(0 \leq x \leq \frac{a}{2}\right) = \int_0^{a/2} \frac{2}{a} \sin^2\left(\frac{\pi x}{a}\right) dx = \frac{1}{2}$$

By symmetry, the particle is equally likely to be in either half. But for the $n = 2$ state, the probability density has a node at $x = a/2$, and the particle spends zero time at the center — a purely quantum-mechanical result.

[!NOTE] **Concept Check 7.1** For the PIB ground state ($n = 1$), calculate the probability of finding the particle in the middle third of the box ($a/3 \leq x \leq 2a/3$). Is it greater or less than $1/3$? Why does this make physical sense given the shape of $|\psi_1|^2$?

2. Normalization

Since the particle must be found *somewhere*, the total probability over all space must equal 1:

$$\int_{-\infty}^{\infty} |\Psi(x, t)|^2 dx = 1$$

This is the **normalization condition**.

Normalization Procedure

If a solution to the Schrödinger equation gives an unnormalized function $\psi(x)$, we find the normalization constant N :

$$N^2 \int_{-\infty}^{\infty} |\psi(x)|^2 dx = 1 \implies N = \left[\int_{-\infty}^{\infty} |\psi(x)|^2 dx \right]^{-1/2}$$

The normalized wavefunction is $\psi_{\text{norm}}(x) = N\psi(x)$.

PIB Example: Deriving the Normalization Constant

The unnormalized PIB solutions are $\psi_n(x) = A \sin(n\pi x/a)$. To normalize:

$$A^2 \int_0^a \sin^2\left(\frac{n\pi x}{a}\right) dx = A^2 \cdot \frac{a}{2} = 1 \implies A = \sqrt{\frac{2}{a}}$$

This is why the PIB wavefunctions carry the factor $\sqrt{2/a}$. Notice it depends on the box size a but not on n .

Worked Example: Gaussian Normalization

Normalize $\psi(x) = Ae^{-\alpha x^2}$ where $\alpha > 0$:

$$\int_{-\infty}^{\infty} |A|^2 e^{-2\alpha x^2} dx = |A|^2 \sqrt{\frac{\pi}{2\alpha}} = 1$$

$$\implies A = \left(\frac{2\alpha}{\pi}\right)^{1/4}$$

Three Dimensions

$$\iiint |\Psi(\mathbf{r}, t)|^2 d^3r = 1$$

3. Constraints on Acceptable Wavefunctions

Not every mathematical function qualifies as a wavefunction. Physical requirements:

The wavefunction must be:

1. **Single-valued:** ψ has only one value at each point in space
2. **Continuous:** ψ has no discontinuities
3. **Continuously differentiable:** $d\psi/dx$ is continuous (except where V is infinite)
4. **Square-integrable (normalizable):** $\int |\psi|^2 dx < \infty$
5. **Finite:** ψ does not diverge at any point

PIB Example: Seeing the Constraints in Action

The PIB solutions $\psi_n = \sqrt{2/a} \sin(n\pi x/a)$ satisfy every constraint:

- **Single-valued:** sin is single-valued ✓
- **Continuous:** ψ matches the boundary conditions $\psi(0) = \psi(a) = 0$ with the $\psi = 0$ regions outside ✓
- **Derivative discontinuity at walls:** $d\psi/dx$ is discontinuous at $x = 0$ and $x = a$ — but this is allowed because $V = \infty$ there
- **Normalizable:** $\int_0^a |\psi_n|^2 dx = 1$ ✓

The constraint of continuity at the boundaries is precisely what forces *quantization* — only specific k values satisfy $\psi(a) = 0$.

