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Homework 4

- 1a) ? = U-239
- b) ? = C-14
- c) ? = Rn-222
- d) ? = U-234
- 2) 931.5 MeV
- 3a) 7.976 MeV
- b) 15.664 MeV
- c) 7.751 MeV
- d) 4.143 MeV
- e) These answers indicate that the extra neutron makes the isotope more unstable than if it had 1 less neutron.
- 4a) 3.27 MeV released
 - b) 17.56 MeV released
- 5) Reactions 1, 4, and 6 are exothermic.

Reactions 2 and 5 are endothermic

Reaction 3 is neither

- .1) q1 = 6.586 MeV
- .2) q2 = -1.85 MeV
- .3) q3 = 0 MeV
- .4) q4 = 0.559 MeV
- .5) q5 = -12.083 MeV
- .6) q6 = 2.125 MeV

6) Reactions 1 and 5 are exothermic

Reactions 3 and 4 are endothermic

Reaction 2 is neither

$$.1) q1 = 7.250 MeV$$

.2)
$$q2 = 0 \text{ MeV}$$

$$.3) q3 = -2.726 MeV$$

.4)
$$q4 = -2.342 \text{ MeV}$$

$$.5) q5 = 4.783 MeV$$

7a)
$$q = 1.820 \text{ MeV } (B + \text{decay})$$

b)
$$q = 4.9165 \text{ MeV (B- decay)}$$