

H W #2

1) a) $q_x'' = \frac{-k}{L} (T_2 - T_1) = \frac{-100 \frac{W}{mK}}{0.1 m} (600 - 400) K = \boxed{-200000 \frac{W}{m^2}}$

$\frac{dT}{dx} = \frac{T_2 - T_1}{x_2 - x_1} = \frac{600 - 400}{0.1} = \boxed{2000 \frac{K}{m}}$

b) $q_x'' = \frac{-100}{0.1} (T_1 - T_2) = \frac{-100}{0.1} (400 - 600) = \boxed{200000 \frac{W}{m^2}}$

$\frac{dT}{dx} = \frac{T_1 - T_2}{0.1} = \frac{400 - 600}{0.1} = \boxed{-2000 \frac{K}{m}}$

c) $q_x'' = \frac{-100}{0.1} (600 - 400) = \boxed{-200000 \frac{W}{m^2}}$

$\frac{dT}{dx} = \frac{600 - 400}{0.1} = \boxed{2000 \frac{K}{m}}$

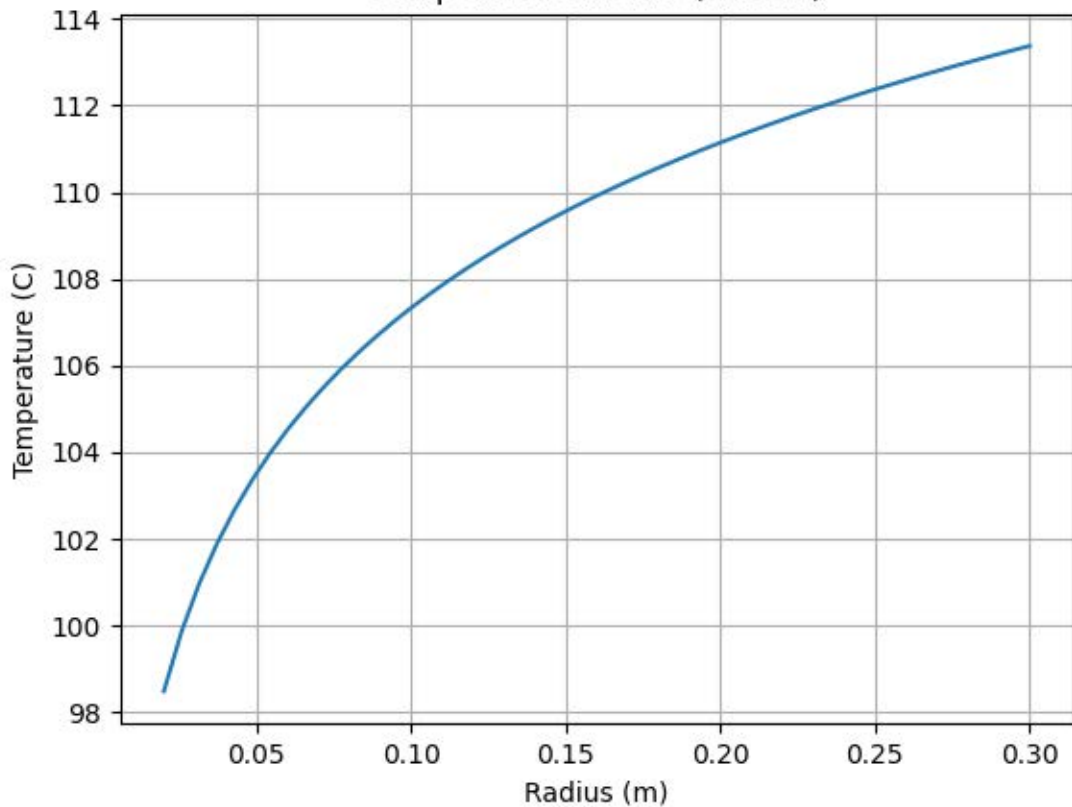
2 a) $q_r'' = -k \frac{dT}{dr} = -k \cdot \frac{5.5}{r} \quad @ 0.2 m$

