

Heat Flux

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1 Gas Properties

Combustion gas properties are commonly calculated using NASA's Chemical Equilibrium with Applications (CEA) program, however this FORTRAN program is difficult to work with. Instead the combustion gas properties were calculated using Cantera as described by Youngblood.

Species included in the chemical equilibrium calculations were based off of those used in CEA and kerosene combustion models as well as the availability of transport properties. The species that were included in this analysis are: RP-1, LOX, CO, CO₂, H, H₂, H₂O, O, OH, and O₂. RP-1 and LOX were modelled with the same stoichiometry and heat of formation as used in CEA. Cantera requires transport parameters for all species, so RP-1 and LOX were assigned placeholder values with the understanding that equilibrium solutions contain essentially zero concentration of these species.

Thermodynamic properties and Lennard-Jones transport parameters were taken from a high-temperature version of the GRI MECH 3.0 combustion mechanism included with Cantera that has been extended from an upper temperature of 3500K to 6000K using thermodynamic properties from NASA.

1.1 Property Evaluation

The procedure used to calculate the gas properties differs significantly from CEA. Instead of using a user-specified chamber pressure, the properties are calculated from engine geometry and flow rates.

The first step is to calculate the stagnation and throat properties, which are solved for together. At the throat of the engine, the following two conditions must be met:

$$v = a \tag{1}$$

$$h_0 = h + \frac{v^2}{2} \tag{2}$$

The velocity and sound speed are calculated as follows:

$$v = \frac{\dot{m}}{A\rho} \tag{3}$$

$$a = \sqrt{\left(\frac{\partial p}{\partial \rho}\right)_s} \tag{4}$$

The partial derivative required to calculate the sound speed is evaluated by perturbing the pressure at constant entropy. If equilibrium expansion is used, the perturbed gas is equilibrated.

For a given stagnation pressure, the stagnation gas properties can be fully calculated as described by Youngblood. The throat conditions can then be solved for by varying the throat pressure isentropically

until condition 1 is met. If condition 2 is met, the stagnation pressure is correct, if not, another stagnation pressure must be guessed. Two root finders were used to solve for the chamber and throat properties in this manner.

With the stagnation enthalpy known, the conditions at all other points of the engine can be solved for by varying the pressure isentropically until condition 2 is met. Both subsonic and supersonic solutions exist; specific solutions are forced by using the throat pressure to limit the solution range.

1.2 Frozen Boundary Layer Chemistry

CEA calculates “frozen” and “equilibrium” values for specific heat and thermal conductivity. Equilibrium values are taken to be the sum of the frozen value and a reaction term that accounts for chemical reactions occurring in the boundary layer. Equilibrium values for both are several times higher than frozen, and using the equilibrium values gives a calculated heat flux approximately 4.5 times higher (based on the Bartz equation).

Bartz **page 46** discusses chemical reactions occurring in rocket engine boundary layers and the resulting increase in heat transfer. When dissociated species such as O, H, and OH are present in the free stream, they can cool against combustion walls and recombine exothermically in the boundary layer. Bartz notes that this effect would likely be maximum for a hydrogen and oxygen rocket at 100% combustion efficiency, yet the increase in heat flux over a frozen layer is shown to be only 32% and rapidly decreases with reduced combustion efficiency.

The presence of dissociated species is not insignificant for this engine, as can be seen in figure ???. It is unclear why equilibrium values calculated by CEA are so high and reproducing them is difficult, so based off of the Bartz information the boundary layer was considered to be chemically frozen.

