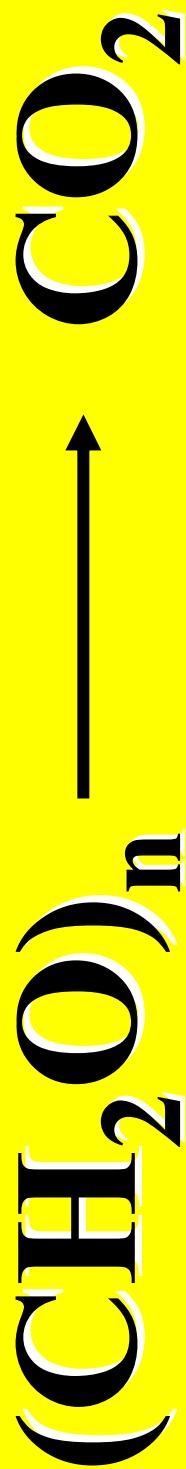


“Eutrophication” means there is more respiration (because there is more organic fuel produced in, or washed into, the lake).



*If it lives...it performs respiration!*

All living organisms “combust” organic fuel to carbon dioxide. We often measure the rate of oxygen loss in a lake hypolimnion. That measures *aerobic respiration*. We rarely measure the rate of increase in dissolved inorganic carbon ( $\text{CO}_2$ ) which would account for *all respiration of all living organisms*.

*Autochthonous and Allochthonous*

# Synthesis



Photosynthesis  
Chemosynthesis

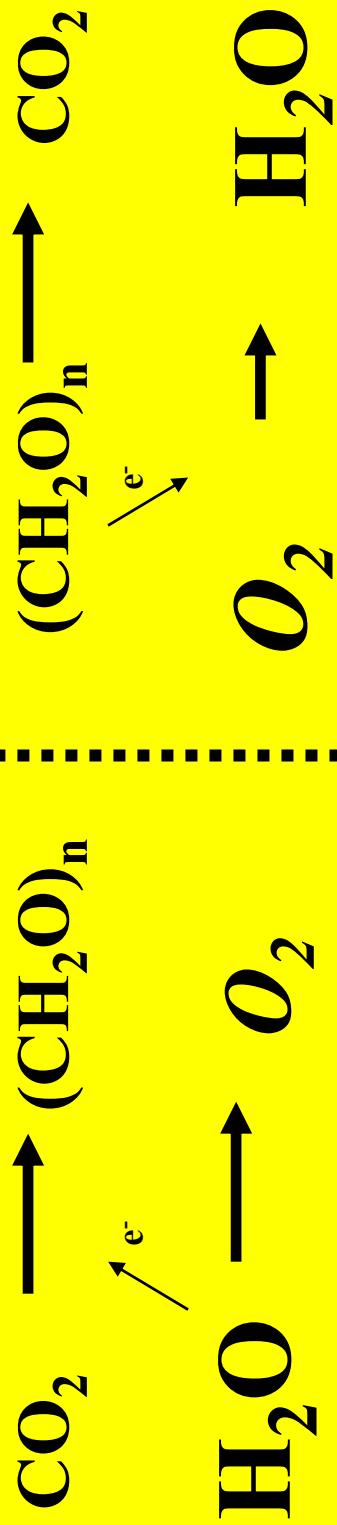


# Respiration

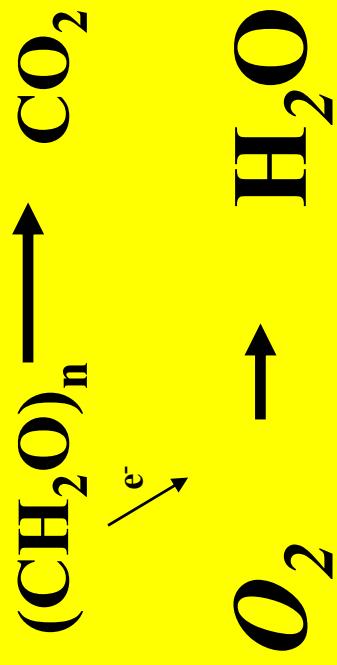


Aerobic Respiration  
Anaerobic Respiration  
Fermentation

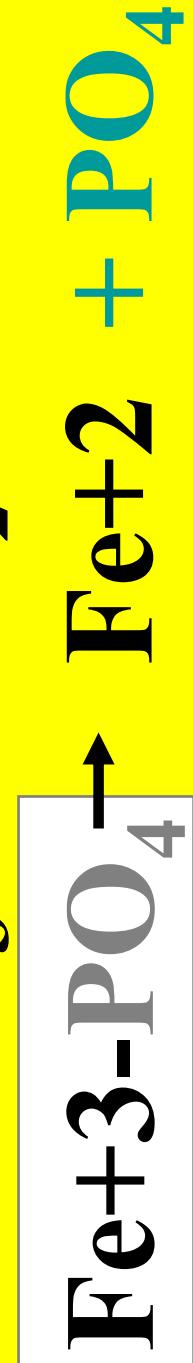
## Photosynthesis



## Aerobic Respiration

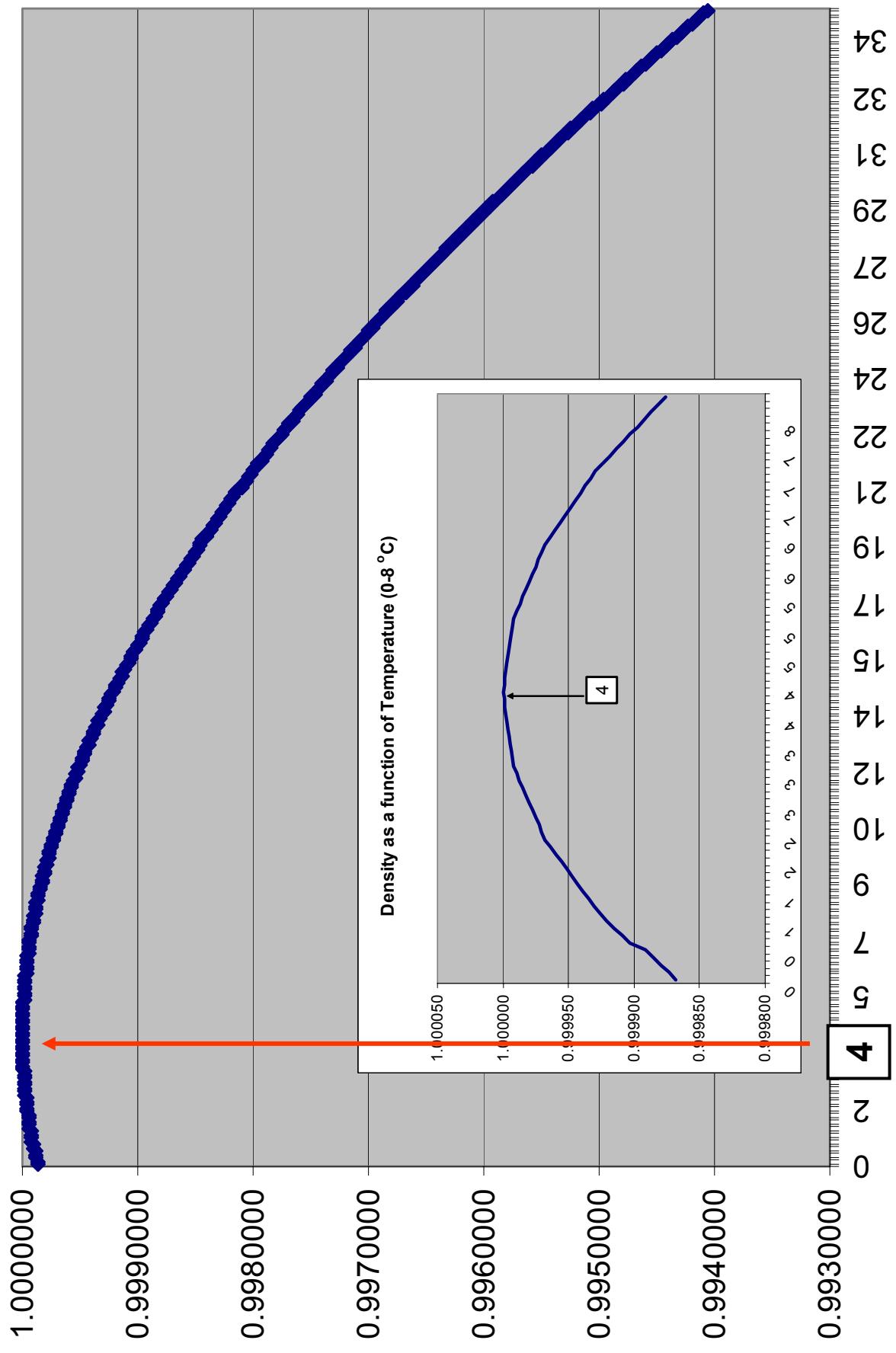


*In Order*



## Anaerobic Respiration

# Density as a function of Temperature ( $^{\circ}\text{C}$ )



RTRM

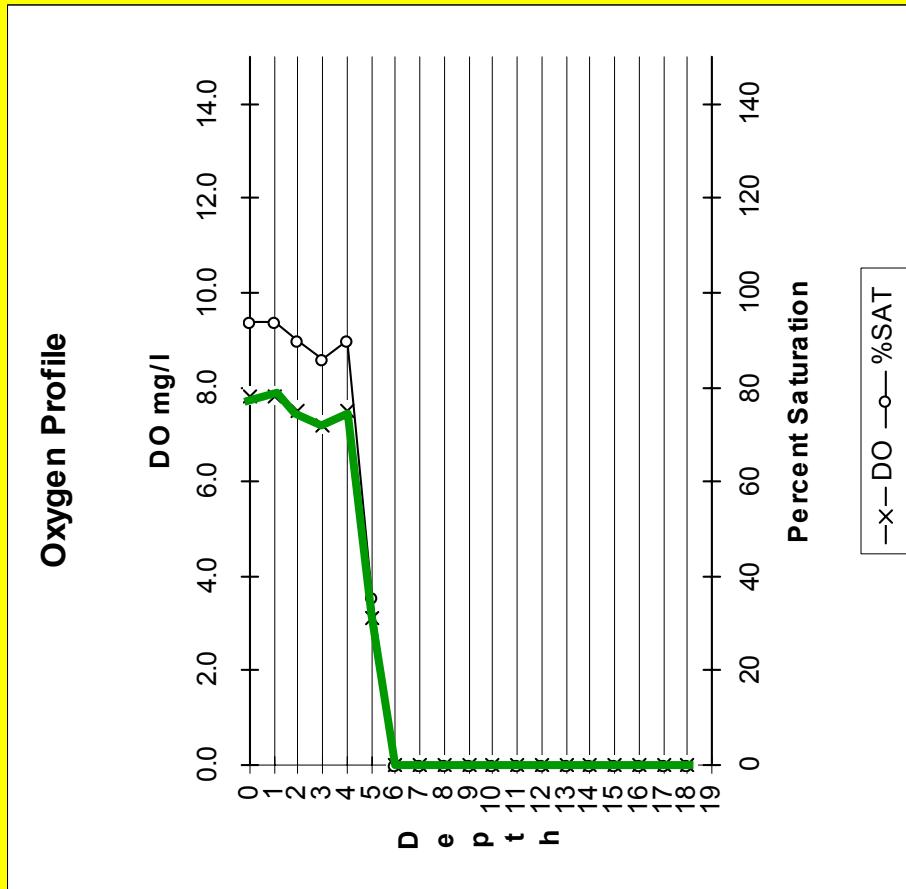
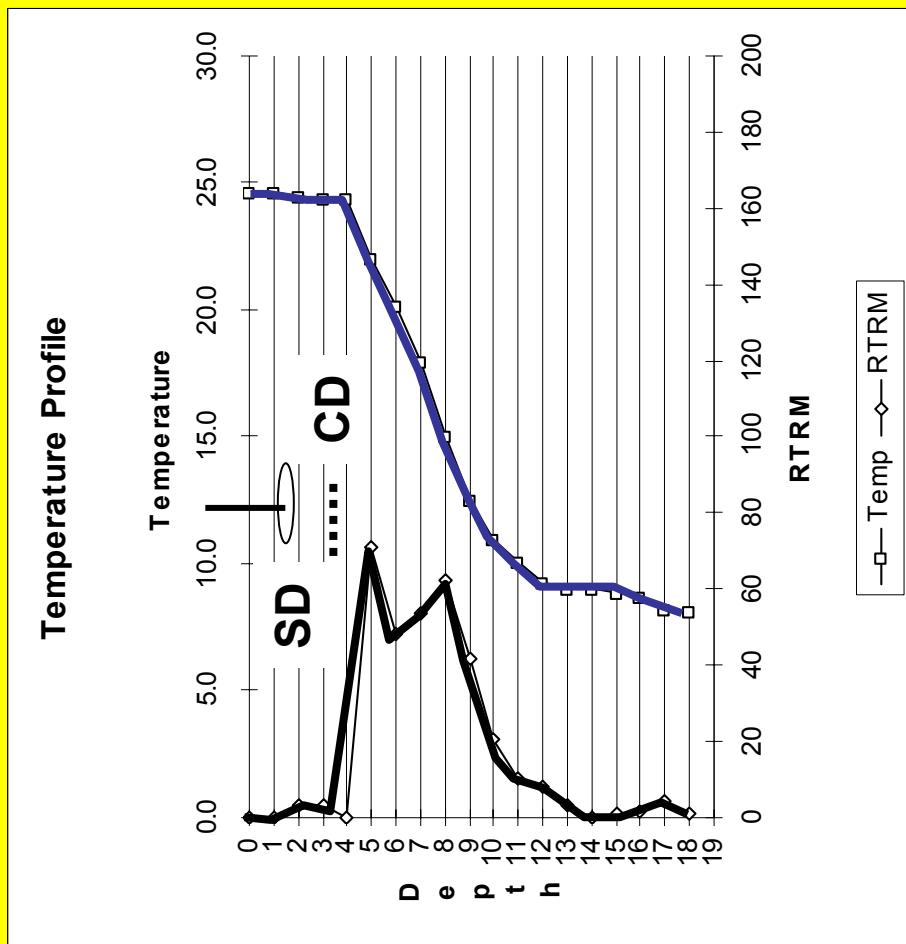
Relative Thermal Resistance to Mixing

$$\text{RTRM} = \frac{\text{Density of Upper Layer} - \text{Density of Lower Layer}}{\text{Density at } 5^\circ\text{C} - \text{Density at } 4^\circ\text{C}}$$

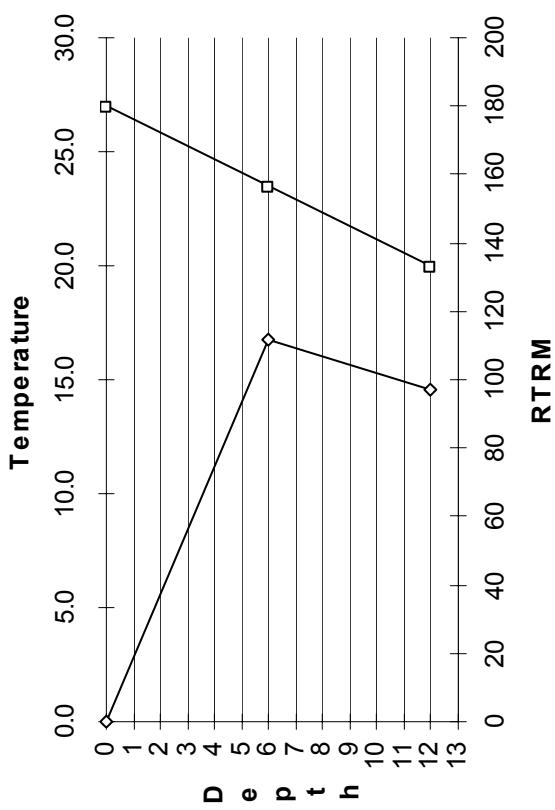
RTRM

- ...is a relative, non-dimensional, value which quantifies stratification as a function of temperature differential.
- ...is independent of the units of depth used.
- ...is additive (RTRM<sub>sum</sub> indicates total stratification intensity).
- ...the Peak RTRM identifies the location and intensity of the thermocline (steepest density gradient).

## “Typical Stratification”

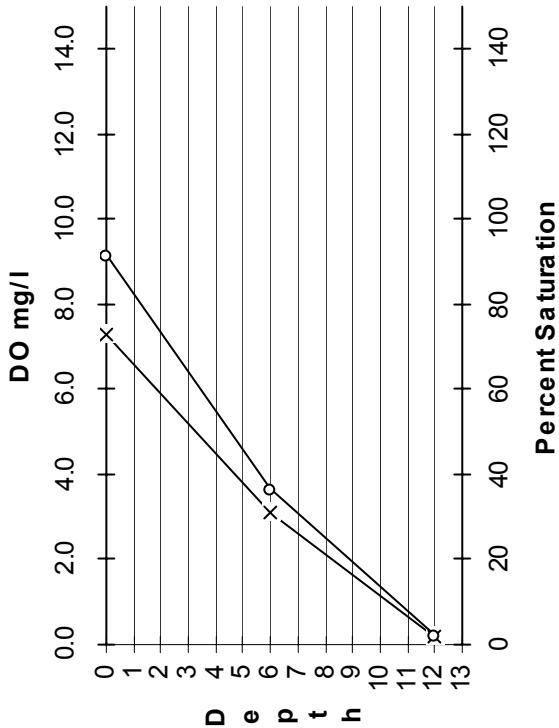


### Temperature Profile



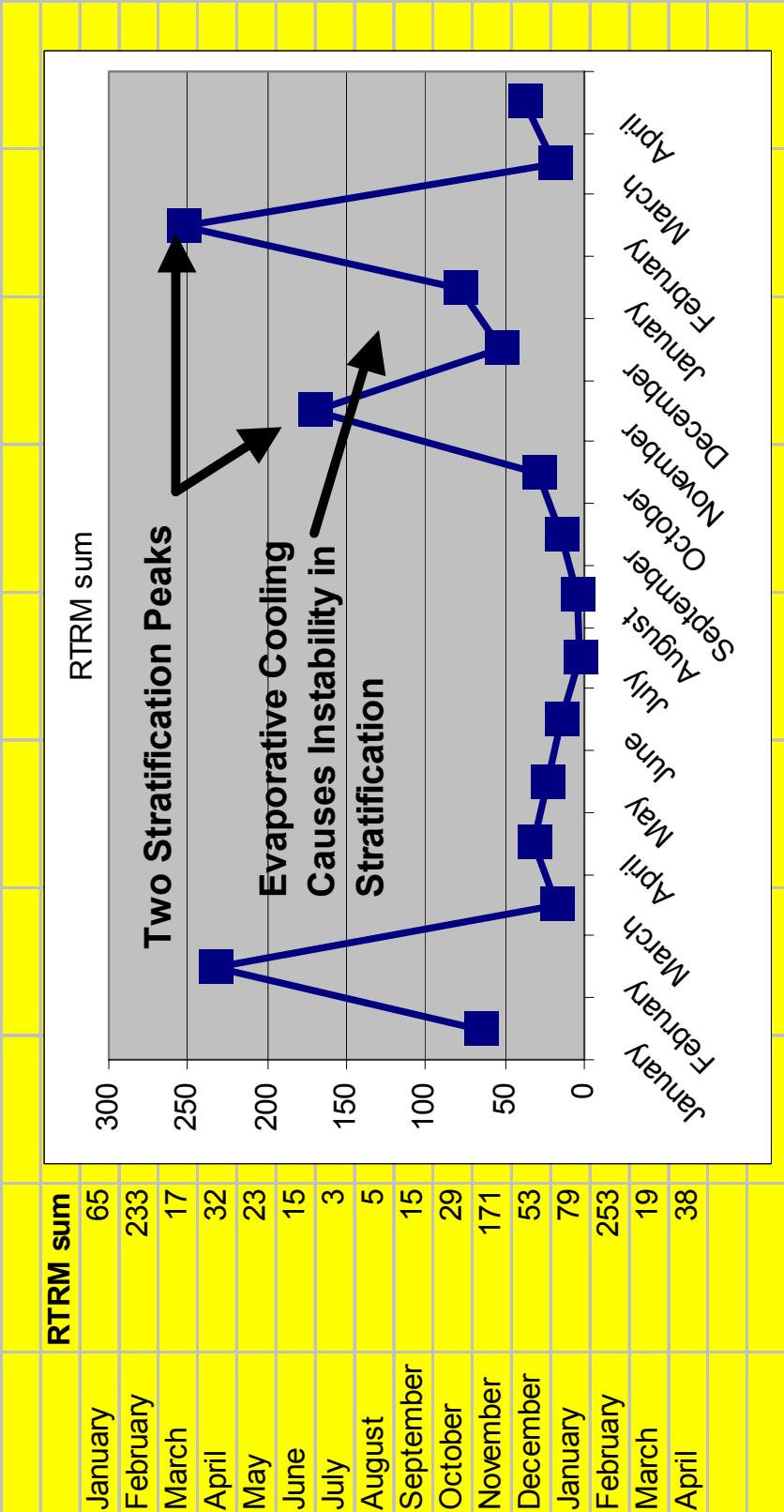
—□— Temp —◇— RTRM

### Oxygen Profile



—×— DO —○— %SAT

December 1993 (their summer)  
Rio Grande Reservoir, Sao Paulo, Brazil  
Depth in Meters RTRMsum= 209 RTRM Units



## Guarapiranga Reservoir, São Paulo, Brazil

During the summer, two stratification peaks are observed (and result in two anoxia seasons, separated by a mixing episode). The evaporative cooling during mid-summer represents enough temperature change (hence density change at the warm lake temperatures) To cause instability in stratification. The RTRM peaks during February and March are typically Stronger because overall temperatures are higher (remember the higher the temperature, the stronger The resistance to mixing per degree difference). The reservoir typically exhibits bluegreen algae Blooms in January-April.

