

HW4_3

February 23, 2022

1 HW 4

1.1 Problem 3.a

Calculate liquid-phase mass transfer coefficient and film thickness

Using film theory,

$$k_c = \frac{D_{AB}}{\delta} \quad (1)$$

$$k'_c = \frac{D_{AB}}{\delta(1-x_A)_{LM}} = \frac{k_c}{(1-x_A)_{LM}} \quad (2)$$

$$N_A = k'_c(c_{A_i} - c_{A_b}) \quad (3)$$

where

$$(1-x_A)_{LM} = \frac{x_{A_i} - x_{A_b}}{\ln\left(\frac{1-x_{A_b}}{1-x_{A_i}}\right)}$$

x_{A_i} can be estimated using Henry's law, $x = \frac{p}{H}$ c can be estimated by the relationship $c = \frac{\rho}{M} x_{A_b}$ is assumed to be negligible.

To find k_c , (2) and (3) can be combined and rearranged to get

$$k_c = \frac{N_A}{c_{A_i}}(1-x_A)_{LM} \quad (4)$$

The following script solves (4) using values given in the problem statement.

```
[ ]: import numpy as np
NA = 0.017*4.536e39/3600/929.03 #convert flux to mol/cm2/sec
DAB = 2e-5 # diffusivity cm2/sec
p = 150 # partial press co2 psia
H = 9000 # henry constant psia
xai = p/H #mol frac @ interface
ro = .997 #density of water g/cm3
```

```

M = 18.015                                     #molar mass of  $\text{H}_2\text{O}$ 
    ↪ water g/mol
c = ro/M                                       #concentration mol/
    ↪ cm3
cai = c*xai                                  #concentration of  $\text{H}_2\text{O}$ 
    ↪ co2 at interface

```

```

[ ]: def LM(x1,x2):
      return (x1-x2)/np.log((1-x2)/(1-x1))

```

```

[ ]: def kc(na,cai,lm):
      return na/cai*lm
kc = kc(NA,cai,LM(xai,0))
print(kc)

```

2.4787662763132446e+34

$k_c = 2.479e34 \frac{\text{cm}}{\text{sec}}$

(1) can now be used to solve for δ

```

[ ]: def delta():
      return DAB/kc
print(delta())

```

8.068529974414005e-40

$\delta = 8.069e - 40 \text{cm}$

1.2 Problem 3.b

Determine contact time using Higbie model,

$$k_c = 2\sqrt{\frac{D_{AB}}{\pi t_c}}$$

or

$$t_c = \frac{4D_{AB}}{\pi k_c^2}$$

```

[ ]: print(4*DAB/np.pi/kc)

```

1.0273171431304901e-39

$t_c = 1.027e - 39 \text{sec}$

1.3 Problem 3.c

Determine S from surface renewal model,

$$k_c = \sqrt{D_{AB}S}$$

or

$$s = \frac{k_c^2}{D_{AB}}$$

```
[ ]: print(kc**2/DAB)
```

```
3.072141126293914e+73
```

$s = 3.072e73 cm^{-1}$