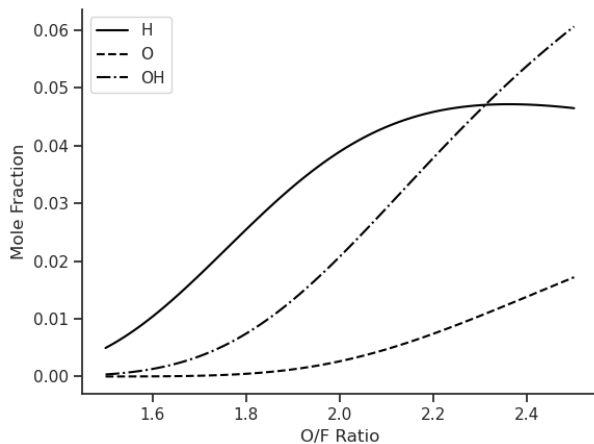


Figure 1: Chamber dissociated species.

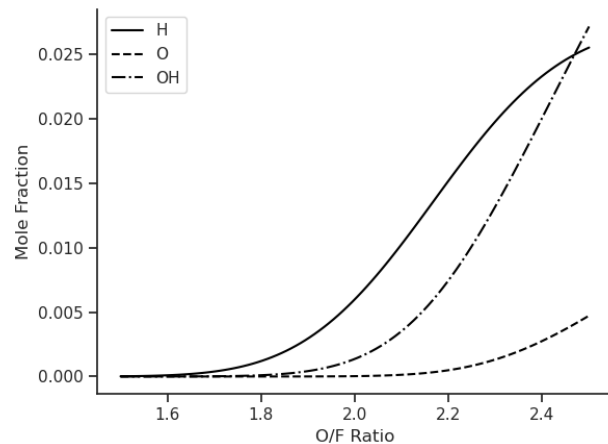


Figure 2: Nozzle dissociated species.

1.3 Frozen Contraction and Expansion

Through the contraction and expansion of the engine, combustion gas can be considered to either be in equilibrium (mole fractions change along length of engine) or the composition could be frozen. Both methods were implemented, but this was found to make little difference to heat flux calculations.

1.4 Property Comparison

Table 1: Cantera and CEA property comparison, OF = 2.29.

	Stagnation		Throat		Exit	
	Cantera	CEA Diff.	Cantera	CEA Diff.	Cantera	CEA Diff.
T (K)	3307.3	−0.09%	3135.7	0.23%	2660.9	1.16%
C_p (10^3 J/kg/K)	2.0766	−0.05%	2.0665	−0.04%	2.0291	0.06%
ρ (10^{-1} kg/m ³)	6.2405	0.06%	3.7268	3.26%	0.8617	10.5%
γ	1.2256	−4.09%	1.2228	−4.63%	1.2185	−5.23%
μ (10^{-5} Pa s)	9.4636	8.03%	9.1093	8.42%	8.0947	9.60%
k (10^{-1} W/m/K)	3.5662	−3.87%	3.3775	−3.33%	2.8629	−1.72%

Table 2: Cantera and CEA property comparison, OF = 2.29.

	Stagnation		Throat		Exit	
	Cantera	CEA Diff.	Cantera	CEA Diff.	Cantera	CEA Diff.
P (Bar)	8.3785	N/A	4.8249	0.37%	0.99181	0.01%
T (K)	3314.8	−0.09%	3151.2	−0.06%	2697.0	−0.09%
C_p (10^3 J/kg/K)	2.0771	−0.05%	2.0675	−0.07%	2.0325	0.09%
ρ (10^{-1} kg/m ³)	6.6174	0.06%	4.0651	0.39%	1.0093	0.07%
μ (10^{-5} Pa s)	9.4775	8.06%	9.1404	8.21%	8.1726	8.64%
k (10^{-1} W/m/K)	3.5689	−3.79%	3.3898	−3.55%	2.8982	−2.86%

All major species were found to have less than 1% difference in mole fractions between Cantera and CEA throughout the engine. Stagnation thermodynamic properties match very well, which is expected as GRI MECH uses the same thermodynamic data as CEARUN. Transport properties are consistently different; CEA uses empirical fits but Cantera uses the Lennard-Jones model so this isn't unexpected.

