Jonathan G. V. Ström^a, Shuai Xie^a, Eric M. Suuberg^a

^aBrown University, School of Engineering, Providence, RI, USA

Abstract

Abstract here

- 5 Keywords: Vapor intrusion, Temporal variability, Sorption, Attenuation
- 6 factor

7 1. Introduction

Many vapor intrusion (VI) contaminants has the capacity to sorb onto soil and various common indoor materials, but the role and more importantly - the consequences of these sorption processes in VI are poorly understood. The migration of contaminant vapors from its source into the affected building and potential indoor sources are usually the prime concern in VI investigations. Rarely is the sorbed contaminant vapors in the soil or indoor considered in an investigation, but these may potentially act as a capacitor, storing and releasing contaminant vapors in response to a change in contaminant concentration. Consequently, contaminant vapors may be much more persistent at a site that has undergone remediation, potentially reducing the effectiveness of mitigation systems, or impeding site investigations.

Measuring contaminant sorption onto materials or soils is not a regular part of VI investigations and thus very little is known of the potential impact of this. In this paper we will therefore begin to explore this topic through a combined effort of gathering experimental sorption data and incorporating it into three-dimensional finite element models of VI scenarios. These type of models have been used extensively before and are useful to illuminate some of these issues. The potential impact of sorption may perhaps be most significant in the application of the controlled pressure method and various mitigation schemes.

The controlled pressure method (CPM) is the sequential forced overand depressurization of a building which in theory max- and minimizes the contaminant entry to the building. Helping the investigator to ascertain the worst-case VI scenario and help identify potential indoor contaminant sources. However, if the building has a large capacity to sorb contaminant vapors onto various materials, these may be ad-/absorbed and desorbed in response to the changing condition, potentially preventing coresponding changes in indoor air contaminant concentrations. The same is true for various mitigation schemes, while they may successfully prevent contaminant vapors from entering the house, these may still be released from the interior over an unknown period of time.

[1]

39

2. Methods

2.1. Experimental Setup

The TCE dynamic sorption process of different building materials were determined by use of a method schematically shown in Figure 1. This method involved a selected material contained in an adsorption column through which TCE-containing gas was passed, and subsequent thermal desorption and measurement of the total amount of adsorption. During the adsorption part of the process, stainless steel tubes were packed with building materials held in place by glass wool. The amount of building material normally held in the tube was around 1 g. It was determined that neither the glass wool nor the stainless steel tube would retain significant amounts of TCE. The samplecontaining tubes were first exposed desired low concentrations of TCE in nitrogen, which were then allowed to interact with the flow for varying periods of time. The typical flow rate of the nitrogen was 60 ml/min and the concentrations of TCE was around 1.1 ppbv. All of these adsorption experiments were conducted at room temperature. After a given time of exposure to the TCE-containing flow, that flow was stopped, and the sample tube was attached to a sorbent tube placed downstream of the sample tube. The sample tube was arranged such that the direction of the nitrogen flow in the subsequent desorption process was opposite that of the TCE-containing nitrogen flow during the adsorption process. During the thermal desorption step, the sample containing tube was covered by a heating mantle which permitted its heating at 100 C. This allowed fully desorbing the TCE which had been held on the sample into a pure nitrogen flow, which carried it to the room temperature downstream sorbent tube, where it was again fully

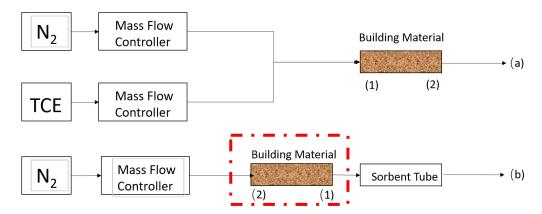


Figure 1: Schematic of experimental setup.

adsorbed. These tubes fully capture all of the TCE desorbed, from the samples, and the amount of TCE was analyzed by Gas Chromatography (GC) with an Electron Capture Detector(ECD).

2.2. Numerical Model

77

81

82

83

85

To investigate the role of sorption in VI, we consider a simple VI scenario. Here we consider a house with a 10 by 10 m footprint, with the foundation bottom located 1 m below ground surface (bgs). The sole contaminant source is an uniformly TCE contaminated groundwater located 4 bgs, and the soil surrounding the house is assumed to homogenous and of a singular type. All contaminant vapors are assumed to enter the house through breaches in the foundation, modeled as a 1 cm wide crack that runs along the perimeter of the house. Finally we assume that sorption processes can occur both in the soil matrix and in the indoor environment (on various indoor materials).

