

A section of the periodic table showing the first 18 elements is below

${}_1\text{H}$

${}_3\text{Li}$	${}_4\text{Be}$	${}_5\text{B}$	${}_6\text{C}$	${}_7\text{N}$	${}_8\text{O}$	${}_9\text{F}$	${}_2\text{He}$
${}_{11}\text{Na}$	${}_{12}\text{Mg}$	${}_{13}\text{Al}$	${}_{14}\text{Si}$	${}_{15}\text{P}$	${}_{16}\text{S}$	${}_{17}\text{Cl}$	${}_{10}\text{Ne}$
							${}_{18}\text{Ar}$

Using only the symbol, write down the symbols for an element which forms an ion with 1 negative charge.