$q_r'' = -2 \cdot \frac{5.5}{0.2} = \boxed{-55.0 \frac{W}{m^2}}$

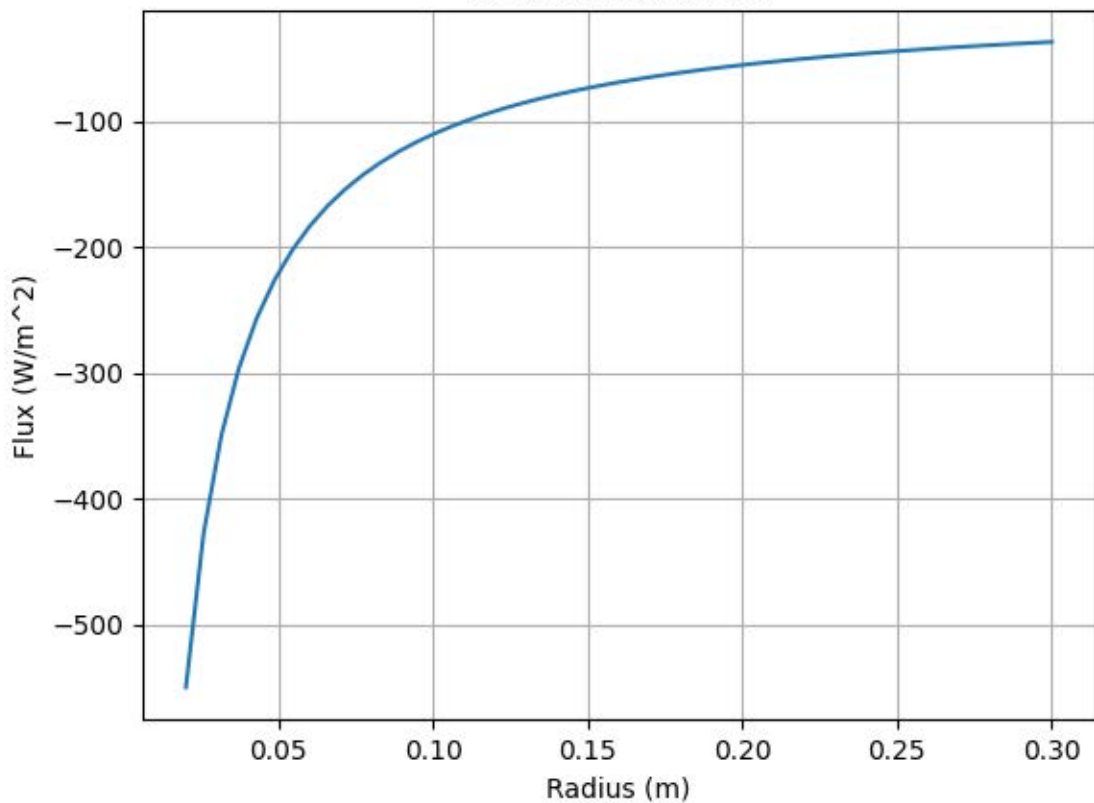
b) $q_r = q_r'' A = q_r'' \cdot 2\pi r L = -55 \cdot 2\pi (0.2) \cdot 1 = \boxed{-69.1 W}$