**BLUEGREEN ALGAE FORCING FACTORS**

**N:P Ratio   Organic-N   Organic-P**

**SiO<sub>2</sub> : P Ratio**

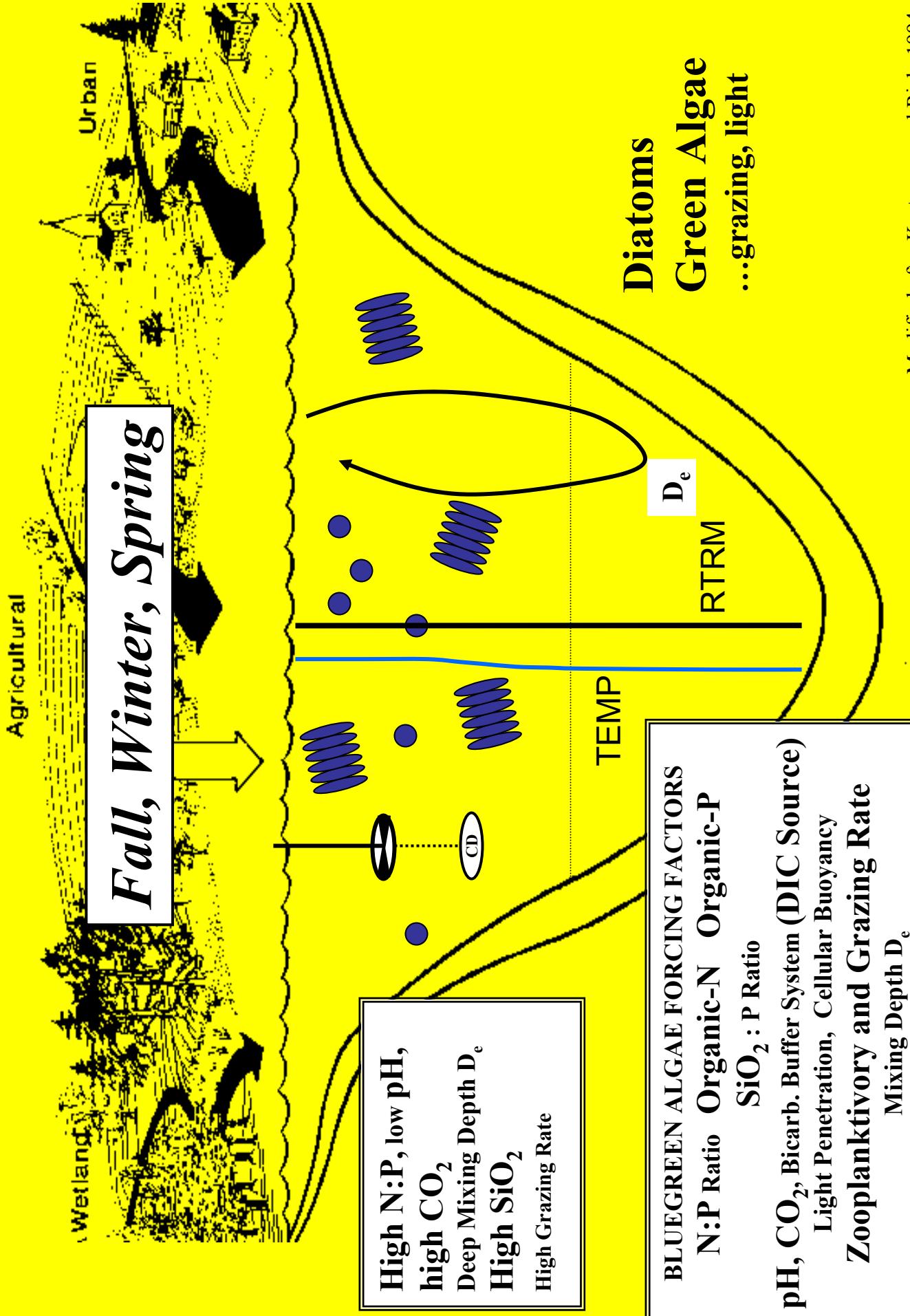
**pH, CO<sub>2</sub>, Bicarb. Buffer System (DIC Source)**

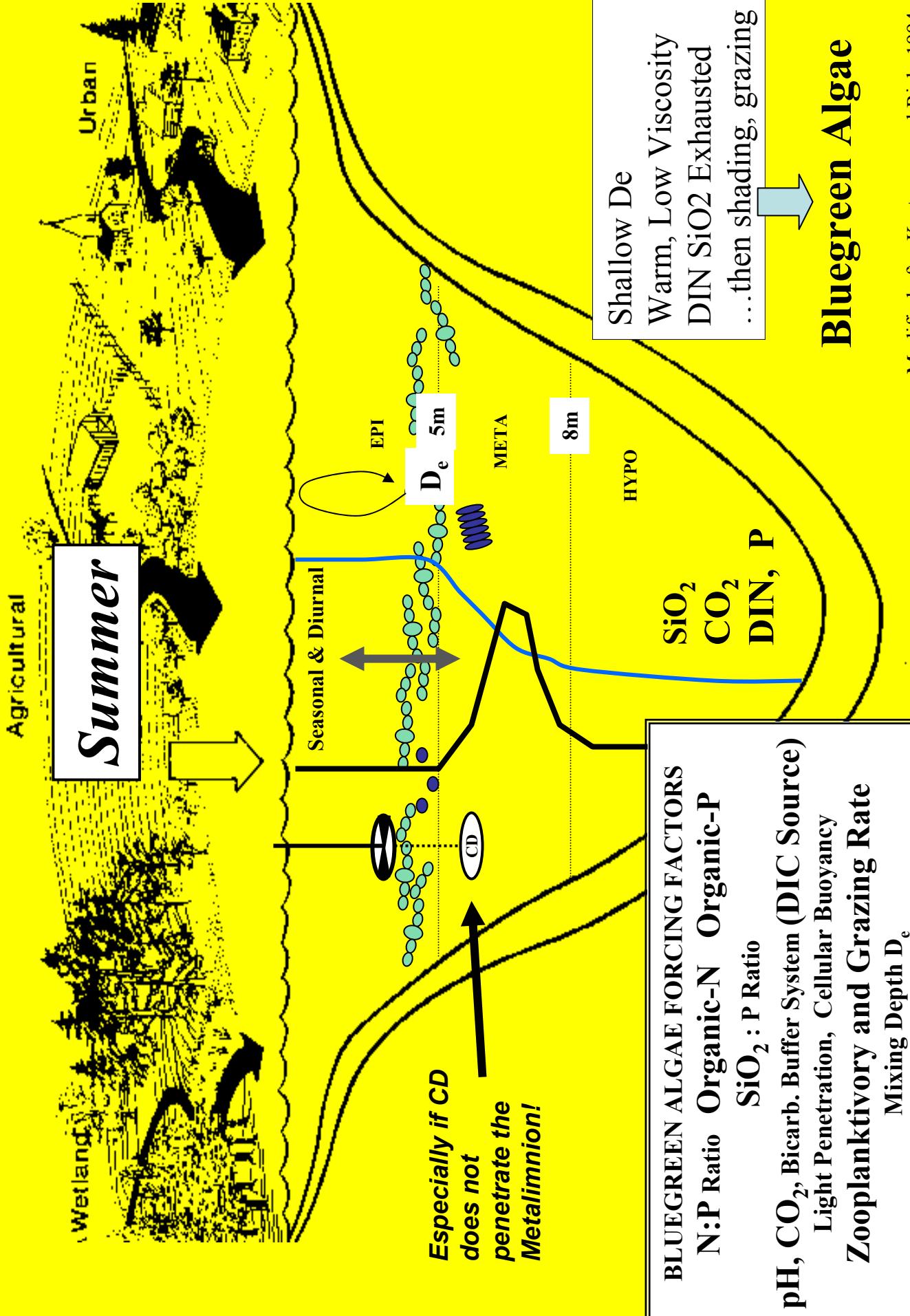
**Light Penetration, Cellular Buoyancy**

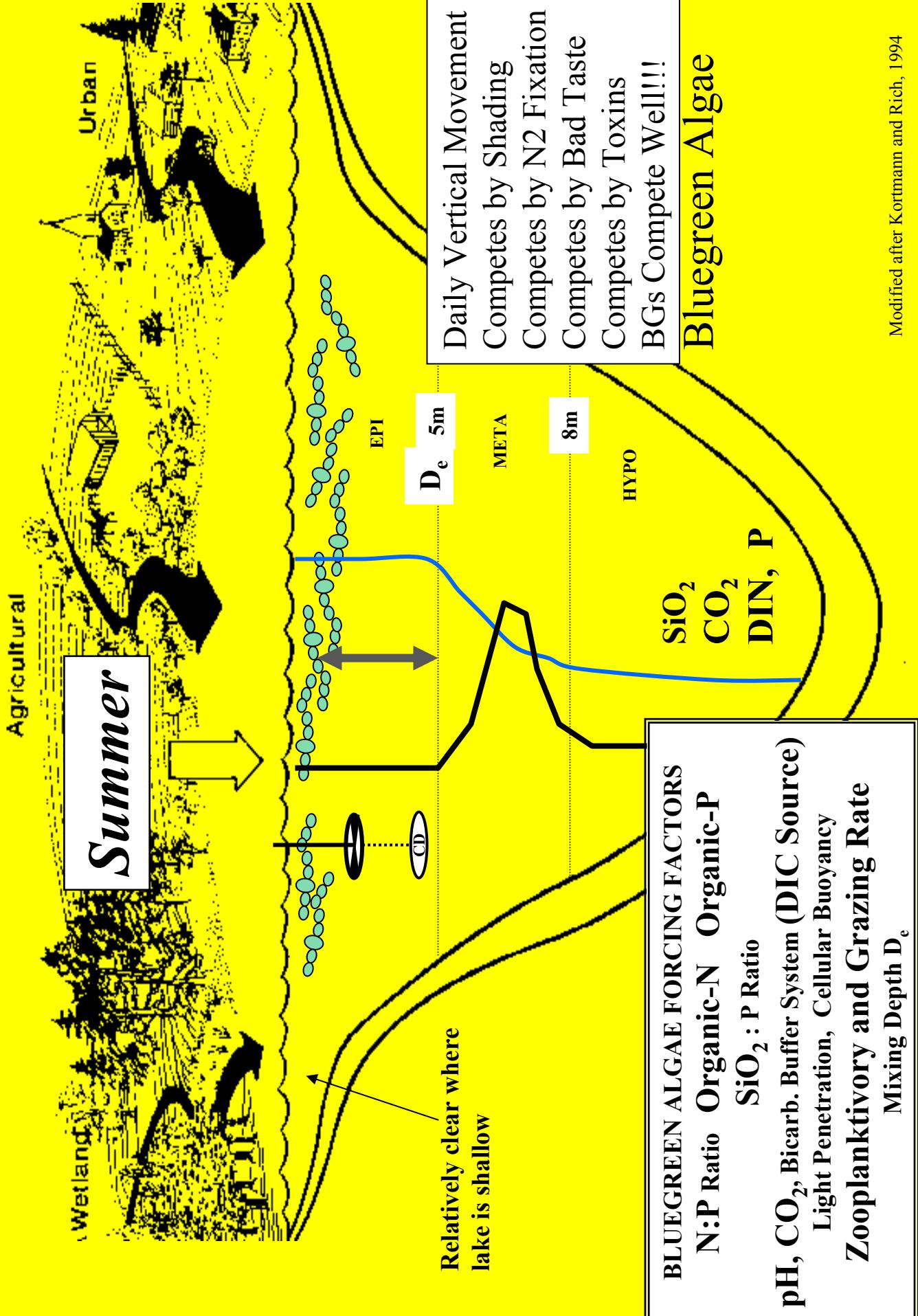
**Zooplanktivory and Grazing Rate**

**Mixing Depth D<sub>e</sub>**

**Bluegreen Algae (Cyanobacteria, prokaryotic cells)**







**Now for a “closer look” ...**

**... with diagnostic and management implications.**

**Synthesis – Respiration: Anaerobic Pathways**

**Physical Limnology: Managing the Heat Budget and Stratification Dynamics**

**Food Web Dynamics: Productivity vs. Biomass**

**Raw Water Quality Relationships:**

Anaerobic Respiration Products- Fe, Mn, S=; NTU, Filter Runs

Alkalinity, Buffering Capacity

Relates to Coagulation Flocculation/Dosing, Enhanced Coag.

Temperature Swings- Disruption of Settling Processes

Disinfection Byproducts and Precursors- THMs and HAAs

Where to place the first Disinfection Step

Taste and Odor- Geosmin and MIB – Phytoplankton vs Epibenthic Algae

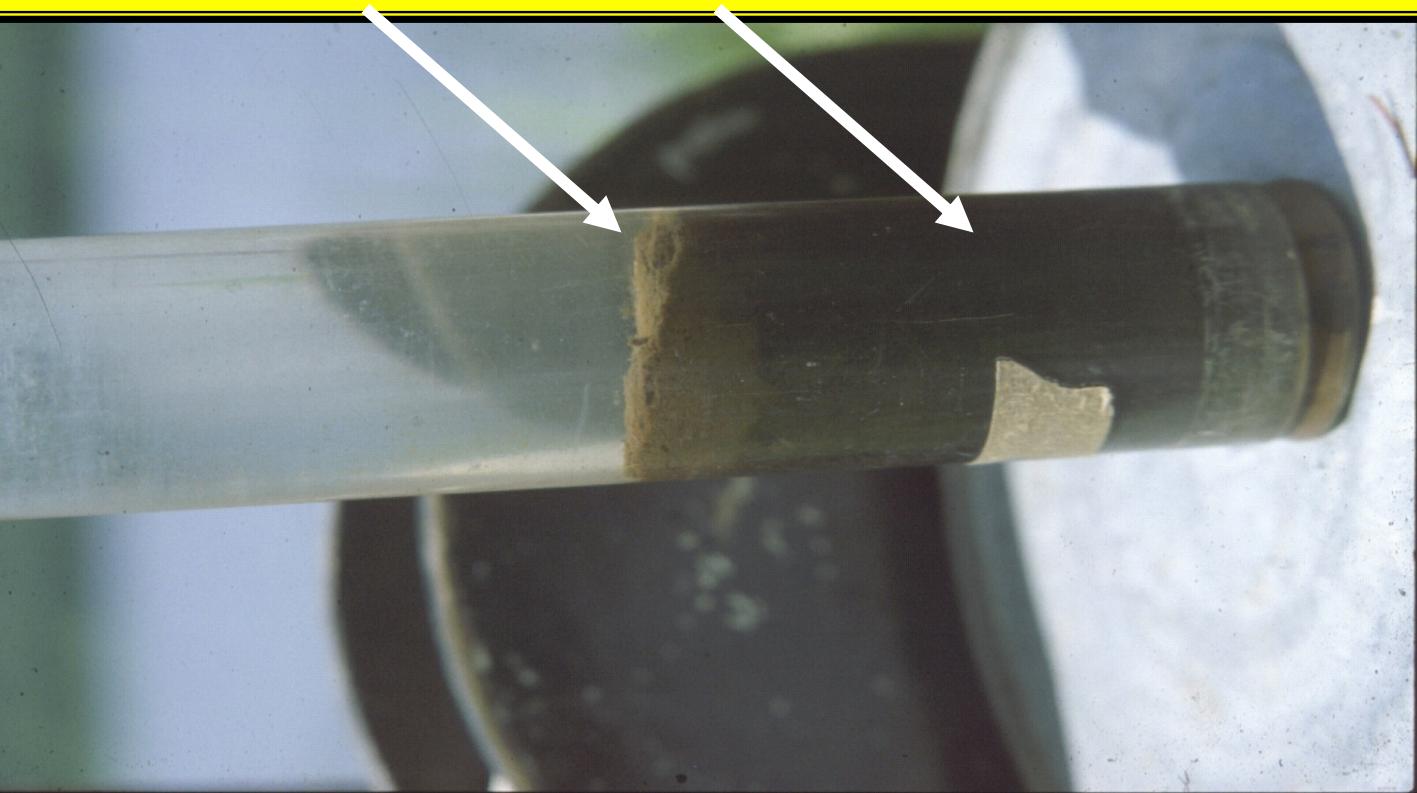
“Nutrient” Source Management – Internal vs. External

*In Fe-dominated systems:*

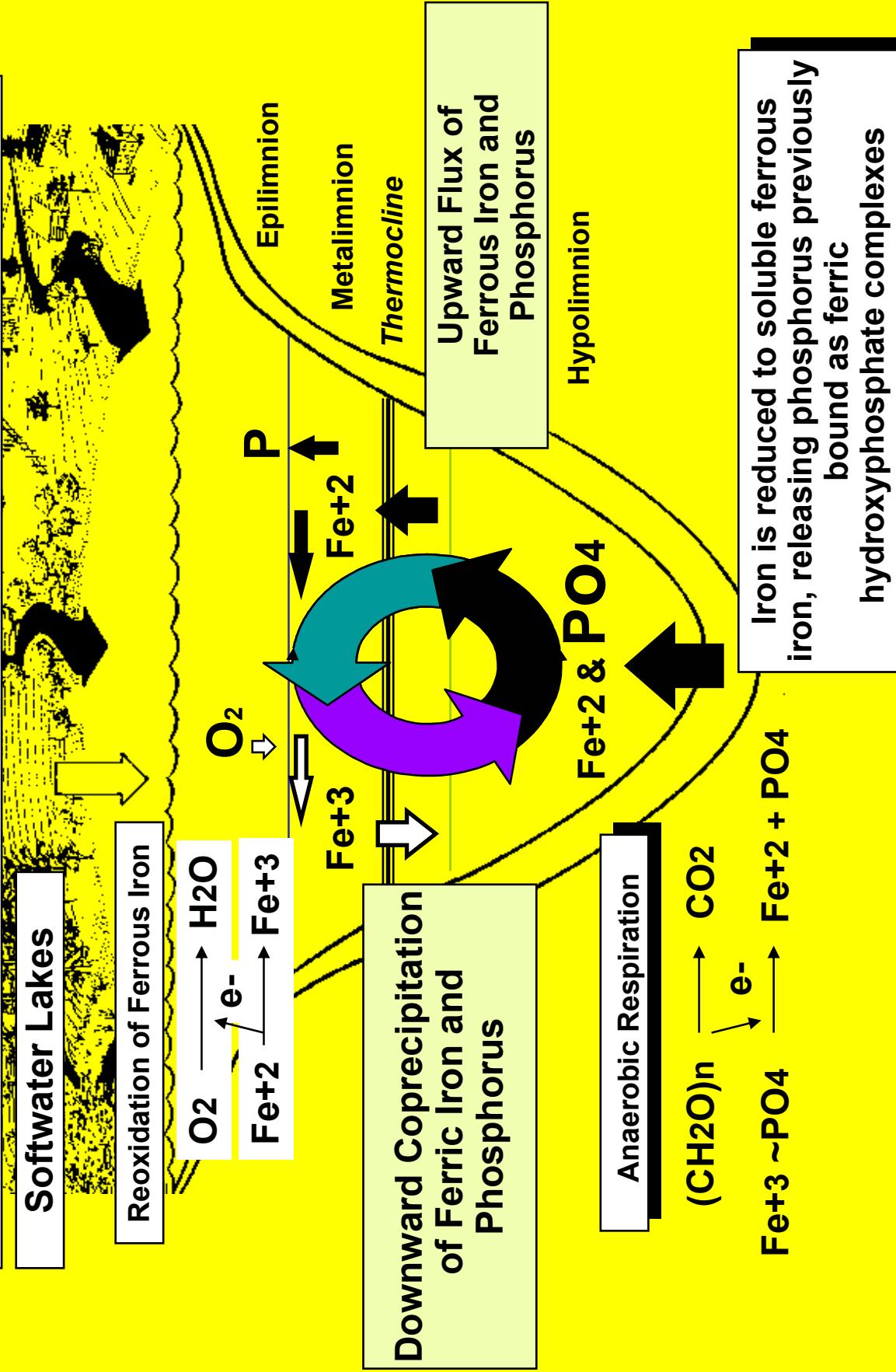
Oxidized Microzone  
Ferric Iron Complexes  
(Some of the Fe binds P)

Deep Anaerobic Sediments  
(with *Ferrous Sulfide* solids;  
Note: both the iron and  
sulfur are chemically  
reduced and do not  
participate in biotic redox or  
P-binding any longer!)

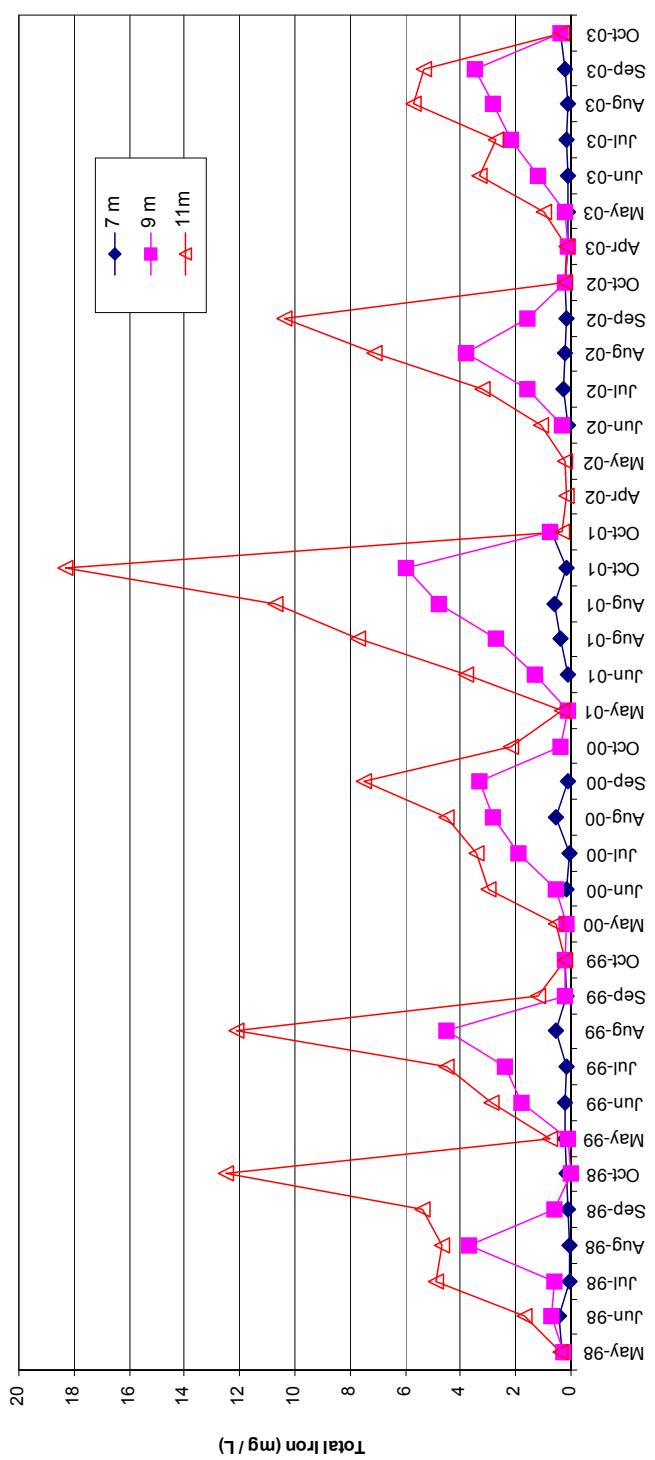
(Re: Roger Doyle, 1968)



# The Ferrous Wheel

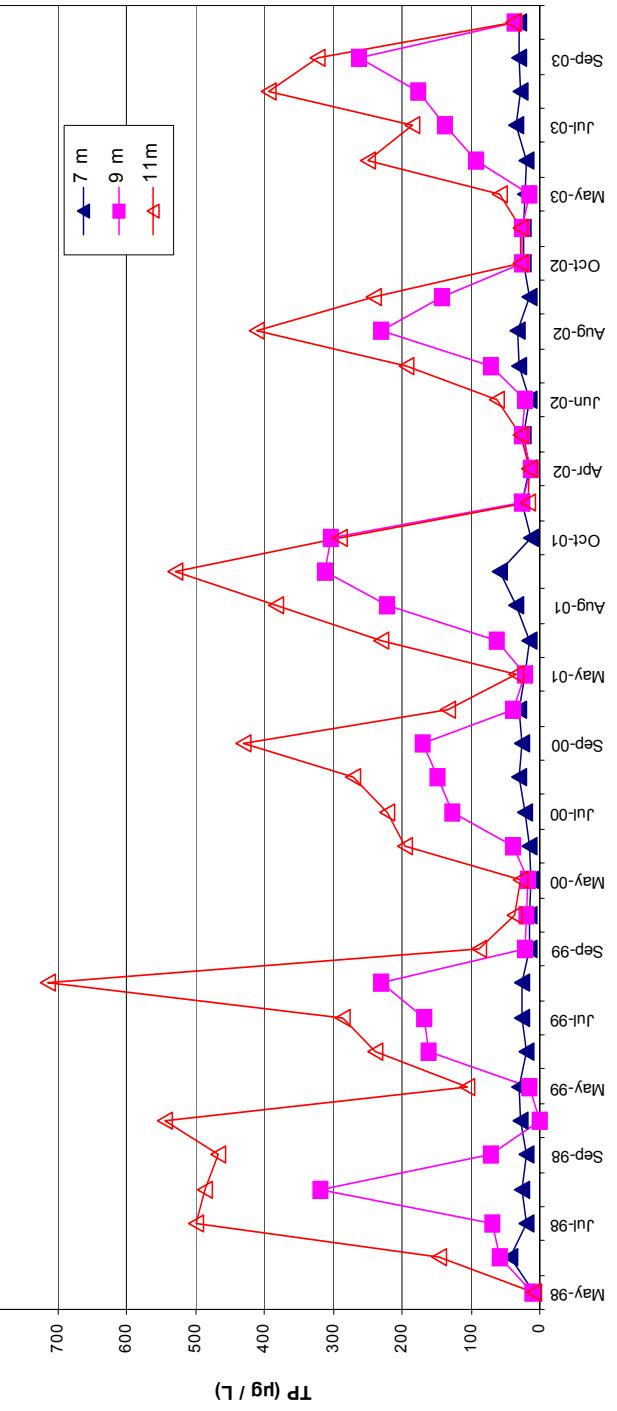


### Lake Waramaug Seasonal Hypolimnetic Total Iron



Fe

### Lake Waramaug Seasonal Hypolimnetic Total Phosphorus



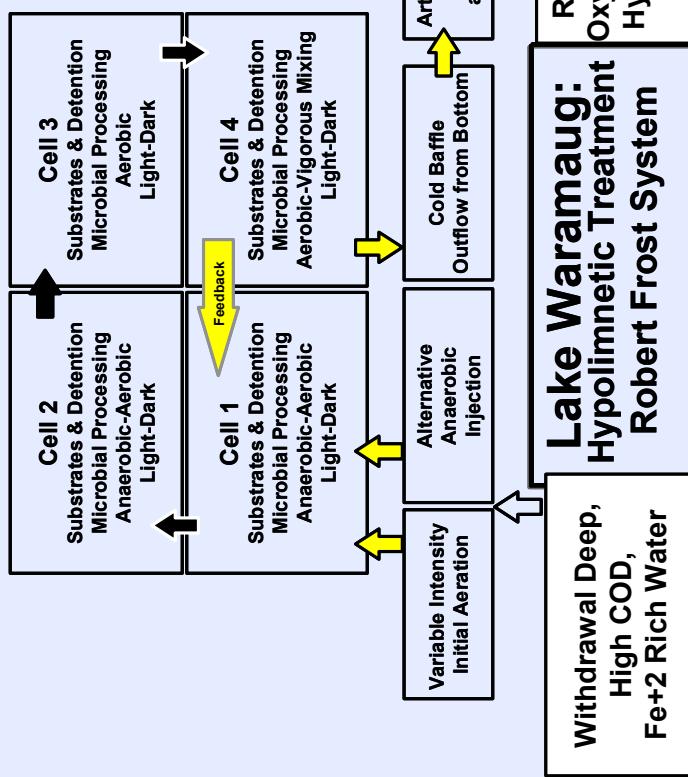
TP

# Lake Waramaug

(Fe still carries Anaerobic Respiration)

Iron, generated by anaerobic respiration, is used as a coagulant to remove P from lake water.

