

# Algorithm to Accurately Determine Free Lift and Float Altitude For Zero Pressure and Super Pressure Balloons

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## *Inputs and Givens*

An atmosphere model, 5 point or more to give  $T_{air}$ ,  $P_{air}$ , and density at altitude  $Z$ .

### Specific Gas Constants

$R_{air} := 287.1$      $R_{gas} := 2077.2$      $g := 9.807$     Acceleration of gravity, m/s/s

$M_{gas}$             Mass of lift gas, kg

$M_{gross}$            Gross mass, kg

$T_{AirLaunch}$        Ambient air temperature at launch, deg K

$P_{AirLaunch}$        Ambient air pressure at launch, Pascals

$V_{design}$            Design volume at design super pressure, m<sup>3</sup>

$\Delta P_{design}$           Design super pressure, Pa

$EA_{tendon}$           Tendon stiffness, Newtons

$N_{gore}$             Number of gores

$\Delta P_{launch}$           Initial assumed average deltaP at launch, Pa

$\Delta T_{launch}$           Assumed super temperature at launch, deg C

$\Delta T_{float}$            Assumed super temperature at float, deg C

### Note:

Superpressure, deltaP, and differential pressure are used interchangeably. It refers to the average differential pressure for the volume of lift gas. Even a zero pressure balloon has differential pressure due to the aerostatic buoyancy gradient, and this does slightly compress the lift gas density making it have less buoyancy.

### ***Launch Buoyancy, Pressure, and Free Lift***

Ambient air density at launch, kg/m<sup>3</sup>

$$\rho_{\text{AirLaunch}} := \frac{P_{\text{AirLaunch}}}{R_{\text{air}} \cdot T_{\text{AirLaunch}}}$$

Lift gas density at launch, kg/m<sup>3</sup>

$$\rho_{\text{GasLaunch}} := \frac{(P_{\text{AirLaunch}} + \Delta P_{\text{launch}})}{R_{\text{gas}} \cdot (T_{\text{AirLaunch}} + \Delta T_{\text{launch}})}$$

$$V_{\text{Launch}} := \frac{M_{\text{gas}}}{\rho_{\text{GasLaunch}}} \quad \text{Launch volume, m}^3$$

$$\text{Radius}_{\text{Launch}} := \left( \frac{3}{4} \cdot \frac{V_{\text{Launch}}}{\pi} \right)^{\frac{1}{3}} \quad \text{Spherical radius of launch bubble, m}$$

Specific buoyancy at launch, N/m<sup>3</sup>

$$b_{\text{Launch}} := g \cdot (\rho_{\text{AirLaunch}} - \rho_{\text{GasLaunch}})$$

$$\Delta P := b_{\text{Launch}} \cdot \text{Radius}_{\text{Launch}} \quad \text{average super pressure at launch, Pa due to aerostatic buoyancy pressure}$$

$$\Delta P_{\text{launch}} := \Delta P \quad \text{iterate until agreement}$$

$$GI := b_{\text{Launch}} \cdot V_{\text{Launch}} \quad \text{Gross inflation, Newtons}$$

$$\text{FreeLift}_{\text{ratio}} := \frac{GI}{g \cdot M_{\text{gross}}} \quad \text{Free Lift Ratio}$$

$$100 \cdot (\text{FreeLift}_{\text{ratio}} - 1) = \quad \% \text{ Free Lift}$$

## Conditions at Float

### Determine Un-Stretched Float Volume

Elasticity effects on the volume-----

$$A_{\text{barT}} := \frac{\Delta P_{\text{design}} \cdot V_{\text{design}}^{\frac{2}{3}}}{g \cdot M_{\text{gross}}} \quad \text{Non dimensional pressure head}$$

$$T_{\text{barT}} := -0.528 + 1.604 \cdot A_{\text{barT}} \quad \text{Non dimensional apex tension}$$

$$T_{\text{ension}} := T_{\text{barT}} \cdot g \cdot M_{\text{gross}} \quad \text{Total tension at apex, N}$$

$$\text{Strain}_{\text{design}} := \frac{T_{\text{ension}}}{N_{\text{gore}} \cdot EA_{\text{tendon}}} \quad \text{Design elastic strain}$$

$$V_o := \frac{V_{\text{design}}}{(1 + \text{Strain}_{\text{design}})^3} \quad \text{Unstretched volume}$$

### Assume an Initial Float Volume Stretch Factor

The stretch factor is defined as actual float volume/design volume

$$\text{Stretch}_{\text{factor}} := 1.0017 \quad \text{Assume an initial volume elastic stretch factor: 1.0 for a ZPB, and 1.0017 for a pumpkin}$$

