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09 Feb 2022

### HW 3

$$P_1 = P_2 = 101 \text{ kPa}$$

$$T_1 = T_2 = 25^\circ\text{C}$$

1<sup>st</sup> order in A, B



$$y_c^{\text{sat}} = 0.25$$

$$F_{A0} = 1 \frac{\text{mol}}{\text{sec}}$$

$$F_{B0} = 2 \frac{\text{mol}}{\text{sec}} = 2F_{A0}$$

$$\dot{V} = 1.5 \frac{\text{L}}{\text{min}}$$

$$K = 0.035 \frac{\text{L}}{\text{mol} \cdot \text{sec}}$$

a) First order in A, B  $\therefore -r_A = K C_A C_B$

If elementary then  $-r_A = K C_A C_B^2$

$\therefore$  this is not an elementary rxn/step

Species	I	Δ	F (before Condense)
A	$F_{A0}$	$-X_A F_{A0}$	$F_{A0}(1-X_A)$
B	$2F_{A0}$	$-2X_A F_{A0}$	$2F_{A0}(1-X_A)$
C	0	$+X_A F_{A0}$	$X_A F_{A0}$
D	0	$+X_A F_{A0}$	$X_A F_{A0}$
Total	$3F_{A0}$	$-X_A F_{A0}$	$3F_{A0}(1-X_A) + 2X_A F_{A0}$ $= F_{A0}(3-X_A)$

c)  $P_{\text{vap}} = y_c \cdot P_{\text{tot}} = 0.25 \cdot 101 \text{ kPa} = 25.3 \text{ kPa}$

$$y_c = \frac{F_C}{F_{\text{tot}}} = \frac{X_A F_{A0}}{F_{A0}(3-X_A)} \rightarrow X_A = \frac{3y}{1+y} = 0.6$$



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1) d) Finish table:

F (after Condens.) - C is only one that Condenses

A  $F_{A0}(1-X_A)$

B  $2F_{A0}(1-X_A)$

C  $\gamma_c F_{tot}$

D  $X_A F_{A0}$

tot:  $3F_{A0}(1-X_A) + X_A F_{A0} + \gamma_c F_{tot} = \gamma_c F_T + 3F_{A0} - 2F_{A0}X_A$

$-r_A = C_A C_B$

$$C_i = \frac{C_{A0} \left( \theta_j - \frac{\gamma_j X_A}{\gamma_A} \right)}{1 + \epsilon X_A} \quad \therefore C_A = \frac{C_{A0} (1-X_A)}{1 - \frac{1}{3}X_A}$$

$$\epsilon = \gamma_{A0} \delta = \frac{1}{3} (1 + 1 - 2\frac{1}{3}) = -\frac{1}{3}$$

$$C_B = \frac{C_{A0} (2 - 2X_A)}{1 - \frac{1}{3}X_A}$$

$$\therefore -r_{A \text{ before}} = 2K C_{A0}^2 \left( \frac{(1-X_A)^2}{(1-\frac{1}{3}X_A)^2} \right)$$



HW 3

1) d)

$$C_A = \frac{F_A}{\dot{V}}$$

$$C_B = \frac{F_B}{\dot{V}}$$

where  $\dot{V} = \dot{V}_0 \left( \frac{F_T}{F_{T0}} \right)$

$$F_T = \gamma_c F_T + 3F_{A0} - 2F_{A0} X_A$$

$$\rightarrow F_T = \frac{2F_{A0}(1.5 - X_A)}{1 - \gamma_c}$$

$$\dot{V} = \dot{V}_0 \left( \frac{2F_{A0}(1.5 - X_A)}{3F_{A0}(1 - \gamma_c)} \right) = \frac{\dot{V}_0 2 \cdot (1.5 - X_A)}{3(1 - \gamma_c)}$$

$$\therefore C_A = \frac{F_{A0}(1 - X_A) \cdot 3(1 - \gamma_c)}{2\dot{V}_0(1.5 - X_A)}$$

$$C_B = \frac{2F_{A0}(1 - X_A) \cdot 3(1 - \gamma_c)}{2\dot{V}_0(1.5 - X_A)}$$

$$\therefore -r_A = \frac{K 4.5 C_{A0}^2 (1 - \gamma_c)^2 (1 - X_A)^2}{(1.5 - X_A)^2}$$

↑  
after condense

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HW #3

1) e)  $V = \frac{F_A \cdot X_A}{-r_A}$

$$\frac{(1-x)(1-x)}{(1-x)(1-x)} = \frac{(1-x)(1-x)}{(1-x)(1-x)}$$

↑  
simplify



## Hw 3

2) a) Given  $\bar{K}_f$ ,  $K_f$ , and  $T$ , find $E_A$ ,  $A$ ,  $\Delta H_{rxn}$ for  $E_A$ 

$$\bar{K}_f(T) = K_f(T_0) \exp\left(\frac{+E}{R} \left(\frac{1}{T_0} - \frac{1}{T}\right)\right)$$

$$\rightarrow E_A = R \ln \left( \frac{K_f(T)}{K_f(T_0)} \right) \cdot \frac{1}{\frac{1}{T_0} - \frac{1}{T}}$$

Using  $T_0 = 30^\circ$ 

- Solve for each set of data and average!

$$E_A = 111. \frac{\text{kJ}}{\text{mol}}$$

$$\text{for } A \quad K_f(T) = A \exp\left(\frac{-E}{RT}\right)$$

$$\rightarrow A = \frac{K_f(T)}{\exp\left(\frac{-E}{RT}\right)}$$

$$\Rightarrow 2.42 \times 10^{12}$$

for  $\Delta H_{rxn}$ 

$$\bar{K}(T) = \bar{K}(T_0) \exp\left(\frac{\Delta H}{R} \left(\frac{1}{T_0} - \frac{1}{T}\right)\right)$$

$$\rightarrow \Delta H_{rxn} = R \ln \left( \frac{\bar{K}(T)}{\bar{K}(T_0)} \right) \cdot \frac{1}{\frac{1}{T_0} - \frac{1}{T}}$$

$$\Rightarrow -63.0 \frac{\text{kJ}}{\text{mol}}$$