2 Heat Transfer Correlations

It appears to be accepted to use the difference between the adiabatic wall temperature and the wall temperature as the driving potential for heat flux, so that was adopted here as well. The adiabatic wall temperature is given by:

$$T_{aw} = T + r(T_0 - T) \quad (5)$$

The recovery factor r is commonly given as

$$r = (Pr)^{1/3} \quad (6)$$

The gas wall temperature was taken to be a constant value of 500K

Many models require property evaluation at temperatures other than the free stream temperature. For these models, the gas temperature was changed in Cantera while keeping the pressure constant.

The largest difference was between the Dittus-Boelter and Bartz models, at 30%. This was found to be because of the difference in constants (11%), the throat curvature factor (10%), and the breakdown of the assumption of constant Prandtl number and specific heat (9%).

2.1 Bartz

The Bartz equation is commonly used for calculating rocket engine heat flux. The following form is common:

$$h_g = \left[\frac{0.026}{D_t^{0.2}} \left(\frac{\mu^{0.2} c_p}{Pr^{0.6}} \right)_0 \left(\frac{p_0}{c^*} \right)^{0.8} \left(\frac{D_t}{R_c} \right)^{0.1} \right] \left(\frac{A_t}{A} \right)^{0.9} \sigma \quad (7)$$

$$\sigma = \frac{1}{\left[\frac{1}{2} \left(\frac{T_w}{T_0} \right) \left(1 + \frac{\gamma-1}{2} M^2 \right) + \frac{1}{2} \right]^{0.8-m/5} \left[1 + \frac{\gamma-1}{2} M^2 \right]^{m/5}}$$

Here m is the exponent of the temperature dependence of viscosity, and it is commonly taken to be 0.6. The σ term accounts for property variation in the boundary layer and along the length of the engine.

This form shows the pressure dependence of heat flux at a constant characteristic velocity, however if the equation is rearranged into nondimensional parameters the fact that it is a modified version of the Dittus-Boelter equation is more clear.

$$Nu_t = 0.026 Re_t^{0.8} Pr^{0.4} \left(\frac{D_t}{R_c} \right)^{0.1} \left(\frac{A_t}{A} \right)^{0.9} \sigma \quad (8)$$

Here all gas properties are evaluated at stagnation and the throat diameter is the characteristic dimension.

Another form given by Bartz also uses the stagnation gas properties but the characteristic dimension is the local diameter. It has a boundary layer property variation term that uses the gas viscosity and density at the arithmetic mean of the free stream and wall temperatures:

$$Nu_D = 0.026 Re_D^{0.8} Pr^{0.4} \left(\frac{D_t}{R_c} \right)^{0.1} \left[\left(\frac{\rho_{am}}{\rho} \right)^{0.8} \left(\frac{\mu_{am}}{\mu} \right)^{0.2} \right] \quad (9)$$

2.2 Dittus-Boelter

The unmodified Dittus-Boelter equation the Bartz equation is based on is given by:

$$Nu_D = 0.023 Re_D^{0.8} Pr^{0.4} \quad (10)$$

The Dittus-Boelter equation is known to be ill-suited to modelling flows with large property variations in the boundary layer. A proposed solution is to evaluate all properties at the arithmetic mean of the free stream and wall temperatures. Another solution is to evaluate all properties at the reference temperature given by:

$$T_{ref} = 0.5(T + T_w) + 0.22 Pr^{1/3} (T_0 - T) \quad (11)$$

2.3 Sieder-Tate

$$Nu_D = 0.027 Re_D^{0.8} Pr^{1/3} \left(\frac{\mu}{\mu_w} \right)^{0.14} \quad (12)$$

2.4 Colburn

$$Nu_x = 0.0296 Re_x^{0.8} Pr^{1/3} \quad (13)$$

$$x_e = 3.53D \left[1 + \left(\frac{x}{3.53D} \right)^{-1.2} \right]^{-1/1.2} \quad (14)$$

3 OF Ratio

3.1 Constant Pressure

3.2 Changing Pressure

4 Experimental Data

5 Conclusion

Pintle recirculation zones

Film cooling