



THE UNIVERSITY OF THE STATE OF NEW YORK • THE STATE EDUCATION DEPARTMENT • ALBANY, NY 12234 Reference Tables for Physical Setting/PHYSICS 2006 Edition

List of Physical Constants				
Name	Symbol	Value		
Universal gravitational constant	G	$6.67 \times 10^{-11} \mathrm{N} \cdot \mathrm{m}^2 / \mathrm{kg}^2$		
Acceleration due to gravity	g	9.81 m/s^2		
Speed of light in a vacuum	c	$3.00 \times 10^8 \text{ m/s}$		
Speed of sound in air at STP		$3.31 \times 10^2 \text{ m/s}$		
Mass of Earth		$5.98 \times 10^{24} \mathrm{kg}$		
Mass of the Moon		$7.35 \times 10^{22} \mathrm{kg}$		
Mean radius of Earth		$6.37 \times 10^6 \mathrm{\ m}$		
Mean radius of the Moon		$1.74 \times 10^6 \text{ m}$		
Mean distance—Earth to the Moon		$3.84 \times 10^8 \text{ m}$		
Mean distance—Earth to the Sun		$1.50 \times 10^{11} \; \mathrm{m}$		
Electrostatic constant	k	$8.99 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2$		
1 elementary charge	e	$1.60 \times 10^{-19} \text{ C}$		
1 coulomb (C)		6.25×10^{18} elementary charges		
1 electronvolt (eV)		$1.60 \times 10^{-19} \text{ J}$		
Planck's constant	h	$6.63 \times 10^{-34} \text{ J} \cdot \text{s}$		
1 universal mass unit (u)		$9.31\times10^2~\mathrm{MeV}$		
Rest mass of the electron	m_e	$9.11 \times 10^{-31} \text{ kg}$		
Rest mass of the proton	m_p	$1.67 \times 10^{-27} \text{ kg}$		
Rest mass of the neutron	m_n	$1.67 \times 10^{-27} \text{ kg}$		

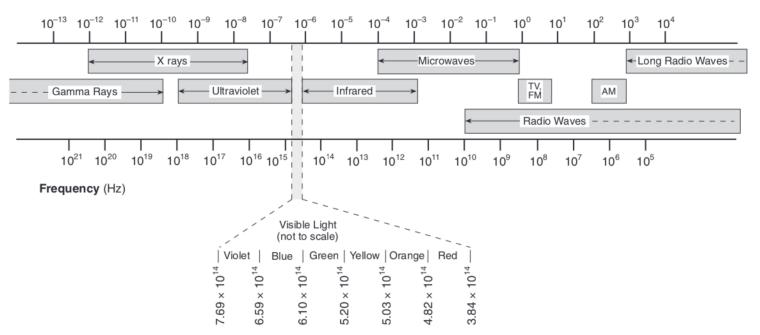
Prefixes for Powers of 10				
Prefix	Symbol	Notation		
tera	T	10^{12}		
giga	G	10^{9}		
mega	M	10^{6}		
kilo	k	10^{3}		
deci	d	10^{-1}		
centi	С	10^{-2}		
milli	m	10^{-3}		
micro	μ	10-6		
nano	n	10^{-9}		
pico	р	10^{-12}		

Approximate Coefficients of Friction			
	Kinetic	Static	
Rubber on concrete (dry)	0.68	0.90	
Rubber on concrete (wet)	0.58		
Rubber on asphalt (dry)	0.67	0.85	
Rubber on asphalt (wet)	0.53		
Rubber on ice	0.15		
Waxed ski on snow	0.05	0.14	
Wood on wood	0.30	0.42	
Steel on steel	0.57	0.74	
Copper on steel	0.36	0.53	
Teflon on Teflon	0.04		



The Electromagnetic Spectrum

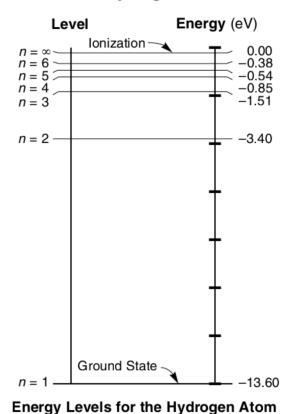




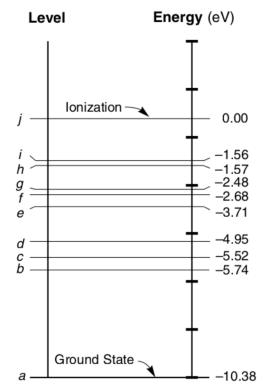
Absolute Indices of $f = 5.09 \times 10^{14}$	
Air	1.00
Corn oil	1.47
Diamond	2.42
Ethyl alcohol	1.36
Glass, crown	1.52
Glass, flint	1.66
Glycerol	1.47
Lucite	1.50
Quartz, fused	1.46
Sodium chloride	1.54
Water	1.33
Zircon	1.92

Energy Level Diagrams

Hydrogen

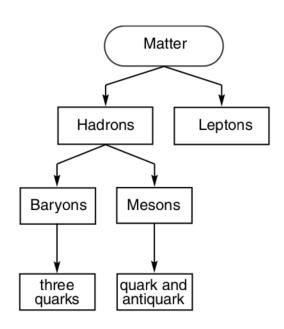


Mercury



A Few Energy Levels for the Mercury Atom

Classification of Matter



Particles of the Standard Model

Quarks

Name Symbol Charge

charm $c + \frac{2}{3}e$

 $top
t
+\frac{2}{3}e$

strange s $-\frac{1}{3}e$

bottom b $-\frac{1}{3}e$

Leptons

electron e -1e

 $\begin{array}{c} \text{muon} \\ \mu \\ -1 \text{e} \end{array}$

tau τ –1e

electron neutrino v_e

muon neutrino u_{μ} 0 tau neutrino v_{τ}

Note: For each particle, there is a corresponding antiparticle with a charge opposite that of its associated particle.