d) Heat flows towards origin

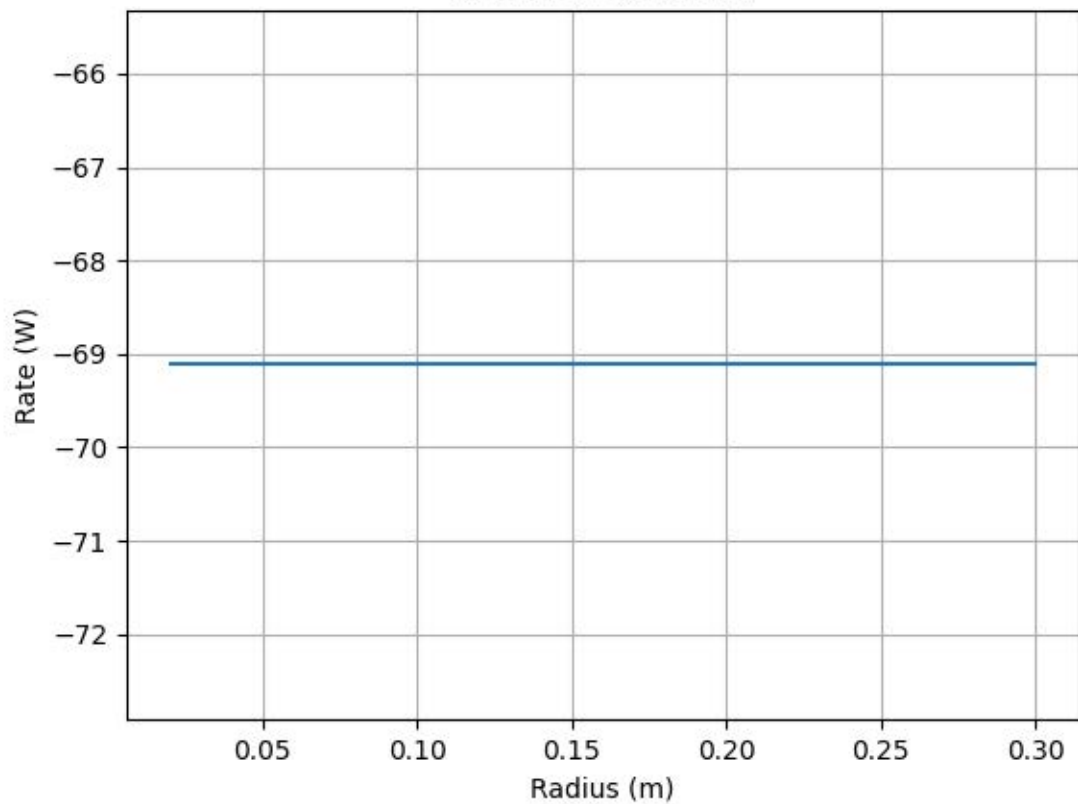
Temperature Profile (HW6.2)



Heat Flux (HW6.2)



Heat Rate (HW6.2)



HW #2

3) $r = 0.30 \text{ m}$ $k = 5 \frac{\text{W}}{\text{mK}}$ $T(r) = 500 - 100r^2 - 275r$

$$\frac{dT}{dr} = -200r - 275 \quad q_r'' = -k \frac{dT}{dr} \quad q_r'' A = q_r$$

$$q_r'' = -5 \frac{\text{W}}{\text{mK}} (-200(0.3) - 275) = 1675 \frac{\text{W}}{\text{m}^2}$$

$$q_r = q_r'' \cdot 4\pi r^2 = 1675 \cdot 4\pi (0.3)^2 = 1.89 \times 10^3 \text{ W}$$

4) $A = (0.3 \text{ m})^2 = 0.09$ $\Delta T = 80$ $L = 0.01 \text{ m}$

Poly: $q_x'' = - \frac{0.21 \frac{\text{W}}{\text{mK}} (80) \text{K}}{0.01 \text{ m}} = -1680 \frac{\text{W}}{\text{m}^2}$

$$q_x = -1680 \cdot 0.09 \text{ m}^2 = -151.2 \text{ W} \rightarrow -0.1512 \text{ kW}$$

$$\rightarrow -0.1512 \text{ kW} \cdot 8 \text{ hrs} \cdot 130 \text{ windows} = \$157.7$$

loosing

Aerogel: $q_x'' = - \frac{0.014 \frac{\text{W}}{\text{mK}} (80) \text{K}}{0.01 \text{ m}} \cdot 0.09 \text{ m}^2 = -10.08 \text{ W}$

$$= -0.01008 \text{ kW} \cdot 8 \text{ hrs} \cdot 130 \text{ windows} = \$-10.5$$