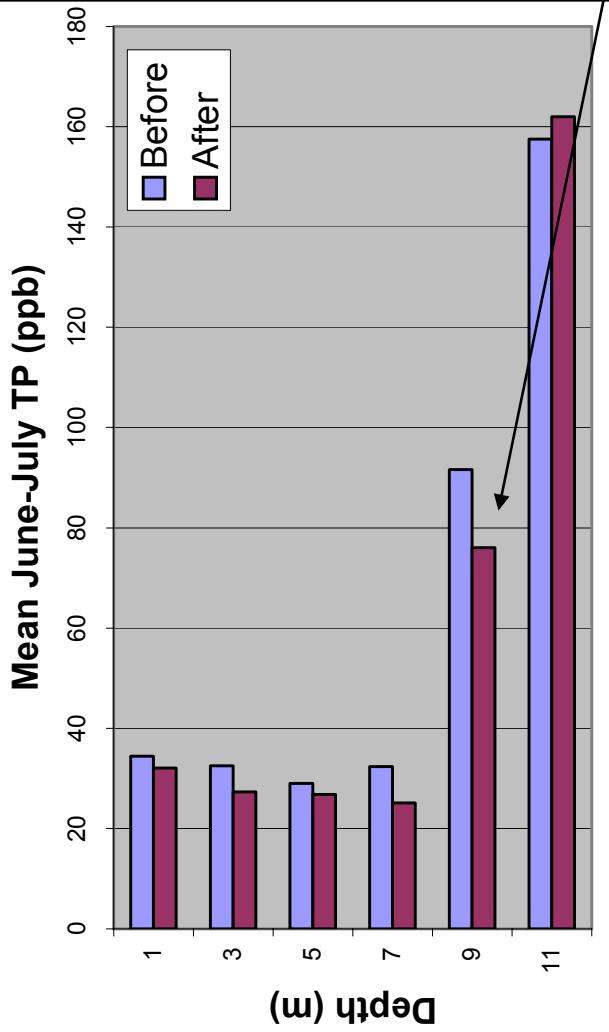
(Akin to an *annual* alum treatment without adding any chemicals.)



In-lake treatment focused on the “nature of the iron cycle”; ferrous iron generated deep in the hypolimnion was oxidized and “rained back down through the hypolimnion”, taking P with it.



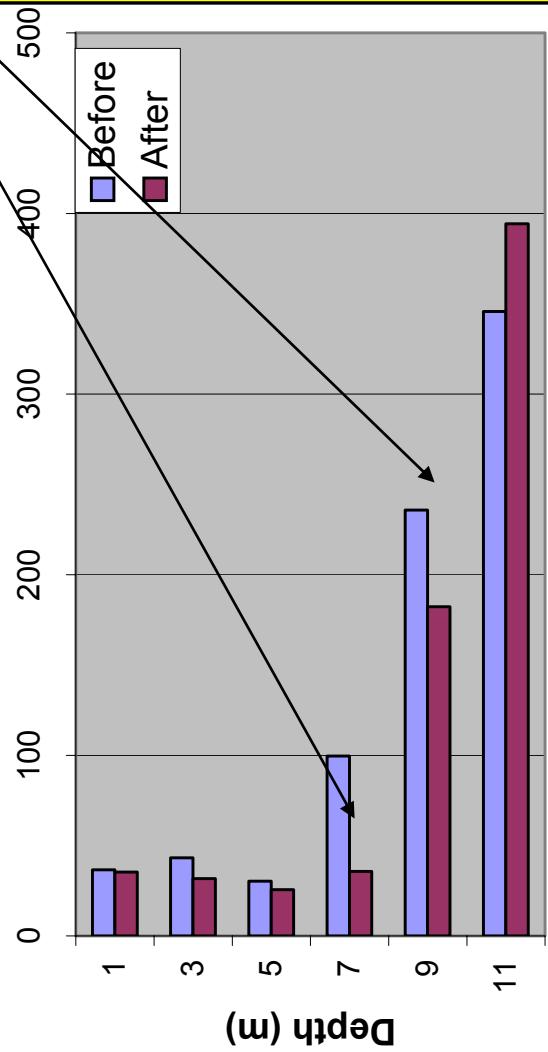
### Total P Lake Waramaug 1980-1987



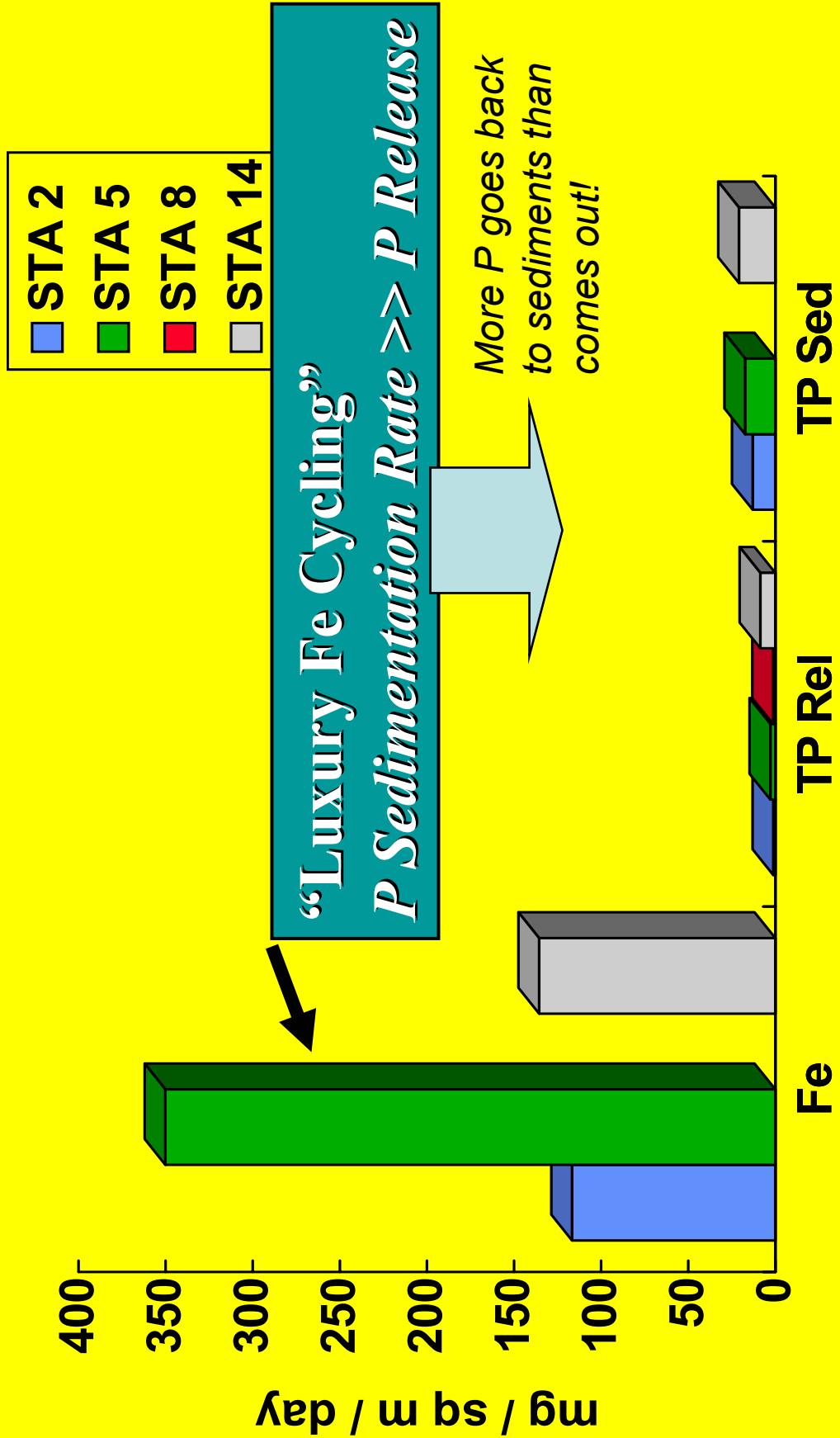
During the initial years of “treatment”, TP concentrations decreased, especially in the upper hypolimnion.

### Total P Lake Waramaug 1980-87

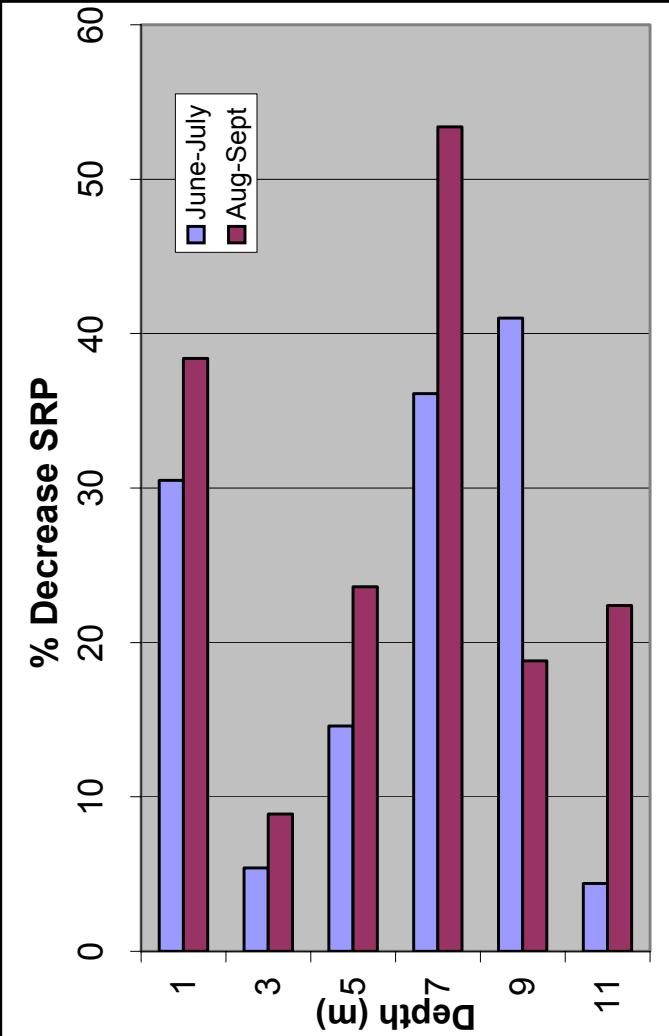
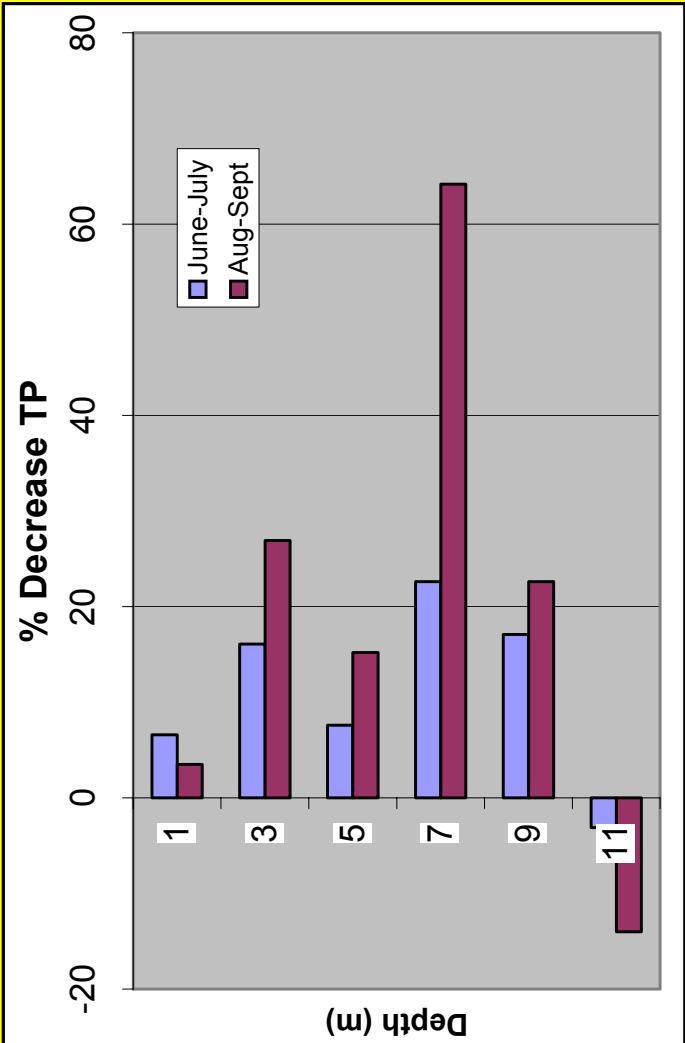
Mean Aug-Sept TP (ppb)



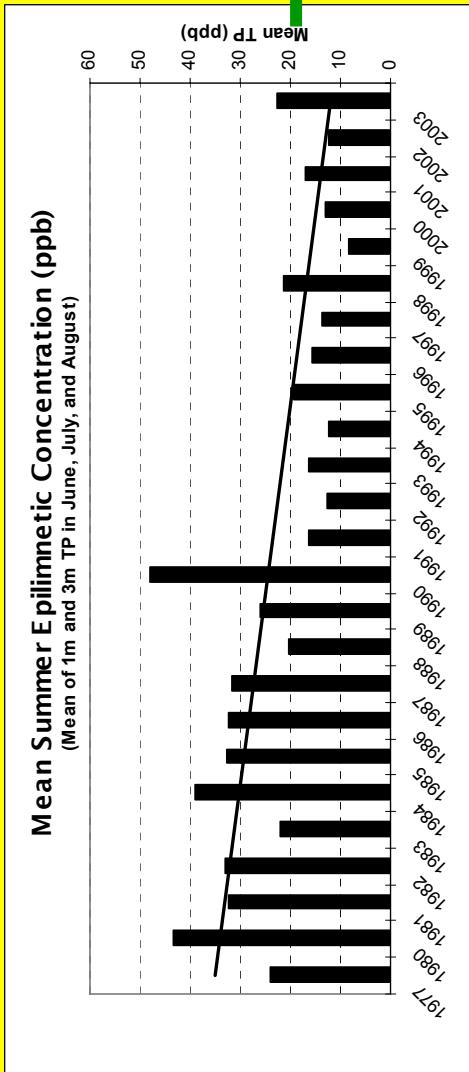
# TP Fe Sedimentation 1994



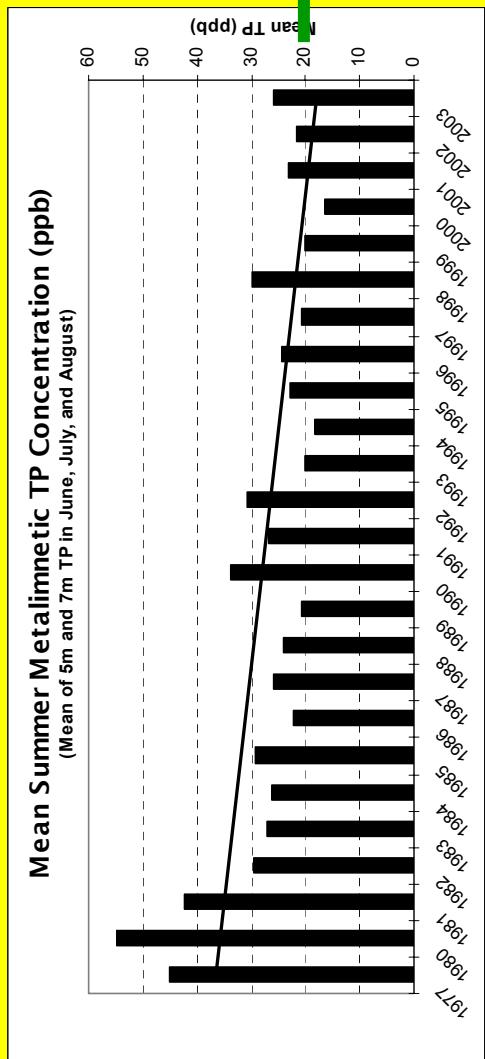
# % Decrease in TP and SRP at Lake Waramaug



In addition to decreased TP concentrations in the upper hypolimnion, *much less of the TP was readily assimilable SRP.*



## LAKE WARAMAUG Mean Summer TP Concentrations



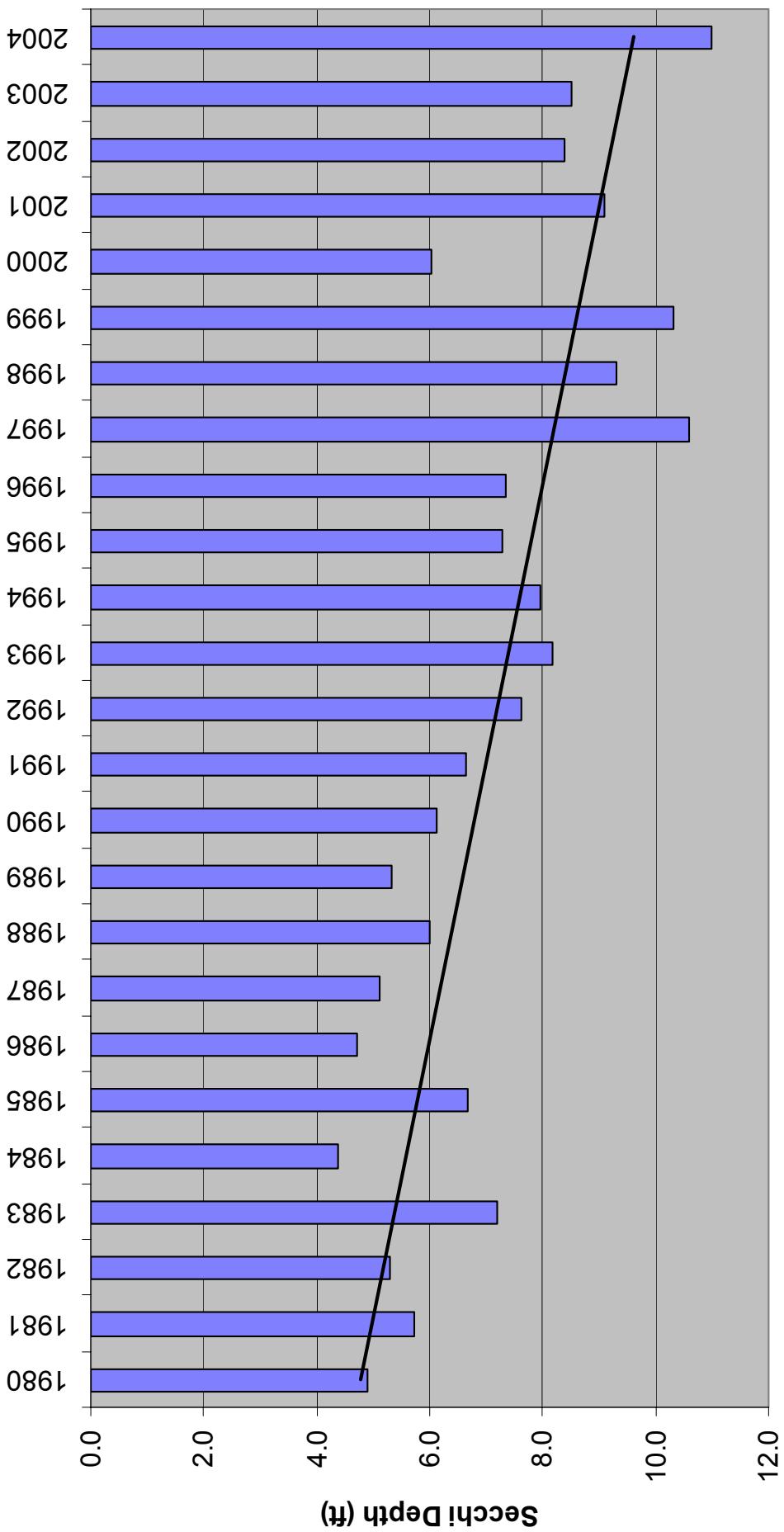
Mid-Depth Water

20 ppb

Surface Water

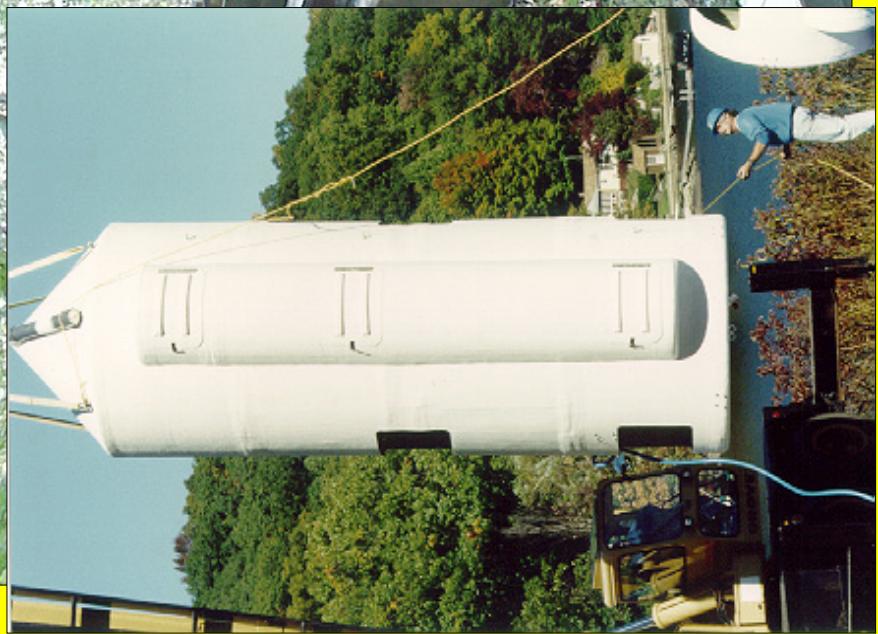
20 ppb

### Mean Summer Transparency - Lake Waramaug 1980 - 2003



## **Robert Frost Hypolimnetic Treatment System**

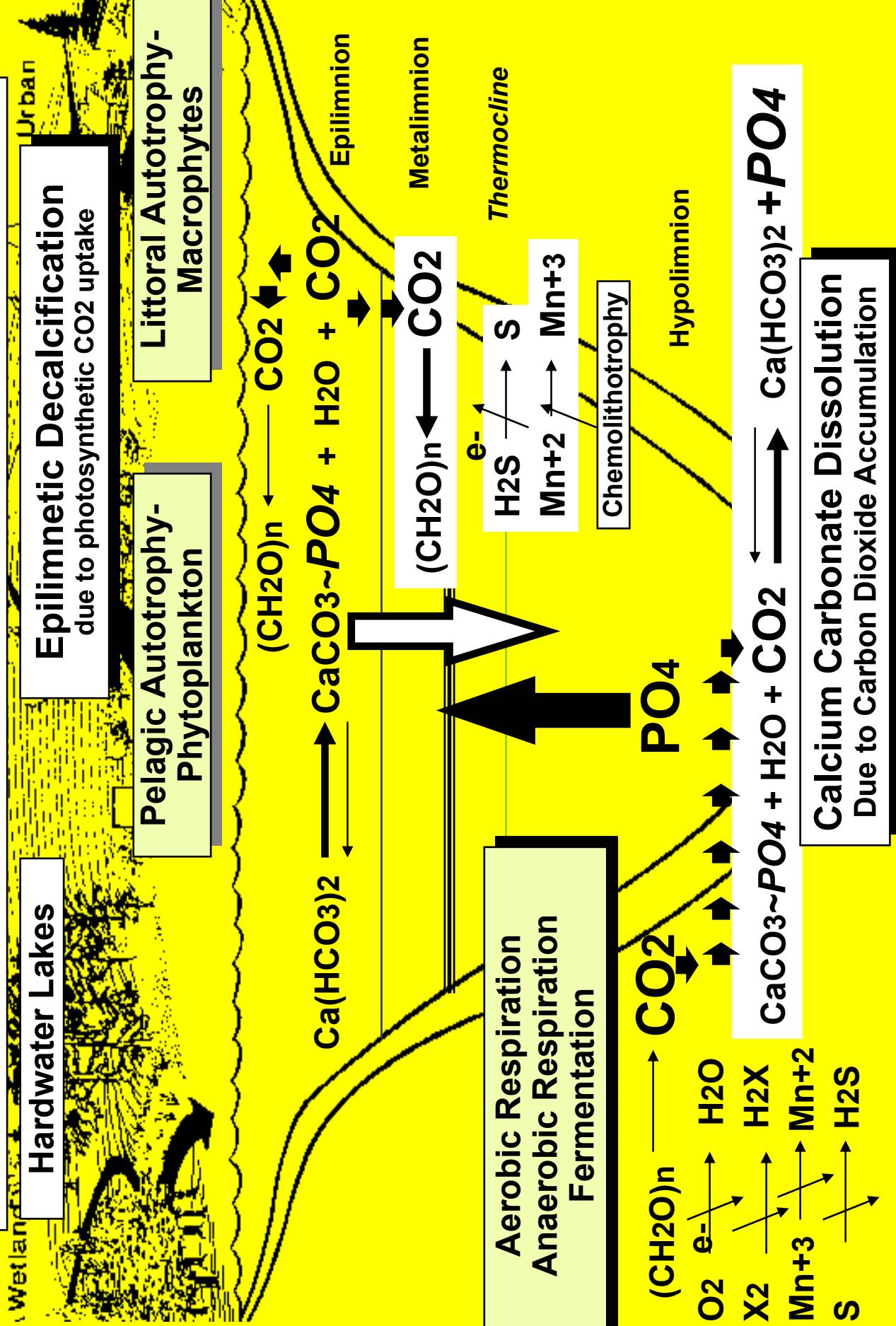
**“Uses iron generated by the lake to remove P from the lake.”**