Examples of Unacceptable Functions

Function	Problem
$\psi = e^x$	Not normalizable (diverges as $x \rightarrow \infty$)
$\psi = 1/x$	Diverges at $x = 0$; not continuous
$\psi = \begin{cases} x & x > 0 \\ -2x & x < 0 \end{cases}$	Discontinuous derivative at $x = 0$

[!NOTE] **Concept Check 7.2** The function $\psi(x) = N \cos(3\pi x/a)$ satisfies the Schrödinger equation inside a box of width a . Explain why it is *not* an acceptable PIB wavefunction. Which constraint does it violate?

4. Expectation Values

The **expectation value** (average of many measurements) of an observable A with corresponding operator \hat{A} :

$$\langle A \rangle = \int_{-\infty}^{\infty} \Psi^* \hat{A} \Psi dx$$

PIB Example: Average Position

For the PIB ground state $\psi_1 = \sqrt{2/a} \sin(\pi x/a)$:

$$\langle x \rangle = \int_0^a \psi_1^* x \psi_1 dx = \frac{2}{a} \int_0^a x \sin^2\left(\frac{\pi x}{a}\right) dx$$

Using the identity $\sin^2 \theta = \frac{1-\cos(2\theta)}{2}$: $\langle x \rangle = \frac{2}{a} \int_0^a x \left(\frac{1-\cos(2\pi x/a)}{2} \right) dx = \frac{1}{a} \left[\int_0^a x dx - \int_0^a x \cos\left(\frac{2\pi x}{a}\right) dx \right]$

The first integral is trivial: $\int x dx = \frac{1}{2}x^2$. For the second, we use the **indefinite integral**: $\int x \cos(kx) dx = \frac{\cos(kx)}{k^2} + \frac{x \sin(kx)}{k}$

$$\text{Applying this with } k = 2\pi/a: \langle x \rangle = \frac{1}{a} \left[\left(\frac{a^2}{2} - 0 \right) - \left(\left[\frac{\cos(2\pi x/a)}{(2\pi/a)^2} + \frac{x \sin(2\pi x/a)}{2\pi/a} \right]_0^a \right) \right] \langle x \rangle = \frac{1}{a} \left[\frac{a^2}{2} - \left(\left(\frac{a^2}{4\pi^2} + 0 \right) - \left(\frac{a^2}{4\pi^2} + 0 \right) \right) \right] = \frac{1}{a} \left[\frac{a^2}{2} \right] = \frac{a}{2}$$

The average position is exactly at the center of the box.

$$\langle p \rangle = \int_0^a \psi_1^* (-i\hbar \frac{d}{dx}) \psi_1 dx = -i\hbar \frac{2}{a} \int_0^a \sin\left(\frac{\pi x}{a}\right) \frac{d}{dx} \left[\sin\left(\frac{\pi x}{a}\right) \right] dx$$

$$\langle p \rangle = -i\hbar \frac{2}{a} \left(\frac{\pi}{a}\right) \int_0^a \sin\left(\frac{\pi x}{a}\right) \cos\left(\frac{\pi x}{a}\right) dx$$

Using the **indefinite integral**: $\int \sin(kx) \cos(kx) dx = \frac{\sin^2(kx)}{2k}$

$$\text{Applying this with } k = \pi/a: \langle p \rangle = -\frac{2i\hbar\pi}{a^2} \left[\frac{\sin^2(\pi x/a)}{2\pi/a} \right]_0^a = -\frac{2i\hbar\pi}{a^2} \frac{a}{2\pi} [\sin^2(\pi) - \sin^2(0)] \langle p \rangle = -\frac{i\hbar}{a}(0 - 0) = 0$$

The average momentum is zero because the particle is a standing wave (equal probability of moving left and right).

Common Expectation Values

Kinetic energy: $\langle T \rangle = \int \Psi^* \left(-\frac{\hbar^2}{2m} \frac{d^2}{dx^2} \right) \Psi dx$

For PIB state ψ_n : $\langle T \rangle = E_n = \frac{n^2 \hbar^2}{8ma^2}$ (since $V = 0$ inside the box, all energy is kinetic).