### Calculate the Spherical Radius Equivalent at Float

$$\text{Radius}_{\text{float}} := \left( \frac{3}{4} \cdot \frac{V_{\text{design}}}{\pi} \right)^{\frac{1}{3}} \quad \text{Design equivalent spherical radius}$$

*Determine the Float Altitude:*

⇒ OUTER LOOP (Do until stretch factor converges)

$$\rho_{\text{system}} := \frac{M_{\text{gross}} + M_{\text{gas}}}{V_{\text{design}} \cdot \text{Stretch factor}}$$

$$\rho_{\text{air}} := \rho_{\text{system}} \quad \text{System density at float = air density at float}$$

$$Z_{\text{float}} := -6317.12 \cdot \ln\left(\frac{\rho_{\text{air}}}{2.077608}\right) \quad \text{Approximate float altitude, m}$$

$$\frac{d}{dz} \rho_{\text{air}} = -\rho_{\text{air}} \cdot 0.0001175 \quad \text{Approximate density gradient, kg/m}^3 \text{ per meter}$$

Call on Atmosphere Model for  $T_{\text{air}}$  and  $P_{\text{air}}$  at approximate altitude  $Z_{\text{float}}$

$\Delta P_{\text{float}} := 2$  Initial guess for ZPB average buoyancy pressure, Pa

⇒ INNER LOOP (loop 25 times to converge altitude)

If the balloon is a ZPB, then

$$\rho_{\text{gas}} = \frac{(P_{\text{air}} + \Delta P_{\text{float}})}{R_{\text{gas}} \cdot (T_{\text{air}} + \Delta T_{\text{float}})} \quad \text{Gas density at float}$$

$$M_{\text{gasFloat}} := \rho_{\text{gas}} \cdot V_{\text{design}} \quad \text{Mass of lift gas}$$

$$\rho_{\text{system}} := \frac{M_{\text{gross}} + M_{\text{gasFloat}}}{V_{\text{design}}} \quad \text{System density with float gas mass}$$

$$b_{\text{float}} := g \cdot (\rho_{\text{air}} - \rho_{\text{gas}}) \quad \text{Specific buoyancy at float}$$

$$\Delta P_{\text{float}} := 0.834 \cdot \text{Radius}_{\text{float}} \cdot b_{\text{float}} \quad \text{Better estimate on average buoyancy pressure}$$

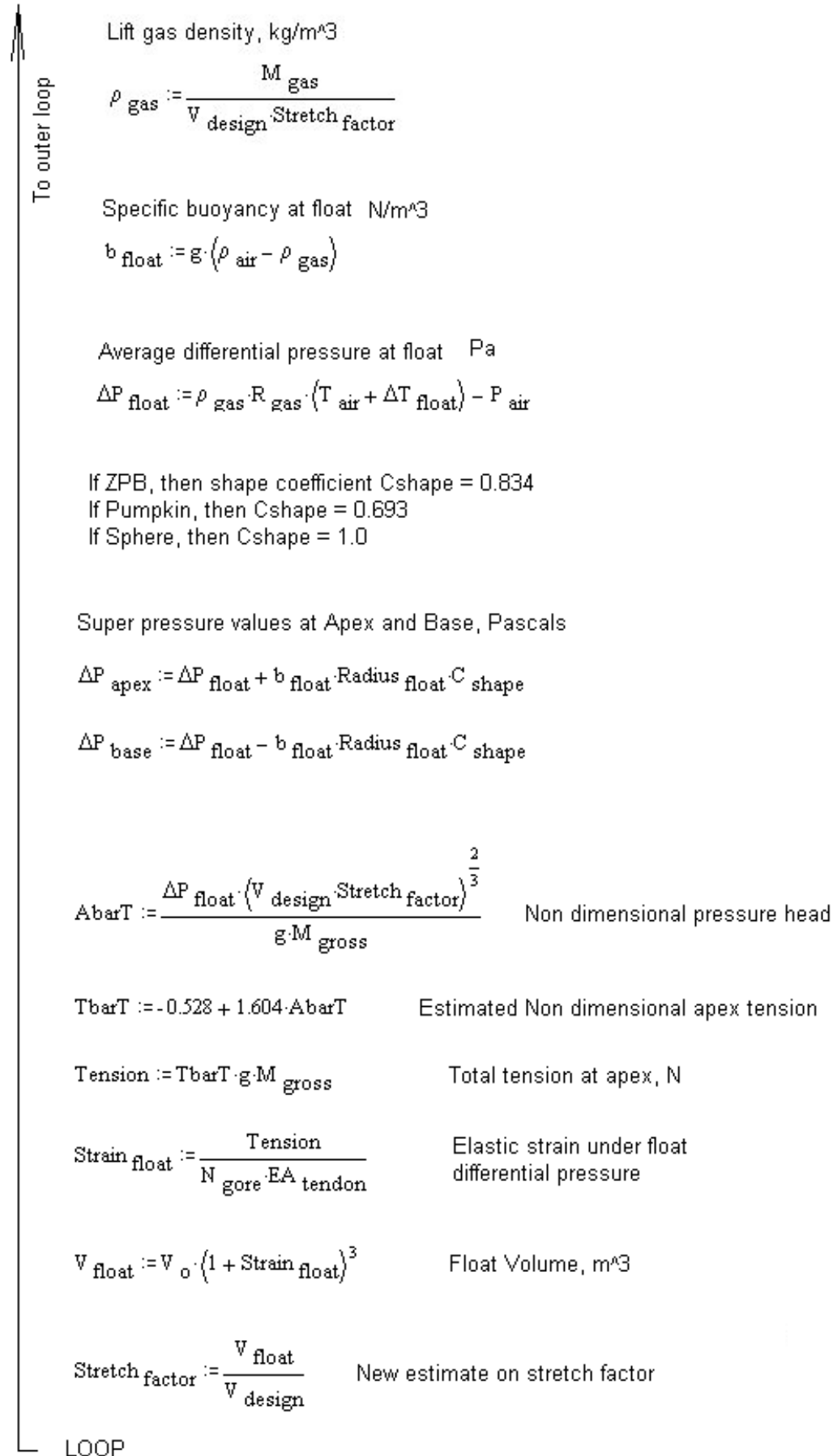
Call on Atmosphere Model for  $T_{\text{air}}$  and  $P_{\text{air}}$  and  $\rho_{\text{air}}$  at new altitude  $Z_{\text{float}}$