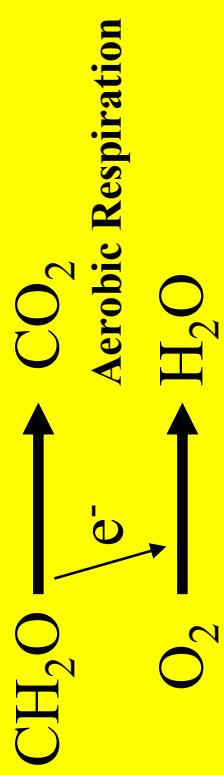


## **Layer Aeration System**

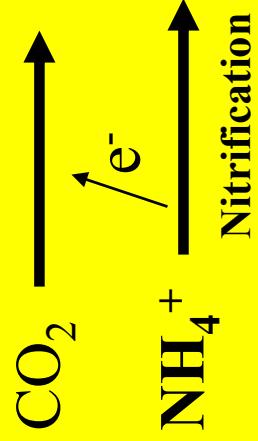
**“Uses oxygen produced by plants and algae to aerate deeper waters.”**

# Carbonate System P Dynamics

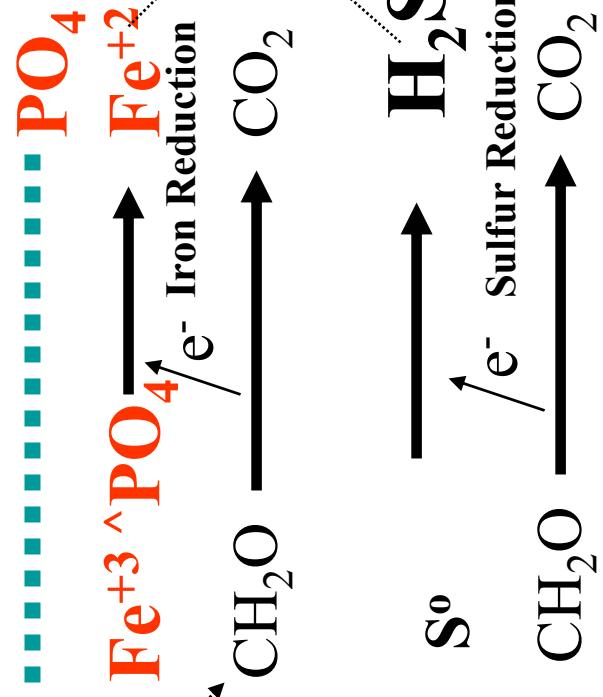
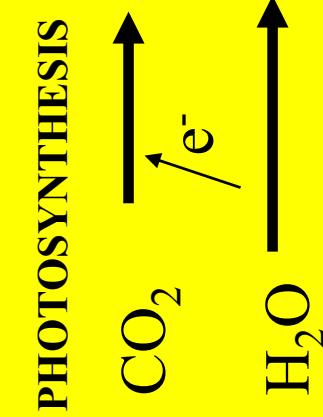




### CHEMOSYNTHESES

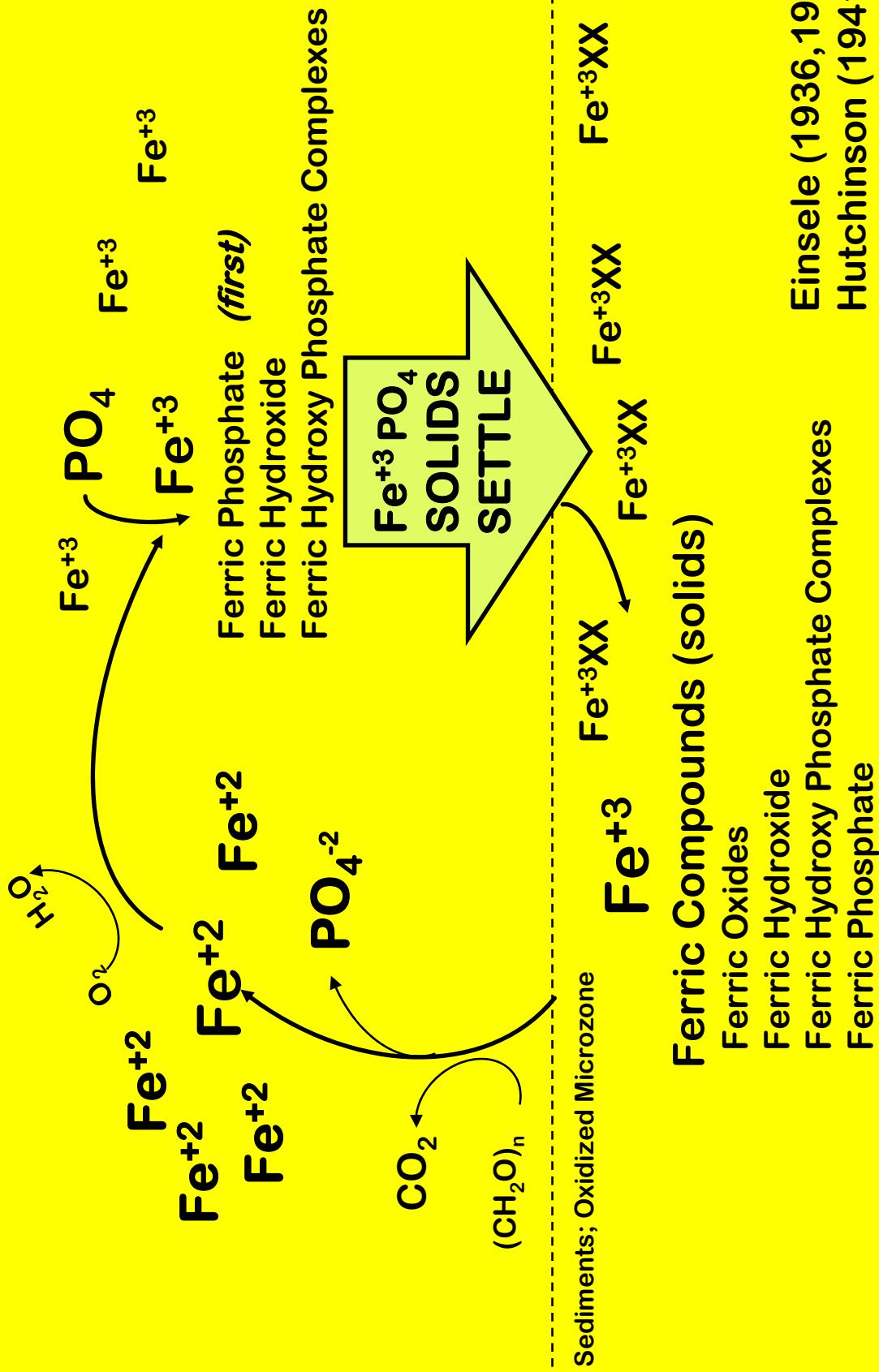


Very little P Release while operating in the N Cycle



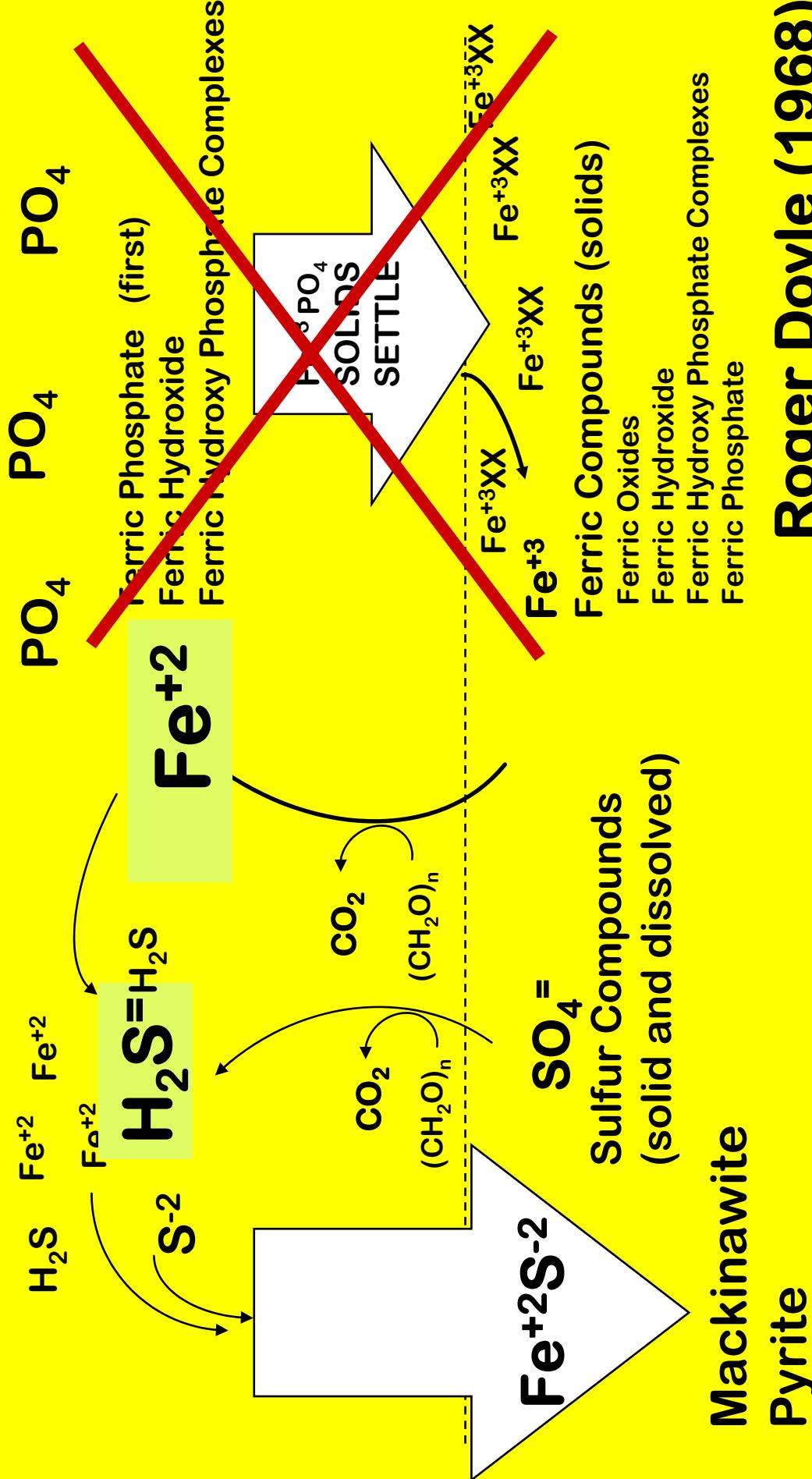
Anaerobic

- Ferrous Iron is typically in excess compared to phosphorus.
- Iron is precipitated as phosphate before hydroxide is formed.



- Ferric Iron gives sediments a “Phosphorus Binding Capacity”.

- Iron no longer participates in anaerobic respiration (it is permanently deposited as ferrous sulfide).
- Iron no longer cycles between the sediments and the water column.
- Iron no longer removes phosphorus from the water column, (phosphorus is left behind in the water).
- More Hydrogen Sulfide production occurs to replace the respiration role previously filled by iron.

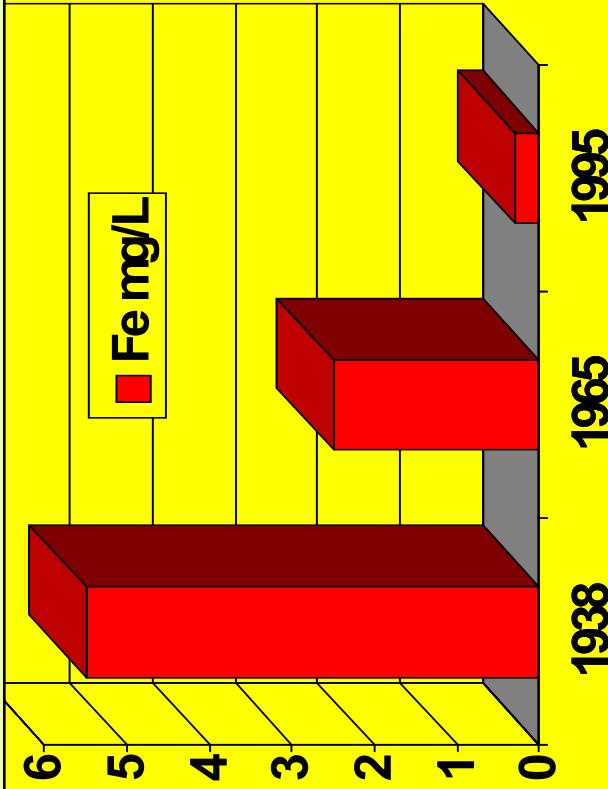


**Roger Doyle (1968)**

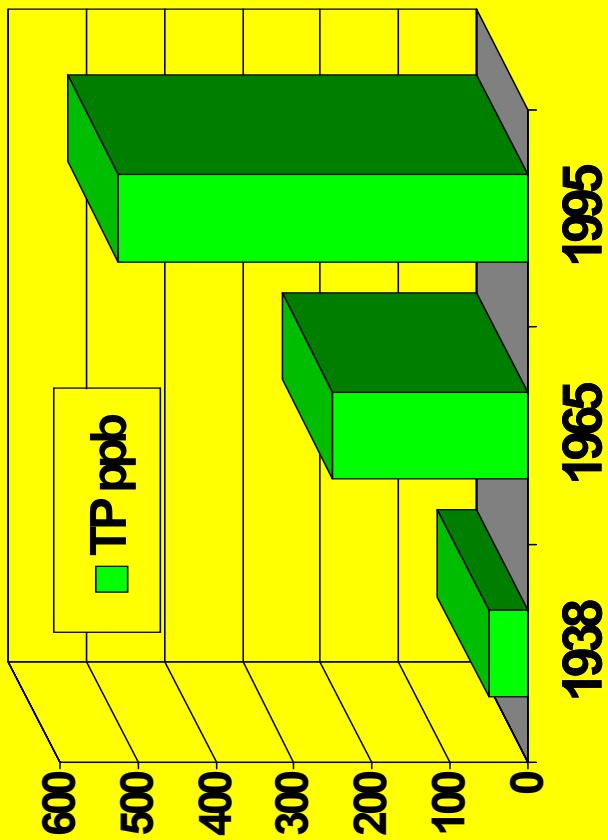
- The Phosphorus Binding Capacity of sediments is lost.
- $\text{AlSO}_4 = \text{ALUM}$ , Might the sulfate load have an adverse impact?

# Linsley Pond 13m Fe TP Concentrations

## Sulfur Cycle Became Dominant



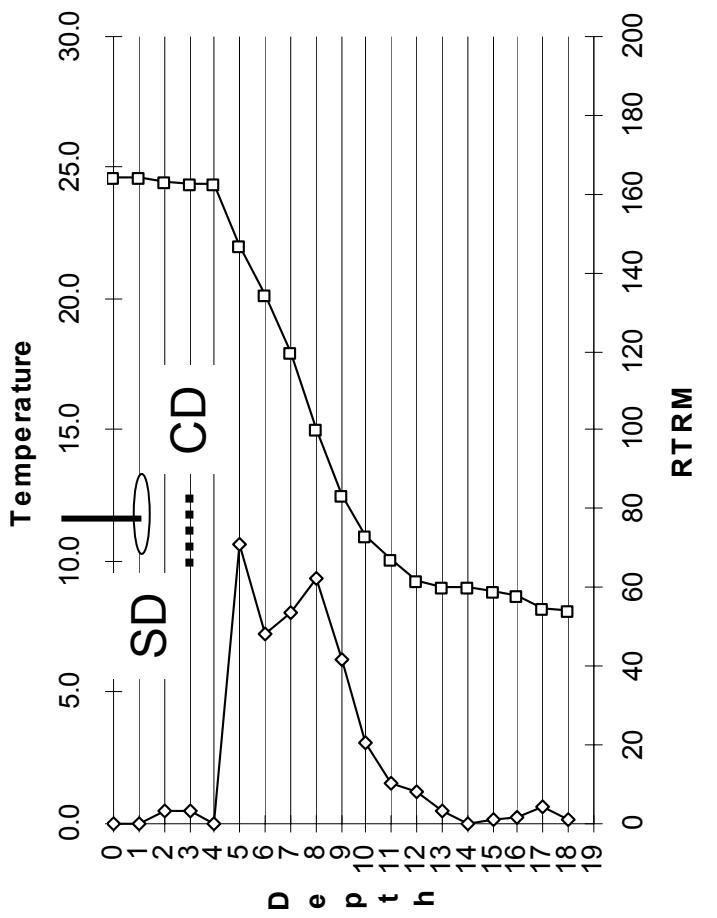
It is “ironic” that this was observed at Roger Doyle’s study lake.



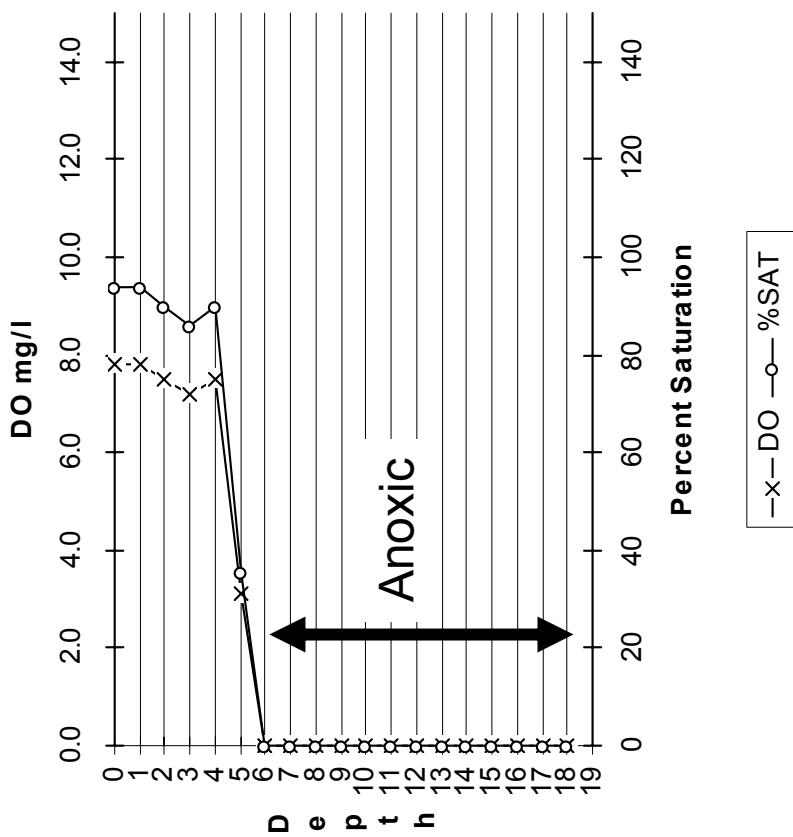
Increasing Eutrophication,  
Increasing Organic Supply (Autoth. And Allochth.),  
Increasing Carbonaceous Oxygen Demand  
Increasing demand for ATEAs...more S= production which permanently removes Fe.

**Loss** of a “Functional Iron Cycle” is a consequence of Eutrophication, and leads to more intense eutrophic conditions.