Variance and Uncertainty

$$\sigma_A^2 = \langle A^2 \rangle - \langle A \rangle^2$$

$$\Delta A = \sqrt{\langle A^2 \rangle - \langle A \rangle^2}$$

5. The Free Particle and Wave Packets

Free Particle Solutions

For $V = 0$, the TISE gives:

$$\psi_k(x) = Ae^{ikx}, \quad E = \frac{\hbar^2 k^2}{2m}$$

These are plane waves with definite momentum $p = \hbar k$ but completely delocalized (not normalizable over all space).

Wave Packets

A **wave packet** — a superposition of many plane waves — can represent a localized particle:

$$\Psi(x, t) = \frac{1}{\sqrt{2\pi}} \int_{-\infty}^{\infty} \phi(k) e^{i(kx - \omega(k)t)} dk$$

where $\phi(k)$ determines the distribution of momenta.

Gaussian Wave Packet

A Gaussian momentum distribution $\phi(k) \propto e^{-(k-k_0)^2/(2\sigma_k^2)}$ gives a spatial wave packet:

$$|\Psi(x, 0)|^2 \propto e^{-(x-x_0)^2/(2\sigma_x^2)}$$

with $\sigma_x \cdot \sigma_k = 1/2$, or equivalently $\Delta x \cdot \Delta p = \hbar/2$ — the minimum uncertainty product.

Group Velocity vs. Phase Velocity

- **Phase velocity:** $v_p = \omega/k$ (speed of individual wavefronts)
- **Group velocity:** $v_g = d\omega/dk$ (speed of the wave packet envelope = speed of the particle)

For a free particle: $\omega = \hbar k^2/(2m)$, so $v_g = \hbar k/m = p/m = v_{\text{classical}}$. The wave packet moves at the classical particle velocity.

[!NOTE] **Concept Check 7.3** Explain why a "wave packet" is necessary to represent a localized particle, whereas a single plane wave e^{ikx} describes a particle with a perfectly defined momentum but unknown position.

Key Equations Summary

Equation	Expression
Born interpretation	$P(x, t) dx = \ \Psi(x, t)\ ^2 dx$
Normalization	$\int_{-\infty}^{\infty} \ \Psi\ ^2 dx = 1$
Expectation value	$\langle A \rangle = \int \Psi^* \hat{A} \Psi dx$
Uncertainty	$\Delta A = \sqrt{\langle A^2 \rangle - \langle A \rangle^2}$
Group velocity	$v_g = d\omega/dk$
Min. uncertainty	$\Delta x \cdot \Delta p = \hbar/2$ (Gaussian)

Recent Literature Spotlight

"Logical Quantum Processor Based on Reconfigurable Atom Arrays" D. Bluvstein, S. J. Evered, A. A. Geim, S. H. Li, et al., Nature, 2024, 626, 58–65. DOI

This landmark paper demonstrates a programmable quantum processor that encodes and manipulates logical qubits using reconfigurable atom arrays. Each atom is effectively a "particle in a trap" — confined in optical tweezers with quantized energy levels, just like our particle in a box. The Born interpretation governs the readout: the probability of measuring each qubit state is given by $|c_0|^2$ and $|c_1|^2$, directly applying the concepts from this lecture.

Practice Problems

1. **Normalization.** Normalize the wavefunction $\psi(x) = A x e^{-\alpha x}$ for $x \geq 0$ (and $\psi = 0$ for $x < 0$), where $\alpha > 0$.
 2. **PIB probability.** For a particle in the $n = 2$ state of a box of width a , calculate the probability of finding the particle between $x = 0$ and $x = a/4$.
 3. **Acceptable wavefunctions.** Which of the following are acceptable wavefunctions on $(-\infty, \infty)$? Explain.
(a) $\psi = e^{-|x|}$ (b) $\psi = \tan(x)$ (c) $\psi = (x^2 + 1)^{-1}$
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Next lecture: Postulates I–III — State Functions, Operators & Measurement