

HW#4

1) FCC Fe, 0.55 wt% C, 1325 K, 0 wt% C @ 1 atm

a) $\frac{\partial C}{\partial t} - \nabla \cdot (D \nabla C) - \dot{\alpha} = 0$ assume i) $\nabla D = 0$, constant D is good

ii) $\dot{\alpha} = 0$, no source

$\hookrightarrow \frac{\partial C}{\partial t} = D \nabla^2 C$

$\hookrightarrow C(x,t) = (C_s - C_0) \left(1 - \operatorname{erf} \left[\frac{x}{2\sqrt{Dt}} \right] \right) + C_0$

BC) $C(0,t) = 0 = C_s$

IC) $C(x,0) = 0.55 \text{ wt\%} = C_0$

$C(\infty,t) = 0.55 \text{ wt\%} = C_0$

$\Rightarrow C(x,t) = C_0 \operatorname{erf} \left(\frac{x}{2\sqrt{Dt}} \right)$

b) w/ $D_{\text{Fe, alloy}} @ 1325 \text{ K} = 3.30 \cdot 10^{-11} \text{ m}^2/\text{s}$, after 10 hrs or 36,000 s

$C = 0.25 = 0.55 \operatorname{erf} \left(\frac{x}{2\sqrt{Dt}} \right) \Rightarrow \operatorname{erf} \left(\frac{x}{2\sqrt{Dt}} \right) = \frac{0.25}{0.55} = 0.455$

$\operatorname{erf}(z) = 0.455$ when $z \approx 0.4275$

$\therefore \frac{x}{2\sqrt{Dt}} = z \Rightarrow x = 2z \sqrt{Dt} = 2(0.4275) \sqrt{(3.30 \cdot 10^{-11})(36,000)}$

$x = 9.319 \cdot 10^{-4} \text{ m} = 0.9319 \text{ mm}$

HW# 4 - cont

- 2) - pelbble bed of Li_2TiO_3 spheres, $\phi = 2 \text{ cm}$, $\rho = 3.43 \text{ g/cc}$, $M = 110 \text{ g/mol}$
- T produced uniformly @ rate of 5 mol/day/cc
 - T diffuses & gets swept away by H_2 @ 600°C
 - SS: molar fraction $10^{-6} \text{ mol T/mol LiTiO}_3$
 - $D = 10 \times 10^{-7} \text{ cm}^2/\text{s}$

a) $\frac{\partial C}{\partial t} - \nabla(\nabla C) = 2 = 0$ assume 1) SS, $\frac{\partial C}{\partial t} = 0$

2) $\nabla D = 0$, spatially invariant D

3) spherical symmetry, $\frac{\partial C}{\partial \theta} = \frac{\partial C}{\partial \phi} = 0$

$\rightarrow -\nabla \nabla^2 C = 2$

$\rightarrow \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial C}{\partial r} \right) = -\frac{2}{D}$

$\Rightarrow \frac{\partial}{\partial r} \left(r^2 \frac{\partial C}{\partial r} \right) = -\frac{2r^2}{D} \Rightarrow r^2 \frac{\partial C}{\partial r} = -\frac{2r^3}{3D} + A_1 \Rightarrow \frac{\partial C}{\partial r} = -\frac{2r}{3D} + \frac{A_1}{r^2}$

$C = -\frac{2r^2}{6D} - \frac{A_1}{r} + A_2$

Know $[C] = \frac{\text{mol}}{\text{V}}$, & $C_s = C_{\text{H}_2} = 10^{-6} C_{\text{LiTiO}_2} = 3.118 \times 10^{-8} \text{ mol/cc}$ @ surface

know, $V = \frac{4}{3}\pi r^3 = 0.004189 \text{ cm}^3$, $m = \rho V = 0.01436 \text{ g}$

sphere $\text{mol Li}_2\text{TiO}_3 = m/M = 0.0001306 \text{ mol}$ $C_{\text{LiTiO}_2} = 0.03118 \text{ mol/cc}$

next, get 2 in right units, $2 = 5 \text{ mol/day/cc} \cdot \frac{1000 \text{ mm}^3}{\text{cc}} \cdot \frac{1 \text{ day}}{26,400 \text{ s}} = 5.787 \times 10^{-10} \frac{\text{mol}}{\text{cc} \cdot \text{s}}$