### Temperature Profile



### Oxygen Profile

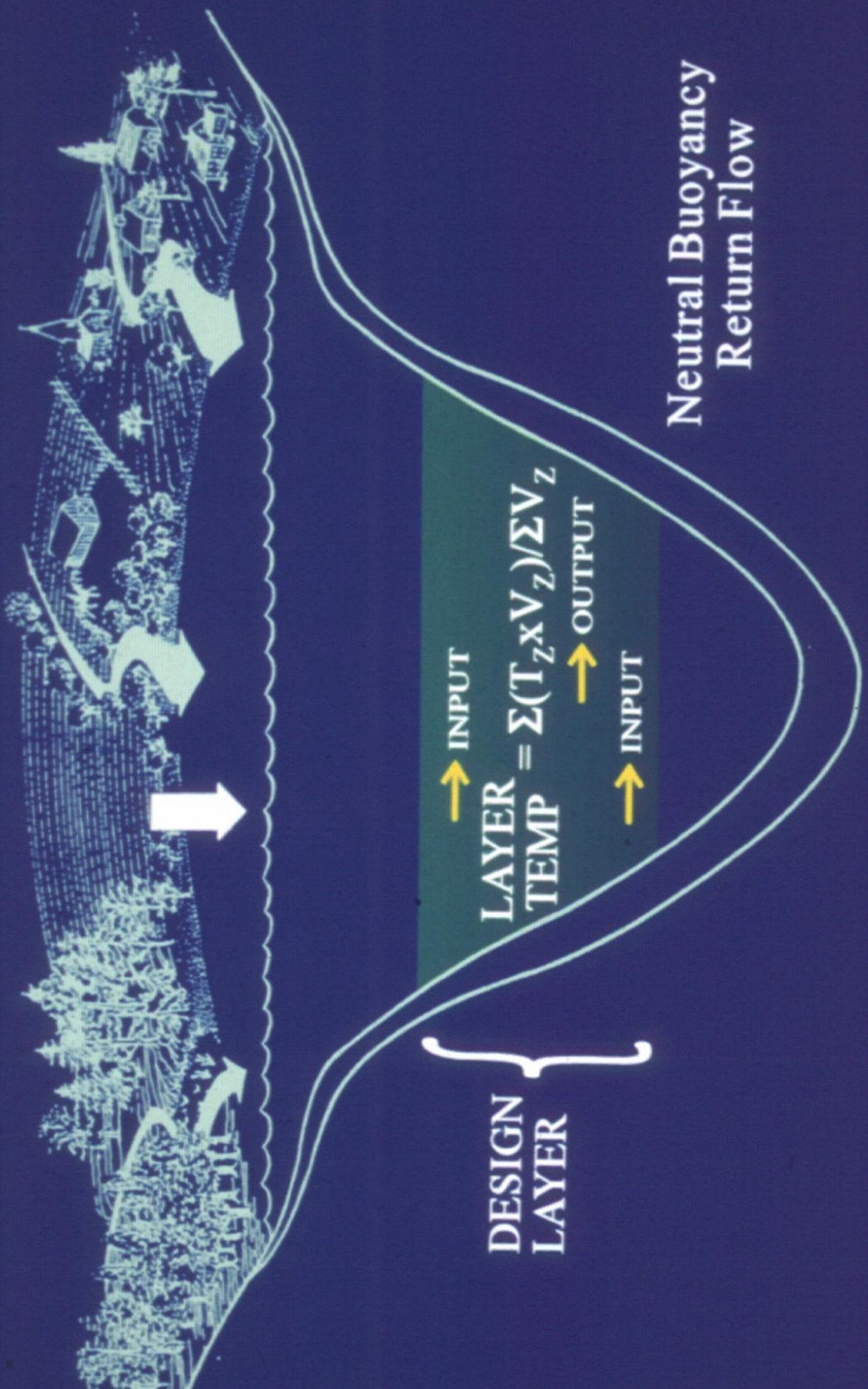


Lake Shennipsit 8/19/1985 (Before Multiple Layer Aeration)

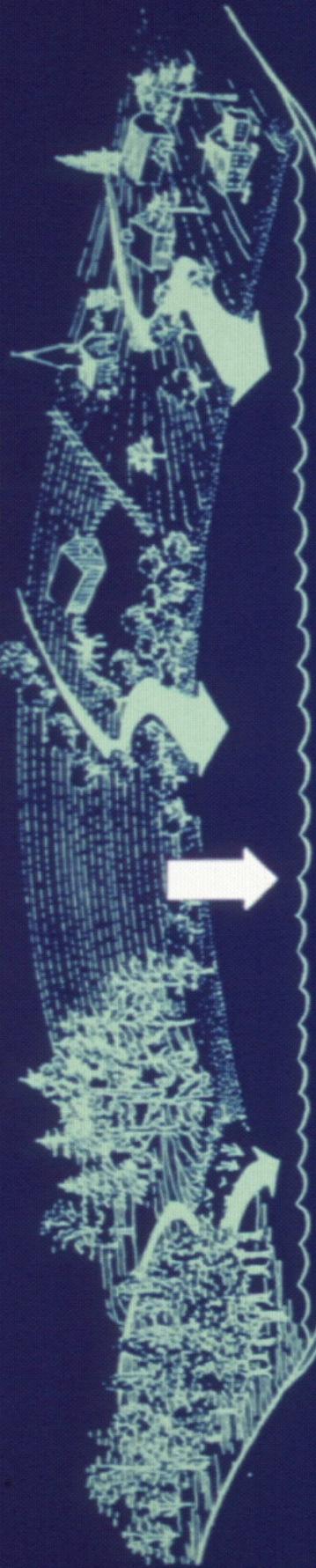
SD= 1.5 m RTRMsum= 333 RTRM Units



# MEAN LAYER TEMPERATURE



# MEAN LAYER DISSOLVED OXYGEN



## DESIGN LAYER

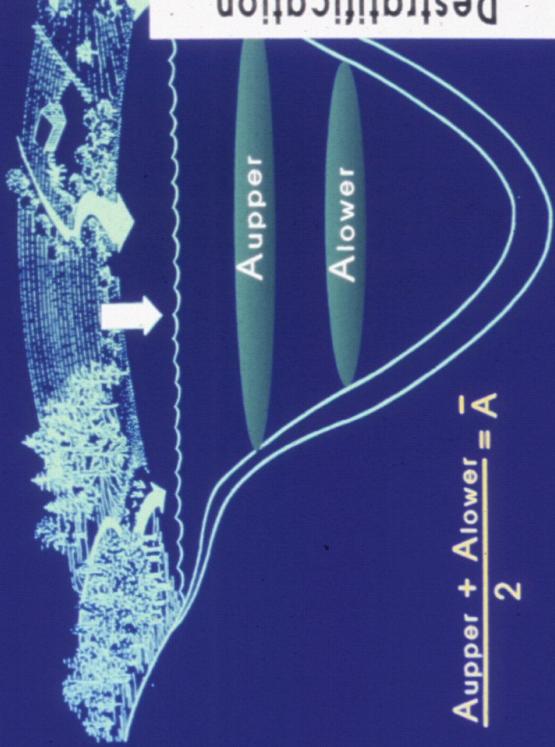
$$\text{MEAN D.O.} = \frac{\sum (D_0 z_i V_i)}{\sum V_i}$$

→ INPUT      → OUTPUT

Exclusive of  
(Solute Phase Transfer)

Expected 1.5 - 4.0 mg/L

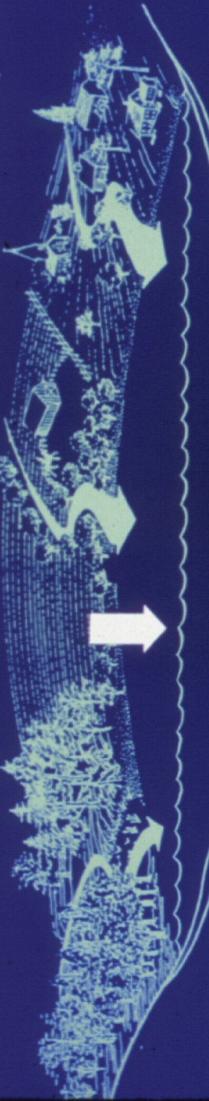
## LAYER AERATION : MORPHOMETRY BASED SIZING



**Depth-Discrete Circulation Capacity estimated from empirical artificial circulation case results; using area of the layer.**



$$\text{TOTAL OXYGEN DEMAND} = ( \text{SOD} \times \text{Asc} ) + ( \text{CODflux} * \text{A}_{lower} ) + \text{TOD( vol )}$$



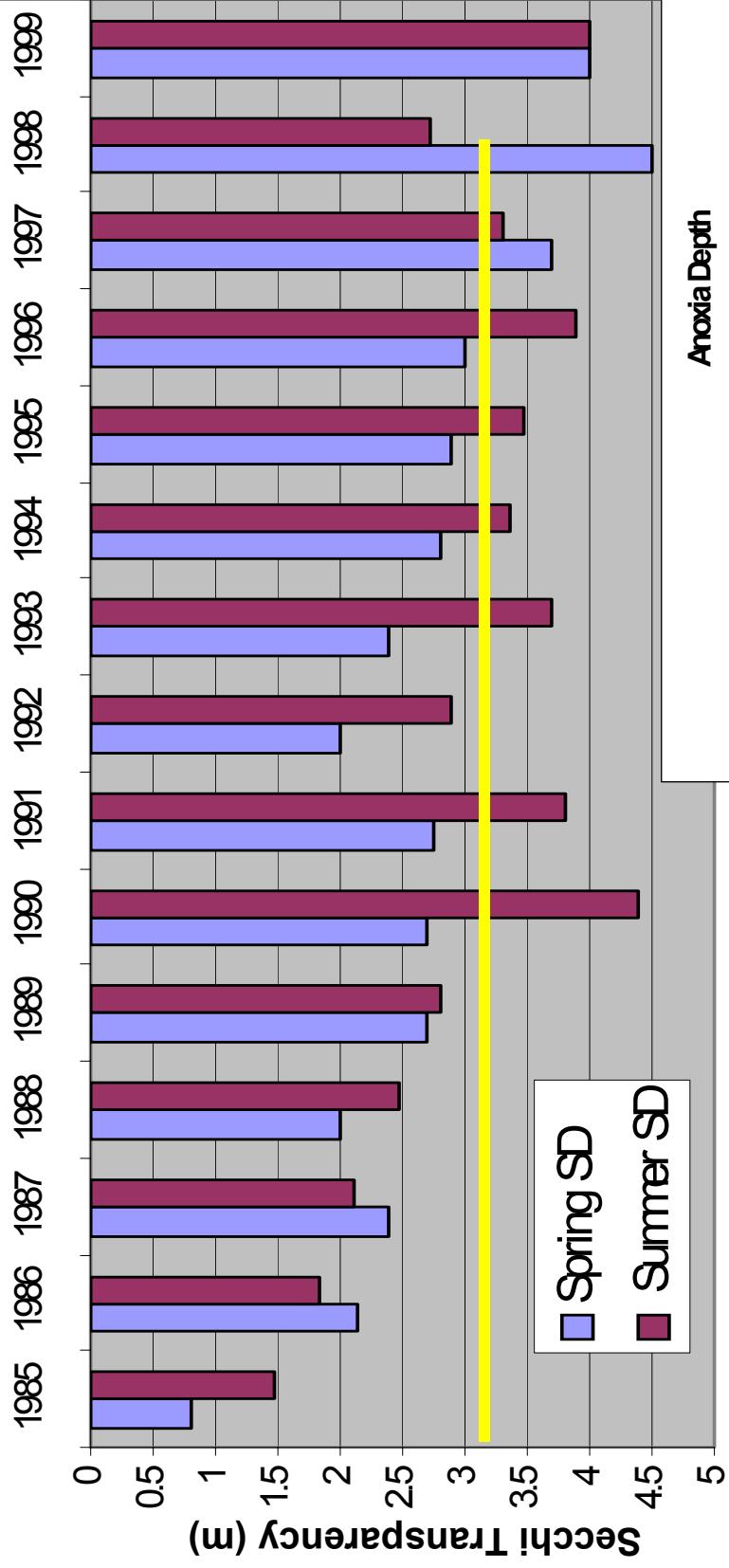
- Volumetric Oxygen Demand
- Sediment Oxygen Demand
- COD Flux Rates
- Nitrogenous Oxygen Demand

**How much more DO needs to enter solute phase?**

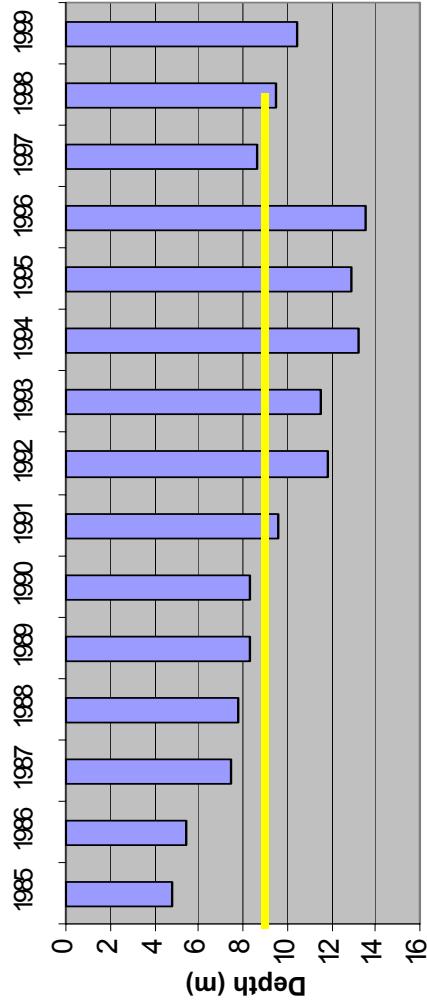


*If it lives, it combusts CH<sub>2</sub>O to CO<sub>2</sub>*

# Lake Shennipsit



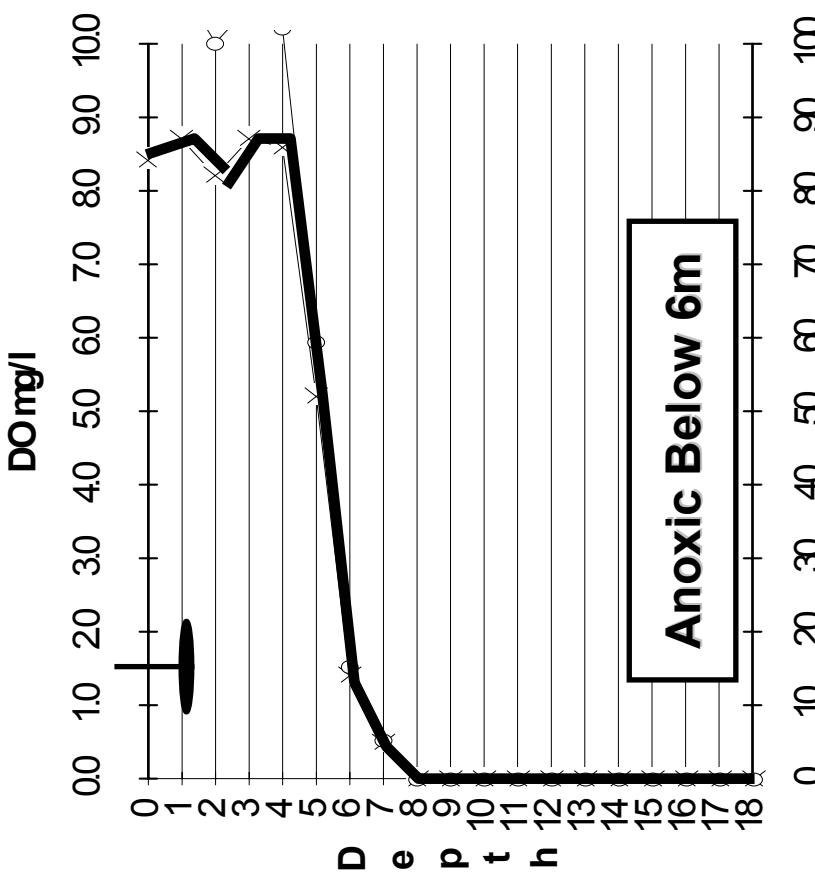
Anoxia Depth



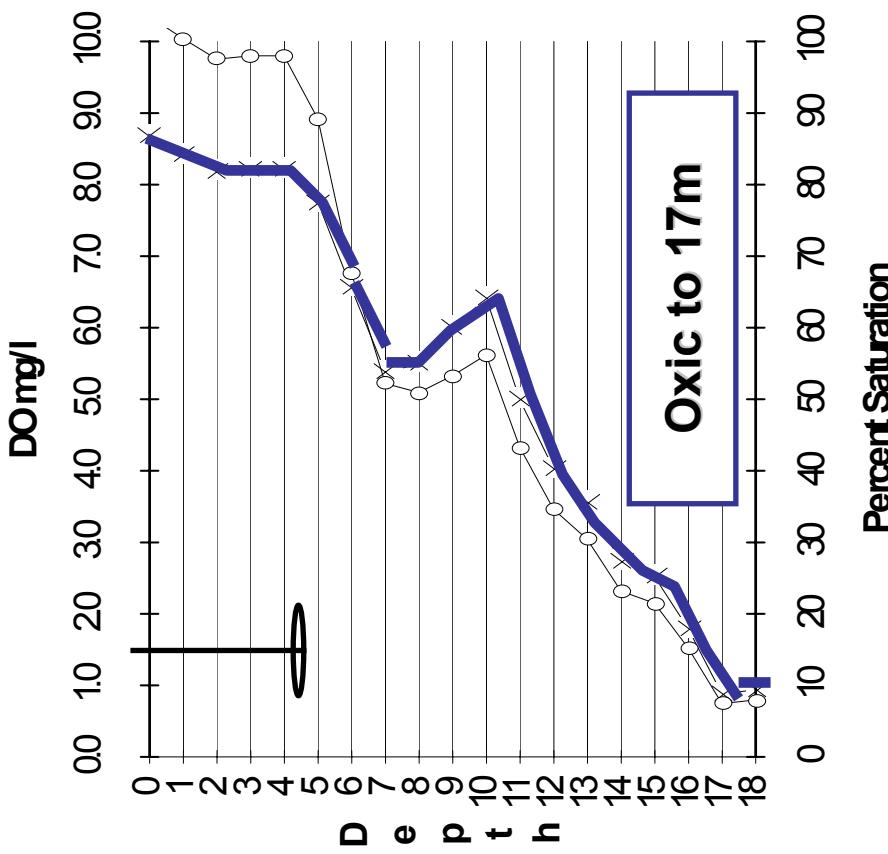
## Layer / Hypolimnetic Aeration – Effect of Dissolved Oxygen Distribution

**July 22, 1985****Secchi= 1.3m**

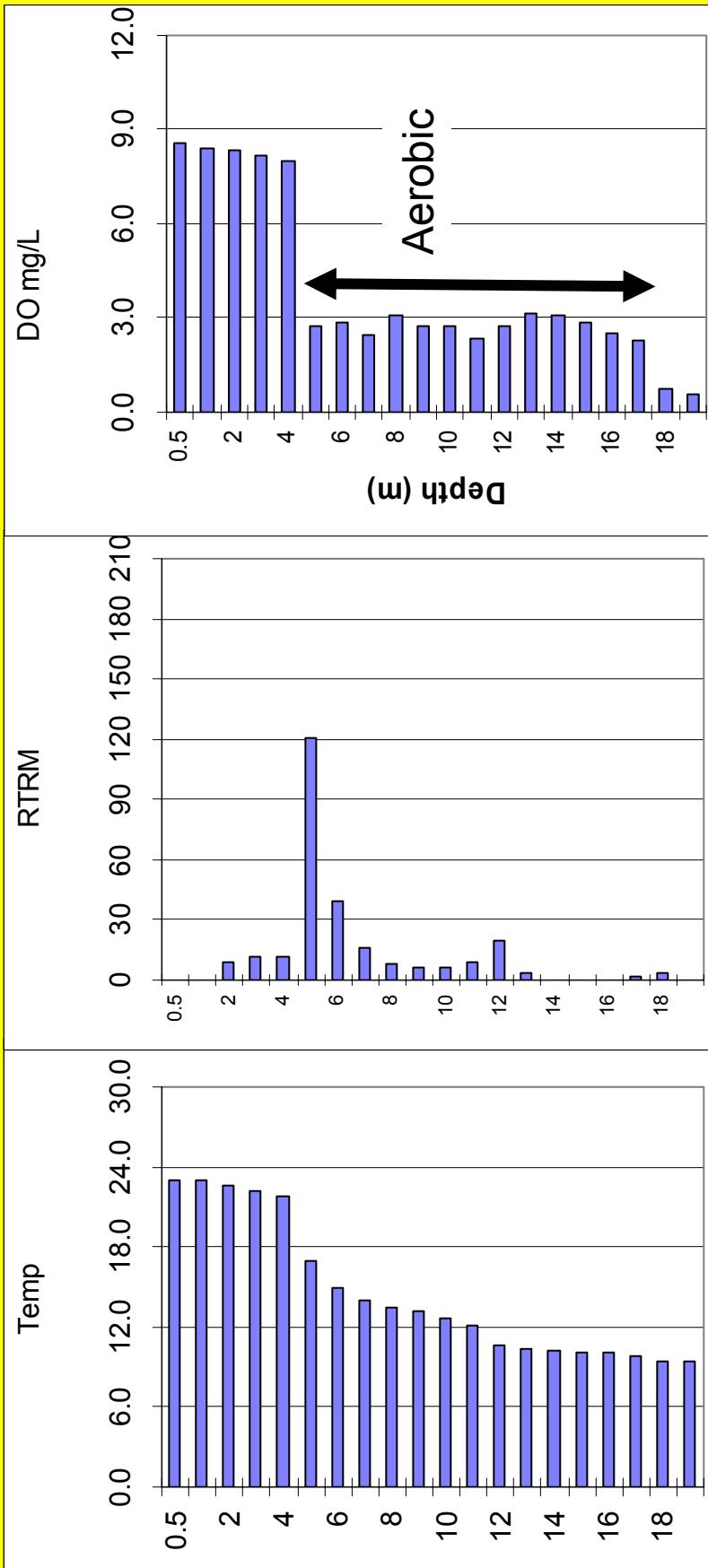
## Oxygen Profile

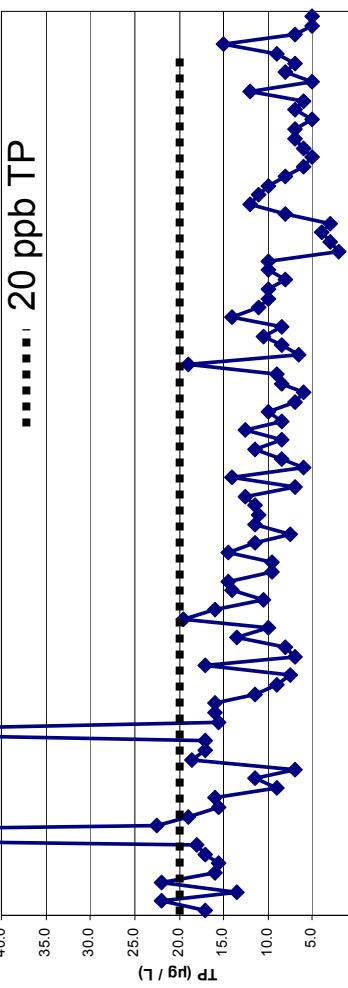
**Percent Saturation****—X— DO    —○— %SAT****July 22, 1993****Secchi= 4.0m**

## Oxygen Profile

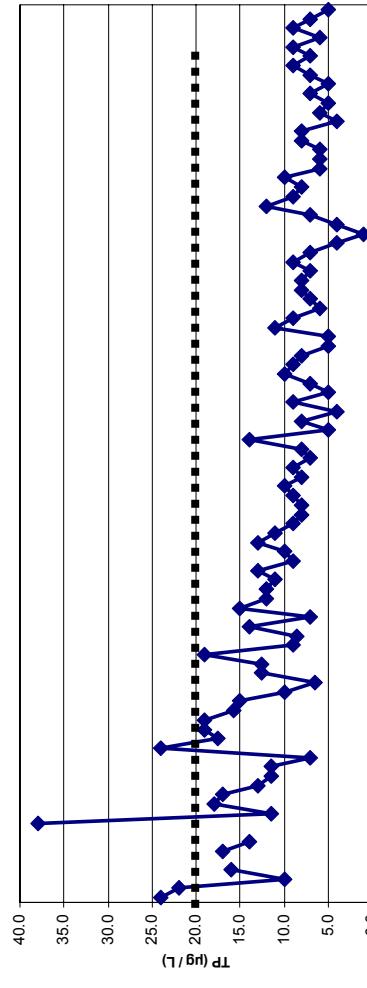
**Percent Saturation****—X— DO    —○— %SAT**

# Lake Shenipsit 2000 (During Multiple Layer Aeration)

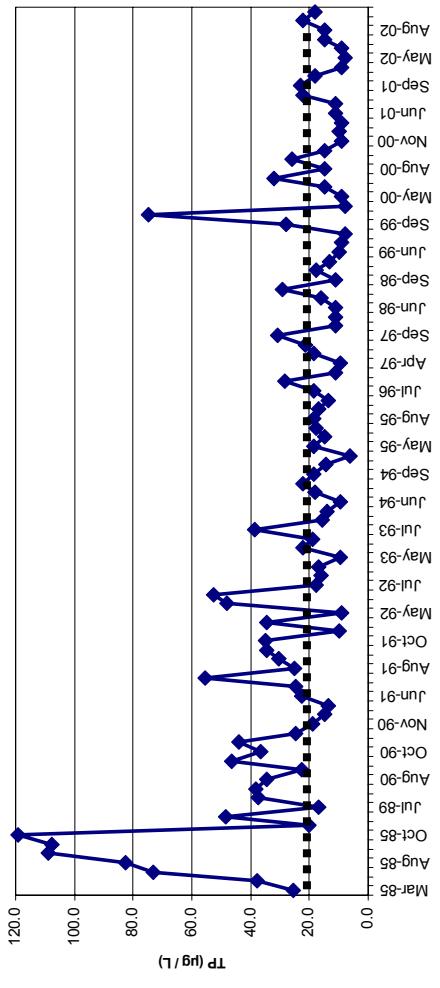




Shenipsit Reservoir Total Phosphorus Concentrations of the Metalimnion



Shenipsit Reservoir Total Phosphorus Concentrations of the Hypolimnion



1985

2002

TP Hypolimnion  
("Anaerobic Aeration")

TP Epilimnion

TP Metalimnion  
(Aerated Layer)

## **Lake Shenipsit: Additional Effects on Ecosystem Structure and Function:**

- **3000+ acre-ft of CW Habitat Restored**
  - **Fish (piscivory, zooplanktivory)**
  - **Herbivorous Zooplankton Refuge**
- **Increased Nitrification (less Ammonia-N)**
- **Lower TOC, lower pH peaks**
- **Deepened SD and CD**



