### Chapter 20

Organic Functional Groups

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#### Overhead 20 # 1 Organic Functional Groups

• A functional group is a specific arrangement of atoms in an organic compound that is capable of characteristic chemical reactions.

Table R: Organic Functional Groups			
Class of	Functional	General	
Compound	Group	Formula	Example
alcohol	-ОН	R-OH	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> OH
			1-propanol
	Ö	O	O
aldehyde	— <sup>  </sup> — Н	$R - \overset{\parallel}{C} - H$	$CH_3CH_2-\ddot{\mathbb{C}}-H$
			propanal
	O	O R'	O
amide	-C $-$ N $-$ H	O R'      	$CH_3CH_2-\ddot{C}-NH_2$
			propanamide
		R'	
amine	N	R - N - R''	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> NH <sub>2</sub>
			1-propanamine
ether	-O-	R - O - R'	CH <sub>3</sub> OCH <sub>2</sub> CH <sub>3</sub>
			methyl ethyl ether
	O	O	O
ester		$R - \overset{\parallel}{C} - O - R'$	$\left  \text{CH}_3\text{CH}_2 - \ddot{\text{C}} - \text{O} - \text{CH}_3 \right $
			methyl propanoate
	-F (fluoro-)	R-X	
halide	-Cl (chloro-)	(X represents	CH <sub>3</sub> CHClCH <sub>3</sub>
(halocarbon)	-Br (bromo-)	any halogen)	2-chloropropane
	−I (iodo-)		
	O	O	O
ketone		$R - \ddot{C} - R'$	$\left  \text{CH}_3 - \overset{\text{"}}{\text{C}} - \text{CH}_2 \text{CH}_2 \text{CH}_3 \right $
			2-pentanone
	O	O	0
organic acid	— <sup>  </sup> — ОН	$R - \ddot{C} - OH$	$CH_3CH_2-\ddot{\mathbb{C}}-OH$
			2-propanoic acid

$$\operatorname{CH}_3$$
  $\operatorname{CH}$   $\operatorname{CH}_2$   $\operatorname{CH}_3$ 

$$\begin{array}{c}
\text{O} \\
\parallel \\
\text{CH}_3-\text{C}-\text{CH}_2-\text{CH}_3
\end{array}$$

$$CH_3CH_2CH_2-O-CH_2CH_3$$

CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CONH<sub>2</sub>

$$O_2N$$
 $NO_2$ 
 $NO_2$ 

#### Overhead 20 # 1 Organic Functional Groups, con't

Name the following organic compounds.

$$\begin{array}{c} H \\ C = C \\ H \end{array}$$

$$CH_3-CH_2-C-C-CH_2-CH_3$$

$$\mathrm{CH_3-C-NH_2}^{\mathrm{O}}$$

$$\begin{array}{c} H - C - C \\ H \end{array}$$

# Overhead 20 # 1 Organic Functional Groups, con't Name the following organic compounds.

$$\begin{array}{ccc} H & H \\ N-C-H \\ H & H \end{array}$$

$$H_3C$$
  $CH_2$   $CH_2$   $\ddot{N}H_2$ 

$$\begin{array}{c} H \\ H \\ C \\ C \\ O \\ O \\ \end{array}$$

$$\begin{array}{c|cccc} H & CH_3H \\ & | & | \\ H-C-C-C-C-H \\ & | & | \\ H & I & H \end{array}$$

# Overhead 20 # 1 Organic Functional Groups, con't Name the following organic compounds.

$$\mathrm{H-C}_{\mathrm{OH}}$$

$$H-C$$
 $NH_2$ 

$${\rm H-C-H}$$

## Overhead 20 # 1 Organic Functional Groups, con't Draw structural formulas of the following organic compounds.

4-iodo-2-hexene pentanamine hexanoic acid

ethyl methanoate 2-methyl-2-propanol pentanamide

pentyl methyl ether diethyl ether 3-hexanone

#### Overhead 20 # 2 Isomers

• Structural isomers have the same functional group in a different position.

Draw the structural formula of the following compounds then its structural isomer.

2-butanol Isomer

3-pentanone Isomer

• Functional isomers have the same chemical formula but different functional groups.

Draw the structural formula of the following compounds then its functional isomer.

methyl ethyl ether Isomer

propanal Isomer