$$\Delta Z := \frac{(\rho_{\text{system}} - \rho_{\text{air}})}{\frac{d}{dz} \rho_{\text{air}}} \quad \text{Estimate of change of altitude needed}$$

$$Z_{\text{float}} = Z_{\text{float}} + \Delta Z \quad \text{New float altitude estimate}$$

LOOP

Continue with the Loop to Determine Differential Pressure and Elastic Stretch...



## *The Gas Mass Calculator*

FreeLift ratio = Gross Inflation / Gross Weight

Here are the formulas to determine the actual free lift ratio, and one based on the standard assumption of no super pressure or super temperature:

$$\text{FreeLift}_{\text{ratio}} = \frac{M_{\text{gas}}}{M_{\text{gross}}} \cdot \left[ \frac{\left(1 + \frac{\Delta T}{T}\right) R_{\text{gas}}}{\left(1 + \frac{\Delta P}{P}\right) R_{\text{air}}} - 1 \right]$$

Free lift fraction expression as a function of ambient temperature T and pressure P, differential temperature dT and differential pressure dP, the mass of the gas and gross mass

$$M_{\text{gas}} = M_{\text{gross}} \cdot \frac{\text{FreeLift}_{\text{ratio}}}{\left( \frac{R_{\text{gas}}}{R_{\text{air}}} - 1 \right)}$$

Gas mass for a standard assumption of no super temperature and no super pressure

## The Supply Tank Calculator

First, since under high pressure gases don't quite follow the ideal gas law, we need a correction factor. Such a correction factor is known as the Z factor, and this graph of Z factors describes the corrections for helium, oxygen, and air.

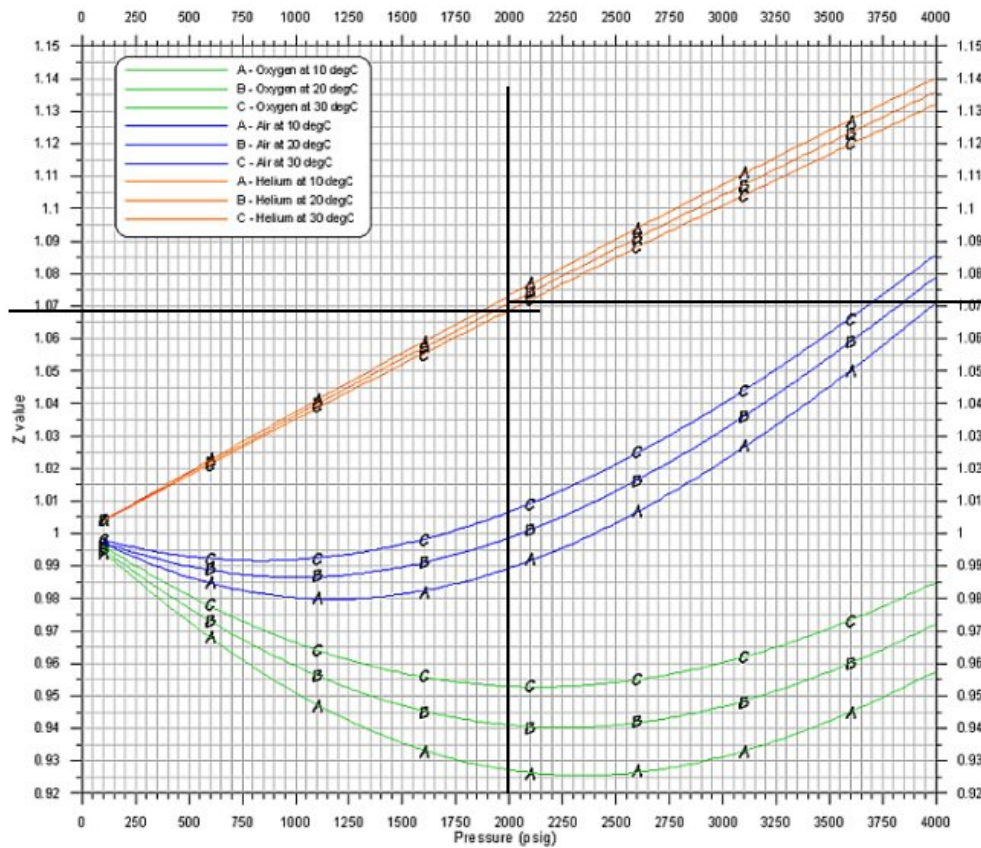
Recall, the Z factor accounts for differences with the ideal gas law:

$$\text{Pressure } P = \rho \cdot R_{\text{gas}} \cdot T \cdot Z_{\text{factor}} \quad \text{Pascals}$$

$\rho$  is the density,  $\text{kg/m}^3$

$R_{\text{gas}}$  is the specific gas constant = 2077.2 for helium

T is the absolute temperature, deg K



If you know the initial tank temperature and pressure, and final temperature and pressure:

$V$  is the individual tank volume,  $m^3$

$N_{\text{tank}}$  is the number of gas bottles being used

The total trailer tank volume is referred as the “water volume”, so the individual tank volume would simply be the trailer water volume divided by the number of tanks. For a PraxAir T-73 12 cylinder trailer, the individual volume = 78.592 cubic feet.

$P_1$  Initial absolute pressure, Pa

$T_1$  Initial temperature, deg K

$Z_1$  Z factor for the initial condition

$P_2$  Final absolute pressure, Pa

$T_2$  Final temperature, deg K

$Z_2$  Z factor for the final condition

+

$$\rho_1 = \frac{P_1}{R_{\text{gas}} \cdot T_1 \cdot Z_1} \quad \text{Initial gas density in the tank, kg/m}^3$$

$$M_1 = \rho_1 \cdot V \quad \text{Initial gas mass in the individual supply tank, kg}$$

$$\rho_2 = \rho_1 \cdot \frac{P_2}{P_1} \cdot \frac{T_1}{T_2} \cdot \frac{Z_1}{Z_2} \quad \text{Final gas density in the tank, kg/m}^3$$

$$M_2 = \rho_2 \cdot V \quad \text{Final gas mass in the individual supply tank, kg}$$

$$M_{\text{gas}} = N_{\text{tank}} \cdot (M_1 - M_2) \quad \text{Total lift gas mass delivered, kg}$$

Conversion from PSIG to Pascals: Pascals = 6895\* (PSIG+14.7)

Conversion from Pascals to PSIG: PSIG = Pascals/6895 - 14.7



If you know the initial tank temperature and pressure, and gas mass required:

$P_1$  Initial absolute pressure, Pa

$T_1$  Initial temperature, deg K

$Z_1$  Z factor for the initial condition

$M_{gas}$  The total lift gas mass required, kg

$T2T1_{ratio}$  The assumed ratio of final temperature to initial temperature, deg K/deg K

$$M_1 = \frac{P_1 \cdot V}{R_{gas} \cdot T_1 \cdot Z_1} \quad \text{Initial mass of lift gas in a single tank, kg}$$

$$M_2 = M_1 - \frac{M_{gas}}{N_{tank}} \quad \text{Final mass of lift gas in a single tank, kg}$$

$$\rho_1 = \frac{M_1}{V} \quad \text{Initial gas density in the tank, kg/m}^3$$

$$\rho_2 = \frac{M_2}{V} \quad \text{Final gas density in the tank, kg/m}^3$$

$$T_2 = T2T1_{ratio} \cdot T_1 \quad \text{Final gas temperature in the tank, deg K}$$

→ LOOP 25 times to converge the final Z factor

$$P_2 = \rho_2 \cdot R_{gas} \cdot T_2 \cdot Z_2$$

Call subroutine to determine Z as function (P,T)

Loop

$P_2$  is the desired cut-off pressure, Pascals

Divide by 6895 then subtract 14.7 to get gauge pressure PSIG