**Ecosystem**  
Consulting Service, Inc.

## BTMUA Reservoir Aeration Management System



## Supply Intakes

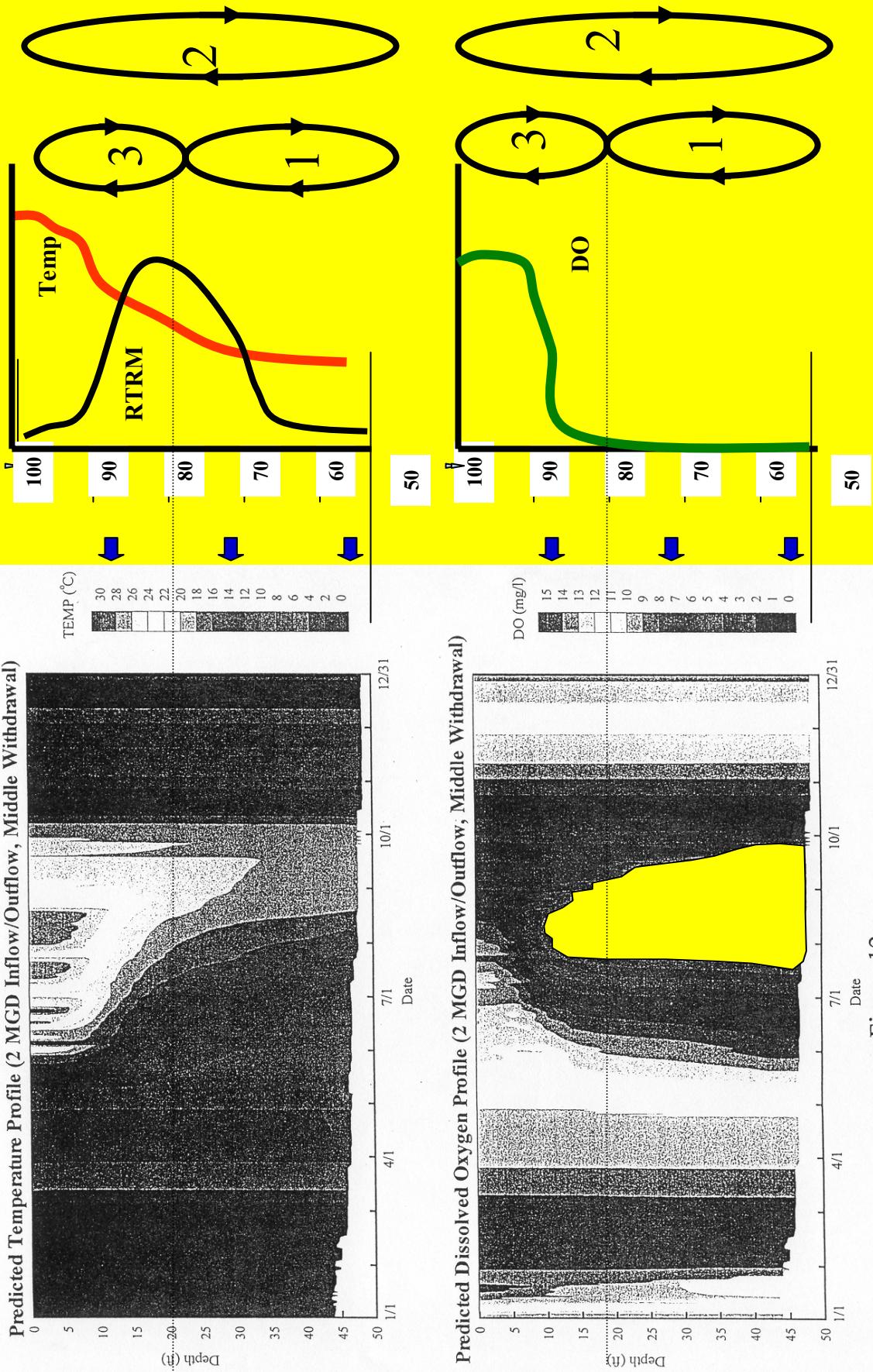
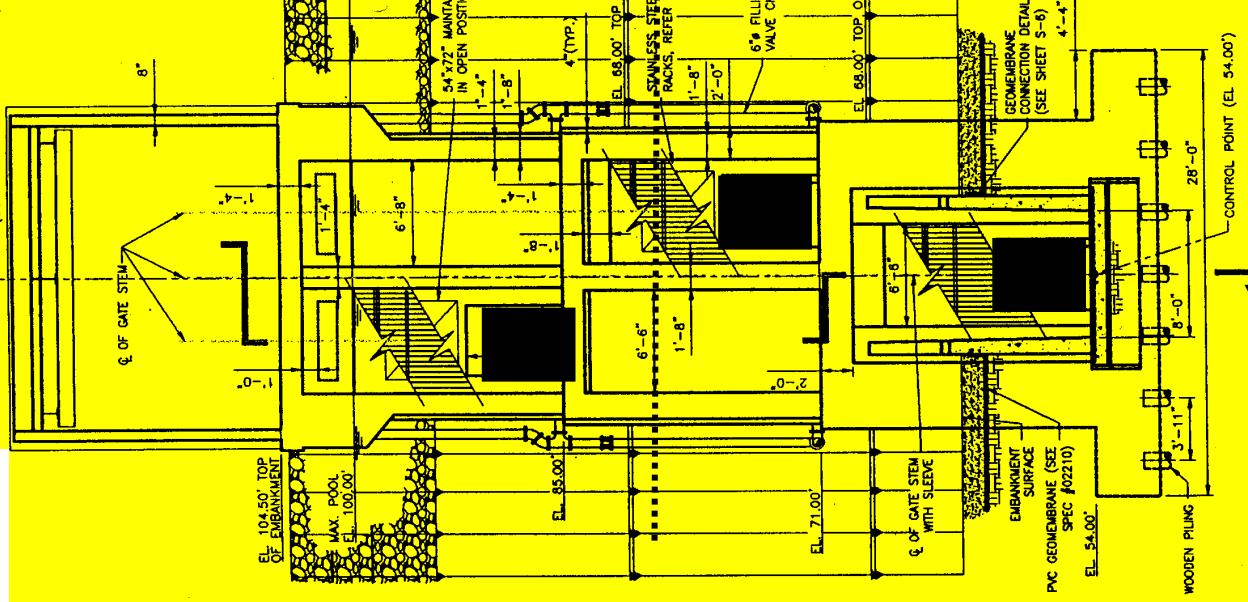


Figure 12

# Process Schematic – BTMUA Storage Reservoir Management / Treatment System

**Ecosystem**  
Consulting Service, Inc.

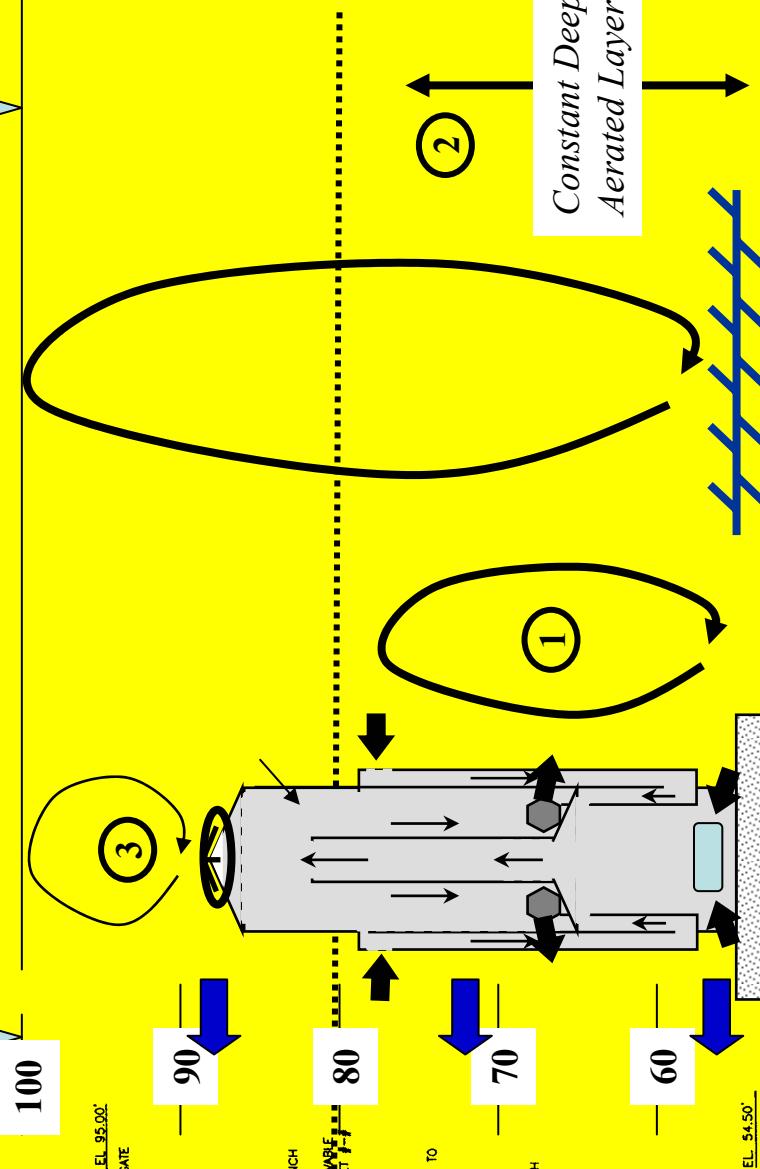


## Layer Aeration (Partial Lift)

Add a ChemFeed  
Manifold to the  
diffused vent release

Artificial Circulation

**Full Artificial Circulation Diffuser Module  
with Dual Chem Feed  
Installed at each Layer Aerator Slab**



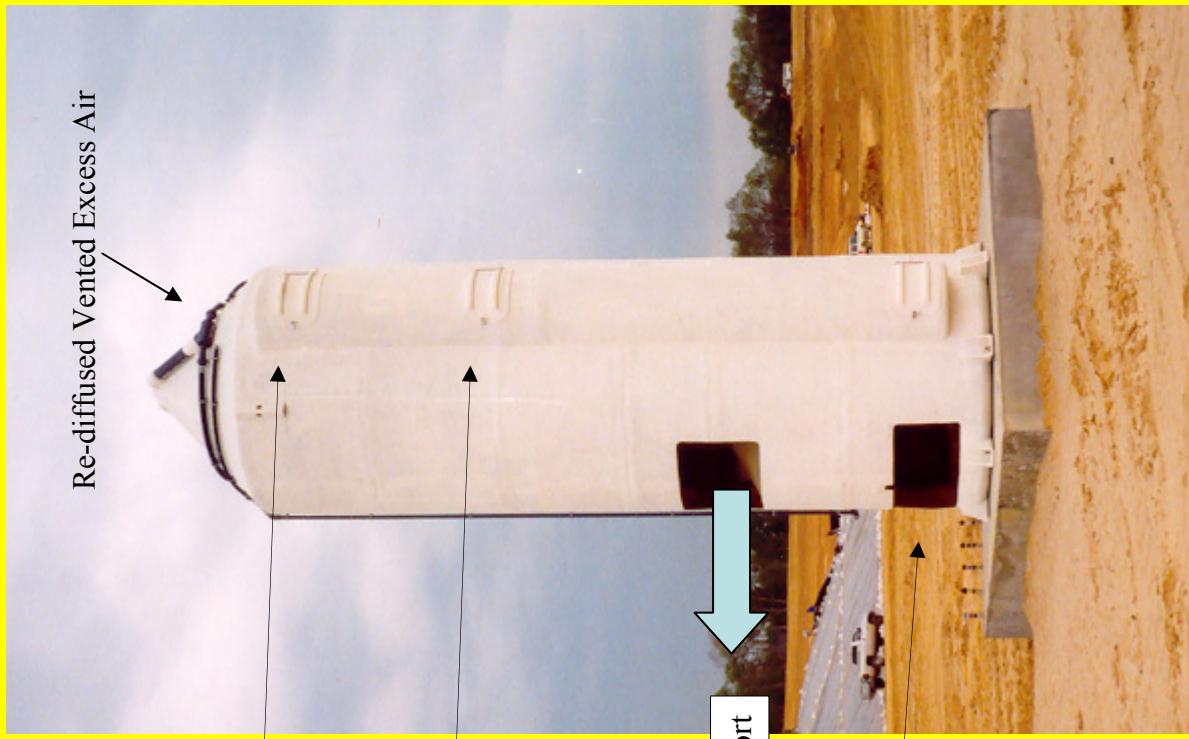
5

### *Functions Provided:*

1. Layer Aeration below elev. 78ft with Dual ChemFeed
  2. Complete Artificial Circulation down to elev. 50 ft with Dual Chem Feed
  3. Enhanced Epilimnetic Circulation down to elev. 80ft (using vented air)

## LayerAir PIP8 Layer Aerator

Three shallow intakes are available to establish the thickness of the aerated layer. The deep intake is always used. The initial set-up has the upper shallow intake port open (other two are closed). As lake data is collected, the upper intake could be closed and the middle intake opened which will focus aeration on a more narrow water layer. Opening only the lower shallow intake (or closing all three intakes) turns the aerator function into a “Hypolimnetic” mode of operation.

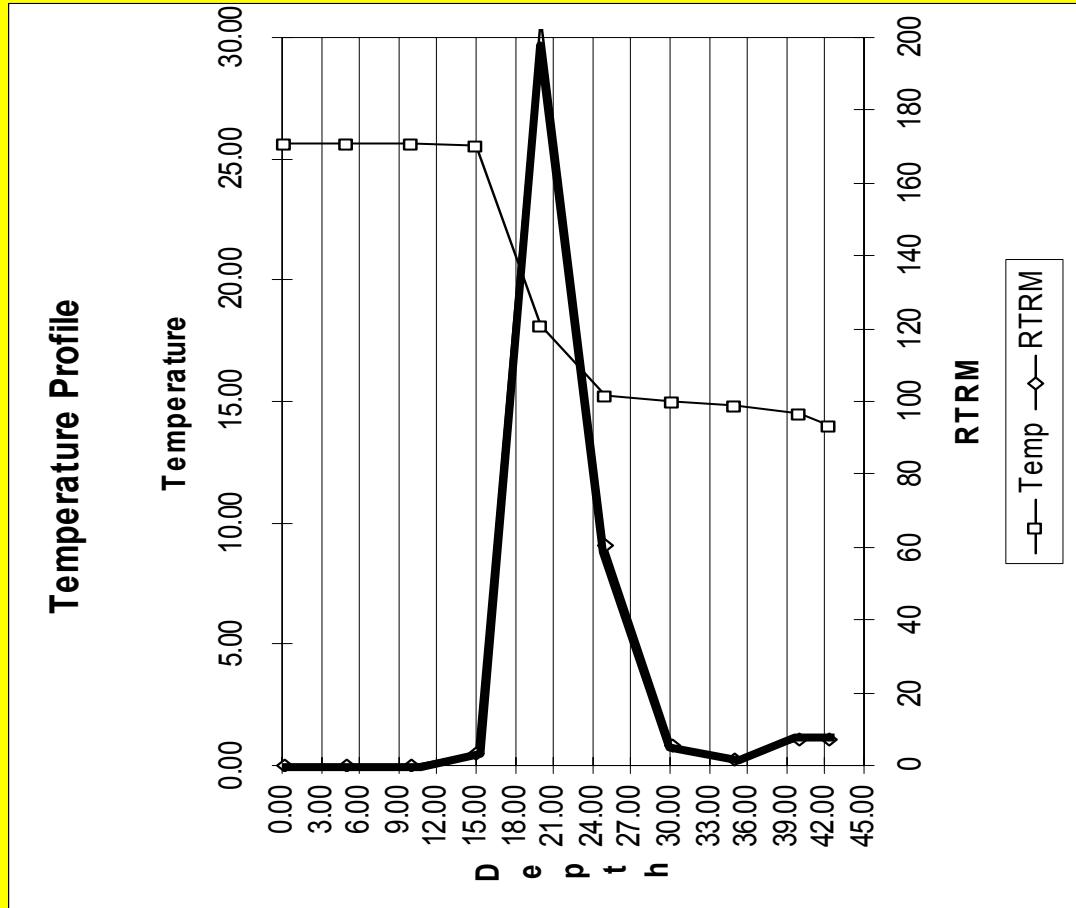
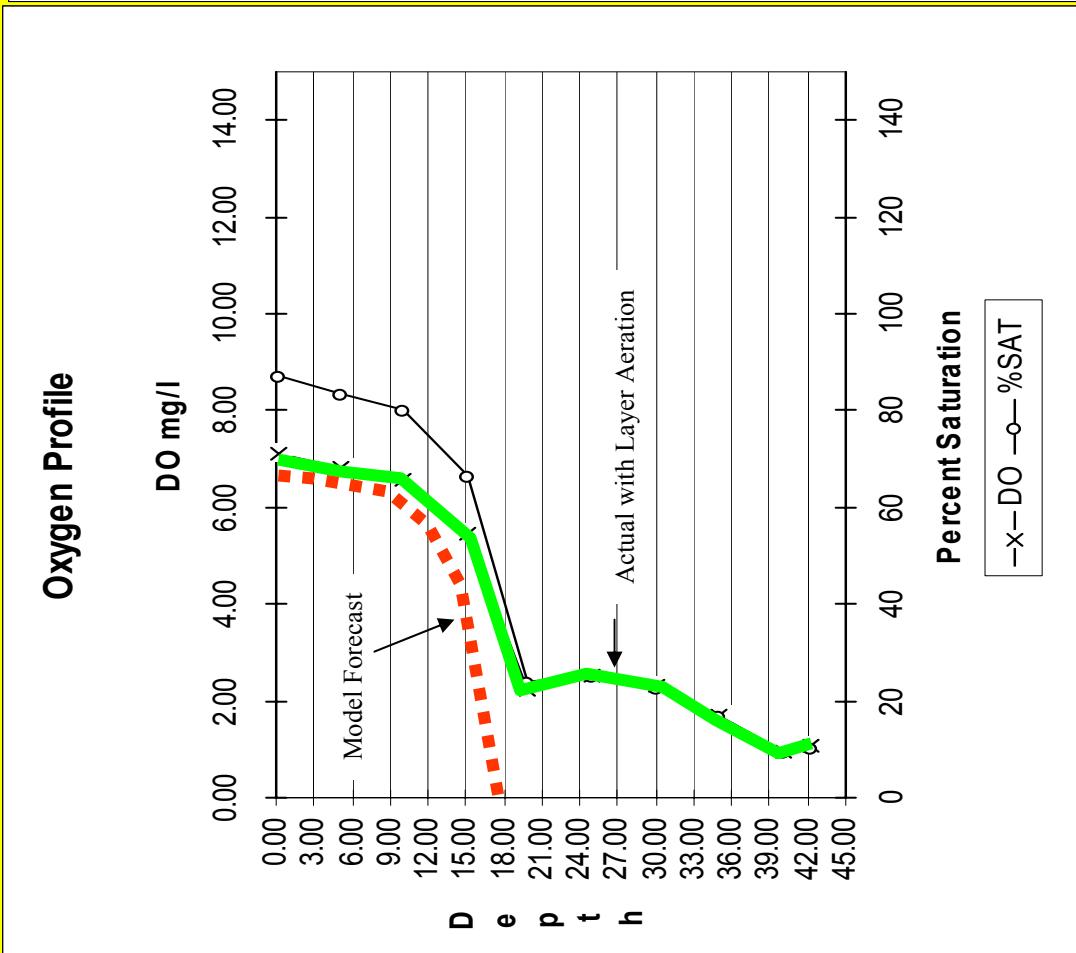


## Chemical Feed Options

1. Into the Layer Aeration Tower ( to treat the deep layer),
2. Into the re-diffused vent air plume ( to treat the surface layer),
3. Into the Full Artificial Circulation Diffuser Module (to mix and treat the entire water column).

All operational modifications and treatments can be performed from the land-based compressor facility.

# BTMUA Reservoir R-8 8/6/2004



**Remained Aerobic;** would have been anoxic below 21 ft without aeration.

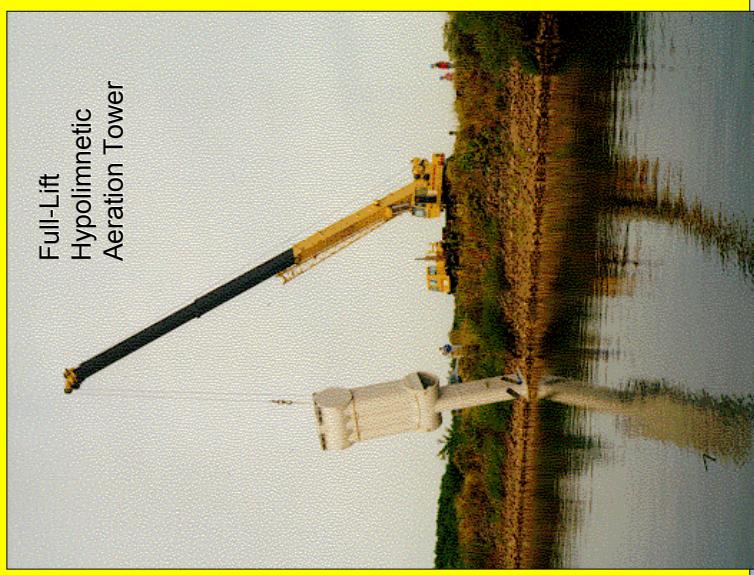
**Very Strong Thermocline @ 21 ft,**  
Good circulation of surface and deep layers.



Partial-Lift (submerged)  
Layer Aeration Tower



Full-Lift  
Hypolimnetic  
Aeration Tower

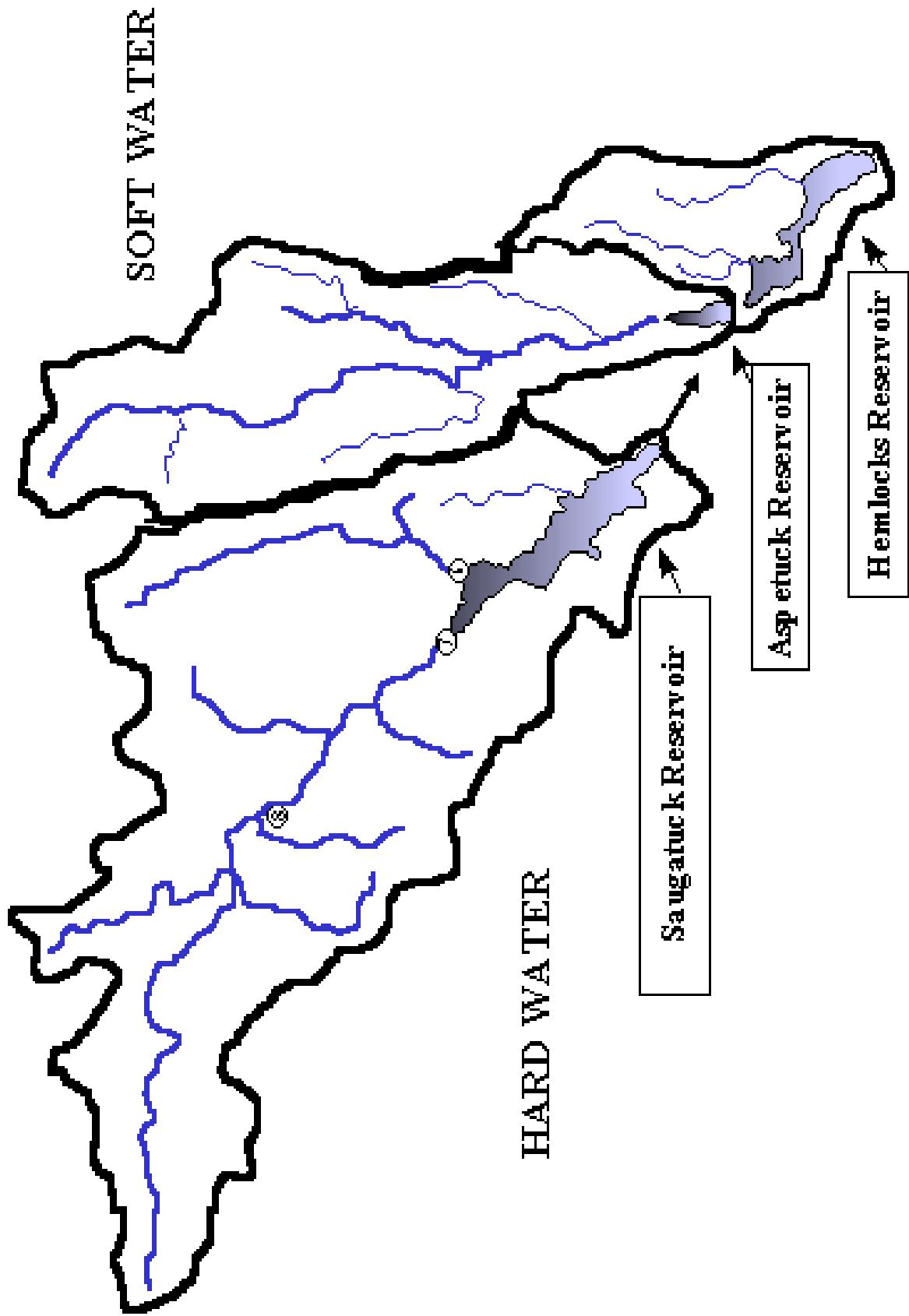


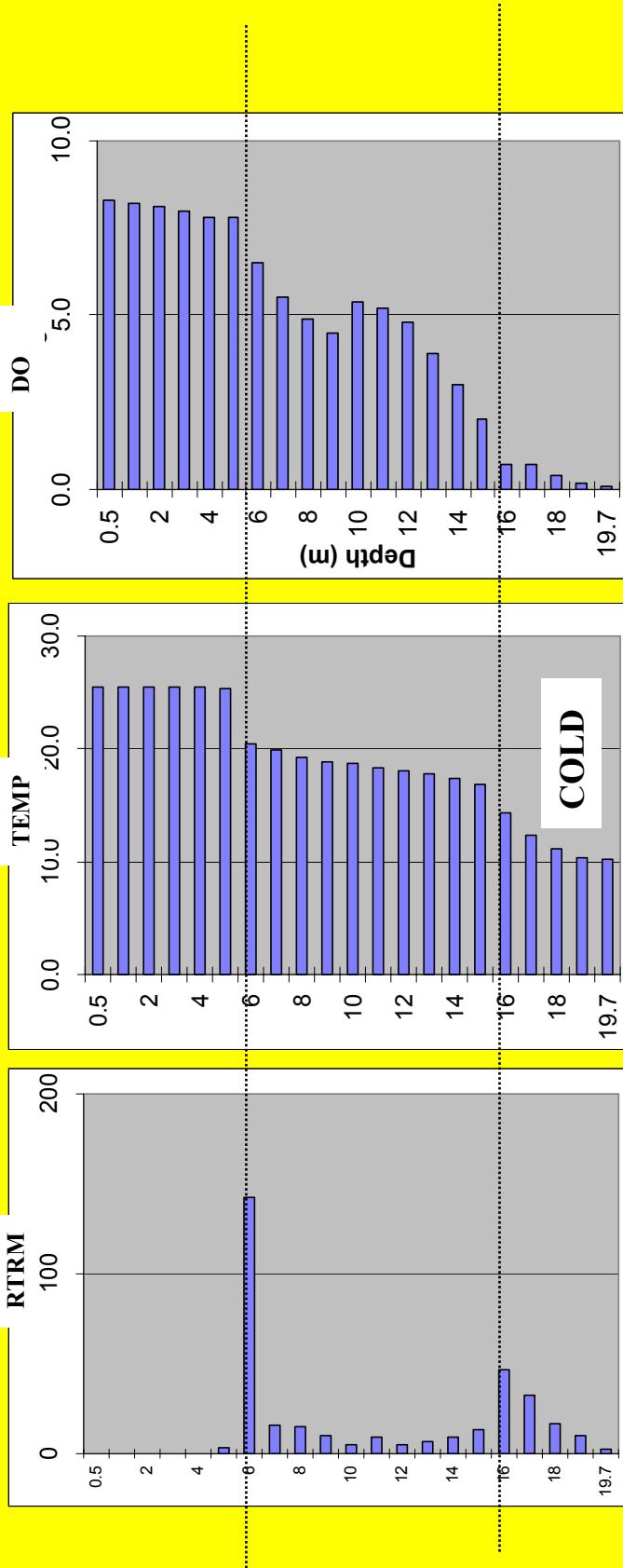
Full-Lift  
Hypolimnetic  
Aeration Tower

