Aspirin ($C_9H_8O_4$), can be made in the reaction below.

$$C_7H_6O_3 + C_4H_6O_3 \rightarrow C_9H_8O_4 + C_2H_4O_2$$

(Formula masses: $C_7H_6O_3 = 138$ $C_4H_6O_3 = 102$ $C_9H_8O_4 = 180$ $C_2H_4O_2 = 60$) Use the equation below to calculate the atom economy for the production of aspirin. Atom economy =

mass of required product ×100% total mass of products