Modeling this scenario requires us to simulate a couple of physics, many of which depend and interact with each other. The governing equations and the physics they govern are:

- 1. van Genuchten retention model soil moisture.
- 2. Darcy's Law air flow in the porous media.
- 3. Transport equation contaminant transport in porous media.
- 4. Continuously stirred tank reactor (CSTR) contaminant concentration in the indoor environment.

These physics are implemented in COMSOL Multiphysics, a commercial finite-element method package, which is used to solve our model. It is impor-

tant to note that the indoor environment is implicitly modeled, but instead only given by the CSTR equation; the soil domain is explicitly modeled.

2.2.1. Vadose Zone Moisture Content

91

101

103

105

106

108

Since the contaminant transport occurs through three-phased the vadose zone, it is important that we correctly account for soil moisture content and its effect on advective and diffusive transport. In this modeled scenario, we assume that the soil moisture is at steady-state and does not change, and thus the soil moisture content is given by the retention model developed by van Genuchten.

The van Genuchten retention model gives the soil water saturation as a function of elevation above groundwater. In turn this gives the water and gas filled porosities, and the relative permeability of the soil matrix.

$$Se = \begin{cases} \frac{1}{(1+\alpha z^n)^m} & z < 0\\ 1 & z \ge 0 \end{cases}$$
 (1)

$$\theta_w = \begin{cases} \theta_r + \operatorname{Se}(\theta_s - \theta_r) & z < 0\\ \theta_s & z \ge 0 \end{cases}$$

$$k_r = \begin{cases} \operatorname{Se}^l \left[1 - \left(1 - \operatorname{Se}^{\frac{1}{m}}\right)\right]^2 & z < 0\\ 0 & z \ge 0 \end{cases}$$

$$(2)$$

$$k_r = \begin{cases} \operatorname{Se}^l \left[1 - \left(1 - \operatorname{Se}^{\frac{1}{m}} \right) \right]^2 & z < 0 \\ 0 & z \ge 0 \end{cases}$$
 (3)

Se is the saturation, and ranges from 0 to 1, which represent completely unto fully saturated; z is the elevation above the groundwater in meter; θ_r , θ_s , θ_w , and θ_g are the residual moisture content, saturated porosity (or just porosity), and water and air filled porosities respectively. All units are in volume of phase divided by the volume of soil; k_r is the relative permeability of water, which modifies the saturated permeability. This too ranges from 0 to 1, indicating completely im- and permeable respectively. $1-k_r$ gives the relative permeability of air.

2.2.2. Gas Flow In The Vadose Zone

The gas flow in the vadose zone is governed by a modified version of Darcy's Law. Originally, Darcy's Law was developed to describe flow in saturated porous media, but since we're interested in flow in unsaturated media - modification is necessary. An effective permeability that depends on the relative permeability from van Genuchten is introduced to allow for correct flow profiles in unsaturated porous media.

The vapor flow governing equation is given by

$$\frac{\partial}{\partial t}(\rho\theta_s) + \nabla \cdot \rho \left(-\frac{(1-k_r)\kappa}{\mu} \nabla p \right) = 0 \tag{4}$$

Here ρ is the fluid density; ∇ is the del operator; κ is the saturated permeability; μ is the fluid viscosity; and p is the fluid pressure. We assume that the contaminant vapors are so dilute that the gas flow properties can be taken to be those of air, and specifically at 20 Celsius.

To solve (4) we need to specify some boundary and initial conditions. In this VI scenario, we assume that there is a pressure difference between the indoor and outdoor. To reflect this we assume that the ground surface is the datum and thus pressure here is always zero. Likewise, to reflect the pressure difference, we assign its value directly to the foundation crack boundary. The rest of the boundaries are no flow boundaries and the initial condition is zero Pa. These conditions are summarized in Table ??.

 24 2.2.3. Mass Transport In The Vadose Zone

2.2.4. Indoor Environment

Table ?? summarizes the governing equations, boundary and initial conditions, and parameters used in the numerical model.

3. Results & Discussion

²⁹ 4. Conclusions

112

115

117

131

133

134

30 Acknowledgements

This project was supported by grant ES-201502 from the Strategic Environmental Research and Development Program and Environmental Security Technology Certification Program (SERDP-ESTCP).

Declaration of interest: none

References

[1] D. N. Bekele, R. Naidu, S. Chadalavada, Influence of soil properties on vapor-phase sorption of trichloroethylene, Journal of Hazardous Materials 306 (2016) 34–40. doi:10/f8kswb.