Glass: $k = 1.4$

$$q_x = \frac{-1.4 (80)}{0.01} \cdot 0.09 = -1008 \text{ W} = -1.008 \text{ kW}$$

$$-1.008 \text{ kW} \cdot 8 \text{ hrs} \cdot 130 \text{ windows} = \$-1048.$$

~~\$-10~~ because loosing money

HW#2

$$5.) a) q'' = \frac{-50 \frac{W}{mK}}{0.25 m} (-20 - 50) K = \boxed{14000 \frac{W}{m^2} = q''}$$

$$\frac{dT}{dx} = \frac{-20 - 50}{0.25} = \boxed{-280 \frac{K}{m}}$$

$$b) q'' = \frac{-50}{0.25} (-10 + 30) = \boxed{-4000 \frac{W}{m^2} = q''}$$

$$\frac{dT}{dx} = \frac{-10 + 30}{0.25} = \boxed{80 \frac{K}{m}}$$

$$c) \frac{dT}{dx} = 160 = \frac{T_2 - 70}{0.25} \rightarrow T_2 = 0.25(160) + 70 = \boxed{110^\circ C = T_2}$$

$$q'' = \frac{-50}{0.25} (110 - 70) = \boxed{-8000 \frac{W}{m^2} = q''}$$

$$d) \frac{dT}{dx} = -80 = \frac{40 - T_1}{0.25} \rightarrow T_1 = -[0.25(-80) - 40] = \boxed{60^\circ C}$$

$$q'' = \frac{-50}{0.25} (-20) = \boxed{4000 \frac{W}{m^2}}$$

$$e) \frac{dT}{dx} = 200 = \frac{30 - T_1}{0.25} \rightarrow T_1 = -[0.25(200) - 30] = \boxed{20^\circ C}$$

$$q'' = \frac{-50}{0.25} (30 - 20) = \boxed{-2000 \frac{W}{m^2}}$$

Case	heat flux direction
1	left \rightarrow right
2	right \rightarrow left
3	right \rightarrow left
4	left \rightarrow right
5	right \rightarrow left

HW #2

6.) b.) $q_r'' = -k \nabla T = -k \frac{dT}{dr}$

$$T(r) = \frac{200 - 50}{\ln(0.1/0.5)} \ln\left(\frac{r}{0.5}\right) + 50$$

$$\frac{dT}{dr} = \frac{150}{\ln(0.2)} \cdot \frac{1}{r} = \frac{-93.2}{r}$$

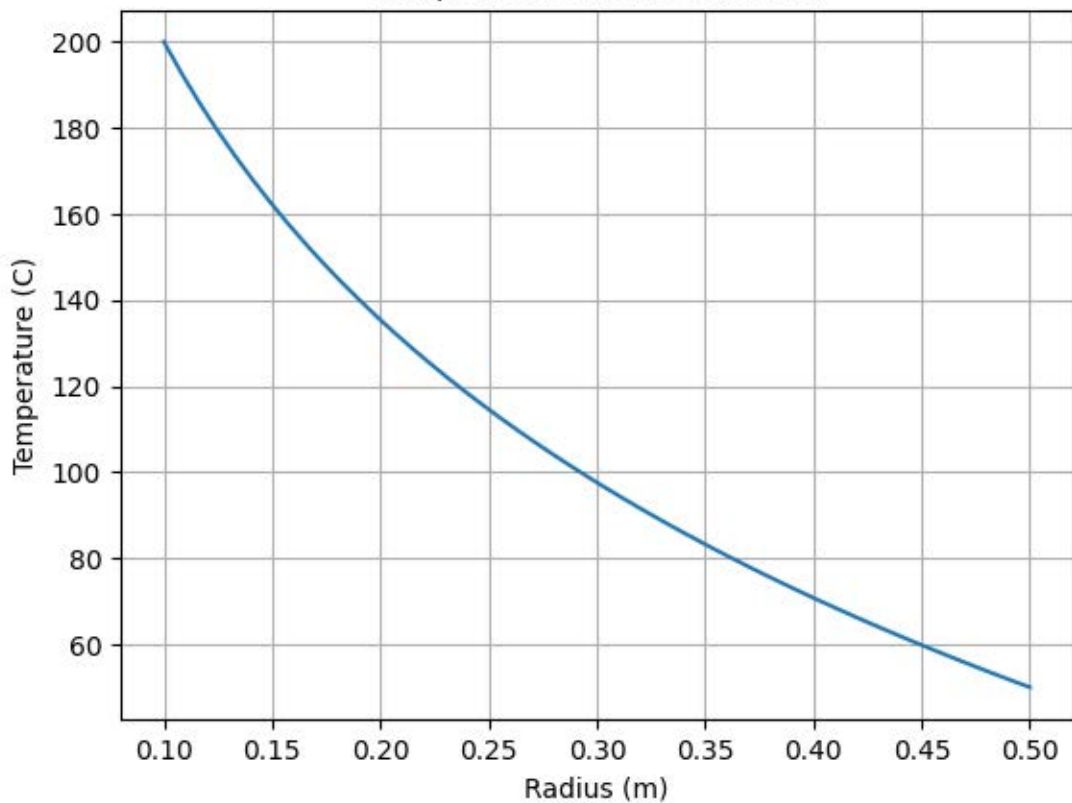
$$\therefore q_r'' = -2.4 \frac{\text{W}}{\text{mK}} \cdot \frac{-93.2}{r}$$