next, BCs & dCs

BC) i) $C(r=1) = C_s = 3.118 \times 10^{-8} \text{ mol/cc}$

ii) $C(r=\infty) \neq 0$

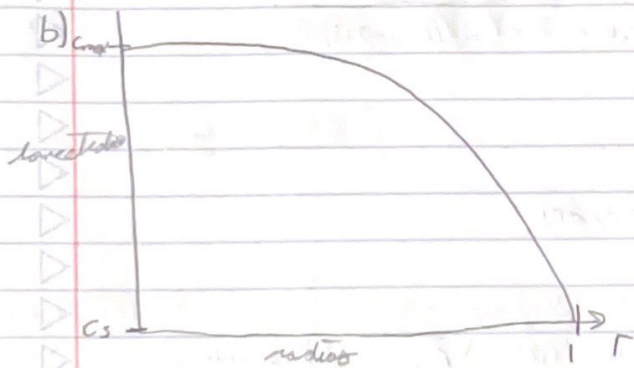
HW#4-carb

Applying BCs

$$ii) 0 = 0 - \frac{A_1}{\infty} + A_2 \quad \therefore A_1 = 0 \quad \Rightarrow C = \frac{-2r^2}{6D} + A_2$$

$$i) C_s = \frac{-2(1)^2}{6D} + A_2 \Rightarrow A_2 = C_s + \frac{2}{6D} = 3.118 \times 10^{-2} \frac{\text{mol}}{\text{cc}} + 9.645 \times 10^{-5} \frac{\text{mol}}{\text{cc}} = 9.648 \times 10^{-5} \frac{\text{mol}}{\text{cc}}$$

$$\Rightarrow C(r) = A_2 - \frac{2r^2}{6D} \quad \text{w/ } A_2 = 9.648 \times 10^{-5} \frac{\text{mol}}{\text{cc}} \\ 2 = 5.787 \times 10^{-10} \frac{\text{mol}}{\text{cc} \cdot \text{s}}$$



$$c) \text{ know loss / store} = 2\psi = \frac{5 \times 10^{-5} \text{ mol}}{\text{cc} \cdot \text{day}} \cdot \frac{4}{3} \pi r^3 = 2.0944 \times 10^{-7} \frac{\text{mol}}{\text{day}}$$

$$|R| = 2.0944 \times 10^{-7} \frac{\text{mol}}{\text{day}} \quad \& \quad |Q| = 2\psi \quad \text{in SS}$$

answers are somewhat off because of density used

HW#4

3) Cu, FCC, $a = 0.36 \text{ nm}$, $\gamma = 1.3 \text{ eV}$, $z = 1.3 \text{ eV}$

a) know (4.1) $N_v = N_{\text{exp}} \left(\frac{-z\gamma}{kT} \right) = 2.6955 \times 10^{21} \frac{\text{vacancies}}{\text{m}^3}$

$\rho = \frac{n M_{\text{Cu}}}{V_{\text{c}} N_A} = \frac{(4)(63.5 \text{ g/mol})}{(a^3)(6.022 \times 10^{23})} = 9.04 \text{ g/cc}$

$N = \frac{N_A \rho}{M_{\text{Cu}}} = \frac{(6.022 \times 10^{23})(9.04)}{(63.5)} \cdot (10^{-6} \frac{\text{cc}}{\text{m}^3}) = 8.5734 \times 10^{28} \frac{\text{atoms}}{\text{m}^3}$

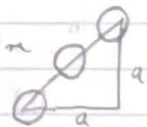
$N_v = N_{\text{exp}} \left(\frac{-z\gamma}{kT} \right) = (8.57 \times 10^{28}) \exp \left(\frac{-1.3 \text{ eV}}{(8.62 \times 10^{-5} \text{ eV/K})(600+273) \text{ K}} \right) = 2.696 \times 10^{21} \frac{\text{vacancies}}{\text{m}^3}$

b) $D_0 = 2.5 \times 10^{-5} \text{ m}^2/\text{s}$, $z = 2.1 \text{ eV}$ $\Gamma = D_0 \exp \left(\frac{-\Delta H_m + T \Delta S_m}{kT} \right)$

know $D = \frac{1}{6} \lambda^2 \Gamma$ $D = D_0 \exp(-z/kT)$

$\Rightarrow \Gamma = \frac{6D_0}{\lambda^2} \exp(-z/kT) = \frac{6(2.5 \times 10^{-5} \text{ m}^2/\text{s})}{(2.546 \times 10^{-10} \text{ m})^2 (12)} \exp \left(\frac{-2.1 \text{ eV}}{(8.62 \times 10^{-5} \text{ eV/K})(600+273) \text{ K}} \right) = 146.52 \text{ s}^{-1}$

$\lambda = \text{spacing b/w planes} = \frac{a}{\sqrt{2}} = \frac{0.36 \text{ nm}}{\sqrt{2}} = 2.546 \times 10^{-10} \text{ m}$
 $z = CN = 12$



$\Rightarrow \Gamma = D_0 \exp(-z/kT) \Rightarrow z = \Delta H_m \Rightarrow \Delta H_m = \ln \left(\frac{D_0}{\Gamma} \right) kT = 1.897 \text{ eV} = \Delta H_m$

z as thermal entropy change is $\rightarrow \Delta S_m \approx 0$
 negligible...