**A few Examples of Aeration  
Towers for Layer Aeration or  
Hypolimnetic Aeration.**

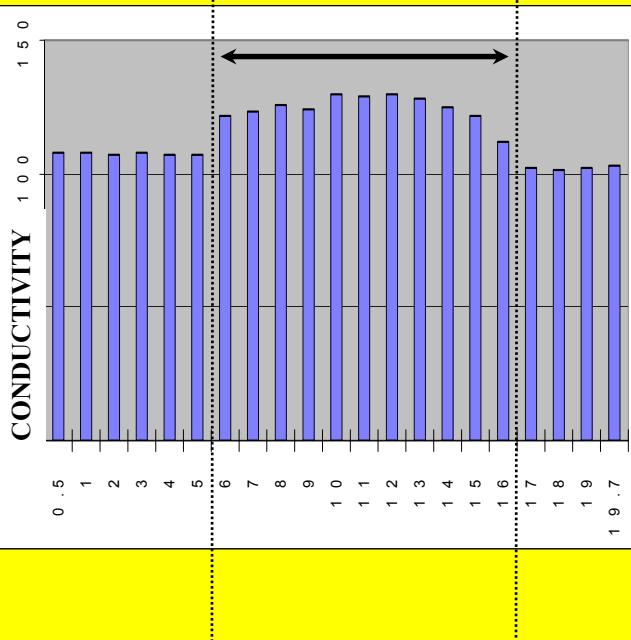
Partial-Lift (submerged)  
Layer Aeration Tower  
Installed during  
construction of a new off-  
line storage reservoir







## Hemlocks Reservoir August 19, 1998



Saugatuck Reservoir water, released from about 50 feet deep, flows through Aspetuck Reservoir as an “Underflow”, and then it flows through Hemlocks Reservoir as an “Interflow” between thermoclines at 6 and 16 meters deep.

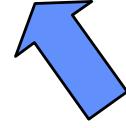
## **Seasonal Source System Optimization – Hemlocks System (Based on Water Quality – Quantity needs balancing)**

**“Season 1”**  
**Winter**

**Approx. December 15 to March 15**

- Use the Aspetuck River source to maintain/refill Hemlocks Reservoir;
- Winter Drawdown of Hemlocks for Milfoil Control (option)
- Winter Circulation for Alewife Control (option)
- Allow Refill of Saugatuck Reservoir Storage

**Utilize Aspetuck River during Higher Flow/Water Quality Season  
(while refilling Saugatuck Storage Volume)**



**“Season 2”**  
**Spring**

**Approx. October 15 to December 15**

(From the beginning of leaf-fall)

- Consume Hemlocks Storage
- Close the Aspetuck Reservoir Gate;
- Close the diversion from Saugatuck Reservoir;
- Allow refill of Saugatuck Storage

**Avoid High Color, TOC/DOC, in Aspetuck River**

**“Season 4”**  
**Autumn**



**Approx. March 15 to May 15**

- Close the Aspetuck Reservoir Gate;
- Consume Hemlocks Reservoir Storage;
- Allow Aspetuck Dam Overflow
- Allow Refill of Saugatuck Reservoir Storage

**Minimize Spring Nutrient Input to Hemlocks Reservoir**

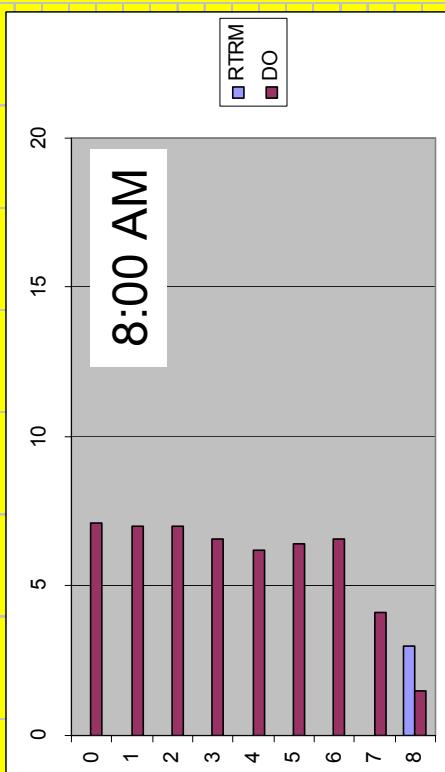
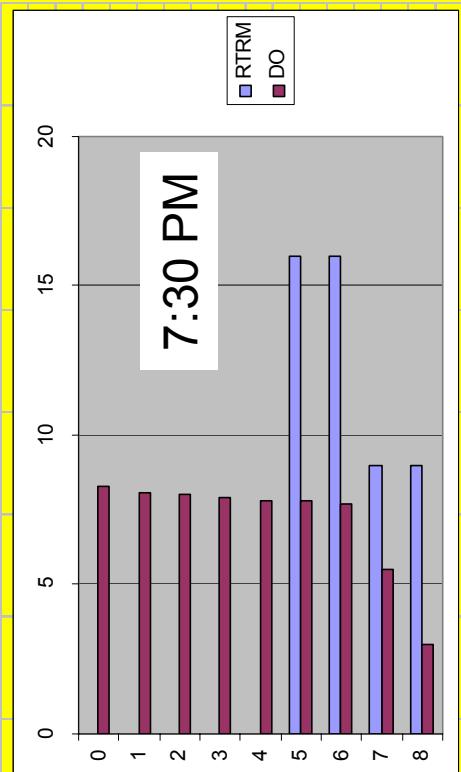
**“Season 3”**  
**Summer**

- Approx. May 15 to October 15**
- Open the Aspetuck Reservoir Gate to Hemlocks Reservoir;
  - Divert from Saugatuck Reservoir Storage at 50 ft;
  - Layer Aeration and Interflow to/thru Hemlocks Reservoir

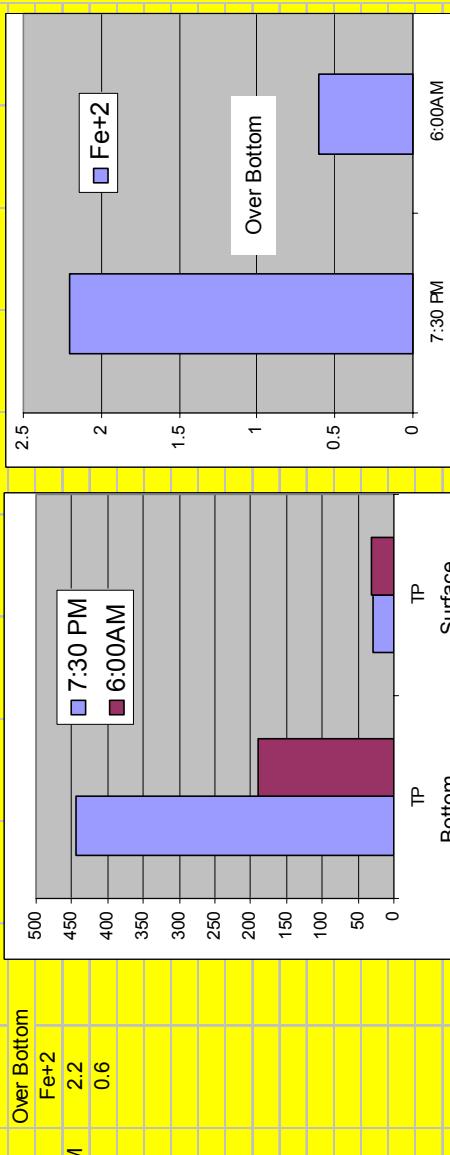
**Maximize use of Saugatuck Reservoir Water**

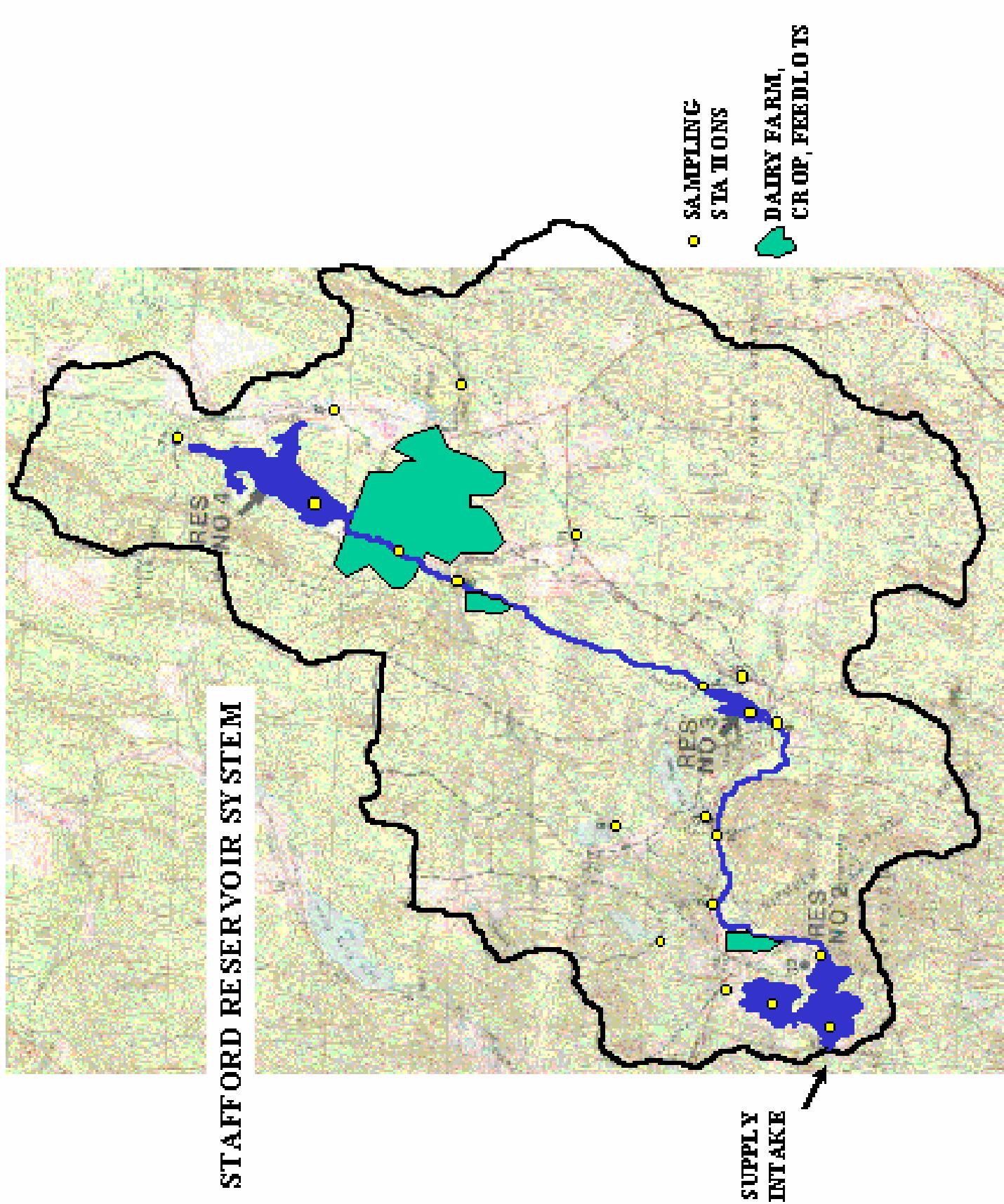
# *When is oxygen loss most intense, during day or night?*

Is there a difference during night vs. day?



Develop a hypothesis...  
...Test it...  
...and USE THE DATA!



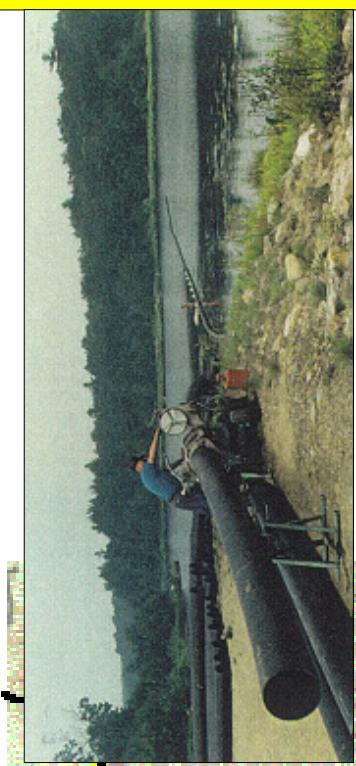


**MAXIMIZE THE USE OF THE  
TOTALLY UNDISTURBED  
WOODED WATERSHED SUBBASIN**

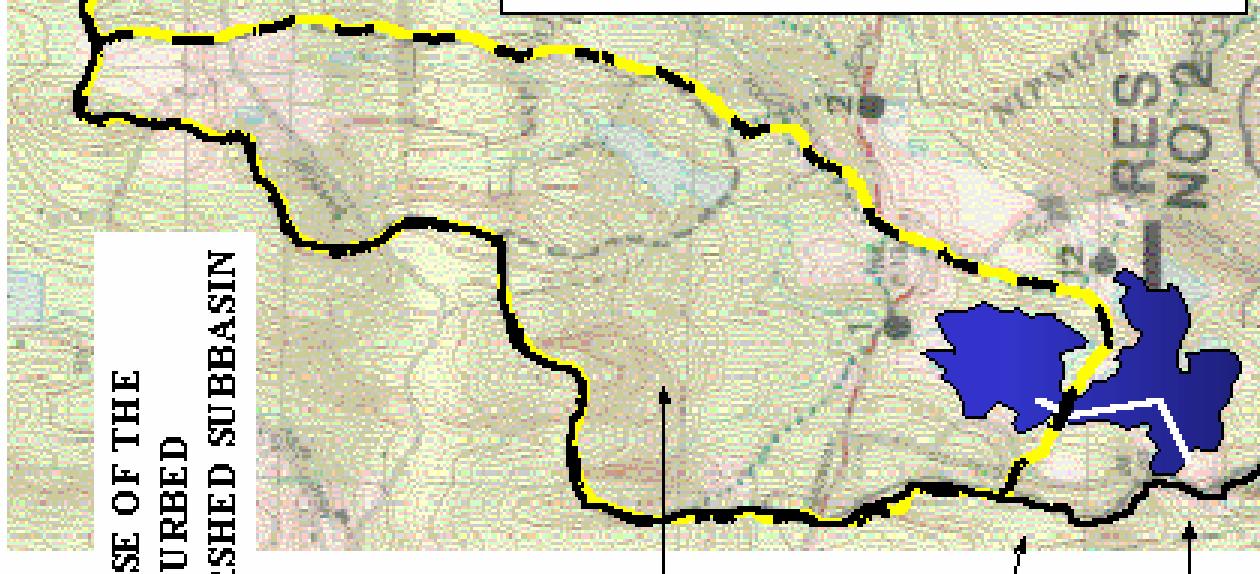
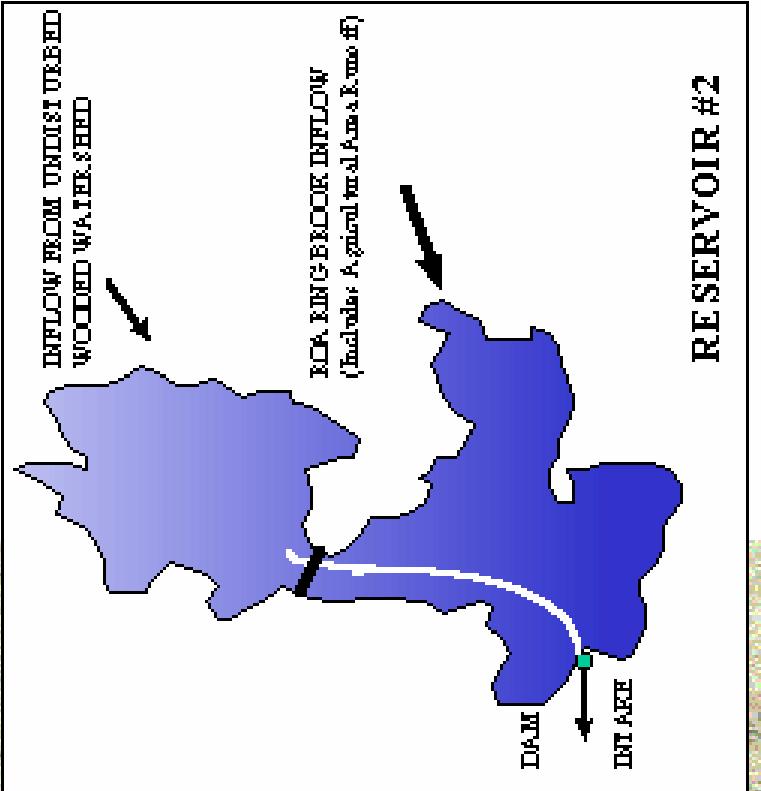
Watershed Area  
Draining Directly to  
the Northern Bay of  
the Reservoir

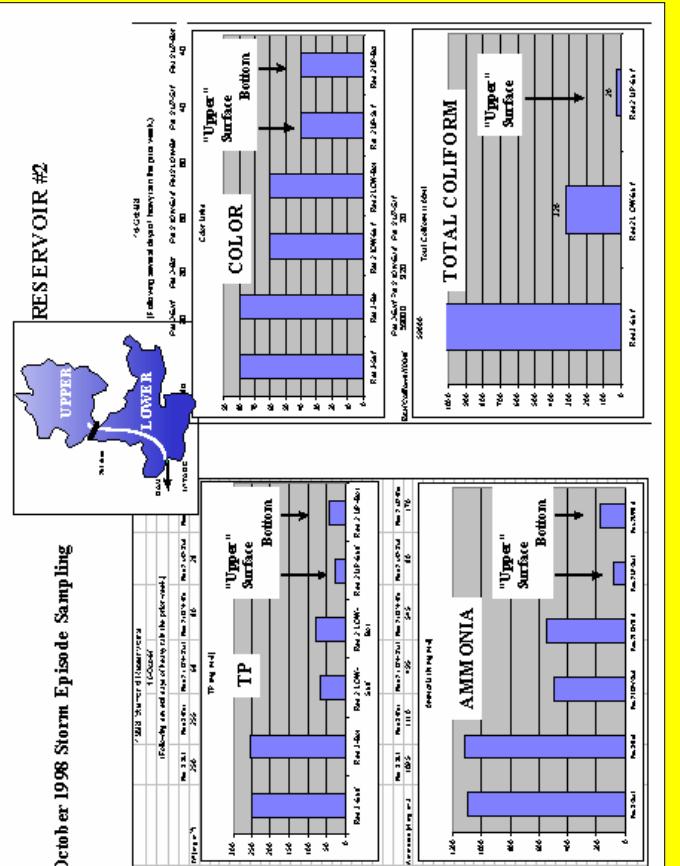
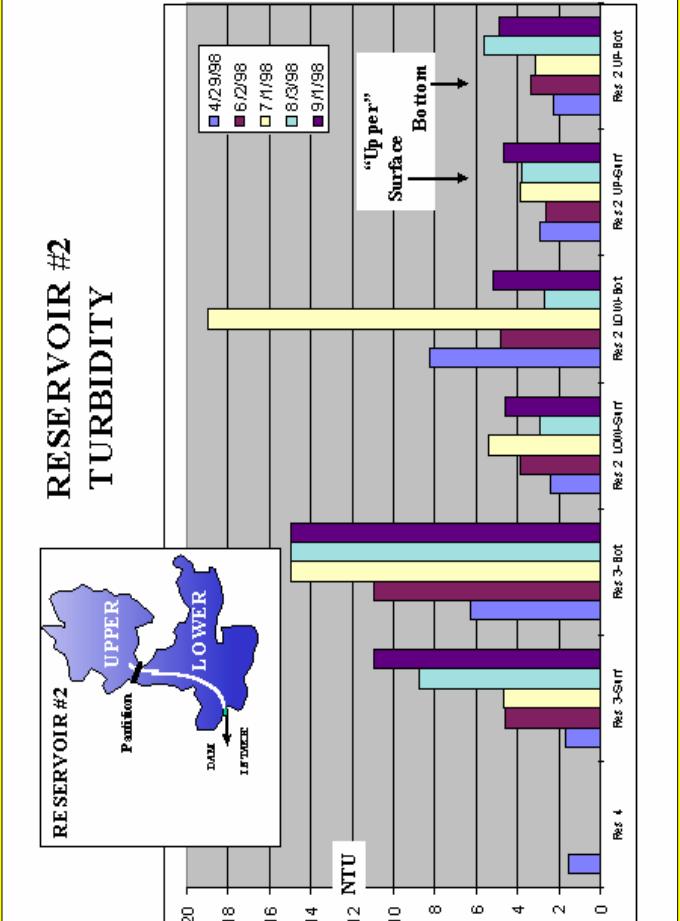
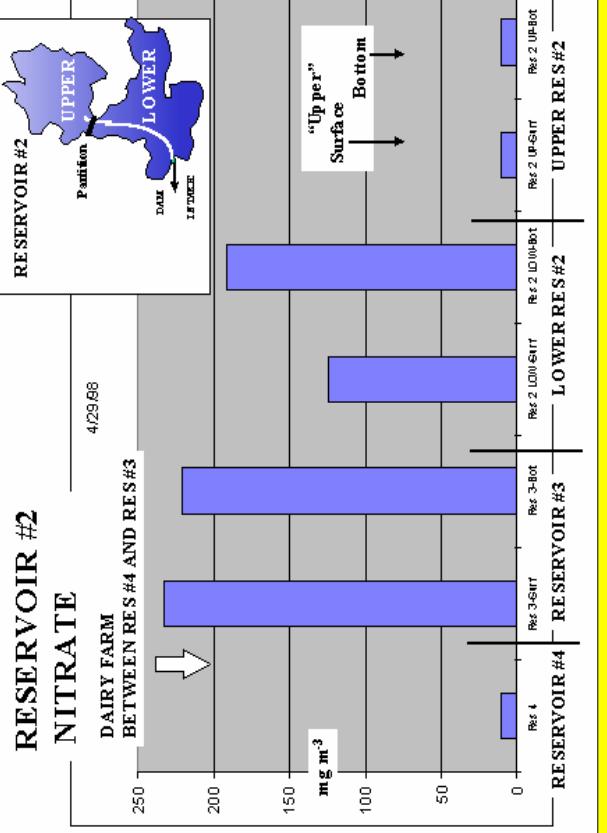
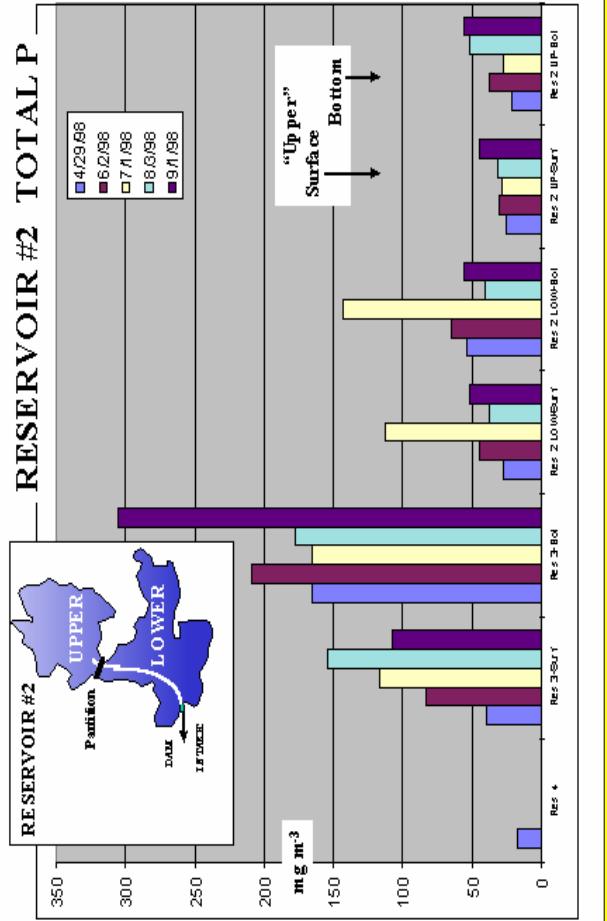
"PARTITION  
CURTAIN"

