Effect of Mineral Buildup in Closed Aquaculture Systems on Hairy Shore Crab (*Hemigrapsus oregonensis*)

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Background - Chemistry

- Shelled invertebrates incorporate CaCO₃
 - Molting rate is tied to [Ca⁺] (Zanotto 2002; Perry *et al.* 2000); increasing [Ca⁺] sped up molt cycle (Zhang *et al.* 2024)
 - Crabs absorb the minerals and molt for homeostasis
 - Unknown to which degree this process "shuts down"--if at all-or cannot remove sufficient calcium
- Heavy metal uptake (Cd, Hg, Pb) facilitated through calcium incorporation (Averina *et al.* 2022)
 - Heavy metals can be detrimental for organism development and human consumption through aquaculture (Martins *et al.* 2009)
- In closed systems, this buildup is more prominent due to lack of removal methods

Background - Aquaculture

- Aquaculture systems are important for farming shelled invertebrates
- NOAA (2021) reported that global harvests in 2018 amounted to:
 - 16.1 million tonnes of shellfish (\$19 billion USD)
 - 6.9 million tonnes of crustaceans (\$36.1 billion USD)

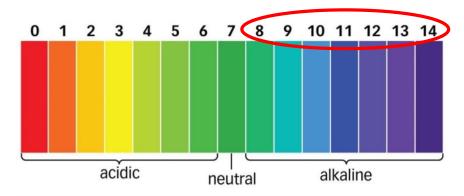
Typical research usually centers around ocean acidification, but often do not address alkaline conditions

Resources

- ➤ Lab Provided
 - Hairy Shore Crab (30-60)
 - Tank (2-4)
 - Temperature Manipulation Mechanism
 - Lactate and BCA Protein Physiology Assays
- > Independently Sourced
 - Calcium Carbonate Powder
 - Salt Water Calcium Concentration Test Kit

Research Question: What impact will increased dissolved calcium carbonate levels have?

- Sub Question: How will increased pH/Alkalinity affect our crab's physiology?



- Sub Question: Will the dissolved calcium carbonate in the water buildup on



Hypotheses

Research Question 1: Alkalinity changes

- Null Hypothesis: Increased pH levels from calcium carbonate will lead to no impact on the crab's physiological functions

Alternative Hypothesis: Increased pH levels from calcium carbonate will lead to increased hemolymph lactate and decreased hemolymph protein levels.



Hypotheses (Cont.)

Research Question 2: Calcification

- Null Hypothesis: Increased calcium carbonate levels in the water will not affect the crab's shells
- Alternative Hypothesis: Increased calcium carbonate levels in the water will lead to calcification on the crab's shells
 - Null Sub-Hypothesis: Calcification of the crab's shells will lead to no physiological responses
 - Alternative Sub-Hypothesis: Calcification of the crab's shells will cause negative physiological effects such as increased molting, increased response to heat stress, and longer righting times.

Experimental Design

- ➤ Tank Setup
 - 15 replicates per treatment
 - Sand substrate
 - Treatment Options
 - **♦** Control
 - ♦ Calcium Carbonate Supersaturation
 - ♦ Heat stress (Optional)
 - ♦ Heat Stress With Calcium Carbonate Hyper Saturation (Optional)

Experimental Design (continued)

- > Control
 - Average Temperature (7.6 °C)
 - Average Calcium
 Concentration (400 ppm)
- ➤ Heat Stress
 - Augmented Temperature (17.6 °C)
 - Average Calcium
 Concentration (400 ppm)

- > Calcium Carbonate Hypersaturation
 - Average Temperature (7.6 °C)
 - Augmented Calcium
 Concentration (800 ppm)
- Heat Stress with Calcium Carbonate Hypersaturation
 - Augmented Temperature (17.6 °C)
 - Augmented Calcium
 Concentration (800 ppm)

Experimental Design (continued)

- Calcium Carbonate Application
 - 1 mg Ca/L = 1 ppm
 - 1mg CaCO₃ = 0.4 mg Ca
 - Weekly addition to maintain concentration (daily if possible)
 - Weekly calcium concentration test
- > Heat Application
 - Set to 17.6 °C (if digital) or apply heat lamp and take temperature reading

Assessment

- Acute Stress (Open Systems)
 - Lactate Physiology Assay

 - Indicative of anaerobic respiration
 Within 24 hours of final calcium carbonate application
- Chronic Stress (Closed Systems)
 - **BCA Protein Physiology Assay**
 - Indicative of protein consumption
- Behavior
 - Righting
 - Indicative of overall health

References

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