d) $q_r = q_r'' \cdot 2\pi r L = -2.4 \frac{\text{W}}{\text{mK}} \cdot \frac{-93.2}{r} \cdot 2\pi r L$

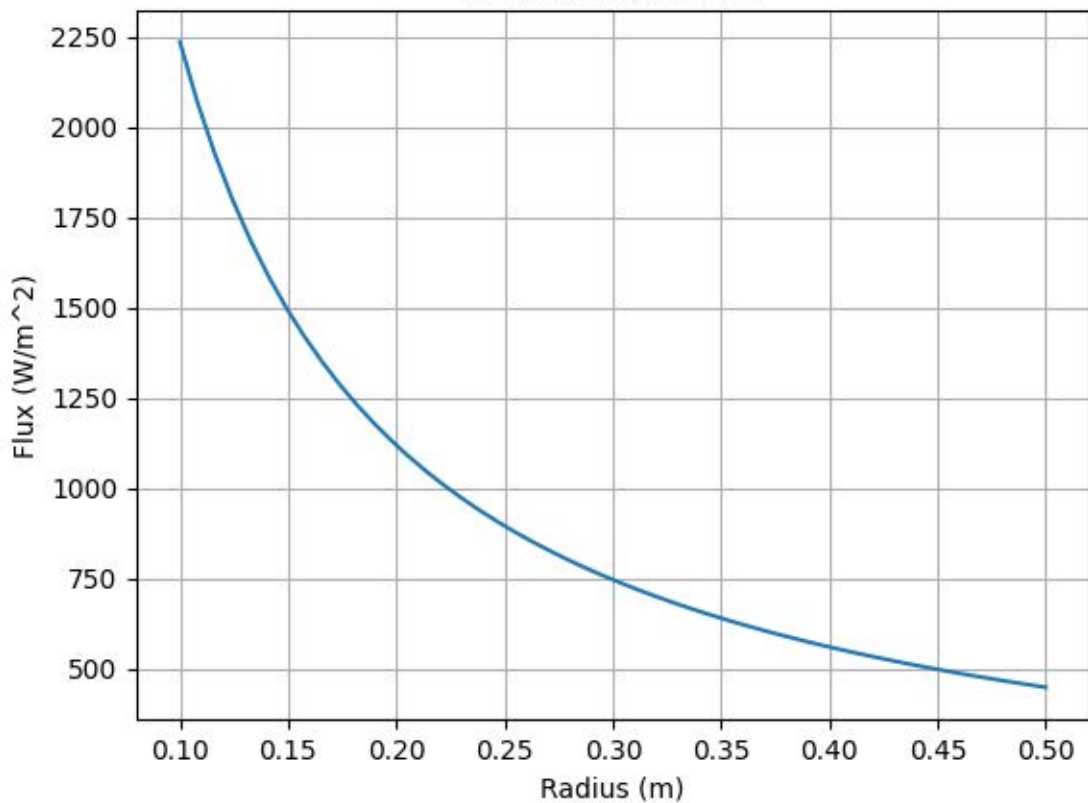
$$q_r = -2.4 \frac{\text{W}}{\text{mK}} \cdot -93.2 \cdot 2\pi L$$

$$q_r = 14054 \text{ W}$$

Temperature Profile (HW6.6)



Heat Flux (HW6.6)



Heat Rate (HW6.6)

