

Thermo Quiz 7

1.)

$$dU = TdS - PdV$$

$$dH = TdS + VdP$$

$$\frac{dU}{T} = dS = \frac{C_v dT}{T} \Rightarrow \Delta S = C_v \ln\left(\frac{T_2}{T_1}\right)$$

$$\Delta U = Q + W_{EC} + W_s \therefore \Delta U = 0 = \Delta U_1 + \Delta U_2$$

$$\therefore m_1 C_1 (T_2 - T_1)_1 + m_2 C_2 (T_2 - T_1)_2 = 0$$

$$m_1 C_1 T_2 - m_1 C_1 T_{11} + m_2 C_2 T_2 - m_2 C_2 T_{12}$$

$$m_1 C_1 T_2 + m_2 C_2 T_2 = m_1 C_1 T_{11} + m_2 C_2 T_{12}$$

$$T_2 = \frac{m_1 C_1 T_{11} + m_2 C_2 T_{12}}{m_1 C_1 + m_2 C_2}$$

2.)

$$\Delta S = C_v \ln\left(\frac{T_2}{T_1}\right) = C_1 \ln\left(\frac{T_2}{T_{11}}\right) (m_1) + C_2 m_2 \ln\left(\frac{T_2}{T_{12}}\right)$$

3.)

$$\Delta S = S_{gen} + \int \frac{\delta Q}{T} \therefore \Delta S = S_{gen}$$

4.)

$$V_1 = 0.08 \text{ m}^3$$

$$V_2 = 0.001 \text{ m}^3$$

$$P = 100 \text{ kPa} \quad T_1 = 300 \text{ K} = T_2$$

$$P_2 = 0.001 \text{ m}^3$$

$$dU = TdS - PdV$$

$$dH = TdS + VdP$$

$$\Delta U = Q + W_{EC} + W_s \quad W_{EC} = -\int_1^2 PdV$$

$$dU = TdS$$

$$dU = C_v dT$$

$$C_v dT + PdV = Q$$

$$\text{Reversible: } S_{gen} = 0$$

5.) $T_H = 1200\text{K}$ $Q_H = 500 \frac{\text{kJ}}{\text{s}} = \text{kW}$
 $T_C = 300\text{K}$
 $\eta_{\theta} = 1 - \frac{T_C}{T_H} = 0.75$