INTAKE  
EXTENSION



RESERVOIR PARTITIONING  
INTAKE ISOLATION





EXAMPLE

# Example Watershed Management Plan

- Designation of Lake Protection Areas
- Septic System Mgmt Approaches
  - 1. Pump Out Reporting Ordinance
  - 2. Effluent Filters - “Consider for Use”
  - 3. P Removal Minimums for Sand Fill
  - 4. Leaching Field Geometry
  - 5. “Ban” Septic System Additives
  - 6. Innovative Septic Designs- P Removal
- “Ban” on High-P Fertilizers
- Standard Conditions/Policies PZC, IW
  - Nutrient Allocation Evaluation
  - Road / Drainage Maint. Schedule
  - Wetland Restoration - Erdoni Basin

**EXAMPLE**

## Nutrient Allocation - Land Use Worksheet

Watershed Areal P Export Allocation = **0.32 lb P Acre<sup>-1</sup> Year<sup>-1</sup>**

### Base Loading Areal TP Allocations:

Undisturbed Woodlands=	<b>0.09</b> lb P Acre <sup>-1</sup> Year <sup>-1</sup>
Impervious Area=	<b>1.5</b> lb P Acre <sup>-1</sup> Year <sup>-1</sup>
Semi-Pervious Area=	<b>1.25</b> lb P Acre <sup>-1</sup> Year <sup>-1</sup>

Parcel Area= **67485** sq ft

Parcel Allocation=

<b>1.549242</b> Acres
<b>0.495758</b> lb P Year <sup>-1</sup>
<b>0.495758</b> lb P Year <sup>-1</sup>

**67485** Sq. Feet

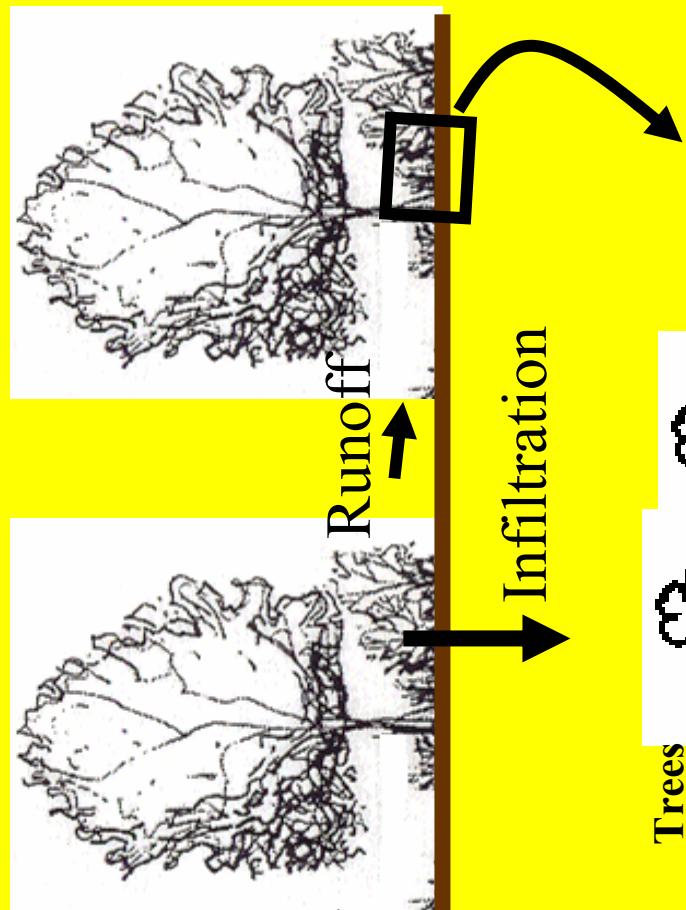
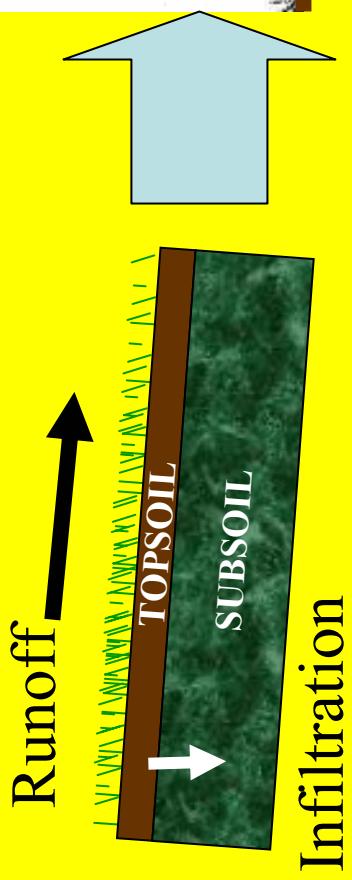
### Existing Single Family Residential Use:

sq ft	Est P Load (lb P / acre / yr)	LOAD
1860	1.5	0.06405
1530	1.5	0.052686
7000	1.25	0.200872
43120	0.09	0.089091
13975	0.75	0.240616
TOTAL		<b>0.647315</b>

**Compare:**  
Does not meet allocation.

**Need to apply Best Management Practices to comply with the Total Phosphorus Allocation:**  
Reduce disturbed area, reduce impervious areas, increase retained woodland,  
Rain Gardens, Roof Water Harvesting / Rain Barrels, "Simple" Infiltrators,

**0.20 lb  
P / Acre / Year**



## Convert Semi-Pervious to “Simulated Woodland”

- Mimic the Function of Woodlands

### Ornamental Landscaping

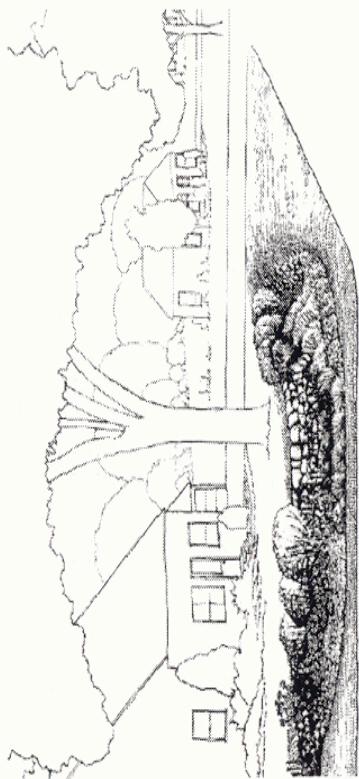
Example: Convert a Lawn



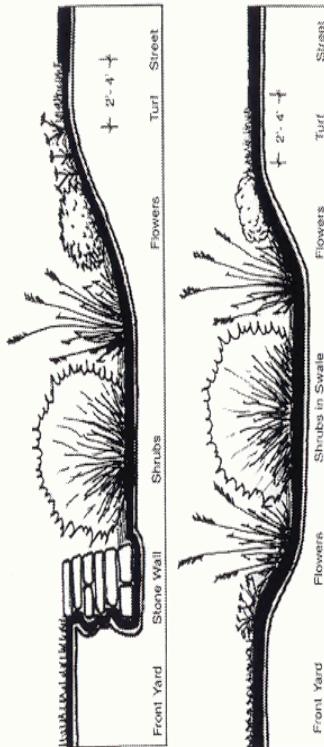
- Spread 3" of Peat Moss over the area
- Rototill to > 6 inches
- Cover with Mulch
- Plant Groundcover
- Plant Shrubs (50% cover when grown)
- Plant Tree Saplings (> 1" DBH; 30' Center Spacing)

**Figure 4-6. Residential Rain Gardens**

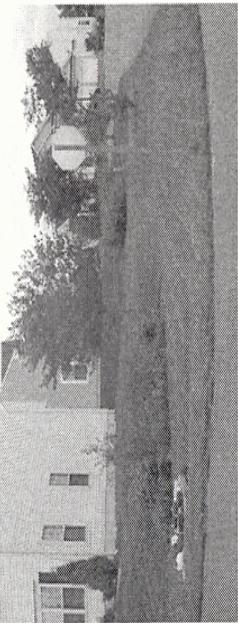
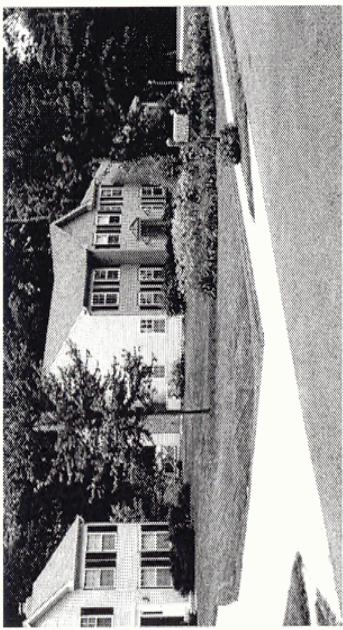
**Typical Residential Rain Garden (With and Without Masonry Wall)**



**First  $\frac{1}{2}$  Inch  
0.40 lb P / Acre / Year  
First 1 Inch  
0.20 lb P / Acre / Year**



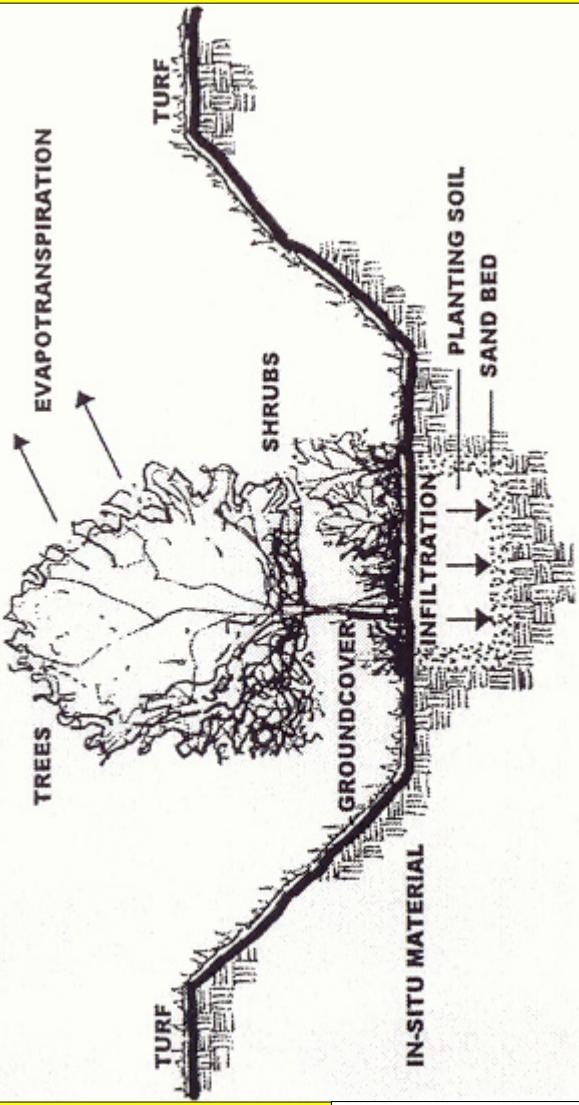
**Examples of Residential Rain Gardens**



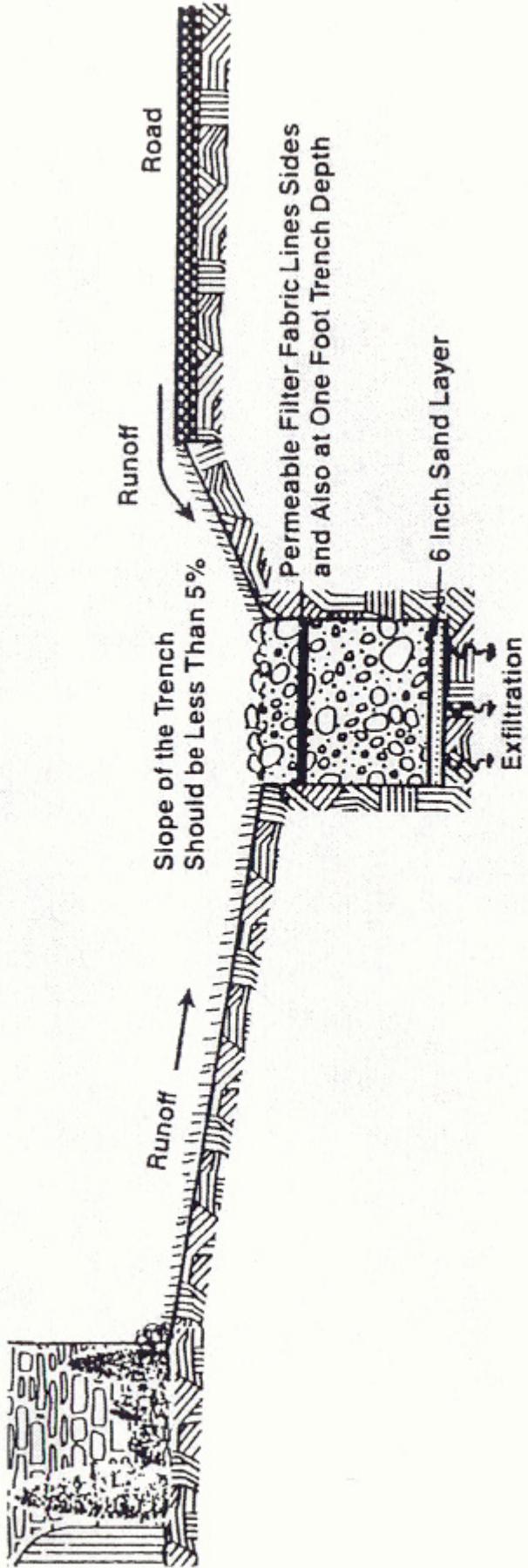
Source: Metropolitan Council, 2001 (Adapted from Nassauer et al., 1997) and Low Impact Development Center ([www.lowimpactdevelopment.org](http://www.lowimpactdevelopment.org)), 2001.

**First  $\frac{1}{2}$  Inch  
0.40 lb P / Acre / Year**

**First 1 Inch  
0.20 lb P / Acre / Year**

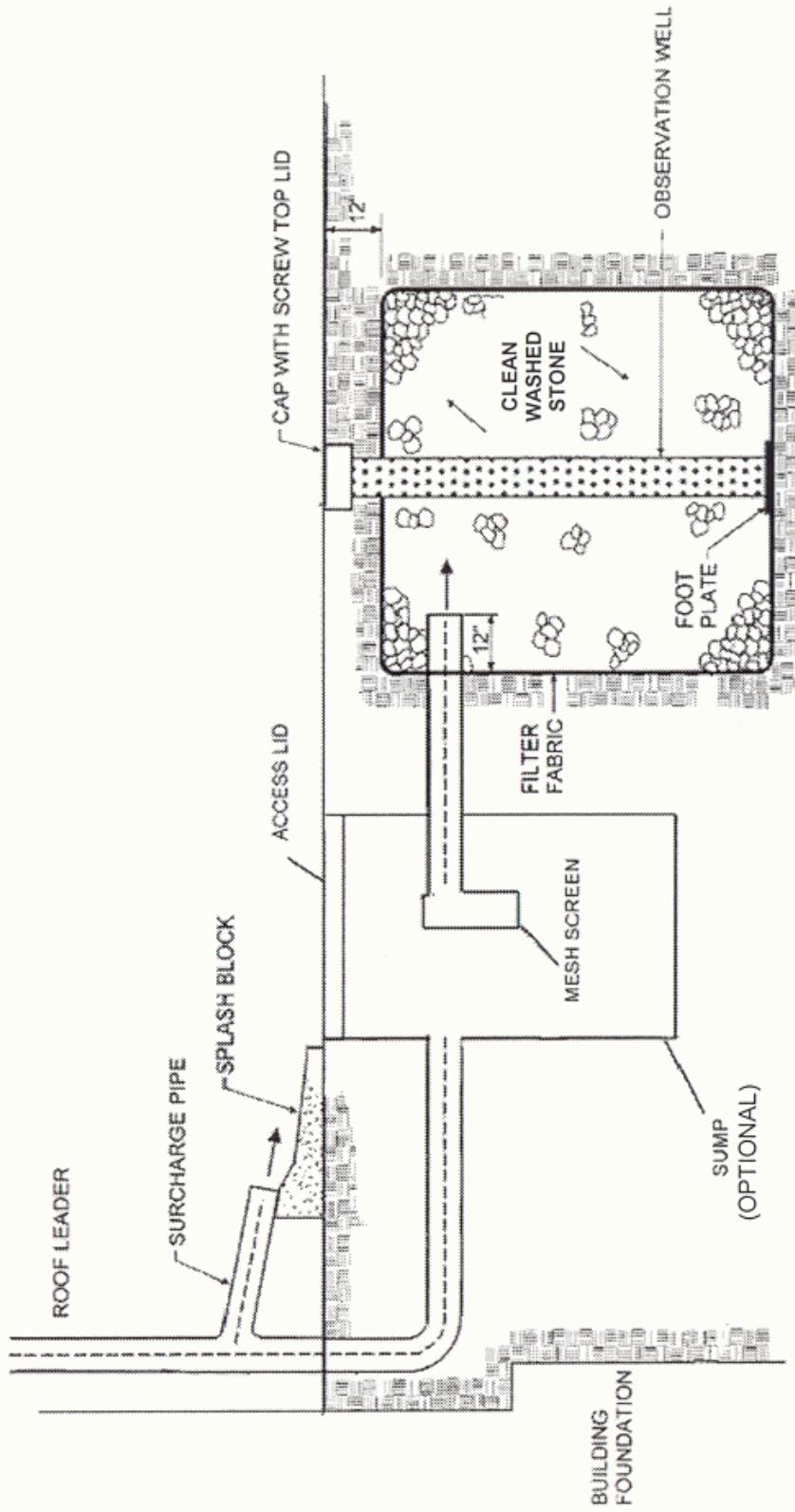


Bioretention Basin



Enhanced Swale

**Figure 4-7. Schematic of Typical Dry Well**



Source: Adapted from NYDEC, 2001.

**First  $\frac{1}{2}$  Inch**

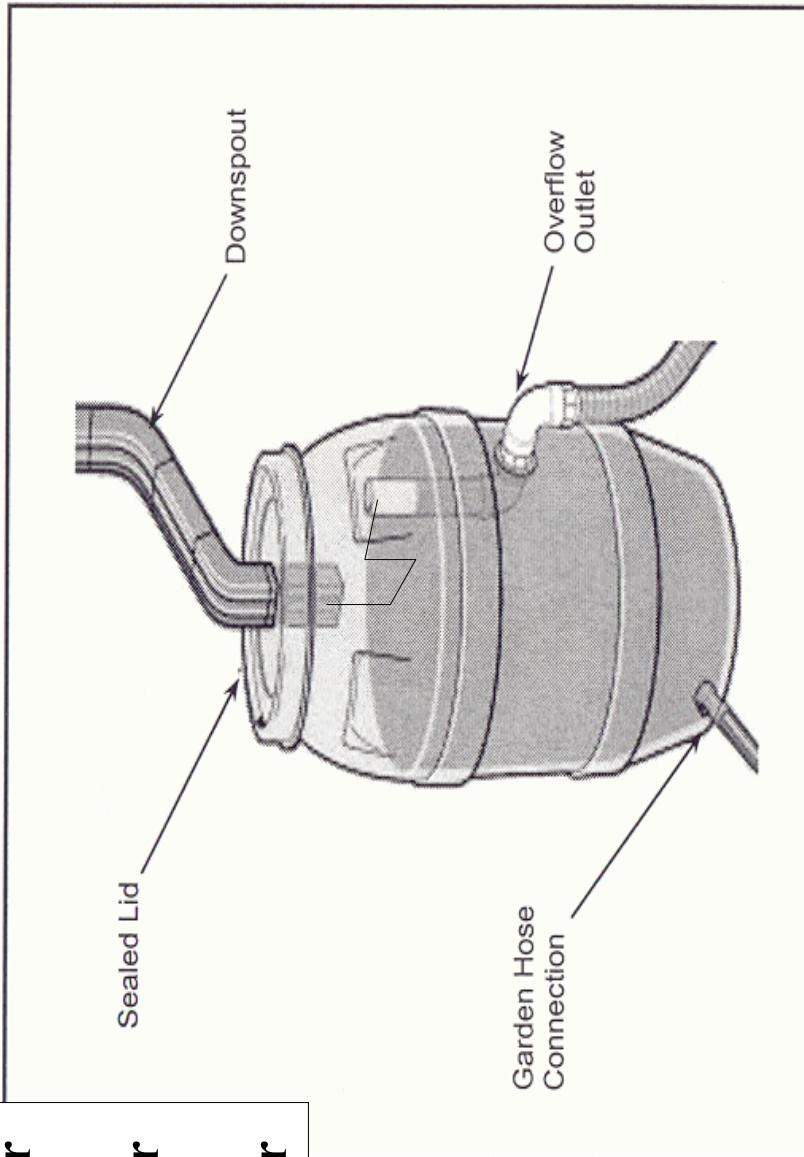
**0.40 lb P / Acre / Year**

**First 1 Inch**

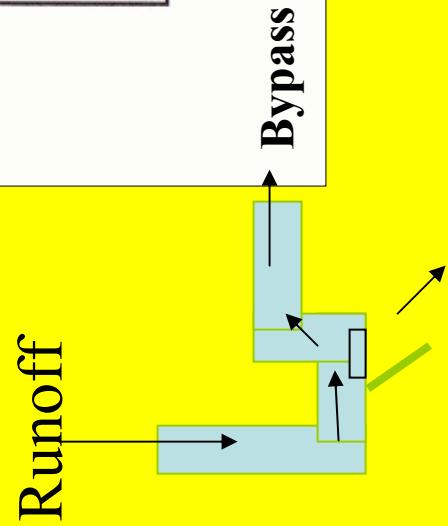
**0.20 lb P / Acre / Year**

<b>First <math>\frac{1}{4}</math> Inch</b>	<b>0.60 lb P / Acre / Year</b>
<b>First <math>\frac{1}{2}</math> Inch</b>	<b>0.40 lb P / Acre / Year</b>
<b>First 1 Inch</b>	<b>0.20 lb P / Acre / Year</b>

**Figure 4-8. Typical Rain Barrel**



Source: Adapted from [urbangardencenter.com](http://urbangardencenter.com) (D&P Industries, Inc., 2001).



**First Flush Fills Barrel**



The End  
Questions?  
Remember...you are what you ...DRINK!